## Supporting Information

for

# Biomimetic synthesis and HPLC-ECD analysis of the isomers of dracocephins A and B 

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## Details of physicochemical measurements

## Potentiometric $p K_{\mathrm{a}}$ determination

T3-Sirius automated $\mathrm{p} K_{\mathrm{a}}$ analyser (Sirius Analytical Instruments Ltd., Forest Row, UK) fitted with combina,tion $\mathrm{Ag} / \mathrm{AgCl} \mathrm{pH}$ electrode was used for determination of dissociation constants. The $\mathrm{p} K_{\mathrm{a}}$ values were calculated by RefinementProTM software (Sirius Analytical Instruments Ltd., Forest Row, UK). Methodologies used by the software have been described in earlier publications [1,2].

## Titration of dracocephins $A$ and $B$

The titration of dracocephins $A$ and $B$ was carried out in a similar procedure as described in the literature [3]. In each experiment, 1.50 mL of a 1 mM aqueous solution of sample was prealkalified to $\mathrm{pH} 12.0-12.5$ with 0.5 M KOH , and then titrated with 0.5 M HCl to an appropriately low pH , usually 2.0. The titrations were carried out at constant ionic strength $(I=0.15 \mathrm{M} \mathrm{KCl})$ and temperature ( $T=25.0 \pm$ $0.1^{\circ} \mathrm{C}$ ), and under nitrogen atmosphere. Three to five parallel measurements were carried out and the $\mathrm{p} K_{\mathrm{a}}$ values of samples were calculated by RefinementProTM software.

## Potentiometric determination of partition coefficients

Partition coefficients were determined in a similar manner as described in the literature [4]. Typically, $0.8-2.5 \mathrm{~mL}$ of $0.5-10 \mathrm{mM}$ solutions of samples were titrated under the same conditions as in $\mathrm{p} K_{\mathrm{a}}$ determinations but in the presence of various amounts of the partitioning solvent, water-saturated $n$-octanol. The phase ratio applied was varied from 1.5 mL water -0.05 mL n-octanol to 1.5 mL water -1.0 mL
$n$-octanol. From the $n$-octanol containing titrations the $\mathrm{p}_{0} K_{\mathrm{a}}$ (the apparent ionization constant in the presence of $n$-octanol) and then $\log P$ values were estimated and refined by a weighted non-linear least-squares procedure, where the aqueous $\mathrm{p} K_{a}$ values (taken from aqueous titrations) were used as unrefined contributions. For each compound a minimum of three to six titrations at different phase volume ratios were measured, and the respective average $\log P$ values were calculated. $\log D$ values were also calculated by $\log P$ and $\mathrm{p} K_{\mathrm{a}}$ values using the HendersonHasselbalch equation. The relevant relationships between $\log P, \mathrm{p} K_{\mathrm{a}}$, and $\mathrm{p}_{0} K_{\mathrm{a}}$, for mono- and multiprotic substances, including cases of ion-pair formation, have been described in detail earlier [5].

Antioxidant activity: The radical scavenging capacity of dracocephins $A$ and $B$ were tested in the microplate format of the 2,2-diphenyl-1-picrylhydrazyl (DPPH) assay [6]. Naringenin and quercetin were obtained from PhytoLab (Vestenbergsgreuth, Germany), and used as controls.

Permeability measurements: The PAMPA-BBB method was previously published [7]. Briefly, a 96-well acceptor plate and a 96-well filter plate are assembled into a sandwich. The hydrophobic filter material of the 96 well filter plate is coated with $5 \mu \mathrm{~L}$ of a $2.6 \%(\mathrm{w} / \mathrm{v})$ dodecane/hexane ( $25: 75 \mathrm{v} / \mathrm{v} \%$ ) solution of porcine brain lipid (PBL). Subsequently, the acceptor wells at the bottom of the sandwich are filled with $300 \mu \mathrm{~L}$ of 10 mM PBS solution with $5 \%$ DMSO adjusted to pH 7.4 . The donor wells at the top of the sandwich are hydrated with $150 \mu \mathrm{~L}$ of test compound solution. The test compound solution is prepared by diluting $\times 100$ from a 10 mM stock solution in DMSO using PBS solution at pH 7.4 with $5 \%$ DMSO followed by filtration through a MultiScreen Solubility filter plate. The resulting sandwich is then incubated at $37^{\circ} \mathrm{C}$
for 4 h . After the incubation, PAMPA sandwich plates are separated and compound concentrations in donor and acceptor solutions are determined by HPLC-DAD.

## Biology

## Cell lines

SH-SY5Y neuroblastoma was cultured in EMEM media supplemented with nonessential amino acids, 1 mM Na pyruvate and $10 \%$ inactivated fetal bovine serum, nystatin, 2 mM L-glutamine, 100 U penicillin and 0.1 mg streptomycin, purchased from Sigma. Cells were cultured at $37^{\circ} \mathrm{C}$ and $5 \% \mathrm{CO}_{2}$.

Two mouse lymphoma cell lines were used to test for the capacity of the compounds to inhibit the function of the ABCB1 transporter: a parental (L5178 PAR) cell line, L5178 mouse T-cell lymphoma cells (ECACC catalog no. 87111908, U.S. FDA, Silver Spring, MD, U.S.), and a multi-drug resistant (L5178 MDR) cell line derived from L5178 by transfection with pHa MDR1/A retrovirus [8]. Cells were cultured in McCoy's 5A media supplemented inactivated horse serum and antibiotics as above. MDR cell line was selected by culturing the infected cells with $60 \mu \mathrm{~g} / \mathrm{L}$ colchicine (Sigma).

## Materials and methods:

Compounds ( $\pm$ )-2a-d and ( $\pm$ )-3a-d were dissolved in DMSO at a stock concentration of 10 mM .

Cytotoxicity on SH-SY5Y cells: Cytotoxicity on SH-SY5Y cells was measured in an analogous manner as described in the literature [9]. 10,000 cells per well were seeded overnight. Serial dilutions of the compounds were prepared and added the following day to the plate. Cells were then incubated for 48 h , after which $10 \%$ MTT was added to each well. After 4 h , SDS was added to the medium and the results
were read after o/n incubation. Fifty percent inhibitory concentrations ( $\mathrm{IC}_{50}$ ) were calculated using nonlinear regression curve fitting of log (inhibitor) versus normalized response and variable slope with a least squares (ordinary) fit of GraphPad Prism 5 software, for three independent samples.

Evaluation of compounds effect on rhodamine 123 accumulation: Rd123 concentration inside L5178 PAR and L5178 MDR cells was determined by flow cytometry. Briefly, $2 \times 10^{6}$ cells $/ \mathrm{mL}$ were treated with 2 and $20 \mu \mathrm{M}$ of each compound and incubated for 10 min at rt . Rhodamine 123 (Sigma, Germany) was added to a final concentration of $5.2 \mu \mathrm{M}$. The samples were incubated for 20 min at $37^{\circ} \mathrm{C}$ in water bath and then centrifuged ( $2000 \mathrm{rpm}, 2 \mathrm{~min}$ ). The pellet was resuspended in 0.5 mL of phosphate buffer saline (PBS) (Sigma, Germany). The washing step was repeated twice. The fluorescence of the samples was measured by flow cytometry (Becton Dickinson FACScan, BD, U.S.). Tariquidar at $2 \mu \mathrm{M}$ was used as positive control. Fluorescence activity ratio measures the capacity of inhibition (accumulation of Rd123) and it is equal to the ratio between the FL-1 values of the $\mathrm{L} 5178_{\text {MDR }}$ cells treated and untreated.

## Computational section

Mixed torsional/low-frequency mode conformational searches were carried out by means of the Macromodel 9.9.223 software using the Merck Molecular Force Field (MMFF) with an implicit solvent model for $\mathrm{CHCl}_{3}[10]$. Geometry reoptimizations were carried out at the B3LYP/6-31G(d) level in vacuo, B3LYP/TZVP, B97D/TZVP, CAMB3LYP/TZVP and M06-2X/TZVP levels with the PCM solvent model for MeCN or $\mathrm{CHCl}_{3}$. TDDFT ECD calculations were run with various functionals (B3LYP, BH\&HLYP, CAM-B3LYP, PBE0) and the TZVP basis set as implemented in the Gaussian 09 package with the same or no solvent model as in the preceding DFT
optimization step [11]. ECD spectra were generated as sums of Gaussians with 3000 $\mathrm{cm}^{-1}$ widths at half-height (corresponding to ca. 24 nm at 280 nm ), using dipole-velocity-computed rotational strength values [12]. Boltzmann distributions were estimated from the ZPVE-corrected B3LYP/6-31G(d) energies in the gas-phase calculations and from the B3LYP/TZVP, B97D/TZVP, CAM-B3LYP/TZVP and M062X/TZVP energies in the solvated ones. The MOLEKEL software package was used for visualization of the results [13].

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Figure S1: HPLC-ECD spectra of ( $2 R$ )- and ( $2 S$ )-naringenin (1).





Figure S2: Structure and population of the low-energy B3LYP/6-31G(d) in vacuo conformers ( $>2 \%$ ) of ( $R$ )-1.


Figure S3: Experimental HPLC-ECD spectrum of $(R) \mathbf{- 1}$ compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers at various levels.


Figure S4: Structure and population of the low-energy B3LYP/TZVP PCM/ $\mathrm{CHCl}_{3}$ conformers (>2\%) of (R)-1.


Figure S5: Experimental HPLC-ECD spectrum of $(R)-1$ compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/ $\mathrm{CHCl}_{3}$ low-energy conformers at various levels.


Figure S6: Structure and population of the low-energy B97D/TZVP PCM/CHCl ${ }_{3}$ conformers (>2\%) of ( R )-1.


Figure S7: Experimental HPLC-ECD spectrum of $(R) \mathbf{- 1}$ compared with the Boltzmannweighted ECD spectra computed for the $\mathrm{B} 97 \mathrm{D} / \mathrm{TZVP} \mathrm{PCM} / \mathrm{CHCl}_{3}$ low-energy conformers at various levels.


Figure S8: Structure and population of the low-energy M06-2X/TZVP PCM/ $\mathrm{CHCl}_{3}$ conformers ( $>2 \%$ ) of ( $R$ )-1.


Figure S9: Experimental HPLC-ECD spectrum of $(R) \mathbf{- 1}$ compared with the Boltzmannweighted ECD spectra computed for the M06-2X/TZVP $\mathrm{PCM} / \mathrm{CHCl}_{3}$ low-energy conformers at various levels.


Figure S10: Experimental HPLC-ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the CAM-B3LYP/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime}, R\right)-\mathbf{2}$ at various levels.


Figure S11: Experimental HPLC-ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $\left(2 R, 5^{\prime \prime} R\right)-\mathbf{2}$ at various levels.


Figure S12: Experimental HPLC-ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $\left(2 R, 5^{\prime} ' S\right)$ - $\mathbf{2}$ at various levels.


Figure S13: Experimental HPLC-ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime \prime} R\right)-\mathbf{2}$ at various levels.


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Figure S15: Experimental HPLC-ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B97D/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime} \times R\right)-\mathbf{2}$ at various levels.


Figure S16: Experimental HPLC-ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B97D/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime} ' S\right)$ - $\mathbf{2}$ at various levels.


Figure S17: Experimental HPLC-ECD spectra of 3b and 3c compared with the Boltzmannweighted ECD spectra computed for the CAM-B3LYP/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime \prime} R\right)-\mathbf{3}$ at various levels.


Figure S18: Experimental HPLC-ECD spectra of $\mathbf{3 b}$ and $\mathbf{3 c}$ compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $\left(2 R, 5^{\prime} \times R\right)-3$ at various levels.


Figure S19: Experimental HPLC-ECD spectra of 3b and 3c compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $\left(2 R, 5^{\prime} ' S\right)-\mathbf{3}$ at various levels.


Figure S20: Experimental HPLC-ECD spectra of 3b and 3c compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime} ' R\right)$-3 at various levels.


Figure S21: Experimental HPLC-ECD spectra of 3b and 3c compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of $\left(2 R, 5^{\prime} S\right)-3$ at various levels.


Figure S22: ${ }^{1} \mathrm{H}$ NMR spectrum of ( $\mathbf{\pm}$ )-3a-d in $\mathrm{CD}_{3} \mathrm{OD}$.
IV-3-1 Hazai Lasz
5h-8650
mixN 800 ms
COC $13+$ MeO

| $\operatorname{SAMPLE}_{\text {feb }}$ | ns flags |
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| 3434.1 |  |
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| ss 32 | F2 Processin |
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| ${ }^{n t}{ }_{20}$ AcQuisition ${ }^{32}$ | ${ }_{\text {fin }}^{\text {fis }}$ |
| sw1 ${ }^{3434.1}$ | F1 Processing |
| ${ }^{\text {ni }}$ transmitter ${ }^{200}$ | $\begin{array}{ll}\text { sb1 } \\ \text { sbs1 } 1 & -0.058 \\ -0.058\end{array}$ |
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| Sfra tof | fni display |
| tpwr ${ }^{\text {a }}$ | sp 815 |
| pw 6.500 | wp ${ }^{\text {a }}$ 33 |
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| $\mathrm{mixN}_{\text {PRESATURATION }}{ }^{0.800}$ | wp 1 rf |
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| dn decoupler $\mathrm{Cl}^{\text {a }}$ | rfpl plot |
| dm ${ }_{\text {dm }}$ | wc |
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|  | wc2 sc2 che |
|  | vs 3.80 |
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Figure S23: NOESY spectrum of ( $\mathbf{\pm}$ )-3a-d (No. 1).

Iv-3-1 Hazai Lasz10
Szigetvar
v5h- 8650

exp84 NOESY
date $\underset{\text { Feb }}{\text { SAMPLE }}{ }_{5} 2015$ hs FLAGS










 wet decoupler
$\underset{\substack{d n \\ d m}}{\substack{\text { din }}}$


Figure S24: NOESY spectrum of ( $\mathbf{\pm}$-3a-d (No. 2).

IV-3-1 Hazai Laszlo
Szigetvari Aron
v $5 \mathrm{~h}-8650$
5h-8650



Figure S25: gCOSY spectrum of ( $\mathbf{\pm} \mathbf{)} \mathbf{- 3 a} \mathbf{- d}$



Figure S27: ${ }^{1} \mathrm{H}$ NMR spectrum of $( \pm) \mathbf{- 3 a - d}$ in $\mathrm{CDCl}_{3}: \mathrm{CD}_{3} \mathrm{OD}(2: 1)$ - zoom from 5.24 to 7.74 ppm .


Figure S28: ${ }^{1} \mathrm{H}$ NMR spectrum of $( \pm)$-3a-d in $\mathrm{CDCl}_{3}: \mathrm{CD}_{3} \mathrm{OD}(2: 1)$ - zoom from 2.04 to 4.20 ppm .


Figure S29: gHSQCad spectrum of ( $\mathbf{\pm}$ )-3a-d.


Figure S30: gHMBCad spectrum of ( $\mathbf{\pm}$ )-3a-d.


Figure S31: ${ }^{1} \mathrm{H}$ NMR spectrum of ( $\pm$ )-2a-d.


Figure S32: ${ }^{1} \mathrm{H}$ NMR spectrum of $\mathbf{( \pm ) - 2 a - d ~ - ~ z o o m ~ f r o m ~} 5.30$ to 7.38 ppm .


Figure S33: ${ }^{1} \mathrm{H}$ NMR spectrum of $(\mathbf{\pm}) \mathbf{- 2 a - d}$ - zoom from 2.14 to 3.50 ppm .


Figure S34: gCOSY spectrum of ( $\mathbf{\pm}$ )-2a-d


Figure S35: NOESY spectrum of ( $\mathbf{~}$ )-2a-d.


Figure S36: gHSQCad spectrum of ( $\mathbf{\pm}$ )-2a-d.

IV-3-2 Haza Laszlo
Szigetvar
v $8 h-5310$
exp34 ghmbcad
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Figure S37: gHMBCad spectrum of ( $\mathbf{\pm}$ )-2a-d.


Figure S38: ${ }^{13} \mathrm{C}$ NMR spectrum of $( \pm)$-2a-d.

