

# Unexpected degradation of bisphosphonate P-C-P bridge under mild conditions

Petri A. Turhanen\* and Jouko J. Vepsäläinen

*University of Kuopio, Department of Biosciences, Laboratory of Chemistry, P.O. Box 1627, FIN-70211, Kuopio, Finland.*

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**General experimental details:**  $^1\text{H}$ ,  $^{31}\text{P}$  and  $^{13}\text{C}$  NMR spectra were recorded on a Bruker Avance 500 spectrometer operating at 500.1, 202.5 and 125.8 MHz, respectively. TMS or TSP (for  $\text{D}_2\text{O}$  solutions) was used as an internal standard for  $^1\text{H}$  and  $^{13}\text{C}$  measurements, and 85%  $\text{H}_3\text{PO}_4$  was used as an external standard for  $^{31}\text{P}$  measurements. The  $^3J_{\text{HH}}$  couplings are indicated by the letter “*J*”. The  $^nJ_{\text{HP}}$  couplings were calculated from proton spectra and all *J* values are given in Hz. The  $^nJ_{\text{CP}}$  couplings were calculated from proton decoupled carbon spectra with the coupling constants given in parenthesis as hertz. The purity of the compound **1a** was determined from  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra and was ca. 96 %. Electrospray ionization mass spectra were acquired by an LCQ quadrupole ion trap mass spectrometer with an electrospray ionization source.

**Procedure for the preparation of etidronate derivative 1a:** Acetylated etidronic acid (**5**) (150 mg, 0.60 mmol),  $\text{Na}_2\text{CO}_3$  (388 mg, 3.66 mmol) and ethyl chloroformate (3 ml) was refluxed for overnight under the nitrogen atmosphere. Reaction mixture was evaporated to dryness, diethyl ether (8 ml) was added and reaction mixture was stirred for 15 minutes before the precipitate was removed by centrifugation. The remaining solution was evaporated to dryness and compound **1a** was obtained as colorless oil in 55% yield.

**[1-(diethoxycarbonyloxy)phosphoryl-1-acetyloxyethyl]-1-(ethoxycarbonyloxy)phosphonic acid ethyl ester (1a):** Pair of diastereomers (ratio 50:50).  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  4.49-4.41 (m, 2H), 4.36-4.18 (m, 6H), 2.14 (s, 3H), 1.973 (dd,  $^3J_{\text{HP}} = 15.4$ ,  $^3J_{\text{HP}'} = 17.0$ ) and 1.965 (dd,  $^3J_{\text{HP}} = 15.5$ ,  $^3J_{\text{HP}'} = 17.0$ , 3H), 1.41-1.34 (m, 12H).  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ )  $\delta$  168.91 (dd,  $^3J_{\text{CP}} = 7.8$ ,  $^3J_{\text{CP}'} = 8.9$ ), 168.76 (dd,  $^3J_{\text{CP}} = 7.0$ ,  $^3J_{\text{CP}'} = 8.8$ ), 147.67 (d,  $^3J_{\text{CP}} = 7.0$ ), 147.66 (d,  $^3J_{\text{CP}} = 7.3$ ), 147.59 (d,  $^3J_{\text{CP}} = 7.3$ ), 78.79 (dd,  $^1J_{\text{CP}} = 156.1$ ,  $^1J_{\text{CP}'} = 157.4$ ), 78.71 (dd,  $^1J_{\text{CP}} = 155.2$ ,  $^1J_{\text{CP}'} = 159.5$ ), 66.27 (d,  $^2J_{\text{CP}} = 7.6$ ), 66.14 (d,  $^3J_{\text{CP}} = 7.6$ ), 65.82, 65.81, 64.33 (d,  $^4J_{\text{CP}} = 6.9$ ), 64.29 (d,  $^4J_{\text{CP}} = 7.0$ ), 64.23 (d,  $^4J_{\text{CP}} = 7.2$ ), 63.91 (d,  $^4J_{\text{CP}} = 7.5$ ), 21.26, 21.24, 18.61 (t,  $^2J_{\text{CP}} = 2.8$ ), 18.18 (t,  $^2J_{\text{CP}} = 2.0$ ), 16.44 (d,  $^5J_{\text{CP}} = 1.8$ ), 16.39 (d,  $^5J_{\text{CP}} = 2.6$ ), 16.39 (d,  $^5J_{\text{CP}} = 2.6$ ), 16.34 (d,  $^5J_{\text{CP}} = 2.6$ ), 15.99 (d,  $^4J_{\text{CP}} = 6.7$ ), 13.98, 13.97.  $^{31}\text{P}$  NMR ( $\text{CDCl}_3$ )  $\delta$  13.19 d ( $^2J_{\text{PP}} = 22.5$ ) 10.28 and 13.12 d ( $^2J_{\text{PP}} = 21.0$ ) 10.23 d. ESI-MS: 515.3 (M + Na; 100%).

**Typical example of 1a degradation experiment:** Compound **1a** (70.5 mg, 0.143 mmol) was dissolved in MeOH (2 ml), 40% NaOH (58  $\mu\text{l}$ , 4 eq) was added and the reaction mixture was stirred for 0.5 h at room temperature. Reaction mixture was evaporated to dryness and dried in vacuo. Product (ca. 63 mg) was obtained as white powder and contained mixture of acetate and phosphites **2-4** as identified from  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra (see pages S12-13).

**Typical example of 1b degradation experiment:** Compound **1b** (50 mg) was dissolved in  $\text{H}_2\text{O}$  (1 ml), pH was adjusted to  $\geq 11$  with 6 M NaOH (this needed only 1 pasteur pipette drop) and the reaction mixture was stirred for 1 h at room temperature before evaporating it to dryness in vacuo. Product was obtained as white powder and contained mixture of acetate and phosphite **4** as

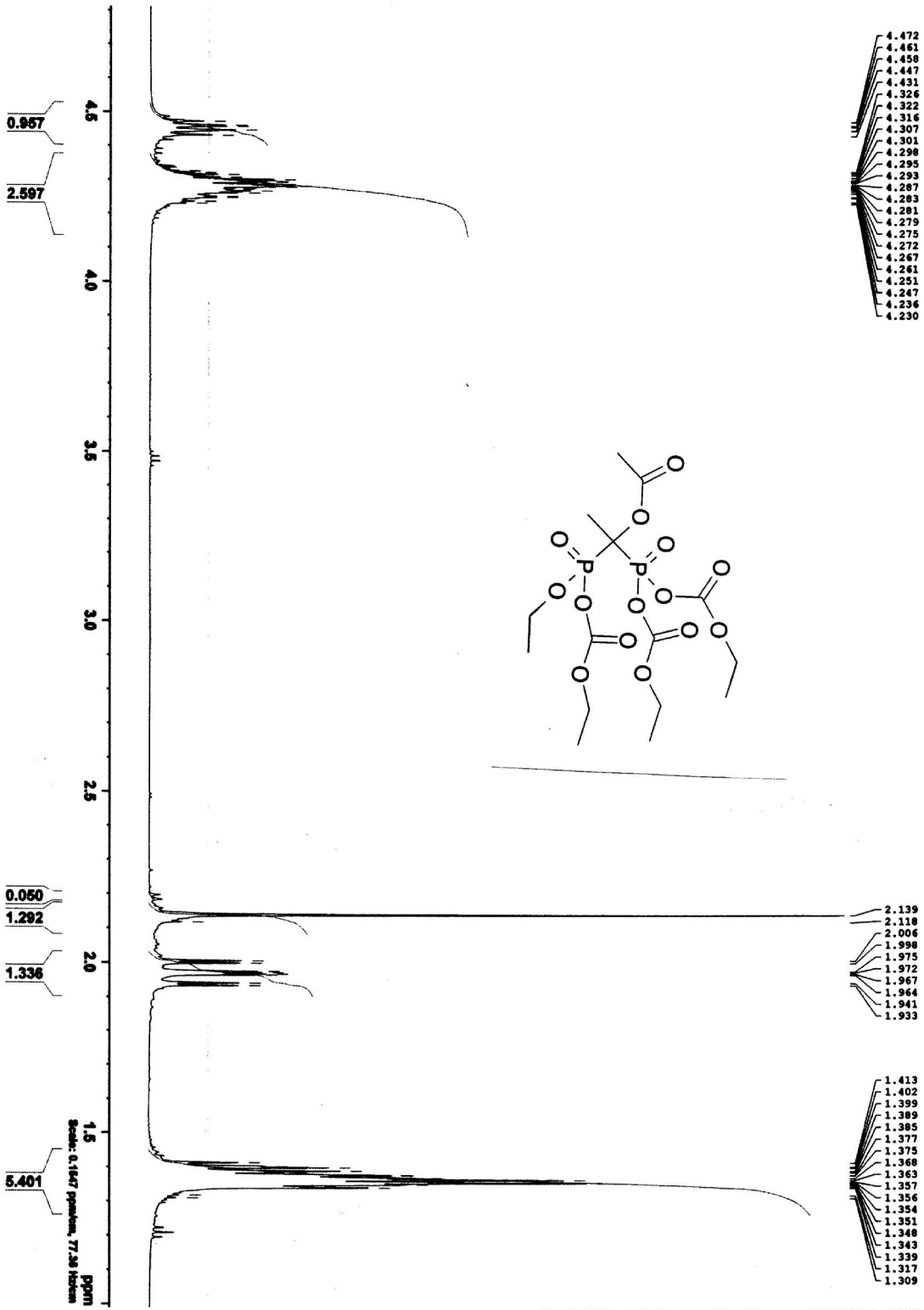
identified from  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra (see pages S14-15; in  $^{31}\text{P}$  NMR spectrum the remaining *H-P* coupling can be seen because of insufficient decoupling power).

**Typical example of 1b degradation experiment in triethyl amine:** Compound **1b** (50 mg) was dissolved in  $\text{H}_2\text{O}$  (1 ml), triethyl amine (122  $\mu\text{l}$ , 5 eq) was added and the reaction mixture was stirred for 1 h at 60  $^\circ\text{C}$  before evaporating it to dryness in vacuo. Product was obtained as slightly yellow solid and contained mixture of acetylphosphonate **7** and phosphite **3** as identified from  $^1\text{H}$  and  $^{31}\text{P}$  NMR spectra (see pages S16-17; in  $^{31}\text{P}$  NMR spectrum the remaining *H-P* coupling can be seen for the compound **3** because of insufficient decoupling power).

Compound 1a

PT/MS-170205-3 cdcl3 tms

<sup>1</sup>H NMR spectrum



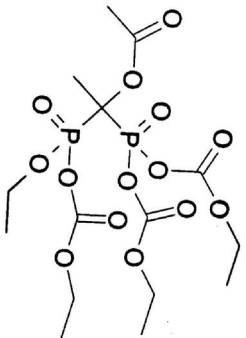
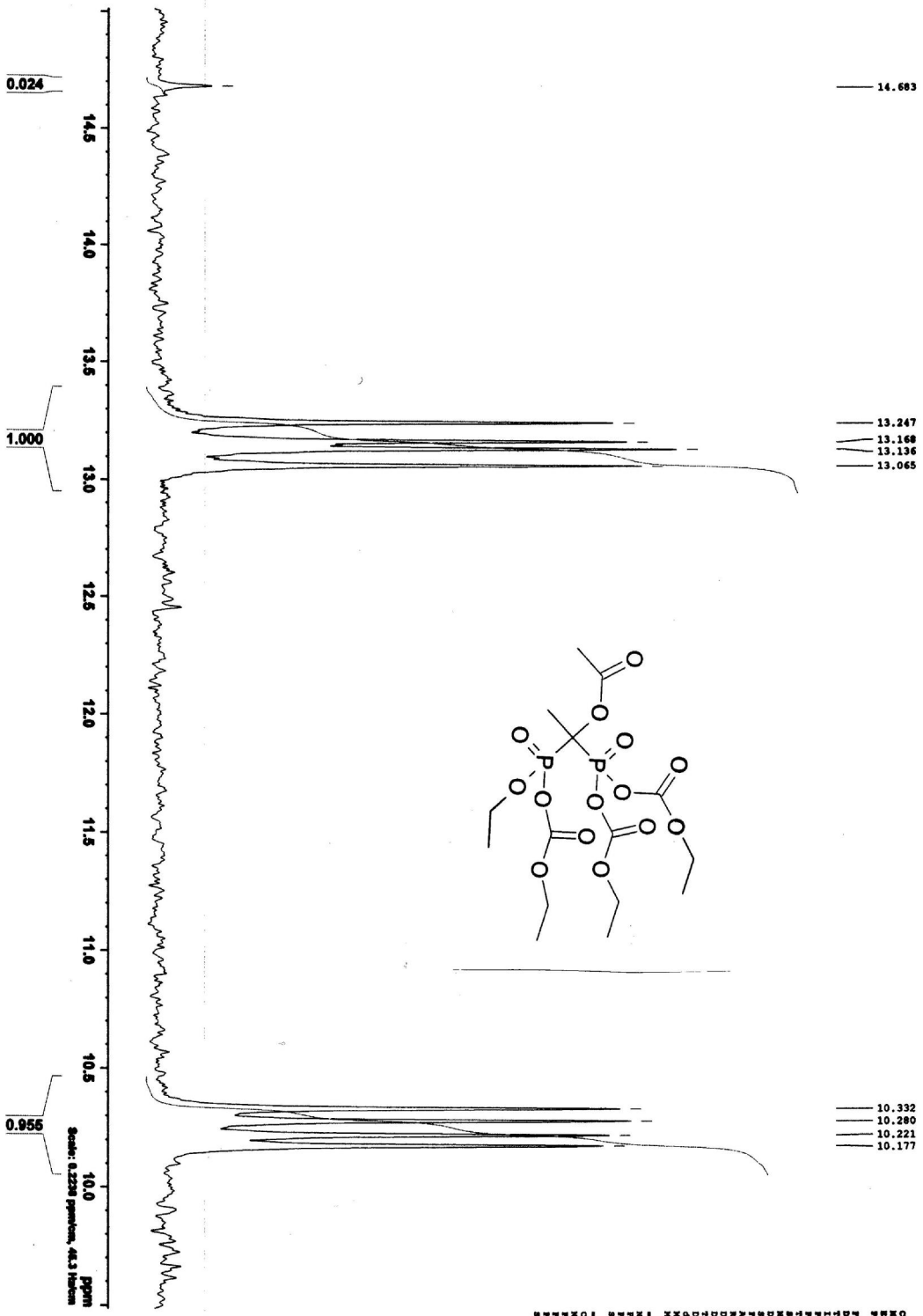
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Compound 1a

PTMMS-170205-3 cdcl3 tms

31P NMR spectrum



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Compound 1a

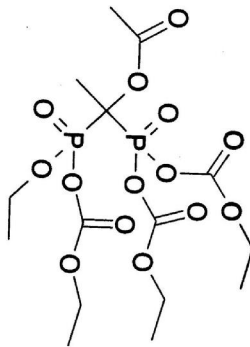
PTMMS-170206-3 cdCl3 rms

<sup>13</sup>C NMR spectrum



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168.925  
168.862  
168.837  
168.782  
168.767  
168.711

147.710  
147.650  
147.632  
147.574



80.061  
79.980  
78.822  
78.806  
78.752  
78.711  
77.617  
77.568  
77.479  
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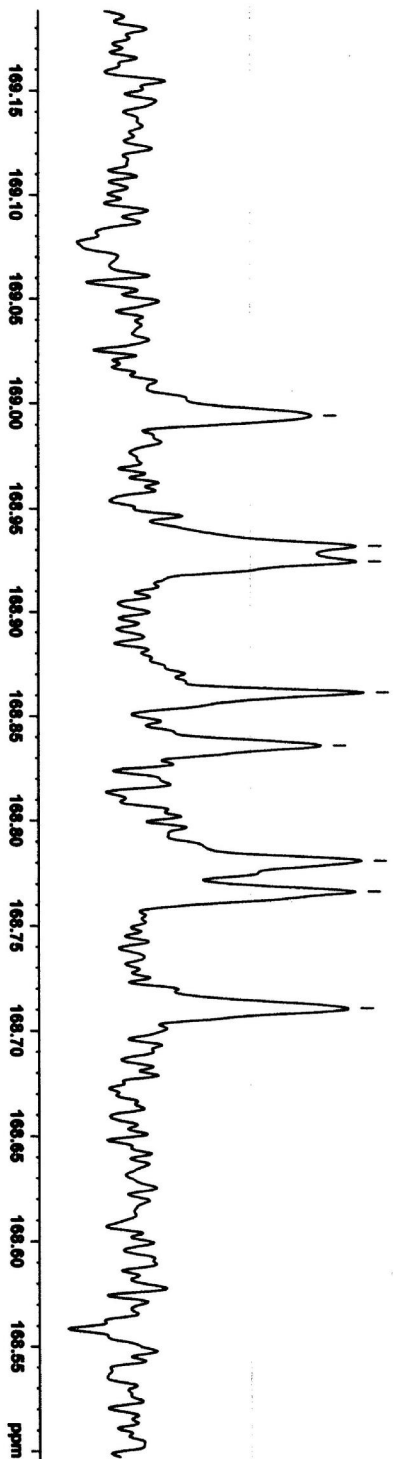
Compound 1a

PTMMS-170205-3 cdcl3 tms

<sup>13</sup>C NMR spectrum



- 168.995
- 168.933
- 168.925
- 168.862
- 168.837
- 168.782
- 168.767
- 168.711



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Compound 1a

PT/MS-170205-3 cdcl3 tms

<sup>13</sup>C NMR spectrum

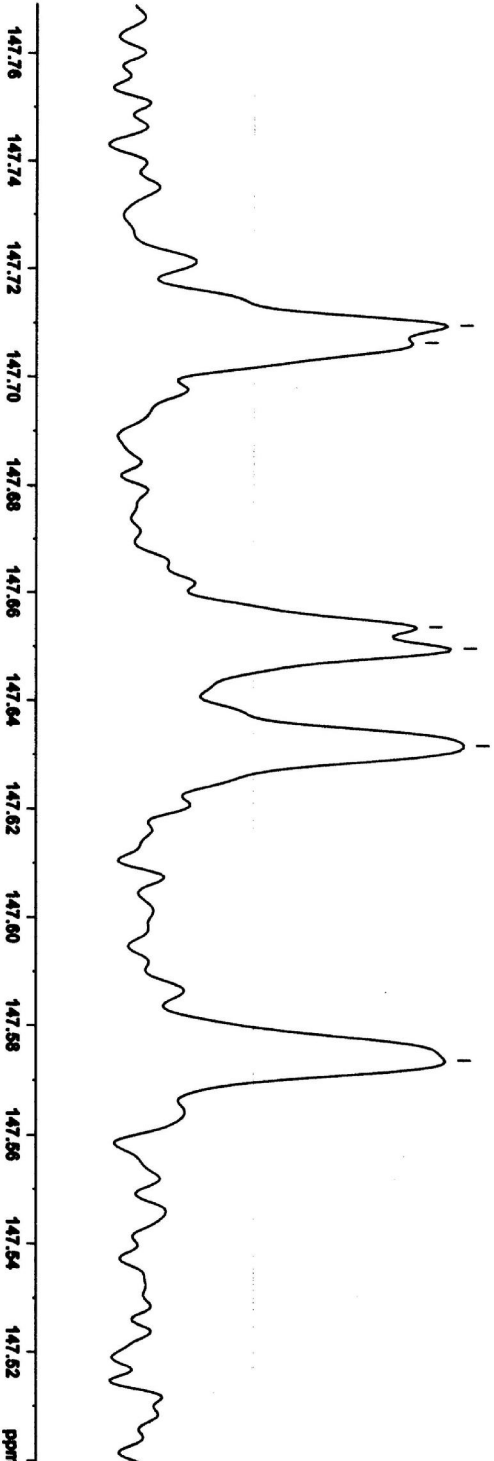


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147.707

147.654  
147.650

147.632

147.574



Current Data Parameters  
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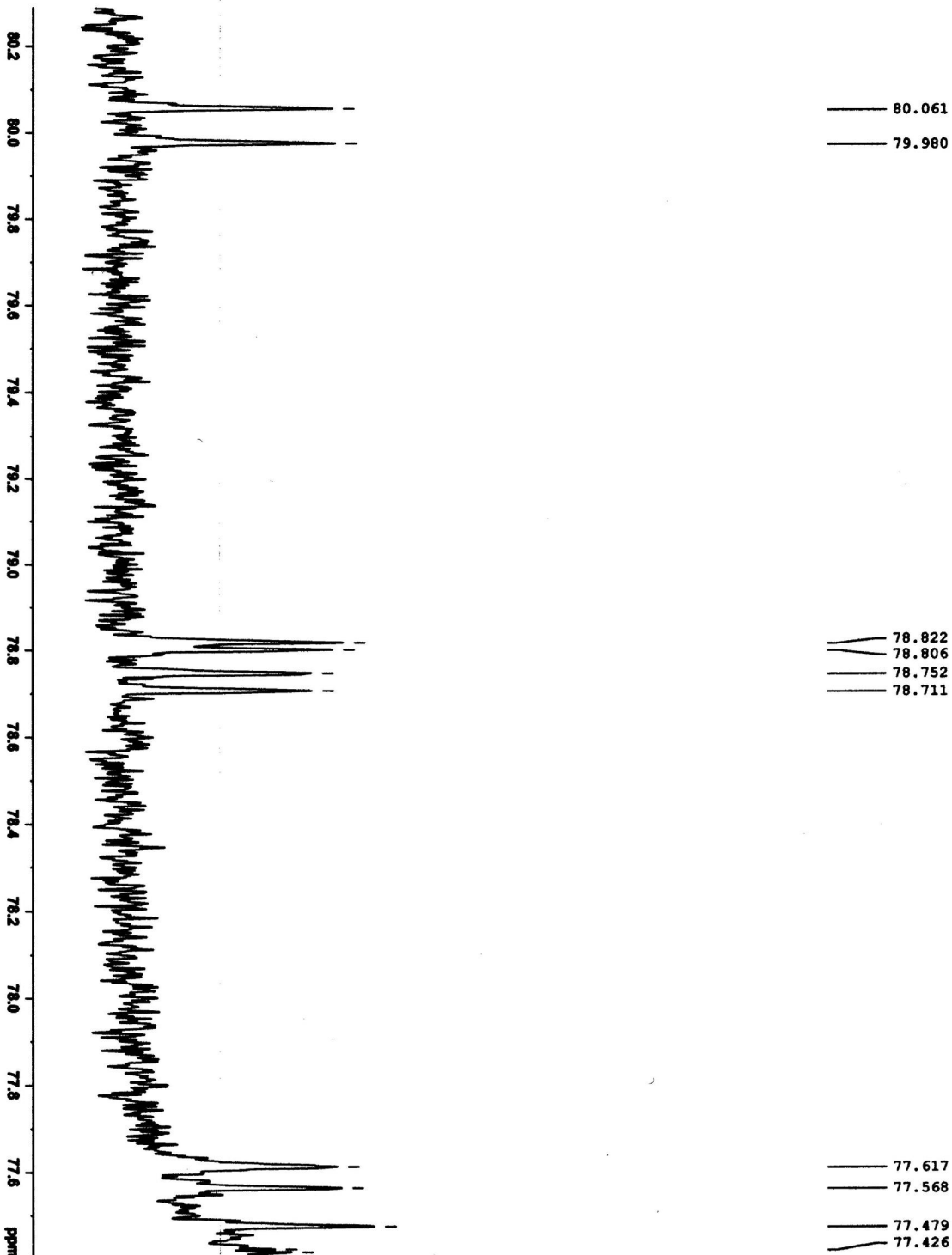
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Compound 1a

PT/MS-170205-3 cdcl3 tms

<sup>13</sup>C NMR spectrum

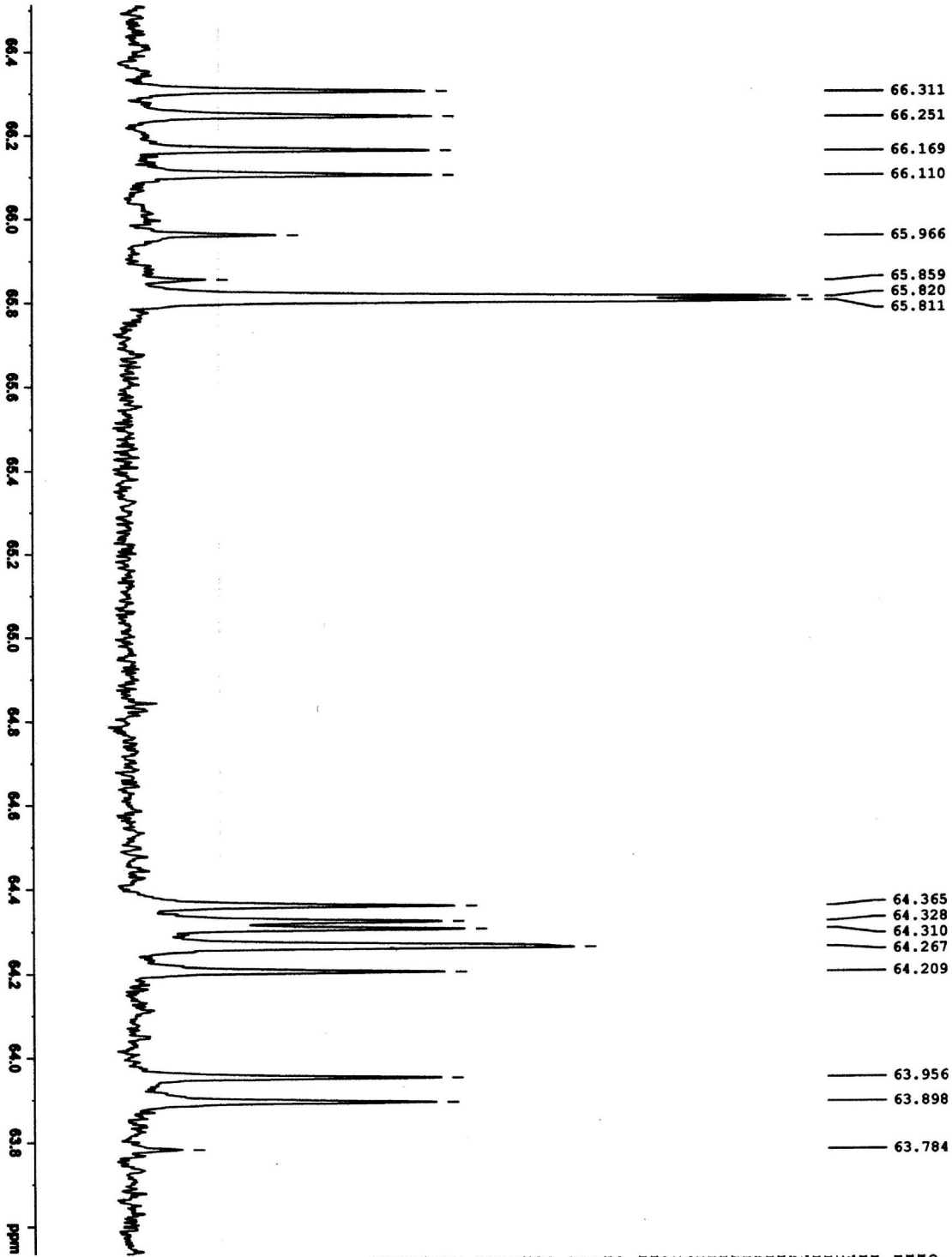


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Compound 1a

PT/MS-170205-3 cdcl3 tms

<sup>13</sup>C NMR spectrum



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Compound 1a

PTMS-170205-3 cdcl3 tms

<sup>13</sup>C NMR spectrum



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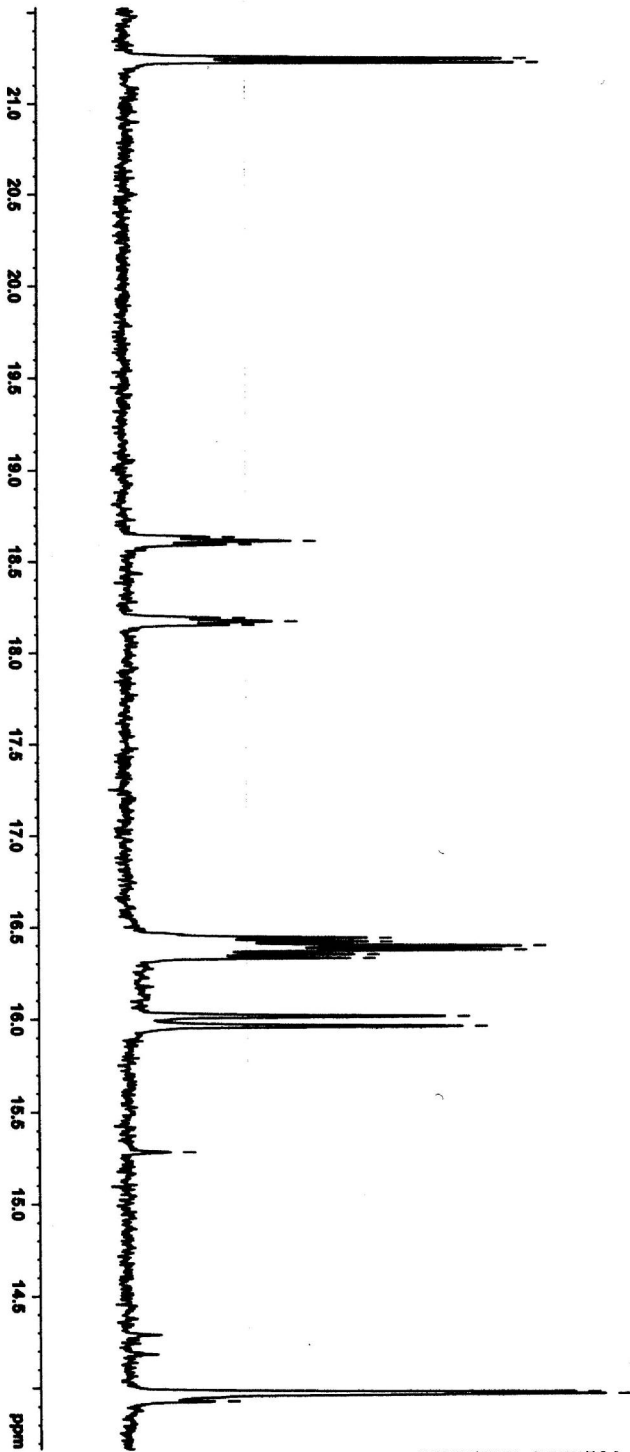
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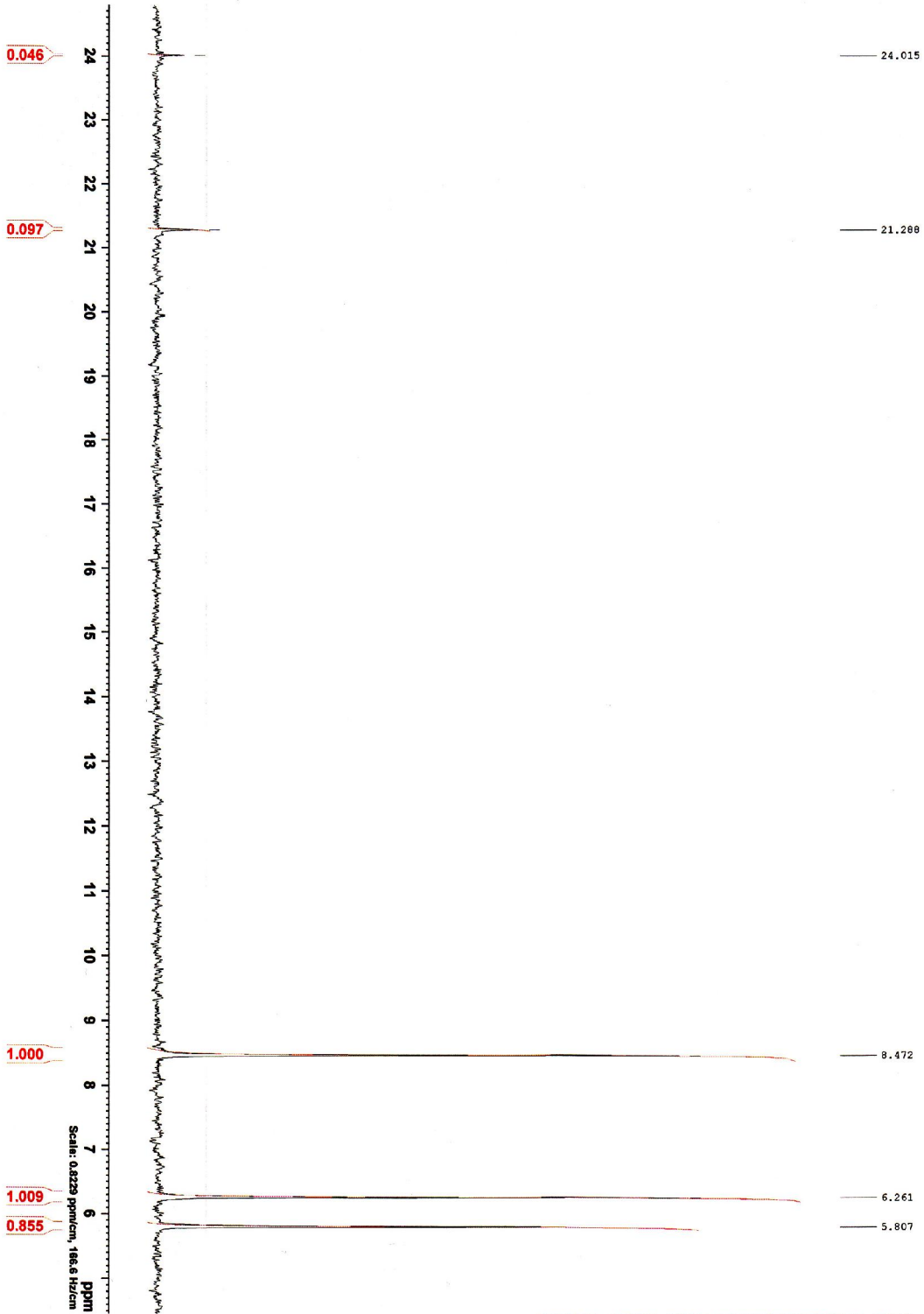


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Typical example of 1a decomposition; 31P NMR spectrum

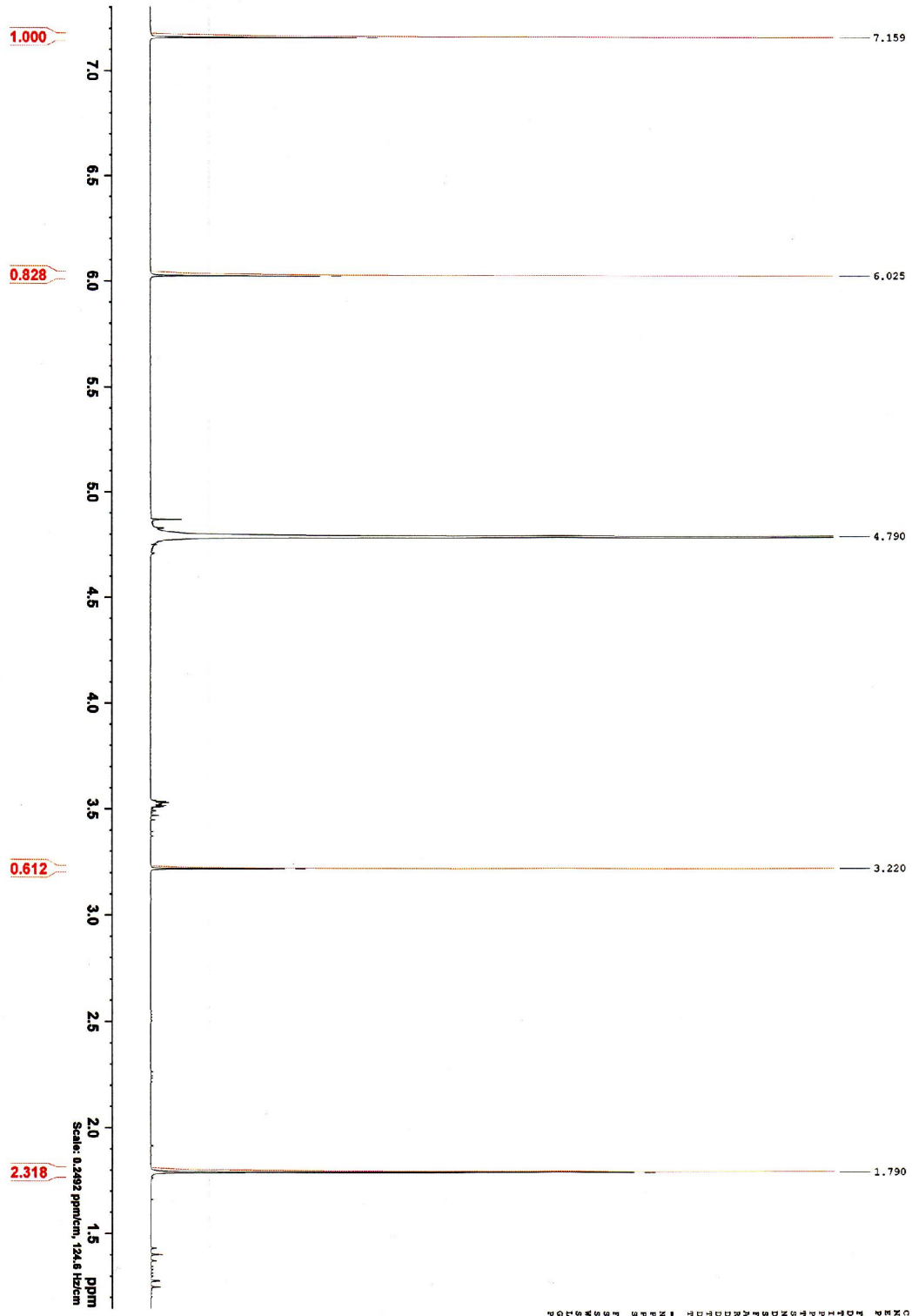


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PT/MS-190906-2b d2o

Typical example of 1b decomposition; 1H NMR spectrum



Current Data Parameters  
NAME: MS-190906-2b  
EXNO: 1  
PROCNO: 2  
F2 - Acquisition Parameters  
Date\_ Time: 2001.13.28  
INSTRUM: spect  
PROBHD: 5 mm QNP 1H/2  
PULPROG: zgpg30  
TD: 65536  
SFO: 500.1364518  
AQ: 1.0000000  
RG: 327.680  
DE: 6.00  
TE: 300.2  
D1: 1.5000000  
D11: 0.0500000  
D12: 0.0500000  
D13: 0.0500000  
D14: 0.0500000  
D15: 0.0500000  
D16: 0.0500000  
D17: 0.0500000  
D18: 0.0500000  
D19: 0.0500000  
D20: 0.0500000  
CHANNEL: f1  
NUC1: 1H  
P1: 8.75  
PL1: 0.00  
SFO1: 500.1364518  
F2 - Processing Parameters  
SI: 32768  
SF: 500.1364518  
AQ: 1.0000000  
RG: 327.680  
DE: 6.00  
TE: 300.2  
D1: 1.5000000  
D11: 0.0500000  
D12: 0.0500000  
D13: 0.0500000  
D14: 0.0500000  
D15: 0.0500000  
D16: 0.0500000  
D17: 0.0500000  
D18: 0.0500000  
D19: 0.0500000  
D20: 0.0500000

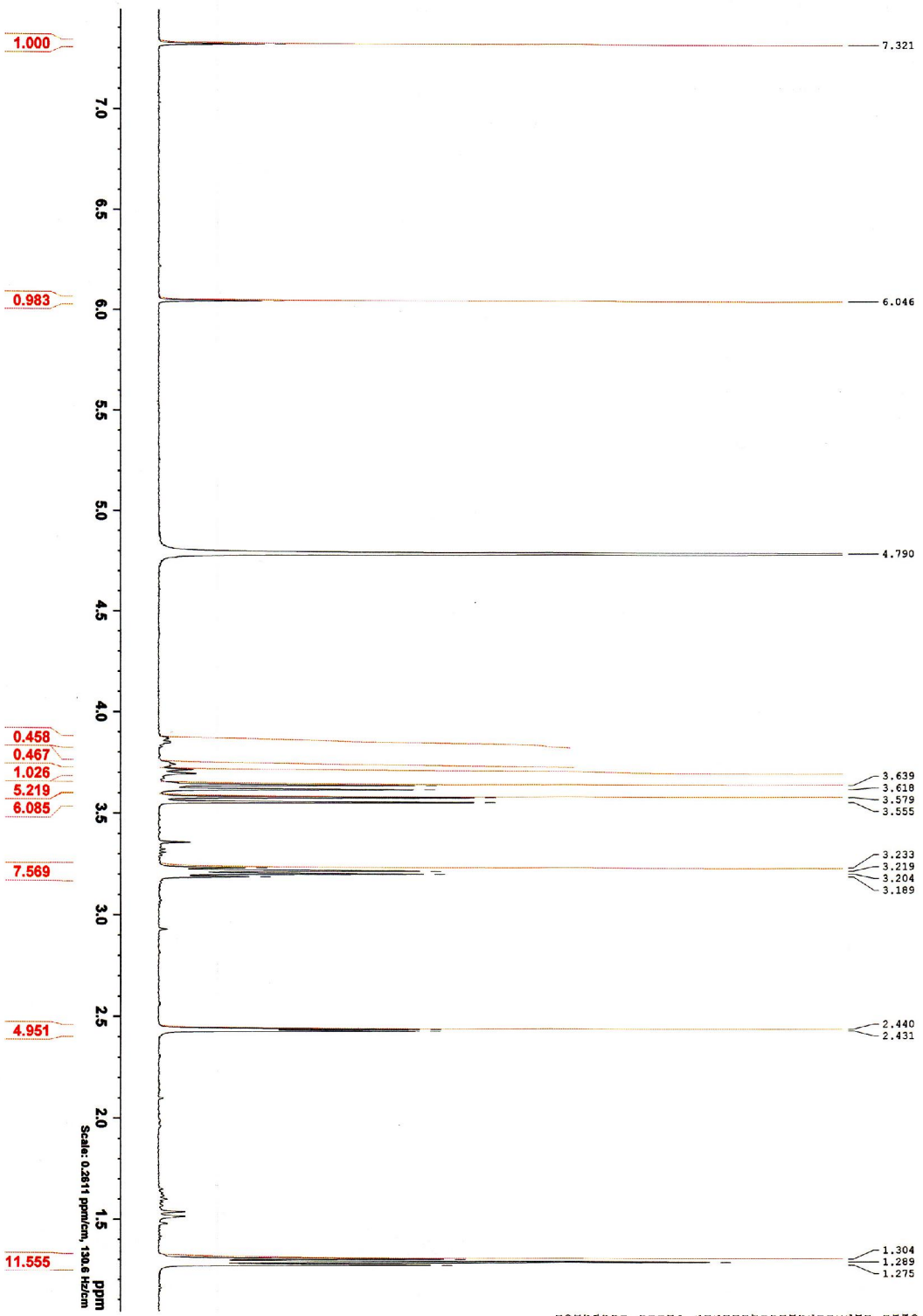
PTMS-190906-2b d2o Typical example of 1b decomposition; 31P NMR spectrum



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=====
Report Date Parameters
NAME MS-190906-2b
EXPTNO 1
PROCNO 1
F2 - Acquisition Parameters
Date_ 20080928
Time 14:28
INSTRUM spect
PROBHD 5 mm QNP 1H/1
PULPROG zgpg30
TD 65536
SOLVENT d2o
NUC1 31P
NUC2 15N
SWH 40830.406 Hz
FIDRES 0.181 Hz
AQ 1.999931 sec
RG 2048
AQ 12.2048 usec
DE 6.00 usec
TE 293.2 K
D1 0.3000000 sec
d11 0.0300000 sec
TDO 1
=====
CHANNEL F1
NUC1 31P
P1 12.00 usec
PL1 0.00 dB
SFO1 202.1586930 MHz
=====
CHANNEL F2
CPDPRG2 waltz16
NUC2 15N
P2 90.00 usec
PL2 24.00 dB
SFO2 500.1325007 MHz
=====
F2 - Processing parameters
    
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PTMS-190107-1a d2o Typical example of 1b decomposition in triethyl amine; 1H NMR spectrum

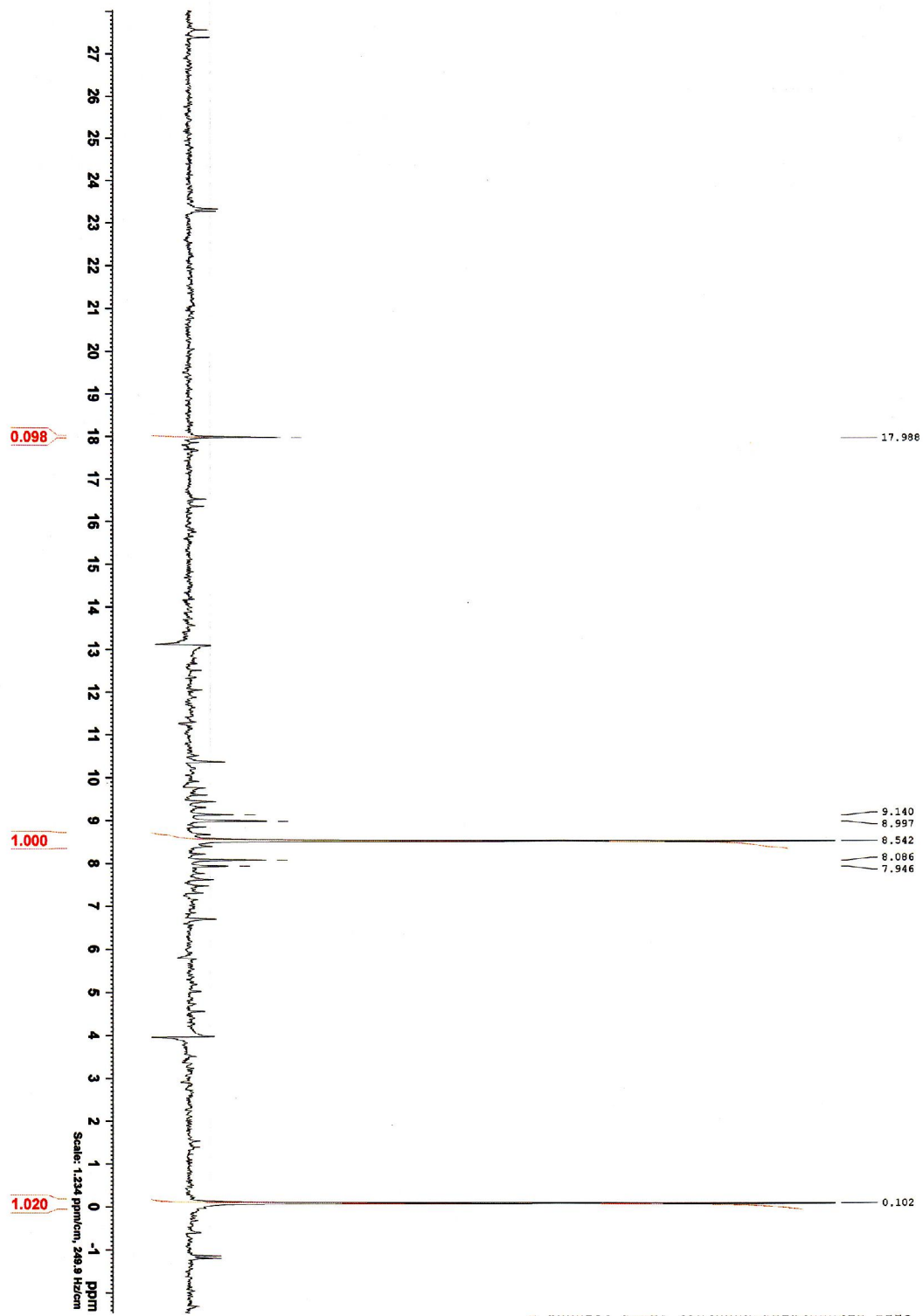


Current Data Parameters  
 Name: PTMS-190107-1a  
 ExpNO: 2  
 PROCNO: 1  
 F2 - Acquisition Parameters  
 Date\_: 2007123  
 Time: 12.43  
 INSTRUM: spect  
 PROBRD: 5 mm QNP 1H/1  
 TD: 65536  
 TO: 65536  
 SOLVENT: D2O  
 NS: 0  
 DS: 0  
 SWH: 10000.000 Hz  
 FIDRES: 0.000000 Hz  
 AQ: 59.64 usec  
 RG: 64  
 DE: 3.2769001 sec  
 TE: 300.2 K  
 T1: 300.0 K  
 T1RHO: 1.50000001 sec  
 T1rho: 1.50000001 sec  
 ===== CHANNEL f1 =====  
 NUCL1: 1H  
 P1: 8.75 usec  
 PL1: 0.00 dB  
 SFO1: 500.1360010 MHz  
 F2 - Processing parameters  
 SI: 32768  
 SF: 500.1360010 MHz  
 SD: 32768  
 SFO: 500.1360010 MHz  
 AS: 0.30 Hz  
 PC: 5.00



PTMS-190107-1a d2o

Typical example of 1b decomposition in triethyl amine; 31P NMR spectrum



===== CHANNEL F1 =====  
NUC1 31P  
PUL 8  
PL 10.00 dB  
SFO1 202.4586930 MHz  
===== CHANNEL F2 =====  
CPDPRG2 waltz16  
RFRF2 50.00 usec  
PL2 24.00 dB  
SFO2 500.132607 MHz  
===== Processing parameters =====  
Date\_ 2007122  
Time\_ 10.42  
INSTRUM spect  
PROBHD 5 mm QNP 1H/1  
PULPROG zgpg30  
TD 65536  
SOLVENT D2O  
NS 1  
DS 0  
SWH 40650.406 Hz  
FIDRES 0.001351 sec  
AQ 2.048 usec  
RG 2048  
DE 6.00 usec  
TE 300.1 K  
DQ 0.000000 sec  
d11 0.0300000 sec  
TPO 1