



Bifurcated synthesis of methylene-lactone- and methylene-lactam-fused spiro-lactams via electrophilic amide allylation of γ -phenylthio-functionalized γ -lactams

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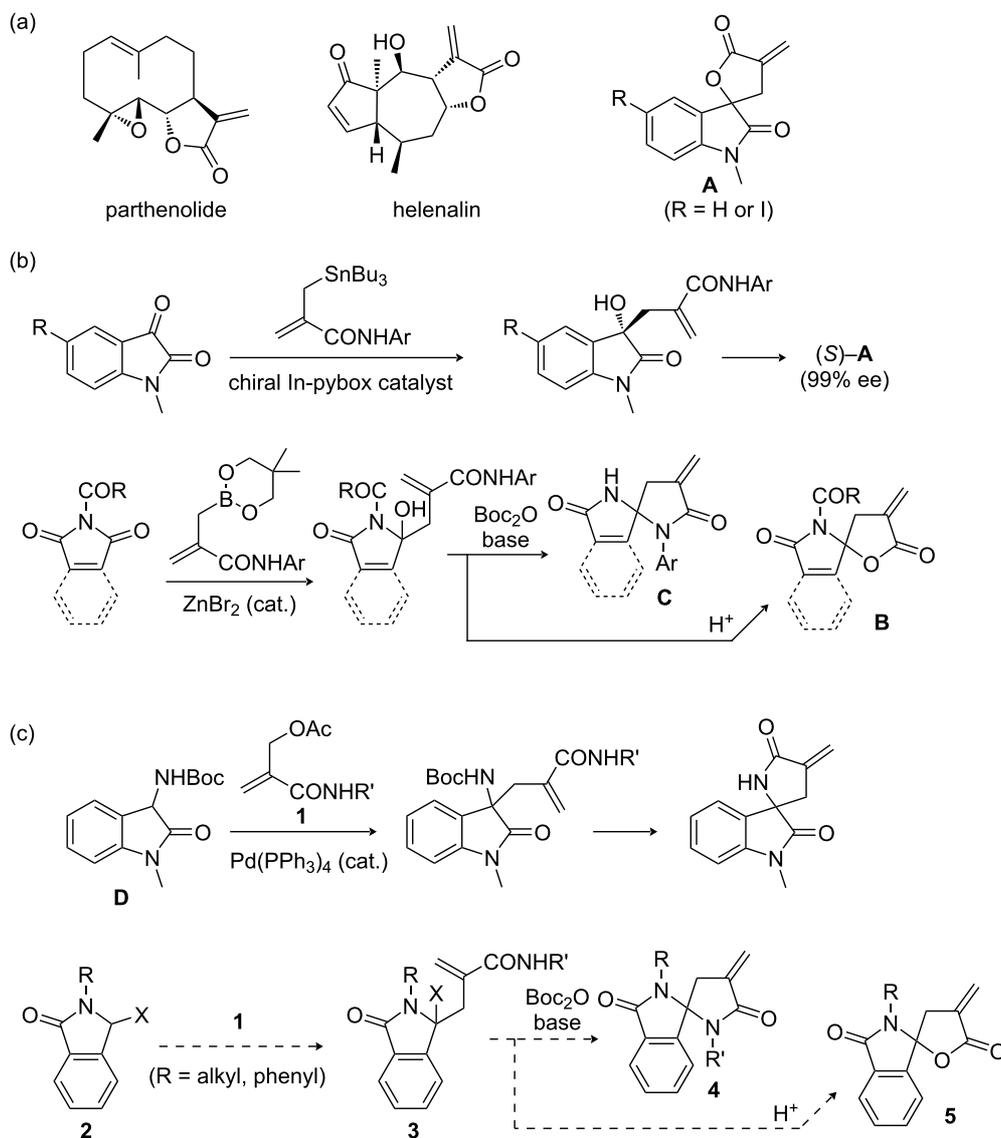
Abstract

New synthetic methods for spiro-lactams bearing an α -methylene- γ -butyrolactone or its analogous methylene-lactam have been developed. The allylation of γ -phenylthio-functionalized γ -lactams with 2-(acetoxy)methyl acrylamides was accomplished by using 2.5 equivalents of NaH to give the corresponding adducts in excellent yields. The remaining phenylthio group was substituted with a hydroxy group by treatment with CuBr, and the resulting γ -hydroxyamides were cyclized under acidic conditions to afford the corresponding methylene-lactam-fused spiro-lactams in high yields. On the other hand, methylene-lactone-fused spiro-lactams could be delivered from the allyl adducts in high yields through a sequential *N*-Boc protection/desulfinate lactonization.

Introduction

α -Methylene- γ -butyrolactone is a pharmaceutically important motif which is found in a wide variety of bioactive molecules including natural products (Scheme 1a) [1-3]. For example, sesquiterpene lactones represented by parthenolide and helinalin have attracted interest because of their useful biological activities, and many synthetic efforts have been made so far [4-6]. On the other hand, syntheses and biological evaluation of nonnatural methylene-lactones also have been reported [4,6-8]. For instance, Heindel and his co-worker synthesized methylene-

lactone derivatives spiro-fused to an oxindole (**A**), one of which exhibited a potent cytotoxic activity [9]. Our research group succeeded the enantiopure synthesis of these compounds, in which enantioselective allylation of an isatin derivative with an amido-functionalized allylstannane was employed as a key reaction (Scheme 1b) [10,11]. More recently, an amide-functionalized allyl boronate was developed as an alternative reagent for the allylstannane [12-16], which led to the syntheses of not only cytotoxic methylene-lactones spiro-fused to a lactam



Scheme 1: Examples of (a) bioactive compounds bearing an α -methylene- γ -butyrolactone structure, (b) syntheses of spirocyclic compounds through nucleophilic amide allylation, and (c) syntheses of spirocyclic compounds through electrophilic amide allylation.

ring (**B**) but also their analogous methylene-lactams (**C**) through zinc-catalyzed addition to *N*-carbonyl imides [13,14,16].

Meanwhile, we also developed an umpolung electrophilic allylation of 3-heterosubstituted oxindole **D** for the synthesis of lactam analog of **A** (Scheme 1c) [17]. The oxindole **D** readily reacted with 2-(acetoxy)methyl acrylamides **1** in the presence of a catalytic amount of $\text{Pd(PPh}_3)_4$ to give the corresponding adducts which could be delivered into an methylene-lactam-fused oxindole. On the basis of this umpolung strategy, spiro-lactams **4** and **5**, which are *N*-alkyl or *N*-phenyl-substituted analogs of **B** and **C** and unable to be obtained under the condi-

tions depicted in Scheme 1b, seems to be accessible from the common precursor **3** prepared through the reaction between **1** and 3-heterosubstituted isoindolinones **2** followed by cyclization. Considering that **2** is estimated to be less reactive compared with the oxindole derivative **D** which show excellent nucleophilicity arising from the oxindole–hydroxyindole tautomerization [18,19], this synthetic approach is undoubtedly attractive because it could complement our previous one using nucleophilic amido-functionalized allyl boronates in terms of the structural diversity as well as understanding of the structure–cytotoxicity relationship [14]. Therefore, we were motivated to develop a new synthetic methodology for *N*-alkyl and *N*-phenyl derivatives **4** and **5** starting from **2** with **1**.

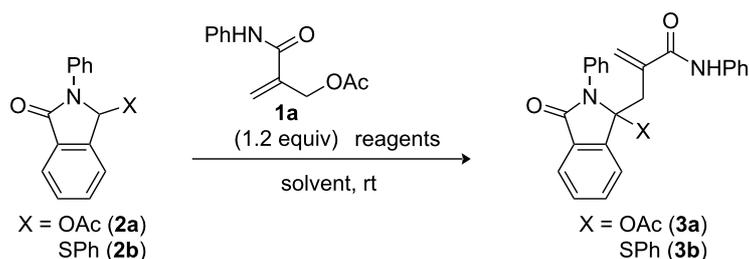
Results and Discussion

Our studies began with an exploration of a promising substrate for the desired electrophilic allylation. We initially chose *N*-phenyl-3-(acetoxyl)isoindolinone (**2a**) as a reactant. However, **2a** showed no reactivity under the reaction conditions determined for the palladium-catalyzed allylation in our previous work and was recovered in 98% yield [17] (Table 1, entry 1). The use of strong bases such as potassium *tert*-butoxide or sodium hydride resulted in no formation of the desired product **3a** because **2a** was decomposed under the harsh reaction conditions (Table 1, entries 2 and 3). Thus we opted to employ another substrate **2b** bearing a phenylthio group which polarizes the α -C–H bond to lead the corresponding stabilized anion [20] and is readily transformed into a hydroxy group by treatment with copper bromide according to our previous work [21]. Allylation of **2b** with **1a** at room temperature proceeded in the presence of a catalytic amount of Pd(PPh₃)₄ by using potassium *tert*-butoxide or sodium hydride, affording the correspond-

ing adduct in 50% and 81% yields, respectively (Table 1, entries 6 and 7). Surprisingly, the desired allylation underwent even in the absence of palladium catalyst, probably due to high nucleophilicity of the deprotonated intermediate, to give **3b** in 87% yield (Table 1, entry 8). Optimization studies were conducted by screening solvents, reagent amount, and reaction temperature, showing that **3b** was produced in the highest yield of 97% when the reaction was carried out with 2.5 equivalents of sodium hydride in THF at –10 °C (Table 1, entries 9–13) [22].

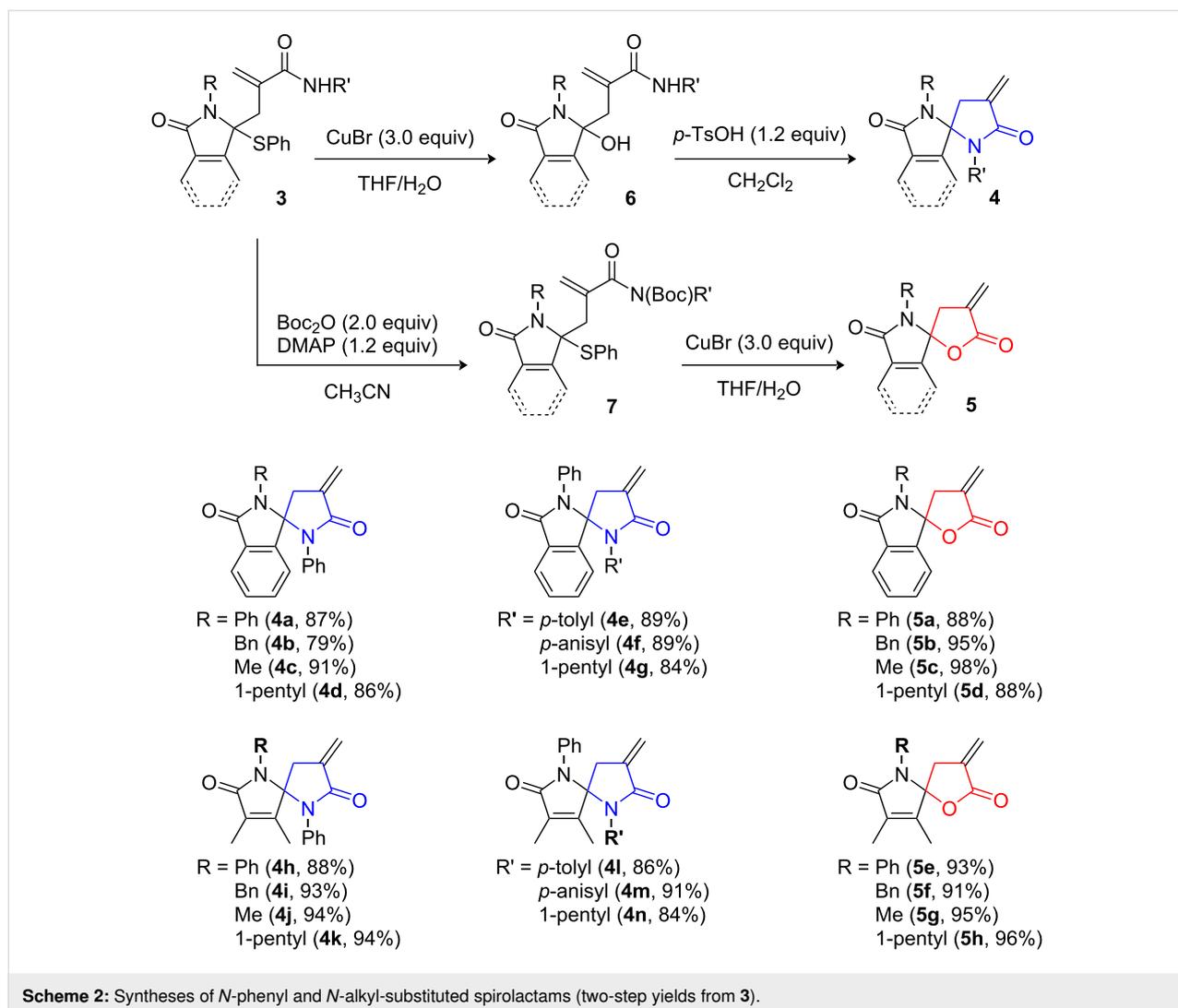
With the optimal reaction conditions for electrophilic allylation of **2b** with **1a** in hand, we turned to demonstrate the versatility of this method. In our previous work relating to nucleophilic allylation of imide derivatives, we succeeded the syntheses of *N*-carbonyl-functionalized γ -hydroxy amides bearing an amide side chain [14]. Therefore, we here examined reactions using *N*-alkyl derivatives in order to gain the structural diversity. Reactions of 3-(phenylthio)isoindolinone derivatives bearing an

Table 1: Screening of reaction conditions for electrophilic amide allylation.



entry	2	reagents (equiv)	solvent	time (h)	3 (%)
1	2a	Et ₃ N (2.5) Pd(PPh ₃) ₄ (0.1)	THF	24	3a (0)
2	2a	<i>t</i> -BuOK (2.5) Pd(PPh ₃) ₄ (0.1)	THF	24	3a (0)
3	2a	NaH (2.5) Pd(PPh ₃) ₄ (0.1)	THF	24	3a (0)
4	2b	Pyridine (2.5) Pd(PPh ₃) ₄ (0.1)	THF	24	3b (0)
5	2b	Et ₃ N (2.5) Pd(PPh ₃) ₄ (0.1)	THF	24	3b (0)
6	2b	<i>t</i> -BuOK (2.5) Pd(PPh ₃) ₄ (0.1)	THF	24	3b (50)
7	2b	NaH (2.5) Pd(PPh ₃) ₄ (0.1)	THF	1	3b (81)
8	2b	NaH (2.5)	THF	1	3b (87)
9	2b	NaH (1.0)	THF	2	3b (62)
10	2b	NaH (2.5)	toluene	1	3b (74)
11	2b	NaH (2.5)	DMF	1	3b (18)
12 ^a	2b	NaH (2.5)	THF	1	3b (97)
13 ^b	2b	NaH (2.5)	THF	1	3b (87)

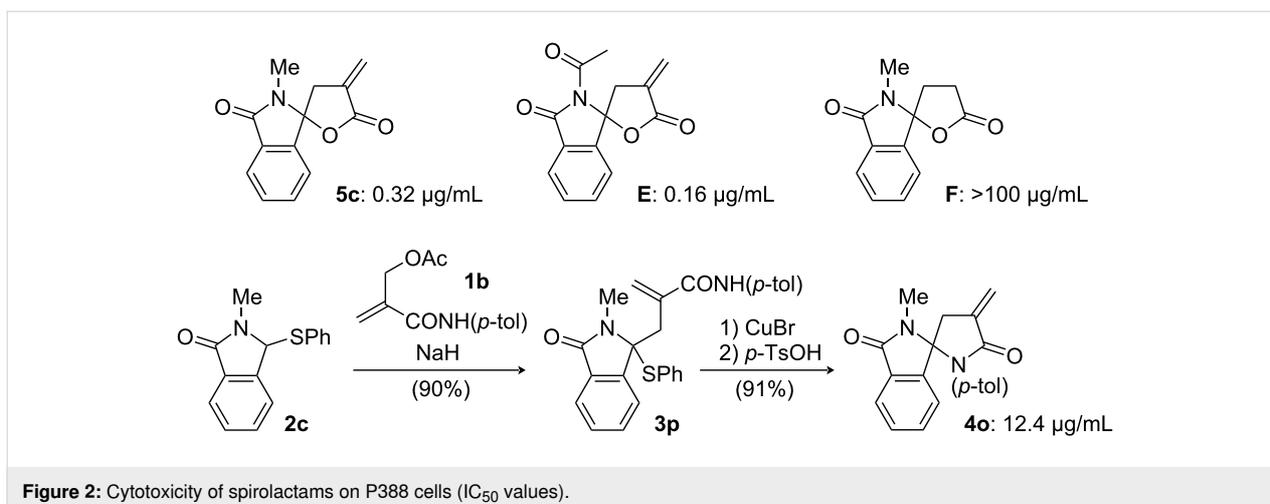
^aThe reaction was performed at –10 °C. ^bThe reaction was performed at –20 °C.



Since the cyclization under acidic conditions led to the predominant formation of the methylene-lactam-fused spirolactam, we set another route toward **5** via cyclization of *N*-Boc-functionalized γ -hydroxymethacrylamide [26]. We initially treated the hydroxylactam prepared through CuBr-mediated hydroxylation of **3b** with di-*tert*-butyl dicarbonate (Boc_2O) in the presence of *N,N*-dimethyl-4-aminopyridine (DMAP), but a mixture of structurally unidentified products was formed. Meanwhile, when the reaction was carried out with 2.0 equivalents of *n*-butyllithium and 1.1 equivalents of Boc_2O in THF at -78°C , **5a** was obtained as a mixture including unreacted starting material (<11% yield), indicating that installation of an *N*-Boc group on the terminus amide should trigger the desired lactonization. Therefore, we next examined the transposed reaction sequence, *N*-Boc protection followed by hydroxylation (Scheme 2). *N*-Boc amides were readily obtained by treatment of **3b–o** with Boc_2O and DMAP, which were successively subjected to hydroxylation in the presence of CuBr. Expectedly, the desired

lactonization occurred spontaneously under the reaction conditions and led to methylene-lactone-fused spirolactams **5a–h** in excellent yields without exception (88–98% two-step yields). Thus, we established the bifurcated synthetic routes toward lactams spiro-fused to α -methylene- γ -butyrolactone or α -methylene- γ -butyrolactam by using 3-phenylthiolactams bearing an acrylamide side chain as a common intermediate.

Finally, we subjected methylene-lactone-fused spirolactams to the assay of cytotoxicity on P388 cells (Figure 2). *N*-Methyl-substituted spirolactam **5c** exhibited potent cytotoxicity (IC_{50} 0.32 $\mu\text{g}/\text{mL}$) which was comparable to that of the *N*-acetyl analog **E** (0.16 $\mu\text{g}/\text{mL}$) [14]. It should be noted that the IC_{50} values of our recently reported non-conjugated lactone **F** [27] was >100 $\mu\text{g}/\text{mL}$. In addition to **5c**, we tested the cytotoxicity of methylene-lactam-based compound **4o**, which exhibited weak activity (IC_{50} 12.4 $\mu\text{g}/\text{mL}$) against the P388 cell line. These results suggest that cytotoxic activity of this type of spirolac-



tams against P388 cells is simply related to the methylene lactone structure [28].

Conclusion

In conclusion, we have achieved a new bifurcated syntheses of *N*-alkyl and *N*-phenyl-substituted spirolactams bearing a methylene-lactone or methylene-lactam structure. The key intermediates were synthesized by electrophilic allylation of γ -phenylthio-functionalized γ -lactam derivatives with 2-(acetoxymethyl)acrylamides in the absence of any metal catalyst and delivered into each type of spirolactams through desulfurative hydroxylation/lactamization or *N*-Boc protection/desulfurative lactonization. The increase of diversity of structure accessible led to further understanding of the structure–cytotoxicity relationship of spirolactams on P388 cells.

Supporting Information

Supporting Information File 1

Experimental procedures and characterization data.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-227-S1.pdf>]

Supporting Information File 2

Copies of 1H and ^{13}C NMR spectra of all new compounds.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-227-S2.pdf>]

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28. Cytotoxic evaluation of compound **A** also supports this suggestion, in which the comparable level of IC_{50} value (0.27 $\mu\text{g/mL}$) was obtained.

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