### Supporting Information hat

# Efficient mechanochemical synthesis of regioselective persubstituted cyclodextrins

"Ncu| nq"Lleukpu| nf(,."Octkpc"Ecrqtcuq."Mcvkc"Octvkpc."Gocpwgnc"Ecnekq"Icwfkpq"cpf""Ikcpectnq"Etcxqwq,"

Department of Drug Science and Technology and NIS - Centre for Nanostructured Interfaces and Surfaces, University of Turin, Via P. Giuria 9, 10125 Turin (Italy);

Email: Ncu| m/Ikeukpu| m("/"ljicsinszky@gmail.com="I kcpectm/Etcxqwq"/"giancarlo.cravotto@unito.it

, Eqttgur qpf kpi "cwj qt"

""Gzr gtko gpvcn'f gvcku'cpf "ej ctcevgtk cvkqp"qh'vj g'r tgr ctgf "eqo r qwpf u"

"

..

This *Supporting Information* file contains the experimental details of the syntheses and IR/NMR and MS characterization (**3a**, **3b**, **5a**, **5b** only) of the prepared compounds.

#### **Table of Contents Reaction Scheme S**3 **Experimental Details S**3 Instrumentation and reagents **S**3 Syntheses in solutions **S**4 High Energy Ball Milling reactions **S**7 **IR** spectra Table S4 Characteristic IR absortion bands (cm<sup>-1</sup>) of the prepared compounds S14 S15 Fig. S5 IR spectrum of heptakis(6-deoxy-6-iodo)-βCD 2a Fig. S6 IR spectrum of heptakis(6-deoxy-6-chloro)-βCD 2a' S15 Fig. S7 IR spectrum of octakis(6-deoxy-6-iodo)-γCD 2b S16 Fig. S8 IR spectrum of octakis(6-bromo-6-deoxy)-γCD 2b' S16 S17 Fig. S9 IR spectrum of heptakis(6-deoxy-6-S-thiuronium)-BCD iodide 3a Fig. S10 IR spectrum of octakis(6-deoxy-6-S-thiuronium)-yCD iodide 3b S17 Fig. S11 IR spectrum of octakis(6-deoxy-6-S-thiuronium)-yCD bromide 3b' **S18** S18 Fig. S12 IR spectrum of heptakis(6-azido-6-deoxy)-βCD 4a Fig. S13 IR spectrum of octakis(6-azido-6-deoxy)- $\gamma$ CD 4b S19 Fig. S14 IR spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl))-βCD NH4<sup>+</sup> 5a S19 Fig. S15 IR spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl))-γCD NH4<sup>+</sup> 5b S20 S20 Fig. S16 IR spectrum of heptakis(6-deoxy-6-S-(1-dodecylthio))-βCD 6 **MS** spectra S21 Fig. S17 ESI-MS spectrum of heptakis(6-deoxy-6-S-thioureido)-βCD 3a Fig. S18 ESI-MS spectrum of octakis(6-deoxy-6-S-thioureido)-yCD 3b S21 Fig. S19 ESI-MS spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl)-βCD 5a S21 Fig. S20 ESI-MS spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl)-yCD 5b S21 NMR spectra Table S5 Proton NMR assignment (ppm) of compounds (from 1H and HSQC spectra) S22 Table S6 Carbon NMR assignment (ppm) of compounds (from HSQC spectra) S23

Fig. S21 1H-NMR spectrum of heptakis(6-deoxy-6-iodo)-βCD 2a	S24
Fig. S22 HSQC-NMR spectrum of heptakis(6-deoxy-6-iodo)-βCD 2a	S24
Fig. S24 1H-NMR spectrum of heptakis(6-deoxy-6-chloro)-βCD 2a'	S25
Fig. S25 HSQC-NMR spectrum of heptakis(6-deoxy-6-chloro)-βCD 2a'	S25
Fig. S25 1H-NMR spectrum of octakis(6-deoxy-6-iodo)-γCD 2b	S26
Fig. S26 HSQC-NMR spectrum of octakis(6-deoxy-6-iodo)-γCD 2b	S26
Fig. S27 1H-NMR spectrum of octakis(6-bromo-6-deoxy)-γCD 2b'	S26
Fig. S28 HSQC-NMR spectrum of octakis(6-bromo-6-deoxy)-γCD 2b'	S27
Fig. S29 1H-NMR spectrum of heptakis(6-deoxy-6-thioureido)-βCD iodide <b>3a</b> , thiourea/I	S28
ratio=3.5	
Fig. S30 HSQC-NMR spectrum of heptakis(6-deoxy-6-thioureido)-βCD iodide <b>3a</b> , thiourea/I	S28
ratio=3.5	
Fig. S31 1H-NMR spectrum of octakis(6-deoxy-6-thioureido)-γCD iodide <b>3b</b> , thiourea/I	S29
ratio=1.5	
Fig. S32 HSQC-NMR spectrum of octakis(6-deoxy-6-thioureido)-γCD <b>3b</b> , thiourea/I	S29
ratio=1.5	
Fig. S33 1H-NMR spectrum of octakis(6-deoxy-6-thioureido)-γCD iodide <b>3b</b> , thiourea/I	<b>S</b> 30
ratio=3.5	
Fig. S34 HSQC-NMR spectrum of octakis(6-deoxy-6-thioureido)-γCD iodide <b>3b</b> , thiourea/I	S30
ratio=3.5	~ ~ .
Fig. S35 1H-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD bromide <b>3b'</b> , thiourea/I	<b>S</b> 31
ratio=3.5	~ ~ .
Fig. S36 HSQC-NMR spectrum of octakis(6-deoxy-6-thioureido)-γCD bromide <b>3b'</b> ,	\$31
thiourea/I ratio=3.5	<b>G 2 2</b>
Fig. S37 1H-NMR spectrum of heptakis(6-azido-6-deoxy)-βCD 4a	S32
Fig. S38 HSQC-NMR spectrum of heptakis(6-azido-6-deoxy)-βCD 4a	S32
Fig. S39 1H-NMR spectrum of octakis(6-azido-6-deoxy)-γCD <b>4b</b>	<b>S</b> 33
Fig. S40 HSQC-NMR spectrum of octakis(6-azido-6-deoxy)-γCD <b>4b</b>	S33
Fig. S41 1H-NMR spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl)-βCD NH4 <sup>+</sup>	S34
5a	
Fig. S42 HSQC-NMR spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl)-βCD NH4 <sup>+</sup>	S34
5a	~ ~ ~
Fig. S43 1H-NMR spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl)-γCD NH4 <sup>+</sup> <b>5b</b>	S35
Fig. S44 HSQC-NMR spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl)-γCD NH4 <sup>+</sup>	S35
5b	<b>a a a</b>
Fig. S45 1H-NMR spectrum of heptakis(6-deoxy-6-S-(1-dodecylthio)-βCD 6	S36
Fig. S46 HSQC-NMR spectrum of heptakis(6-deoxy-6-S-(1-dodecylthio)-βCD 6	<b>S</b> 36
	627

S37

## **Reaction Scheme**



For 2a' TsCl, for 2b' Br2 and N-methylpirrolidone was used instead of I2 and DMF

## **Experimental Details**

Instrumentation and reagents

**Ball mill**: Retsch PM100 High Speed Planetary Ball Mill, 1500 steel balls of 1 mm diameter (44.94 g) and 50 steel balls of 5 mm diameter (25.54 g, total weight of balls=70.5 g, V=15 mL), in a jar of 50 ml, sun wheel speed 650 min-1 for 120 min, weight = 780 g (jar, cap, and balls). Reagents other than cyclodextrins and solvents were purchased from AlfaAesar and Sigma-Aldrich. Cyclodextrins were the generous gift of Roquette, France

**Thermometer**: Lafayette TRI-88 no-contact thermometer, built-in laser pointer, with  $\pm 2$  °C reading accuracy, distance to spot size = 8:1, measuring distance 18-23 cm. The measurement matrix formed "a five on a die", two measurements were made at each point and the values were averaged.

**NMR**. 1H and HSQC-DEPT spectra were recorded at 20 °C with a Bruker Avance 300 MH at 300.13 MHz and 75.47 MHz, respectively. The 1H NMR spectra were obtained by using standard pulse programs from Bruker library using 16 K data points of 64 transients in a 4496.40 Hz spectral width.

The HSQC-DEPT experiments were acquired by pulse sequence *hsqcedtpg* from Bruker library using 128x64 data points of 32 transients. **2a'** NMR analysis were performed on a Varian System 600 MHz spectrometer. Solvent signals used for referencing. NMR data were processed with ACD/NMR Processor Academic Edition Release 12.00 product version 12.01 (Build 39104), Advanced Chemistry Development Inc. (www.acdlabs.com).

IR spectra were recorded on a PerkinElmer 1005 reflection IR spectrometer in KBr matrix, except2a' which was recorded on a Nicolet IR spectrometer.

Melting points were measured with Büchi 545 and values are uncorrected.

TLC: TLC experiments used Merck 5554 Silicagel 60, saturated chamber, regular running distance was 7 cm. Samples were dissolved in DMF or water at 10% concentration.

Syntheses in solutions

#### Synthesis of per-6-halogenated $\beta$ - and $\gamma$ CD derivatives 2a, 2a', 2b, 2b':

The syntheses of per-6-iodo- $\beta$ - and - $\gamma$ -CD, I<sub>7</sub> $\beta$ CD, (**2a**) and I<sub>8</sub> $\gamma$ CD (**2b**), were performed using a small modification to the known method [1], from freshly dried  $\beta$ CD (**1a**) and  $\gamma$ CD (**1b**) on a 0.01 mol scale with triphenylphosphine (TPP, 0.15 and 0.17 mol, respectively) and iodine (0.15 and 0.17 mol, respectively) in DMF (115-130 mL). Yields: **2a** 16.4 g, 86% and **2b** 17.1 g, 79%. Per-6-bromo- $\gamma$ CD (**2b**') was prepared in *N*-methylpyrrolidone by the same method using bromine. The prepared compounds contained < 1 % TPP and approx. 1 mol of DMF by IR/NMR.

Per-6-chloro- $\beta$ CD (**2a'**) was synthesized as per-6-iodo-CDs using p-toluenesulfonyl chloride [2]. The prepared compounds contained < 0.3 % TPP, < 5 % p-toluenesulfonic acid related compounds, and approx. 1 mol of DMF.

**2a**  $R_f$ =0.61-0.64 (max. 25 µg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); m.p. 205-208 °C [dec]; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, 20°C):  $\delta$ = 5.03 (d, <sup>2</sup>J<sub>1,2</sub>(H,H)=2.4 Hz, 7H; H-1), 3.33-3.50 (m, 7H; H-2), 3.56-3.73 (m, 7H; H-3), 3.25-3.41 (m, 7H; H-4), 3.65-3.76 (m, 7H; H-5), 3.77-3.92, 3.40-3.55 ppm (m, 14H; H-6a,b); <sup>13</sup>C NMR (from HSQC, 75 MHz, DMSO-*d*<sub>6</sub>, 25°C):  $\delta$ =103.23 (C-1), 73.14 (C-2), 72.53 (C-3), 87.11 (C-4), 73.75 (C-5), 9.68, 9.91 ppm (C-6), IR (KBr), cm<sup>-1</sup>: 2912-3032 (C-H), 1658 (O-C-O).

**2a'** R<sub>f</sub>=0.73-0.77 (max. 25 μg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); m.p. 207-210 °C [dec]; <sup>1</sup>H NMR (600 MHz, DMSO-*d*<sub>6</sub>, 25°C): δ= 4.96 (d, <sup>2</sup>J<sub>1,2</sub>(H,H)=3.4 Hz, 7H; H-1), 3.32-3.46 (m, 7H; H-2), 3.55-3.69 (m, 7H; H-3), 3.29-3.46 (m, 7H; H-4), 3.76-3.91 (m, 7H; H-5), 4.02-4.13, 3.73-3.85 ppm (m, 14H; H-6a,b); <sup>13</sup>C NMR (from HSQC, 100 MHz, DMSO-*d*<sub>6</sub>, 25°C): δ=101.94 (C-1), 71.85 (C-2), 72.37 (C-3), 83.47 (C-4), 71.03 (C-5), 44.73, 44.78 ppm (C-6), IR (KBr), cm<sup>-1</sup>: 2884-2925 (C-H), 1675 (O-C-O).

**2b** R<sub>f</sub>=0.69-0.73 (max. 25 µg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); m.p. 215-216 °C [dec]; <sup>1</sup>H NMR (300 MHz, DMSO- $d_6$ , 20°C):  $\delta$ = 5.07 (d, <sup>2</sup>J<sub>1,2</sub>(H,H)=3.3 Hz, 8H; H-1), 3.35-3.48 (m, 8H; H-2), 3.57-3.72 (m, 8H; H-3), 3.25-3.38 (m, 8H; H-4), 3.57-3.72 (m, 8H; H-5), 3.86,3.45 ppm (m, 16H; H-6a,b); <sup>13</sup>C NMR (from HSQC, 75 MHz, DMSO- $d_6$ , 25°C):  $\delta$ =102.22 (C-1), 72.8 (C-2), 71.99 (C-3), 85.73 (C-4), 71.99 (C-5), 9.75, 9.87 ppm (C-6), IR (KBr), cm<sup>-1</sup>: 2905-3032 (C-H), 1658 (O-C-O).

**2b'** R<sub>f</sub>=0.72-0.75 (max. 50 μg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); m.p. 235-237 °C [dec]; <sup>1</sup>H NMR (300 MHz, DMSO-*d*<sub>6</sub>, 20°C): δ= 5.06 (d, <sup>2</sup>J<sub>1,2</sub>(H,H)=3.3 Hz, 8H; H-1), 3.26-3.53 (m, 8H; H-2), 3.57-3.75 (m, 8H; H-3), 3.22-3.58 (m, 8H; H-4), 3.72-3.97 (m, 8H; H-5), 3.85-4.17, 3.52-3.85 ppm (m, 16H; H-6a,b); <sup>13</sup>C NMR (from HSQC, 75 MHz, DMSO-*d*<sub>6</sub>, 25°C): δ=102.11 (C-1), 72.77 (C-2), 72.47 (C-3), 84.47 (C-4), 71.82 (C-5), 35.22 ppm (C-6), IR (KBr), cm<sup>-1</sup>: 2832-3038 (C-H), 1654 (O-C-O).

#### Preparation of 3a/b-6 using the classic method:

3a and 3b. Per-6-S-thiouronium-CDs were prepared according to the known method [3].

**4a** and **4b**. The known method [4] for the preparation of per-6-azido-CDs ( $\beta$ : 9.5 g,  $\gamma$ : 10.9 g, 0.0075 mol) was modified considerably: the required amount of sodium azide ( $\beta$ : 2.8 g, 0.044 mol;  $\gamma$ : 3.3 g, 0.050 mol; 1.25 eq. to CH<sub>2</sub>-I) was divided into 5 equal parts, the first portion was dissolved in DMF (150 mL) together with freshly dried per-6-halogeno-CDs and the solution was heated to 100 °C in 30-45 min. The subsequent NaN<sub>3</sub> portions were added at 50 °C, 75 °C, 85 °C and 95 °C at a rate that would keep the reaction mixture homogenous. The reaction mixture was stirred for a further 4 h at 100-105 °C once the additions were completed. After this time, the reaction mixture was cooled to around 40-45 °C and the majority of the DMF was removed by evaporation under reduced pressure

at 45-47 °C. (The obtained liquid turned solid below ~ 40 °C.) The warm residue was poured onto r.t. MeOH (150 mL) under ultrasonication and allowed to crystallize overnight. The solid was filtered and washed with MeOH (3x10 mL) and dried at 75-80 °C. Yields, **4a**: 5.9 g, 90.1%; **4b**: 6.3 g, 84.2%. Preparation of **4a** was repeated in 0.5 mmol scale (0.95 g, 0.0005 mol) using identical molar ratio with dissolved NaN<sub>3</sub> (0.29 g, 0.0044 mol) in DMF solution (50 ml), stirring for 5 h at 100-105 °C. Workup as before, yield: 0.5 g, 76.3%.

**5a** and **5b**. Literature data [5,6] were used and **5b** was prepared *via* the modification of the known method when trimethylamine was used as base instead of NaH.

**6**. Synthesis of heptakis(6-deoxy-6-S-(1-dodecylthio))- $\beta$ CD, due to the lacking detailed literature batch data [7,8], was carried out analogously of the known method for heptakis(6-deoxy-6-S-(1-hexylthio))- $\beta$ CD [9] on 0.5 mmol scale The work-up needed some modifications due to technical difficulties in filtration. Yield: 1.0 g, 82.5%.

#### Thioureido (TU) CDs 3a and 3b.

Air-dry per-6-halogeno CD were mixed in the jar with a spatula and then ball milled for 120 min. The temperature was checked after 30, 60, 90 and 120 min using the infra thermometer.

The ground material was dissolved in water (15 mL) and the equipment washed with water (3x5 mL) and freeze-dried. The obtained solid was suspended in abs EtOH (10 ml), filtered and washed in abs EtOH (4x3 mL) and then with acetone (5x3 mL). The obtained solid was dissolved in water, filtered through charcoal and the filtrate was freeze-dried. TLC showed 30-40% products in the mother liquor after the organic solvents removal.

**Scaled-up preparation of 3a and 3b'** Ten-fold scale-up of ball milling reactions were performed using identical conditions with the original scale. The work-up used identical amounts of solvents as described in the previous paragraph. TLC showed 15-30% products in the mother liquor after the organic solvents removal.

**3a**, *entry*  $4^*$  R<sub>f</sub>=0.43-0.47 (max. 50 µg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); ESI-MS (neutralized with NaOH, m/z): 1542.26 [M+H]<sup>+</sup>, 371.34 [M-3(CN<sub>2</sub>H<sub>3</sub>)+HCOONa+4H]<sup>4+</sup>, 236.32 [M+Na<sub>2</sub>CO<sub>3</sub>+7H]<sup>7+</sup>. **3b** *entry* 9 R<sub>f</sub>=0.39-0.44 and **3b'** *entry* 11 R<sub>f</sub>=0.38-0.42 (max. 50 µg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); ESI-MS (after neutralization, m/z): 595.78 [M+2H+Na]<sup>3+</sup>, 441.33 [M+4H]<sup>4+</sup>, 353.44 [M +5H]<sup>5+</sup>. IR and NMR assignments are in Table S 4, Table S 5, and Table S 6

\* Entry numbers are corresponding to the entry numbers in **Table 1** of the main text.

	Table S 1 Amounts of reagents and yields (scaled up experiments are in italics):												
	$CH_2$ - <b>X</b>	Compnd.	Entries in Table 1	CD [g (mol)]	TU/CH <sub>2</sub> – X molar	Balls/ substance	TU [g, (mol)]	Yield [g (%)]					
_					ratio	mass ratio							
	Ι	<b>3</b> a	2	0.190	1.5	230	0.116	0.03					
				(0.00010)			(0.00105)	(12.3)					
	Ι	<b>3</b> a	3	0.190	3.5	153	0.270	0.06					
βCD				(0.00010)			(0.00245)	(24.6)					
	Ι	3a	4	1.904	3.5	15	2.700	1.48					
				(0.00100)			(0.02450)	(61)					
	Cl	3a'	5	0.126	3.5	178	0.270	traces					
				(0.00010)			(0.00245)						
	Ι	<b>3</b> b	7	0.218 g,	1.5	248	0.066	0.04					
				(0.00010)			(0.00120)	(14.4)					
	Br	<b>3</b> b'	8	0.090 g,	1.5	451	0.066	0.022					
				(0.00005)			(0.00060)	(9.1)					
	Ι	<b>3</b> b	9	0.109 g,	3.5	268	0.154	0.046					
γCD				(0.00005)			(0.00140)	(33.0)					
	Br	<b>3</b> b'	10	0.090 g,	3.5	288	0.154	0.047					
				(0.00005)			(0.00140)	(39.0)					
	Br	3b'	11	0.900 g,	3.5	29	1.543	0.69					
				(0.00050)			(0.01400)	(57.3)					

T (°C) 90 80 70 60 βCD • 3a entry 2 50 • 3a entry 3 derivatives ■ 3a entry 4 40 ▲ 3a entry 5 30 20 0 30 60 90 t (min) 120



Fig. S 1 Temperature-ball milling time curves of thiourea/per-6-halogenated CDs

**Per-6-azido CDs 4a and 4b.** The air-dry per-6-iodo-CD were mixed in the jar with a spatula and then ball milled for 120 min. The temperature was checked after 30, 60, 90 and 120 min using the infra thermometer. The ground material was suspended in water (30 mL) and the equipment was washed with water (3x10 mL). The filtered material was dissolved in DMF (0.5 ml) and precipitated with methanol (6 ml) and the solid was dried at around 70 °C, under reduced pressure, in the presence of P<sub>2</sub>O<sub>5</sub> and KOH.

Scaled-up preparation of 4a and 4b (entries 17 and 20, respectively): The air-dry per-6-iodo-CD were mixed in the jar with a spatula and then ball milled for 120 min. The work-up used double amounts of organic solvents than as it is described in the previous paragraph.

**4a** entry 17  $R_f=0.59-0.62$  (max. 75  $\mu$ g; presumably as I<sup>-</sup> complex, 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7). **4b** entry 20  $R_f=0.68-0.72$  (max. 75  $\mu$ g; presumably as I<sup>-</sup> complex, 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7). IR and NMR assignments are in Table S 4, Table S 5, and Table S 6

			0	•	` 1	1		
	$CH_2$ -X	Compnd.	Entries in	CD	NaN <sub>3</sub> /CH <sub>2</sub> -	Balls/	NaN <sub>3</sub>	Yield
			Table 1	[g (mol)]	X molar	substance	[g, (mol)]	[g (%)]
					ratio	mass ratio		
	Cl	4a'	15	0.126	5	199	0.228	traces
				(0.00010)			(0.0035)	
	Ι	<b>4a</b>	16	0.190	5	169	0.228	0.06
				(0.00010)			(0.0035)	(68.7)
	Ι	<b>4</b> a	17	0.952	5	34	1.138	1.48
0CD				(0.00050)			(0.0175)	(71.8)
pCD	$\mathbf{I}^1$	I <sup>1</sup> 4a	-	0.190	5	169	0.228	traces
				(0.00010)			(0.0035)	
	$\mathbf{I}^2$	<b>4a</b>	-	0.190	5	169	0.228	traces
				(0.00010)			(0.0035)	
	<b>I</b> <sup>3</sup>	<b>4</b> a	-	0.190	5	169	0.228	traces
				(0.00010)			(0.0035)	
	Ι	<b>4b</b>	19	0.109 g,	10	191	0.260	0.05
CD				(0.00005)			(0.0040)	(66.8)
γCD	Ι	<i>4b</i>	20	1.088 g,	10	19	2.600	0.532
				(0.00050)			(0.0400)	(71.1)

Table S 2 Amounts of reagents an	<i>id yields</i> (scaled	up experiments	are italics):
----------------------------------	--------------------------	----------------	---------------

 $^1$  added 100 µl 1:1 (v/v) EtOH water before milling  $^2$  added 50 µl water before milling  $^3$  added 50 µl 1-pentanol before milling





Fig. S 2 Temperature-ball milling time curves of NaN<sub>3</sub>/per-6-halogenated CDs

**Per-6-S-carboxyethyl CDs 5a and 5b.** MPA and potassium t-butoxide (double molar amount of MPA) were added to the jar, the formed solid was cracked using a spatula, and air-dry per-6-iodo-CD was added, then mixed with a spatula in the jar, balls were added and ball milling was carried out for 120 min. The temperature was checked after 30, 60, 90 and 120 min using the infra thermometer. The ground material was dissolved in water (15 mL) and the equipment washed with water (3x5 mL) filtered and then freeze-dried. The solid was dissolved in water (1 mL), 1 M HCl was added (2 mL) and the product was precipitated in acetone (30 mL). The solid was filtered, washed with acetone and then dissolved in ammonia containing water (5 mL), filtered through charcoal and then the filtrate was freeze-dried. No CH<sub>2</sub>-I signal was found in the HSQC spectrum and products contained some of complexed MPA.

The  $\gamma$ CD reaction was repeated, with altered reagent addition sequence and practically identical yields were found, data are not inserted to the tables.

Scaled-up preparation of 5a and 5b (entries 23 and 28, respectively). Ten-fold scale-up of ball milling reactions carried out as described in the previous paragraph but the isolation method was a little different: the larger amount allowed the filtration of the protonated products without addition of acetone. The freeze-dried crude was dissolved in 5 mL water (pH ~7.5-8.0) and then the pH was adjusted <2 with 2 M HCl when the product precipitated. The solid was filtered, washed 3x2 mL

water. The solid was dissolved aqueous ammonia solution, filtered and washed 3\*1 mL water then freeze dried.

**5a** entry 23 R<sub>f</sub>=0.02-0.05 (max. 25  $\mu$ g; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); ESI-MS (m/z): 362.85 [M+HCOONa-5H]<sup>5-</sup>, 360.83 [M+NaCl-5H]<sup>5-</sup>, 323.23 [M+HCOONa+MPANa-5H]<sup>5-</sup>.

**5b** *entry* 28 R<sub>f</sub>=0.02-0.05 (max. 25 μg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7); ESI-MS (m/z): 444.85 [M+5HCOOH-5H]<sup>5-</sup>, 420.95 [M+MPA-5H]<sup>5-</sup>, 363.22 [M+4HCOOH-6H]<sup>6-</sup>, 317.07 [M+5HCOOH-7H]<sup>7-</sup>, 277.09 [M+2(HCOOH+HCOONa)-8H]<sup>8-</sup>.

IR and NMR assignments are in Table S 4, Table S 5, and Table S 6

Table S 3 Amounts of reagents and yields (scaled up experiments are italics):

	CH <sub>2</sub> -X	Compnd.	Entries in Table 1	CD [g (mol)]	MPA/CH <sub>2</sub> – X molar ratio	Balls/ substance mass ratio	MPA [g, (mol)]	Yield [g (%)]
RCD	Ι	5a	22	0.190 (0.00010)	1.43	124	$0.122 \\ (0.00115)^1$	0.16 (85.5)
βCD	Ι	5a	23	1.900 (0.00100)	1.5	12	1.274 $(0.01200)^2$	1.33 (71.1)
	Ι	5b	26	0.218 g, (0.00010)	1.25	129	0.106 $(0.0010)^3$	0.087 (81.4)
γCD	Br	5b'	27	0.090 g, (0.00005)	1.5	243	0.064 $(0.0006)^4$	0.092 (86.0)
	Br	5b'	28	0.900 g, (0.00050)	1.5	24.4	$0.637$ $(0.0060)^5$	0.774 (72.4)

<sup>1</sup> KO<sup>t</sup>Bu: 0.258 g, 0.0023 mol; <sup>4</sup> KO<sup>t</sup>Bu: 0.135 g, 0.0012 mol; <sup>2</sup> KO<sup>t</sup>Bu: 2.693 g, 0.0240 mol; <sup>5</sup> KO<sup>t</sup>Bu: 1.347 g, 0.0120 mol <sup>3</sup> KO<sup>t</sup>Bu: 0.224 g, 0.0020 mol;





Fig. S 3 Temperature-ball milling time curves of 3-mercaptopropionic acid/KO<sup>t</sup>Bu/per-6-halogenated CDs **Reaction of per-6-iodo-\betaCD and 1-dodecanthiol (DDS).** DDS (0.22 g, 0.0011 mol) and potassium t-butoxide (0.144 g, 0.0012 mol) were added to the jar, the formed solid was cracked using a spatula and air-dry per-6-iodo- $\beta$ CD (0.19 g, 0.0001 mol) was added and then mixed with a spatula in the jar. Balls/substance mass ratio was ~127. The balls were added and ball milling was carried out for 120 min. The temperature was checked after 30, 60, 90 and 120 min using the infra thermometer. The jar content was washed MeOH, then acetone, and finally dissolved in methylene chloride and MeOH addition resulted in an impossible-to-filter milk-like solution-suspension. After centrifuging the supernatant was removed and treated with chloroform. The product was only partially soluble in ethanol-free chloroform and DMSO. Yield: 0.23 g, 94.8%.

6a entry 31 Rf=0.00-0.03 (max. 25 µg; 1,4-dioxane/cc. NH<sub>3</sub>-H<sub>2</sub>O=10:7).

IR and NMR assignments are in Table S 4, Table S 5, and Table S 6



Fig. S 4 Temperature-ball milling time curve of 1-dodecanethiol/KO<sup>t</sup>Bu/per-6-iodo-βCDs

Table S 4	Characteristic	IR absorption	bands (cm <sup>-</sup>	<sup>1</sup> ) of the	prepared	compound	ls
	~			~ ~ ~	* ~ .		~

	N-H	С-Н	NH3 <sup>+</sup>	<b>N</b> 3	0-C-0	$N=C-NH_2$	0=C-0	C-S
				cm <sup>-1</sup>				
<b>2a</b> (I)		2912-3032			1658			
2a' (Cl)		2884-2925						
<b>2b</b> (I)		2906-3032			1658			
<b>2b'</b> (Br)		2832-3038			1654			
<b>3a</b> (TU*HI) <i>entry 4</i>	3123	2829-2901	2755		1651	1545		636
<b>3b</b> (TU*HI) <i>entry 9</i>	3120	2934	2747		1651	1556		647
<b>3b'</b> (TU*HBr) <i>entry</i> 11	3101	2930-2902	2757		1647	1547		655
<b>4a</b> (N3) entry 17		2914		2104	1633			
<b>4b</b> (N3) entry 20		2927		2107	1658			
<b>5a</b> (MPA) entry 23	3054	2819-2901	~2690-2700		1703			668
<b>5b</b> (MPA) <i>entry</i> 28	3176	2904	2738		1647		1610	643
<b>6</b> (DDS) <i>entry 31</i>		2848-2950			1662			719

\* Overlapped with DMF signals (compounds 2 and 4) and overlapped with carboxylic signals (compounds 5a and 5b).







Fig. S 6. IR spectrum of heptakis(6-deoxy-6-chloro)-βCD 2a'







Fig. S 8. IR spectrum of octakis(6-bromo-6-deoxy)-yCD 2b'



Fig. S 9. IR spectrum of heptakis(6-deoxy-6-S-thiuronium)- \beta CD iodide 3a entry 4



Fig. S 10. IR spectrum of octakis(6-deoxy-6-S-thiuronium)-  $\gamma$ CD iodide 3b entry 9



Fig. S 11. IR spectrum of octakis(6-deoxy-6-S-thiuronium)-7CD bromide 3b' entry 11



Fig. S 12. IR spectrum of heptakis(6-azido-6-deoxy)- \beta CD 4a entry 17







Fig. S 14. IR spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl))-  $\beta$ CD NH4<sup>+</sup> 5a entry 23



Fig. S 15. IR spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl))-  $\gamma$ CD NH4<sup>+</sup> 5b entry 28



Fig. S 16. IR spectrum of heptakis(6-deoxy-6-S-(1-dodeclythio))-βCD 6 entry 31







Fig. S 18. ESI-MS spectrum of octakis(6-deoxy-6- S-thioureido)-  $\gamma$ CD 3b entry 11







Fig. S 20. ESI-MS spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl)-  $\gamma CD$  5b entry 28

	H1	H2	H3	H4	Н5	H6	α	β		ω
<b>2a</b> (I)	5.03	3.33-3.50	3.56-3.73	3.25-3.41	3.65-3.76	3.77-3.92, 3.40-3.55				
2a' (Cl)	4.95	3.38	3.63	3.37	3.83	3.79,4.07				
<b>2b</b> (I)	5.07	3.35-3.48	3.57-3.72	3.25-3.38	3.57-3.72	3.86,3.45				
<b>2b'</b> (Br)	5.06	3.26-3.53	3.57-3.75	3.22-3.58	3.72-3.97	3.85-4.17, 3.52-3.85				
<b>3a</b> (TU*HI) <i>entry 4</i>	5.17	3.61-3.76	3.87-4.03	3.52-3.68	4.31-4.45	3.65-3.80, 3.39-3.54				
<b>3b</b> (TU*HI) <i>entry 9</i>	5.20	3.60-3.76	3.84-3.99	3.48-3.68	4.25-4.40	3.66-3.82, 3.38-3.53				
<b>3b'</b> (TU*HBr) <i>entry 11</i>	5.16	3.58-3.72	3.81-3.94	3.50-3.63	4.16-4.28	3.62-3.72, 3.37-3.49				
<b>4a</b> (N <sub>3</sub> ) entry 17	4.94	3.29-3.52	3.54-3.76	3.28-3.48	3.71-3.90	3.52-3.92				
<b>4b</b> (N <sub>3</sub> ) entry 20	4.97	3.36-3.50	3.54-3.67	3.32-3.46	371-3.84	3.69-3.83, 3.54-3.68				
<b>5a</b> (MPA) entry 23	5.04	3.47-3.57	3.78-3.92	3.46-3.62	3.88-3.98	2.98-3.16, 2.84-2.98	2.72- 2.86	2.34- 2.50		
<b>5b</b> (MPA) <i>entry</i> 28	5.13	3.50-3.65	3.88	3.40-3.64	3.98	3.05-3.20, 3.86-2.98	2.75- 2.88	2.40- 2.58		
<b>6</b> (DDS) entry 31	4.86-5.03	3.32-3.46	3.58-3.76	3.26-3.43	3.74-3.90	3.00-3.15, 2.80-2.88	2.65- 2.74	1.57- 1.72	1.18- 1.58	0.83- 0.96

Table S 5 Proton NMR assignment (ppm) of compounds (from <sup>1</sup>H and HSQC spectra)

Table S 6 Carbon NMR assignment (ppm) of compounds (from HSQC spectra)

	C1	C2	C3	C4	C5	C6	α	β		ω
<b>2a</b> (I)	103.23	73.14	72.53	87.11	73.75	9.68,9.91				
2a' (Cl)	101.94	71.85	72.37	83.47	71.03	44.73,44.78				
<b>2b</b> (I)	102.22	72.8	71.99	85.73	71.99	9.75,9.87				
<b>2b'</b> (Br)	102.11	72.77	72.47	84.47	71.82	35.22				
<b>3a</b> (TU*HI) <i>entry 4</i>	102.09	72.26	72.85	85.25	71.65	33.53,33.57				
<b>3b</b> (TU*HI) <i>entry 9</i>	102.70	73.03	72.96	85.10	72.31	34.14,34.23				
<b>3b'</b> (TU*HBr) <i>entry 11</i>	101.94	71.96	72.12	84.18	71.18	33.12,33.15				
<b>4a</b> (N <sub>3</sub> ) entry 17	103.24	73.05	72.79	84.37	71.41	52.07				
<b>4b</b> (N <sub>3</sub> ) entry 20	102.41	72.74	73.04	83.42	70.99	51.70,51.88				
5a (MPA) entry 23	100.67	71.92	71.31	82.90	72.18	33.83,32.56	34.7	37.57		
<b>5b</b> (MPA) <i>entry</i> 28	101.5	72.67	72.69	82.97	72.59	34.38,33.90	30.22	37.76		
<b>6</b> (DDS) entry 31	102.48	72.58	72.84	85.74	71.53	33.68,33.92	32.81	29.95	29.99	14.91



Fig. S 21. <sup>1</sup>H-NMR spectrum of heptakis(6-deoxy-6-iodo)- $\beta$ CD 2a



Fig. S 22. HSQC-NMR spectrum of heptakis(6-deoxy-6-iodo)- \beta CD 2a



Fig. S 23. <sup>1</sup>H-NMR spectrum of heptakis(6-deoxy-6-chloro)-  $\beta$ CD 2a'



Fig. S 24. HSQC-NMR spectrum of heptakis(6-deoxy-6-chloro)- \beta CD 2a'



Fig. S 25. <sup>1</sup>H-NMR spectrum of octakis(6-deoxy-6-iodo)-  $\gamma$ CD 2b



Fig. S 26. HSQC-NMR spectrum of octakis(6-deoxy-6-iodo)- yCD 2b



Fig. S 27. <sup>1</sup>H-NMR spectrum of octakis(6-bromo-6-deoxy)-<sub>7</sub>CD 2b'



Fig. S 28. HSQC-NMR spectrum of octakis(6-bromo-6-deoxy)-yCD 2b'



Fig. S 29. <sup>1</sup>H-NMR spectrum of heptakis(6-deoxy-6-thioureido)-βCD iodide 3a, entry 4, thiourea/I ratio=3.5



*Fig. S 30. HSQC-NMR spectrum of heptakis*(6-deoxy-6-thioureido)-βCD iodide **3a**, entry 4, thiourea/I ratio=3.5





Fig. S 31. <sup>1</sup>H-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD iodide 3b, entry 7, thiourea/I ratio=1.5



Fig. S 32. HSQC-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD iodide **3b**, entry 7, thiourea/I ratio=1.5



Fig. S 33. <sup>1</sup>H-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD iodide **3b** entry 9, thiourea/I ratio=3.5



Fig. S 34. HSQC-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD iodide 3b, entry 9, thiourea/I ratio=3.5



Fig. S 35. <sup>1</sup>H-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD bromide **3b**', entry 11, thiourea/I ratio=3.5



Fig. S 36. HSQC-NMR spectrum of octakis(6-deoxy-6-thioureido)- $\gamma$ CD bromide **3b**', entry 11, thiourea/I ratio=3.5





Fig. S 37. <sup>1</sup>H-NMR spectrum of heptakis(6-azido-6-deoxy)- $\beta$ CD 4a entry 17



Fig. S 38. HSQC-NMR spectrum of heptakis(6-azido-6-deoxy)-  $\beta$ CD 4a entry 17



Fig. S 39. <sup>1</sup>H-NMR spectrum of octakis(6-azido-6-deoxy)-<sub>7</sub>CD 4b entry 20



Fig. S 40. HSQC-NMR spectrum of octakis(6-azido-6-deoxy)-yCD 4b entry 20



Fig. S 41. <sup>1</sup>H-NMR spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl)- $\beta$ CD NH<sub>4</sub><sup>+</sup> 5a entry 23



Fig. S 42. HSQC-NMR spectrum of heptakis(6-deoxy-6-S-(3-mercapto)propionyl)- \beta CD NH4<sup>+</sup> 5a entry 23



Fig. S 43. <sup>1</sup>H-NMR spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl)-<sub>2</sub>CD NH<sub>4</sub><sup>+</sup> 5b entry 28



Fig. S 44. HSQC-NMR spectrum of octakis(6-deoxy-6-S-(3-mercapto)propionyl)- $\gamma$ CD NH4<sup>+</sup> 5b entry 28



Fig. S 45. <sup>1</sup>H-NMR spectrum of heptakis(6-deoxy-6-S-(1-dodecylthio)- $\beta$ CD 6 entry 31



Fig. S 46. HSQC-NMR spectrum of heptakis(6-deoxy-6-S-(1-dodecylthio)- $\beta$ CD 6 entry 31

## References

- 1. Gadelle, A.; Defaye, J. Angew. Chem. Int. Ed. Eng., 1991, 30, 78-80.
- Tuza, K. Synthesis of Cyclodextrin based C. Perfringens Antidotes. (Hung), Master's Thesis Univ. L. Eötvös, Budapest, Hungary, 2011
- 3. Uccello-Barretta, G.; Evangelisti, C.; Balzano, F.; Vanni, L.; Aiello, F.; Jicsinszky, L. *Carbohydr. Res.*, **2011**, *346*, 753-758.
- 4. Szurmai, Z.; Lipták, A.; Debrecen; Szejtli, J. Starch Stärke, 1990, 42, 447-449.
- Adam, J. M.; Bennett, D. J.; Bom, A.; Clark, J. K.; Feilden, H.; Hutchinson, E. J.; Palin, R.; Prosser, A.; Rees, D. C.; Rosair, G. M.; Stevenson, D.; Tarver, G. J.; Zhang, M.-Q. *J. Med. Chem.*, 2002, 45, 1806-1816.
- Bom, A.; Bradley, M.; Cameron, K.; Clark, J. K.; Egmond, J. van; Feilden, H.; MacLean, E. J.; Muir, A. W.; Palin, R.; Rees, D. C.; Zhang, M.-Q. Angew. Chem., 2002, 114, 275-280.
- 7. Mazzaglia, A.; Donohue, R.; Ravoo, B. J.; Darcy, R. Eur. J. Org. Chem., 2001, 2001, 1715-1721.
- 8. Ling, C.-C.; Darcy, R.; Risse, W. J. Chem. Soc., Chem. Commun., 1993, 438-440.
- 9. Darcy, R.; Penkler, L., US20060148756, July 6, 2006