BEILSTEIN JOURNAL OF ORGANIC CHEMISTRY

Bromotyrosine-derived alkaloids from the Caribbean sponge Aplysina lacunosa

Qun Göthel, Thanchanok Sirirak and Matthias Köck*

Full Research Paper		Open Access
Address: Alfred-Wegener-Institut, Helmholtz-Zentrum für Polar- und Meeresforschung, Am Handelshafen 12, 27570 Bremerhaven,	Beilstein J. Org. Chem. 2015, <i>11</i> , 2334–2342. doi:10.3762/bjoc.11.254	
Germany	Received: 25 March 2015	
	Accepted: 01 September 2015	
Email:	Published: 26 November 2015	
Matthias Köck [*] - mkoeck@awi.de		
	Associate Editor: A. Kirschning	
* Corresponding author		
	© 2015 Göthel et al; licensee Beilstein-Institut.	
Keywords:	License and terms: see end of document.	
alkaloids; <i>Aplysina lacunosa</i> ; bromotyrosine; marine natural products; NMR spectroscopy		

Abstract

Three new bromotyrosine-derived alkaloids 14-debromo-11-deoxyfistularin-3 (1), aplysinin A (2), and aplysinin B (3), together with 15 known compounds (4–18) were isolated from the sponge *Aplysina lacunosa* collected from Stirrup Cay, Bahamas. The structures of the isolated compounds were identified on the basis of MS and NMR data analysis. The ¹³C NMR assignment of spirocyclohexadienylisoxazoline moieties of 1 and 2 were confirmed by an 1,1-ADEQUATE experiment. Compounds 1 and 2 showed a mild to moderate cytotoxic activities against KB-31 and FS4-LTM cell lines. Only aplysinin A (2) exhibited cytotoxicity against MCF-7 cells.

Introduction

Bromotyrosine-derived alkaloids are unique brominated metabolites which were isolated mainly from marine sponges of the order Verongida. For more than 50 years, bromotyrosine alkaloids raised the interests of synthetic and natural products chemists due to their high chemical diversity (besides the common spirocyclohexadienylisoxazoline moiety) and interesting biological activities. Since the first derivative 2,6-dibromo-4-acetamido-4-hydroxycyclohexadienone was isolated in 1967 [1], a series of bromotyrosine alkaloids were discovered with various biological activities, including antiviral [2], antibiotic [3-5], Na⁺/K⁺ ATPase inhibition [6-8], anti-HIV [9,10], anti-fungal [11], histidine-H3 antagonist [12], cytotoxic [13,14], and

antimalarial activities [15-17]. During our investigation of the chemical constituents of *Aplysina lacunosa* (Aplysinidae, Verongida), three new bromotyrosine-derived alkaloids: 14-debromo-11-deoxyfistularin-3 (1), aplysinin A (2), and aplysinin B (3) (Figure 1), together with 15 known compounds were obtained. In this report we describe the structure elucidation of 1 to 3 and the biological activities of all the isolated compounds.

Results and Discussion

The freeze-dried sponge was extracted three times with $CH_2Cl_2/MeOH$ (1:1, v/v). The resulting crude extract was parti-



tioned between n-hexane and MeOH. The MeOH extract was further partitioned between ethyl acetate and H2O. The resulting ethyl acetate phase was purified by vacuum liquid chromatography using silica gel with a stepwise gradient eluent from 100:0 to 80:20 (CH₂Cl₂/MeOH, v/v). The two fractions eluting with 97:3 and 94:6 (CH₂Cl₂/MeOH, v/v) were further purified by HPLC and yielded the three new bromotyrosine derivatives 14-debromo-11-deoxyfistularin-3 (1), aplysinins A (2) and B (3), as well as the 15 known compounds: 14-debromoaraplysillin I (4) [18], fistularin-3 (5) [19], 11,19-dideoxyfistularin-3 (6) [20], 19-deoxyfistularin-3 (7) [21], 11-deoxyfistularin-3 (8) [22], 11-ketofistularin-3 (9) [2], hexadellin B (10) [23], aerothionin (11) [13,24,25], 11-hydroxyaerothionin (12) [20], 11-oxoaerothionin (13) [13], 11-oxo-12-hydroxyaerothionin (14) [26], N-methyl-aerophobin-2 (15) [27], aeroplysinin-2 (16) [28], subereaphenol B (17) [29], and the unnamed bromotyrosine 18 [30]. Compounds 4 to 18 were identified by comparison of their MS data as well as ¹H and ¹³C NMR chemical shifts with those reported in the literature (Figure 2).

Compound **1** was obtained as a white solid. The MS–ESI(+) showed a characteristic pentabrominated ion peak cluster at m/z 1037/1039/1041/1043/1045/1047 [M + Na]⁺ (1:5:10:10:5:1). The molecular formula of C₃₁H₃₁Br₅N₄O₁₀ was deduced from HRMS–ESI(+) at m/z 1036.7844 [M + Na]⁺ (calcd for C₃₁H₃₁⁷⁹Br₅N₄O₁₀Na, 1036.7855) which required 16 double bond equivalents (DBEs). The ¹³C NMR spectrum of **1** indicated two amide groups at δ 158.9 (C-9') and 159.5 (C-9), 14 olefinic carbons, two hetero-olefinic carbons at δ 154.5 (C-8') and 155.0 (C-8), and 5 ring systems to fulfill the DBEs. The comparison of the ¹H and ¹³C NMR data from positions C-1, C-1' to C-9, C-9' (Table 1) of **1** with those of aerothionin (**11**) [13,24,25] allowed the assignment of the two dibromospirocyclohexadienylisoxazole carbonyl groups which was

further confirmed by ¹H,¹³C-HMBC and 1,1-ADEQUATE experiments. The ¹³C NMR assignment of C-2 and C-4 were reversed before Ciminiello's revision in 1994 [26]. Nevertheless, the wrong assignment has still continued to be used as reference in the literature [29,31]. We therefore applied an 1,1-ADEQUATE experiment which allows the selective observation of two-bond H,C correlations [32]. The signals in the 1,1-ADEQUATE spectrum from $\delta_{\rm H}$ 3.92 (H-1, H-1') to $\delta_{\rm C}$ 113.6 (C-2, C-2') and from $\delta_{\rm H}$ 6.57 (H-5, H-5') to $\delta_{\rm C}$ 120.9 and 120.8 (C-4 and C-4', respectively) confirmed the assignments of C-2, C-2' and C-4, C-4' from 1994 (Figure 3). ¹H, ¹H-COSY correlations were observed among 9-NH (& 8.60)/H-10 (& 3.36, 2H)/H-11 (δ 1.95, 2H)/H-12 (δ 4.06, 2H) establishing an propanamine moiety. In a similar manner, ¹H, ¹H-COSY correlations among 9'-NH (\$ 8.35)/H-20 (\$ 3.35)/H-19 (\$ 4.65, 2H) revealed the presence of a hydroxylethylamine moiety. The remaining signals at $\delta_{\rm H}$ 7.05 (d, J = 8.6 Hz, H-14), 7.28 (dd, J = 1.5, 8.6 Hz, H-15), and 7.52 (d, J = 1.5 Hz, H-17) together with ¹H,¹³C-HMBC correlations indicated the presence of a 1,2,4trisubstituted phenoxy group. The phenoxy group was connected to the propanamine and the hydroxyethylamine substructures according to ¹H, ¹³C-HMBC correlations from H-12 to δ_C 153.8 (C-13) and from H-19 to δ_C 137.2 (C-16). Both sides of the linear fragment were connected to dibromospirocyclohexadiene moieties through amide bonds according to ¹H,¹³C-HMBC correlations from H-7 (δ 3.21, 3.63) and 9-NH to C-9 (δ 159.1) as well as from H-7' (δ 3.19, 3.62) and 9'-NH to C-9' (δ 159.0). The structure of **1** is closely related to 11-deoxyfistularin-3 (8) which was originally isolated from the Caribbean sponge Aplysina fistularis insularis [22]. The only difference between 1 and 8 is the lack of one bromine atom in the central benzene ring of compound 1 at C-14. Therefore, compound 1 was named 14-debromo-11-deoxyfistularin-3. The ¹³C chemical shifts assignment of **1** according to the 1,1-ADEQUATE suggested a revision of the chemical shifts of C-2,



C-2', C-6, and C-6' of the two related compounds 11-deoxy-fistularin-3 (8) and 14-debromoaraplysilin I (4) [18] (Table 1 and Table 2, respectively).

The relative configuration of the spiroisoxazoline rings of 1 was investigated using a NOESY experiment. NOEs were observed between $\delta_{\rm H}$ 6.36 (1-OH) and 3.60 (H-7); 6.37 (1'-OH) and 3.63 (H-7'); 3.57 (2H, H-5-and H-5') and 3.21 (2H, H-7 and H-7') indicated a trans-hydroxyspiroisoxazoline ring which was supported by a W-coupling between the olefinic proton H-5 and the methine proton H-1 (${}^{4}J \sim 0.7$ Hz) [28,33]. An NOE was also observed between $\delta_{\rm H}$ 5.54 (19-OH) and $\delta_{\rm H}$ 6.37 (1'-OH) suggested that both hydroxy groups are on the same side of the molecule. The absolute configuration of compound 1 was investigated by CD spectroscopy. The CD spectrum of 1 showed positive Cotton effects (λ_{max} 248, $\Delta \epsilon$ +5.16, λ_{max} 285, $\Delta \epsilon$ +4.58) with the same sign and magnitude as observed for (+)-aerothionin (11) [25,34,35]. Thus, the absolute configuration of spiroisoxazoline moieties were assigned as 1,1'-(R),6,6'-(S) (Figure 1). The absolute configuration of C-19

was assigned as 19-(R) according to NOE data which is in agreement with the proposed configuration by Molinski and co-workers [34,36]. The configuration of C-19 of fistularin-3 (5) was proposed to be the same as of **19**, chemical fragmentation of **5** releasing from the sponge *Aplysina* spp. after induction by tissue damage (Figure 4). However, the conversion from **5** to **19** has not been confirmed. A single data set in the ¹³C NMR spectrum supported the presence of one diastereomer of **1**.

Compound **2** was isolated as a white solid. MS–ESI(+) data of **2** showed a pseudomolecular ion cluster at m/z 793/795/797/799/ 801 [M + Na]⁺ (1:4:6:4:1) indicating a tetrabrominated compound. HRMS–ESI(+) of **2** at m/z 793.8320 [M + Na]⁺ suggested a molecular formula of C₂₃H₂₅Br₄N₃O₇ (calcd for C₂₃H₂₅⁷⁹Br₄N₃O₇Na, 793.8324). The ¹H and ¹³C NMR spectra of **2** resemble to **1** except that the spectrum of **2** showed one extra methyl group ($\delta_{\rm H}$ 1.80 s; $\delta_{\rm C}$ 22.6) and one aromatic proton less (C-14, $\delta_{\rm H}$ 7.05 for **1**). According to one set of $\delta_{\rm H}$ 3.17 (H-7), 3.61 (H-7), 3.93 (H-1), 6.56 (H-5), $\delta_{\rm C}$ 39.7 (C-7), 73.5

4 0 03					03	
nosition		1			8	8 ^a δ _C
position	δ_{C}	δ _H	1,1-ADEQ	δ_{C}	δ _H	
1, 1′	73.6, CH; 73.5, CH	3.92, 2H s	2, 2′, 6, 6′	74.1; 74.0	3.93, d (7.9)	74.67; 74.60
2, 2 <i>′</i>	113.6, 2C	-	-	113.6	-	121.66 ^b
3, 3´	147.1, 2C	-	_	147.6	_	147.92
4, 4 <i>′</i>	120.9, C; 120.8, C	-	_	121.4; 121.3	-	115.16 ^b
5, 5´	131.2, CH; 131.1, CH	6.57, 2H s	6, 6′	131.7; 131.6	6.57, s; 6.59, s	132.31; 132.15
6, 6´	90.3, C; 90.2, C	-	_	90.8; 90.7	_	91.78; 91.72
7	40.0 ^c , CH ₂	3.21, d (18.2)	6, 8	39.9	3.22, d (18.2)	40.27
		3.63, d (18.2)			3.63, d (18.2)	
7′	39.9 ^c , CH ₂	3.19, d (18.2) 3.62, d (18.2)	6′, 8′	39.6	3.18, d (18.1) 3.62, d (18.1)	
8, 8 <i>′</i>	154.5, C; 154.6, C	-	_	155.0; 154.9	_	155.23; 155.10
9, 9′	159.1, C; 159.0, C	-	_	159.5; 159.4	_	160.44; 160.05
10	36.1, CH ₂	3.36 ^c , 2H, overlapped	11	36.7	_	37.13
11	28.6, CH ₂	1.95, 2H qui (6.4)	10, 12	29.9	2.01, qui (7.2)	30.37
12	66.5, CH ₂	4.06, 2H t (6.2)	11	71.7	3.98, t (6.4)	71.51
13	153.8, C	-	_	151.8	_	152.27
14	113.3, CH	7.05, d (8.6)	13, 15	117.8	_	118.35
15	126.7, CH	7.28, dd (1.5, 8.6)	_	130.9	7.58, s	130.90
16	137.2, C	-	_	143.1		143.35
17	130.5, CH	7.52, d (1.5)	_	130.9	7.58, s	130.90
18	110.8, C	-	_	117.8	_	118.35
19	69.9, CH	4.65, dt (4.5, 7.2)	16, 20	69.9	4.69, q (5.3)	70.70
20	46.8, CH ₂	3.35 ^c , 2H overlapped	-	46.8	3.29, m 3.34 ^c	47.99
3, 3´-OMe	59.7, 2CH ₃	3.64, 6H s	_	60.1	3.66, s	59.75
9-NH	_	8.60, t (5.8)	_	_	8.57, t (5.7)	_
9´-NH	_	8.35, t (5.8)	_	_	8.39, t (5.7)	_
1-OH	_	6.36 ^d	_	_	6.36, d (7.9)	_
1′-OH	_	6.37 ^d	_	_	6.37, d (7.9)	_
19-OH	_	5.54. d (4.5)	_	_	5.73. d (5.3)	_

^aThe ¹³C NMR data were obtained in pyridine-*d*₅ at 67.5 MHz [22]. ^bAssignments should be reversed. ^cSignal obscured by the H₂O residual signal in DMSO-*d*₆; chemical shift was obtained from 2D NMR spectra. ^dChemical shifts were obtained from the sample before purification. Peaks were not observed in the purified sample.

Table 2: 13 C NMR data (600 MHz, DMSO- d_6) of the central benzene ring of the isolated compounds 1 and 4.

position	1	4	4 ^a
13	153.8	153.3	153.4
14	113.3	111.3	134.4 ^b
15	126.7	133.8	133.4
16	137.2	133.4	112.2 ^b
17	130.5	129.6	128.8
18	110.8	113.6	113.3

 ^{a}The ^{13}C NMR data were obtained in CDCl_{3} [18]. $^{b}\text{Assignments}$ should be reversed.

(C-1), 90.3 (C-6), 113.1 (C-2), 120.9 (C-4), 131.1 (C-5), 147.1 (C-3), 154.3 (C-8), and 159.0 (C-9) together with ¹H,¹³C-HMBC correlations revealed that **2** consist of only one spirocyclohexadienylisoxazoline moiety in comparison with **1**. The structure determination of **2** was accomplished based on ¹H,¹H-COSY and ¹H,¹³C-HMBC correlations in the same manner as for **1**. Once again, ¹H,¹H-COSY revealed propanamine [H-18 (δ 3.96, 2H)/H-19 (δ 1.91, 2H)/H-20 (δ 3.25)/20-NH (δ 7.86)] and hydroxyethylamine [9-NH (δ 8.39)/H-10 (δ 3.33, 2H)/H-11 (δ 4.67)/11-OH (δ 5.72)] substructures. The signals at $\delta_{\rm H}$ 7.57 (2H, s, H-13,17), $\delta_{\rm C}$ 117.3 (C-14, 16), 130.4 (C-13, 17), 142.5 (C-12), and 151.4 (C-15) and ¹H,¹³C-HMBC correlations among those signals showed a 1,2,4,6-dibromophenyl moiety. The substructures were assem-



Figure 3: 1,1-ADEQUATE spectrum of 14-debromo-11-deoxyfistularin-3 (1).



bled by ¹H, ¹³C-HMBC correlations from H-10 to C-9 and C-12, from H-11 to C-12 and C-13 as well as from $\delta_{\rm H}$ H-18 to C-15. The terminal of side chain was connected to an acetamide moiety according to ¹H, ¹³C-HMBC correlations from H-20, 20-NH, and $\delta_{\rm H}$ 1.80 (3H; H-22) to $\delta_{\rm C}$ 169.1 (C-21). Compound **2** was named aplysinin A. The structure of **2** is similar to rightside portions of **1** and **8** (start at C-10). However, compound **2** contained an acetamide in the left-side portion instead of a ring system in comparison with **8** and showed great similarity with hexadellin B (**10**) isolating from the same organism. Hexadellin B (**10**) was originally isolated from the sponge *Hexadella* sp. The spectroscopic data of **10** was coincidently obtained from diacetylhexadellin B (**20**, Figure 5 and Table 3) [23] which supported the assignment of the acetamide moiety. The relative



Figure 5: Diacetylhexadellin B (20) isolated from sponge *Hexadella* sp.

	2		10		20 ^a	
position	δ _C	δ _H	δ _C	δ _H	δ _C	
1	73.5, CH	3.93, d (8.2)	74.0	3.92, d (7.3)	73.1	
2	113.1, C	_	113.5	_	122.1 ^b	
3	147.1, C	_	147.6	-	149.7	
4	120.9, C	_	121.4	-	107.8 ^b	
5	131.1, CH	6.56, s	131.6	6.58, d (0.8)	130.2	
6	90.3, C	_	90.7	_	89.9	
7	39.7, CH ₂	3.17, d (18.4);	39.6	3.19, d (18.1);	39.9	
		3.61, d (18.4)		3.60, d (18.1)		
8	154.3, C	_	154.9	_	153.5	
9	159.0, C	_	159.4	-	158.6	
10	46.3, CH ₂	3.33, 2H, overlapped	40.4	3.38, m	40.4	
11	69.3, CH	4.67, t (6.1)	33.6	2.77, t (7.1)	34.4	
12	142.5, C	_	139.5	-	137.2	
13,17	130.4, 2CH	7.57, 2H, s	133.5	7.54, s	132.8	
14,16	117.3, 2C	_	117.7	-	118.2	
15	151.4, C	_	150.9	-	151.5	
18	71.3, CH ₂	3.96, 2H, t (6.2)	70.8	4.00, t (6.1)	72.1	
19	29.8, CH ₂	1.91, 2H, q (6.9)	28.2	2.08, m	29.4	
20	35.7, CH ₂	3.25, 2H, q (6.7)	37.0	3.08, br s	37.7	
21	169.1, C	_	-	-	170.0	
22	22.6, CH ₃	1.80, 3H, s	_	-	23.6	
3-OMe	59.6, CH ₃	3.65, 3H, s	60.1	3.65, s	60.3	
9-NH	-	8.39, t (5.9)	_	8.59, t (5.9)	-	
1-OH	-	6.36, d (8.2)	-	6.37, d (7.8)	-	
11-OH	-	5.72, d (4.4)	-	-	_	
20-NH	_	7.86, t (5.3)	_	_	_	

configuration of the spiroisoxazoline moiety (C-1 and C-6) was determined by comparison of the ¹H and ¹³C NMR data with 10 and **20**. The NOESY spectrum showed correlations between δ_H 6.36 (1-OH) and 3.61 (C-7) as well as between 6.56 (H-6) and 3.17 (H-7) suggesting a trans-hydroxyspiroisoxazoline ring similar to compound 1. An NOE was also observed between δ_{H} 5.72 (11-OH) and 1-OH indicating the same planar alignment of both hydroxy groups. The absolute configuration of spiroisoxazoline moiety was confirmed as 1-(R), 6-(S) by positive Cotton effects (λ_{max} 252, $\Delta \epsilon$ +4.77, λ_{max} 283, $\Delta \epsilon$ +3.34) comparing to (+)-aerothionin (11) in the same manner of 1 [25,37]. The arrangement of 1-OH and 19-OH on the same side of the structure allowed assigning the configuration of C-19. The presence of one diastereomer of 2 was confirmed by a single data set in the ¹³C NMR spectrum.

Compound 3 was isolated together with N-methylaerophobin-2 (15) as a mixture (approximate ratio 1:5). The ESIMS spectrum exhibited a 1:2:1 ion cluster at m/z 457/459/461, indicating the

presence of two bromine atoms. The HRMS-ESI(+) spectrum revealed a pseudomolecular ion $[M + H]^+$ at m/z 456.9892, which indicated a molecular formula of C16H18Br2N4O2 (calcd for C₁₆H₁₉⁷⁹Br₂N₄O₂, 456.9875), containing nine DBEs. Two singlet aromatic protons δ_H 7.88 (δ_C 131.7) suggested a tetrasubstituted benzene pattern which was confirmed by ¹H,¹³C-HMBC correlations; from $\delta_{\rm H}$ 7.88 (2H, H-2, H-6) to δ_{C} 118.5 (C-3, C-5), δ_{C} 154.4 (C-4), δ_{C} 131.7 (C-2, C-6), and $\delta_{\rm C}$ 135.7 (C-7), from $\delta_{\rm H}$ 3.82 (4-OMe) to C-4. The connection of the benzene fragment to the E-vinyl moiety was confirmed by HMBC correlations from two olefinic protons $\delta_{\rm H}$ 7.33 (d, J = 15.8 Hz, H-7) and $\delta_{\rm H}$ 6.66 (d, J = 15.8 Hz, H-8) to $\delta_{\rm C}$ 134.9 (C-1). ¹H, ¹H-COSY correlations in between 9-NH (δ 8.17)/ H-10 (δ 3.20, 2H)/H-11 (δ 1.71, 2H)/H-12 (δ 2.45, 2H) indicated a propanamine fragment which was connected to a 2-aminoimidazole moiety according to ¹H, ¹³C-HMBC correlations from H-12 to δ_{C} 126.8 (C-13) and 109.2 (C-14) (Table 4). The two substructures are connected through an amide bond according to the ¹H,¹³C-HMBC correlations from H-7, 9-NH,

Table 4: NMF pound 21.	R data of aplysinin	B (3) (600 MHz, DMS	O-d ₆) and com-
position		21 ^a	
	δ _C	δ _H	δ _C
1	134.9, C	-	136.36
2, 6	131.7, CH	7.88, 2H, s	132.98
3, 5	118.5, C	-	119.48
4	154.4, C	_	154.46
7	135.7, CH	7.33, d (15.8)	138.42
8	124.9, CH	6.66, d (15.8)	123.95
9	165.0, C	-	167.50
10	38.4, CH ₂	3.20, 2H, m	39.27
11	28.1, CH ₂	1.71, 2H, m	25.92
12	22.0, CH ₂	2.45, 2H, m	125.89
13	126.8, C	-	110.87
14	109.2, CH	6.60, s	146.43
15	147.4, C	-	_
4-OMe	61.0, CH ₃	3.82, s	61.26
9-NH	_	8.17, t (5.58)	-



The new compounds 14-debromo-11-deoxyfistularin-3 (1) and aplysinin A (2) were tested for their antimicrobial activity against different Gram-positive and Gram-negative bacteria, fungi, and for their antiproliferative activity. Aplysinin B (3) was not subjected to any biological activity test due to the minute amount and its existence as the minor compound of a mixture. The results showed that 1 and 2 exhibited mild cytotoxic activity against KB-31 epidermoid carcinoma cells (IC₅₀ = 69 and 26 μ M, respectively). Only 2 showed mild toxicity against the breast cancer cell line MCF-7 and to FS4-LTM conditional immortalization human fibroblasts (IC₅₀ = 78 and 32 μ M, respectively). The cytotoxicities of the known compounds (4–17) are also listed in Table 5. None of the isolated compounds showed any antimicrobial activity.

Table 5: Cytotoxicity of the isolated compounds (IC_{50}) . ^a					
compound	IC ₅₀ [μM]				
compound	L929	KB-31	MCF-7	FS4-LTM	
1	_	68.8	_	_	
2	-	25.8	77.5	32.2	
4	94.3	-	78.6	-	
5	-	-	206.9	-	
6	117.6	88.2	-	-	
7	-	-	60.0	-	
8	-	-	47.2	87.3	
10	_	-	90.6	73.4	
15	55.9	48.9	-	-	
16	_	-	96.3	-	
17	-	-	64.8	-	

compounds with no deavity are not instea in the tab

Experimental

UV spectra were recorded during HPLC separation with a DAD detector (JASCO MD-2010 Plus). CD spectra were recorded on a JASCO J-810 spectropolarimeter. Low and high resolution ESIMS was performed with a Bruker micrOTOF_{LC} mass spectrometer. Mass calibration was performed using sodium formate cluster ions prior each measurement. ¹H and ¹³C spectra were recorded on a Bruker Avance 600 NMR spectrometer equipped with a cryo platform (¹H at 600 MHz, ¹³C at 150 MHz) and a Bruker Avance NMR spectrometer (¹H at 400 MHz, ¹³C at 100 MHz). All NMR experiments were measured at a temperature of 303 K using DMSO-*d*₆ ($\delta_{\rm H}$ 2.50, $\delta_{\rm C}$ 39.5) as internal standard. HPLC separation was achieved by Jasco PU-1580 using a Kromasil RP18 column (16 mm × 250 mm, 5 µm) and a Kromasil RP18 column (11 mm × 50 mm, 5 µm) and was eluted with gradient H₂O (0.1% TFA) and MeCN (0.1% TFA).

The sponge Aplysina lacunosa was collected by SCUBA diving at a depth of 8 m from Stirrup Cay in the Bahamas in June 2008. The sample was immediately frozen and kept at -20 °C until extraction. A voucher specimen of this species is deposited in AG Köck, Alfred-Wegener-Institut, Helmholtz-Zentrum für Polar- und Meeresforschung (voucher number: Aplysina lacunosa 08/21). The freeze-dried sponge (200 g) was extracted three times with CH₂Cl₂/MeOH (1:1, v/v) at room temperature. The filtrates were pooled and evaporated to yield 24.3 g of crude extract which was further partitioned between *n*-hexane and MeOH. The MeOH extract was then partitioned between EtOAc and H₂O. The EtOAc fraction was further purified by vacuum liquid chromatography using silica gel eluting with stepwise gradient from 100:0 to 80:20 (CH₂Cl₂/MeOH, v/v). The fraction eluted with 97:3 (CH₂Cl₂/MeOH, v/v) was concentrated and further purified by HPLC using an RP C18 column (stepwise gradient 60:40, 40:60, and 20:80 H₂O/MeCN, v/v) to

yield 5 (198.0 mg), 11 (263.0 mg), 12 (445.0 mg), 13 (55.0 mg), 14 (7.9 mg), 16 (15.0 mg), 17 (9.1 mg), and two other fractions. The first fraction was purified using reversedphase HPLC [52:48 H2O (0.1% TFA)/MeCN (0.1% TFA), v/v] to obtain 7 (3.2 mg), 8 (4.5 mg), 9 (9.6 mg) and following with analytical RP18 column [gradient 60:40 to 20:80 H₂O (0.1% TFA)/MeCN (0.1% TFA), v/v] yielding 1 (0.8 mg). The other fraction was purified using RP18 HPLC [60:40 H₂O (0.1% TFA)/MeCN (0.1% TFA), v/v] and follow with an analytical RP18 column [65:35 H₂O (0.1% TFA)/MeCN (0.1% TFA), v/v] yielding 2 (0.8 mg) and 18 (5.3 mg). The 94:6 (CH₂Cl₂/MeOH, v/v) fraction was purified by HPLC using a RP C18 column (gradient 80:20 to 40:60 H₂O/MeCN, v/v) yielding 6 (10.7 mg), 10 (25.5 mg), 15 (5.0 mg), and one fraction which was purified using reversed-phase HPLC [70:30 H₂O (0.1% TFA)/MeCN (0.1% TFA), v/v] to get a mixture of **3** and **15** (3.5 mg) as well as **4** (3.0 mg).

Biological activity test

The antimicrobial activities of isolated compounds were evaluated against five microorganisms [Gram-positive: *Straptococcus aureus* (MRSA and MSSA) and *Micrococcus luteus*; Gram-negative: *Peumonia aruginosa* and *Klebsiella pneumonia*] and antifungal *Candida albicans* using microdilution technique. The MIC was defined as lowest concentration that shows 50% growth inhibition after 24 hour incubation.

Cytotoxicity assay

The cytotoxicity was determined using WST-1 cell proliferation assays. Targeting cell lines are L929 mouse fibroblasts, KB-31 epidermoid carcinoma, and MCF-7 breast cancer cell lines which were incubated for 5 days with the test substances. The acute toxicity was determined using the FS4-LTM conditional immortalization human fibroblasts cell line which was incubated for 24 hours with the test compounds.

Experimental data

14-Debromo-11-deoxyfistularin-3 (1): white solid; UV (DAD) λ_{max} 226 nm; CD (MeOH) λ_{max} 248 nm ($\Delta \epsilon$ +5.16), 288 nm ($\Delta \epsilon$ +4.55); ¹H NMR and ¹³C NMR see Table 1; HRMS–ESI(+) $m/z = 1036.7844 \ [M + Na]^+$ (calcd for $C_{31}H_{31}^{79}Br_5N_4O_{10}Na$, 1036.7855, $\Delta m = 1.1$ ppm).

Aplysinin A (2): white solid; UV (DAD) λ_{max} 225 nm; CD (MeOH) λ_{max} 252 nm ($\Delta\epsilon$ +4.77), 283 nm ($\Delta\epsilon$ +3.34); ¹H NMR and ¹³C NMR see Table 3; HRMS-ESI(+) *m*/*z* = 793.8320 [M + Na]⁺ (calcd for C₂₃H₂₅⁷⁹Br₄N₃O₇Na, 793.8324, $\Delta m = 0.4$ ppm).

Aplysinin B (3): white solid; UV (DAD) λ_{max} 227 nm; ¹H NMR and ¹³C NMR see Table 4; HRMS-ESI(+) $m/z = 456.9892 [M + H]^+$ (calcd for $C_{16}H_{19}^{79}Br_2N_4O_2$, 456.9875, $\Delta m = 1.7$ ppm).

Supporting Information

Supporting Information File 1

1D, 2D NMR, and CD spectra of three new compounds. 1D NMR, mass and CD spetra of all known isolated compounds.

[http://www.beilstein-journals.org/bjoc/content/ supplementary/1860-5397-11-254-S1.pdf]

Acknowledgements

Financial support from the Deutsche Forschungsgemeinschaft (DFG) (Ko 1314/5-1 and 5-2, DFG-Forschergruppe FOR 934) is gratefully acknowledged. Sponge collection was carried out by Dr. Gesine Schmidt and Dr. Achim Grube during a scientific expedition to the Bahamas in 2008. We would like to acknowledge the support of Prof. Dr. Joseph R. Pawlik (University of North Carolina, Wilmington, USA) who gave members of the Köck research group the opportunity to participate in the research trips to the Bahamas. We further thank Dr. Sven Zea (Universidad Nacional de Colombia) for identification of the sponge samples, Dr. Florenz Sasse (Helmholtz Center for Infection Research, Braunschweig) for biological activity tests, and Dr. Andreas Hennig (Jacobs University, Bremen) for CD spectrometer access. The photograph in the graphical abstract was reproduced with permission from Sven Zea (http:// www.spongeguide.org).

References

- Sharma, G. M.; Burkholder, P. R. *Tetrahedron Lett.* **1967**, *8*, 4147–4150. doi:10.1016/S0040-4039(01)89710-0
- Gunasekera, S. P.; Cross, S. S. J. Nat. Prod. 1992, 55, 509–512. doi:10.1021/np50082a020
- Kobayashi, J.; Tsuda, M.; Agemi, K.; Shigemori, H.; Ishibashi, M.; Sasaki, T.; Mikami, Y. *Tetrahedron* **1991**, *47*, 6617–6622. doi:10.1016/S0040-4020(01)82314-0
- 4. Andersen, R. J.; Faulkner, D. J. *Tetrahedron Lett.* **1973**, *14*, 1175–1178. doi:10.1016/S0040-4039(01)95788-0
- Gao, H.; Kelly, M.; Hamann, M. T. *Tetrahedron* 1999, *55*, 9717–9726. doi:10.1016/S0040-4020(99)00553-0
- Okamoto, Y.; Ojika, M.; Kato, S.; Sakagami, Y. *Tetrahedron* 2000, *56*, 5813–5818. doi:10.1016/S0040-4020(00)00544-5
- Tsuda, M.; Shigemori, H.; Ishibashi, M.; Kobayashi, J. Tetrahedron Lett. 1992, 33, 2597–2598. doi:10.1016/S0040-4039(00)92253-6
- Yagi, H.; Matsunaga, S.; Fusetani, N. *Tetrahedron* 1993, 49, 3749–3754. doi:10.1016/S0040-4020(01)90227-3
- Ichiba, T.; Scheuer, P. J.; Kelly-Borges, M. J. Org. Chem. 1993, 58, 4149–4150. doi:10.1021/jo00067a062

- Ross, S. A.; Weete, J. D.; Schinazi, R. F.; Wirtz, S. S.; Tharnish, P.; Scheuer, P. J.; Hamann, M. T. *J. Nat. Prod.* 2000, *63*, 501–503. doi:10.1021/np980414u
- 11. Tsukamoto, S.; Kato, H.; Hirota, H.; Fusetani, N. *Tetrahedron* **1996**, *52*, 8181–8186. doi:10.1016/0040-4020(96)00387-0
- Mierzwa, R.; King, A.; Conover, M. A.; Tozzi, S.; Puar, M. S.; Patel, M.; Coval, S. J.; Pomponi, S. A. *J. Nat. Prod.* **1994**, *57*, 175–177. doi:10.1021/np50103a029
- Acosta, A. L.; Rodriguez, A. D. J. Nat. Prod. 1992, 55, 1007–1012. doi:10.1021/np50085a031
- 14. Tabudravu, J. N.; Jaspars, M. J. Nat. Prod. 2002, 65, 1798–1801. doi:10.1021/np020275n
- Yang, X.; Davis, R. A.; Buchanan, M. S.; Duffy, S.; Avery, V. M.; Camp, D.; Quinn, R. J. *J. Nat. Prod.* **2010**, *73*, 985–987. doi:10.1021/np900834g
- Galeano, E.; Martínez, A.; Thomas, O. P.; Robledo, S.; Munoz, D. *Quim. Nova* **2012**, *35*, 1189–1193. doi:10.1590/S0100-40422012000600023
- Galeano, E.; Thomas, O. P.; Robledo, S.; Munoz, D.; Martinez, A. Mar. Drugs 2011, 9, 1902–1913. doi:10.3390/md9101902
- James, D. M.; Kunze, H. B.; Faulkner, D. J. J. Nat. Prod. 1991, 54, 1137–1140. doi:10.1021/np50076a040
- Gopichand, Y.; Schmitz, F. J. *Tetrahedron Lett.* **1979**, *41*, 3921–3924. doi:10.1016/S0040-4039(01)86465-0
- Kernan, M. R.; Cambie, R. C.; Bérgquist, P. R. J. Nat. Prod. 1990, 53, 615–622. doi:10.1021/np50069a012
- Mancini, I.; Guella, G.; Laboute, P.; Debitus, C.; Pietra, F. J. Chem. Soc., Perkin Trans. 1 1993, 3121–3125. doi:10.1039/p19930003121
- Compagnone, R. S.; Avila, R.; Suárez, A. I.; Abrams, O. V.; Rangel, H. R.; Arvelo, F.; Piña, I. C.; Merentes, E. *J. Nat. Prod.* **1999**, 62, 1443–1444. doi:10.1021/np9901938
- 23. Morris, S. A.; Andersen, R. J. Can. J. Chem. 1989, 67, 677–681. doi:10.1139/v89-102
- 24. Fattorusso, E.; Minale, L.; Sodano, G.; Moody, K.; Thomson, R. H. J. Chem. Soc. D 1970, 752–753. doi:10.1039/c29700000752
- McMillan, J. A.; Paul, I. C.; Goo, Y. M.; Rinehart, K. L., Jr.; Krueger, W. C.; Pschigoda, L. M. *Tetrahedron Lett.* **1981**, *22*, 39–42. doi:10.1016/0040-4039(81)80035-4
- Ciminiello, P.; Costantino, V.; Fattorusso, E.; Magno, S.; Mangoni, A. J. Nat. Prod. 1994, 57, 705–712. doi:10.1021/np50108a004
- 27. Assmann, M.; Wray, V.; van Soest, R. W. M.; Proksch, P. Z. Naturforsch., C 1998, 53, 398–401.
- Minale, L.; Sodano, G.; Chan, W. R.; Chen, A. M. J. Chem. Soc., Chem. Commun. 1972, 674–675. doi:10.1039/c39720000674
- Abou-Shoer, M. I.; Shaala, L. A.; Youssef, D. T. A.; Badr, J. M.; Habib, A.-A. M. *J. Nat. Prod.* 2008, *71*, 1464–1467. doi:10.1021/np800142n
- Nishiyama, S.; Yamamura, S. Bull. Chem. Soc. Jpn. 1985, 58, 3453–3456. doi:10.1246/bcsj.58.3453
- Tilvi, S.; Rodrigues, C.; Naik, C. G.; Parameswaran, P. S.; Wahidhulla, S. *Tetrahedron* **2004**, *60*, 10207–10215. doi:10.1016/j.tet.2004.09.009
- Köck, M.; Reif, B.; Fenical, W.; Griesinger, C. Tetrahedron Lett. 1996, 37, 363–366. doi:10.1016/0040-4039(95)02206-6
- 33. Fulmor, W.; van Lear, G. E.; Morton, G. O.; Mills, R. D. *Tetrahedron Lett.* **1970**, *11*, 4551–4552. doi:10.1016/S0040-4039(00)89414-9

- 34. Rogers, E. W.; de Oliveira, M. F.; Berlinck, R. G. S.; König, G. M.; Molinski, T. F. J. Nat. Prod. 2005, 68, 891–896. doi:10.1021/np050050n
- Rogers, E. W.; Molinski, T. F. J. Nat. Prod. 2007, 70, 1191–1194. doi:10.1021/np0701091
- 36. Ebel, R.; Brenzinger, M.; Kunze, A.; Gross, H. J.; Proksch, P. *J. Chem. Ecol.* **1997**, *23*, 1451–1462. doi:10.1023/B:JOEC.0000006475.10310.3a
- Ciminiello, P.; Dell'Aversano, C.; Fattorusso, E.; Magno, S.; Pansini, M. J. Nat. Prod. **1999**, 62, 590–593. doi:10.1021/np9805138
- Ciminiello, P.; Fattorusso, E.; Magno, S.; Pansini, M. J. Nat. Prod. 1994, 57, 1564–1569. doi:10.1021/np50113a016

License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/2.0), which

permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (http://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at: doi:10.3762/bjoc.11.254