Supporting Information

for

Tetrathiafulvalene-based azine ligands for anion and metal cation coordination

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Additional analytical data

Figure S1: Partial crystal packing of ligand **L1** with columns of stacked head to tail molecules that are connected laterally through S…O heteroatom contacts.



Figure S2: Electronic delocalization scheme for ligands L1 and L2.



Figure S3: HOMO-LUMO Frontier orbitals representation for compounds 1 and 2.



Figure S4: Color change of ligand L1 upon addition of inorganic anions.



Figure S5: Color change of ligand L2 upon addition of inorganic anions.



Figure S6: UV-visible absorption spectra of L1 upon addition of 1 equiv of anion.



Figure S7: UV-visible absorption spectra of L2 upon addition of 1 equiv of anion.



Figure S8: Cyclic voltammograms of ligands L1 (left) and L2 (right) (2 × 10⁻⁵ M) without (black, bold lines first cycle and dashed lines for second cycle) and with (red, bold lines first cycle and dashed lines for second cycle) 2 equiv of F⁻. CVs recorded at 100 mV·s⁻¹ on a glassy carbon electrode in CH₂Cl₂/CH₃CN (9/1, v/v) with *n*-

Bu₄NPF₆ (0.1 M).



Figure S9: ¹H NMR spectra of ligand L1 (4×10^{-3} M in DMSO-*d*₆) upon addition of successive aliquots of TBAF (DMSO-*d*₆).



Figure S10: UV–visible absorption spectra of complex **3** (c 1.1 × 10⁻⁴ M in dichloromethane/acetonitrile, 9/1, (v/v)), room temperature. Ligand **L2** is added for comparison.



Figure S11: Cyclic voltammograms of **L2** (red, added for comparison) and complex **3** (black) (1.1 × 10^{-4} M) at 100 mVs⁻¹ on a glassy carbon electrode in CH₂Cl₂/CH₃CN (9/1, v/v) with *n*-Bu₄NPF₆ (0.1 M).