Preparation and in situ use of unstable N-alkyl α-diazo-γ-butyrolactams in RhII-catalyzed X–H insertion reactions

Maria Eremeyeva, Daniil Zhukovsky, Dmitry Dar’in and Mikhail Krasavin*§

Abstract

N-Alkyl α-diazo-γ-butyrolactams, previously found to be unstable and to undergo unproductive dimerization to bishydrazones, were successfully converted immediately to various X–H insertion products with alcohols, aromatic amines and thiols via an in situ RhII-catalyzed reaction. With aliphatic amines or unreactive, sterically hindered anilines, the reactions tended to yield enamine adducts.

Introduction

Recently, we described the first synthesis and subsequent transformations of a rare type of cyclic α-diazocarbonyl compounds, namely, α-diazo-γ-butyrolactams [1]. In particular, N-aryl-α-diazo-γ-butyrolactams 1 were efficiently transformed into pyrrolinones 2 upon the treatment with AgOTf (1 mol %) and into α-alkoxy derivatives 3 via Rh2(OAc)4-catalyzed O–H insertion reactions with various alcohols. In contrast, N-alkyl-α-diazo-γ-butyrolactams 4 did not undergo these reactions typical of α-diazocarbonyl compounds, as they rapidly dimerized to give bishydrazones 5 (Figure 1). The instability of N-alkyl-α-diazo-γ-butyrolactams 4 compared to the N-aryl counterparts 1, was most likely related to the reduced electron-withdrawing character of the lactam carbonyl group in the former compared to the latter. This assumption is further supported by the fact that ortho-substituted N-aryl derivatives 1 (in which the conjugation of the aromatic ring with the lone pair of the lactam nitrogen atom is reduced due to the sterically forced loss of the coplanarity between the aromatic ring and the aminocarbonyl moiety) are as unstable as the N-alkyl derivatives 4 [1].

Faced with this serious limitation of the reactivity scope, we set off to investigate the possibility of using unstable compounds 4 in situ, promptly after their formation, in various RhII-catalyzed X–H insertion reactions, particularly the recently described rhodium carbene insertion into O–H [1], N–H [2] and S–H [3] bonds of alcohols, aromatic amines, and thiols, respectively. Herein, we report the results of these studies.
Results and Discussion

Three $N$-alkyl-$\alpha$-ethoxalyl-$\gamma$-lactams 6a–c, prepared by oxalyla-
tion of the respective $\gamma$-lactams as described previously [1],
underwent a rapid diazo transfer reaction via the conventional
protocol [4,5] employing 4-nitrobenzenesulfonyl azide and
DBU. A quick filtration through a plug of alumina (in lieu of
silica gel, which led to decomposition of the diazo compounds
4a–c), and addition of an alcohol, a thiol, or an aromatic amine
along with a Rh$^{II}$ catalyst resulted in a rapid insertion reaction
and the isolation of the desired $\alpha$-substituted $\gamma$-lactams 7a–o in
modest yields (Scheme 1). It should be noted that, after some
experimentation, the reactions with alcohols and thiols were
found to be efficiently catalyzed by 1 mol % of Rh$_2$(OAc)$_4$ and
completed within 30 min. For aromatic amines, this catalyst
proved inefficient and was replaced by 0.5 mol % of Rh$_2$(esp)$_2$.
The attempted change of the catalyst to Rh$_2$(esp)$_2$ in the reac-

---

**Scheme 1:** Generation and in situ Rh$^{II}$-catalyzed $X$–H insertion reactions of the diazo compounds 4a–c. Conditions: $^a$Rh$^{II}$ catalyst: 1 mol % Rh$_2$(OAc)$_4$ for $X = O$ or $S$ and 0.5 mol % Rh$_2$(esp)$_2$ for $X = NR^\prime$. 

---

Figure 1: Previously reported uses of $\alpha$-diazo-$\gamma$-butyrolactams 1 and 4.
tions with alcohols and thiols (which earlier gave a remarkable improvement of the product yields of NH-insertion reactions [2]) resulted in no notable improvement in this case.

The only attempt to employ an aliphatic amine, cyclopropylamine (which would presumably be less reactive in the RhIII-catalyzed insertion reaction [2]) resulted in the formation of a sole identifiable product – enamine 8a, isolated chromatographically from a complex mixture of unidentified byproducts. The formation of 8a (that was also observed previously, along with the expected, saturated coupling product of the RhIII-catalyzed reaction of cyclopropylamine with N-phenyl-α-diazo-2-pyrrolidone [2]) can be rationalized, as proposed previously [2], either by the oxidation of diazolactam 6c to a respective ketone (a process described in the literature for other α-diazocarbonyl compounds [6]), followed by a nucleophilic attack of cyclopropylamine. Alternatively, the formation of the enamine product could be envisaged via the reaction of the amine with bis-hydrazone 5, which would have combined, if the N–H insertion pathway was not sufficiently rapid. Both assumptions are in line with the formation of the similar enamine coupling product 8b, that we observed with 2,6-dimethylaniline. With this unreactive, sterically hindered aromatic amine, 6c is likely to undergo either the unwanted N<sub>2</sub> → O oxidation or dimerize to bis-hydrazone 5, whereupon the resulting intermediate would be eventually trapped by the aniline to give 8b (Scheme 2). The viability of either (or both) of these possibilities is currently investigated. It should be noted that a similar Rh<sub>2</sub>(esp)<sub>2</sub>-catalyzed reaction of one of the N-aryl-α-diazo-γ-butyrolactams 1 with 2,6-dimethylaniline previously gave an excellent yield of the N–H insertion product [2].

**Conclusion**

We demonstrated that the scope of α-diazo-γ-butyrolactams being capable to undergo RhIII-catalyzed X–H insertion reactions with alcohols, thiols, and aromatic amines can be extended to unstable N-alkyl derivatives, for which rapid, unproductive dimerization was previously observed. This was achieved through the immediate addition of the X–H insertion partner and a RhIII catalyst to the solution of the diazo compound. The reactions are rapid, albeit moderately yielding. Despite the latter drawback, the range of 1,3-disubstituted 2-pyrrolidones attainable via the intermediate formation of α-diazo-γ-butyrolactams was substantially expanded, thereby making this approach more useful for potential medicinal chemistry exploration of these disubstituted γ-lactams.

**Supporting Information**

**Supporting Information File 1**

General experimental information, synthetic procedures, analytical data and NMR spectra for the reported compounds.

[https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-16-55-S1.pdf]

**Acknowledgements**

We are grateful to the Research Centre for Magnetic Resonance and the Centre for Chemical Analysis and Materials Research of Saint Petersburg State University Research Park for the analytical data.

**Funding**

This research was supported by the Russian Science Foundation (project grant 19-33-90016).

**ORCID iDs**

Dmitry Dar’in - https://orcid.org/0000-0002-0413-7413
Mikhail Krasavin - https://orcid.org/0000-0002-0200-4772

**Preprint**

A non-peer-reviewed version of this article has been previously published as a preprint doi:10.3762/bxiv.2020.15.v1

**References**


License and Terms

This is an Open Access article under the terms of the Creative Commons Attribution License (http://creativecommons.org/licenses/by/4.0). Please note that the reuse, redistribution and reproduction in particular requires that the authors and source are credited.

The license is subject to the *Beilstein Journal of Organic Chemistry* terms and conditions: (https://www.beilstein-journals.org/bjoc)

The definitive version of this article is the electronic one which can be found at: doi:10.3762/bjoc.16.55