

Supporting Information

for

Alloyed Pt_3M (M = Co, Ni) nanoparticles supported on S- and N-doped carbon nanotubes for the oxygen reduction reaction

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Additional experimental data

This file includes:

Table S1: Binding energies and ratio of Pt and Co species obtained from XPS spectra after washing with EDTA solution.

Figure S1: TEM micrographs of: a) CNT; b) N-CNT; c) S-CNT, and d) N-CNT_{HT}. (Scale bar = 50 nm).

Figure S2: Raman spectra at an incident laser energy of 2.33 eV (532 nm) of: a) CNT; b) N-CNT; and c) S-CNT.

Figure S3: Deconvolution of XPS spectra for S-CNT S_{2p}.

Figure S4: Deconvolution of XPS spectra for a) N-CNT N_{1s} , b) N-CNT S_{2p} , and c) N-CNT_{HT} N_{1s} .

Figure S5: a) TEM images of Pt_3Co/CNT synthetized with hexadecyl trimetyl ammonium bromide as stabilizer, and size distribution for the catalyst (the mean particle size is 1.90 ± 0.77 nm). b) TEM images of Pt_3Co/CNT synthetized with [bmim][Tf_2N] as stabilizer, and size distribution for the catalyst (the mean particle size is 2.28 ± 0.82 nm).

Figure S6: Size distribution for a) Pt₃Co/N-CNT, b) Pt₃Co/N-CNT_{HT}, c) Pt₃Co/S-CNT,

d) Pt₃Ni/N-CNT, e) Pt₃Ni/N-CNT_{HT} and f) Pt₃Ni/S-CNT.

Figure S7: HRTEM images of a) Co/N-CNT and c) Ni/N-CNT_{HT}; and STEM-HAADF images of b) Co/N-CNT and d) Ni/N-CNT_{HT}.

Figure S8: a) HREM image and b) STEM-HAADF micrograph of Pt₃Co/N-CNT and EDX spectrum of the selected area.

Figure S9: a) STEM-HAADF micrograph of Pt₃Co/N-CNT and b) EDX spectrum of the selected area.

Figure S10: Deconvolution of XPS spectra for Pt₃Co/N-CNT a) Pt_{4f}, and b) Co_{2p}.

Figure S11: Deconvolution of XPS spectra for Pt₃Ni/N-CNT_{HT} a) Pt_{4f}, and b) Ni_{2p}.

Figure S12: Results of the EDX analysis on Pt₃Co/N-CNT a) before and b) after washing with EDTA solution.

Figure S13: WAXS analyses – a) diffractogram from Pt_3Co/N -CNT before (red) and after (green) washing with EDTA and simulation for a spherical pure model; b) experimental PDF from Pt_3Co/N -CNT before (red) and after (green) washing with EDTA.

Figure S14: Polarization curves of MEA based on Pt_3Co/CB (red) and $Pt_3Co/N-CNT$ (dark blue); recorded under O_2 , $P_{inlet} = 2.5$ bar, T = 80 °C, $St_{H2} = 1.2$; $St_{O2} = 5$, $RH_{anode} = 50\%$; $RH_{cathode} = 30\%$.

Figure S15: Nyquist diagrams of EIS for MEA based on Pt₃Co/CB (red) and Pt₃Co/N-CNT (dark blue) registered under air, $P_{inlet} = 2.5$ bar, T = 80 °C, $St_{H2} = 1.2$; $St_{Air} = 3.5$, $RH_{anode} = 50\%$; $RH_{cathode} = 30\%$.

Tabl	e S1:	Binding	energies	and	ratio	of	Pt	and	Co	species	obtained	from	XPS	spectra	after
wash	ng w	ith EDTA	A solutior	1.											

	S	Binding e	nergy (eV)	Ratio (atom %)	
Pt3C0/N-CN I	Species	Pt 4f _{7/2}	Pt 4f _{5/2}		
Pt	Pt(0)	71.1	74.3	57.8	
	PtO	72.1	75.2	15.8	
	Pt(OH) ₂	73.1	76.4	26.4	
		Co 2p _{3/2}	Co 2p1/2		
Со	Co(0)	778.5	792.5	16.4	
	CoO	780.0	794.5.0	48.9	
	Co(OH) ₂ /CoN ₄	781.8	796.4	34.9	



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Figure S6: Size distribution for a) $Pt_3Co/N-CNT$, b) $Pt_3Co/N-CNT_{HT}$, c) $Pt_3Co/S-CNT$, d) $Pt_3Ni/N-CNT$, e) $Pt_3Ni/N-CNT_{HT}$ and f) $Pt_3Ni/S-CNT$.



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