

## **Supporting Information**

for

# An efficient electrode material for high performance solidstate hybrid supercapacitors based on a Cu/CuO/porous carbon nanofiber/TiO<sub>2</sub> hybrid composite

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Additional experimental data

### Preparation of PVA/H<sub>2</sub>SO<sub>4</sub> gel electrolyte

To prepare  $PVA/H_2SO_4$  electrolyte, in the first step 1 M  $H_2SO_4$  (20 mL) aqueous solution is prepared. Then PVA (2 g) is added into 20 mL of 1 M  $H_2SO_4$  solution to maintain 10 wt % of PVA. The obtained mixture is stirred continuously at 40 °C until a transparent gel is formed.

#### **Fabrication of SSHSC**

To fabricate SSHSC, first supercapacitor electrodes are coated with slurry with paint brush. The slurry is prepared in the manner as prepared for three-electrode configuration maintaining same weight percentage of PVDF binder, conductive carbon and active material. The slurry is coated on carbon cloth (2 cm  $\times$  2 cm) and vacuum dried at 70 °C for 12 h. The SSHSC fabrication requires two electrodes, a polypropylene sheet as separator, PVA/H<sub>2</sub>SO<sub>4</sub> as gel electrolyte, stainless sheets as current collector, perpex sheet and binding clips. Coated carbon clothes are used as both positive and negative electrode. PVA/H<sub>2</sub>SO<sub>4</sub> gel electrolyte is coated on both sides of separator with brush and pressed between two electrodes. Further, this assembly is sandwiched between current collectors along with perpex sheets. Finally, two big size clips are used to bind the fabricated SSHSC for mechanical stability.



**Figure S1:** Typical EDX elemental mapping graphs of a) Cu/CuO/PCNF/TiO<sub>2</sub>, b) C K, c) O K, d) TiK and e) CuK. The scale bar is 500 nm

#### Half-cell studies of Cu/CuO/PCNF/TiO<sub>2</sub> composite

CV test for Cu/CuO/PCNF/TiO<sub>2</sub> composite in the potential range -0.2 to 0.8 for three-electrode configuration were performed using 1 M H<sub>2</sub>SO<sub>4</sub> aqueous electrolyte at different scan rates of 5, 10, 20, 50 and 100 mV s<sup>-1</sup> (Figure S2).



**Figure S2:** CV curves of Cu/CuO/PCNF/TiO<sub>2</sub> composite in three-electrode configuration at different scan rates in 1 M H<sub>2</sub>SO<sub>4</sub>.

#### Effect of amount of TiO<sub>2</sub> on capacitance

To study the effect of amount of  $TiO_2$  on capacitance, we synthesized three different composite materials by varying concentration of titanium tetraisopropoxide 0.001, 0.003 and 0.005 M and labelled as C0001, C0003 and C0005, respectively. In result, as amount of  $TiO_2$  increase, the capacitance values decrease.



Figure S3: (a) CV and (b) GCD curves of C0001, C0003 and C0005, respectively.



Figure S4: FE-SEM images of a) C0001, b) C0003 and c) C0005.



Figure S5: TGA curves for C0001, C0003 and C0005.