

Supporting Information

for

Self-assembly of chiral fluorescent nanoparticles

based on water-soluble L-tryptophan derivatives of *p*-

***tert*-butylthiacalix[4]arene**

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Additional experimental parameters and results

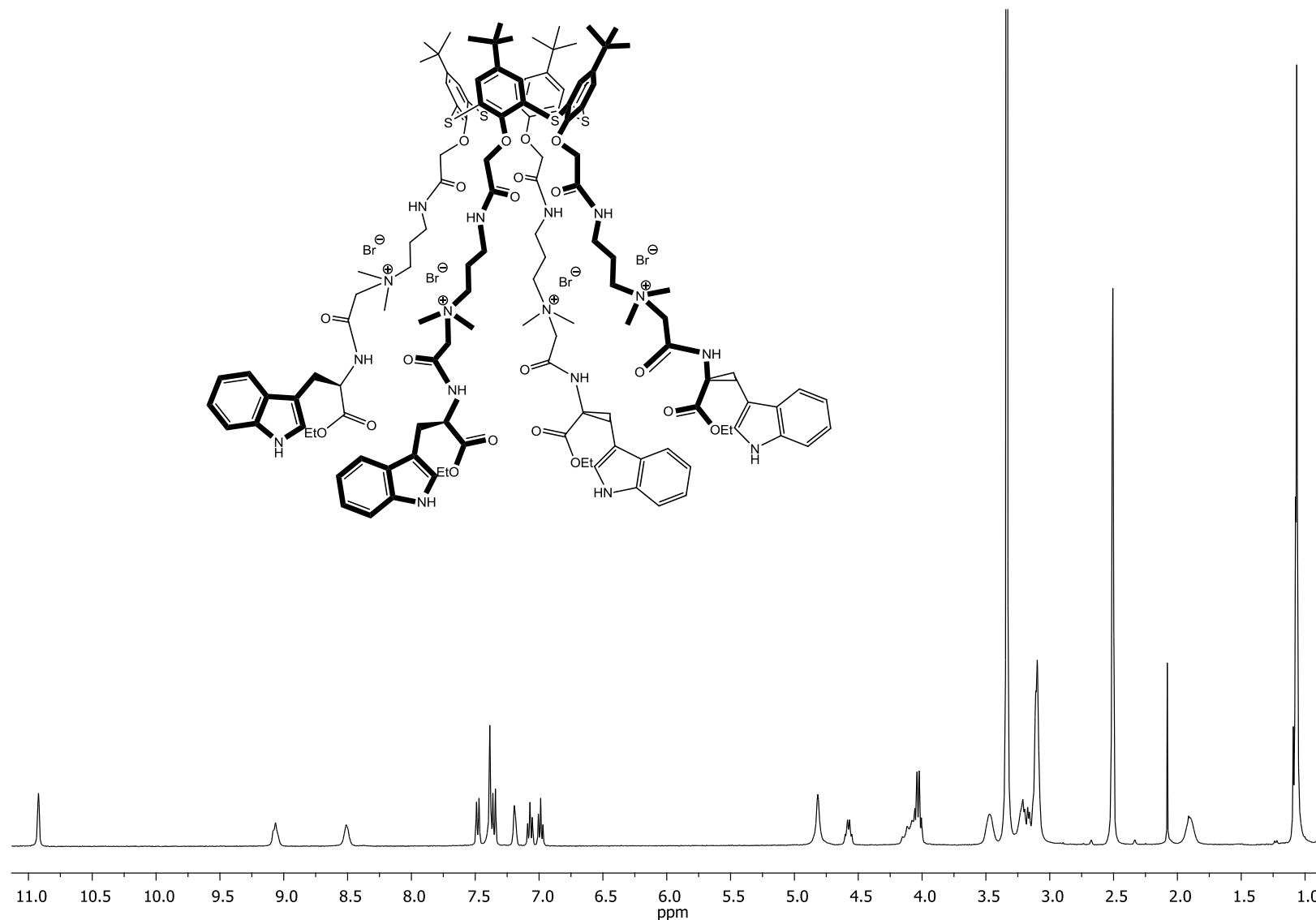


Figure S1: ^1H NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone-8*), $\text{DMSO-}d_6$, 298 K, 400 MHz

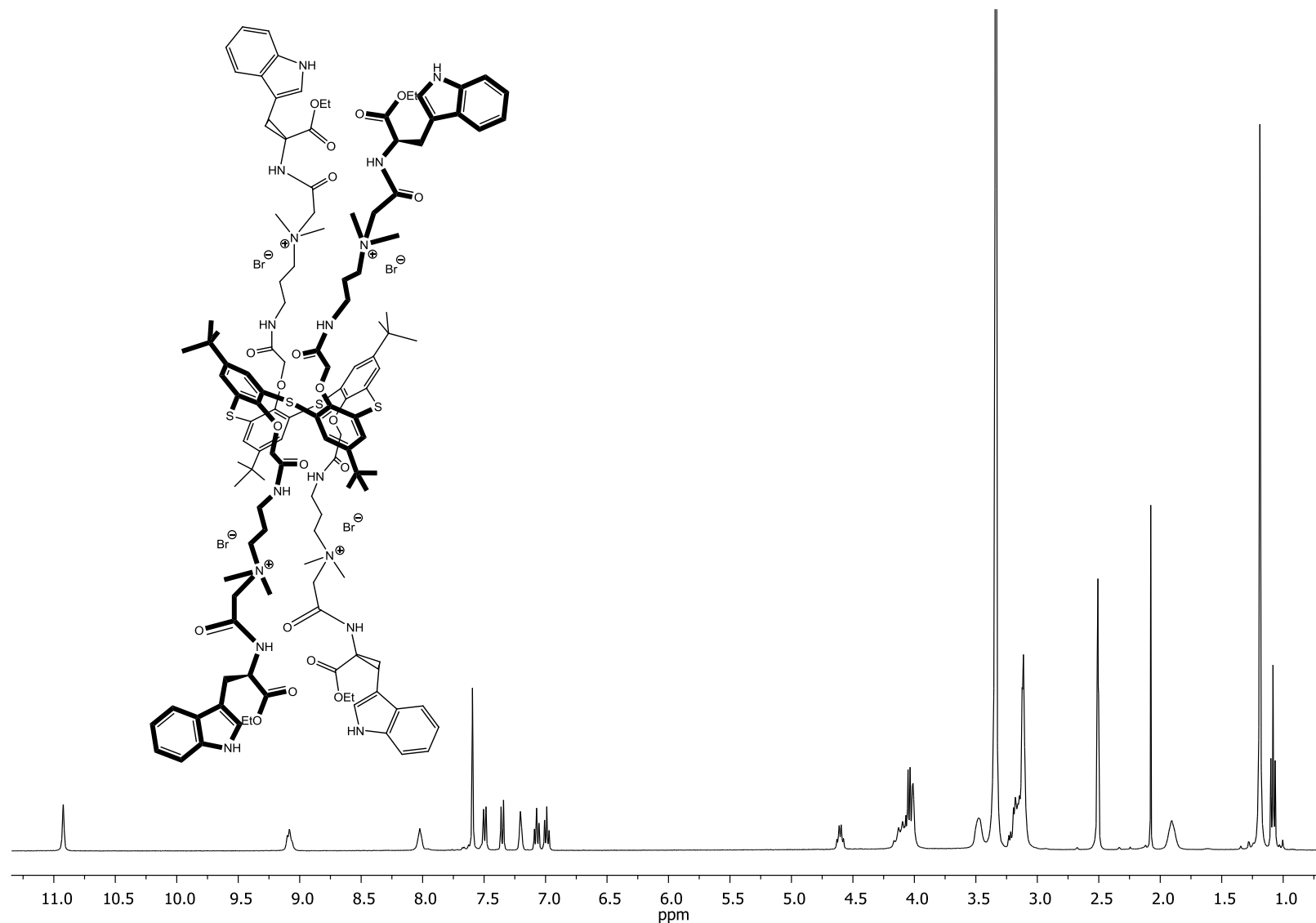


Figure S2: ^1H NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-**9**), DMSO- d_6 , 298 K, 400 MHz.

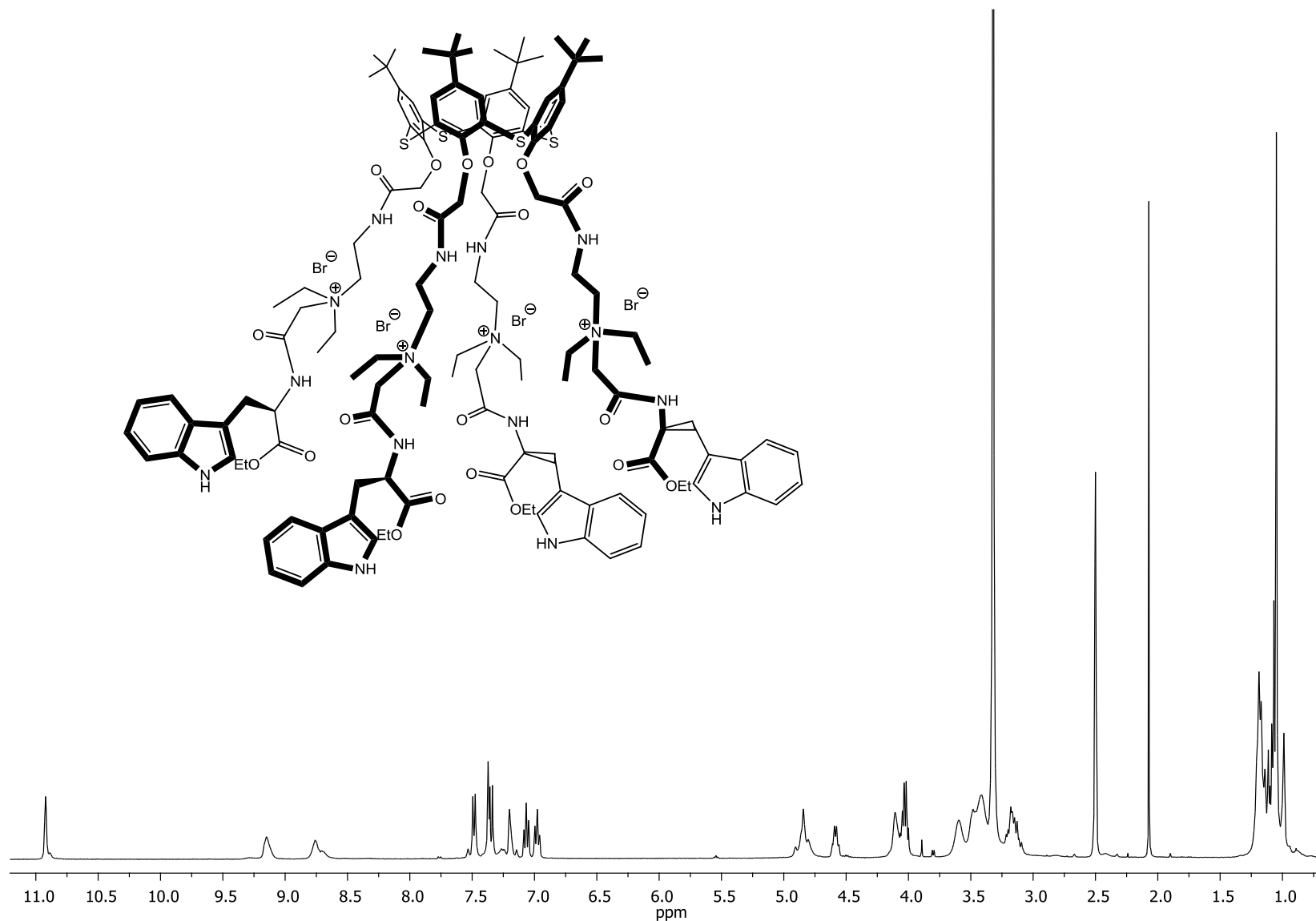


Figure S3: ^1H NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[(*N*-(2',2'-diethyl-2'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl}ammoniummethyl)carbamoylmethoxy)-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone*-10), DMSO- d_6 , 298 K, 400 MHz.

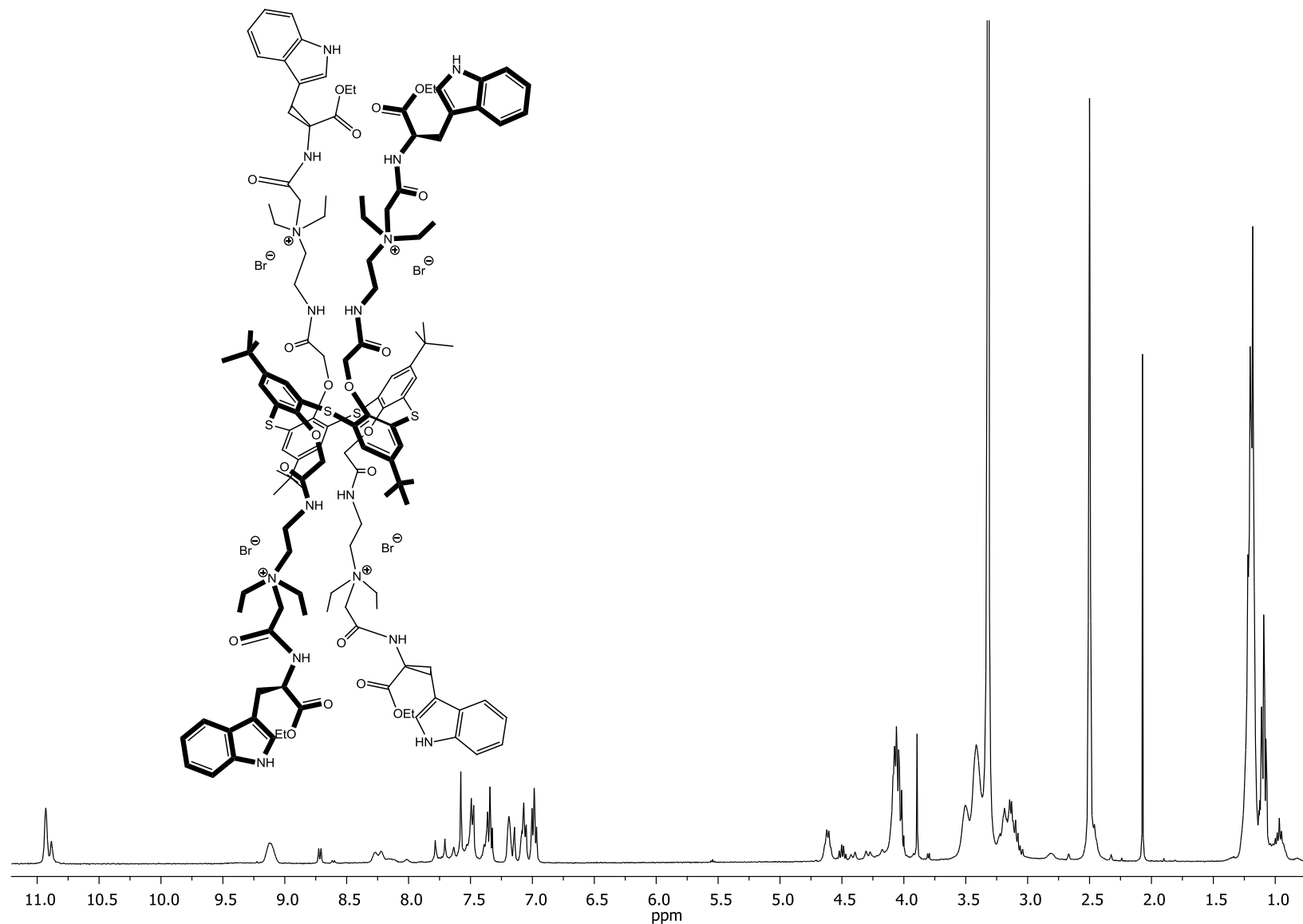


Figure S4: ^1H NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(2',2'-diethyl-2'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3'')-yl)methyl)methyl)amidocarbonylmethyl)ammoniummethyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-**11**), DMSO- d_6 , 298 K, 400 MHz.

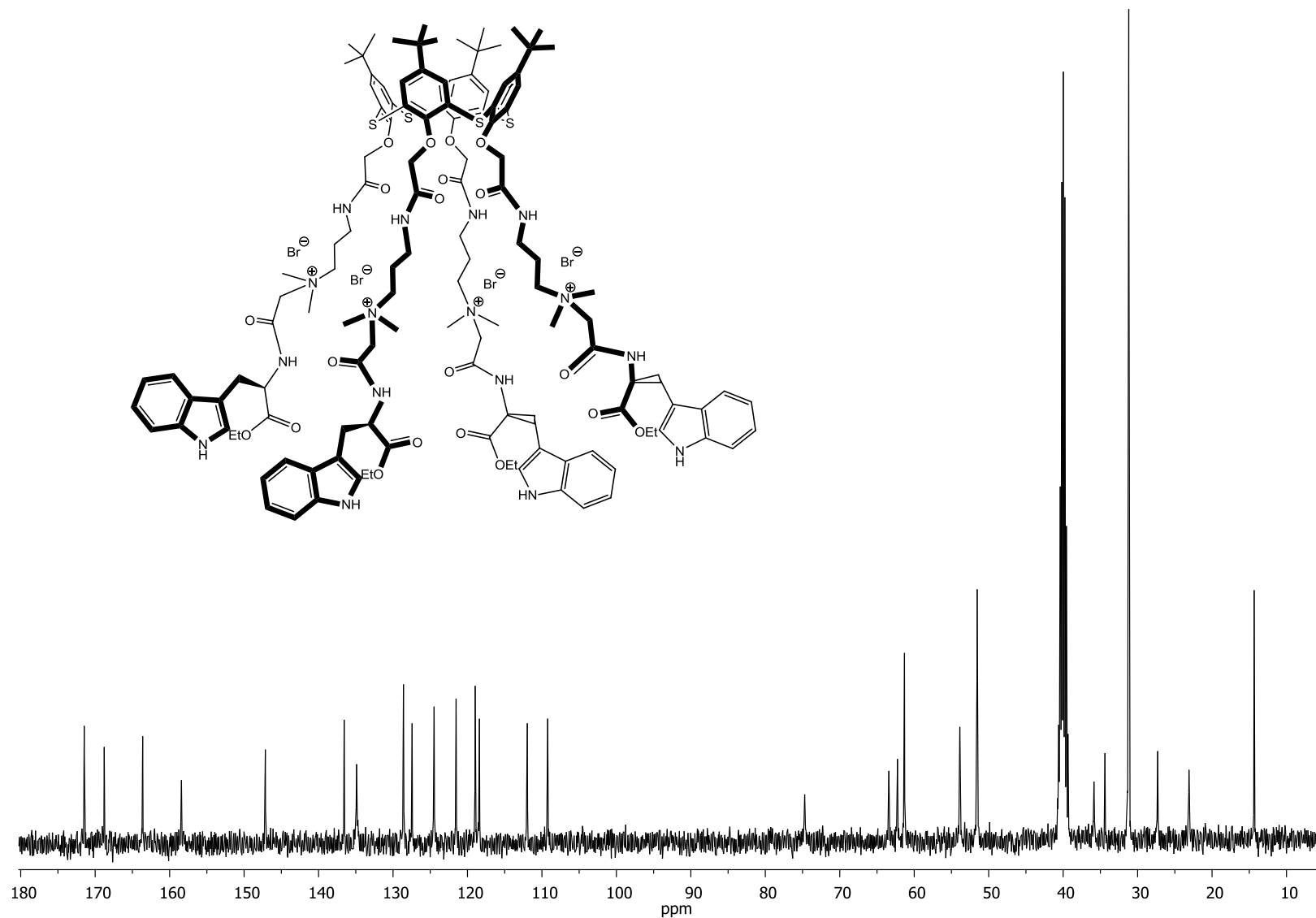


Figure S5: ^{13}C NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl}ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone-8*), $\text{DMSO-}d_6$, 298 K, 100 MHz.

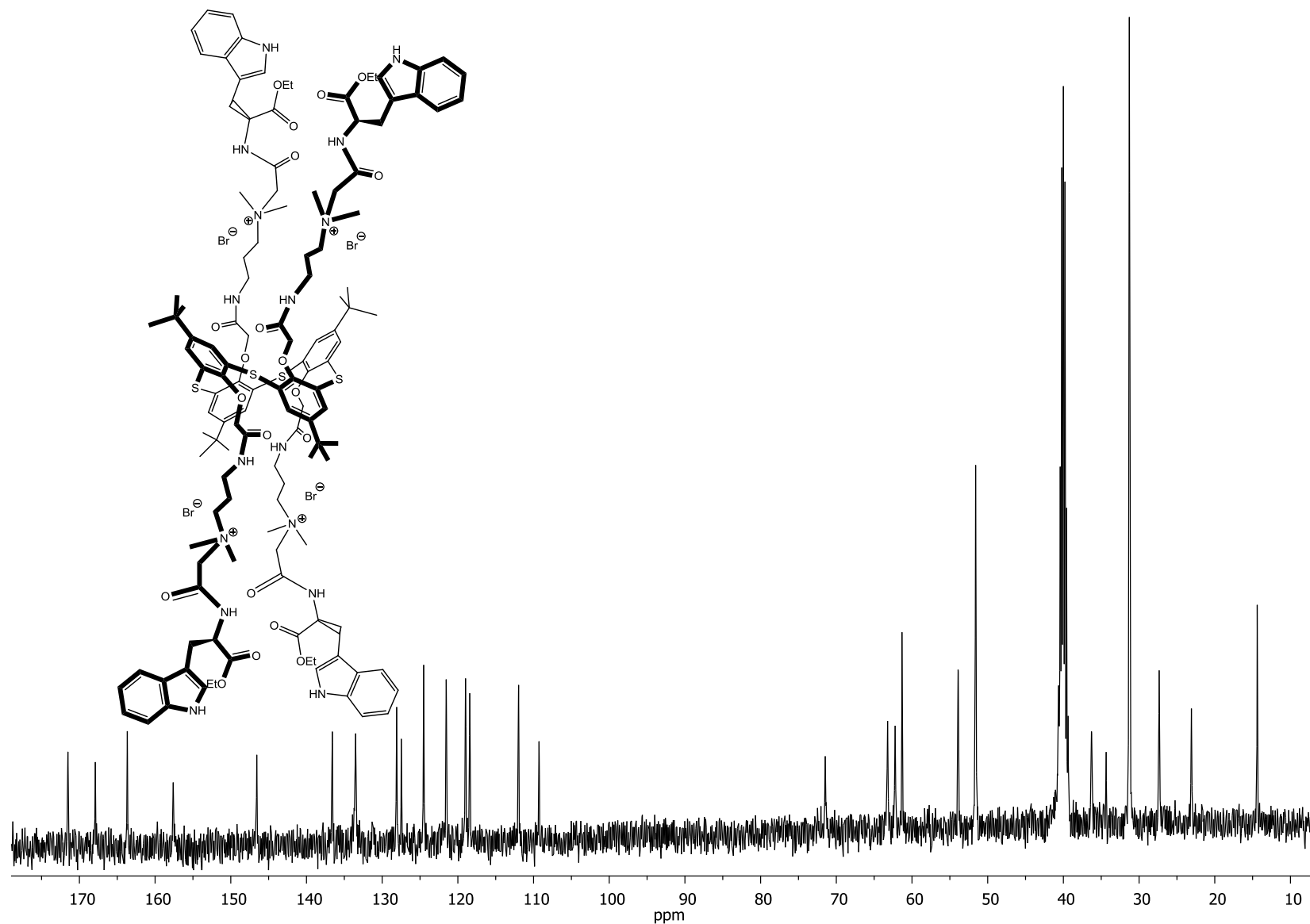


Figure S6: ^{13}C NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-9), DMSO- d_6 , 298 K, 100 MHz.

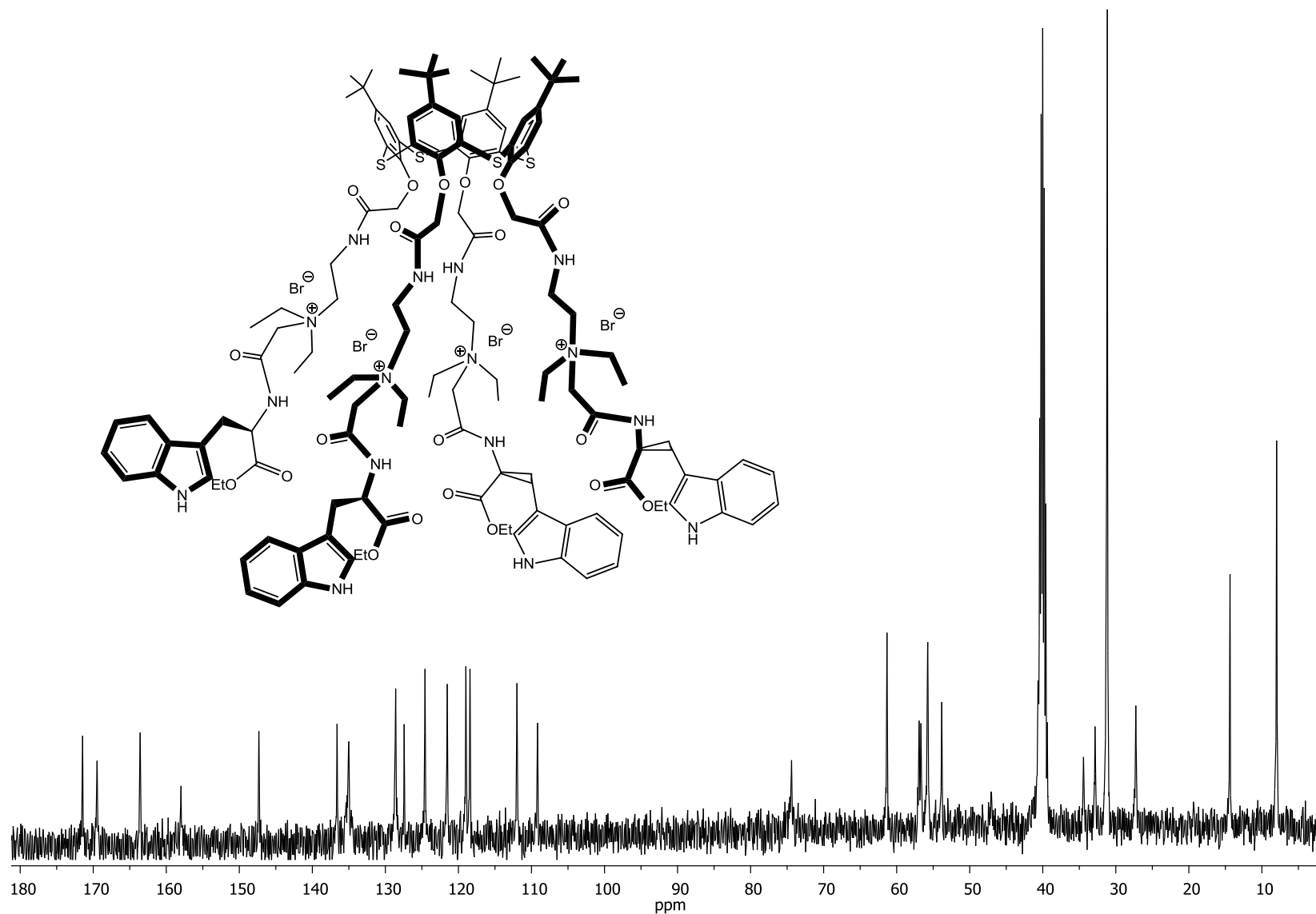


Figure S7: ^{13}C NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(2',2'-diethyl-2'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniummethyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone-10*), DMSO- d_6 , 298 K, 100 MHz.

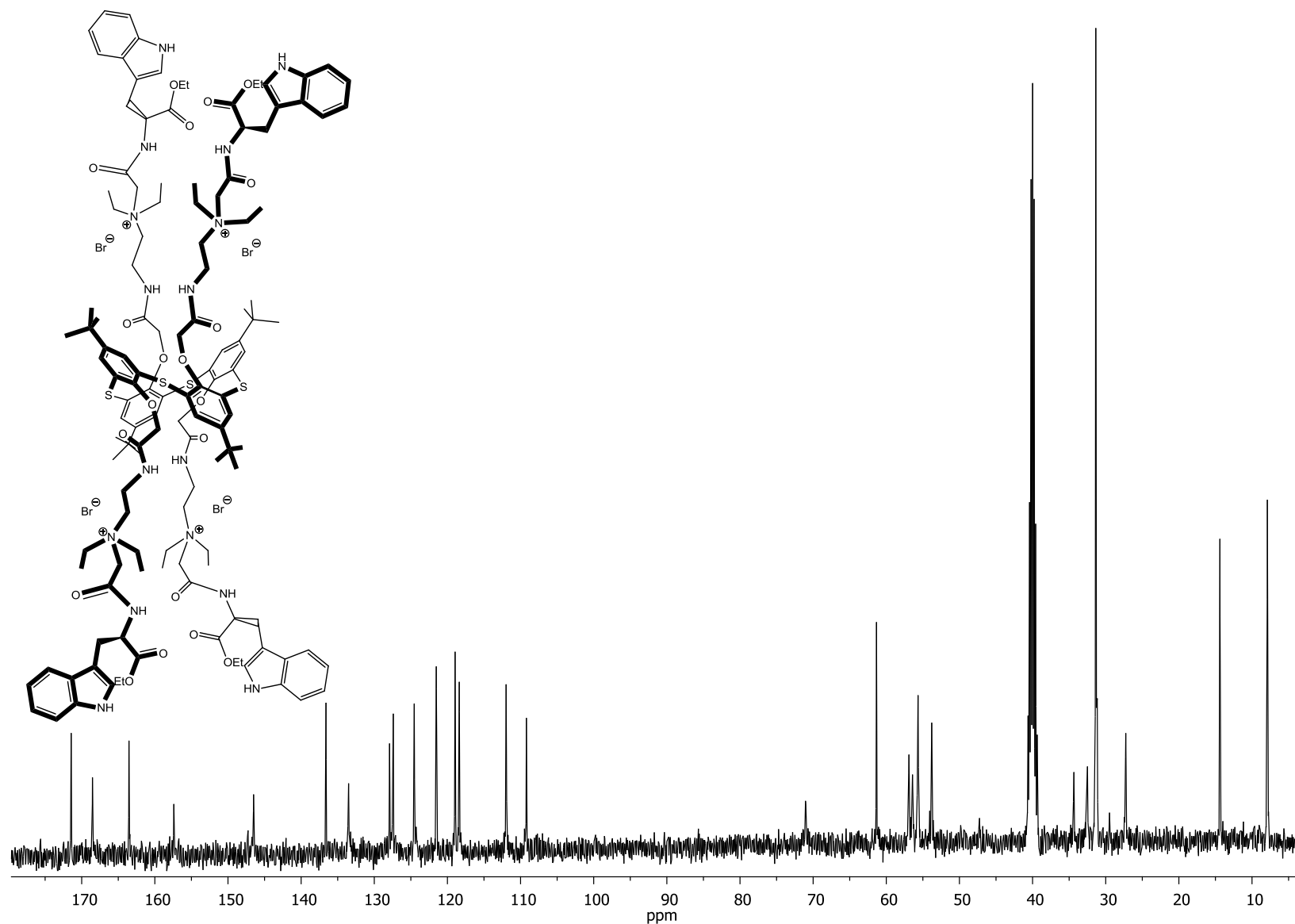


Figure S8: ^{13}C NMR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(2',2'-diethyl-2'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl)methyl)amidocarbonylmethyl}ammoniummethyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-**11**), DMSO- d_6 , 298 K, 100 MHz.

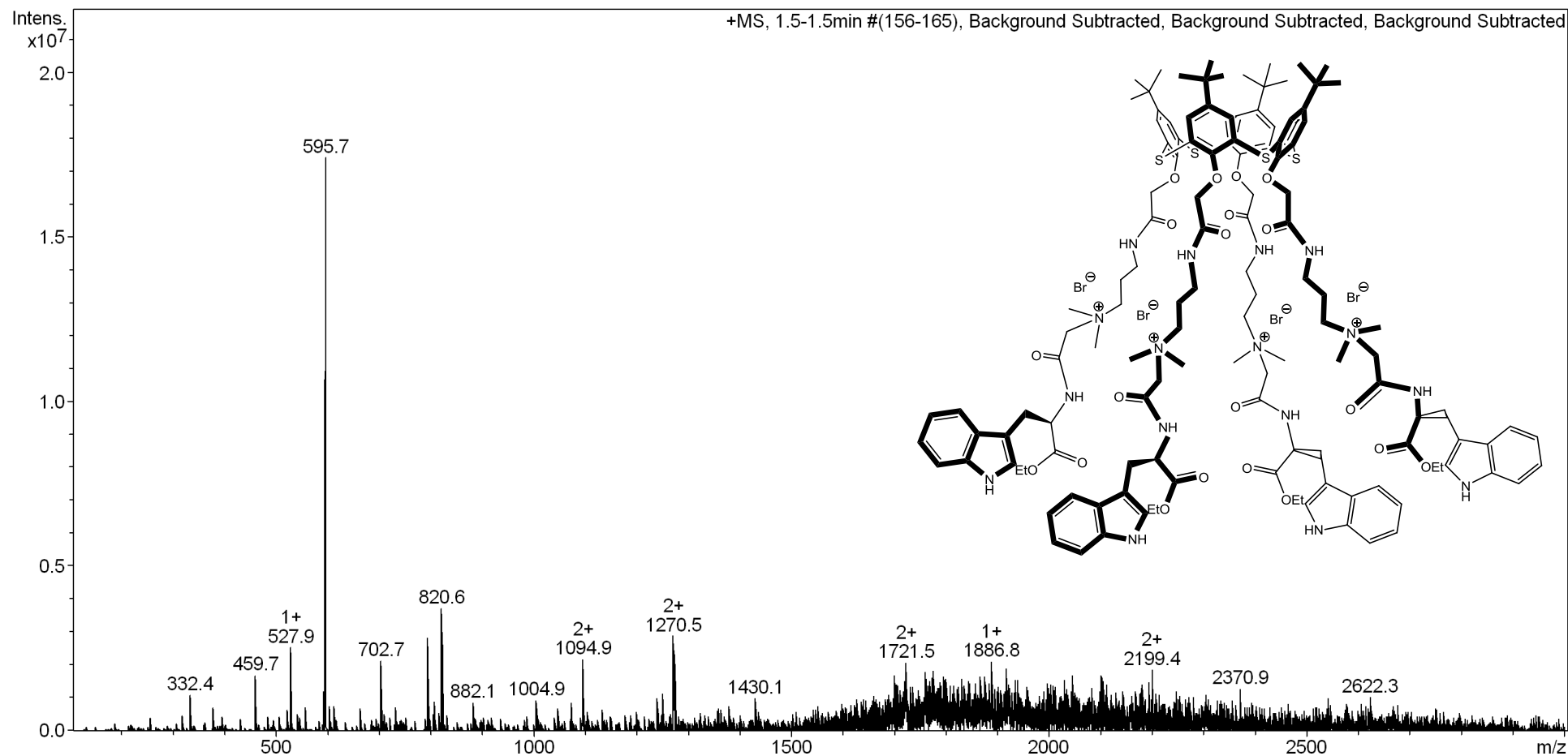


Figure S9: Mass spectrum (ESI) of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone-8*).

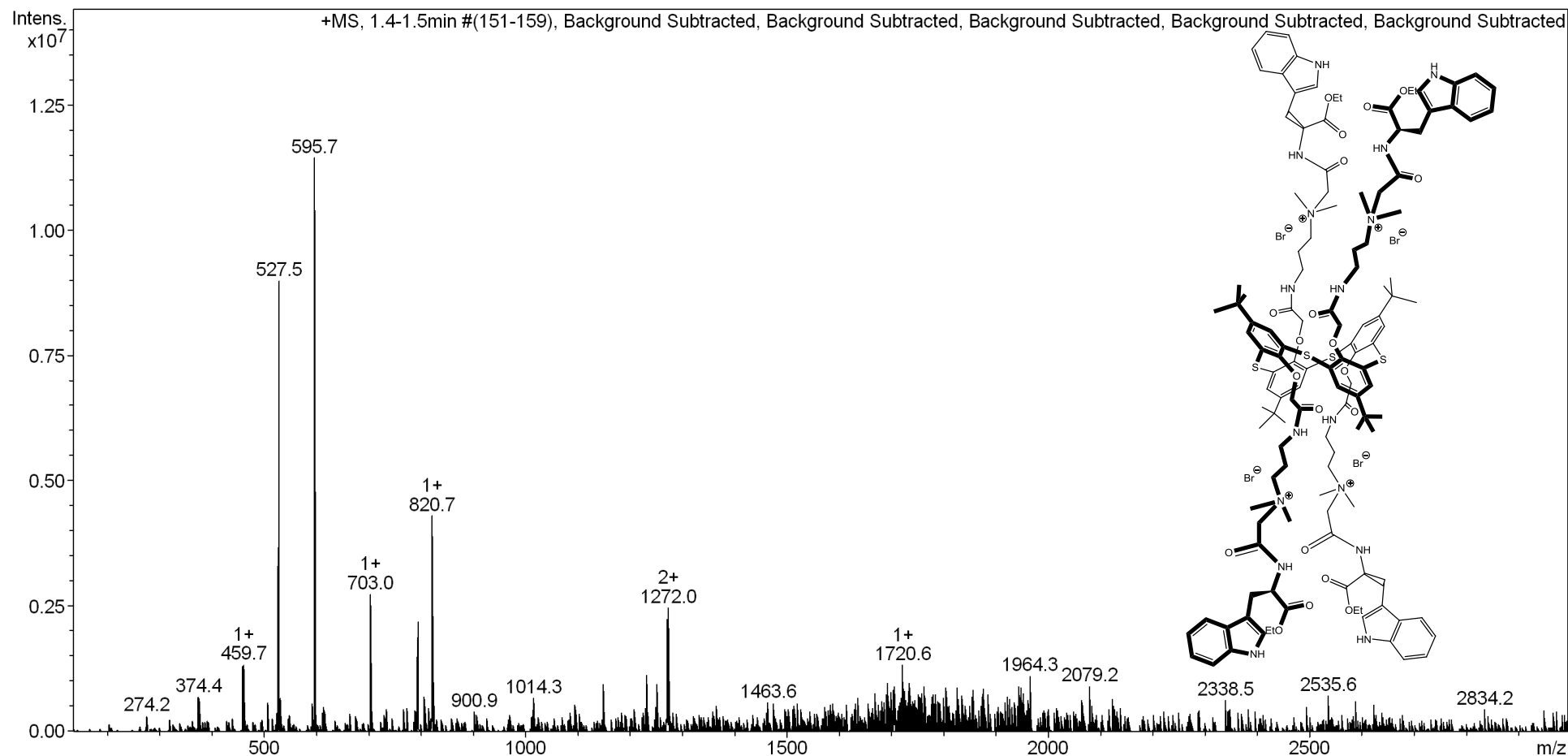


Figure S10: Mass spectrum (ESI) of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-9).

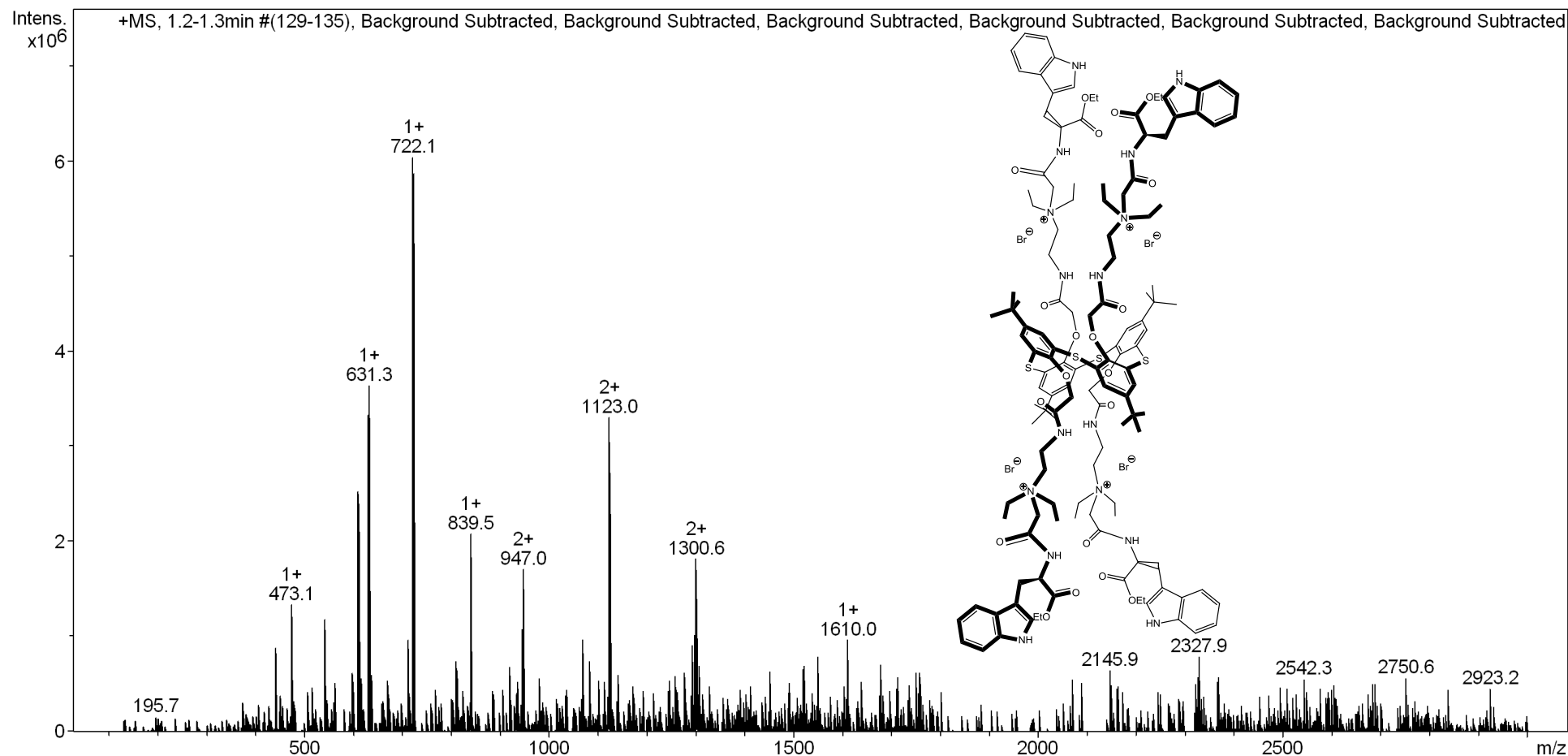


Figure S12: Mass spectrum (ESI) of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(2',2'-diethyl-2'-{(ethoxycarbonyl[S-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniummethyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-**11**).

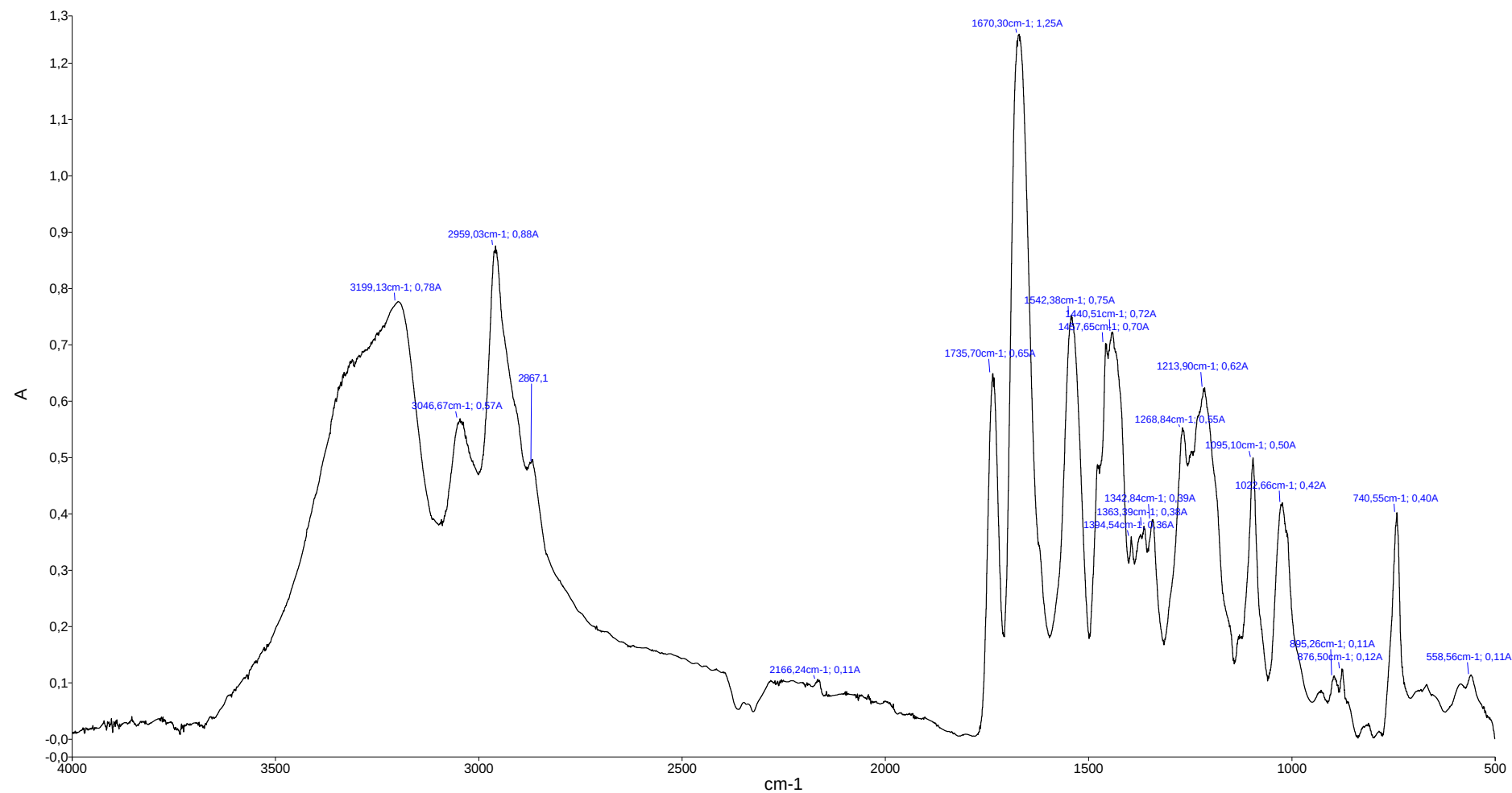


Figure S13: IR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[(*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone-8*).

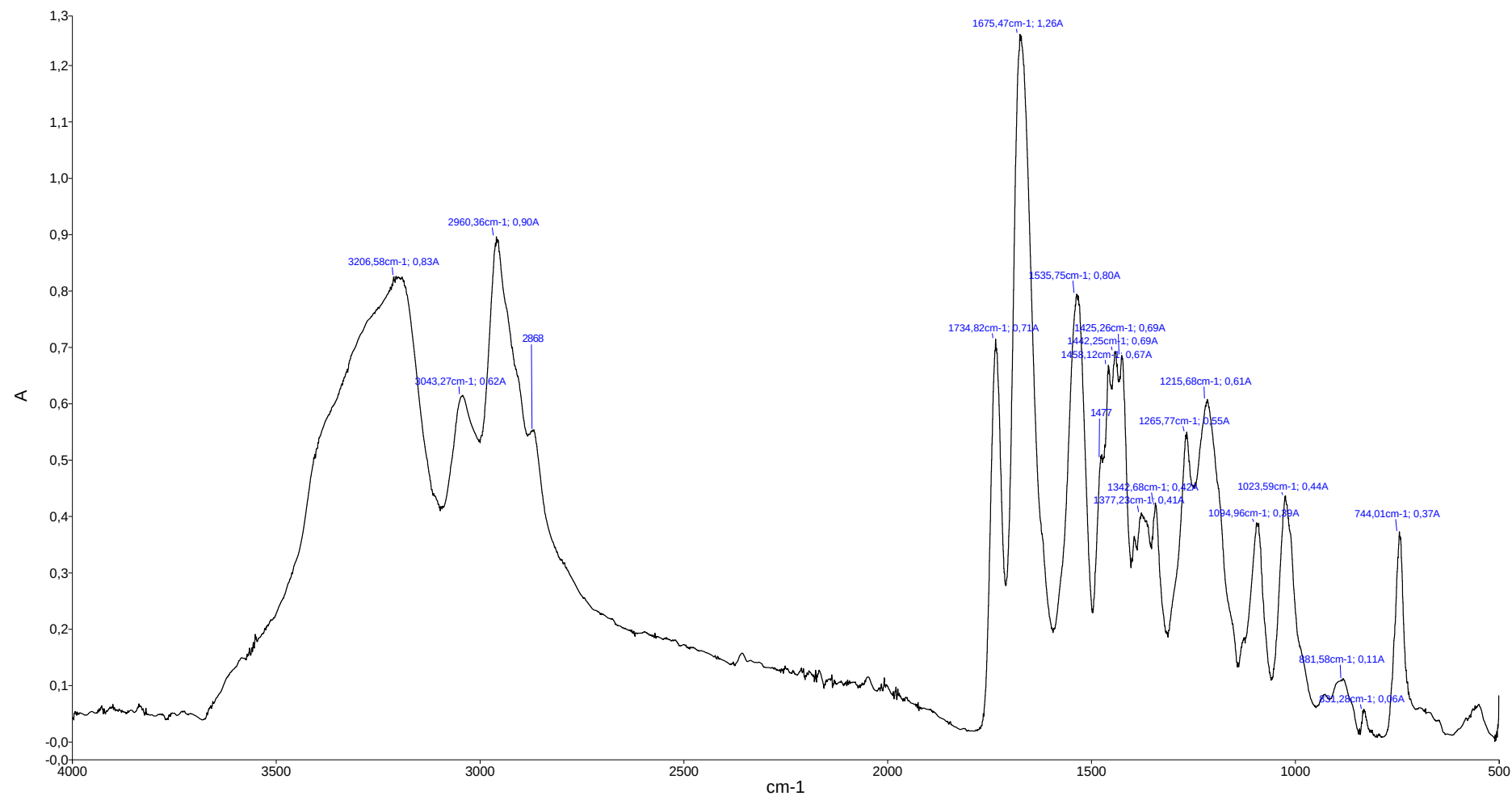


Figure S14: IR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[(*N*-(3',3'-dimethyl-3'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniumpropyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-9).

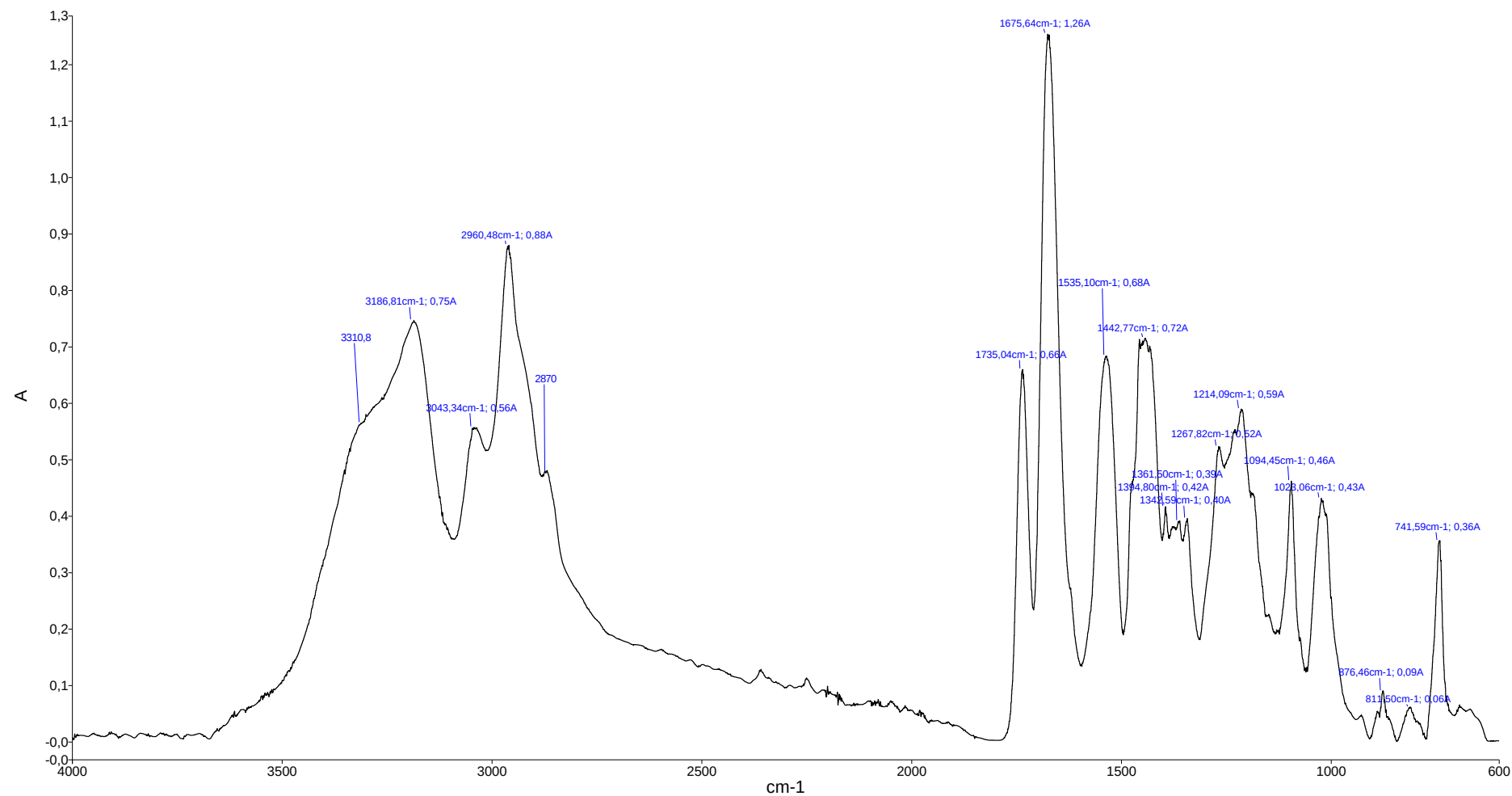


Figure S15: IR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[*N*-(2',2'-diethyl-2'-{(ethoxycarbonyl[*S*-(1''*H*-indol-3''-yl)methyl]methyl)amidocarbonylmethyl)ammoniummethyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (*cone*-**10**).

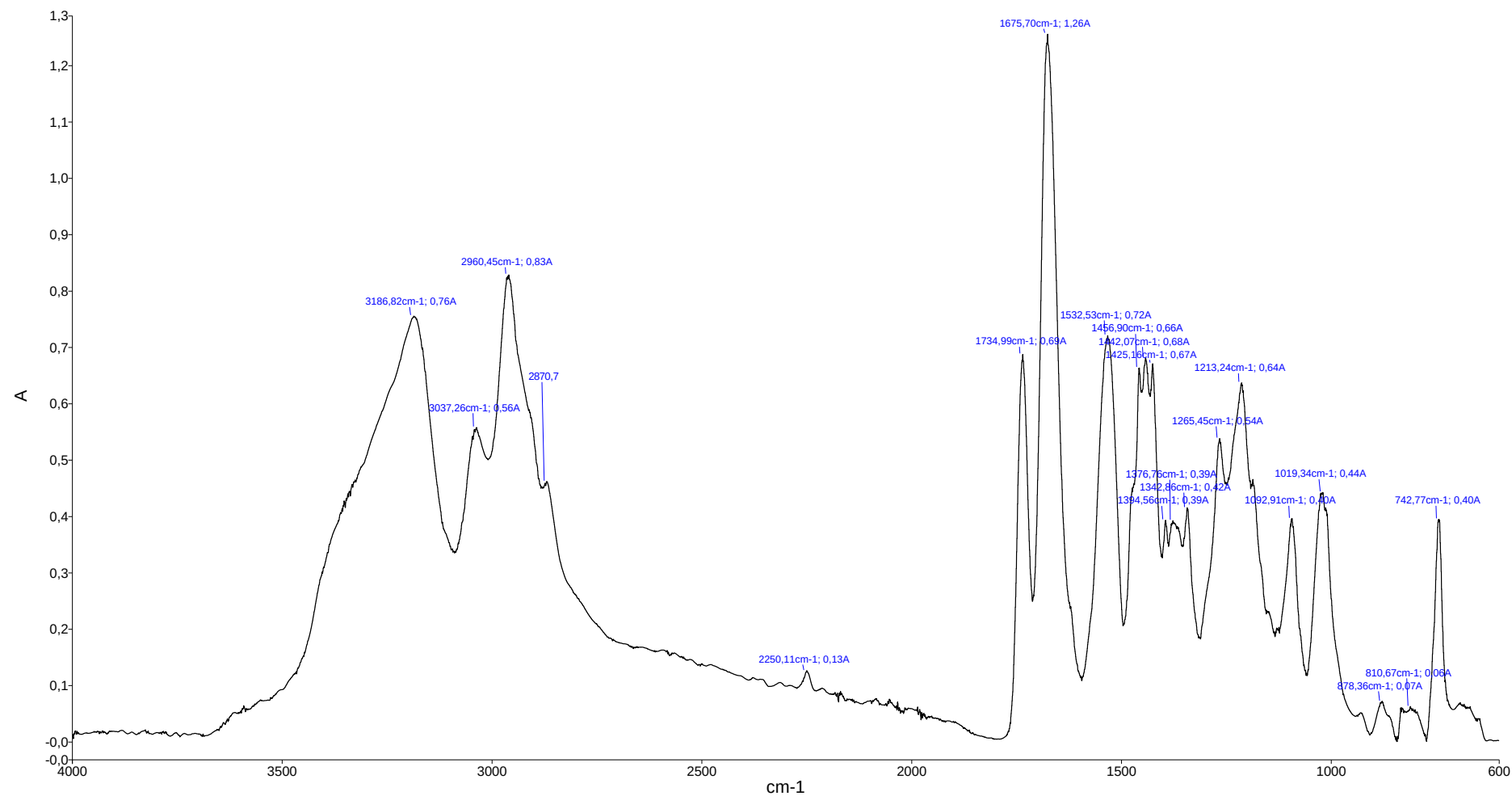


Figure S16: IR spectrum of 5,11,17,23-tetra-*tert*-butyl-25,26,27,28-tetrakis[(*N*-(2',2'-diethyl-2'-(ethoxycarbonyl[S-(1''*H*-indol-3''-yl)methyl)methyl]amidocarbonylmethyl)ammoniummethyl)carbamoylmethoxy]-2,8,14,20-tetra thiacalix[4]arene tetrabromide (1,3-*alternate*-**11**).

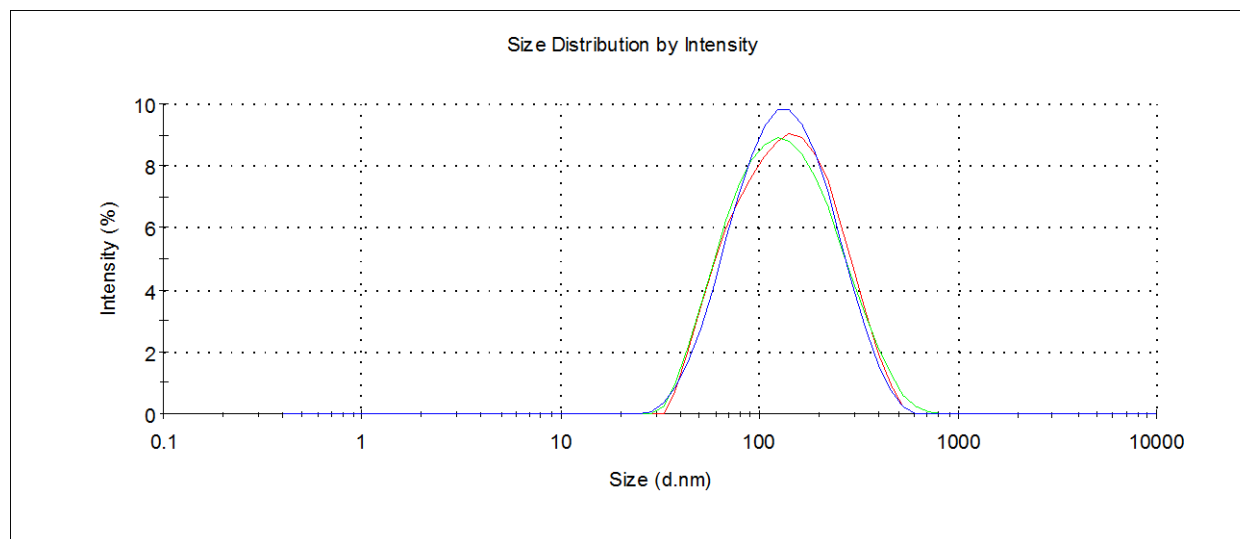


Figure S17: Size distribution by intensity of solution of the macrocycle **8** in the water (1×10^{-4} M).

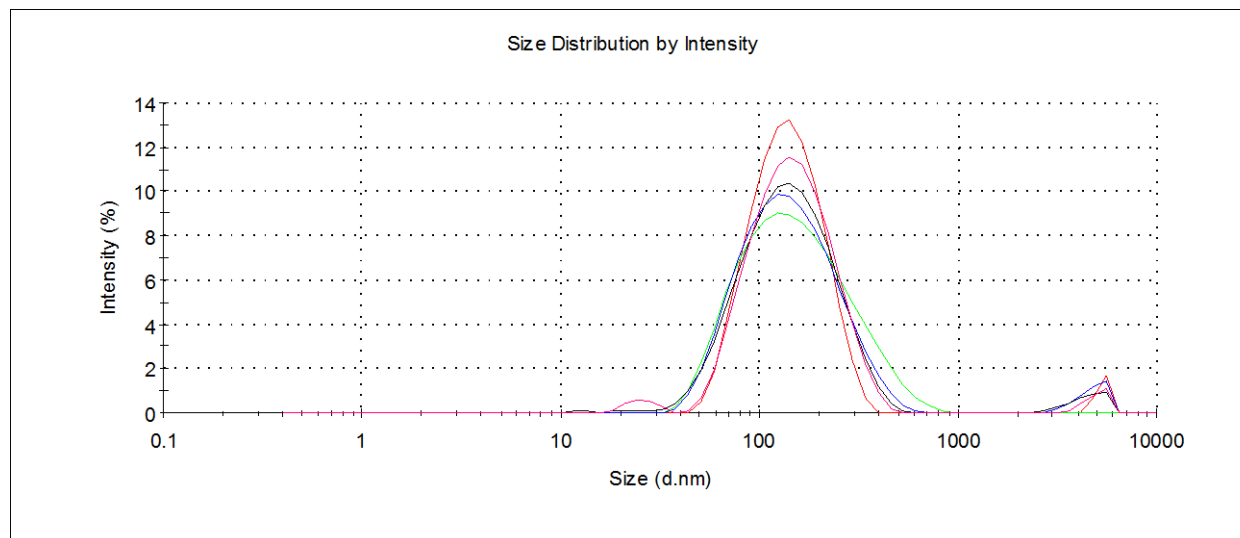


Figure S18: Size distribution by intensity of solution of the macrocycle **9** in the water (1×10^{-4} M).

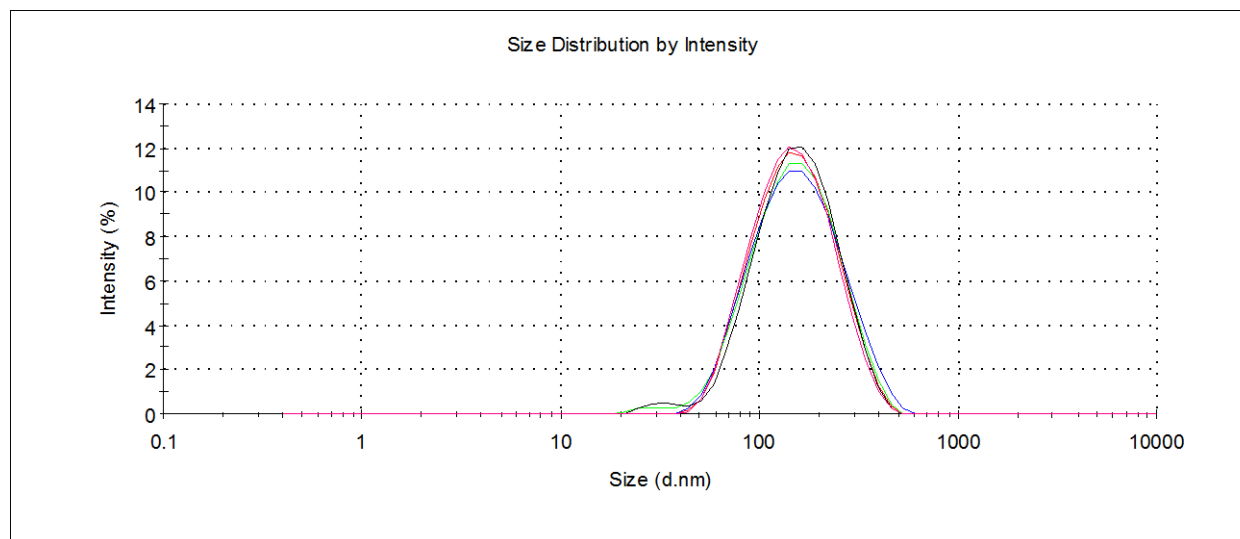


Figure S19: Size distribution by intensity of solution of the macrocycle **10** in the water (1×10^{-4} M).

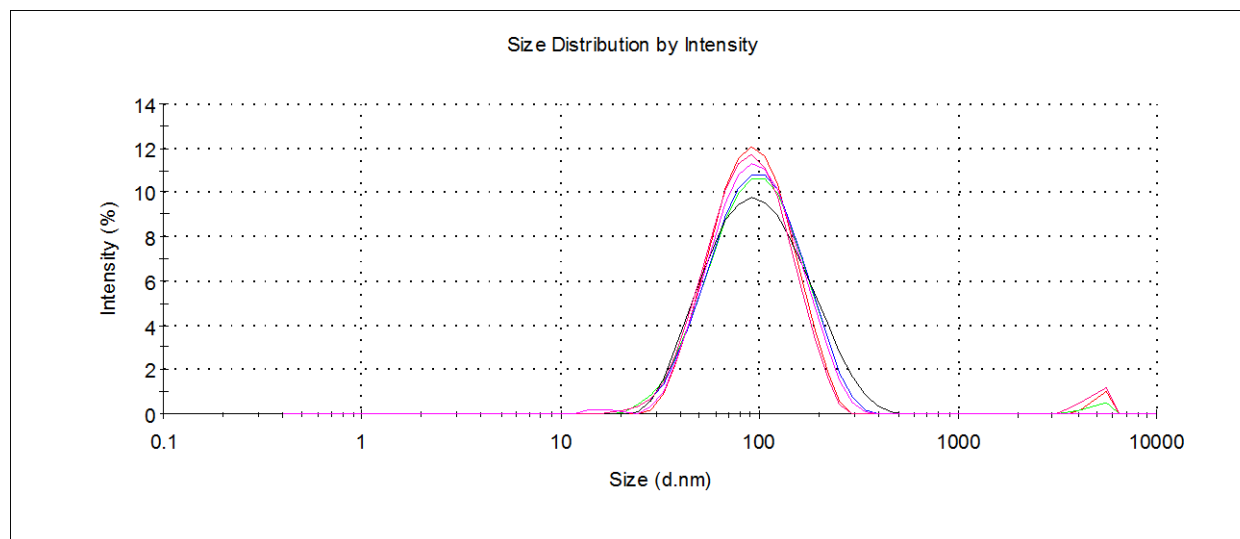


Figure S20: Size distribution by intensity of solution of the macrocycle **11** in the water (1×10^{-4} M).

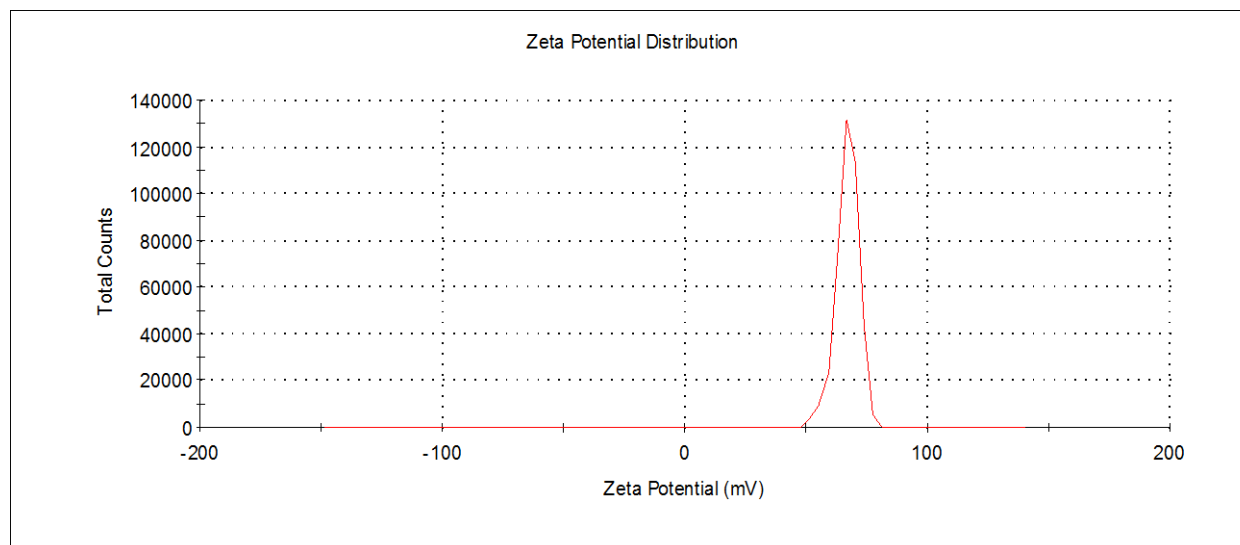


Figure S21: Zeta potential of solution of the macrocycle **8** in the water (1×10^{-4} M).

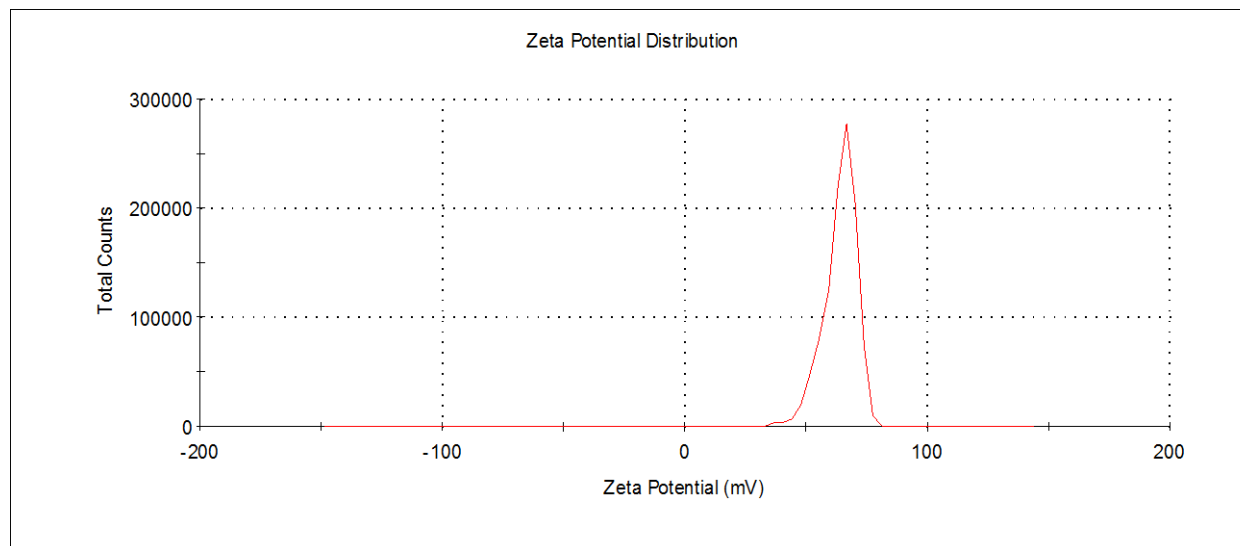


Figure S22: Zeta potential of solution of the macrocycle **9** in the water (1×10^{-4} M).

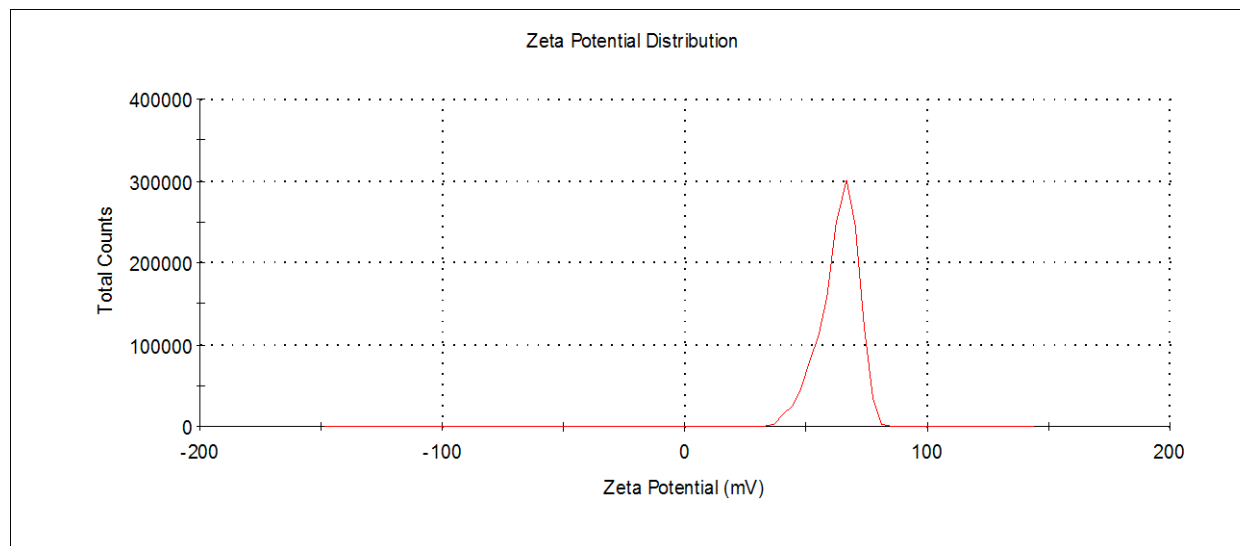


Figure S23: Zeta potential of solution of the macrocycle **10** in the water (1×10^{-4} M).

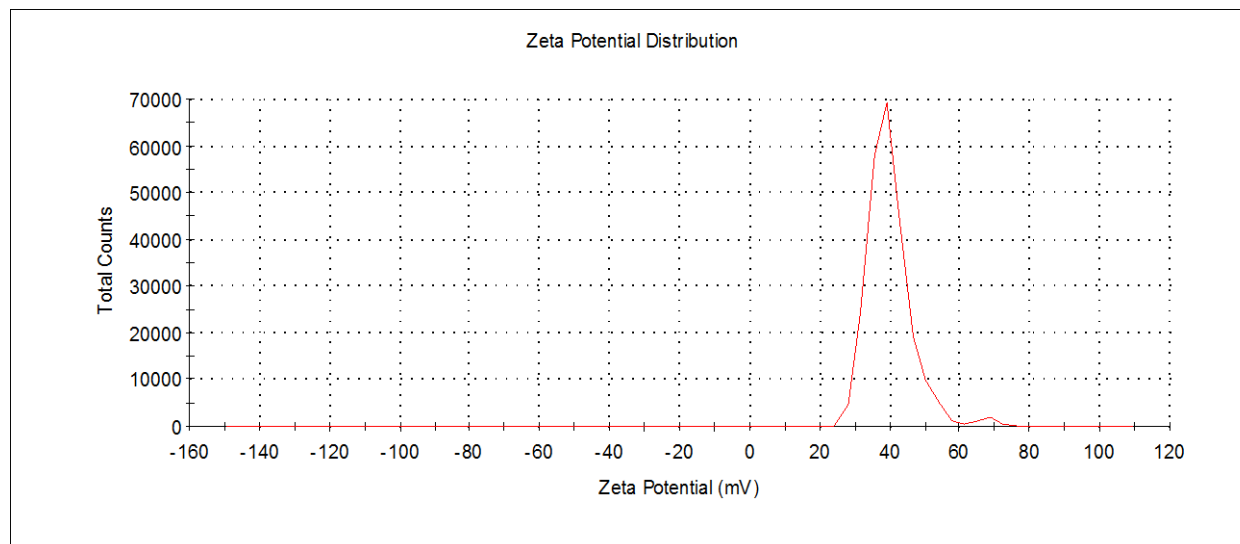


Figure S24: Zeta potential of solution of the macrocycle **11** in the water (1×10^{-4} M).

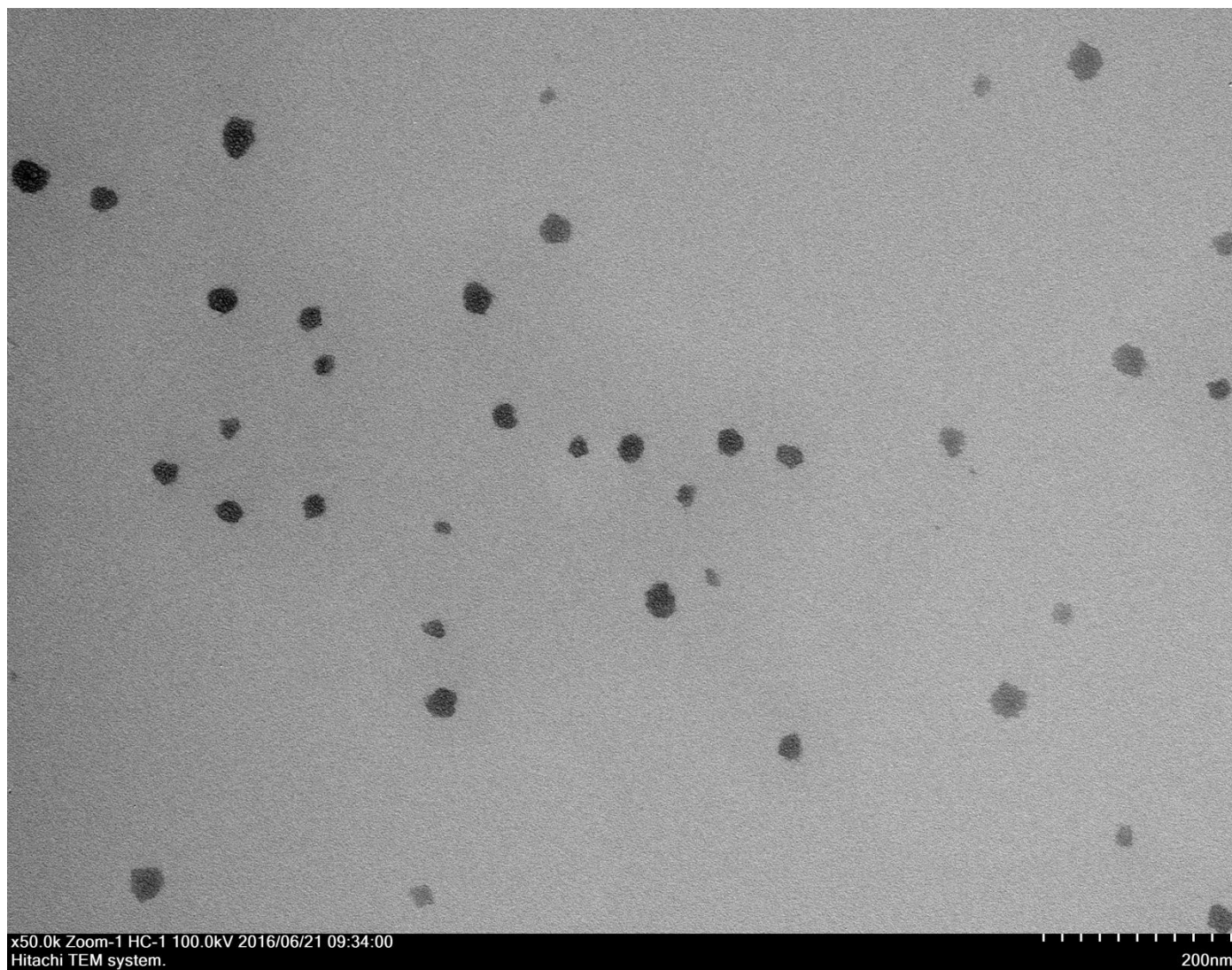


Figure S25: TEM image of self-associates based on thiacalix[4]arene *cone-8* in water (1×10^{-4} M). Scale bar 200 nm.

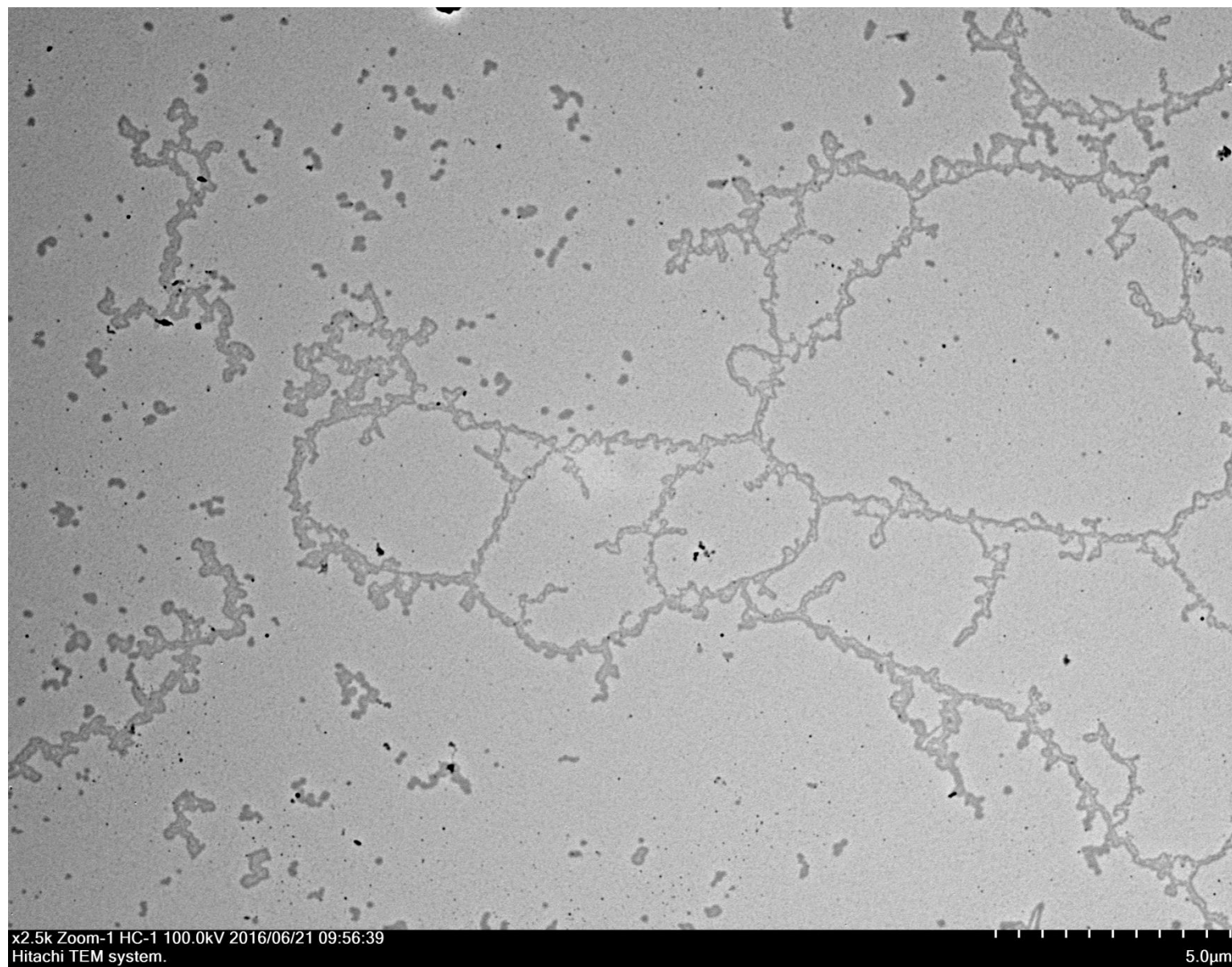


Figure S26: TEM image of self-associates based on thiacalix[4]arene 1,3-*alternate*-9 in water (1×10^{-4} M). Scale bar 5 μ m.

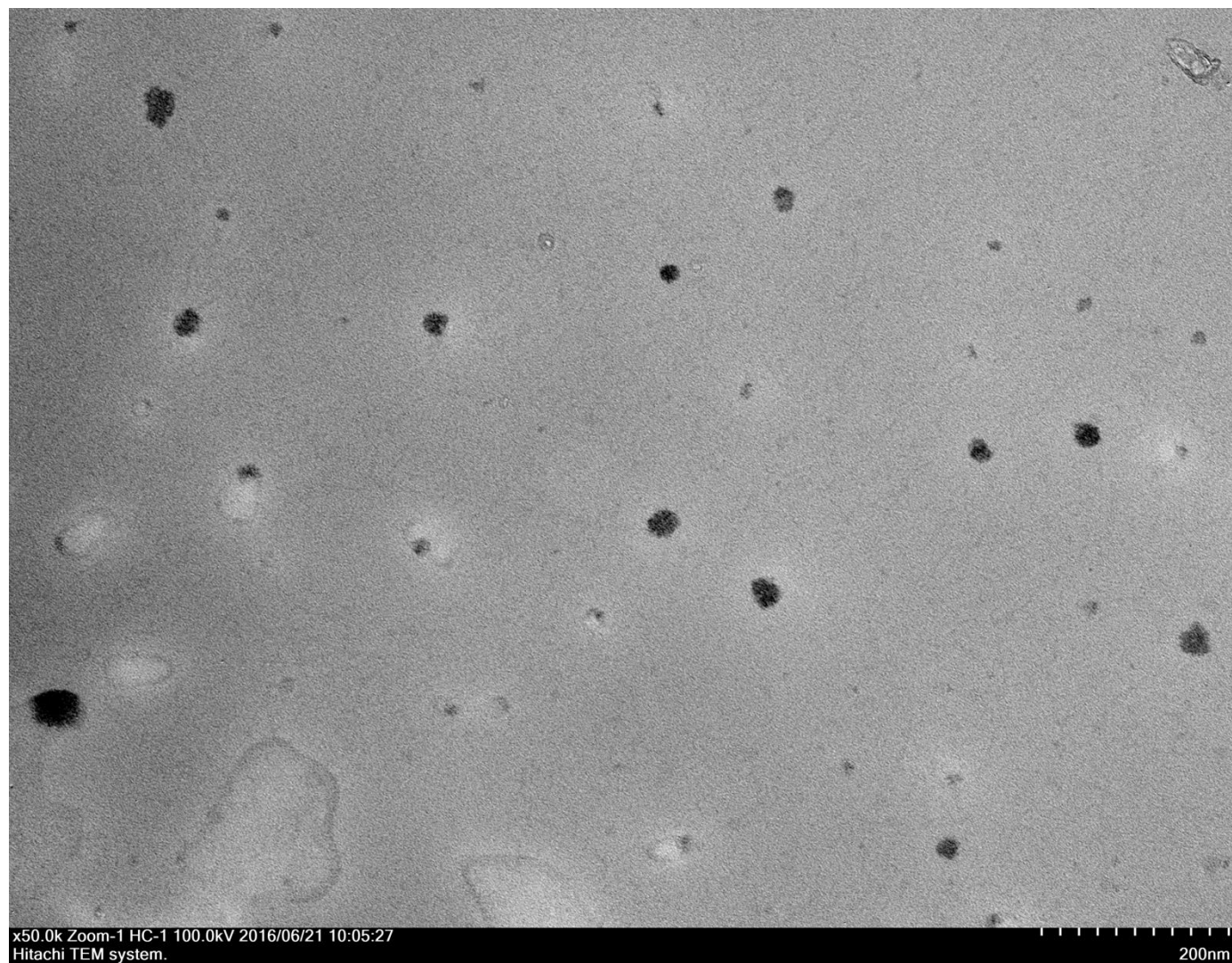


Figure S27: TEM image of self-associates based on thiacalix[4]arene *cone-10* in water (1×10^{-4} M). Scale bar 200 nm.

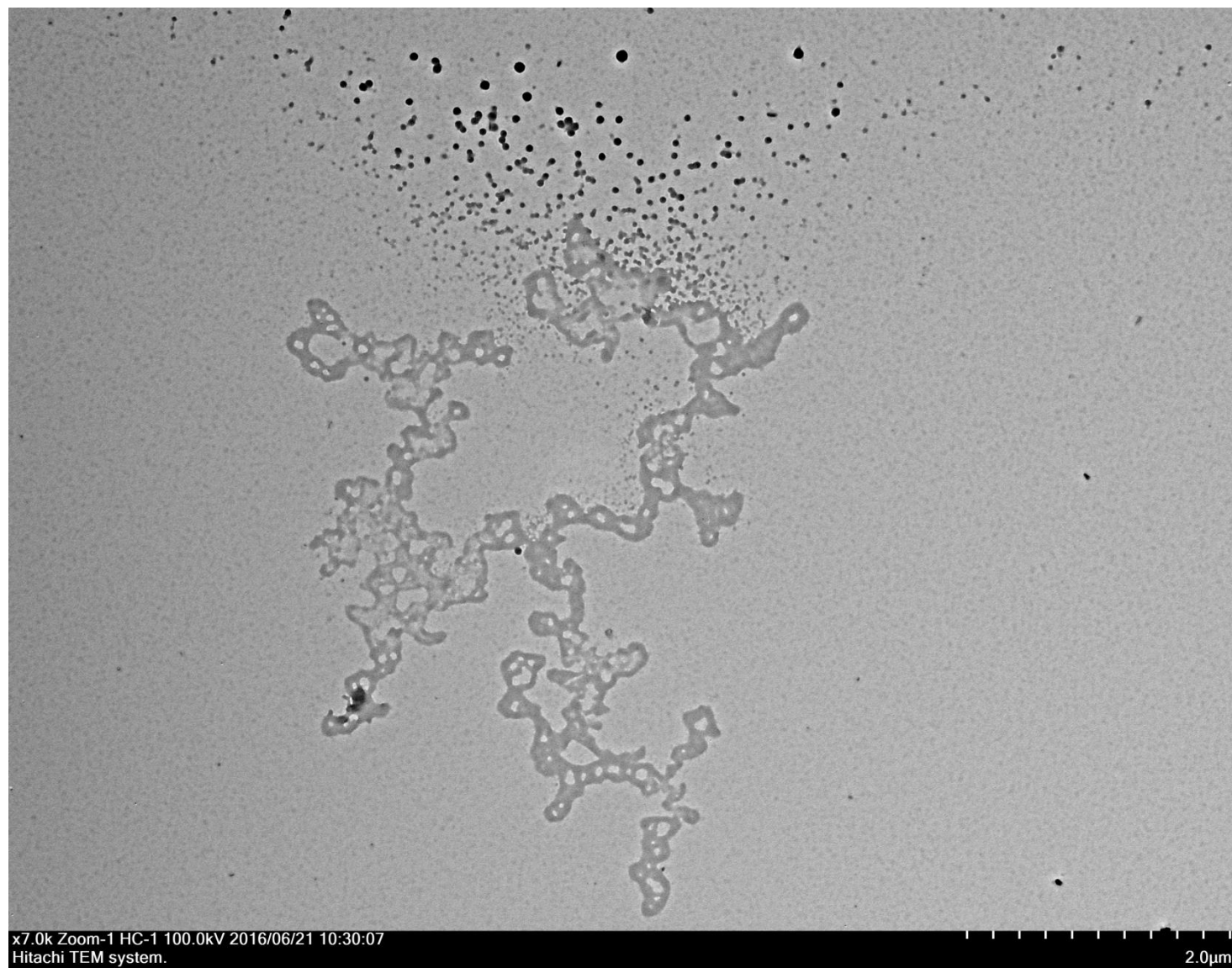


Figure S28: TEM image of self-associates based on thiacalix[4]arene 1,3-*alternate*-**11** in water (1×10^{-4} M). Scale bar 2 μm .

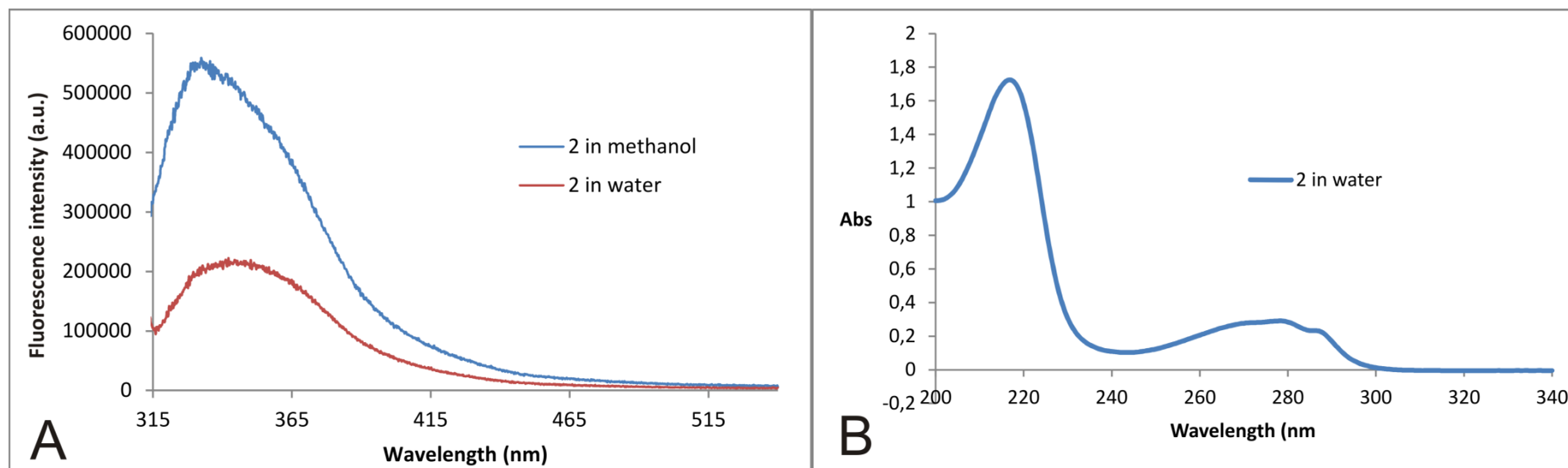


Figure S29: A) Fluorescence spectra of compound **2** in water and methanol (1×10^{-5} M), B) UV spectrum of compound **2** in water (1×10^{-5} M).

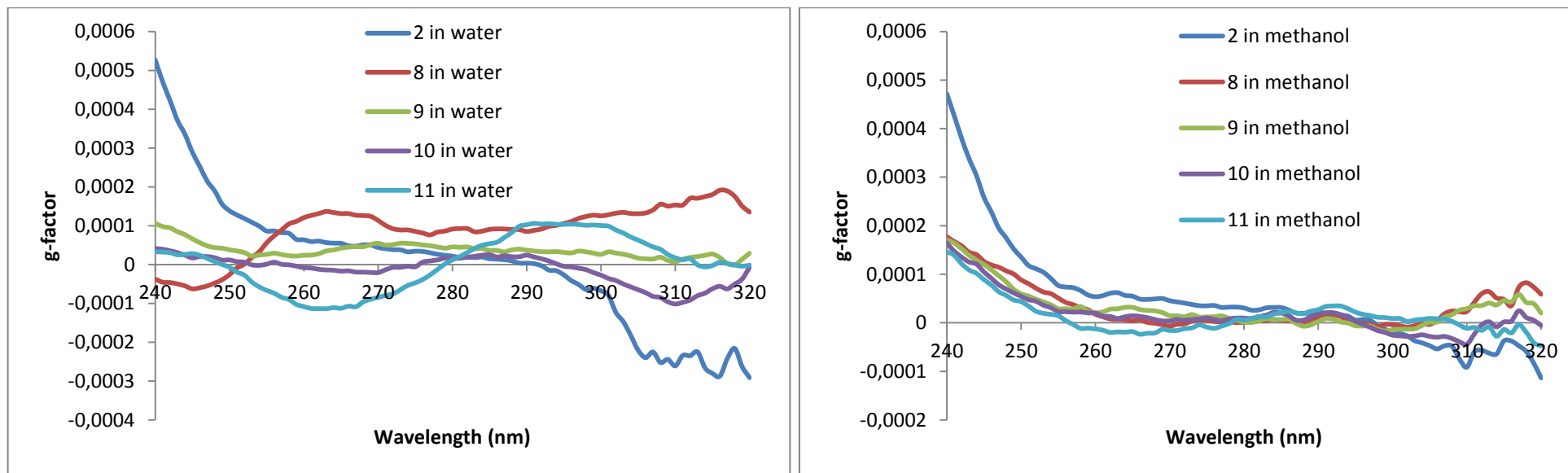


Figure S30: g -Factors of compounds **2** and **8–11** in water and methanol (1×10^{-5} M).

The optical activity of chiral systems is quantified using the anisotropy factor (g -factor) [1,2]:

$$g = \Delta\epsilon / \epsilon$$

where $\Delta\epsilon$ and ϵ are the molar circular dichroism and molar extinction coefficient, respectively.

References

1. Yan, W.; Xu, L.; Xu, C.; Ma, W.; Kuang, H.; Wang, L.; Kotov, N. A. *J. Am. Chem. Soc.*, **2012**, *134*, 15114–15121.
2. Cheng, J.; Le Saux, G.; Gao, J.; Buffeteau, T.; Battie, Y.; Barois, P.; Ponsinet, V.; Delville, M.-H.; Ersen, O.; Pouget, E.; Oda, R. *ACS Nano*, **2017**, *11*, 3806–3818.