Supporting Information

for

Simultaneous cancer control and diagnosis with magnetic nanohybrid materials

Reza Saadat*1 and Franz Renz1

Address: ¹Institute for Inorganic Chemistry, Leibniz University Hannover, Callinstr. 3–9, 30169 Hannover, Germany

Email: Reza Saadat* - rezasaadat.chem-uni.hannover@mail.de

Additional experimental data

Table of contents

General information	2
2. Experimental procedures	2
3. Spectra	5

^{*} Corresponding author

1. General information

All commercially available reagents and solvents were deployed as received from Sigma Adrich Corporation without further purification and without drying. FTIR spectra were performed as KBr pellets with a Bruker Tensor 27 spectrometer. TG-DTA measurements were obtained with a STA 409 from Netzsch with a gas flow of 80% N₂ and 20% O₂. DLS and zeta potential measurements were performed with a Zetasizer 6.01 from Malvern Instruments at pH 7 in water. XRD measurements were obtained with a Stoe StadiP PSD with Cu Kα radiation.

2. Experimental procedures

Synthesis of magnetite nanoparticles

FeCl₂·4H₂O (5 g, 25.1 mmol) and FeCl₃·6 H₂O (13.6 g, 50.3 mmol) were solved in 60 mL distilled water. The solution was stirred 20 min after adding 10 mL of an NH₃ solution (25%) at 25 °C. The resulted black deposit was filtered, washed three times with water and dried under vacuum. The yield was 5.46 g.

Silanization of magnetite nanoparticles

5 g of the magnetite nanoparticles were dispersed in 400 mL Ethanol in an ultrasonic bath. Subsequently, 120 mL of an NH₃ solution (25%), 2.33 mL TEOS (10.5 mmol) and 80 mL distilled water was slowly added within 2 h at 25 °C. The resulted black-brown deposit was separated with a hand-held neodymium magnet, washed twice with ethanol and dried under vacuum. The yield was 5.15 g.

Aminofunctionalisation of silanized magnetite nanoparticles

3.5 g of silanized nanoparticles were dispersed with an ultrasonic bath in 500 mL ethanol. Afterwards 15 mL (3-aminopropyle)triethoxysilane (60 mmol) was added and treated with ultrasound for 12 h at 25 °C. The resulted brown deposit was separated with a hand-held neodymium magnet, washed twice with ethanol and dried under vacuum. The yield was 3.37 g.

Conjunction of the aminofunctionalized nanoparticles with CI-bzimpy

0.03 g of aminofunctionalized nanoparticles were dispersed with an ultrasonic bath in 20 mL ethanol. Subsequently 0.06 g Cl-bzimpy (bzimpy = 2,6-bis(benzimidazol-2-yl)-pyridine, 0.2 mmol) was added and treated with ultrasound for 4 h at 25 °C. The resulted bright-brown deposit was separated with a hand-held neodymium magnet, washed twice with ethanol and dried under vacuum. The yield was 0.029 g.

Conjunction of the nanoparticles to each other

0.02 g of CI-bzimpy nanoparticles were dispersed with an ultrasonic bath in 40 mL ethanol. 0.27 g FeCI₂·4H₂O (1 mmol) and 0.1 g ascorbic acid were added to this dispersion and treated with ultrasound for 2 h at 25 °C. The resulted brown deposit was separated with a hand-held neodymium magnet, washed twice with ethanol and dried under vacuum. The yield was 0.025 g.

Conjunction of Ga-DOTA with the aminofunctionalized nanoparticles

0.15 g of aminofunctionalized nanoparticles were dispersed with an ultrasonic bath in 300 mL DMF. Subsequently, 0.39 g DCC (0.19 mmol), 0.18 g 4-DMAP (1.5 mmol) and 0.1 g Ga-DOTA (0.2 mmol) was added and treated with ultrasound for 24 h at 80 °C. The resulted bright-brown deposit was separated with a hand-held neodymium magnet, washed twice with DMF and dried under vacuum. The yield was 0.152 g.

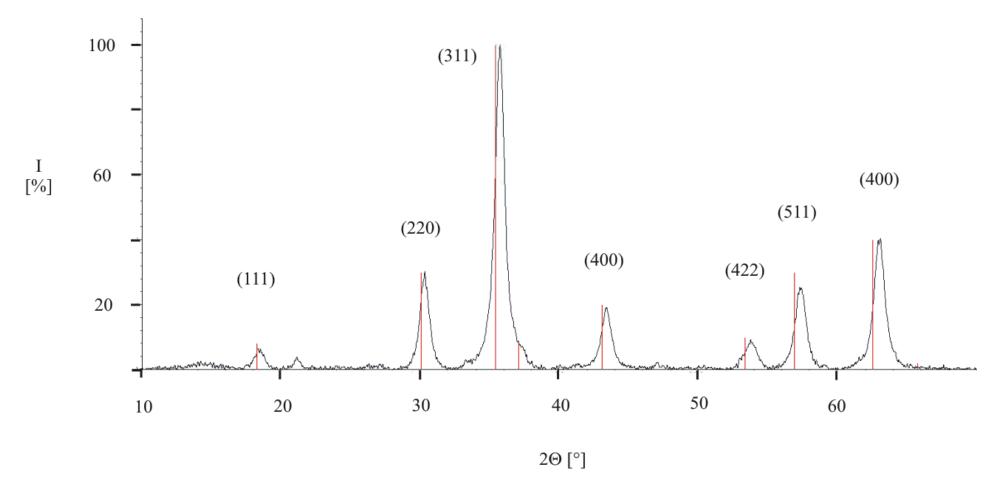


Figure S1: XRD of Fe₃O₄

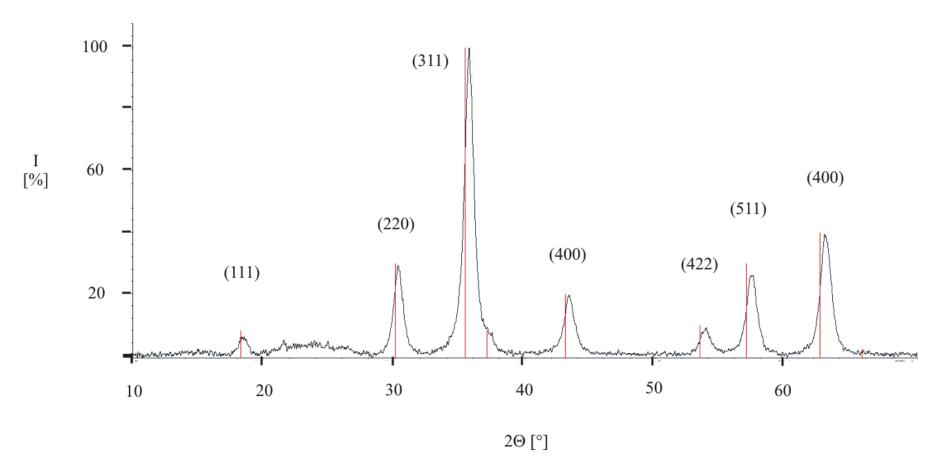


Figure S2: XRD of $Fe_3O_4@SiO_2$

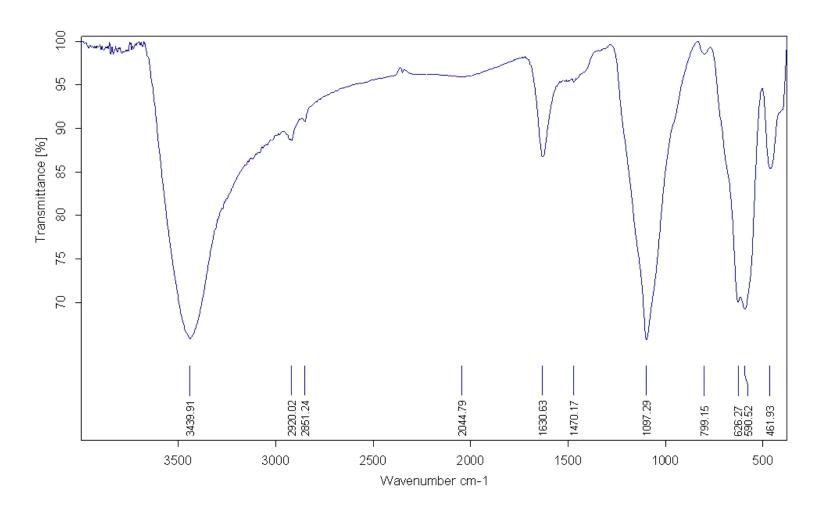


Figure S3: IR spectrum of aminofunctionalized Fe $_3O_4@SiO_2.$

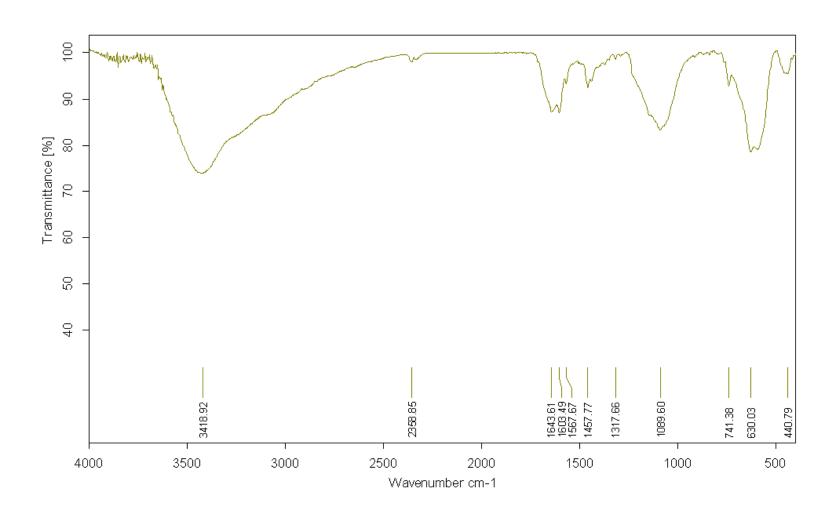


Figure S4: IR spectrum of Fe₃O₄-bzimpy.

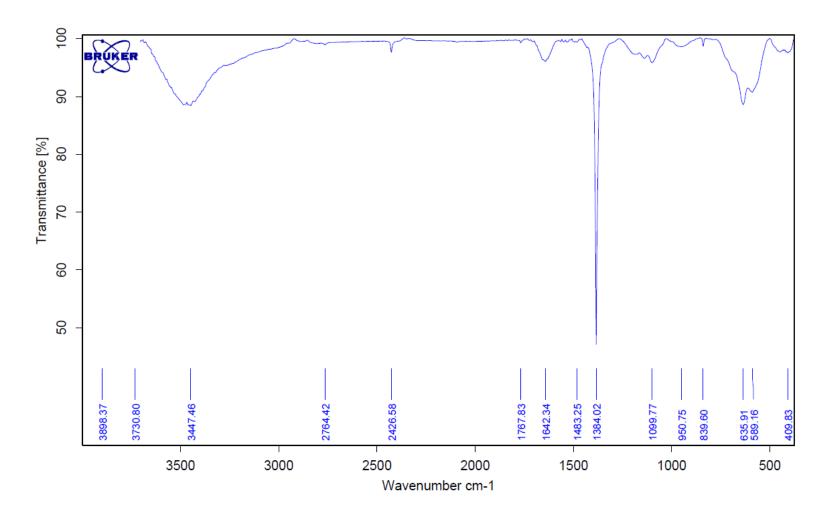


Figure S5: IR spectrum of Fe₃O₄-DOTA