

**Supporting Information**

for

**Molecular solubilization of fullerene C<sub>60</sub> in water by**

**γ-cyclodextrin thioethers**

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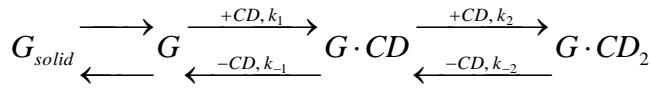
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**Detailed dissolution kinetics**

## 2-step model



### Approximations:

concentrations:  $[G] = \text{const} = [G]_0, [CD] = \text{const} = [CD]_0$

stationarity of  $G \cdot CD$ :  $[G \cdot CD_2] \gg [G \cdot CD] \approx \text{const}$

rate determining steps:  $k_1[G]_0[CD]_0 \ll k_2[G \cdot CD][CD]_0, k_2[G \cdot CD_2] \ll k_{-1}[G \cdot CD]$

rate equation:  $\frac{d[G \cdot CD_2]}{dt} = k_1[CD]_0[G]_0 - k_{-2}[G \cdot CD_2]$

equilibrium state:  $k_1[CD]_0[G]_0 \equiv k_{-2}[G \cdot CD_2]_{equ}$

equilibrium constant:  $K' = \frac{k_1}{k_{-2}} = \frac{[G \cdot CD_2]_{equ}}{[G][CD]} \approx \frac{[G \cdot CD_2]_{equ}}{[G]_0[CD]_0}$

rate equation of complexation:

$$\frac{d[G \cdot CD_2]}{dt} = k_1[G]_0[CD]_0 - k_{-2}[G \cdot CD_2] = k_1[G]_0[CD]_0(1 - c'[G \cdot CD_2])$$

$$\text{with } c' = \frac{1}{[G]_0[CD]_0 K'} = \frac{1}{[G \cdot CD_2]_{equ}}$$

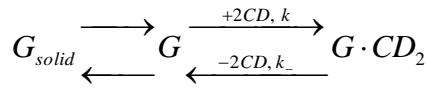
$$\text{integrated form: } \frac{\ln(1 - c'[G \cdot CD_2])}{c'} = -k_1[G]_0[CD]_0 t$$

$$\text{kinetics of complexation: } [G \cdot CD_2] = (1 - \exp(-k_1 c'[G]_0[CD]_0 t)) / c'$$

$$[G \cdot CD_2] = [G \cdot CD_2]_{equ}(1 - \exp(-k_{-2}t))$$

$$\text{initial slope: } \left( \frac{d[G \cdot CD_2]}{dt} \right)_{t=0} = [G \cdot CD_2]_{equ} k_{-2} = k_1[G]_0[CD]_0$$

## 1-step model



### Approximations:

concentrations:  $[G] = \text{const} = [G]_0, [CD] = \text{const} = [CD]_0$

rate equation:  $\frac{d[G \cdot CD_2]}{dt} = k_+[CD]_0^2[G]_0 - k_- [G \cdot CD_2]$

equilibrium state:  $k_+[CD]_0^2[G]_0 \equiv k_- [G \cdot CD_2]_{equ}$

equilibrium constant:  $K = \frac{k_+}{k_-} = \frac{[G \cdot CD_2]_{equ}}{[G]_0 [CD]_0^2} \cong \frac{[G \cdot CD_2]_{equ}}{[G]_0 [CD]_0^2}$

Rate equation of complexation:

$$\frac{d[G \cdot CD_2]}{dt} = k_+[G]_0 [CD]_0^2 - k_- [G \cdot CD_2] = k_+[G]_0 [CD]_0^2 (1 - c [G \cdot CD_2])$$

$$\text{with } c = \frac{1}{[G]_0 [CD]_0^2 K} = \frac{1}{[G \cdot CD_2]_{equ}}$$

$$\text{integrated form: } \frac{\ln(1 - cx)}{c} = -k_+[G]_0 [CD]_0^2 t$$

$$\text{Kinetics of complexation: } [G \cdot CD_2] = (1 - \exp(-kc[G]_0 [CD]_0 t)) / c$$

$$[G \cdot CD_2] = [G \cdot CD_2]_{equ} (1 - \exp(-k_- t))$$

$$\text{Initial slope: } \left( \frac{d[G \cdot CD_2]}{dt} \right)_{t=0} = [G \cdot CD_2]_{equ} k_- = k_+[G]_0 [CD]_0^2$$