

Supporting Information

for

Free-radical gases on two-dimensional transition-metal disulfides (XS_2 , $\text{X} = \text{Mo/W}$): robust half-metallicity for efficient nitrogen oxide sensors

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Additional computational data

Method

The spin-polarized structural relaxation and electronic property calculations were carried out using density functional theory (DFT) with the Perdew–Burke–Ernzerhof (PBE) generalized gradient approximation [1] exchange–correlation potential as implemented in the Vienna ab initio simulation package (VASP) [2-4]. A dispersion correction of the total energy (DFT-D3 method) [5] was used to incorporate the long-range van der Waals interactions. In addition to MoS₂, and WS₂, we expand our 2D material to GaS, GaSe, SnS, SnSe, WSe and Bi₂Se₃ (Figure S1). A cut-off energy of 500 eV for the plane-wave basis set and a Monkhorst–Pack mesh [6] of $7 \times 7 \times 1$ for the unit cell, $3 \times 3 \times 1$ for the 3×3 supercell, and $3 \times 3 \times 1$ for the 6×6 supercell for the Brillouin zone integration are employed to find the most favorable adsorption sites for NO and NO₂. For simplicity, only the PDOS of a single NO gas adsorption on unit cell surface is given in Figure S1. The unit cell for a single NO₂ adsorption is not incorporated in order to guarantee a certain distance between two neighbouring NO₂ molecules (Figure S2).

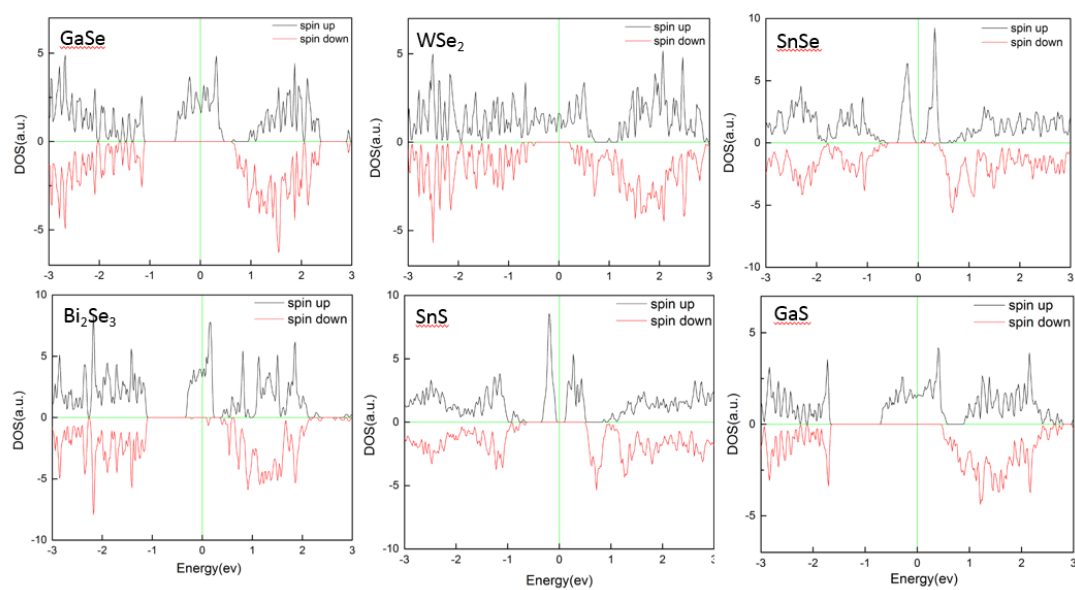


Figure S1: Total density of states (DOS) of NO adsorbed GaSe, WSe₂, SnSe, Bi₂Se₃, SnS and GaS. The Fermi level is indicated by the green line.

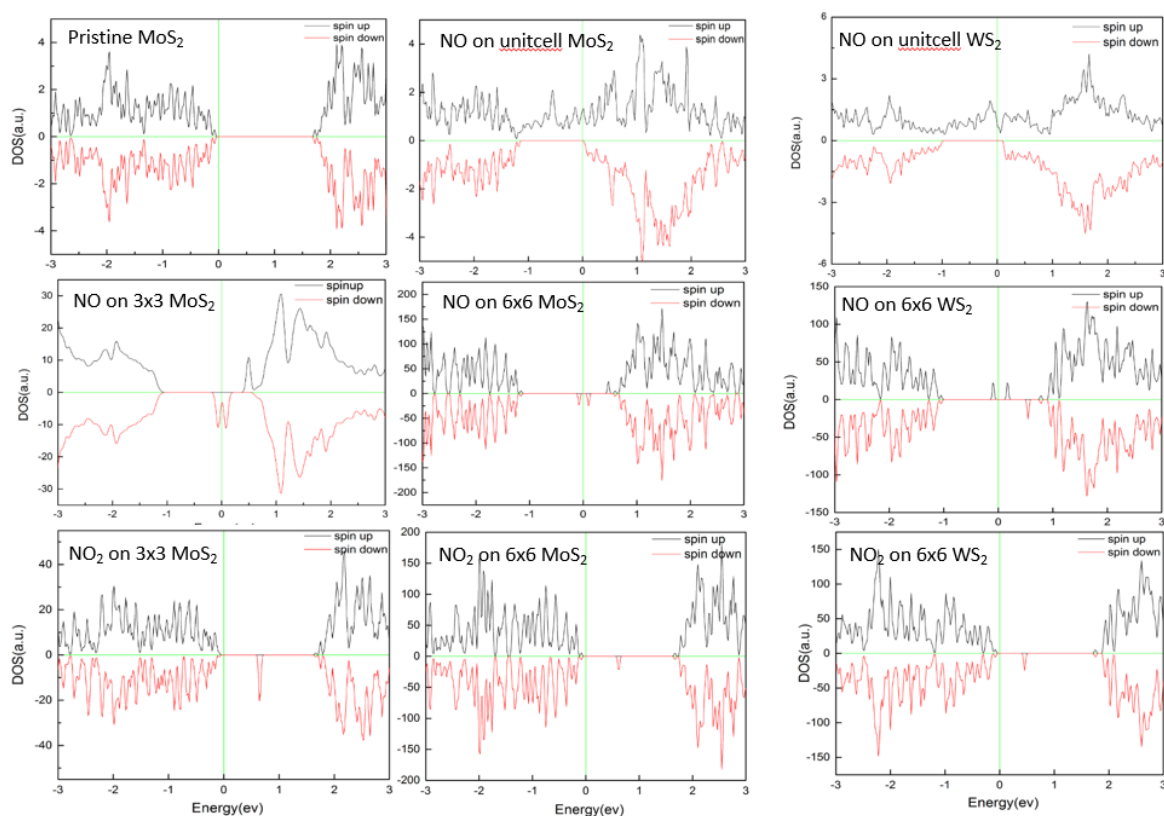


Figure S2: Total density of states (DOS) of the pristine 2D MoS₂ unit cell, 3 × 3 supercell and 6 × 6 MoS₂ supercell and with and with NO_x (x = 1, 2) adsorbed. The Fermi level is indicated by the green line.

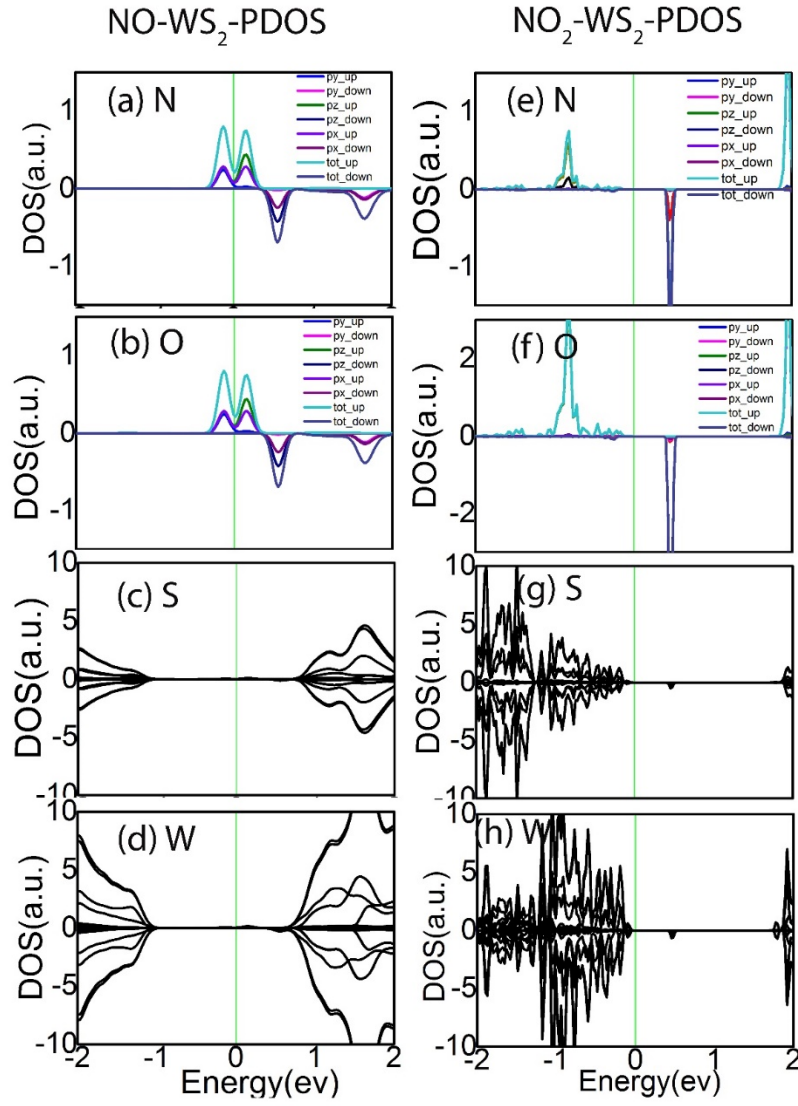


Figure S3: The N, O, W and S PDOS of WS₂ with (a–d) adsorbed NO and with (e–h) adsorbed NO₂. The Fermi level is indicated by the green line.

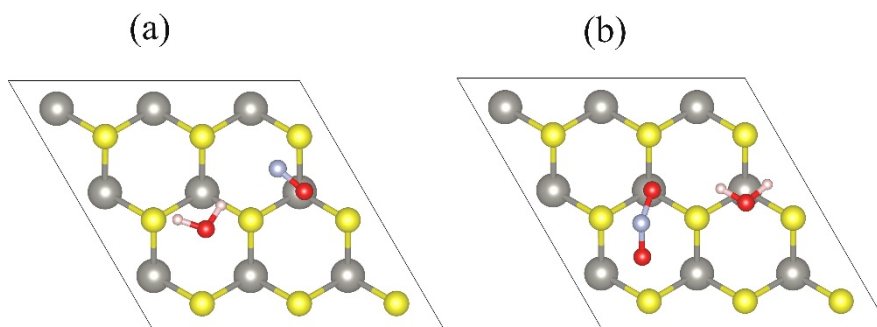


Figure S4: WS₂ with (a) adsorbed NO and with (b) adsorbed NO₂ in humid environment. The white, red, yellow, grey, dark grey balls represent H, O, S, N, W atoms, respectively.

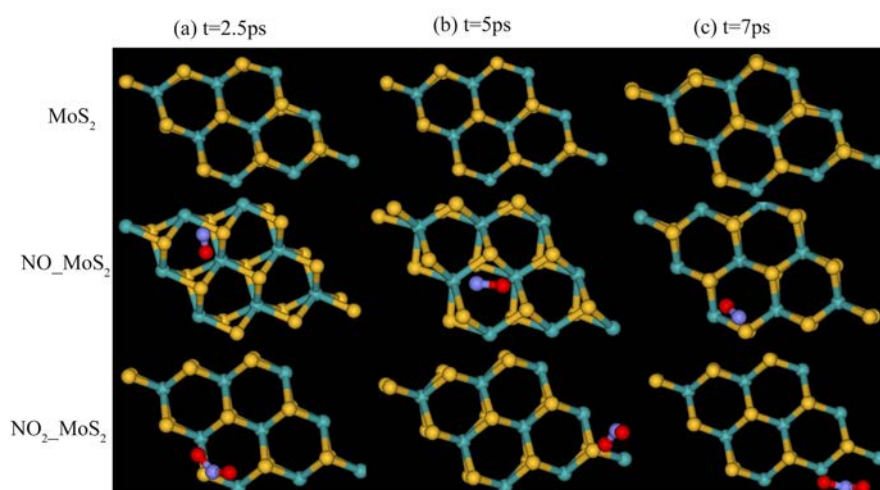


Figure S5: The thermodynamic stability at 300 K of MoS₂ monolayer, NO adsorbed MoS₂ (NO_MoS₂) and NO₂ adsorbed MoS₂ (NO₂_MoS₂) are evaluated through ab initio molecular dynamics (AIMD) simulations. The figures present configurations after (a) 2.5 ps, (b) 5 ps, and (c) 7 ps.

References

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