Supporting Information

for

Biomimetic synthesis and HPLC–ECD analysis of the isomers of dracocephins A and B

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Details of physicochemical measurements

Potentiometric pK_a determination

T3-Sirius automated p K_a analyser (Sirius Analytical Instruments Ltd., Forest Row, UK) fitted with combina,tion Ag/AgCl pH electrode was used for determination of dissociation constants. The p K_a values were calculated by RefinementProTM software (Sirius Analytical Instruments Ltd., Forest Row, UK). Methodologies used by the software have been described in earlier publications [1,2].

Titration of dracocephins A and B

The titration of dracocephins A and B was carried out in a similar procedure as described in the literature [3]. In each experiment, 1.50 mL of a 1 mM aqueous solution of sample was prealkalified to pH 12.0–12.5 with 0.5 M KOH, and then titrated with 0.5 M HCl to an appropriately low pH, usually 2.0. The titrations were carried out at constant ionic strength (I = 0.15 M KCl) and temperature ($T = 25.0 \pm 0.1 \text{ °C}$), and under nitrogen atmosphere. Three to five parallel measurements were carried out and the p K_a values of samples were calculated by RefinementProTM software.

Potentiometric determination of partition coefficients

Partition coefficients were determined in a similar manner as described in the literature [4]. Typically, 0.8–2.5 mL of 0.5–10 mM solutions of samples were titrated under the same conditions as in pK_a determinations but in the presence of various amounts of the partitioning solvent, water-saturated *n*-octanol. The phase ratio applied was varied from 1.5 mL water – 0.05 mL *n*-octanol to 1.5 mL water – 1.0 mL

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n-octanol. From the *n*-octanol containing titrations the p_0K_a (the apparent ionization constant in the presence of *n*-octanol) and then $\log P$ values were estimated and refined by a weighted non-linear least-squares procedure, where the aqueous pK_a values (taken from aqueous titrations) were used as unrefined contributions. For each compound a minimum of three to six titrations at different phase volume ratios were measured, and the respective average $\log P$ values were calculated. $\log D$ values were also calculated by $\log P$ and pK_a values using the Henderson–Hasselbalch equation. The relevant relationships between $\log P$, pK_a , and p_0K_a , for mono- and multiprotic substances, including cases of ion-pair formation, have been described in detail earlier [5].

Antioxidant activity: The radical scavenging capacity of dracocephins A and B were tested in the microplate format of the 2,2-diphenyl-1-picrylhydrazyl (DPPH) assay [6]. Naringenin and quercetin were obtained from PhytoLab (Vestenbergsgreuth, Germany), and used as controls.

Permeability measurements: The PAMPA-BBB method was previously published [7]. Briefly, a 96-well acceptor plate and a 96-well filter plate are assembled into a sandwich. The hydrophobic filter material of the 96 well filter plate is coated with 5 μ L of a 2.6% (w/v) dodecane/hexane (25:75 v/v %) solution of porcine brain lipid (PBL). Subsequently, the acceptor wells at the bottom of the sandwich are filled with 300 μ L of 10 mM PBS solution with 5% DMSO adjusted to pH 7.4. The donor wells at the top of the sandwich are hydrated with 150 μ L of test compound solution. The test compound solution is prepared by diluting ×100 from a 10 mM stock solution in DMSO using PBS solution at pH 7.4 with 5% DMSO followed by filtration through a MultiScreen Solubility filter plate. The resulting sandwich is then incubated at 37 °C

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for 4 h. After the incubation, PAMPA sandwich plates are separated and compound concentrations in donor and acceptor solutions are determined by HPLC-DAD.

Biology

Cell lines

SH-SY5Y neuroblastoma was cultured in EMEM media supplemented with nonessential amino acids, 1 mM Na pyruvate and 10% inactivated fetal bovine serum, nystatin, 2 mM L-glutamine, 100 U penicillin and 0.1 mg streptomycin, purchased from Sigma. Cells were cultured at 37 °C and 5% CO₂.

Two mouse lymphoma cell lines were used to test for the capacity of the compounds to inhibit the function of the ABCB1 transporter: a parental (L5178_{PAR}) cell line, L5178 mouse T-cell lymphoma cells (ECACC catalog no. 87111908, U.S. FDA, Silver Spring, MD, U.S.), and a multi-drug resistant (L5178_{MDR}) cell line derived from L5178 by transfection with pHa MDR1/A retrovirus [8]. Cells were cultured in McCoy's 5A media supplemented inactivated horse serum and antibiotics as above. MDR cell line was selected by culturing the infected cells with 60 µg/L colchicine (Sigma).

Materials and methods:

Compounds (\pm) -**2a**–**d** and (\pm) -**3a**–**d** were dissolved in DMSO at a stock concentration of 10 mM.

Cytotoxicity on SH-SY5Y cells: Cytotoxicity on SH-SY5Y cells was measured in an analogous manner as described in the literature [9]. 10,000 cells per well were seeded overnight. Serial dilutions of the compounds were prepared and added the following day to the plate. Cells were then incubated for 48 h, after which 10% MTT was added to each well. After 4 h, SDS was added to the medium and the results

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were read after o/n incubation. Fifty percent inhibitory concentrations (IC_{50}) were calculated using nonlinear regression curve fitting of log (inhibitor) versus normalized response and variable slope with a least squares (ordinary) fit of GraphPad Prism 5 software, for three independent samples.

Evaluation of compounds effect on rhodamine 123 accumulation: Rd123 concentration inside L5178_{PAR} and L5178_{MDR} cells was determined by flow cytometry. Briefly, 2×10^{6} cells/mL were treated with 2 and 20 µM of each compound and incubated for 10 min at rt. Rhodamine 123 (Sigma, Germany) was added to a final concentration of 5.2 µM. The samples were incubated for 20 min at 37 °C in water bath and then centrifuged (2000 rpm, 2 min). The pellet was resuspended in 0.5 mL of phosphate buffer saline (PBS) (Sigma, Germany). The washing step was repeated twice. The fluorescence of the samples was measured by flow cytometry (Becton Dickinson FACScan, BD, U.S.). Tariquidar at 2 µM was used as positive control. Fluorescence activity ratio measures the capacity of inhibition (accumulation of Rd123) and it is equal to the ratio between the FL-1 values of the L5178_{MDR} cells treated and untreated.

Computational section

Mixed torsional/low-frequency mode conformational searches were carried out by means of the Macromodel 9.9.223 software using the Merck Molecular Force Field (MMFF) with an implicit solvent model for CHCl₃ [10]. Geometry reoptimizations were carried out at the B3LYP/6-31G(d) level in vacuo, B3LYP/TZVP, B97D/TZVP, CAM-B3LYP/TZVP and M06-2X/TZVP levels with the PCM solvent model for MeCN or CHCl₃. TDDFT ECD calculations were run with various functionals (B3LYP, BH&HLYP, CAM-B3LYP, PBE0) and the TZVP basis set as implemented in the Gaussian 09 package with the same or no solvent model as in the preceding DFT

optimization step [11]. ECD spectra were generated as sums of Gaussians with 3000 cm⁻¹ widths at half-height (corresponding to ca. 24 nm at 280 nm), using dipole-velocity-computed rotational strength values [12]. Boltzmann distributions were estimated from the ZPVE-corrected B3LYP/6-31G(d) energies in the gas-phase calculations and from the B3LYP/TZVP, B97D/TZVP, CAM-B3LYP/TZVP and M06-2X/TZVP energies in the solvated ones. The MOLEKEL software package was used for visualization of the results [13].

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Figure S1: HPLC–ECD spectra of (2*R*)- and (2*S*)-naringenin (1).



Figure S2: Structure and population of the low-energy B3LYP/6-31G(d) in vacuo conformers (>2%) of (R)-1.



Figure S3: Experimental HPLC–ECD spectrum of (R)-1 compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers at various levels.



Figure S4: Structure and population of the low-energy B3LYP/TZVP PCM/CHCl₃ conformers (>2%) of (R)-1.



Figure S5: Experimental HPLC–ECD spectrum of (*R*)-1 compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/CHCl₃ low-energy conformers at various levels.



Figure S6: Structure and population of the low-energy B97D/TZVP PCM/CHCl₃ conformers (>2%) of (R)-1.



Figure S7: Experimental HPLC–ECD spectrum of (R)-1 compared with the Boltzmannweighted ECD spectra computed for the B97D/TZVP PCM/CHCl₃ low-energy conformers at various levels.



Figure S8: Structure and population of the low-energy M06-2X/TZVP PCM/CHCl₃ conformers (>2%) of (R)-1.



Figure S9: Experimental HPLC–ECD spectrum of (R)-1 compared with the Boltzmannweighted ECD spectra computed for the M06-2X/TZVP PCM/CHCl₃ low-energy conformers at various levels.



Figure S10: Experimental HPLC–ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the CAM-B3LYP/TZVP PCM/MeCN low-energy conformers of (2R,5''R)-2 at various levels.



Figure S11: Experimental HPLC–ECD spectra of **2a** and **2d** compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of (2R,5"R)-**2** at various levels.



Figure S12: Experimental HPLC–ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $(2R,5^{"S})$ -2 at various levels.



Figure S13: Experimental HPLC–ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of (2R,5''R)-2 at various levels.



Figure S14: Experimental HPLC–ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of (2R,5''S)-2 at various levels.



Figure S15: Experimental HPLC–ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B97D/TZVP PCM/MeCN low-energy conformers of (2R,5"R)-2 at various levels.



Figure S16: Experimental HPLC–ECD spectra of 2a and 2d compared with the Boltzmannweighted ECD spectra computed for the B97D/TZVP PCM/MeCN low-energy conformers of $(2R,5^{"S})$ -2 at various levels.



Figure S17: Experimental HPLC–ECD spectra of 3b and 3c compared with the Boltzmannweighted ECD spectra computed for the CAM-B3LYP/TZVP PCM/MeCN low-energy conformers of (2R,5''R)-3 at various levels.



Figure S18: Experimental HPLC–ECD spectra of **3b** and **3c** compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $(2R,5^{"}R)$ -**3** at various levels.



Figure S19: Experimental HPLC–ECD spectra of **3b** and **3c** compared with the Boltzmannweighted ECD spectra computed for the B3LYP/6-31G(d) in vacuo low-energy conformers of $(2R,5^{,*}S)$ -**3** at various levels.



Figure S20: Experimental HPLC–ECD spectra of 3b and 3c compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of (2R,5''R)-3 at various levels.



Figure S21: Experimental HPLC–ECD spectra of **3b** and **3c** compared with the Boltzmannweighted ECD spectra computed for the B3LYP/TZVP PCM/MeCN low-energy conformers of (2R,5''S)-**3** at various levels.



Figure S22: ¹H NMR spectrum of (±)-3a–d in CD₃OD.

IV-3-1 Hazai Laszlo Szigetvari Aron v5h-8650 mixN=800 ms CDCl3+MeOD (2:1) exp85 NOESY ALL LING
 SAMPLE
 F

 date
 Feb
 5 2015
 hs

 solvent
 cd3od
 sspul

 sample
 PFGf1g

 ACQUISITION
 hsglv1

 Sw
 3434.1
 SP
 F1 FLAGS nn (ppm) .
 sample
 PFGflg
 y

 ACQUISITION
 hsglv2
 5796

 sw
 0.150
 temp
 25.0

 np
 1030
 gain
 10

 fb
 4000
 spin not used

 ss
 32
 F2 PROCESSING

 d1
 1.000
 sb
 -0.150

 nt
 32
 sbs
 -0.150

 nt
 323
 sbs
 -0.150

 nt
 323
 sbs
 -0.150

 nt
 3434.1
 F1 PROCESSING
 fi

 ni
 200
 sb1
 -0.058

 tn
 H1
 procl
 1p

 sfrq
 499.915
 fn1
 2048

 tof
 2.0
 procl
 1p

 sfrq
 499.915
 fn1
 2048

 tof
 2.0
 procl
 1p

 sfrq
 99.915
 fn1
 2048

 tof
 2.0
 procl
 1p

 stof
 0.800
 wp1
 3430.7< 5796 . 10 3 4. 0 4 4 C13 nnn PLOT wc sc2 sc2 vs 3.804 th ai cdc ph 155.0 1 155.0 0 . 3.80469e+06 2 onop 141 0 ٥ 0 6 0 P -----. 0 . -0 No. -8 14 B . 1AP 1 5 3 2 4 8 7 6 F2 (ppm)

d n d m

Figure S23: NOESY spectrum of (±)-3a-d (No. 1).

IV-3-1 Hazai Laszlo Szigetvari Aron v5h-8650 mixN=200 ms CDCl3+MeOD (2:1)

exp84 NOESY

SAMPLE		FLAGS	
date F	Feb 5 2015	hs nn	
solvent	cd3od	sspul y	
sample		PFGflq V	
ACQUISITION		hsalv1 5796	
sw	3434.1	SPECIAL	
at	0.150	temp 25.0	
np	1030	gain 10	
fb	4000	spin not used	
SS	32	F2 PROCESSING	
d 1	1.000	sb -0.150	
nt	32	sbs -0.150	
2D ACC	DUISITION	fn 2048	
sw1	3434.1	F1 PROCESSING	
ni	200	sb1 -0.058	
TRANSMITTER		sbs1 -0.058	
tn	H1	proc1 lp	
sfrq	499.915	fn1 2048	
tof	20.2	DISPLAY	
tpwr	52	sp 819.2	
pw	6.500	wp 3102.1	
NO	DESY	sp1 825.9	
nixN	0.200	wp1 3216.1	
PRESATURATION		rfl -812.5	
satmode	n	rfp 0	
wet	n	rfl1 -812.5	
DECO	DUPLER	rfp1 0	
dn	C13	PLOT	
dm	nnn	wc 155.0	
		sc 0	
		wc2 155.0	
		sc2 0	
		vs 1494	
		th 2	
		ai cdc ph	



Figure S24: NOESY spectrum of (±)-3a–d (No. 2).

IV-3-1 Hazai Laszlo Szigetvari Aron v5h-8650

exp7 gCOSY gCOS1 SAMPLE Feb 5 2015 hs cd3od sspul hsglvl SPECIAL SAMVLL date Feb 5 2015 solvent cd3od sample hsglvl ACQUISITION SPECIAL sw 2289.4 temp 25.0 at 0.150 gain 10 np 686 Spin not used fb 4000 F2 PROCESSING ss 32 sb -0.075 d1 1.000 sbs not used nt 2 fn 2048 an ACQUISITION F1 PROCESSING 2289.4 sb1 -0.056 266 sbs1 not used an 2048 nt 2 fn 2D ACQUISITION F1 P Sw1 2283.4 sb1 ni 256 sbs1 d2 0 proc1 PRESATURATION fn1 satmode n D wet n sp TRANSMITTER wp tn H sp1 SATURATION fn. Jde n DISr... n sp FRANSMITTER wp rq 499.914 wp1 of -495.1 rf1 cpwr 6.500 rf11 gzlvlE 4830 gtE 0.001000 wc EDratio 1.000 sc gstab 0.000500 wc7 DECOUPLER sc dn C13 v nnn f 1p 2048 DISPLAY Y 1046.1 1667.8 1032.7 1748.3 -869.5 0 -869.5 0 PLOT PLOT wc sc wc2 sc2 vs th ai cdc av 155.0 0 155.0 15722 4



Figure S25: gCOSY spectrum of (±)-3a–d.



Figure S26: ¹H NMR spectrum of (\pm) -3a–d in CDCl₃ : CD₃OD (2:1).



Figure S27: ¹H NMR spectrum of (±)-**3**a–**d** in CDCl₃ : CD₃OD (2:1) – zoom from 5.24 to 7.74 ppm.



Figure S28: ¹H NMR spectrum of (±)-**3**a–**d** in CDCl₃ : CD₃OD (2:1) – zoom from 2.04 to 4.20 ppm.



Figure S29: gHSQCad spectrum of (±)-3a–d.



Figure S30: gHMBCad spectrum of (±)-3a–d.



Figure S31: ¹H NMR spectrum of (±)-2a–d.



Figure S32: ¹H NMR spectrum of (\pm) -2a–d – zoom from 5.30 to 7.38 ppm.



Figure S33: ¹H NMR spectrum of (±)-2a–d – zoom from 2.14 to 3.50 ppm.











Figure S36: gHSQCad spectrum of (±)-2a–d.



Figure S37: gHMBCad spectrum of (±)-2a-d.



Figure S38: ¹³C NMR spectrum of (±)-2a–d.