

Decarboxylative 1,3-dipolar cycloaddition of amino acids for the synthesis of heterocyclic compounds

Xiaofeng Zhang^{*1,2}, Xiaoming Ma^{*3} and Wei Zhang^{*1}



Abstract

semi-stabilized azomethine ylide

The [3 + 2] cycloadditions of stabilized azomethine ylides (AMYs) derived from amino esters are well-established. However, the reactions of semi-stabilized AMYs generated from decarboxylative condensation of α -amino acids with arylaldehydes are much less explored. The [3 + 2] adducts of α -amino acids could be used for a second [3 + 2] cycloaddition as well as for other post-condensation modifications. This article highlights our recent work on the development of α -amino acid-based [3 + 2] cycloaddition reactions of *N*–H-type AMYs in multicomponent, one-pot, and stepwise reactions for the synthesis of diverse heterocycles related to some bioactive compounds and natural products.

Introduction

The 1,3-dipolar cycloaddition of azomethine ylides (AMYs) [1-6] is a powerful method for the synthesis of bioactive pyrrolidine-containing compounds and natural product analogs [7-15]. AMYs generated from the reaction of aldehydes and α -amino esters (via dehydration) or α -amino acids (via decarboxylation) could be classified based on the substitution groups on the N atom to: 1) N-substituted (*N*–R type), 2) hydrogen containing (*N*–H type), and 3) metal complexes (*N*–M type) (Figure 1) [16,17]. These AMYs could also be classified as stabilized (A1–A4) which contain an electron-withdrawing group (EWG), semi-stabilized (B1–B4) which have an aryl (Ar) substituent, and non-stabilized (C1 and C2) which have neither an Ar group nor an EWG on the α -carbon atoms.

The routes to access AMYs of different classes are shown in Scheme 1: A1-type AMYs can be generated from the condensa-



tion of aldehydes with α - and *N*-dialkylglycine esters, **A2**-type AMYs are derived from α -alkylglycine esters, **A3**-type AMYs are derived from *N*-alkylglycine esters, and **A4**-type AMYs are derived from glycine esters. Stabilized zwitterions **A1**–**A4** have the anionic charge on the α -carbon connecting to the EWG. They are popular AMYs for 1,3-diploar [3 + 2] cycloaddition reactions with alkenes to generate pyrrolidines **1a–d** with high regio- and stereoselectivities. They have been reported in a huge number (1,000+) of publications [18-28]. It is worth noting that among the products **1a–d**, only compound **1d** has hydrogen atoms on both the nitrogen and α -carbon atoms, which makes it suitable to be used for a second cycloaddition to form double [3 + 2] cycloaddition products **2** [29,30].

The *N*–R-type AMYs **B1** and **B2** bearing an Ar group on the α -carbon atom are semi-stabilized (Scheme 1) [16]. The **B1**-type AMYs can be generated from the decarboxylative condensation of aldehydes with α - and *N*-dialkylglycines or from cyclic amino acids (such as proline) [31-33], while AMYs of type **B2** are accessible through the decarboxylative condensation of *N*-dialkylglycines [34-51]. The *N*–H-type semi-stabilized AMYs **B3** are generated through decarboxylative condensation of arylaldehydes with α -alkylglycines, while **B4**-type AMYs are derived from the reaction of glycine [52-58]. The

[3 + 2] cycloadditions of AMYs **B1–B4** with alkenes lead to the formation of cycloaddition products **3a–d** with attenuated regioand stereoselectivity, since the Ar group is not strong enough to fully localize the negative charge on the carbon connecting to Ar in the 1,3-dipoles. Both products **3c** and **3d** can be used for a second cycloaddition to form products **4a** and **4b**. The nonstabilized AMYs **C1** and **C2** have neither an EWG nor an Ar group to localize the negative charge. The 1,3-dipolar cycloadditions of C-type AMYs lead to the formation of [3 + 2] adducts **5** or **6** with low regio- and stereoselectivity which limits the synthetic utility of non-stabilized AMYs of type C.

There are over 300 papers on the amino acid-based decarboxylative [3 + 2] cycloadditions of *N*–R-type AMYs **B1** (such as that derived from proline) and **B2** [31-51]. However, to the best of our knowledge, there are only few examples on the reactions of *N*–H-type semi-stabilized AMYs **B3** or **B4** which were either derived from special carbonyl compounds (such as isatin) [52-55] or the AMYs were reacted with uncommon alkenes as the 1,3-dipolarophiles (such as C_{60}/C_{70} fullerenes) [56-58].

Other than amino esters and amino acids shown in Scheme 1, cyclic amines can also react with arylaldehydes to form **B1**-type semi-stabilized AMYs. In this context, the Seidel group re-



Scheme 1: Aminoester- and amino acid-based AMYs for single and double [3+2] cycloadditions.

ported the reactions of pyrrolidines **5** with arylaldehydes for the formation of AMYs **B1** which then were reacted with nucleophiles to form C–H-functionalized pyrrolidines or subjected to the 1,3-dipolar cycloaddition with olefins to afford bicyclic compounds (Scheme 2A and B) [59,60]. We employed cyclic amines for the synthesis of spirooxindole-pyrrolidines **7a** or **7b** in good stereoselectivity (Scheme 3) [61,62].

From the results shown in Scheme 2 and Scheme 3, we envisioned that pyrrolidines **3c** or **3d** generated from the cycloaddition of AMYs **B3** or **B4** could undergo a second cycloaddition to form double cycloaddition products **4a** or **4b** (Scheme 4). The double cycloaddition process involves two kinds of AMYs, with the first ones (*N*-H-type **B3** or **B4**) derived from amino acids, while the second ones (*N*-R-type **B1**) derived from pyrrolidines **3c** or **3d**. It is worth noting that the double cycloaddition reaction is a pseudo-five-component reaction of amino acids with two equivalents each of aldehydes and alkenes. The first cycloaddition products **3c** or **3d** can also be used as intermediates for other transformations to synthesize novel heterocyclic rings via multicomponent, one-pot, and stepwise synthesis [63,64].

Presented in the following sections is our work on the development of amino acid-based decarboxylative [3 + 2] cycloadditions of *N*–H-type AMYs **B3** and **B4** for double cycloadditions. The stereochemistry of the cycloadditions and the combination of the cycloaddition with other transformations to be one-pot or stepwise reactions are also presented.

Perspective

One-step synthesis of trifluoromethylated pyrrolidines

As mentioned above, the unexpected double cycloaddition and low stereoselectivity are the major challenges for [3 + 2] cyclo-







addition reactions of semi- and non-stabilized AMYs derived from the condensation of amino acids with aldehydes. However, the reactions of amino acids with ketones can result in a different kind of AMYs to address the issue. The reaction of trifluoromethyl ketones with glycine or α -substituted amino acids generated stabilized AMY **8** which underwent cycloaddition with maleimides to give 2-CF₃-substituted pyrrolidines **9** in 50–76% yield (Scheme 5) [65]. Both the Ar and CF₃ groups can



Scheme 5: [3 + 2] Cycloaddition for the synthesis of trifluoromethylated pyrrolidines 9.

localize the negative charge and also provide steric effects to afford stereoselective cycloaddition products with 3:1 to 6:1 dr. The steric hindrance also prevents products **9** from undergoing a second cycloaddition. The control reactions of methyl ketone or benzaldehydes gave much lower yields and stereoselectivity because of the lacking CF₃ group. This was the first example of synthesizing 2-CF₃-substituted pyrrolidines via decarboxylative [3 + 2] cycloaddition which is more efficient than multistep and metal-assisted syntheses reported in the literature [66,67].

Pseudo-five-component double cycloadditions for polycyclic pyrrolizidines

With the success of the three-component [3 + 2] cycloadditions shown in Scheme 5, we then explored the double cycloaddition reactions proposed in Scheme 4. The reaction has synthetic significance since the resulting pyrrolizidine scaffold can be found in many biologically active compounds and natural products such as 1-epiaustraline, hyacinthacine A1, (-)-isoretronecanol, and (-)-supinidine (Figure 2) [68,69].

After the method development work, a pseudo-five-component double cycloaddition reaction of glycine with two equivalents each of arylaldehydes and *N*-substituted maleimides was carried out in EtOH as a protic solvent at 90 °C for 3 h to afford pyrrolizidines **10** in 73–93% yield with greater than 9:1 dr (Scheme 6). The scope of the reaction could be readily extended for α -substituted amino acids, such as alanine, leucine, serine, and norvaline to give products **11a–f** in 53–88% yields with greater than 8.5 dr (Scheme 7). The reactions with leucine and phenylglycine (R² = iPr and Ph) as amino acids gave mainly mono-cycloaddition products and very little double cycloaddition products **11g** and **11h** due to the steric hindrance of the R² group.

The stereochemistry of products **10** and **11** was confirmed by X-ray crystal structure and the ¹H NMR analysis of both the major and minor diastereomers [69]. The first cycloaddition gives adducts **12** and **12'** as a diastereomeric mixture. At the second cycloaddition, both major and minor adducts from the

first cycloaddition generate the same products 10 or 11 (Scheme 8).

We also evaluated the double cycloadditions in two operational steps by using two different sets of aldehydes and maleimides to afford products **13a–d** in 45–60% yields with 2:1 to 3:1 dr (Scheme 9). The low diastereoselectivity is caused by the different $R^2/R^{2'}$ and $R^3/R^{3'}$ groups which no longer have the same stereochemistry as that shown in Scheme 8.

Double cycloadditions for bis[spirooxindolepyrrolizidine] compounds

After completing the pseudo-five-component double cycloaddition reactions leading to polycyclic pyrrolizidines shown in Scheme 6 and Scheme 7, we then conducted similar reactions in order to synthesize spirooxindole-pyrrolidines. This unique ring skeleton exists in some natural products and biologically active compounds such as (–)-horsfiline, (+)-alstonisine, pteropodine and spirotryprostatin A (Figure 3) [70].

We expected that using olefinic oxindoles 14 as alkenes for the [3 + 2] cycloaddition could afford spirooxindole-pyrrolizidines. The method development revealed that recyclable zeolite HY acid is a good catalyst for the cycloaddition [70]. Thus, the zeolite HY-catalyzed reaction of glycine with two equiv each of arylaldehydes and olefinic oxindoles 14 in EtOH at 90 °C for 6 h gave bis[spirooxindole-pyrrolizidine] compounds 15a-g in 60-73% yields with up to 6:1 dr (Scheme 10). It is worth noting that this pseudo-five-component reaction gives butterfly-shaped molecules which have a plane of symmetry. The stereochemistry of the products was confirmed by X-ray crystal structure and NMR analysis. The reaction mechanism shown in Scheme 11 suggests that a semi-stabilized AMY 16 generated from the reaction of glycine and arylaldehydes undergoes a [3 + 2] cycloaddition with 14a via the favorable endo-transition state A to give spirooxindole-pyrrolizidine 17 which spontaneously reacts with another equiv of arylaldehyde to form ylide 18 in the presence of zeolite HY. The second [3 + 2] cycloaddition of 18 with 14a affords product 15a as a major product through an endocycloaddition and 15a' as a minor diastereomeric product through an exo-cycloaddition.





Scheme 6: Double cycloadditions with glycine for the synthesis of products 10 (dr > 9:1).

One-pot synthesis of triazolobenzodiazepines

Other than the multicomponent double cycloaddition reactions shown in the last section, we also utilized the first cycloaddition products for post-condensation reactions to generate new heterocyclic scaffolds. α -Substituted amino acids, such as 2-aminoisobutyric acid, could be used to block the second cycloaddition. Shown in Scheme 12 is a method development for the stepwise synthesis of triazolobenzodiazepines. The reaction of 2-azidobenzaldehyde, 2-aminoisobutyric acid and *N*-ethylmaleimide in MeCN under the catalysis of AcOH at 110 °C for 6 h afforded the monocycloaddition product **19a** in 93% LC yield [71]. The isolated compound **19a** was used for an *N*-propargylation to produce compound **20a** in 94% LC yield. The following Cu-catalyzed click reaction afforded triazolobenzodiazepine **21a** in 88% LC yield (Scheme 12).

Our next goal was to convert the stepwise reaction process into a one-pot synthesis. After optimizing the reaction conditions, a one-pot two-step reaction was developed by the reaction of 2-azidobenzaldehydes, 2-substituted amino acids and maleimides with AcOH as a catalyst in MeCN at 110 °C for 6 h to afford the monocycloaddition compounds. Without isolation, the reaction mixtures were then used for the *N*-propargylation in the presence of K₂CO₃ under microwave heating at 110 °C for 1 h to give triazolobenzodiazepines **21a**–**f** in 35–65% yields with 2:1 to 7:1 dr (Scheme 13). Other than 2-aminoisobutyric acid, phenylglycine and valine with Ph or iPr groups could also



Scheme 7: Double cycloadditions with α -substituted amino acids leading to products 11 (\approx 8.5:1 dr).



Scheme 8: Stereochemistry for the formation of products 10 or 11.

be used for the synthesis of the monocycloaddition products for the post-condensation reactions. It is worth noting that in the one-pot synthesis involving an intramolecular click reaction, no Cu catalyst was used. A similar reaction sequence using stabilized AMYs was also reported from our lab [72]. The triazolobenzodiazepines obtained through this highly efficient one-pot synthesis have structure similarity with some drug molecules shown in Figure 4 [71].

One-pot synthesis of pyrroloquinazolines and pyrrolobenzodiazepines

We developed a 2-azidobenzladehyde-based reaction sequence including a one-pot [3 + 2] cycloaddition, *N*-acylation and Staudinger/aza-Wittig reactions for the construction of pyrroloquinazolines and pyrrolobenzodiazepines [73]. The AcOH-catalyzed reaction of 2-azidobenzaldehydes, α -substituted amino acids and maleimides in MeCN at 110 °C for 6 h afforded the



Scheme 9: One-pot and stepwise double cycloadditions. Conditions: i) MeCN (0.02 M), 90 °C, 6 h; ii) then AcOH (0.5 equiv), 125 °C, 12 h.



corresponding monocycloaddition compounds followed by acylation to yield intermediates **22**. The subsequent sequential Staudinger/aza-Wittig reaction of intermediates **22** gave products **23a–g** in 48–75% yields with 5:1 to 6:1 dr (Scheme 14). This one-pot reaction could also be applied for the synthesis of pyrrolobenzodiazepines when using 2-bromoketones instead of the acid chlorides affording products **24a–g** in 59–77% yields with 3:1 to 6:1 dr (Scheme 15). The pyrroloquinazolines and pyrrolobenzodiazepines made by this route have structure similarity with bioactive compounds and natural products such as PB1-5 [74], lixivaptan, and (+)-anthramycin (Figure 5) [73].

Stepwise synthesis of pyrrolo[2,1-*a*]isoquinolines

A stepwise synthesis involving [3 + 2] cycloaddition, *N*-allylation and Heck reactions has been developed for the synthesis of pyrrolo[2,1-a]isoquinolines. The reaction of 2-bromobenzalde-

hydes, 2-aminoisobutyric acid, and maleimides in MeCN under the catalysis of AcOH at 110 °C for 6 h afforded the cycloaddition products **26**. The purified intermediates were used for the one-pot *N*-allylation with allyl bromide to afford intermediate **25** followed by a Pd-catalyzed Heck reaction to give products **26** in 65–78% yields (Scheme 16) [75]. The pyrrolo[2,1*a*]isoquinoline core installed by this route can be found in some natural products and synthetic compounds with antitumor, antibacterial, antiviral, antioxidizing, and other biological activities (Figure 6) [75].

One-pot double annulations for the synthesis of tetrahydropyrrolothiazoles

The unique tetrahydropyrrolothiazole and spiro[indole-tetrahydropyrrolothiazole] scaffolds are found in bioactive compounds such as those shown in Figure 7 [76,77]. Using cysteine as a key reactant, we developed a pseudo-four-component reaction



Scheme 10: Double cycloadditions for the synthesis of bis[spirooxindole-pyrrolizidine]s.















Scheme 15: One-pot synthesis of tetrahydropyrrolobenzodiazepines.



R² R Br Pd(OAc)₂, PPh₃ AcOH K₂CO₃ K₂CO₃, NaOAc 110 °C. 6 h 105 °C, 4 h MeCN, 105 °C, 3 h O₂H MeCN MeCN Ŕ 26 25 Et Bn Ph **26a**, 75% 26b, 78% 26c, 72% 26d, 73% Et Bn MeÓ MeÓ CF3 26e, 70% 26f, 71% **26g**, 68% 26h, 72%

Scheme 16: Stepwise synthesis of pyrrolo[2,1-a]isoquinolines.



Figure 6: Bioactive pyrrolo[2,1-a]isoquinolines and hexahydropyrrolo[2,1-a]isoquinolines.

for the synthesis of tetrahydropyrrolothiazole derivatives. The reaction of cysteine with two equiv of arylaldehydes and one equiv of maleimides in EtOH at 90 °C for 12 h afforded tetrahydropyrrolothiazoles 29 in 66-79% yields with up to 7:1 dr (Scheme 17) [76]. Using olefinic oxindoles to replace maleimides, the reactions gave spiro[indoline-tetrahydropyrrolothia-



Figure 7: Bioactive tetrahydropyrrolothiazoles.



Scheme 17: Pseudo-four-component reaction for the synthesis of tetrahydropyrrolothiazoles 29 and 30 (>4:1 dr).

zole] products **30** in 55–70% with greater than 4:1 dr [76]. The reaction mechanism suggests that the reaction of cysteine with arylaldehydes gives *N*,*S*-acetals **27** which convert to AMYs **28** after decarboxlyation. Cycloaddition of **28** with maleimides or olefinic oxindoles gives products **29** and **30**, respectively. The reactions could be carried out as a two-step synthesis using two different arylaldehydes to give products **31** in 43–72% yields with greater than 4:1 dr (Scheme 18). A similar reaction sequence based on a [3 + 2] cycloaddition of stabilized AMYs has been reported by our lab [78].

Conclusion

The amino acid-based decarboxylative [3 + 2] cycloaddition reactions developed from our lab are summarized in this paper. The semi-stabilized *N*–H-type azomethine ylides derived from amino acids could be used for multicomponent, one-pot, and multistep reactions in the synthesis of heterocyclic compounds. The methods have advantages of using readily available starting materials, performing streamlined reactions, producing diverse product structures, and having high pot, atom, and step economy (PASE) [79-81] for the diversity-oriented synthesis (DOS) [82-88]. The work presented in this paper may also be helpful to understand the reaction mechanism and stereoselectivity of semi-stabilized *N*–H-type AMYs. We hope the new development for 1,3-dipolar cycloaddition chemistry can be used for the synthesis of bioactive heterocyclic compounds in medicinal and drug discovery programs.

Acknowledgements

We would like to thank all our co-workers who worked on the research projects highlighted in this paper, and the Centre for Green Chemistry at the University of Massachusetts Boston.

ORCID[®] iDs

Xiaofeng Zhang - https://orcid.org/0000-0003-4529-1158 Xiaoming Ma - https://orcid.org/0000-0003-1358-428X Wei Zhang - https://orcid.org/0000-0002-6097-2763

References

- Baunach, M.; Hertweck, C. Angew. Chem., Int. Ed. 2015, 54, 12550–12552. doi:10.1002/anie.201507120
- Coldham, I.; Hufton, R. Chem. Rev. 2005, 105, 2765–2810. doi:10.1021/cr040004c
- Otero-Fraga, J.; Montesinos-Magraner, M.; Mendoza, A. Synthesis 2017, 49, 802–809. doi:10.1055/s-0036-1588662
- Wu, S.; Zhu, G.; Wei, S.; Chen, H.; Qu, J.; Wang, B. Org. Biomol. Chem. 2018, 16, 807–815. doi:10.1039/c7ob03051g
- Cayuelas, A.; Larrañaga, O.; Selva, V.; Nájera, C.; Akiyama, T.; Sansano, J. M.; de Cózar, A.; Miranda, J. I.; Cossío, F. P. *Chem. – Eur. J.* **2018**, *24*, 8092–8097. doi:10.1002/chem.201801433
- Zhang, C.; Das, D.; Seidel, D. Chem. Sci. 2011, 2, 233–236. doi:10.1039/c0sc00432d
- Snider, B. B.; Ahn, Y.; O'Hare, S. M. Org. Lett. 2001, 3, 4217–4220. doi:10.1021/ol0168840
- Vidadala, S. R.; Golz, C.; Strohmann, C.; Daniliuc, C.-G.; Waldmann, H. Angew. Chem., Int. Ed. 2015, 54, 651–655. doi:10.1002/anie.201409942
- Gollner, A.; Weinstabl, H.; Fuchs, J. E.; Rudolph, D.; Garavel, G.; Hofbauer, K. S.; Karolyi-Oezguer, J.; Gmaschitz, G.; Hela, W.; Kerres, N.; Grondal, E.; Werni, P.; Ramharter, J.; Broeker, J.; McConnell, D. B. *ChemMedChem* **2019**, *14*, 88–93. doi:10.1002/cmdc.201800617
- Narayan, R.; Potowski, M.; Jia, Z.-J.; Antonchick, A. P.; Waldmann, H. Acc. Chem. Res. 2014, 47, 1296–1310. doi:10.1021/ar400286b
- 11. Pearson, W. H.; Lovering, F. E. *J. Org. Chem.* **1998**, *63*, 3607–3617. doi:10.1021/jo972255+
- Narayan, R.; Bauer, J. O.; Strohmann, C.; Antonchick, A. P.; Waldmann, H. Angew. Chem., Int. Ed. 2013, 52, 12892–12896. doi:10.1002/anie.201307392



Scheme 18: One-pot two-step synthesis of spirooxindole-pyrrolothiazoles 31 (>4:1 dr).

- Pearson, W. H.; Postich, M. J. J. Org. Chem. 1994, 59, 5662–5671. doi:10.1021/jo00098a026
- 14. Bhat, C.; Tilve, S. G. *RSC Adv.* **2014**, *4*, 5405–5452. doi:10.1039/c3ra44193h
- 15. Milen, M.; Abranyi-Balogh, P.; Keglevich, G. *Curr. Org. Synth.* **2014**, *11*, 889–901. doi:10.2174/1570179411666140818210247
- Tang, S.; Zhang, X.; Sun, J.; Niu, D.; Chruma, J. J. Chem. Rev. 2018, 118, 10393–10457. doi:10.1021/acs.chemrev.8b00349
- Banks, H. D. Org. Biomol. Chem. 2011, 9, 6335–6342. doi:10.1039/c1ob05588g
- Gothelf, K. V.; Jørgensen, K. A. Chem. Rev. 1998, 98, 863–910. doi:10.1021/cr970324e
- Hashimoto, T.; Maruoka, K. Chem. Rev. 2015, 115, 5366–5412. doi:10.1021/cr5007182
- Pandey, G.; Banerjee, P.; Gadre, S. R. Chem. Rev. 2006, 106, 4484–4517. doi:10.1021/cr050011g
- Chogii, I.; Njardarson, J. T. Angew. Chem., Int. Ed. 2015, 54, 13706–13710. doi:10.1002/anie.201506559
- 22. Peng, L.; Wang, H.; Guo, C. J. Am. Chem. Soc. 2021, 143, 6376–6381. doi:10.1021/jacs.1c02697
- 23. Adrio, J.; Carretero, J. C. *Chem. Commun.* **2014**, *50*, 12434–12446. doi:10.1039/c4cc04381b
- 24. Najera, C.; de Garcia Retamosa, M.; Sansano, J. M. Angew. Chem., Int. Ed. 2008, 47, 6055–6058. doi:10.1002/anie.200801690
- Potowski, M.; Schürmann, M.; Preut, H.; Antonchick, A. P.; Waldmann, H. *Nat. Chem. Biol.* **2012**, *8*, 428–430. doi:10.1038/nchembio.901
- 26. Esteban, F.; Cieślik, W.; Arpa, E. M.; Guerrero-Corella, A.; Díaz-Tendero, S.; Perles, J.; Fernández-Salas, J. A.; Fraile, A.; Alemán, J. ACS Catal. **2018**, *8*, 1884–1890. doi:10.1021/acscatal.7b03553
- 27. Zhang, W. Chem. Lett. 2013, 42, 676–681. doi:10.1246/cl.130504
- 28. Zhang, X.; Zhang, W. *Curr. Opin. Green Sustainable Chem.* **2018**, *11*, 65–69. doi:10.1016/j.cogsc.2018.04.005
- Lu, Q.; Song, G.; Jasinski, J. P.; Keeley, A. C.; Zhang, W. Green Chem. 2012, 14, 3010–3012. doi:10.1039/c2gc36066g
- 30. Zhang, W.; Lu, Y.; Geib, S. Org. Lett. 2005, 7, 2269–2272. doi:10.1021/ol0507773
- Gayen, B.; Banerji, A.; Dhara, K. Synth. Commun. 2016, 46, 293–308. doi:10.1080/00397911.2015.1135954
- 32. Madhavan, S.; Okamoto, S. ChemCatChem 2018, 10, 2014–2018. doi:10.1002/cctc.201702035
- Malatesti, N.; Boa, A. N.; Clark, S.; Westwood, R. *Tetrahedron Lett.* 2006, 47, 5139–5142. doi:10.1016/j.tetlet.2006.05.064
- 34. De Luca, L.; Chiminazzo, A.; Sperni, L.; Strukul, G.; Scarso, A. Chem. – Eur. J. 2017, 23, 3474–3478. doi:10.1002/chem.201605878
- Bolognesi, M. L.; Bartolini, M.; Cavalli, A.; Andrisano, V.; Rosini, M.; Minarini, A.; Melchiorre, C. *J. Med. Chem.* **2004**, *47*, 5945–5952. doi:10.1021/jm049782n
- 36. Higashino, T.; Yamada, T.; Yamamoto, M.; Furube, A.; Tkachenko, N. V.; Miura, T.; Kobori, Y.; Jono, R.; Yamashita, K.; Imahori, H. Angew. Chem., Int. Ed. 2016, 55, 629–633. doi:10.1002/anie.201509067
- Takano, Y.; Herranz, M. Á.; Martín, N.; Radhakrishnan, S. G.; Guldi, D. M.; Tsuchiya, T.; Nagase, S.; Akasaka, T. *J. Am. Chem. Soc.* 2010, *132*, 8048–8055. doi:10.1021/ja100665q
- 38. Guldi, D. M.; Swartz, A.; Luo, C.; Gómez, R.; Segura, J. L.; Martín, N. J. Am. Chem. Soc. 2002, 124, 10875–10886. doi:10.1021/ja012694x

- 39. Hölzel, H.; Haines, P.; Kaur, R.; Lungerich, D.; Jux, N.; Guldi, D. M. J. Am. Chem. Soc. 2022, 144, 8977–8986. doi:10.1021/jacs.2c00456
- Guldi, D. M.; Spänig, F.; Kreher, D.; Perepichka, I. F.; van der Pol, C.; Bryce, M. R.; Ohkubo, K.; Fukuzumi, S. *Chem. – Eur. J.* 2008, *14*, 250–258. doi:10.1002/chem.200700837
- 41. Che, J.; Ma, C.; Lu, J.; Chen, B.; Shi, Q.; Jin, X.; Song, R.; Xu, F.; Gan, L.; Li, J.; Hu, Y.; Dong, X. *Eur. J. Med. Chem.* **2022**, *228*, 113954. doi:10.1016/j.ejmech.2021.113954
- 42. Sato, S.; Takei, T.; Matsushita, Y.; Yasuda, T.; Kojima, T.; Kawano, M.; Ohnuma, M.; Tashiro, K. *Inorg. Chem.* **2015**, *54*, 11581–11583. doi:10.1021/acs.inorgchem.5b01183
- Wong, W. W. H.; Vak, D.; Singh, T. B.; Ren, S.; Yan, C.; Jones, D. J.; Liaw, I. I.; Lamb, R. N.; Holmes, A. B. *Org. Lett.* **2010**, *12*, 5000–5003. doi:10.1021/ol102166m
- 44. Clark, R. B.; Pearson, W. H. Org. Lett. 1999, 1, 349–352. doi:10.1021/ol990677v
- 45. Kanemasa, S.; Sakamoto, K.; Tsuge, O. Bull. Chem. Soc. Jpn. 1989, 62, 1960–1968. doi:10.1246/bcsj.62.1960
- 46. Tsuge, O.; Kanemasa, S.; Hatada, A.; Matsuda, K. Bull. Chem. Soc. Jpn. 1986, 59, 2537–2545. doi:10.1246/bcsj.59.2537
 47. Vedejs, E.; West, F. G. Chem. Rev. 1986, 86, 941–955.
- doi:10.1021/cr00075a014
- Ardill, H.; Grigg, R.; Sridharan, V.; Surendrakumar, S. *Tetrahedron* 1988, 44, 4953–4966. doi:10.1016/s0040-4020(01)86199-8
- 49. Pearson, W. H.; Postich, M. J. J. Org. Chem. 1992, 57, 6354–6357. doi:10.1021/jo00049a058
- Pearson, W. H.; Walters, M. A.; Oswell, K. D. J. Am. Chem. Soc. 1986, 108, 2769–2771. doi:10.1021/ja00270a055
- 51. Tsuge, O.; Kanemasa, S.; Yamada, T.; Matsuda, K. J. Org. Chem. 1987, 52, 2523–2530. doi:10.1021/jo00388a033
- Gollner, A.; Rudolph, D.; Arnhof, H.; Bauer, M.; Blake, S. M.; Boehmelt, G.; Cockroft, X.-L.; Dahmann, G.; Ettmayer, P.; Gerstberger, T.; Karolyi-Oezguer, J.; Kessler, D.; Kofink, C.; Ramharter, J.; Rinnenthal, J.; Savchenko, A.; Schnitzer, R.; Weinstabl, H.; Weyer-Czernilofsky, U.; Wunberg, T.; McConnell, D. B. *J. Med. Chem.* **2016**, *59*, 10147–10162. doi:10.1021/acs.jmedchem.6b00900
- 53. Rao, M. P.; Gunaga, S. S.; Zuegg, J.; Pamarthi, R.; Ganesh, M. Org. Biomol. Chem. 2019, 17, 9390–9402. doi:10.1039/c9ob01429b
- 54. Dong, H.; Song, S.; Li, J.; Xu, C.; Zhang, H.; Ouyang, L. Bioorg. Med. Chem. Lett. 2015, 25, 3585–3591. doi:10.1016/j.bmcl.2015.06.076
- Rehn, S.; Bergman, J.; Stensland, B. Eur. J. Org. Chem. 2004, 413–418. doi:10.1002/ejoc.200300621
- 56. Wang, N.; Jiang, F.; Du, Z.; Bao, X.; Wang, T.; Yang, R. Supramol. Chem. **2012**, *24*, 819–825. doi:10.1080/10610278.2012.721551
- 57. Andersson, C.-H.; Nyholm, L.; Grennberg, H. Dalton Trans. 2012, 41, 2374–2381. doi:10.1039/c2dt12097f
- Pérez, L.; Lenoble, J.; Barberá, J.; de la Cruz, P.; Deschenaux, R.; Langa, F. Chem. Commun. 2008, 4590–4592. doi:10.1039/b808730j
- 59. Seidel, D. Acc. Chem. Res. 2015, 48, 317-328. doi:10.1021/ar5003768
- Mantelingu, K.; Lin, Y.; Seidel, D. Org. Lett. 2014, 16, 5910–5913. doi:10.1021/ol502918g
- 61. Zhang, X.; Liu, M.; Qiu, W.; Evans, J.; Kaur, M.; Jasinski, J. P.; Zhang, W. ACS Sustainable Chem. Eng. 2018, 6, 5574–5579. doi:10.1021/acssuschemeng.8b00555
- Zhang, X.; Liu, M.; Zhan, D.; Kaur, M.; Jasinski, J. P.; Zhang, W. New J. Chem. 2022, 46, 3866–3870. doi:10.1039/d1nj05538k

- Ma, X.; Zhang, W. *iScience* 2022, 25, 105005. doi:10.1016/j.isci.2022.105005
- Zhang, W.; Yi, W. B. Pot, Atom, and Step Economy (PASE) Synthesis; Springer Nature Switzerland: Cham, Switzerland, 2019. doi:10.1007/978-3-030-22596-4
- Zhang, X.; Liu, M.; Zhang, W.; Legris, M.; Zhang, W. J. Fluorine Chem. 2017, 204, 18–22. doi:10.1016/j.jfluchem.2017.10.003
- Hao, J.; Milcent, T.; Retailleau, P.; Soloshonok, V. A.; Ongeri, S.; Crousse, B. *Eur. J. Org. Chem.* **2018**, 3688–3692. doi:10.1002/ejoc.201800255
- 67. Corbett, M. T.; Xu, Q.; Johnson, J. S. *Org. Lett.* **2014**, *16*, 2362–2365. doi:10.1021/ol500679w
- Donohoe, T. J.; Sintim, H. O. Org. Lett. 2004, 6, 2003–2006. doi:10.1021/ol049397s
- Zhang, X.; Qiu, W.; Evans, J.; Kaur, M.; Jasinski, J. P.; Zhang, W. Org. Lett. 2019, 21, 2176–2179. doi:10.1021/acs.orglett.9b00487
- Zhang, X.; Qiu, W.; Murray, S. A.; Zhan, D.; Evans, J.; Jasinski, J. P.; Wang, X.; Zhang, W. J. Org. Chem. 2021, 86, 17395–17403. doi:10.1021/acs.joc.1c01797
- 71. Ma, X.; Zhang, X.; Qiu, W.; Zhang, W.; Wan, B.; Evans, J.; Zhang, W. Molecules 2019, 24, 601. doi:10.3390/molecules24030601
- 72. Zhang, X.; Zhi, S.; Wang, W.; Liu, S.; Jasinski, J. P.; Zhang, W. *Green Chem.* **2016**, *18*, 2642–2646. doi:10.1039/c6gc00497k
- 73. Ma, X.; Zhang, X.; Awad, J. M.; Xie, G.; Qiu, W.; Muriph, R. E.; Zhang, W. *Tetrahedron Lett.* **2020**, *61*, 151392. doi:10.1016/j.tetlet.2019.151392
- 74. Sutherell, C. L.; Tallant, C.; Monteiro, O. P.; Yapp, C.; Fuchs, J. E.; Fedorov, O.; Siejka, P.; Müller, S.; Knapp, S.; Brenton, J. D.; Brennan, P. E.; Ley, S. V. *J. Med. Chem.* **2016**, *59*, 5095–5101. doi:10.1021/acs.jmedchem.5b01997
- 75. Ma, X.; Meng, S.; Zhang, X.; Zhang, Q.; Yan, S.; Zhang, Y.; Zhang, W. Beilstein J. Org. Chem. 2020, 16, 1225–1233. doi:10.3762/bjoc.16.106
- 76. Ma, X.; Qiu, W.; Liu, L.; Zhang, X.; Awad, J.; Evans, J.; Zhang, W. Green Synth. Catal. 2021, 2, 74–77. doi:10.1016/j.gresc.2020.11.001
- 77. Lu, J.; Yao, B.; Zhan, D.; Sun, Z.; Ji, Y.; Zhang, X. Beilstein J. Org. Chem. 2022, 18, 1607–1616. doi:10.3762/bjoc.18.171
- 78. Zhang, X.; Ma, X.; Qiu, W.; Awad, J.; Zhang, W. Green Process. Synth. 2022, 11, 1128–1135. doi:10.1515/gps-2022-0088
- 79. Hayashi, Y. Chem. Sci. 2016, 7, 866-880. doi:10.1039/c5sc02913a
- Sydnes, M. O. Curr. Green Chem. 2014, 1, 216–226. doi:10.2174/2213346101666140221225404
- Clarke, P. A.; Santos, S.; Martin, W. H. C. Green Chem. 2007, 9, 438–440. doi:10.1039/b700923b
- O' Connor, C. J.; Beckmann, H. S. G.; Spring, D. R. Chem. Soc. Rev. 2012, 41, 4444–4456. doi:10.1039/c2cs35023h
- Comer, E.; Duvall, J. R.; duPont Lee, M., IV. Future Med. Chem. 2014, 6, 1927–1942. doi:10.4155/fmc.14.111
- Dandapani, S.; Marcaurelle, L. A. Curr. Opin. Chem. Biol. 2010, 14, 362–370. doi:10.1016/j.cbpa.2010.03.018
- Galloway, W. R. J. D.; Isidro-Llobet, A.; Spring, D. R. Nat. Commun. 2010, 1, 80. doi:10.1038/ncomms1081
- 86. Gerry, C. J.; Schreiber, S. L. Curr. Opin. Chem. Biol. 2020, 56, 1–9. doi:10.1016/j.cbpa.2019.08.008
- 87. Mortensen, K. T.; Osberger, T. J.; King, T. A.; Sore, H. F.; Spring, D. R. *Chem. Rev.* **2019**, *119*, 10288–10317. doi:10.1021/acs.chemrev.9b00084
- Schreiber, S. L. Science 2000, 287, 1964–1969. doi:10.1126/science.287.5460.1964

License and Terms

This is an open access article licensed under the terms of the Beilstein-Institut Open Access License Agreement (https://www.beilstein-journals.org/bjoc/terms), which is identical to the Creative Commons Attribution 4.0 International License

(<u>https://creativecommons.org/licenses/by/4.0</u>). The reuse of material under this license requires that the author(s), source and license are credited. Third-party material in this article could be subject to other licenses (typically indicated in the credit line), and in this case, users are required to obtain permission from the license holder to reuse the material.

The definitive version of this article is the electronic one which can be found at: https://doi.org/10.3762/bjoc.19.123