



Cascade intramolecular Prins/Friedel–Crafts cyclization for the synthesis of 4-aryltetralin-2-ols and 5-aryltetrahydro-5*H*-benzo[7]annulen-7-ols

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Abstract

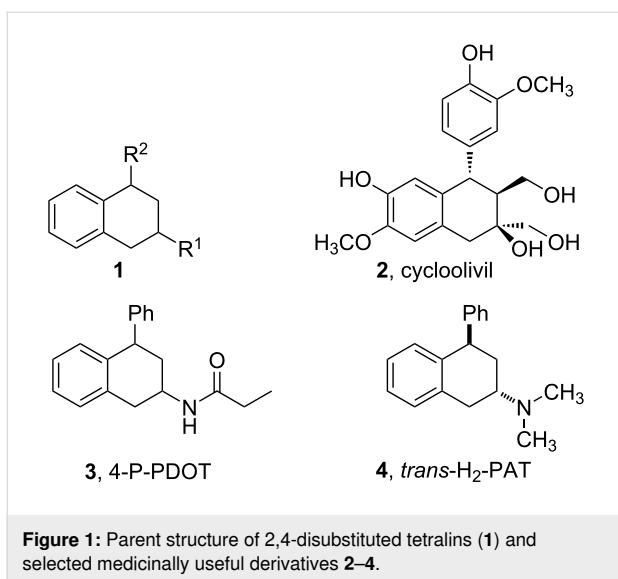
The treatment of 2-(2-vinylphenyl)acetaldehydes or 3-(2-vinylphenyl)propanals with $\text{BF}_3 \cdot \text{Et}_2\text{O}$ results in an intramolecular Prins reaction affording intermediary benzyl carbenium ions, which are then trapped by a variety of electron-rich aromatics via Friedel–Crafts alkylation. This cascade Prins/Friedel–Crafts cyclization protocol paves an expedient path to medicinally useful 4-aryltetralin-2-ol and 5-aryltetrahydro-5*H*-benzo[7]annulen-7-ol derivatives.

Introduction

2,4-Disubstituted tetralins (Figure 1, **1**), especially 2-functionalized tetralins are privileged building blocks for medicinal chemistry applications which are known to exhibit a wide spectrum of biological activities [1–3]. Some representative compounds comprising this skeleton are illustrated in Figure 1. Cyclooolivil (Figure 1, **2**) [4], which is isolated from the stem bark of *Olea europaea*, has been recognized as inhibitor of cyclic AMP dependent phosphodiesterase, can act as a Ca^{2+} antagonist, and exhibits promising anti-oxidant properties. 4-Phenyl-2-propionamidotetralin (4-P-PDOT, **3**, Figure 1) [5] is a melatonin MT_2 selective antagonist that can be used to map melatonin receptor subtypes in tissue and serves as a chemical biology tool

to identify sub-type selective analogues with therapeutic potential. In addition, *trans*-4-phenyl-*N,N*-dimethyl-2-aminotetralin (*trans*- H_2 -PAT, **4**, Figure 1) [6] has been determined to modulate tyrosine hydroxylase activity and dopamine synthesis in rodent forebrain and is also a ligand binding to histamine H_1 receptors, and thus is a potentially useful therapeutic for psychoses, addiction, and other neuropsychiatric disorders.

Although 4-substituted tetralin-2-ols and derivatives with significant pharmaceutical activities have been identified, only a limited number of synthetic methods is documented in the literature (Scheme 1) [7–9]. Moreover, these methods generally



require multiple steps, proceed in low overall yields, and have a limited ability for structural modifications to prepare analogues with new substitution patterns for enhancing activities. Consequently, it is highly desirable to develop new synthetic methods that provide efficient access to 2,4-disubstituted tetralin compounds and thus facilitate their biological investigations.

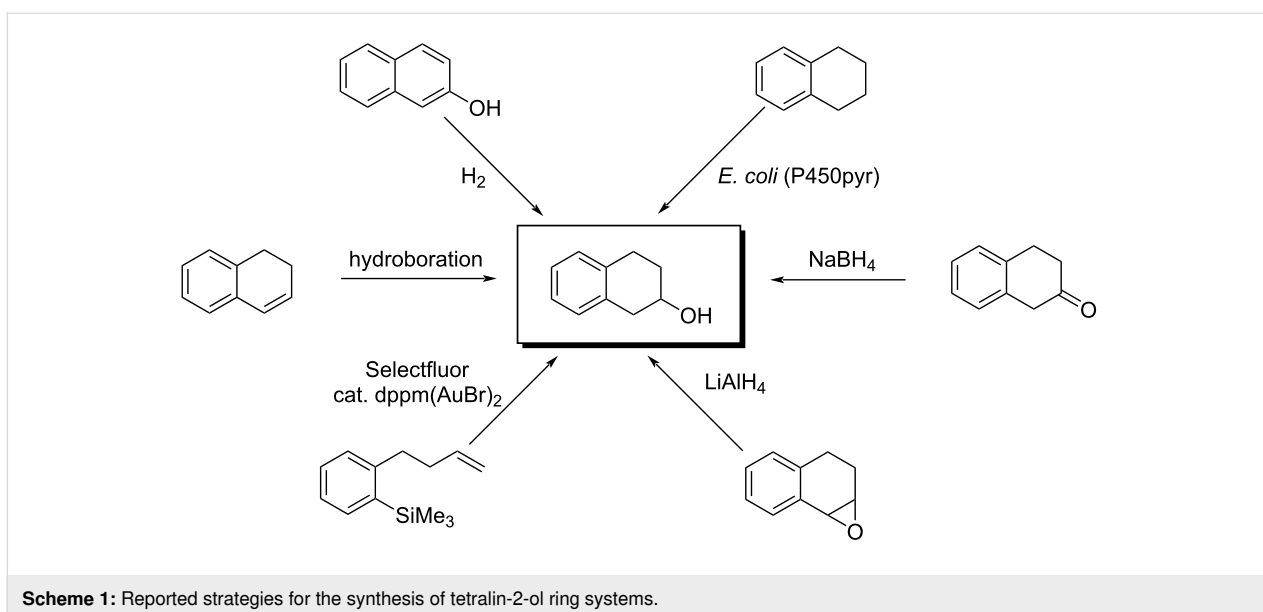
The cascade Prins/Friedel–Crafts reaction to form multiple chemical bonds in one operation has emerged as an atom-economic and straightforward strategy for the construction of oxygen-containing heterocycles [10–14]. For example, Nagumo and coworkers have developed a Prins/Friedel–Crafts cyclization of homocinnamyl alcohols with aromatic aldehydes under the action of $\text{BF}_3 \cdot \text{Et}_2\text{O}$ affording 2*H*-indeno[1,2-*b*]furan deriva-

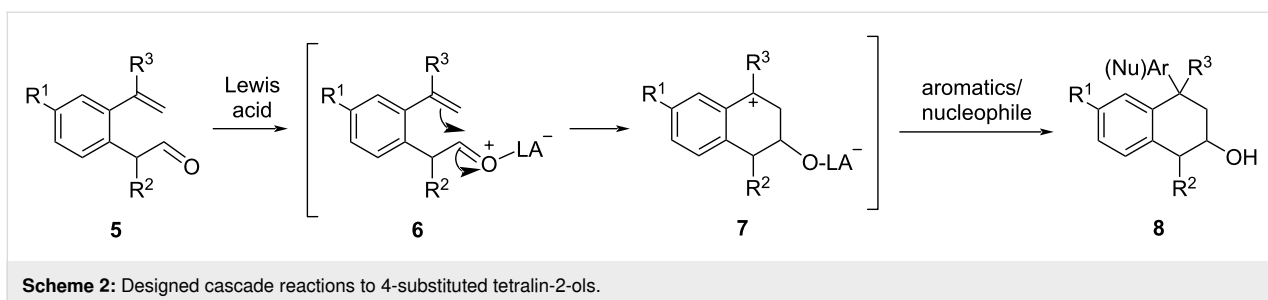
tives [15]. Likewise, Hinkle and coworkers reported in 2017 a three-step domino alkynyl-Prins cyclization, Friedel–Crafts alkylation, and dehydration/aromatization reaction between 1-aryl-3-hexyne-2,6-diol derivatives and aldehydes, that led to the formation of 1,4-dihydro-2*H*-benzo[*f*]isochromenes [16].

The Prins reaction-induced cyclization, inter alia, became a versatile tool for the assembly of complex molecules from relatively simple and inexpensive materials/reagents in a single operation. The reaction continues to be an interesting and profitable field of research with high impact on synthetic organic chemistry [17,18]. Many of the existing protocols rely on an acid-promoted condensation of a homoallylic alcohol and an aldehyde to give an oxocarbenium ion, which is then reacted with an olefinic/alkynic bond generating a carbocation that undergoes a Friedel–Crafts reaction. Given the potential value of tetralin-2-ol scaffolds to drug research programs, we decided to develop a novel Prins/Friedel–Crafts cyclization strategy for the synthesis of 4-aryl-2-hydroxytetralins starting from 2-(2-vinylphenyl)acetaldehydes (Scheme 2). In this protocol, we envisioned that the aldehyde **5** would give rise to an oxocarbenium ion species **6** upon treatment with a Lewis acid. The intermediate **6** then would undergo a Prins-type intramolecular cyclization with the olefinic bond to produce a stable benzyl carbocation **7**, that may be trapped through a Friedel–Crafts alkylation with an aromatic substrate or through the reaction with an external nucleophile to afford the target product **8**.

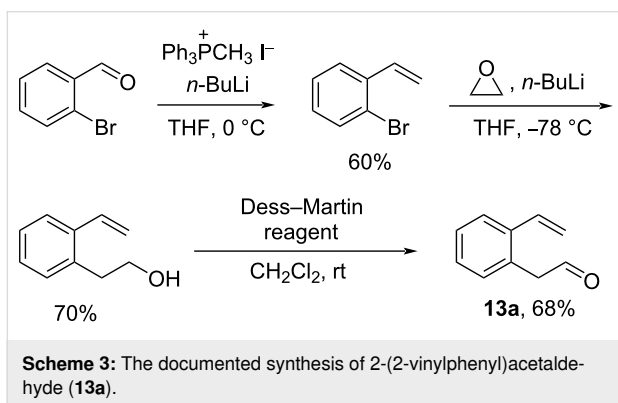
Results and Discussion

Our research began with the preparation of 2-(2-vinylphenyl)acetaldehydes (**13**) required as substrates for the Prins/Friedel–Crafts cyclization reactions. Commonly, these aromat-



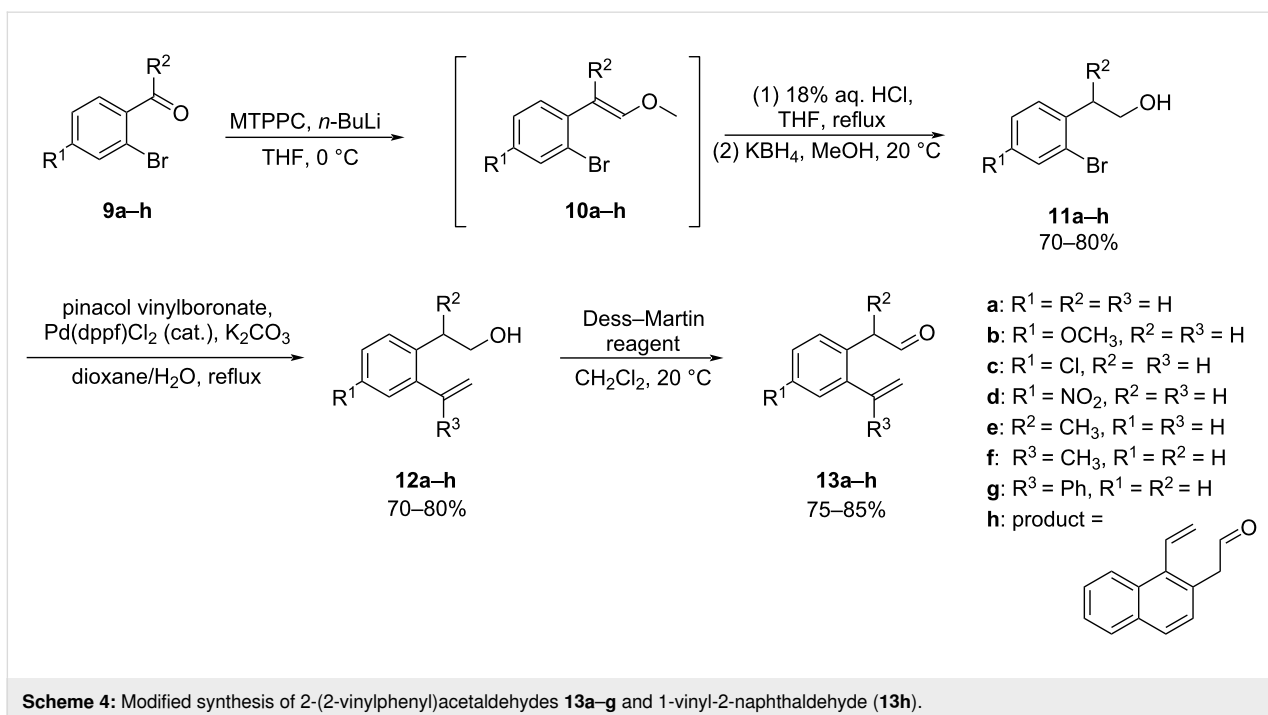


ic alkenyl aldehydes were previously prepared via a three step process as exemplified by **13a** shown in Scheme 3 consisting of the following steps: (i) Wittig reaction of 2-bromobenzaldehyde with methyltriphenylphosphonium iodide ylide, (ii) lithiation of the resultant *o*-bromostyrene with *n*-BuLi and reaction of the aryl lithium species with ethylene oxide, and (iii) oxida-



tion of the resultant primary alcohol using Dess–Martin periodinane [19,20].

The reported methods involved the use of ethylene oxide, a hazardous and carcinogenic gas. This prompted us to work out a more practical and flexible method to access the aromatic enal compounds **13**. At the offset, we examined the synthesis of 2-(2-vinylphenyl)acetaldehyde (**13a**) using the route as outlined in Scheme 4. The synthesis started with the Wittig reaction of 2-bromobenzaldehyde (**9a**) with (methoxymethyl)triphenylphosphonium chloride (MTPPC) upon action with *n*-butyllithium in THF at 0 °C to give the vinyl ether **10a** that was subjected to acidic hydrolysis using 18% aq HCl furnishing the corresponding aldehyde [21]. Without purification, the resultant aldehyde intermediate was then directly reduced using potassium borohydride to the corresponding primary alcohol **11a** in 74% yield starting from **9a**. Pd-catalyzed cross-coupling of **11a** with pinacol vinylboronate afforded the *o*-hydroxyethyl-styrene **12a** in 78% yield [22,23]. Next, Dess–Martin oxidation of the



alcohol **12a** was carried out, and the desired 2-(2-vinylphenyl)acetaldehyde (**13a**) was successfully obtained in 85% yield. Obviously, this modified method has the advantages of mild reaction conditions, operational simplicity, and using cheap and non-toxic reagents.

The modified procedure was then expanded to the synthesis of a set of 2-(2-vinylphenyl)acetaldehydes **13b–f** starting from differently substituted 2-bromobenzaldehydes **9** or 1-(2-bromophenyl)ethan-1-one (**9e**) in comparable yields. Likewise, 2-(1-vinylnaphthalen-2-yl)acetaldehyde (**13h**) was prepared from 1-bromo-2-naphthaldehyde in 48% yield over the three steps. It should be noted that the nitro-substituted intermediate **11d** was prepared by nitration of **11a** with nitric acid under the promotion of acetic anhydride.

With the accessibility of the aromatic vinyl aldehydes **13**, next the cascade Prins/Friedel–Crafts reaction was examined. We started our investigations by applying aldehyde **13a** as the model substrate (Scheme 5). A Lewis acid screening was carried out to identify the best catalyst for the tandem intramolecular Prins/Friedel–Crafts reaction (Table 1). Thus, the portion-wise addition of AlCl₃ (1.1 equiv) to a stirred mixture of **13a** (1.0 equiv) and veratrole (1.05 equiv) in CH₂Cl₂ at 0 °C resulted in the intramolecular Prins reaction to generate a benzyl carbenium ion that concurrently underwent Friedel–Crafts reaction with veratrole, leading to the formation of the expected 4-(3,4-dimethoxyphenyl)-1,2,3,4-tetrahydronaphthalen-2-ol (**14aa**), 51:49 mixture of *cis/trans*-diastereomers) as a colorless oil in 50% yield (Table 1, entry 1). The use of Et₂AlCl as the Lewis acid gave tetralin **14aa** in a slightly improved 55% yield (Table 1, entry 2). However, the reaction with AlMe₃ as the promoter resulted in a competing reduction of **13a** to 2-bromophenylethanol (**12a**) that was obtained as the major product (Table 1, entry 3). Switching to the weaker Lewis acid In(OTf)₃ failed to induce any intramolecular Prins cyclization (Table 1, entry 4), whilst the use of FeCl₃ produced **14aa** in a similar 52% yield as observed for AlCl₃ (Table 1, entry 5). To our delight, 1.1 equivalents of BF₃·Et₂O were found to promote the transformation efficiently, and a 70% isolated yield of **14aa** was obtained (Table 1, entry 6). However, experiments with BF₃·Et₂O at substoichiometric amounts afforded significantly decreased yields of **14aa** (Table 1, entries 7 and 8).

The relative *cis*- and *trans*-configuration of the C-2 hydroxy group and the C-4 aryl substituent (Figure 2) were assigned on the basis of ¹H–¹H COSY analysis. Firstly, the HSQC analysis was used to determine H₃. The ¹³C NMR chemical shift for C₂ is expected to be in the range of 60 to 70 ppm and the assignment of H₃ was based on the HSQC correlation between H₃ and C₂. Then, H₁ and H₂ could be assigned by COSY and HSQC

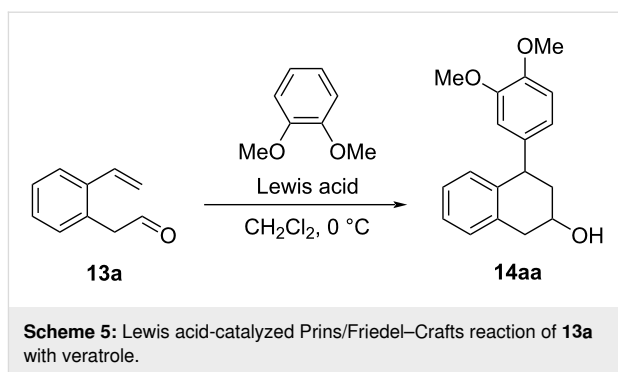
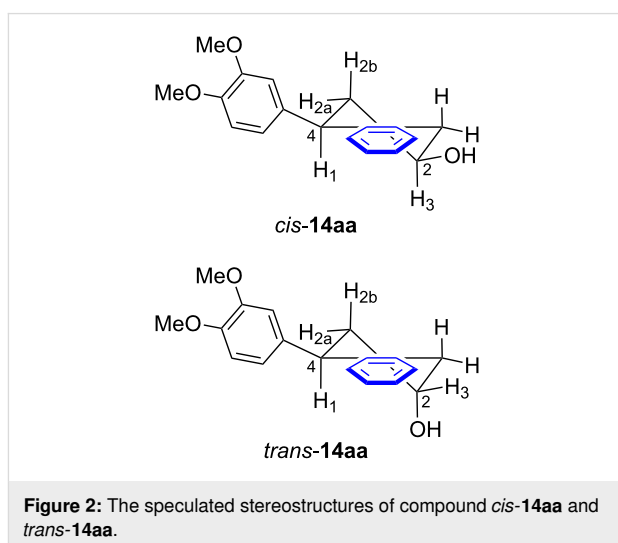


Table 1: Screening of Lewis acid catalysts.^a

entry	Lewis acid	amount of LA [mol %]	<i>cis/trans</i> ratio ^b	yield [%] ^c
1	AlCl ₃	110	51:49	50
2	Et ₂ AlCl	110	50:50	55
3	AlMe ₃	110	NA	0 ^d
4	In(OTf) ₃	110	NA	0
5	FeCl ₃	110	50:50	52
6	BF ₃ ·Et ₂ O	110	49:51	70
7	BF ₃ ·Et ₂ O	80	50:50	50
8	BF ₃ ·Et ₂ O	50	50:50	35

^aReaction conditions: a mixture of **13a** (1.40 mmol), veratrole (1.47 mmol) and Lewis acid (1.54 mmol) in CH₂Cl₂ (6 mL) was stirred at 0 °C for 2 h; ^b*cis/trans* ratios were determined by ¹H NMR spectroscopy; ^cisolated yield after chromatography; ^dreduction product **12a** instead of the desired **14aa** was identified.

experiments. Following that, NOE analysis was applied to analyze the relative *cis*- and *trans*-configuration. If there is an NOE correlation between H₁ and H₃, and meanwhile H₁ and H₃ also have a strong NOE correlation with H_{2a}, the compound is



assigned to be *cis*-configured. Otherwise, it was assigned to be the *trans*-isomer (see Supporting Information File 1 for details).

Having determined the suitable reaction conditions, we surveyed the scope and limitation of the cascade protocol. Initially, we explored the range of nucleophiles that were used to intercept the benzyl carbenium ion and the results are summarized in Scheme 6. All reactions with electron-rich aromatics containing a *p*- and/or *o*-methoxy substituent as the nucleophile proceeded well to give the desired 2-hydroxy-4-aryltetralins **14aa–ae** as 49:51 to 60:40 mixtures of *cis/trans* diastereomers in moderate to good yields.

The electron-rich 5-membered heterocycles like furans and thiophene participated also smoothly in the reaction sequence, leading to the clean formation of the respective 2-hydroxy-4-heteroaryltetralins **14af–ah**, although the yields were somewhat lower than that with substituted anisole derivatives. As an attempt to enlarge the generality, tetraallylsilane was also examined. To our delight, this substrate also participated in the reaction leading to the 4-allyl-substituted tetrahydronaphthalen-2-ol **14ai** in 65% yield.

On comparing the results from the anisole-type nucleophiles or thiophene with that from furans, it was observed that the reac-

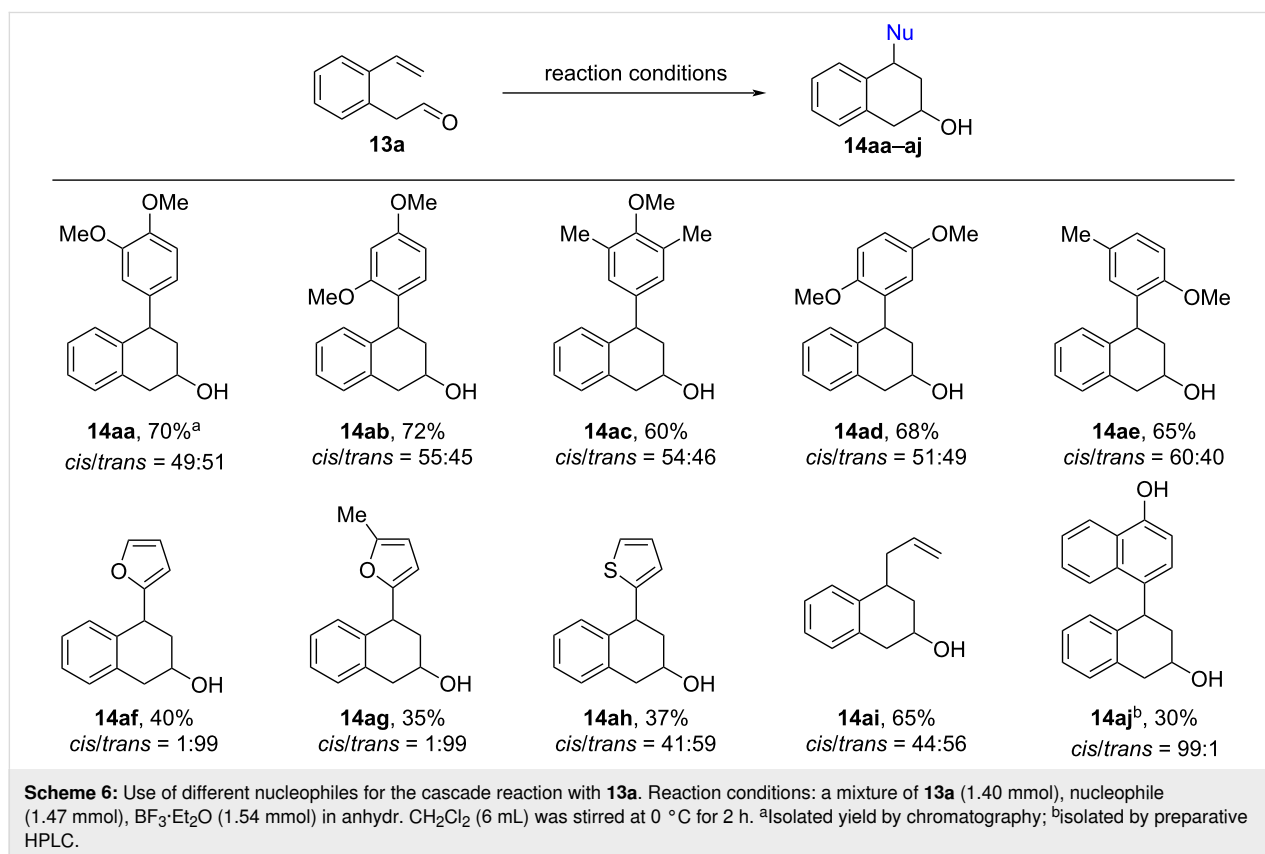
tions with furans furnished predominantly *trans*-**14af** and *trans*-**14ag** with a high degree of diastereoselectivity (*cis/trans* ratio = 1:99). The preferential formation of the *trans*-configured products for furan nucleophiles may be due to the fact that the addition of furan is reversible leading to equilibration to the more stable *trans* product. To test this hypothesis, we monitored the reaction by HPLC (Table 2). As expected, we observed that the initially formed *cis*-isomer of **14af** turned gradually to *trans*-**14af** and finally reached 1:99 after 2 hours (for further details, see Supporting Information File 1).

To further expand the substitution pattern, we then tried the reaction of **13a** with allylsilane as a carbon-nucleophile. As ex-

Table 2: Dependency of *cis/trans* ratio of product **14af** on conditions and time.

entry	conditions/reaction time	<i>cis/trans</i> ratio ^a
1	addition 20% of BF ₃ ·Et ₂ O	29:71
2	addition 100% of BF ₃ ·Et ₂ O	23:77
3	further stirred for 30 min	21:79
4	further stirred for 60 min	12:88
5	further stirred for 120 min	1:99

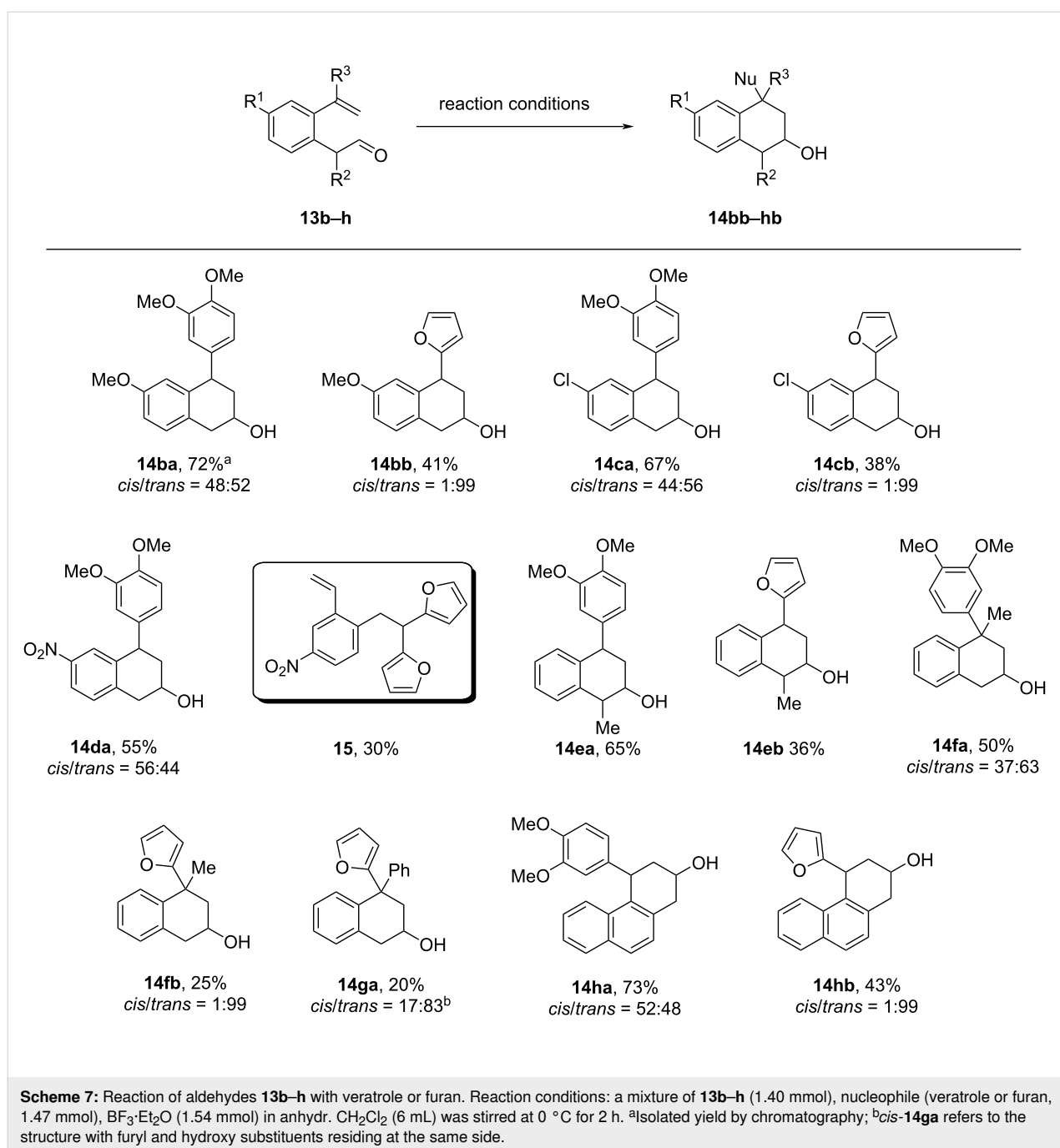
^a*cis/trans* ratio was examined by HPLC.



pected, the 4-allyl-substituted tetrahydronaphthalen-2-ol **14ai** was obtained, again, as a mixture of *cis/trans*-isomers in a ratio of 44:56. This example demonstrates the general synthetic utility of this cascade protocol.

Encouraged by the success of using **13a** as the substrate, the reactions with other 2-(2-vinylphenyl)ethanals **13b–g** carrying different substituents on the benzene ring or on the side chain with veratrole and furan as the nucleophiles were investigated. As can be seen from Scheme 7, under comparable conditions,

most reactions proceeded smoothly with the attempted alkenyl-aldehydes **13** to furnish the corresponding 2,4-disubstituted tetralins **14ba–hb** in acceptable to good isolated yields. For instance, the reaction with aldehydes **13** containing π -donating substituents like methoxy and chloro substituents afforded the 2-hydroxy-4-aryltetralin products **14ba–cb** in 38–72% yield. To our gladness, aldehyde **13d**, with an electron-deficient nitro group residing on the benzene ring reacted with veratrole under the standard conditions, delivering tetralin **14da** in 55% yield. However, using furan as the nucleophile component, the reac-



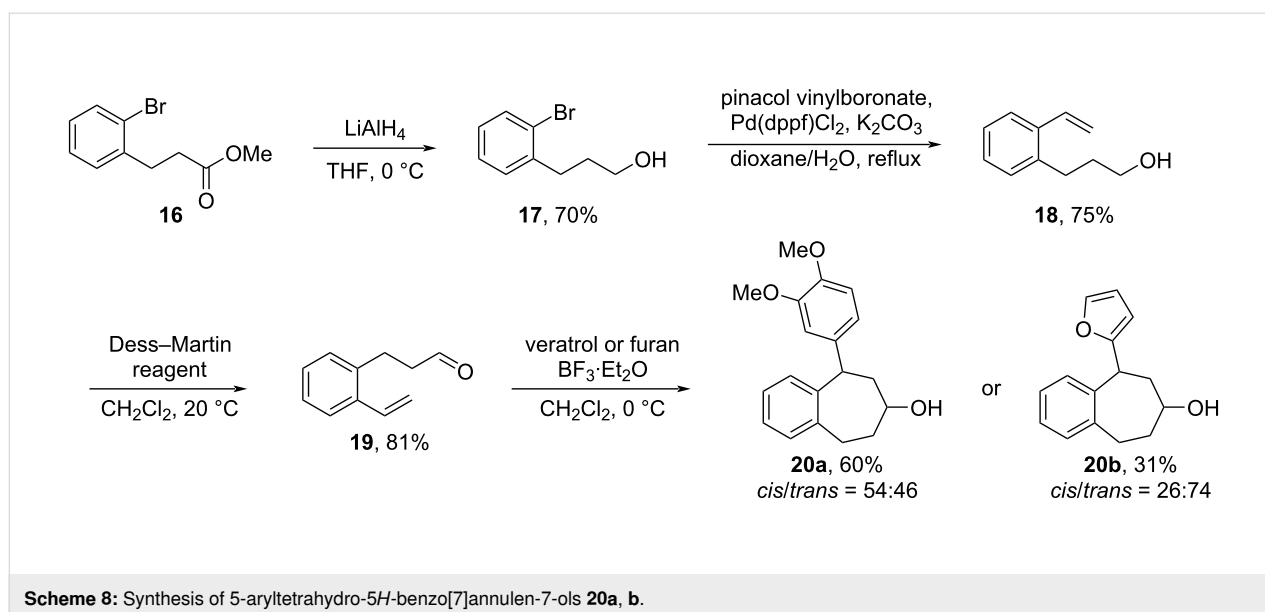
tion sequence with **13d** failed to give the tetralin product. Instead, we only isolated 30% yield of the difuranyl-substituted compound **15** as the major product.

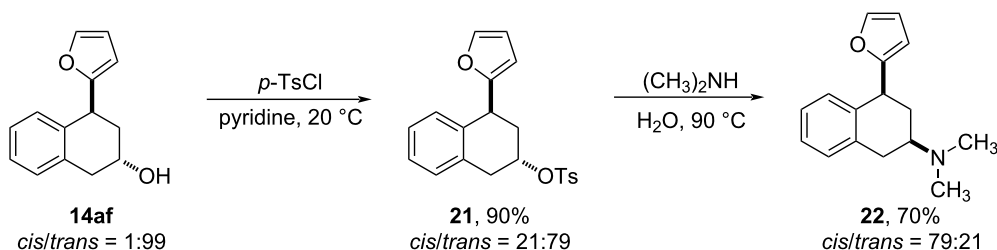
In addition, aldehydes **13e** or **13f** bearing a methyl group at the acetaldehyde side or the benzylic position of the alkene side were also suitable substrates for this cascade strategy: the 1,2,4-trisubstituted tetralins **14ea** and **14eb** as well as the 2,4,4-trisubstituted tetralins **14fa** and **14fb** were obtained in moderate to reasonable yields. The aldehyde **13g** bearing a phenyl group at the benzylic position of the alkene side was also tried. Under the standard conditions, the 1,2,4-trisubstituted tetralin **14ga** was isolated as a 17:83 mixture of *cis/trans* isomers, but with 20% yield. The poor yield may be attributed to the enhanced steric hindrance. This cyclization methodology was also applicable to 2-(1-vinylnaphthalen-2-yl)acetaldehyde (**13h**), for which the reaction with veratrole or furan led to the formation of the respective tricyclic 4-aryl-1,2,3,4-tetrahydrophenanthren-2-ols **14ha** and **14hb** in 73% and 43% yields, respectively.

In order to further explore the generality of this cascade Prins/Friedel–Crafts cyclization, the established methodology was also applied to the formation of tetrahydro-5*H*-benzocyclohepten-7-ol ring systems. As shown in Scheme 8, the required homo-aldehyde substrate **19** was prepared starting from methyl 3-(2-bromophenyl)propionate (**16**) analogously as for **13**. Reduction of the ester **16** with LiAlH₄ in THF at 0 °C afforded the alcohol **17** that was subjected to a Suzuki reaction with pinacol vinylboronate using Pd(dppf)Cl₂ as catalyst to produce 3-(2-vinylphenyl)propan-1-ol (**18**). The oxidation of alcohol **18** with Dess–Martin oxidizing reagent furnished the requisite aldehyde **19** in 43% yield over the three steps.

Under the standard conditions, aldehyde **19** underwent satisfactorily the cascade Prins/Friedel–Crafts cyclization with veratrole or furan as the nucleophile furnishing the tetrahydro-5*H*-benzo[7]annulen-7-ol **20a** (*cis/trans* ratio = 54:46) and **20b** (*cis/trans* ratio = 26:74) in 60% and 31% yield, respectively. The predominance of the *trans*-product for the reaction with furan further verified the oxophilic character of the employed BF₃, although the stereoselectivity considerably decreased in comparison with the formation of tetralin ring system as the distance between the reaction sites is increased. It is worth mentioning that the tetrahydro-5*H*-benzo[7]annulen-7-ol skeleton is also of considerable medicinal significance and has attracted much synthetic efforts [24,25].

Finally, the ability to structurally diversify the 2-hydroxy-4-substituted tetralin skeletons into medicinally useful derivatives was demonstrated by converting 2-hydroxy-4-furyl-tetralin **14af** into the PAT analogue **22** (see Figure 1) [26]. The reaction of **14af** with *p*-toluenesulfonyl chloride in pyridine afforded the tosylate **21** in 90% yield, which was then treated with 40% aqueous dimethylamine to produce the tertiary amine containing PAT analogue **22** (*cis/trans* ratio = 79:21) in 70% yield (Scheme 9). With regard to the partial epimerization of product **21**, it may be due to the action of pyridine. In the preparation of compound **21**, pyridine is used both as solvent and the acid acceptor. Because pyridine itself can show nucleophilic reactivity in addition to basicity, the long reaction time of 20 hours may lead to an ion-pair species with **21** and hence erode the stereochemistry. To prove this idea, we performed the reaction with CH₂Cl₂ as the solvent in the presence of 5.0 equivalents of pyridine and 2.0 equivalents of TsCl. Under these conditions, the tosylate **21** was obtained with full retention of the expected





Scheme 9: Conversion of 2-hydroxy-4-(2-furyl)tetralin (**14af**) into PAT analogue **22**.

stereochemistry (*cis/trans* = 1:99) (see Supporting Information File 1 for details). The conversion of tosylate **21** to product **22** proceeded in a typical $\text{S}_{\text{N}}2$ manner resulting in the expected inversion of the configuration.

To unequivocally support the configuration assignment made by NMR analysis, the sulfonate derivative **21** from **14af** was prepared by reaction with tosyl chloride. For compound **21**, we were able to obtain single crystals suitable for X-ray analysis and the X-ray diffraction studies on **21** confirmed undoubtedly its *trans*-configuration. The ORTEP structure is shown in Figure 3 [27].

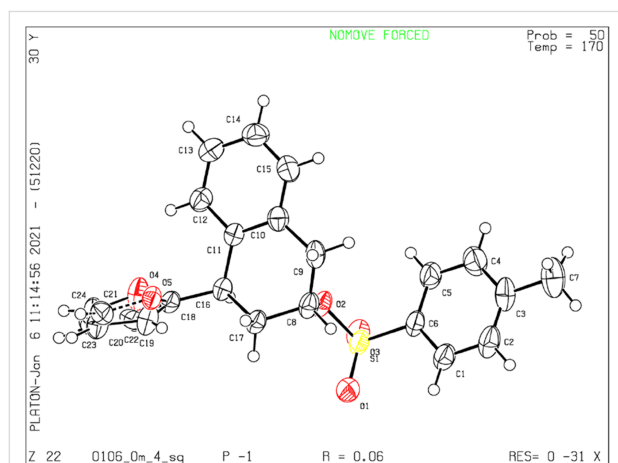


Figure 3: Crystal structure of the tosylate **21**. The displacement ellipsoids are drawn at the 30% probability level.

Conclusion

In summary, a Prins cyclization and Friedel–Crafts cascade reaction strategy for the synthesis of 4-aryl-tetralin-2-ols and 5-aryl-tetrahydro-5*H*-benzo[7]annulen-7-ols has been established. The sequence involved the Prins cyclization of 2-(2-vinylphenyl)acetaldehydes or 3-(2-vinylphenyl)propanal by action with BF_3 to generate benzyl carbenium ions that are captured by a Friedel–Crafts alkylation reaction with a range of electron-rich benzenes or heteroaromatics. The method has a relatively broad applicability allowing variation in the benzene

ring as well as in the side chain. The further manipulation of the hydroxy group affording the PAT analogue demonstrated the synthetic potential for accessing medicinally useful derivatives.

Supporting Information

The Supporting Information contains experimental procedures, characterization data of all isolated products as well as copies of NMR spectra and XRPD data for compound **21**.

Supporting Information File 1

Experimental section.

[<https://www.beilstein-journals.org/bjoc/content/supplementary/1860-5397-17-104-S1.pdf>]

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Preprint

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27. CCDC 2060394 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from the Cambridge Crystallographic Data Centre via http://www.ccdc.cam.ac.uk/data_request/cif.

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