

Open Access

Amyloid-β probes: Review of structure–activity and brain-kinetics relationships

Todd J. Eckroat^{1,2}, Abdelrahman S. Mayhoub^{2,§} and Sylvie Garneau-Tsodikova^{*1,¶}

Review

Address:

¹Department of Pharmaceutical Sciences, University of Kentucky, 789 South Limestone Street, Lexington, KY, 40536-0596, United States and ²Life Sciences Institute and Department of Medicinal Chemistry, University of Michigan, 210 Washtenaw Ave, Ann Arbor, MI, 48109-2216. United States

Email:

Sylvie Garneau-Tsodikova* - sylviegtsodikova@uky.edu

* Corresponding author

§ On leave from Faculty of Pharmacy, Al-Azhar University, Cairo, 11884. Egypt

¶ Phone: 859-218-1686

Keywords:

Alzheimer's disease; in vivo detection; near-infrared fluorescence probes; PET/SPECT imaging; radioactive probes

Beilstein J. Org. Chem. 2013, 9, 1012-1044.

doi:10.3762/bjoc.9.116

Received: 26 January 2013 Accepted: 30 April 2013 Published: 28 May 2013

This article is part of the Thematic Series "Synthetic probes for the study of biological function".

Guest Editor: J. Aubé

© 2013 Eckroat et al; licensee Beilstein-Institut. License and terms: see end of document.

Abstract

The number of people suffering from Alzheimer's disease (AD) is expected to increase dramatically in the coming years, placing a huge burden on society. Current treatments for AD leave much to be desired, and numerous research efforts around the globe are focused on developing improved therapeutics. In addition, current diagnostic tools for AD rely largely on subjective cognitive assessment rather than on identification of pathophysiological changes associated with disease onset and progression. These facts have led to numerous efforts to develop chemical probes to detect pathophysiological hallmarks of AD, such as amyloid-β plaques, for diagnosis and monitoring of therapeutic efficacy. This review provides a survey of chemical probes developed to date for AD with emphasis on synthetic methodologies and structure–activity relationships with regards to affinity for target and brain kinetics. Several probes discussed herein show particularly promising results and will be of immense value moving forward in the fight against AD.

Introduction

Alzheimer's disease (AD) is a progressive neurodegenerative disorder of the central nervous system currently affecting ~5.4 million Americans, a number that could increase to 11–16 million by the year 2050. In the United States, AD repre-

sents the 6th leading cause of death. Between 2000 and 2008, the number of deaths caused by AD increased by 66%, a dramatic rise, especially when compared to other causes of death, such as heart disease, stroke, prostate and breast cancer,

and HIV, which decreased by 3–29% during that time period [1]. As these numbers indicate, AD represents a significant and increasing burden on our population, and efforts towards the development of new and improved diagnostics and therapeutics for this devastating disease are important research endeavors.

Several pathological hallmarks of AD have been identified, and they include decreased cholinergic neurons and acetylcholine (ACh) levels, plaques caused by aggregation of the protein fragment amyloid- β (A β), tangles associated with irregular phosphorylation of tau protein, inflammation and increased oxidative stress from reactive oxygen species (ROS), as well as dyshomeostasis and miscompartmentalization of metal ions such as Cu, Fe, and Zn. Observations of these hallmarks have led to several hypotheses in attempts to explain the underlying cause of the disease, which is likely multifactorial. However, the exact cause of AD still remains unknown.

Postmortem histopathological examination of A β plaques is currently the only way to firmly confirm AD [2]. In view of the limited accessibility to living brain and other central nervous system (CNS) tissues, AD is currently diagnosed through memory tests and/or based on the patients' history [2]. Obviously these kinds of diagnostic tools lack absolute sensitivity and accuracy, especially in the early stages of the disease. Therefore, as A β plaques precede the onset of dementia and cognitive decline in AD patients, their detection by nuclear imaging techniques such as positron emission tomography (PET) or single-photon emission computed tomography (SPECT) represents the presymptomatic diagnostic tool of choice for AD [3-5].

The presence of different binding sites in AB aggregates led medicinal chemists to investigate and develop a variety of chemical scaffolds as Aβ-imaging tracers [6-9]. To provide a high readable signal-to-background ratio, the ideal Aβ-imaging probes should have certain brain kinetics: a rapid initial brain uptake and a fast washout. Early efforts towards developing AB stains focused on dyes such as congo red (1), chrysamine G (2), pinacyanol (3), and thioflavin-T (4) (Figure 1A). However, the bulky and ionic natures of these dyes prevented them from crossing the blood brain barrier (BBB), and consequently, no in vivo benefits were obtained from these initial investigations [10,11]. During the past decade, efforts directed at developing probes that display uptake and retention that differ in healthy and AD-affected brains resulted in a variety of radiolabeled molecular probes for in vivo PET/SPECT imaging. The scaffolds from which these newer radiolabeled probes are derived include chalcone (5) and its conformationally restricted analogues flavone (6) and aurone (7); stilbene (8) and its analogues diphenyl-1,2,4-oxadiazole (9) and diphenyl-1,3,4oxadiazole (10); and thioflavin-T analogues such as benzothiazole (11), benzoxazole (12), benzofuran (13), imidazopyridine (14), and benzimidazole (15); as well as quinoline (16) and naphthalene (17) derivatives (Figure 1B). In this review, we provide an overview of these AD radiolabeled early-diagnostic probes according to their scaffolds, with a special emphasis on their synthesis as well as their structure–activity and brain-kinetics relationships. We also provide a brief summary of the latest developments related to the detection of A β plaques by near-infrared fluorescence (NIRF) imaging.

Review

Radiolabels used in PET/SPECT molecular probes

Even though they decay rapidly, [11 C] ($t_{1/2} = 20$ min) and [18 F] ($t_{1/2} = 110$ min) are the most commonly used radiolabels in PET/SPECT molecular probes for in vivo imaging of A β plaques [4]. With a half-life ($t_{1/2}$) of 6.01 h compatible with the localization and residence time necessary for imaging, technetium-99m [99 mTc] is also a radionuclide of choice that is easily produced by a 99 Mo/ 99 mTc generator [12]. Iodine isotopes such as [125 I] are also employed, although much less frequently [13]. The general synthetic methods utilized to introduce radiolabels into PET probes are outlined in Scheme 1. These general strategies will be abbreviated as **Gs A–D** in all subsequent schemes in this review.

Chalcone and its conformationally restricted analogues

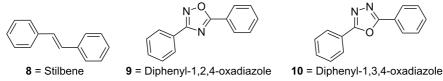
Chalcone derivatives

Chalcones and indolochalcones, such as 18a-l, 19a,b, 20a,b, and 21, have been widely reported as Aβ-imaging tracers (Scheme 2A). Structure–activity-relationship (SAR) studies on fluorinated chalcones 18a-l have shown that, in general, chalcones with tertiary amines in their structures demonstrate good affinity for A β plaques in in vitro models ($K_i = 20-50 \text{ nM}$) (Table 1) [14]. Dimethylation of the amino group seems to be crucial for Aβ binding, since analogues with free amino groups or monomethyl amino groups revealed lower affinity [14]. On the other hand, pegylation is not that essential for plaque binding as tertiary amine analogues with different degrees of pegylation (n = 0-3) all showed similar affinity. In biodistribution experiments using normal mice, the [18F]-labeled chalcone 19a showed high brain uptake rate and good clearance, whereas the [11C]-labeled chalcone 19b revealed reasonable brain uptake rate, but very fast clearance [14]. The [18F]-labeled and [11C]-labeled chalcones 19a and 19b were synthesized using similar methods, and a representative synthesis of 19a is shown (Scheme 2B). Aldol condensation between the appropriate acetophenone 22 and benzaldehyde 23 afforded the chalcone backbone, which was subsequently pegylated to give 24 and

B Newer scaffolds explored for development of radiolabeled Aβ probes:

Chalcone and its conformationally restricted analogues:

Stilbene and its analogues:



Thioflavin-T analogues:

Quinoline and naphthalene analogues:

Figure 1: Structures of A. dyes originally used to stain Aβ and B. newer scaffolds explored for the development of radiolabeled Aβ probes.

radiolabeled to give 19a. Compound 19b was generated by using p-nitrobenzaldehyde instead of the corresponding dimethylamine 23 used in the preparation of 19a [14]. The resultant nitrochalcone was then reduced by $SnCl_2$ in EtOH to yield the free amine, which was monomethylated by controlled addition of an equimolar amount of MeI. The final [^{11}C]-

labeled compound **19b** was produced by reacting [¹¹C]CH₃OTf with the secondary amine precursor.

[Re]- and [99 mTc]-labeled chalcone analogues **20a** and **20b** were also studied (Scheme 2A) [12]. The [Re]-labeled analogue **20a** displayed higher affinity for A β plaque than did the corres-

Scheme 1: General synthetic strategies (Gs) used to introduce A. ¹⁸F, B. ¹¹C, C. ^{99m}Tc/Re, and D. ¹²³I and ¹²⁵I radiolabels into PET probes. Note: Gs A–D will be used in all subsequent schemes to describe these general synthetic strategies.

[¹²⁵]]₂, CH₂Cl₂ or [¹²³]]Nal AcOOH, H₃PO₄

ponding [99mTc]-derived compound **20b**. However, **20b** showed better brain pharmacokinetics than **20a**, as indicated by its high brain-uptake rate (1.48% ID/g) and rapid wash out from the CNS (0.17% ID/g at 60 min). Compounds **20a** and **20b** were synthesized by reacting a Boc-protected metal chelator (Scheme 1C) with 4-*O*-(bromopropyl)hydroxychalcone **25** (Scheme 2C). After removal of the Boc protecting group, the final [Re]- and [99mTc]-labeled chalcones **20a** and **20b** were obtained by treatment with (PPh₃)₂ReOCl₃ and ^{99m}TcGH, respectively [12].

Finally, the radioiodinated indolochalcone 21, among a series of other derivatives, was prepared through condensation of 4-iodoacetophenone (26) and indole-5-carboxaldehyde (27) to give 28, which was radiolabeled to give the target compound (Scheme 2D) [15]. The indolochalcone 21 showed good binding

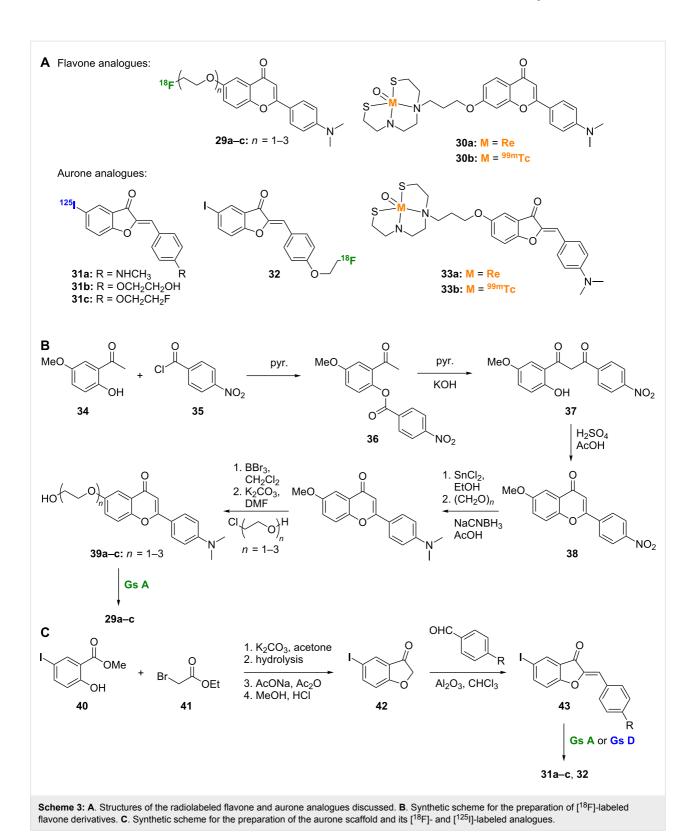
affinity for $A\beta_{1-42}$ aggregates with a $K_i < 10$ nM. Replacement of the iodo substituent with a chloro, bromo, methoxy, or dimethylamino substituent all gave similar results, but replacement with a fluoro, hydroxy, amino, or methylamino substituent all reduced affinity to varying degrees. Autoradiography in sections of brain tissue from an AD animal model showed that 21 specifically labeled A β plaques, but its efficacy was hampered by low in vivo uptake into the brain (0.41% ID/g at 2 min) [15].

Conformationally restricted chalcones: flavones and aurones

Flavones and aurones, such as 29a-c, 30a,b, 31a-c, 32, and 33a,b (Scheme 3A), can be classified as conformationally restricted chalcone derivatives as their basic structures result from insertion of an oxygen atom between the double bond and

Scheme 2: A. Structures of radiolabeled chalcone analogues discussed. B.–D. Synthetic schemes for the preparation of radiolabeled chalcones 19a and 20a,b, and indolochalcone 21.

Compound	n	R ¹	R^2	R^3	$A\beta_{1-42} K_i (nM)$	%ID/g at 2 min	%ID/g at 30 min
18a	1	F	Me	Me	45.7 ± 7.1	_	_
[¹¹ C] 18a	1	F	Me	¹¹ CH ₃	_	6.01 ± 0.61	2.26 ± 0.41
18b	2	F	Me	Me	20.0 ± 2.5	_	_
[¹¹ C] 18b	2	F	Me	¹¹ CH ₃	_	4.73 ± 0.47	1.00 ± 0.19
18c	3	F	Me	Me	38.9 ± 4.2	_	_
19a	3	¹⁸ F	Me	Me	_	3.48 ± 0.47	1.07 ± 0.17
19b	3	F	Me	¹¹ CH ₃	_	4.31 ± 0.33	0.35 ± 0.03
18d	1	F	Н	Н	678.9 ± 21.7	_	_
18e	2	F	Н	Н	1048.0 ± 114.3	_	_
18f	3	F	Н	Н	790.0 ± 132.1	_	_
18g	1	F	Н	Me	197.1 ± 58.8	_	_
18h	2	F	Н	Me	216.4 ± 13.8	_	_
18i	3	F	Н	Me	470.9 ± 100.4	_	_
18j	0	F	Me	Me	49.8 ± 6.2	_	_
[¹¹ C] 18j	0	F	Me	¹¹ CH ₃	_	3.68 ± 0.35	1.04 ± 0.20
18k	0	F	Н	Н	663.0 ± 88.3	_	_
181	0	F	Н	Me	234.2 ± 44.0	_	_



the phenyl ring attached to the carbonyl group of the chalcone scaffold (Figure 1B, with oxygen atoms depicted in red). The affinity of flavonoids towards $A\beta$ aggregates was first established by using fluorescence staining in brain sections of

Tg2576 transgenic mice [10]. The absence of spots in wild-type mouse brain sections indicated the specificity of flavonoids towards A β aggregates in AD mouse models. The [18 F]-labeled pegylated flavones **29a**–**c** showed high affinity towards A β

aggregates with K_i values ranging between 5.3 nM for 29a and 19.3 nM for 29c (Scheme 3A) [16]. SAR studies suggest that, as with chalcones, the tertiary amine in these flavones was important for binding and tracing Aß aggregates in mouse models, as they consistently outperformed secondary and primary amine analogues [16]. Also as with chalcones, the degree of pegylation had only minor effects on binding properties. Compounds 29a-c showed uptake rates indicative of high to sufficient levels for brain imaging (2.89-4.17% ID/g at 2 min) and moderate clearance rates [16]. The flavone backbone of 29a-c was built by acylating 2-hydroxy-5-methoxyacetophenone (34) with 4-nitrobenzoyl chloride (35) and subjecting the resulting 2-acyloxyacetophenone (36) to Baker-Venkataraman rearrangement [17] to afford the 1,3-diarylpropane-1,3-dione 37, which was dehydrated with sulfuric acid to give 38. Subsequent nitro reduction, reductive methylation, methyl ether cleavage, and pegylation gave the nonlabeled precursors 39a-c. The [18F]label was introduced by using the standard [K/K222]¹⁸F in DMSO and acetonitrile reaction conditions (Scheme 3B) [16]. The [Re]- and [99mTc]-labeled flavone complexes 30a and 30b were also prepared by using the procedure described for the synthesis of the [Re]- and [99mTc]-labeled chalcones 20a and 20b (Scheme 2). The [99mTc]-labeled flavone complex 30a displayed high Aβ plaque affinity but limited brain uptake [18].

Aurone derivatives have been investigated for their A β plaque binding affinity [19]. The [125 I]-labeled methylamine aurone **31a** presented great binding affinity to A β aggregates ($K_i = 1.2$ nM), better than all reported flavones to date. It also showed rapid brain uptake rate (3.17% ID/g at 2 min) and rapid clearance (0.24% ID/g at 60 min) [19]. The effect of the tertiary amine in this aurone scaffold was less pronounced than that seen with chalcones or flavones. The dimethylamine analogue of **31a** had approximately six times weaker binding affinity, while the free amine analogue showed only two times weaker affinity. To further enhance the A β plaque traceability of **31a**, its methylamine moiety was replaced with ethylene oxide to provide compound **31b**, which exhibited a K_i value of 1.05 nM in an in vitro binding assay [13]. The brain kinetics of **31b** (brain uptake = 4.51% ID/g at 2 min and washout = 0.09% ID/g

at 60 min) were found to be slightly better than those of 31a [13]. Addition of 2 or 3 ethylene oxide units or replacement with a hydroxy or methoxy group did not significantly improve the plaque binding affinity and modestly affected the brain kinetics. Replacement of the terminal hydroxy group of 31b with a fluorine atom negatively affected the brain uptake property of compound 31c (2.34% ID/g at 2 min) when compared to 31b [20]. However, it did not affect the washout character of the compound. These results were confirmed by preparation and analysis of the [18F]-labeled compound 32 [20]. In general, the aurone derivatives 31a-c and 32 were built from the reaction of methyl 5-iodosalicilate (40) with ethyl bromoacetate (41) followed by ester hydrolysis and cyclization to afford 5-iodo-3coumaranone (42), which, after condensation with the proper benzaldehydes, gave the aurone scaffold 43, which could be radiolabeled (Scheme 3C). As for the chalcone and flavone derivatives, [Re]- and [99mTc]-labeled aurone complexes 33a and 33b were also prepared [18]. The high affinity for Aβ aggregates observed with the [99mTc]-labeled aurone 33b was hampered by its weak brain penetration, which made it unsuitable for in vivo application [18].

Stilbene and its analogues Stilbene derivatives

The SARs of stilbene analogues, such as 44a-f, 45, and 46a,b (Scheme 4A), as AB plaque tracers have been thoroughly investigated. In general, it was found that an electron-donating group at each end of the stilbene derivative is essential for AB plaque binding affinity [21]. Analysis of stilbenes 44a-f shows that a monomethylated or dimethylated amine at one end of the stilbene core leads to strong binding affinity for Aβ₁₋₄₀ aggregates, while a free amine or nitro group reduces affinity. The opposite end of the stilbene core can be substituted with a hydroxy or methoxy substituent with little effect on binding affinity (Table 2). Derivative 44d, which showed good affinity towards A β aggregates in vitro ($K_i = 6.0 \pm 1.5 \text{ nM}$), has been radiolabeled to give N-[11C]methylamino-4'-hydroxystilbene ([¹¹C]**44d**), and this compound shows excellent labeling of Aβ plaques in TgCRDN8 mouse brain sections by in vitro autoradiography [22].

Compound	R ¹	R ²	$A\beta_{1-40} K_i$ (nM)	%ID/g at 2 min	%ID/g at 60 min
44a	NO ₂	OMe	151 ± 30	_	_
44b	NH ₂	OMe	36 ± 5	_	_
44c	NHMe	OMe	1.2 ± 0.5	_	_
44d	NHMe	ОН	6.0 ± 1.5	_	_
[¹¹ C] 44d	NH ¹¹ CH ₃	ОН	_	1.15 ± 0.08	0.30 ± 0.03
44e	NMe ₂	OMe	1.3 ± 0.4	_	_
44f	NMe ₂	ОН	2.2 ± 0.6	_	_

[¹⁸F]-Labeled stilbene derivatives have enhanced brain kinetics rendering them appropriate for clinical use [23-25]. In order to control the lipophilicity and keep the partition coefficient (log P) value between 1 and 3, which reduces brain nonspecific binding and improves signal-to-noise ratio, additional hydroxy or ethylene oxide unit(s) were added [21]. An early fluorinated stilbene was [18F]FMAPO (45), which demonstrated high binding affinity for AB aggregates ($K_i = 5.0 \pm 1.2 \text{ nM}$) in assays using human AD brain homogenates [26]. Even though addition of the fluoroalkyl side chain moiety had little effect on the binding affinity and the clearance rate, it improved brain kinetics significantly (from 1.15% ID/g at 2 min for [11C]44d [22] to 9.75% ID/g at 2 min for 45 [26]). Florbetaben ([¹⁸F]BAY94-9172, **46a**), another member of the stilbene class, showed strong binding affinity for human AD brain homogenates ($K_i = 6.7 \pm 0.3$ nM) and promising pharmacokinetics [21], and this compound has progressed to clinical trials. Compound 46a was tested clinically on 15 AD patients and a similar number of healthy elderly volunteers [4]. Interestingly, all AD patients showed widespread neocortical binding of 46a, which was quantified by using the standardized uptake value ratio (SUVR) technique [4]. This observation was further supported by another study using a wider sample population where AD patients demonstrated significantly higher SUVRs when compared to healthy patients or patients with other neural diseases such as Parkinson's disease, mild cognitive impairment, frontotemporal lobar degeneration, dementia with Lewy bodies, and vascular dementia [27]. More recent phase 2/3 clinical trials collectively showed that compound 46a displays a

high degree of sensitivity and selectivity in discriminating between patients with probable AD and age-matched healthy controls [28].

A pyridine analogue of **46a**, florbetapir ([¹⁸F]AV-45, **46b**) was also prepared using a tosylate precursor with Sumitomo modules for radiosynthesis [23,29]. Compound **46b** displayed strong affinity for A β peptides in AD brain homogenates (K_i = 2.87 ± 0.17 nM), excellent pharmacokinetics [30], and an acceptable safety profile that paved the way to its clinical application in brain imaging [31]. A number of **46b**/PET studies have been conducted [23,32-39]. Using **46b** as an imaging probe, PET indicated that the drug accumulates explicitly in A β -deposition-rich cortical regions in AD patients with minimal accumulation observed in healthy volunteers [40].

In general, the stilbene nucleus was built using the Wadsworth-Emmons reaction, and a representative synthesis of stilbene 46a is shown (Scheme 4B). Initial Wadsworth-Emmons reaction between diethyl (4-nitrobenzyl)phosphonate (47) and 4-methoxybenzaldehyde (48) constructed the stilbene core 49. The target compound 46 was formed from a straightforward sequence of nitro reduction, reductive methylation, methyl ether cleavage, pegylation and radiolabeling. Several synthetic procedures have been described for the preparation of 46a and its precursors in an effort to optimize yield [21,41,42]. The best reported yield and purity was obtained by mixing the mesylate precursor with the fluorinating agent in a modified PET-MF-2V-IT-1 synthesizer and by

purifying using plus C18 Sep-Pak cartridges [41]. In the preparation of [¹¹C]**44d**, the [¹¹C]-methylation of 4-amino-4'-hydroxystilbene was carried out using the "LOOP" method, in which trapping and reaction of [¹¹C]CH₃OTf with the appropriate stilbene analogue takes place inside an HPLC sample loop [43].

Diphenyl-1,2,4- and diphenyl-1,3,4-oxadiazoles

The replacement of the stilbene ethylene linker with different heterocycles is a common strategy in medicinal chemistry used to improve the pharmacokinetics and/or pharmacodynamics of stilbenes (Scheme 5A) [44-46]. In the case of $A\beta$ probes, a series of 2,5-diphenyl-1,3,4-oxadiazoles **50a-f** and 3,5-diphenyl-1,2,4-oxadiazoles **51a-e** have been studied in this respect (Table 3 and Table 4). Among the 2,5-diphenyl-1,3,4-oxadiazoles, the dimethylamine analogue **50a** (K_i = 20.1 \pm 2.5 nM) and methoxy analogue **50b** (K_i = 46.1 \pm 12.6 nM)

and [125|]51c, respectively.

showed the best affinities towards AB aggregates, and radiolabeling has been performed for both of these compounds. In biodistribution studies, the dimethylamine analogue [125I]**50a** showed good brain uptake and washout rates. Although methoxy analogue [125I]50b showed poorer brain uptake, its washout rate was increased compared to its dimethylamine counterpart [47]. Interestingly, changing the heteroatom order in the central ring from 1,3,4 (50a-f) to 1,2,4 (51a-e) has great effects on both the physical characteristics and pharmacokinetics of the compounds. The 3,5-diphenyl-1,2,4-oxadiazole analogue 51c was more lipophilic than its 1,3,4 counterpart 50a $(\log P = 3.22 \text{ for } 51c \text{ and } 2.43 \text{ for } 50a)$ [47]. In general, even though 3,5-diphenyl-1,2,4-oxadiazoles 51a-e show excellent affinity for Aβ aggregates in in vitro binding experiments $(K_i = 4.3-47.1 \text{ nM})$, they show poorer brain uptake rates (1.07-2.06% ID/g at 2 min) and slower washout rates (3.29–2.01% ID/g at 60 min) than their 1,3,4 counterparts [48].

_	_			
Compound	R	$A\beta_{1-42} K_i$ (nM)	%ID/g at 10 min	%ID/g at 60 min
50a	NMe ₂	20.1 ± 2.5	_	<u> </u>
[¹²⁵ l] 50a	NMe ₂	_	5.93 ± 0.76	1.78 ± 0.41
50b	OMe	46.1 ± 12.6	_	_
[¹²⁵ l] 50b	OMe	_	2.74 ± 0.37	0.36 ± 0.13
50c	OH	229.6 ± 47.3	_	_
50d	OCH ₂ CH ₂ OH	282.2 ± 61.4	_	_
50e	(OCH ₂ CH ₂) ₂ OH	348.6 ± 51.7	_	_
50f	(OCH ₂ CH ₂) ₃ OH	257.7 ± 34.8	_	_

Compound	R	$A\beta_{1-42} K_i$ (nM)	%ID/g at 2 min	%ID/g at 60 min
51a	NH ₂	14.2 ± 1.4	_	_
[¹²⁵ l] 51a	NH_2	_	1.61 ± 0.23	3.29 ± 0.58
51b	NHMe	14.3 ± 3.6	_	_
[¹²⁵ l] 51b	NHMe	_	1.44 ± 0.12	2.70 ± 0.33
51c	NMe ₂	15.4 ± 1.4	_	_
[¹²⁵ l] 51 c	NMe ₂	_	1.07 ± 0.23	2.32 ± 0.64
51d	OMe	4.3 ± 2.1	_	_
[¹²⁵ l] 51d	OMe	_	2.06 ± 0.45	2.01 ± 0.33
51e	ОН	47.1 ± 4.1	_	_

These findings, together with the close structural similarities between compounds **50a–f** and **51a–e**, highlight the importance of lipophilicity as a factor in controlling brain kinetics [47].

Representative syntheses of radioiodinated oxadiazoles **50a** and **51c** are shown (Scheme 5B and C). The 1,3,4-oxadiazole core of [¹²⁵I]**50a** was obtained from the reaction between 4-iodobenzhydrazide (**52**) and 4-dimethylaminobenzaldehyde (**23**) in the presence of ceric ammonium nitrate (CAN) followed by subsequent radioiodination of compound **53** (Scheme 5B) [47]. The 1,2,4-oxadiazole core of [¹²⁵I]**51c** was obtained by DCC/HOBt-mediated condensation of 4-bromobenzamidoxime (**54**)

and *p*-nitrobenzoic acid (55). Subsequent nitro reduction, reductive methylation, and radioiodination gave [¹²⁵I]51c (Scheme 5C) [48].

Thioflavin-T analogues Benzothiazoles

Of all the amyloid imaging classes, the benzothiazoles, such as 56a-w, 57, 58a,b, and 59-65 (Figure 2), may well be one of the most prolific and well-studied. The amyloid imaging dye thioflavin-T (4, Figure 1) served as the inspiration for this class of radiotracers in which the ionic charge was removed to increase lipophilicity and to enhance in vivo BBB

permeability. Overall, this class of compounds shows high affinity for $A\beta$ aggregates with promising in vivo pharmacokinetics.

One of the earliest radiolabeled benzothiazoles, $[^{11}C]6$ -Me-BTA-1 ($[^{11}C]$ **56a**; note: BTA = 2-(4'-methylaminophenyl)benzothiazole, Figure 2), was prepared by methylation of 4-(6-methyl-2-benzothiazolyl)aniline using [11C]methyl iodide [49]. Compared to 4, 56a showed greatly increased lipophilicity and improved binding affinity for Aβ₁₋₄₀ $(K_i = 890 \text{ nM for } 4 \text{ and } K_i = 20.2 \text{ for } 56a)$. In postmortem AD brain sections, [11C]56a was able to stain both Aβ plaques and neurofibrillary tangles (NFTs), while pharmacokinetic studies in normal mice showed high brain uptake (7.61% ID/g at 2 min) and good washout (2.76% ID/g at 30 min). Additional modification of this scaffold by removal of the 6-Me group gave [11C]BTA-1 ([11C]**56b**) [50]. Compound **56b** was prepared by coupling of p-nitrobenzoyl chloride (35) and 2-aminothiophenol (66) followed by nitro reduction to 67 and methylation using [11C]methyl iodide (Scheme 6A). While showing a near equal binding affinity for AB, the decreased lipophilicity of [11C]**56b** to the ideal level led to improved pharmacokinetics over [11C]56a as evidenced by improved uptake and washout rates in normal mice (12.9% ID/g at 2 min and 1.7% ID/g at 30 min). Compound [11C]**56b** showed in vivo specificity for Aβ in the brains of PS1/APP transgenic mice, and it was subsequently shown to bind specifically to amyloid deposits in human AD brain homogenates [51].

The addition of a hydroxy group at the 6-position of [11 C]**56b** gave [11 C]6-OH-BTA-1 ([11 C]**56c**) [52]. Compound [11 C]**56c** was synthesized by first coupling p-anisidine (**68**) with p-nitrobenzoyl chloride (**35**) to give the amide **69**, which was subsequently converted to the thioamide by using Lawesson's reagent and cyclized to form the benzothiazole core **70** (Scheme 6B). Demethylation with BBr₃ and protection of the resulting hydroxy moiety as the methoxymethyl (MOM) ether gave **71**. Reduction of the nitro group to **72**, methylation using [11 C]methyl iodide, and cleavage of the MOM ether gave [11 C]**56c**. Compound **56c** showed high affinity for A β_{1-40} ($K_i = 4.3$ nM) (Table 5). This synthesis has since been refined to improve radiochemical yields and eliminate the need for a protecting group by use of [11 C]CH₃OTf as the methylating agent [53].

The 6-OH group of **56c** made it less lipophilic than both **56a** and **56b** and likely contributed to its moderate brain entry (0.21% ID-kg/g at 2 min) but good clearance (0.018% ID-kg/g at 30 min) in normal mice. Interesting SAR findings on this scaffold from comparison of **56c–t** (Table 5) included that the more lipophilic secondary and tertiary amines at the 4'-position

were more potent (K_i) than primary amines. Also, in general, substitution at the 6-position seemed to have only a small effect in terms of K_i as 6-OH, -OCH₃, -CN, and -Br gave similar results. However, substitution at the 6-position had a larger effect on pharmacokinetics in the brain, as 6-OH clearly gave the best results [52]. As one of the most successful radiolabeled A β imaging probes to date, [11 C]56c has subsequently been named Pittsburgh Compound B (PIB).

Additional studies of [11C]56c in humans have been promising and suggest that PET imaging with this compound can provide quantitative information on amyloid deposits in living patients. In postmortem tissue, [11C]56c exhibited specific binding to the amyloid-laden frontal cortex of the AD brain, but little binding to the frontal cortex of the cognitively normal age-matched control brain. Compound [11C]56c also displayed a rapid entry and clearance in the brain of healthy controls, but a marked retention in AD patients in areas of the brain known to contain large amyloid deposits [54]. Additional data suggested that [11C]56c was suitable for early detection of pathological changes in AD patients before a significant loss of cognitive function is apparent [55].

The impact of changing the position of the hydroxy group of 56c was investigated by synthesizing the 4-OH, 5-OH, and 7-OH analogues 56u-w using methods similar to those described above [56]. The K_i values for these analogues in human AD brain homogenates were between 11-19 nM, indicating slightly reduced affinity compared to 56c ($K_i = 2.8 \text{ nM}$) (Table 6). However, each radiolabled analogue was able to stain plaques in sections from transgenic AD mouse brain and human AD brain. The 5-OH analogue [11C]56v showed the best pharmacokinetic profile in normal mice with high brain uptake and a washout rate, that was 8 times faster than that of [11C]56c. Interestingly, it was noted that the 4-OH analogue 56u could form an intramolecular hydrogen bond (i.e. an extra pseudo ring), which could act to increase the lipophilicity of the compound and lead to nonspecific binding and residual background activity in the brain.

A [3 H]-labeled analogue of **56c**, AZD2184 (**57**), was also synthesized to give a higher signal-to-background ratio by virtue of its decreased lipophilicity [57]. This compound was prepared through palladium catalyzed Suzuki coupling of the starting halide **73** and boronic acid **74** followed by *N*-methylation of **75** with [3 H]methyl iodide and *O*-demethylation with sodium thiophenoxide (Scheme 6C). Compound **57** showed high affinity for A β_{1-40} fibrils in vitro ($K_d = 8.4$ nM) and lower background binding levels than **56c**. While **57** was able to label amyloid deposits in APP/PS1 mice, its brain penetration was not as high as that of [11 C]**56c**.

Besides [\$^{11}C\$], other radiolabels have been investigated for benzothiazole imaging agents. Two of the earliest [\$^{125}I\$]-labeled imaging agents reported were [\$^{125}I\$]TZDM (**58a**) and [\$^{125}I\$]TZPI (**58b**) [58]. The synthesis of these agents was achieved in two steps by condensation of 5-bromo-2-aminobenzenethiol (**76**) and the appropriate benzaldehyde **77** followed by

radiolabeling of **78** (Scheme 6D). Both **58a** and **58b** showed high affinity for $A\beta_{1-40}$ and $A\beta_{1-42}$ aggregates with K_d values ≤ 0.15 nM in all cases. However, pharmacokinetics for these agents were less than ideal as both showed long retention in the brains of normal mice, which is indicative of nonspecific binding.

Compound	R ¹	R ²	$A\beta_{1-40}$ K_i (nM)	(%ID-kg)/g at 2 min	(%ID-kg)/g at 30 min
56c	6-OH	NHMe	4.3	_	_
[¹¹ C] 56c	6-OH	NH ¹¹ CH ₃	_	0.21	0.018
56d	6-OH	NH_2	46	_	_
56e	6-OH	NMe ₂	4.4	_	_
[¹¹ C] 56e	6-OH	NMe ¹¹ CH ₃	_	0.32	0.10
56f	6-H	NHMe	11	_	_
[¹¹ C] 56f	6-H	NH ¹¹ CH ₃	_	0.43	0.057
56g	6-H	NH ₂	37	_	_
56h	6-H	NMe ₂	4.0	_	_
[¹¹ C] 56h	6-H	NMe ¹¹ CH ₃	_	0.19	0.078
56i	6-Me	NHMe	10	_	_
[¹¹ C] 56i	6-Me	NH ¹¹ CH ₃	_	0.22	0.083
56j	6-Me	NH_2	9.5	_	_
56k	6-Me	NMe ₂	64	_	_
[¹¹ C] 56k	6-Me	NMe ¹¹ CH ₃	_	0.078	0.15
561	6-OMe	NHMe	4.9	_	_
[¹¹ C] 56I	6-OMe	NH ¹¹ CH ₃	_	0.33	0.10
56m	6-OMe	NH ₂	7.0	_	_
[¹¹ C] 56m	6-O ¹¹ CH ₃	NH ₂	_	0.32	0.084
56n	6-OMe	NMe ₂	1.9	_	_
[¹¹ C] 56n	6-OMe	NMe ¹¹ CH ₃	_	0.16	0.14
56o	6-CN	NHMe	8.6	_	_
[¹¹ C] 56o	6-CN	NH ¹¹ CH ₃	_	0.32	0.063
56p	6-CN	NH ₂	64	_	_
56q	6-CN	NMe ₂	11	_	_
[¹¹ C] 56q	6-CN	NMe ¹¹ CH ₃	_	0.24	0.097
56r	6-Br	NHMe	1.7	_	_
[¹¹ C] 56r	6-Br	NH ¹¹ CH ₃	_	0.12	0.12
56s	6-Br	NH ₂	7.2	_	_
56t	6-Br	NMe ₂	2.9	_	_
[¹¹ C] 56t	6-Br	NMe ¹¹ CH ₃	_	0.054	0.11

Compound	R ¹	R ²	human AD brain homogenates K_i (nM)	%ID/g at 2 min	%ID/g at 60 min
56c	6-OH	NHMe	2.8 ± 0.5	_	_
[¹¹ C] 56c	6-OH	NH ¹¹ CH ₃	_	3.6 ± 1.4	0.6 ± 0.2
56u	4-OH	NHMe	18.8 ± 3.8	_	_
[¹¹ C] 56u	4-OH	NH ¹¹ CH ₃	_	3.8 ± 0.9	0.3 ± 0.3
56v	5-OH	NHMe	11.5 ± 3	_	_
[¹¹ C] 56v	5-OH	NH ¹¹ CH ₃	_	4.3 ± 0.45	0.09 ± 0.02
56w	7-OH	NHMe	11.2 ± 5	_	_
[¹¹ C] 56w	7-OH	NH ¹¹ CH ₃	_	2.6 ± 0.76	0.16 ± 0.03

A series of iodinated benzothiazoles **59a-0** was synthesized using methods similar to those described above and SAR studies were performed (Table 7) [59]. Among the interesting

findings was that the introduction of 3'-iodo increased lipophilicity and binding to $A\beta_{1-40}$ when $R^2=NHMe$. However, the opposite effect on binding was observed when

 $R^2 = OH$. Among the [^{125}I]-labeled derivatives, more polar compounds exhibited better clearance and less nonspecific binding in the brains of normal mice, a typical result for brain imaging probes. One of the most promising compounds identified in this study was [^{125}I]**59d**.

The [125 I]-labeled benzothiazole bithiophene **60** was synthesized by condensation of 5-bromo-2-aminobenzenethiol and 2,2'-bithiophene-5-carbaldehyde followed by installation of the radiolabel. In in vitro binding experiments, **60** displayed high affinity for both A β_{1-40} and A β_{1-42} aggregates with K_i values of 0.25 nM and 0.31 nM, respectively. In addition, it was used to clearly visualize A β plaques in AD brain sections and showed favorable pharmacokinetics in the brain with high uptake (3.42% ID/g at 2 min) and fast washout (0.53% ID/g at 60 min).

The [125 I]-labeled phenyldiazenyl benzothiazole **61** was prepared via a diazo coupling reaction between **79** and **80** to give **81** followed by installation of the radiolabel (Scheme 6E) [60]. Interestingly, in in vitro binding experiments, **61** displayed higher affinity for tau aggregates ($K_i = 0.48$ nM) than for Aβ aggregates ($K_i = 8.24$ nM). Although it was used to clearly visualize NFTs in AD brain sections, further modifications will be necessary to improve the pharmacokinetics of this compound in the brain, as it showed particularly slow washout rate (2.89% ID/g at 60 min).

Three [¹⁸F]-labeled analogues of **56c**, [¹⁸F]O-FEt-PIB (**62**), [¹⁸F]FBTA (**63**), and [¹⁸F]3'-F-PIB ([¹⁸F]GE067, **64**) were also

prepared. Compound 62 was synthesized by using the hydroxy group of **56c** to displace the tosylate of [18F]fluoroethyltosylate. Compound 62 had a K_i value of 0.17 nM for AD brain homogenate and was able to stain Aß plaques in postmortem AD brain [61]. Although its biodistribution was not as good as that of 56c, 62 still showed promise in an in vivo study using a rat model of AD [62]. Moving the [18F]fluoroethoxy substituent of 62 from the 6-position to the 3'-position resulted in a low binding affinity for AB and an inability to stain plaques in postmortem AD brain [63]. In compound 63, the [11C]methylamino group of [11C]56c was replaced by a [18F]fluoroethylamino group, and, while this compound showed better binding affinity than 56c, its brain pharmacokinetics were not as good [64]. Compound 64 also showed promising results in whole-body biodistribution and radiation dosimetry studies [65].

A series of fluorinated benzothiazoles **65a–e** was synthesized by direct substitution of the nitro group of a key synthetic intermediate **82** (prepared using synthetic steps already describe for $[^{11}C]$ **56c**) by an $[^{18}F]$ atom (Scheme 6F) [66,67]. Compounds **65a,b,d** (R = H, Me, and OMe) all showed high binding affinity for AD brain homogenates with K_i values below 10 nM, which is comparable to that of **56c** in the same assay, while **65c** (R = OH) showed slightly reduced affinity (Table 8). In addition to showing a promising ability to stain A β plaques in vivo, $[^{18}F]$ **65a,b** showed high brain uptake and rapid washout in normal mice. In fact, each of these compounds displayed better pharmacokinetics than $[^{11}C]$ **56c** in the same assay.

Compound	R ¹	R^2	$A\beta_{1-40}$ K_i (nM)	%ID/g at 2 min	%ID/g at 30 min
59a	Н	NH ₂	8.32	_	_
[¹²⁵ l] 59a	Н	NH_2	_	9.08	3.4
59b	Н	NHMe	4.94	_	_
[¹²⁵ l] 59b	Н	NHMe	_	4.40	2.68
59c	Н	ОН	19.1	_	_
59d	ОН	NH_2	11.1	_	_
[¹²⁵ l] 59d	ОН	NH_2	_	5.64	0.36
59e	ОН	NHMe	3.22	_	_
[¹²⁵ l] 59e	ОН	NHMe	_	7.76	2.66
59f	ОН	ОН	71.2	_	_
59g	OMe	NH_2	4.4	_	_
59h	OMe	NHMe	1.93	_	_
59i	OMe	ОН	15.8	_	_
59j	NO ₂	NH_2	4.6	_	_
59k	NO_2	NHMe	1	_	_
591	Br	NH_2	0.67	_	_
59m	Br	NHMe	1.6	_	_
59n	OCH ₂ OCH ₃	NH_2	15.1	_	_
59o	CO ₂ Me	NH ₂	3.34		_

Compound	R	human AD brain homogenates K_i (nM)	%ID/g at 2 min	%ID/g at 30 min
·				
65a	Н	9.0 ± 2.0	_	_
[¹⁸ F] 65a	Н	_	3.20 ± 0.38	0.21 ± 0.03
65b	Me	5.7 ± 1.8	_	_
[¹⁸ F] 65b	Me	_	5.33 ± 0.74	0.27 ± 0.06
65c	ОН	22.5 ± 4.5	_	_
[¹⁸ F] 65 c	ОН	_	4.70 ± 0.48	0.57 ± 0.36
65d	OMe	2.2 ± 0.5	_	_
[¹⁸ F] 65d	OMe	_	5.10 ± 0.40	0.43 ± 0.12
65e	CO ₂ H	>4000	_	_

Benzothiazole probes such as 83a,b and 84a,b labeled with [Re] and [99mTc] were also synthesized (Scheme 7A) [68,69]. [Re] and $[^{99m}$ Tc]MAMA-BTA (83a and 83b; MAMA = monoamine-monoamide bisthiol-BTA) were prepared by first linking 2-(4-nitrophenyl)-6-hydroxybenzothiazole (85) via 1,5-dibromopentane (86) to monoamine-monoamide bisthiol protected with p-methoxy benzyl (MAMA-PMB, 87) to give 88 (Scheme 7B) [68]. Nitro reduction of 88 followed by thioether deprotection gave MAMA-BTA (89), which was labeled through reaction with the [Re] (used for in vitro studies) or [99mTc] precursors to give the desired 83a,b. [Re] and $[^{99}$ mTc]BAT-BTA (**84a** and **84b**; note: BAT = bis(aminoethanethiol)) were prepared by addition of ethyl bromoacetate (41) to the unprotected amine of the S,S'-bis-trityl-N-Boc-1,2ethylenedicysteamine chelating agent (90) followed by saponification that gave the free acid intermediate 91, which was coupled with 2-(4-aminophenyl)-1,3-benzothiazole (92) (prepared from 2-aminothiophenol (66) and 4-aminobenzoic acid (93)) by using EDCI·HCl and HOBt (Scheme 7C) [69]. Deprotection followed by reaction with the [Re] or [99mTc] precursors gave **84a** and **84b**. While both **83a** and **84a** showed promise as in vitro A β labeling agents, the [99m Tc] analogues **83b** and **84b** exhibited problems in pharmacokinetic studies in vivo. Compound **83b** showed sufficient initial uptake (1.34% ID/g at 2 min), but delayed washout (0.65% ID/g at 60 min) in normal mice, while **84b** was unable to cross the BBB to a sufficient degree.

Benzoxazoles

Replacement of the sulfur of the benzothiazole backbone by oxygen affords the benzoxazole backbone. Compounds 94, 95a-n, and 96-99 (Figure 3) have also been successfully employed for radioimaging of Aβ plaques. The isosteric replacement of the sulfur of [125I]TZDM (58a) with an oxygen was designed to decrease molecular weight and increase lipophilicity and afforded [125I]IBOX (94) [70]. Compound 94 was prepared via boric acid catalyzed condensation of 5-nitro-2-aminophenol (100) and 4-dimethylaminobenzoic acid (101) to give the nitro intermediate 102, which was reduced through catalytic hydrogenation to the amine (Scheme 8A).

Subsequent conversion to the diazonium ion and displacement with iodide ion gave IBOX, which was radiolabeled to give 94. Compound 94 showed similar affinity for $A\beta_{1-40}$ aggregates when compared to 58a, and it was able

to label A β plaques in postmortem AD brain sections. Importantly, **94** showed superior peak brain uptake (2.08% ID/g at 30 min) and faster brain washout than **58a** in normal mice.

Expanding on this 2-arylbenzoxazole scaffold, a series of benzamide-substituted 2-arylbenzoxazoles **95a-n** was synthesized [71]. A representative synthesis of [¹²³I]**95e** is shown

(Scheme 8B). Boric acid catalyzed condensation of 4-nitro-2-aminophenol (103) and 4-dimethylaminobenzoic acid (101) gave the nitro intermediate 104. Catalytic hydrogenation as

above gave the amine intermediate, and subsequent reaction with 4-iodobenzoyl chloride (105) and installation of the radiolabel gave the target compound. SAR analysis of the compounds indicates that the benzamide moiety is favored at position 5 rather than 6 of the benzoxazole core in terms of binding affinity for A β plaques in vitro (Table 9). The best compound was 95e, which had a K_i value of 9.3 nM, but [^{123}I]95e was unable to cross the BBB in vivo. This disappointing result could be appointed to the excessively high lipophilicity of the compound.

Table 9: Inhibition constants of benzamide-substituted 2-arylbenzoxazole derivatives **95a–n** (values are from [71]).

Compound	R	K _i (nM)
95a	5-phenyl	12.0
95b	6-phenyl	26.0
95c	5-(3,4,5-trimethoxyphenyl)	109
95d	6-(3,4,5-trimethoxyphenyl)	628
95e	5-(4-iodophenyl)	9.3
95f	6-(4-iodophenyl)	60.1
95g	5-(p-tolyl)	13.2
95h	6-(p-tolyl)	86.0
95i	5-(<i>m</i> -tolyl)	13.4
95j	6-(<i>m</i> -tolyl)	31.5
95k	5-(<i>o</i> -tolyl)	18.9
951	6-(<i>o</i> -tolyl)	112
95m	5-(3,4-(methylenedioxy)phenyl)	17.2
95n	6-(3,4-(methylenedioxy)phenyl)	19.7

To improve the pharmacokinetic profile of 94, the [18F]-labeled analogue 96 was designed as an imaging probe [72]. Compound 96, which contains an [18F] end-capped polyethylene glycol chain at position 5 of the benzoxazole core in place of the [125I] of 94 at position 6 to reduce lipophilicity, was prepared by polyphosphoric acid catalyzed condensation of 2-amino-4-methoxyphenol (106) and 4-monomethylaminobenzoic acid (107) to give the benzoxazole core, which was O-demethylated to give 108 (Scheme 8C). Subsequent coupling with 2-[2-(2-chloroethoxy)ethoxy]ethanol (109) gave 110. TBDMS protection of the alcohol and Boc protection of the amine gave 111. Finally, TBAF cleavage, installation of the radiolabel, and acid cleavage gave 96. Compound 96 showed good affinity for $A\beta_{1-42}$ ($K_i = 9.3$ nM). This compound also showed promising pharmacokinetics in normal mice, with greatly improved uptake and washout rates compared to 94, and it successfully labeled AB plaques in vitro. In addition, it showed increased retention in vivo in transgenic AD mice compared to wild-type. A N,N-dimethyl derivative

was also synthesized, and, while it too showed good affinity for $A\beta_{1-42}$, its increased lipophilicity compared to the monomethyl compound gave slightly worse pharmacokinetic properties.

The [11C]-labeled styrylbenzoxazole [11C]BF-145 (97) and the related [18F]-labeled styrylbenzoxazole [18F]BF-168 (98) were prepared and studied for Aβ imaging [73-75]. The simple twostep synthesis of 97 used polyphosphoric acid trimethylsilyl ester (PPSE) catalyzed condensation of 4-fluoro-2-aminophenol (112) with a cinnamic acid 113 to give the benzoxazole core followed by conversion to the primary amine 114 and radiolabeling (Scheme 8D). The synthesis of 98 was more complex and began with a reaction between 2-methyl-6-methoxybenzoxazole (115) and 4-((N-Boc-N-methyl)amino)benzaldehyde (116) followed by a dehydration reaction to give 117 (Scheme 8E). Subsequent removal of the Boc group followed by installation of a trifluoroacetamide and O-demethylation gave the intermediate 118 used in a Mitsunobu reaction with 2-hydroxyethyl tosylate (119). Amine deprotection to 120 and installation of the [18F] label gave the target compound 98.

Both 97 and 98 showed good affinity for $A\beta_{1-42}$ aggregates ($K_i = 4.5$ nM and 6.4 nM, respectively). Interestingly, while 98 was able to selectively stain senile plaques (SPs) and NFTs in AD brain sections, 97 was only able to stain SPs. In addition, 97 and 98 showed substantial brain uptake and fast washout (4.4% and 3.9% ID/g at 2 min and 1.6% and 1.6% ID/g at 30 min, respectively) with promising in vivo imaging results in transgenic mice.

Building on the promising results of 97 and 98, an optimized derivative, [11C]BF-227 (99), was studied for Aβ imaging. The key difference in 99 is the replacement of a phenyl ring with a thiazole ring. Compound 99 demonstrated good affinity for synthetic $A\beta_{1-42}$ aggregates ($K_i = 4.3 \text{ nM}$), rapid uptake (7.9% ID/g at 2 min) and clearance (0.64% ID/g at 60 min) in normal mice, the ability to selectively stain $A\beta$ plaques in AD brain sections, and promising results in a clinical PET study in AD patients [76]. Additional studies suggest that 99 has the possibility to be useful for early detection of AD and also for predicting progression from mild cognitive impairment to AD [77,78]. Interestingly, 99 has also shown promise for diseases other than AD. It has been suggested that 99 may provide a means of diagnosis and disease monitoring in transmissible spongiform encephalopathies [79] and may be useful for monitoring α-synuclein deposits in conditions such as multiple system atrophy and Parkinson's disease [80]. A version of 99 labeled with [18F] rather than [11C] has also been proposed for use in Parkinson's disease [81].

Benzofurans

Replacement of the nitrogen of the benzoxazole backbone with carbon affords the benzofuran backbone of compounds 121-126 (Figure 4), which has also been successfully employed for radioimaging of A β plaques. The [\$^{11}\$C]-labeled benzofuran 121 was prepared via Wittig reaction between the triphenylphosphonium salt of 2-hydroxy-5-methoxybenzyl alcohol (127) and 4-nitrobenzoyl chloride (35) to give 128 followed by nitro reduction and O-demethylation to give 129 and radiolabeling (Scheme 9A). Using AD brain gray matter homogenates, compound 121 showed good binding affinity for A β plaques ($K_i = 0.7$ nM) and was able to stain both SPs and NFTs in vitro. In normal mice, this compound showed rapid uptake (4.8% ID/g at 2 min) and fast washout (0.2% ID/g at 60 min). In vivo plaque labeling in APP transgenic mice was also successful [82].

Using similar chemistry as described above, a series of iodinated 2-arylbenzofurans **122a**–e was prepared and studied [83].

A representative synthesis of [125 I]122a, which uses a similar Wittig as in the synthesis of 121, is shown in Scheme 9A. It was found that the iodo substituent could be varied between the 5-and 6-positions, and the N,N-dimethylamino substituent could be changed to a secondary methylamino or hydroxy moiety with little effect on binding affinity for synthetic A β_{1-40} , as all compounds had a $K_i \leq 8$ nM (Table 10). While the [125 I]-labeled benzofurans in this series showed good brain uptake in normal mice, their washout was rather slow indicating nonspecific binding in vivo.

Several [¹⁸F]-labeled benzofurans have been employed with success for Aβ imaging. [¹⁸F]FPYBF-1 (**123a**), which has a *N*,*N*-dimethyl-2-aminopyridine group attached to the benzofuran core, was synthesized via Suzuki coupling between 5-methoxybenzofuran-2-boronic acid (**130**) and 2-amino-5-iodopyridine (**131**) to give **132**, which was followed by reductive amination and *O*-demethylation to give **133** (Scheme 9B). Finally, reaction with 2-[2-(2-chloroethoxy)ethoxy]ethanol

Table 10: Inhibition co	onstants and biodistr	ibution of the rad	ioactivity of iodinated 2-aryll	penzofuran derivatives 122a-	-e (values are from [83]).
Compound	R ¹	R^2	$A\beta_{1-40}$ K_i (nM)	%ID at 2 min	%ID at 60 min
122a	5-iodo	NMe ₂	7.7 ± 1.2	_	_
[¹²⁵ l] 122a	5-iodo	NMe_2	_	0.51 ± 0.05	1.08 ± 0.15
122b	5-iodo	NHMe	1.1 ± 0.2	_	_
[¹²⁵ l] 122b	5-iodo	NHMe	_	0.78 ± 0.06	1.20 ± 0.34
122c	5-iodo	OMe	4.2 ± 0.8	_	_
122d	5-iodo	ОН	6.5 ± 0.2	_	_
[¹²⁵ l] 122d	5-iodo	ОН	_	1.40 ± 0.04	1.51 ± 0.20
122e	6-iodo	NMe ₂	0.4 ± 0.1	_	_
[¹²⁵ l] 122e	6-iodo	NMe ₂	_	0.48 ± 0.07	1.00 ± 0.22

(134) followed by radiolabeling gave 123a. This compound showed high affinity for A β_{1-42} aggregates ($K_i = 0.9 \text{ nM}$), the ability to label plaques in postmortem AD brains, and suitable pharmacokinetic properties in normal mice (5.16% ID/g at 2 min and 2.44% ID/g at 60 min). In addition, it showed good in vivo binding to plaques in transgenic mice [84]. A closely related compound, [18F]FPHBF-1 (123b), which has a N,Ndimethylaniline group in place of the N,N-dimethylaminopyridine group, was prepared using a Wittig reaction between the triphenylphosphonium salt of 2-hydroxy-5-methoxybenzyl alcohol (127) and 4-dimethylaminobenzoyl chloride (35), followed by O-demethylation and installation of the [18F]-labeled linker (Scheme 9C). Like 123a, compound 123b showed good affinity for Aβ aggregates in vitro and in vivo. However, its slow washout from the brain, which can be attributed to its increased lipophilicity compared to 123a, indicated that additional refinements will be needed [85].

Derivatives of 123a and 123b were also prepared [86]. These compounds, [¹⁸F]FPYBF-2 (**123c**) and [¹⁸F]FPHBF-2 (**123d**), have a secondary methylamino group in place of the dimethylamino group. Introduction of the secondary amine served to reduce lipophilicity. In addition, as the secondary amines are less rapidly metabolized than the tertiary amines, they may help improve the stability of these compounds in vivo. The synthesis of these derivatives used methodology similar to that already described for 123a and 123b. One key difference, however, was the need for orthogonal TBS and Boc protection/ deprotection to prevent the secondary amine from reacting with the MsCl used to introduce the radiolabel. Both 123c and 123d showed good affinity for A β_{1-42} aggregates ($K_i = 2.41$ nM and 3.85 nM, respectively) as well as the ability to label plaques in transgenic mice. Also, both 123c and 123d showed high uptake and rapid washout with improved pharmacokinetic properties when compared to 123a and 123b.

The [3 H]-labeled AZD4694 (**124**) also showed promise for A β imaging [87]. With good affinity for β -amyloid fibrils in vitro ($K_d = 2.3 \text{ nM}$), this compound was able to label plaques in human AD brain sections with little nonspecific binding. In addition, the good pharmacokinetic profile of **124** warrants further investigation in vivo.

[Re] and [99m Tc]-labeled benzofurans, BAT-Bp-2 (**125a,b**), were synthesized from **132** by reductive monoamination and *O*-demethylation to give **135** (Scheme 9D) [88]. Subsequent reaction with the protected chelation ligand TRT-Boc-BAT-Br (**136**) and labeling through reaction with the rhenium (used for in vitro studies) and technetium precursors gave compounds **125a** and **125b**, respectively. Compound **125b** showed decent affinity for A β_{1-42} aggregates ($K_i = 32.8$ nM) in vitro, although,

by comparison to other benzofuran probes of similar structure, it was clear that introduction of the BAT chelator decreased binding affinity. In contrast to other [$^{99\text{m}}$ Tc]-labeled A β probes, **125b** showed decent brain uptake and washout rates in normal mice (1.80% ID/g at 2 min and 0.79% ID/g at 60 min). In addition, it was able to label A β plaques in vivo in transgenic mice, a first for a [$^{99\text{m}}$ Tc]-labeled A β probe.

The [125I]-labeled probe 126 contains the benzofuran core, but could also be classified as a chalcone, specifically a chalcone in which the conformation around the double bond is fixed [89]. Compound 126 was synthesized by using a Rap-Stoermer condensation between the bromo-substituted salicylaldehyde (137) and α-brominated 4-nitroacetophenone (138) to form the benzofuran core (Scheme 9E). Nitro reduction followed by methylation gave 139, and radiolabeling gave 126. This compound showed good affinity for A β_{1-42} aggregates ($K_i = 6.6$ nM). Secondary methylamino and primary amino derivatives showed decreased binding affinity and poorer labeling of plaques in brain sections from transgenic mice. While the pharmacokinetics of this compound in normal mice were promising (3.53% ID/g at 2 min and 0.87% ID/g at 60 min), they were not as good as those previously reported for [125I]-labeled N,N-dimethylamino chalcones and aurones.

Imidazopyridines

The imidazopyridine core has also been used in developing novel Aβ imaging agents such as 140-142 (Scheme 10A). Initial SAR studies were based on derivatives 140a-j. One of the most successful imidazopyridines studied to date has been [125] IMPY ([125]] **140e**). Representative of this scaffold, the synthesis of [125I]140e used a fusion reaction between 2-amino-5-iodopyridine (143) and an α -bromoacetophenone 144 to form 140e, which was then radiolabeled (Scheme 10B) [90]. This preparation has since been improved by Kung et al. who, through the use of a reverse-phase C4 minicolumn with stepwise washing and elution, have simplified the purification process by eliminating the need for HPLC purification [91]. While 140e showed good affinity for $A\beta_{1-40}$ aggregates $(K_i = 15.0 \text{ nM})$, SAR analysis demonstrated that, in general, other modifications to the scaffold were not well tolerated and reduced the binding affinity (Table 11). An exception was the replacement of the 6-iodo substituent with a bromine, as 140f showed similar affinity to the parent compound [90].

In addition to its good binding affinity for A β , [125I]**140e** showed other promising properties for use as an imaging probe. For example, it selectively labeled plaques in postmortem AD brain sections and showed plaque labeling with low background activity in a transgenic mouse model. The pharmacokinetics of [125I]**140e** were also promising. It showed high uptake

Compound	R ¹	R^2	R^3	$A\beta_{1-40}$ K_i (nM)	%ID at 2 min	%ID at 60 min
140a	Н	Н	NHMe	>1000	_	_
140b	Н	Н	NMe ₂	>2000	_	_
140c	Me	Н	NHMe	>2000	_	_
140d	Me	Н	NMe ₂	242 ± 20	_	_
140e	I	Н	NMe ₂	15 ± 5	_	_
[¹²⁵ l] 140e	125	Н	NMe ₂	_	2.88 ± 0.25	0.21 ± 0.03
140f	Br	Н	NMe ₂	10.3 ± 1.2	_	_
140g	Me	Н	Br	638 ± 30	_	_
140h	NMe ₂	Н	Br	339 ± 40	_	_
140i	Н	1	NMe ₂	>2000	_	_
140j	ı	1	NMe ₂	>2000	_	_

(2.9% ID/g at 2 min) and fast washout (0.2% ID/g at 60 min) in normal mice. These kinetic properties represented improvements over both **58a** and **94** [92]. Safety, biodistribution, and dosimetry studies of [¹²³I]IMPY, the [¹²³I]-labeled counterpart of **140e**, have indicated it may be a safe radiotracer with appropriate pharmacokinetics for use in AD patients [93].

More in depth SAR studies of the imidazopyridine scaffold have been conducted by synthesizing the series of analogues **141a**—w [94]. The effect of different substituents on binding affinity (K_i) for human A β plaques was examined (Table 12). In general, it was found that the N,N-dimethylamino analogues ($R^1 = R^2 = Me$) had higher binding affinity for human A β

Compound	R ¹	R ²	R^3	R^4	R^5	human AD brain homogenates <i>K</i> (nM)
140e	Me	Me	Н	1	Н	8.9 ± 0.7
141a	Me	Me	Н	Br	Н	5.9 ± 0.4
141b	Me	Me	Н	CI	Н	24.2 ± 5.6
141c	Me	Me	Н	F	Н	13.0 ± 1.6
141d	Me	Me	Н	CN	Н	8.2 ± 1.0
141e	Me	Me	Н	NO_2	Н	7.6 ± 0.7
141f	Me	Me	Н	OMe	Н	38.5 ± 5.0
141g	Me	Me	Н	SEt	Н	8.3 ± 0.5
141h	Me	Me	Н	Br	1	183 ± 61
141i	Me	Me	Н	Br	CN	>180
141j	Me	Me	Н	ОН	Н	177 ± 31
141k	CF ₃ CO	Me	Br	Br	Н	>1000
1411	CF ₃ CO	Н	Me	Br	Н	>1000
141m	MeCO	Me	Me	Br	Н	>1000
141n	Н	Me	Br	Br	Н	7.4 ± 0.6
141o	Н	Н	Me	Br	Н	658 ± 47
141p	Н	Me	Me	Br	Н	>1000
141q	Н	Me	Н	SCH ₂ CONH ₂	Н	1840 ± 497
141r	Н	Me	Н	S(CH ₂) ₂ OH	Н	645 ± 75
141s	Me	Me	Н	SCH ₂ CONH ₂	Н	391 ± 76
141t	Me	Me	Н	SCH ₂ C ₆ H ₄ -p-OMe	Н	8.3 ± 1.8
141u	Me	Me	Н	S(CH ₂) ₂ OH	Н	88 ± 6
141v	Н	Me	Me	SCH ₂ CONH ₂	Н	>1000
141w	Н	Me	Me	S(CH ₂) ₂ OH	Н	>1000

plaques than did the secondary methylamino analogues ($R^1 = H$, $R^2 = Me$). Little tolerance for substitution at both R^3 and R^5 was seen, as the most potent compounds almost always had a hydrogen atom at these positions. One exception was observed with the secondary methylamino analogue when $R^3 = Br$, as **141n** showed high affinity. For R^4 it was seen that polarizable or electron-withdrawing substituents showed higher affinity than strongly electron-donating substituents. In addition, it was observed that bulky, hydrophobic thioether substituents (such as $R^4 = SCH_2C_6H_4$ -p-OMe) were well tolerated at this position. This finding was of particular interest as it provided a possible means of generating new PET ligands via $[^{11}C]$ - or $[^{18}F]$ -labeling through S-alkylation.

The [18 F]-labeled imidazopyridine, [18 F]FPPIP (142), was prepared starting from 140 e. A palladium-catalyzed coupling with tributyl(vinyl)tin to give an alkene intermediate was followed by hydroboration-oxidation to give the hydroxypropyl intermediate 145 , which was radiolabeled to give 142 (Scheme 10 B). This compound showed good binding affinity for A β ($K_i = 48.3$ nM) in using human AD cortical tissues, as well as specific labeling of A β plaques in postmortem AD

brain. This, coupled with favorable pharmacokinetics observed in a normal rhesus monkey, made **142** a promising compound [95]. However, another [¹⁸F]-labeled imidazopyridine, [¹⁸F]FPM-IMPY, has shown less promising results. This compound, in which one of the *N*-methyl groups of IMPY was replaced with a [¹⁸F]fluoropropyl moiety, showed lower binding affinity than IMPY and poor pharmacokinetics [96].

Benzimidazoles

The benzimidazole scaffold is highly similar to the imidazopyridine scaffold, but the benzimidazole ring has reduced lipophilicity when compared to imidazopyridines. This has the potential to reduce nonspecific binding and enhance signal-to-noise ratio. The [125 I]-labeled benzimidazole analogue of **140e**, compound **146**, was prepared through cyclization of 4-bromobenzene-1,2-diamine (**147**) and *p*-dimethylaminobenzaldehyde (**23**) to give **148** followed by installation of the radio-label (Scheme 11) [97]. Compound **146** showed good binding affinity for A β_{1-42} aggregates ($K_i = 9.8$ nM), as well as high uptake and rapid clearance in normal mice (4.14% ID/g at 2 min and 0.15% ID/g at 60 min). In vitro labeling of A β plaques in AD brain sections showed a strong

signal with low background, and in vivo plaque labeling in transgenic mice was also successful. However, this scaffold is lacking in detailed SAR analysis compared to the imidazopyridine scaffold.

Quinoline and naphthalene analogues Quinolines

Investigation of the quinoline scaffold for imaging in AD has yielded some interesting results, despite there only being a few examples in the literature. The [18F]-labeled 2-fluoroquinolin-8ol [18F]CABS13 (149) has recently been reported (Figure 5) [98]. The straightforward synthesis of this compound began with benzyl protection of 2-chloroquinolin-8-ol followed by installation of the [18F]-label and Pd-catalyzed hydrogenolysis to give the target compound. Compound 149 potently bound to Aβ-Zn aggregates ($K_d = 1.5 \text{ nM}$) and showed rapid uptake and washout in normal mice (10% ID/g at 2 min and 1.1% ID/g at 30 min). Also, delayed washout of 149 was observed in APP/ Ps1 transgenic mice, which was indicative of non-specific binding to Aß plaques. However, two other quinoline probes, [11C]BF-158 (150) and [18F]THK523 (151), had high affinity for tau pathology as opposed to Aβ. Compound 150 showed good uptake and washout in normal mice (11.3% ID/g at 2 min and 2.1% ID/g at 60 min), and was able to label NFTs in postmortem AD brain section while only faintly staining plaques [99]. Compound 151 showed high affinity ($K_d = 1.7 \text{ nM}$) and selectivity for recombinant tau fibrils in vitro, and, with favorable pharmacokinetics, it was able to highlight tau pathology in vivo in transgenic mice [100].

Naphthalenes

Replacement of the cyclic nitrogen in the quinoline scaffold described in the previous section affords the naphthalene scaffold. This scaffold has shown promising results for Aß imaging, particularly [18F]FDDNP (152), although this scaffold is also represented by only a few examples in the literature. Compound 152 was prepared starting from 1-(6-hydroxy-2-naphthyl)-1-ethanone (153) via a Bucherer reaction with 2-(methylamino)ethanol (154) followed by Knoevenagel reaction of 155 with malononitrile (156) and [18F] labeling of 157 (Scheme 12) [101,102]. Compound 152 bound to synthetic $A\beta_{1-40}$ fibrils with high affinity ($K_d = 0.12 \text{ nM}$) and crossed the BBB [103]. In addition, PET imaging studies using 152 demonstrated the ability of this compound to determine the localization and load of both SPs and NFTs in living AD patients [104], as well as the ability to differentiate between patients with no cognitive impairment, mild cognitive impairment, and AD [105].

[Re]- and [99mTc]-labeled derivatives of 152 have also been prepared by bromination of 157 with NBS to give 158 followed by conjugation with MAMA-PMB (87) and deprotection with acid to give 159. Reaction with the technetium or rhenium precursors gave the target derivatives 160a,b (Scheme 12) [106]. In vitro binding studies with the [Re]-labeled compound 160a showed a 14-fold decrease in binding affinity for $A\beta_{1-42}$ aggregates compared to 152. In addition, the [99mTc]-labeled compound 160b showed very low brain uptake in normal mice indicating the need for additional refinements of this compound.

Figure 5: Structures of the quinolines discussed.

Combination of known scaffolds

With the success of the benzothiazole and imidazopyridine scaffolds for A β imaging, it was logical to suspect that combination of the two scaffolds into a single molecule could also provide a good imaging agent. Examples of this combination scaffold can be seen in compounds **161–164** (Scheme 13A). IBT (**161**) was prepared by direct coupling of 6-methoxybenzo[d]thiazol-2-amine (**165**) and the nitro substituted α -bromoacetophenone **138** to give **166** followed by installation of the radiolabel (Scheme 13B) [107]. Compound **161** showed good affinity for both A β_{1-40} and A β_{1-42} (K_i = 3.5 nM and 5.8 nM, respectively) and was comparable to compound **56c** in the same assay. The pharmacokinetics of this compound were also similar to those of [11 C]**56c**. In vivo specific plaque labeling by compound **161**

was confirmed through studies in APP/Ps1 transgenic mice. Derivatives of this combination scaffold **162a**–**n** were also investigated (Table 13) [108]. Of note was derivative **162i** in which the secondary methylamino group of **161** has been replaced with iodine. This derivative showed high affinity for $A\beta_{1-40}$ ($K_d = 10.9$ nM), and the iodo substituent could readily be radiolabeled. However, the high lipophilicity of this compound may lead to nonspecific plaque labeling in vivo.

The [125 I]-labeled styrylindole **163** and styrylquinoline **164** scaffolds synthesized by Yang et al. can be thought of as stilbene combination scaffolds [109]. The synthesis of the [125 I]-labeled styrylindole **163** used a Wittig reaction between the triphenyl phosphonium ylide **167** and 1 H-indole-5-carbalde-

Scheme 13: A. Structures of the radiolabeled analogues resulting from the combination of various scaffolds. B., C. Synthetic schemes for the prepar-

Table 13: Inhibition constants of 2-arylimidazobenzothiazole derivatives **162a**–**n** (values are from [108]).

ation of 161 and 163

Compound	R ¹	R ²	$A\beta_{1-40} K_i (nM)$
162a	OMe	NH ₂	29.8 ± 2.1
162b	OMe	NMe ₂	58.6 ± 4.7
162c	F	NH_2	133 ± 21
162d	F	NHMe	38.1 ± 2.6
162e	F	NMe ₂	42.9 ± 5.7
162f	Br	NH_2	28.8 ± 1.2
162g	Br	NHMe	34.5 ± 3.5
162h	Br	NMe ₂	43.4 ± 5.7
162i	OMe	I	10.9 ± 0.18
162j	F	I	41.9 ± 5.2
162k	Br	I	21.1 ± 0.9
162I	Me	I	17.7 ± 1.9
162m	OMe	Br	9.40 ± 0.07
162n	Me	Br	26.0 ± 0.9

hyde (27) to give 168 followed by radiolabeling (Scheme 13C). The [125 I]-labeled styrylquinoline 164 was prepared by using an identical synthesis with substitution of the indole by quinoline-6-carbaldehyde. Both 163 and 164 showed good affinity for A β_{1-40} aggregates ($K_i = 4.1$ nM and 8.6 nM, respectively). Compound 163 was able to stain A β plaques in in vitro brain

sections from APP/Ps1 transgenic mice and showed high uptake and rapid clearance in normal mice (4.27% ID/g at 2 min and 0.28% ID/g at 60 min). However, compound **164** showed relatively low brain uptake and slow washout by comparison.

Others

Several other less common scaffolds have been evaluated as A β -imaging agents. The [125 I]-labeled N-methyl-4-anilino-phthalimide derivative **169** was prepared and evaluated as a potential probe for A β plaques (Scheme 14A) [110]. This compound was generated via a Cu powder-catalyzed coupling reaction between N-methyl-4-aminophthalimide (**170**) and 1-bromo-4-iodobenzene (**171**) to give **172**, which was then radiolabeled. Compound **169** showed high binding affinity to AD brain homogenates (K_d = 0.21 nM) as well as excellent brain uptake (5.16% ID/g at 2 min) and fast washout (0.59% ID/g at 60 min). SAR studies with other N-methyl-4-anilinophthalimide derivatives demonstrated that a hydrophobic substituent at the 4-position of the aniline ring is important for the binding affinity of this family of compounds.

The [125 I]-labeled quinoxaline derivative **173** was also synthesized and evaluated for in vivo imaging of A β plaques (Scheme 14B) [111]. The quinoxaline backbone of this compound was prepared from the reaction of α -bromoacetophenone **144** and 4-bromobenzene-1,2-diamine (**147**) in DMSO in a one-

pot tandem oxide condensation procedure. This reaction gave the desired 2-aryl-6-substituted quinoxaline 174 as the major product and the isomeric 2-aryl-7-substituted quinoxaline (not shown) as a minor product. The radioiodinated probe was prepared from 174. Compound 173 showed excellent affinity for $A\beta_{1-42}$ aggregates in vitro ($K_i = 4.1$ nM). In addition to being able to specifically label plaques in brain sections from AD patients, 173 readily crossed the BBB showing high uptake

into the brain (6.03% ID/g at 2 min). However, with moderate washout (1.12% ID/g at 120 min), additional refinements will be needed to improve the pharmacokinetics of these molecules.

The boron [125 I]-labeled dipyrro-methane (BODIPY) analogue 175 was prepared to serve as a dual functional SPECT/fluorescence probe for imaging A β (Scheme 14C) [112]. Compound 175 was synthesized through Suzuki coupling of the starting boronic acid 176 with 1-bromo-4-iodobenzene (171). Aldehyde reduction of 177 followed by reaction with triphenylphosphine gave the Wittig reagent 178 for reaction with 2-formylpyrrole (179). The Wittig product 180 was condensed with 3,5-dimethylpyrrole-2-carboxaldehyde (181) to form the BODIPY backbone 182. Subsequent installation of the radiolabel gave 175. Although 175 showed decent affinity for $A\beta_{1-42}$ aggregates ($K_i = 108$ nM) and the ability to label plaques in brain sections from transgenic mice, its in vivo use was limited by extremely low brain uptake, which could be attributed to rapid trapping of the compound in the liver.

Fluorescence probes

Although PET is currently the most promising approach for A β plaque detection, this technique has two main limitations: (i) the short half-life of positron-emitting nuclei ($t_{1/2} = 20$ min for [11 C] and 110 min for [18 F]) and (ii) the narrow availability of this technology that requires a local cyclotron for generating short-lived positron-emitting radionuclides and a synthetic unit to produce radiolabeled agents [113]. Other imaging technologies have been investigated to overcome these problems. Different fluorescence techniques have been reported [114-116]; however, the near-infrared fluorescence (NIRF) imaging

technique is the only one that has an in vivo application. Since normal biological tissues reveal limited photon absorbance in the near-infrared region, NIRF seems to be the second most promising A β deposits tracer tool [117]. In the following sections, we will briefly cover the different scaffolds that have been explored as NIRF ligands using in vivo models.

Oxazines

The oxazine dyes **183–186** were investigated as A β aggregate target-specific probes in the NIRF imaging technique (Scheme 15A) [113]. The preparation of **183** was accomplished through reaction of 4-methyl-3,4-dihydro-2H-benzo[b][1,4]oxazin-6-ol (**187**) with p-nitrobenzenediazonium ion (**188**) to give the key azo intermediate **189**, which afforded the desired oxazine dye **183** upon further reaction with 4-methyl-3,4-dihydro-2H-benzo[b][1,4]oxazin-6-ol (**187**) (Scheme 15B). Compound **183** proved to give a higher fluorescence intensity than other derivatives such as **184–186** [113]. Using APP23 transgenic mice and compound **183**, A β plaques could be traced quantitatively [113].

Thiobarbitals

The thiopental dimer THK-265 (**190**) (maximal emission wavelength >650 nm) was discovered as a good NIRF imaging ligand by screening a large pool of dye candidates (Figure 6) [117]. Compound **190** displayed high binding affinity towards A β aggregates ($K_d = 97$ nM) [117]. Its usefulness in AD diagnosis was confirmed in an animal model as it provided good discrimination between amyloid deposits in the brain and other normal tissues [117]. Compound **190** was also used in a quantitative correlation of different A β aggregation levels with NIRF signals [118].

Scheme 15: A. Structures of the oxazine-derived fluorescence probes discussed. B. Synthetic scheme for the preparation of the oxazine analogue 183.

Figure 6: Structure of THK-265 (190).

Quinoxalines

The use of the radiolabeled quinoxalines for imaging $A\beta$ was discussed in the section "Others" of this review. A quinoxaline derivative, compound 191, was also synthesized and explored for fluorescence imaging (Scheme 16) [119]. The synthesis of 191 began with conversion of the starting lactam 192 to the corresponding chloride by using phosphorus oxychloride followed by reaction with hydrazine. Condensation of the resulting hydrazino-derivative 193 with quinoline-4-carboxaldehyde (194) gave 191. Although 191 has not been tested in vivo yet, this compound warrants further investigation as it has shown the ability to selectively stain amyloid structures in brain sections of transgenic mice, as well as the ability to cross the BBB. In addition, the 7-fluoro substituent of compound 191 could potentially be radiolabeled for in vivo application.

Conclusion

In summary, this review covered the main scaffolds used for radioimaging of $A\beta$ plaques, one of the major pathological hall-marks of AD. Highlighted were important synthetic steps for scaffold formation and introduction of radiolabels, SAR findings where appropriate, and binding affinities and brain kinetics of each scaffold. The synthesis of each scaffold presented was fairly straightforward using well-established reactions, and synthetic complexity will likely not impede future development of $A\beta$ chemical probes.

Most of the scaffolds present compounds with good binding affinity for $A\beta$ in vitro. SARs tended to vary between scaffolds so it is impossible, without extensive computational work, to

declare certain functionalities necessary for this class of chemical probes. However, the dimethylamine structural feature appears in a number of the compounds with high binding affinity, and it is likely that this functionality is important for Aβ interaction. Representative examples of this can be seen in examining SAR trends for chalcones 18a–l, benzothiazoles 56c–t, or imidazopyridines 141a–w among others. With regards to pharmacokinetics in vivo, results varied between scaffolds and radiolabels. In general, [99mTc]-labeled compounds showed poor pharmacokinetic profiles with only 125b being able to label plaques in animal studies. A balance in lipophilicity appeared to be particularly important in terms of pharmacokinetics, as imaging probes need to be lipophilic enough to easily penetrate the BBB, but not too lipophilic to avoid nonspecific binding in the brain.

Many of the molecules described in this review showed very promising results for the in vivo imaging of AB plaques in humans. For example, stilbenes 46a and 46b, benzothiazole [¹¹C]**56c**, and naphthalene **152** have been studied in clinical trials with favorable results. The half-life of the radiolabel and overall lipophilicity will continue to be two of the biggest factors for the clinical success of these probes. Future development and testing of these molecules will be of critical importance as the development of Aß imaging probes will provide an effective means of monitoring new treatments for AD. While not the focus of this review, it should be noted that the current treatments for AD only treat cognitive symptoms and have little to no effect on slowing or reversing the progression of the disease. Current research efforts aimed at developing molecules that target AB plaques, specifically inhibition of plaque formation and disaggregation of already formed plaques, could lead to new therapeutics capable of reversing the progression of AD. Probes such as those described herein will play an important role in evaluating the effectiveness of such drugs. Additionally, the development of Aβ imaging probes will likely lead to better and earlier diagnosis of AD, which in turn will allow future treatments to be more effective.

Acknowledgements

Our work on AD is supported by an Alzheimer's Art Quilt Initiative (AAQI) grant (S.G.-T.). We would like to acknowledge the work on the development of probes for AD of those that are not cited in this review due to the scope of the manuscript.

References

- Thies, W.; Bleiler, L. Alzheimers Dement. 2011, 7, 208. doi:10.1016/j.jalz.2011.02.004
- Ono, M.; Saji, H. J. Pharmacol. Sci. 2012, 118, 338 doi:10.1254/jphs.11R08FM
- Klohs, J.; Rudin, M. Neuroscientist 2011, 17, 539. doi:10.1177/1073858410383433
- Rowe, C. C.; Ackerman, U.; Browne, W.; Mulligan, R.; Pike, K. L.; O'Keefe, G.; Tochon-Danguy, H.; Chan, G.; Berlangieri, S. U.; Jones, G.; Dickinson-Rowe, K. L.; Kung, H. P.; Zhang, W.; Kung, M. P.; Skovronsky, D.; Dyrks, T.; Holl, G.; Krause, S.; Friebe, M.; Lehman, L.; Lindemann, S.; Dinkelborg, L. M.; Masters, C. L.; Villemagne, V. L. Lancet Neurol. 2008, 7, 129. doi:10.1016/S1474-4422(08)70001-2
- Pike, K. E.; Savage, G.; Villemagne, V. L.; Ng, S.; Moss, S. A.; Maruff, P.; Mathis, C. A.; Klunk, W. E.; Masters, C. L.; Rowe, C. C. *Brain* 2007, 130, 2837. doi:10.1093/brain/awm238
- Lockhart, A.; Ye, L.; Judd, D. B.; Merritt, A. T.; Lowe, P. N.; Morgenstern, J. L.; Hong, G.; Gee, A. D.; Brown, J. J. Biol. Chem. 2005, 280, 7677. doi:10.1074/jbc.M412056200
- Thompson, P. W.; Ye, L.; Morgenstern, J. L.; Sue, L.; Beach, T. G.; Judd, D. J.; Shipley, N. J.; Libri, V.; Lockhart, A. J. Neurochem. 2009, 109, 623. doi:10.1111/j.1471-4159.2009.05996.x
- Ye, L.; Morgenstern, J. L.; Lamb, J. R.; Lockhart, A. Biochem. Biophys. Res. Commun. 2006, 347, 669. doi:10.1016/j.bbrc.2006.06.126
- Ye, L.; Morgenstern, J. L.; Gee, A. D.; Hong, G.; Brown, J.; Lockhart, A. J. Biol. Chem. 2005, 280, 23599. doi:10.1074/jbc.M501285200
- Shan, L. In Molecular Imaging and Contrast Agent Database (MICAD) Bethesda (MD), 2004.
- Sabaté, R.; Estelrich, J. Biopolymers 2003, 72, 455. doi:10.1002/bip.10485
- Ono, M.; Ikeoka, R.; Watanabe, H.; Kimura, H.; Fuchigami, T.; Haratake, M.; Saji, H.; Nakayama, M. ACS Chem. Neurosci. 2010, 1, 598. doi:10.1021/cn100042d
- Maya, Y.; Ono, M.; Watanabe, H.; Haratake, M.; Saji, H.; Nakayama, M. *Bioconjugate Chem.* **2009**, *20*, 95. doi:10.1021/bc8003292
- Ono, M.; Watanabe, R.; Kawashima, H.; Cheng, Y.; Kimura, H.;
 Watanabe, H.; Haratake, M.; Saji, H.; Nakayama, M. *J. Med. Chem.* 2009, 52, 6394. doi:10.1021/jm901057p
- Cui, M.; Ono, M.; Kimura, H.; Liu, B. L.; Saji, H. Bioorg. Med. Chem. Lett. 2011, 21, 980. doi:10.1016/j.bmcl.2010.12.045
- Ono, M.; Watanabe, R.; Kawashima, H.; Kawai, T.; Watanabe, H.; Haratake, M.; Saji, H.; Nakayama, M. *Bioorg. Med. Chem.* **2009**, *17*, 2069. doi:10.1016/j.bmc.2009.01.025

- Ares, J. J.; Outt, P. E.; Randall, J. L.; Murray, P. D.; Weisshaar, P. S.;
 O'Brien, L. M.; Ems, B. L.; Kakodkar, S. V.; Kelm, G. R.;
 Kershaw, W. C.; Werchowski, K. M.; Parkinson, A. J. Med. Chem.
 1995, 38, 4937. doi:10.1021/jm00025a011
- Ono, M.; Ikeoka, R.; Watanabe, H.; Kimura, H.; Fuchigami, T.; Haratake, M.; Saji, H.; Nakayama, M. *Bioorg. Med. Chem. Lett.* 2010, 20, 5743. doi:10.1016/j.bmcl.2010.08.004
- Ono, M.; Maya, Y.; Haratake, M.; Ito, K.; Mori, H.; Nakayama, M. Biochem. Biophys. Res. Commun. 2007, 361, 116. doi:10.1016/j.bbrc.2007.06.162
- Watanabe, H.; Ono, M.; Kimura, H.; Kagawa, S.; Nishii, R.; Fuchigami, T.; Haratake, M.; Nakayama, M.; Saji, H. Bioorg. Med. Chem. Lett. 2011, 21, 6519. doi:10.1016/j.bmcl.2011.08.063
- Zhang, W.; Oya, S.; Kung, M.-P.; Hou, C.; Maier, D. L.; Kung, H. F. Nucl. Med. Biol. 2005, 32, 799. doi:10.1016/j.nucmedbio.2005.06.001
- Ono, M.; Wilson, A.; Nobrega, J.; Westaway, D.; Verhoeff, P.;
 Zhuang, Z.-P.; Kung, M.-P.; Kung, H. F. *Nucl. Med. Biol.* 2003, 30, 565. doi:10.1016/S0969-8051(03)00049-0
- Camus, V.; Payoux, P.; Barré, L.; Desgranges, B.; Voisin, T.; Tauber, C.; La Joie, R.; Tafani, M.; Hommet, C.; Chételat, G.; Mondon, K.; de La Sayette, V.; Cottier, J. P.; Beaufils, E.; Ribeiro, M. J.; Gissot, V.; Vierron, E.; Vercouillie, J.; Vellas, B.; Eustache, F.; Guilloteau, D. Eur. J. Nucl. Med. Mol. Imaging 2012, 39, 621. doi:10.1007/s00259-011-2021-8
- O'Keefe, G. J.; Saunder, T. H.; Ng, S.; Ackerman, U.; Tochon-Danguy, H. J.; Chan, J. G.; Gong, S.; Dyrks, T.; Lindemann, S.; Holl, G.; Dinkelborg, L.; Villemagne, V.; Rowe, C. C. J. Nucl. Med. 2009, 50, 309. doi:10.2967/jnumed.108.055756
- Barthel, H.; Luthardt, J.; Becker, G.; Patt, M.; Hammerstein, E.; Hartwig, K.; Eggers, B.; Sattler, B.; Schildan, A.; Hesse, S.; Meyer, P. M.; Wolf, H.; Zimmermann, T.; Reischl, J.; Rohde, B.; Gertz, H.-J.; Reininger, C.; Sabri, O. *Eur. J. Nucl. Med. Mol. Imaging* 2011, 38, 1702. doi:10.1007/s00259-011-1821-1
- Zhang, W.; Oya, S.; Kung, M.-P.; Hou, C.; Maier, D. L.; Kung, H. F.
 J. Med. Chem. 2005, 48, 5980. doi:10.1021/jm050166g
- Villemagne, V. L.; Ong, K.; Mulligan, R. S.; Holl, G.; Pejoska, S.; Jones, G.; O'Keefe, G.; Ackerman, U.; Tochon-Danguy, H.; Chan, J. G.; Reininger, C. B.; Fels, L.; Putz, B.; Rohde, B.; Masters, C. L.; Rowe, C. C. J. Nucl. Med. 2011, 52, 1210. doi:10.2967/jnumed.111.089730
- 28. Barthel, H.; Sabri, O. J. Alzheimer's Dis. 2011, 26 (Suppl. 3), 117.
- Yao, C.-H.; Lin, K.-J.; Weng, C.-C.; Hsiao, I.-T.; Ting, Y.-S.;
 Yen, T.-C.; Jan, T.-R.; Skovronsky, D.; Kung, M.-P.; Wey, S.-P.
 Appl. Radiat. Isot. 2010, 68, 2293. doi:10.1016/j.apradiso.2010.07.001
- Choi, S. R.; Golding, G.; Zhuang, Z.; Zhang, W.; Lim, N.; Hefti, F.; Benedum, T. E.; Kilbourn, M. R.; Skovronsky, D.; Kung, H. F. J. Nucl. Med. 2009, 50, 1887. doi:10.2967/jnumed.109.065284
- Lin, K.-J.; Hsu, W.-C.; Hsiao, I.-T.; Wey, S.-P.; Jin, L.-W.;
 Skovronsky, D.; Wai, Y.-Y.; Chang, H.-P.; Lo, C.-W.; Yao, C. H.;
 Yen, T.-C.; Kung, M.-P. Nucl. Med. Biol. 2010, 37, 497.
 doi:10.1016/j.nucmedbio.2010.02.003
- Sperling, R. A.; Johnson, K. A.; Doraiswamy, P. M.; Reiman, E. M.; Fleisher, A. S.; Sabbagh, M. N.; Sadowsky, C. H.; Carpenter, A.; Davis, M. D.; Lu, M.; Flitter, M.; Joshi, A. D.; Clark, C. M.; Grundman, M.; Mintun, M. A.; Skovronsky, D. M.; Pontecorvo, M. J. Neurobiol. Aging 2013, 34, 822. doi:10.1016/j.neurobiolaging.2012.06.014
- 33. Tateno, A.; Okubo, Y. Nippon Rinsho 2011, 69 (Suppl. 8), 561.

- 34. Kung, M.-P.; Weng, C.-C.; Lin, K.-J.; Hsiao, I.-T.; Yen, T.-C.; Wey, S.-P. Chang Gung Med. J. **2012**, *35*, 211.
- Hsiao, I.-T.; Huang, C.-C.; Hsieh, C.-J.; Hsu, W.-C.; Wey, S.-P.;
 Yen, T.-C.; Kung, M.-P.; Lin, K.-J. Eur. J. Nucl. Med. Mol. Imaging
 2012, 39, 1513. doi:10.1007/s00259-012-2137-5
- Ballinger, J. R. Eur. J. Nucl. Med. Mol. Imaging 2012, 39, 1512. doi:10.1007/s00259-012-2136-6
- Huang, K.-L.; Lin, K.-J.; Ho, M.-Y.; Chang, Y.-J.; Chang, C.-H.;
 Wey, S.-P.; Hsieh, C.-J.; Yen, T.-C.; Hsiao, I.-T.; Lee, T.-H.
 J. Neurol. Sci. 2012, 319, 124. doi:10.1016/j.jns.2012.04.014
- Poisnel, G.; Dhilly, M.; Moustié, O.; Delamare, J.; Abbas, A.; Guilloteau, D.; Barré, L. Neurobiol. Aging 2012, 33, 2561. doi:10.1016/j.neurobiolaging.2011.12.024
- Hsiao, I.-T.; Huang, C.-C.; Hsieh, C.-J.; Hsu, W.-C.; Wey, S.-P.;
 Yen, T.-C.; Kung, M.-P.; Lin, K.-J. Eur. J. Nucl. Med. Mol. Imaging
 2012, 39, 613. doi:10.1007/s00259-011-2051-2
- Wong, D. F.; Rosenberg, P. B.; Zhou, Y.; Kumar, A.; Raymont, V.; Ravert, H. T.; Dannals, R. F.; Nandi, A.; Brašić, J. R.; Ye, W.; Hilton, J.; Lyketsos, C.; Kung, H. F.; Joshi, A. D.; Skovronsky, D. M.; Pontecorvo, M. J. J. Nucl. Med. 2010, 51, 913. doi:10.2967/inumed.109.069088
- Wang, H.; Shi, H.; Yu, H.; Jiang, S.; Tang, G. Nucl. Med. Biol. 2011, 38, 121. doi:10.1016/j.nucmedbio.2010.06.009
- Huynh, T. H. V.; Mantel, M. L.; Mikkelsen, K.; Lindhardt, A. T.;
 Nielsen, N. C.; Otzen, D.; Skrydstrup, T. Org. Lett. 2009, 11, 999. doi:10.1021/ol8029593
- Wilson, A. A.; Garcia, A.; Jin, L.; Houle, S. Nucl. Med. Biol. 2000, 27, 529. doi:10.1016/S0969-8051(00)00132-3
- Mayhoub, A. S.; Marler, L.; Kondratyuk, T. P.; Park, E.-J.;
 Pezzuto, J. M.; Cushman, M. *Bioorg. Med. Chem.* 2012, 20, 2427.
 doi:10.1016/j.bmc.2012.01.047
- Mayhoub, A. S.; Marler, L.; Kondratyuk, T. P.; Park, E.-J.;
 Pezzuto, J. M.; Cushman, M. *Bioorg. Med. Chem.* 2012, 20, 510.
 doi:10.1016/j.bmc.2011.09.031
- Mayhoub, A. S.; Kiselev, E.; Cushman, M. Tetrahedron Lett. 2011, 52, 4941. doi:10.1016/j.tetlet.2011.07.068
- Watanabe, H.; Ono, M.; Ikeoka, R.; Haratake, M.; Saji, H.; Nakayama, M. *Bioorg. Med. Chem.* **2009**, *17*, 6402. doi:10.1016/j.bmc.2009.07.020
- 48. Ono, M.; Haratake, M.; Saji, H.; Nakayama, M. *Bioorg. Med. Chem.* **2008**, *16*, 6867. doi:10.1016/j.bmc.2008.05.054
- Klunk, W. E.; Wang, Y.; Huang, G.-F.; Debnath, M. L.; Holt, D. P.; Mathis, C. A. *Life Sci.* 2001, 69, 1471. doi:10.1016/S0024-3205(01)01232-2
- Mathis, C. A.; Bacskai, B. J.; Kajdasz, S. T.; McLellan, M. E.; Frosch, M. P.; Hyman, B. T.; Holt, D. P.; Wang, Y.; Huang, G.-F.; Debnath, M. L.; Klunk, W. E. *Bioorg. Med. Chem. Lett.* 2002, *12*, 295. doi:10.1016/S0960-894X(01)00734-X
- Klunk, W. E.; Wang, Y.; Huang, G. F.; Debnath, M. L.; Holt, D. P.; Shao, L.; Hamilton, R. L.; Ikonomovic, M. D.; DeKosky, S. T.; Mathis, C. A. J. Neurosci. 2003, 23, 2086.
- Mathis, C. A.; Wang, Y.; Holt, D. P.; Huang, G.-F.; Debnath, M. L.;
 Klunk, W. E. J. Med. Chem. 2003, 46, 2740. doi:10.1021/jm030026b
- Solbach, C.; Uebele, M.; Reischl, G.; Machulla, H.-J.
 Appl. Radiat. Isot. 2005, 62, 591. doi:10.1016/j.apradiso.2004.09.003
- 54. Klunk, W. E.; Engler, H.; Nordberg, A.; Wang, Y.; Blomqvist, G.; Holt, D. P.; Bergström, M.; Savitcheva, I.; Huang, G.-F.; Estrada, S.; Ausén, B.; Debnath, M. L.; Barletta, J.; Price, J. C.; Sandell, J.; Lopresti, B. J.; Wall, A.; Koivisto, P.; Antoni, G.; Mathis, C. A.; Långström, B. *Ann. Neurol.* 2004, *55*, 306. doi:10.1002/ana.20009

- Svedberg, M. M.; Hall, H.; Hellström-Lindahl, E.; Estrada, S.;
 Guan, Z.; Nordberg, A.; Långström, B. *Neurochem. Int.* **2009**, *54*, 347. doi:10.1016/j.neuint.2008.12.016
- Serdons, K.; Verduyckt, T.; Vanderghinste, D.; Borghgraef, P.; Cleynhens, J.; Van Leuven, F.; Kung, H.; Bormans, G.; Verbruggen, A. Eur. J. Med. Chem. 2009, 44, 1415. doi:10.1016/j.ejmech.2008.09.038
- Johnson, A. E.; Jeppsson, F.; Sandell, J.; Wensbo, D.;
 Neelissen, J. A. M.; Juréus, A.; Ström, P.; Norman, H.; Farde, L.;
 Svensson, S. P. S. J. Neurochem. 2009, 108, 1177.
 doi:10.1111/i.1471-4159.2008.05861.x
- Zhuang, Z.-P.; Kung, M.-P.; Hou, C.; Skovronsky, D. M.; Gur, T. L.;
 Plössl, K.; Trojanowski, J. Q.; Lee, V. M.-Y.; Kung, H. F.
 J. Med. Chem. 2001, 44, 1905. doi:10.1021/jm010045q
- Wang, Y.; Mathis, C. A.; Huang, G.-F.; Debnath, M. L.; Holt, D. P.;
 Shao, L.; Klunk, W. E. J. Mol. Neurosci. 2003, 20, 255.
 doi:10.1385/JMN:20:3:255
- Matsumura, K.; Ono, M.; Hayashi, S.; Kimura, H.; Okamoto, Y.;
 Ihara, M.; Takahashi, R.; Mori, H.; Saji, H. MedChemComm 2011, 2, 596. doi:10.1039/c1md00034a
- Zheng, M.-Q.; Yin, D.-Z.; Qia, J.-P.; Zhang, L.; Wang, Y.-X.
 J. Fluorine Chem. 2008, 129, 210. doi:10.1016/j.jfluchem.2007.11.005
- Zheng, M.-Q.; Yin, D.-Z.; Zhang, L.; Lei, B.; Cheng, D.-F.; Cai, H.-C.; Han, Y.-J.; Wu, M.-X.; Zhang, H.; Wang, J. *Acta Pharmacol. Sin.* 2008, 29, 548. doi:10.1111/j.1745-7254.2008.00785.x
- Neumaier, B.; Deisenhofer, S.; Sommer, C.; Solbach, C.;
 Reske, S. N.; Mottaghy, F. Appl. Radiat. Isot. 2010, 68, 1066.
 doi:10.1016/j.apradiso.2009.12.044
- Berndt, U.; Stanetty, C.; Wanek, T.; Kuntner, C.; Stanek, J.;
 Berger, M.; Bauer, M.; Henriksen, G.; Wester, H.-J.; Kvaternik, H.;
 Angelberger, P.; Noe, C. J. Labelled Compd. Radiopharm. 2008, 51, 137. doi:10.1002/jlcr.1476
- Koole, M.; Lewis, D. M.; Buckley, C.; Nelissen, N.; Vandenbulcke, M.; Brooks, D. J.; Vandenberghe, R.; Van Laere, K. J. Nucl. Med. 2009, 50, 818. doi:10.2967/jnumed.108.060756
- Serdons, K.; Verduyckt, T.; Vanderghinste, D.; Cleynhens, J.; Borghgraef, P.; Vermaelen, P.; Terwinghe, C.; Van Leuven, F.; Van Laere, K.; Kung, H.; Bormans, G.; Verbruggen, A. *Bioorg. Med. Chem. Lett.* 2009, *19*, 602. doi:10.1016/j.bmcl.2008.12.069
- Serdons, K.; Terwinghe, C.; Vermaelen, P.; Van Laere, K.; Kung, H.; Mortelmans, L.; Bormans, G.; Verbruggen, A. J. Med. Chem. 2009, 52, 1428. doi:10.1021/jm8013376
- Chen, X.; Yu, P.; Zhang, L.; Liu, B. Bioorg. Med. Chem. Lett. 2008, 18, 1442. doi:10.1016/j.bmcl.2007.12.071
- Serdons, K.; Verduyckt, T.; Cleynhens, J.; Terwinghe, C.; Mortelmans, L.; Bormans, G.; Verbruggen, A. *Bioorg. Med. Chem. Lett.* 2007, *17*, 6086. doi:10.1016/j.bmcl.2007.09.055
- Zhuang, Z.-P.; Kung, M.-P.; Hou, C.; Plössl, K.; Skovronsky, D.;
 Gur, T. L.; Trojanowski, J. Q.; Lee, V. M.-Y.; Kung, H. F.
 Nucl. Med. Biol. 2001, 28, 887. doi:10.1016/S0969-8051(01)00264-5
- Hausner, S. H.; Alagille, D.; Koren, A. O.; Amici, L.; Staley, J. K.; Cosgrove, K. P.; Baldwin, R. M.; Tamagnan, G. D. *Bioorg. Med. Chem. Lett.* **2009**, *19*, 543. doi:10.1016/j.bmcl.2008.05.033
- Cui, M.; Ono, M.; Kimura, H.; Ueda, M.; Nakamoto, Y.; Togashi, K.; Okamoto, Y.; Ihara, M.; Takahashi, R.; Liu, B.; Saji, H. *J. Med. Chem.* 2012, 55, 9136. doi:10.1021/jm300251n

- Okamura, N.; Suemoto, T.; Shiomitsu, T.; Suzuki, M.; Shimadzu, H.; Akatsu, H.; Yamamoto, T.; Arai, H.; Sasaki, H.; Yanai, K.; Staufenbiel, M.; Kudo, Y.; Sawada, T. J. Mol. Neurosci. 2004, 24, 247. doi:10.1385/JMN:24:2:247
- Okamura, N.; Suemoto, T.; Shimadzu, H.; Suzuki, M.; Shiomitsu, T.; Akatsu, H.; Yamamoto, T.; Staufenbiel, M.; Yanai, K.; Arai, H.; Sasaki, H.; Kudo, Y.; Sawada, T. J. Neurosci. 2004, 24, 2535. doi:10.1523/JNEUROSCI.4456-03.2004
- Shimadzu, H.; Suemoto, T.; Suzuki, M.; Shiomitsu, T.; Okamura, N.; Kudo, Y.; Sawada, T. J. Labelled Compd. Radiopharm. 2004, 47, 181. doi:10.1002/jlcr.811
- Kudo, Y.; Okamura, N.; Furumoto, S.; Tashiro, M.; Furukawa, K.; Maruyama, M.; Itoh, M.; Iwata, R.; Yanai, K.; Arai, H. *J. Nucl. Med.* 2007, 48, 553. doi:10.2967/jnumed.106.037556
- Furukawa, K.; Okamura, N.; Tashiro, M.; Waragai, M.; Furumoto, S.; lwata, R.; Yanai, K.; Kudo, Y.; Arai, H. J. Neurol. 2010, 257, 721. doi:10.1007/s00415-009-5396-8
- Shao, H.; Okamura, N.; Sugi, K.; Furumoto, S.; Furukawa, K.; Tashiro, M.; Iwata, R.; Matsuda, H.; Kudo, Y.; Arai, H.; Fukuda, H.; Yanai, K. Dementia Geriatr. Cognit. Disord. 2010, 30, 101. doi:10.1159/000318754
- Okamura, N.; Shiga, Y.; Furumoto, S.; Tashiro, M.; Tsuboi, Y.; Furukawa, K.; Yanai, K.; Iwata, R.; Arai, H.; Kudo, Y.; Itoyama, Y.; Doh-ura, K. Eur. J. Nucl. Med. Mol. Imaging 2010, 37, 934. doi:10.1007/s00259-009-1314-7
- Kikuchi, A.; Takeda, A.; Okamura, N.; Tashiro, M.; Hasegawa, T.; Furumoto, S.; Kobayashi, M.; Sugeno, N.; Baba, T.; Miki, Y.; Mori, F.; Wakabayashi, K.; Funaki, Y.; Iwata, R.; Takahashi, S.; Fukuda, H.; Arai, H.; Kudo, Y.; Yanai, K.; Itoyama, Y. Brain 2010, 133, 1772. doi:10.1093/brain/awq091
- Fodero-Tavoletti, M. T.; Mulligan, R. S.; Okamura, N.; Furumoto, S.; Rowe, C. C.; Kudo, Y.; Masters, C. L.; Cappai, R.; Yanai, K.; Villemagne, V. L. Eur. J. Pharmacol. 2009, 617, 54. doi:10.1016/j.ejphar.2009.06.042
- Ono, M.; Kawashima, H.; Nonaka, A.; Kawai, T.; Haratake, M.; Mori, H.; Kung, M.-P.; Kung, H. F.; Saji, H.; Nakayama, M. *J. Med. Chem.* 2006, 49, 2725. doi:10.1021/jm051176k
- Ono, M.; Kung, M.-P.; Hou, C.; Kung, H. F. Nucl. Med. Biol. 2002, 29, 633. doi:10.1016/S0969-8051(02)00326-8
- Cheng, Y.; Ono, M.; Kimura, H.; Kagawa, S.; Nishii, R.; Saji, H. Bioorg. Med. Chem. Lett. 2010, 20, 6141. doi:10.1016/i.bmcl.2010.08.016
- Cheng, Y.; Ono, M.; Kimura, H.; Kagawa, S.; Nishii, R.; Kawashima, H.; Saji, H. ACS Med. Chem. Lett. 2010, 1, 321. doi:10.1021/ml100082x
- Ono, M.; Cheng, Y.; Kimura, H.; Cui, M.; Kagawa, S.; Nishii, R.;
 Saji, H. J. Med. Chem. 2011, 54, 2971. doi:10.1021/jm200057u
- Juréus, A.; Swahn, B.-M.; Sandell, J.; Jeppsson, F.; Johnson, A. E.; Johnström, P.; Neelissen, J. A. M.; Sunnemark, D.; Farde, L.; Svensson, S. P. S. *J. Neurochem.* 2010, 114, 784. doi:10.1111/j.1471-4159.2010.06812.x
- Cheng, Y.; Ono, M.; Kimura, H.; Ueda, M.; Saji, H. J. Med. Chem. 2012, 55, 2279. doi:10.1021/jm201513c
- Cui, M.; Ono, M.; Kimura, H.; Liu, B.; Saji, H. Bioorg. Med. Chem. 2011, 19, 4148. doi:10.1016/j.bmc.2011.04.049
- Zhuang, Z.-P.; Kung, M.-P.; Wilson, A.; Lee, C.-W.; Plössl, K.;
 Hou, C.; Holtzman, D. M.; Kung, H. F. J. Med. Chem. 2003, 46, 237.
 doi:10.1021/jm020351j

- Kung, M.-P.; Hou, C.; Zhuang, Z.-P.; Cross, A. J.; Maier, D. L.; Kung, H. F. Eur. J. Nucl. Med. Mol. Imaging 2004, 31, 1136. doi:10.1007/s00259-004-1487-z
- Kung, M.-P.; Hou, C.; Zhuang, Z.-P.; Zhang, B.; Skovronsky, D.;
 Trojanowski, J. Q.; Lee, V. M.-Y.; Kung, H. F. Brain Res. 2002, 956, 202. doi:10.1016/S0006-8993(02)03436-4
- Newberg, A. B.; Wintering, N. A.; Plössl, K.; Hochold, J.;
 Stabin, M. G.; Watson, M.; Skovronsky, D.; Clark, C. M.; Kung, M.-P.;
 Kung, H. F. J. Nucl. Med. 2006, 47, 748.
- Cai, L.; Cuevas, J.; Temme, S.; Herman, M. M.; Dagostin, C.;
 Widdowson, D. A.; Innis, R. B.; Pike, V. W. J. Med. Chem. 2007, 50, 4746. doi:10.1021/im0702231
- Zeng, F.; Southerland, J. A.; Voll, R. J.; Votaw, J. R.; Williams, L.;
 Ciliax, B. J.; Levey, A. I.; Goodman, M. M. Bioorg. Med. Chem. Lett.
 2006, 16, 3015. doi:10.1016/j.bmcl.2006.02.055
- Cai, L.; Chin, F. T.; Pike, V. W.; Toyama, H.; Liow, J.-S.;
 Zoghbi, S. S.; Modell, K.; Briard, E.; Shetty, H. U.; Sinclair, K.;
 Donohue, S.; Tipre, D.; Kung, M.-P.; Dagostin, C.; Widdowson, D. A.;
 Green, M.; Gao, W.; Herman, M. M.; Ichise, M.; Innis, R. B.
 J. Med. Chem. 2004, 47, 2208. doi:10.1021/jm030477w
- Cui, M.; Ono, M.; Kimura, H.; Kawashima, H.; Liu, B. L.; Saji, H.
 Nucl. Med. Biol. 2011, 38, 313. doi:10.1016/j.nucmedbio.2010.09.012
- Vasdev, N.; Cao, P.; van Oosten, E. M.; Wilson, A. A.; Houle, S.;
 Hao, G.; Sun, X.; Slavine, N.; Ahasan, M.; Antich, P. P.; Bonte, F. J.;
 Kulkarni, P. *MedChemComm* 2012, 3, 1228.
 doi:10.1039/c2md20075a
- Okamura, N.; Suemoto, T.; Furumoto, S.; Suzuki, M.; Shimadzu, H.; Akatsu, H.; Yamamoto, T.; Fujiwara, H.; Nemoto, M.; Maruyama, M.; Arai, H.; Yanai, K.; Sawada, T.; Kudo, Y. J. Neurosci. 2005, 25, 10857. doi:10.1523/JNEUROSCI.1738-05.2005
- 100. Fodero-Tavoletti, M. T.; Okamura, N.; Furumoto, S.; Mulligan, R. S.; Connor, A. R.; McLean, C. A.; Cao, D.; Rigopoulos, A.; Cartwright, G. A.; O'Keefe, G.; Gong, S.; Adlard, P. A.; Barnham, K. J.; Rowe, C. C.; Masters, C. L.; Kudo, Y.; Cappai, R.; Yanai, K.; Villemagne, V. L. *Brain* 2011, 134, 1089. doi:10.1093/brain/awr038
- 101. Agdeppa, E. D.; Kepe, V.; Petri, A.; Satyamurthy, N.; Liu, J.; Huang, S.-C.; Small, G. W.; Cole, G. M.; Barrio, J. R. *Neuroscience* 2003, 117, 723. doi:10.1016/S0306-4522(02)00907-7
- 102. Jacobson, A.; Petric, A.; Hogenkamp, D.; Sinur, A.; Barrio, J. R. J. Am. Chem. Soc. 1996, 118, 5572. doi:10.1021/ja9543356
- 103.Agdeppa, E. D.; Kepe, V.; Liu, J.; Flores-Torres, S.; Satyamurthy, N.; Petric, A.; Cole, G. M.; Small, G. W.; Huang, S. C.; Barrio, J. R. J. Neurosci. 2001, 21, RC189.
- 104. Shoghi-Jadid, K.; Small, G. W.; Agdeppa, E. D.; Kepe, V.; Ercoli, L. M.; Siddarth, P.; Read, S.; Satyamurthy, N.; Petric, A.; Huang, S.-C.; Barrio, J. R. Am. J. Geriatr. Psychiatry 2002, 10, 24.
- 105. Small, G. W.; Kepe, V.; Ercoli, L. M.; Siddarth, P.; Bookheimer, S. Y.; Miller, K. J.; Lavretsky, H.; Burggren, A. C.; Cole, G. M.; Vinters, H. V.; Thompson, P. M.; Huang, S.-C.; Satyamurthy, N.; Phelps, M. E.; Barrio, J. R. N. Engl. J. Med. 2006, 355, 2652. doi:10.1056/NEJMoa054625
- 106.Cui, M.; Tang, R.; Li, Z.; Ren, H.; Liu, B. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 1064. doi:10.1016/j.bmcl.2010.11.096
- 107. Yousefi, B. H.; Manook, A.; Drzezga, A.; von Reutern, B.; Schwaiger, M.; Wester, H.-J.; Henriksen, G. *J. Med. Chem.* **2011**, *54*, 949. doi:10.1021/jm101129a
- 108.Alagille, D.; DaCosta, H.; Baldwin, R. M.; Tamagnan, G. D. Bioorg. Med. Chem. Lett. 2011, 21, 2966. doi:10.1016/j.bmcl.2011.03.052

- 109. Yang, Y.; Jia, H.-M.; Liu, B.-L. *Molecules* **2012**, *17*, 4252. doi:10.3390/molecules17044252
- 110. Duan, X.-H.; Qiao, J.-P.; Yang, Y.; Cui, M.-C.; Zhou, J.-N.; Liu, B.-L. *Bioorg. Med. Chem.* **2010**, *18*, 1337. doi:10.1016/j.bmc.2009.12.023
- 111.Cui, M.; Ono, M.; Kimura, H.; Liu, B.; Saji, H. *Bioorg. Med. Chem. Lett.* **2011**, *21*, 4193. doi:10.1016/j.bmcl.2011.05.079
- 112.Ono, M.; Ishikawa, M.; Kimura, H.; Hayashi, S.; Matsumura, K.; Watanabe, H.; Shimizu, Y.; Cheng, Y.; Cui, M.; Kawashima, H.; Saji, H. *Bioorg. Med. Chem. Lett.* **2010**, *20*, 3885. doi:10.1016/j.bmcl.2010.05.027
- 113. Hintersteiner, M.; Enz, A.; Frey, P.; Jaton, A.-L.; Kinzy, W.; Kneuer, R.; Neumann, U.; Rudin, M.; Staufenbiel, M.; Stoeckli, M.; Wiederhold, K.-H.; Gremlich, H.-U. Nat. Biotechnol. 2005, 23, 577. doi:10.1038/nbt1085
- 114. Hu, Y.; Su, B.; Kim, C.-S.; Hernandez, M.; Rostagno, A.; Ghiso, J.; Kim, J. R. ChemBioChem 2010, 11, 2409. doi:10.1002/cbic.201000435
- 115.Hu, Y.; Su, B.; Zheng, H.; Kim, J. R. *Mol. BioSyst.* **2012**, *8*, 2741. doi:10.1039/c2mb25148e
- 116.Marek, P.; Gupta, R.; Raleigh, D. P. *ChemBioChem* **2008**, *9*, 1372. doi:10.1002/cbic.200800052
- 117. Okamura, N.; Mori, M.; Furumoto, S.; Yoshikawa, T.; Harada, R.; Ito, S.; Fujikawa, Y.; Arai, H.; Yanai, K.; Kudo, Y. *J. Alzheimer's Dis.* 2011, 23, 37. doi:10.3233/JAD-2010-100270
- 118. Schmidt, A.; Pahnke, J. J. Alzheimer's Dis. **2012**, *30*, 651. doi:10.3233/JAD-2012-112168
- 119. Gemma, S.; Colombo, L.; Forloni, G.; Savini, L.; Fracasso, C.; Caccia, S.; Salmona, M.; Brindisi, M.; Joshi, B. P.; Tripaldi, P.; Giorgi, G.; Taglialatela-Scafati, O.; Novellino, E.; Fiorini, I.; Campiani, G.; Butini, S. Org. Biomol. Chem. 2011, 9, 5137. doi:10.1039/c1ob05288h

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/2.0), which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions:

(http://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at:

doi:10.3762/bjoc.9.116