Supporting Information

for

[2.2]Paracyclophane derivatives containing

tetrathiafulvalene moieties

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Detailed experimental procedures, supplementary spectroscopic and X-ray data

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A. Experimental section

General Remarks: Melting points: Büchi 510, uncorrected. IR: Bruker Tensor 27 with ATR. ¹H and ¹³C NMR: Bruker AV300 spectrometer with TMS as internal standard at room temperature. Chemical shifts are reported in ppm downfield from tetramethylsilane. MS: Finnigan MAT 90X, electron impact (EI). Elemental analyses: CE440 Elemental Analyser; the results were found to be in good agreement (±0.30%) with the calculated values. All reagents were commercially available and used without further purification.

2-Bromo-1-([2.2]paracyclophan-4-yl)ethan-1-one (2)

To a solution of 4-acetyl[2.2]paracyclophane (1) [1] (4.67 g, 18.70 mmol) in dioxane (25 mL), a solution of the molecular complex Br₂-dioxane (1 mL Br₂, 1.6 mL dioxane) in dioxane (10 mL) was added dropwise. The reaction mixture was left to react at rt for 30 min and then poured in water. The precipitate was filtered off, washed with water, dried and recrystallized from ethanol to provide compound **2** as a white solid, m.p. 115-116 °C (lit. 101-103 °C [2]); yield 81% (5.0 g); IR (KBr): nu(tilde) = 2929, 2921, 2851, 1694, 1211, 993, 720, 630, 505 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz): δ = 2.80–2.92 (m, 1 H, CH₂), 2.95–3.07 (m, 2 H, CH₂), 3.11–3.28 (m, 4 H, CH₂), 3.38–3.92 (m, 1 H, CH₂), 4.15, 4.36 (ABq, ²J = 12.2 Hz, 2 H, CH₂), 6.38 (dd, ⁴J = 1.8 Hz, ³J = 7.9 Hz, 1 H, CH_{ar}), 6.53 (m, 4 H, CH_{ar}); ¹³C NMR (CDCl₃, 75 MHz) δ = 33.2

^{1.} Truesdale, E. A.; Cram, D. J. J. Org. Chem. 1980, 45, 3974-3981.

^{2.} Mamyrbekova, Zh. A.; Soldatova, Š. A.; Abelentsev, V. I.; Solov'eva, T. I.; Guryshev, V. N.; Soldatenkov, A. T. *Pharm. Chem. J.* **1994**, *28*, 198-202.

(CH₂), 34.8 (CH₂), 35.0 (CH₂), 35.1 (CH₂), 36.0 (CH₂), 131.2 (CH), 132.3 (CH), 132.7 (CH), 132.8 (CH), 133.6 (CH), 134.6 (C), 136.6 (CH), 137.3 (CH), 139.1 (C), 140.0 (C), 141.1 (C), 142.9 (C), 192.7 ppm (s); MS (EI, 70 eV): m/z (%) = 310 (27) [M⁺+2] for ⁸¹Br, 328 (28) [M⁺] for ⁷⁹Br, 249 (32), 145 (67), 104 (100).

2-([2.2]Paracyclophan-4-yl)-2-oxoethyl-N,N-dimethyldithiocarbamate (4a)

General procedure

To a solution of 2-bromo-1-([2.2]paracyclophan-4-yl)ethan-1-one ($\mathbf{2}$, 4.29 g, 13.04 mmol) in acetone (100 mL) a solution of sodium *N*,*N*-dimethyldithiocarbamate ($\mathbf{3a}$, 2.24 g, 13.04 mmol) in acetone/water (40 mL, 1:1 v/v) was added. The reaction mixture was heated under reflux for 10 min, cooled to rt and then poured into water. The precipitate was filtered off, washed with water, dried, and recrystallized from ethanol to provide compound **4a**. Dithiocarbamates **4b** and **4c** were obtained under the same experimental conditions.

Compound 4a: Colorless solid; m.p. 143-144 °C; yield 81% (3.72 g); IR (KBr): nu(tilde) = 2921, 2883, 2850, 1683, 1498, 1295, 978, 720, 633, 502 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz): δ = 2.77–2.89 (m, 1 H, CH₂), 2.95–3.33 (m, 6 H, CH₂), 3.50 (s, 3 H, CH₃), 3.59 (s, 3 H, CH₃), 3.77–3.88 (m, 1 H, CH₂), 4.22, 5.01 (ABq, ²*J* = 17.2 Hz, 2 H, CH₂), 6.41–6.45 (m, 1 H, CH_{ar}), 6.48–6.60 (m, 4 H, CH_{ar}), 6.70 (dd, ⁴*J* = 1.7 Hz, ³*J* = 7.8 Hz, 1 H, CH_{ar}), 7.30 ppm (d, ⁴*J* = 1.7 Hz, 1 H, CH_{ar}); ¹³C NMR (CDCl₃, 75 MHz): δ = 34.9 (CH₂), 35.0 (CH₂), 35.1 (CH₂), 35.9 (CH₂), 41.7 (CH₃), 45.8 (CH₃), 46.7 (CH₂), 131.5 (CH), 132.5 (CH), 132.6 (CH), 132.7 (CH), 134.0 (CH), 136.2 (C), 136.3 (CH), 136.9 (CH), 139.1 (C), 140.0 (C), 140.1 (C), 142.0 (C), 194.6 (C), 195.9 ppm (C); MS (EI, 70 eV): *m/z* (%) = 369 (23) [M⁺], 354 (5), 131 (9), 104 (12), 88 (100). **Compound 4b**: Colorless solid; m.p. 104-105 °C; yield 80% (0.89 g); IR (KBr): nu(tilde) = 2926, 2871, 1681, 1430, 1158, 955, 720, 632 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz): δ = 1.85–2.19 (m, 4 H, CH₂), 2.70–3.40 (m, 8 H, CH₂), 3.61–4.03 (m, 4 H, CH₂), 4.22, 5.01 (ABq, ²*J* = 17.0 Hz, 2 H, CH₂), 6.30–6.82 ppm (m, 7 H, CH_{ar}); ¹³C NMR (CDCl₃, 75 MHz): δ = 24.0 (CH₂), 25.8 (CH₂), 34.5 (CH₂), 34.7 (CH₂), 34.8 (CH₂), 35.5 (CH₂), 45.4 (CH₂), 50.4 (CH₂), 55.0 (CH₂), 131.1 (CH), 132.1 (CH), 132.3 (CH), 132.4 (CH), 133.6 (CH), 135.8 (C), 135.9 (CH), 136.5 (CH), 138.7 (C), 139.6 (C), 139.7 (C), 141.5 (C), 190.1 (C), 194.3 ppm (C); MS (EI, 70 eV)): *m/z* (%) = 395 (20) [M⁺], 146 (21), 114 (100), 104 (33).

Compound 4c: White solid; m.p. 105-106 °C; yield 80% (0.56 g); IR 3011, 2925, 2853, 1688, 1425, 1228, 1112, 990, 723, 634, 540 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.77–2.90 (m, 1 H, CH₂), 2.93–3.33 (m, 7 H, CH₂), 3.77–3.85 (m, 4 H, CH₂), 3.93–4.51 (m, 4 H, CH₂), 4.27, 5.01 (ABq, ²J = 17.1 Hz, 2 H, CH₂), 6.42 (dd, ⁴J = 1.9 Hz, ³J = 8.6 Hz, 1 H, CH_{ar}); 6,48–6,58 (m, 4 H, CH_{ar}); 6,70 (dd, ⁴J= 1.8 Hz; ³J = 7.8 Hz, 1 H, CH_{ar}); 7.26 (d, ⁴J = 1.8 Hz, 1H, CH_{ar}) ppm; ¹³C NMR (CDCl₃, 75 MHz) δ 35.0 (CH₂), 35.1 (2xCH₂), 35.9 (CH₂), 62.1 (CH₂), 62.2 (3xCH₂), 46.0 (CH₂), 131.4 (CH), 132.4 (CH), 132.7 (CH), 132.8 (CH), 133.9 (CH), 136.2 (C), 136.3 (CH), 136.9 (CH), 139.1 (C), 140.0 (C), 140.1 (C), 142.1 (C), 194.4 (C), 196.3 (s) ppm; MS (EI) *m/z* (%) 411 (25) [M⁺], 130 (100), 104 (24), 86 (30).

2-(*N*,*N*-Dimethylamino)-4-([2.2]paracyclophan-4-yl)-1,3-dithiol-2-ylium perchlorate (5a)

General procedure

To a mixture of 6 mL concentrated sulfuric acid and 18 mL glacial acetic acid, 2-([2.2]paracyclophan-4-yl)-2-oxoethyl-*N*,*N*-dimethyldithiocarbamate (**4a**, 2.74 g, 7.42 mmol) was added in small portions. The reaction mixture was heated at 80 °C for 10 min, cooled after which 3 mL of perchloric acid were added. The precipitate obtained after water addition (250 mL) was filtered, dried and recrystallized from ethanol to provide compound **5a**. 1,3-Dithiolium perchlorates **5b** and **5c** were obtained under the same experimental conditions.

Compound 5a: White solid; m.p. 260-261 °C; yield 63% (2.11 g); IR 3049, 2928, 1576, 1404, 1074, 720, 619, 512 cm⁻¹; ¹H NMR (DMSO-*d*6, 300 MHz) δ 2.79–2.91 (m, 1 H, CH₂), 2.96–3.13 (m, 6 H, CH₂), 3.41–3.52 (m, 1 H, CH₂), 3.56 (s, 3 H, CH₃), 3.60 (s, 3 H, CH₃), 6.55 (dd, ⁴*J* = 1.5 Hz, ³*J* = 7.8 Hz, 1 H, CH_{ar}), 6.58–6.72 (m, 6 H, CH_{ar}), 7.69 (s, 1 H, CH) ppm; ¹³C NMR (DMSO-*d*6, 75 MHz) δ 33.5 (CH₂), 34.1 (CH₂), 34.2 (CH₂), 34.6 (CH₂), 46.6 (CH₃), 47.6 (CH₃), 120.8 (CH), 129.5 (CH), 129.7 (C), 131.9 (CH), 132.7 (CH), 132.9 (CH), 133.3 (CH), 134.9 (CH), 136.4 (CH), 138.0 (C), 138.7 (C), 138.8 (C), 139.4 (C), 141.0 (C), 187.0 (C) ppm; MS (ESI) *m/z* (%) 352 [M⁺ - CIO₄].

Compound 5b: White solid; m.p. 235-236 °C; yield 58% (0.62 g); IR 2931, 2857, 1579, 1439, 1257, 1077, 718, 620 cm⁻¹; ¹H NMR (DMSO-*d*6, 300 MHz) δ 2.17–2.28 (m, 4 H, CH₂), 2.76–2.91 (m, 1 H, CH₂), 2.95–3.16 (m, 6 H, CH₂), 3.38–3.55 (m, 1 H, CH₂), 3.71–3.92 (m, 4 H, CH₂), 6.55 (dd, ⁴*J* = 1.5 Hz, ³*J* = 7.8 Hz, 1 H, CH_{ar}), 6.58–6.74 (m, 6 H, CH_{ar}), 7.67 (s, 1 H, CH) ppm; ¹³C NMR (DMSO-*d*6, 75 MHz) δ 26.1 (CH₂), 26.2 (CH₂), 33.5 (CH₂), 34.1 (CH₂), 34.2 (CH₂), 34.6 (CH₂), 56.4 (CH₂), 57.3 (CH₂), 120.6 (CH), 129.4 (CH), 129.7 (C), 131.9 (CH), 132.7 (CH), 132.9 (CH), 133.3 (CH), 134.9 (CH), 136.4 (CH), 138.0 (C), 138.4 (C), 138.8 (C), 139.4 (C), 141.0 (C), 181.3 (C) ppm; MS (ESI) *m/z* (%) 378 [M⁺ - CIO₄].

Compound 5c: White solid; m.p. 213-214 °C; yield 81% (0.24 g); IR 2918, 2849, 1576, 1431, 1258, 1077, 719, 620, 540 cm⁻¹; ¹H NMR (DMSO-*d*6, 300 MHz) δ 2.80–2.93 (m, 1 H, CH₂), 2.94–3.14 (m, 6 H, CH₂), 3.40–3.50 (m, 1 H, CH₂), 3.84–3.92 (m,

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4 H, CH₂), 3.92–4.00 (m, 4 H, CH₂), 6.55 (dd, ${}^{4}J$ = 1.4 Hz, ${}^{3}J$ = 7.7 Hz, 1 H, CH_{ar}), 6.57–6.74 (m, 6 H, CH_{ar}), 7.71 (s, 1 H, CH) ppm; 13 C NMR (DMSO-*d*6, 75 MHz) δ 33.4 (CH₂), 34.1 (CH₂), 34.2 (CH₂), 34.6 (CH₂), 53.4 (CH₂), 54.6 (CH₂), 64.3 (2xCH₂), 119.8 (CH), 129.4 (CH), 129.6 (C), 131.9 (CH), 132.7 (CH), 132.8 (CH), 133.3 (CH), 135.0 (CH), 136.4 (CH), 137.6 (C), 138.0 (C), 138.8 (C), 139.3 (C), 141.0 (C), 188.1 (s) ppm; MS (ESI) *m/z* (%) 394 [M⁺ - CIO₄].

4-([2.2]Paracyclophan-4-yl)-1,3-dithiol-2-thione (6)

To a suspension of 2-(*N*,*N*-dimethylamino)-4-([2.2]paracyclophan-4-yl)-1,3-dithiol-2ylium perchlorate (**5a**, 2.18 g, 4.82 mmol) in ethanol (100 mL) Na₂S-9H₂O (2.32 g, 9.64 mmol) was added. The reaction mixture was allowed to react for 12 h at rt. The precipitate obtained after water addition (300 mL) was filtered, dried and purified by column chromatography using CH₂Cl₂/pentane (1:1) as eluent to provide compound **6** as a yellow solid. M.p. 137-138 °C; R_f = 0.49; yield 41% (0.67 g); IR 2922, 2849, 1054, 799, 718, 645 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) δ 2.81–2.89 (m, 1 H, CH₂), 2.89–3.09 (m, 6 H, CH₂), 3.36–3.49 (m, 1 H, CH₂), 6.32–6.38 (m, 2 H, CH_{ar}), 6.48– 6.58 (m, 4 H, CH_{ar}), 6.61 (dd, ⁴*J* = 1.8 Hz, ³*J* = 7.9 Hz, 1 H, CH_{ar}), 6.75 (s, 1 H, CH) ppm; ¹³C NMR (CDCl₃, 75 MHz) δ 33.8 (CH₂), 34.6 (CH₂), 34.7 (CH₂), 35.0 (CH₂), 124.3 (CH), 129.5 (CH), 130.9 (C), 131.7 (CH), 132.3 (CH), 132.5 (CH), 133.0 (CH), 134.2 (CH), 136.0 (CH), 137.5 (C), 138.9 (C), 139.2 (C), 140.4 (C), 146.2 (C), 212.7 (C) ppm; MS (EI) *m/z* (%) 340 (52) [M⁺], 264 (16), 160 (100), 104 (48).

Tetrathiafulvalenes 7

A mixture of 4-([2.2]paracyclophan-4-yl)-1,3-dithiol-2-thione (**6**, 0.150 g, 0.44 mmol) and trimethyl phosphite (0.52 mL, 4.4 mmol) was heated at 100 °C for 14 h. The reaction mixture was purified by column chromatography using CH_2Cl_2 /pentane (1:1)

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as eluent to provide tetrathiafulvalenes **7** as yellow solids. M.p. 137-138 °C; $R_f = 0.60$; yield 17% (0.023 g); IR 2922, 2889, 2850, 1587, 1499, 939, 720, 644 cm⁻¹; ¹H NMR (CDCl₃, 300 MHz) selected data for the major isomer δ 2.96–3.10 (m, 14 H, CH₂), 3.66 (m, 2 H, CH₂), 6.29–6.63 (m, 14 H, CH_{ar}), 6.79 (m, 2 H, CH) ppm; ¹³C NMR (CDCl₃, 75 MHz) selected data for the major isomer δ 35.1 (CH₂), 35.2 (CH₂), 35.4 (CH₂), 35.5 (CH₂), 132.3 (CH₂), 132.4 (CH₂), 132.6 (CH₂), 132.7 (CH₂), 133.1 (CH₂), 135.9 (C), 136.0 (C), 139.3 (C), 139.4 (C), 139.5 (C), 139.6 (C) ppm; MS (EI) *m/z* (%) 616 (77) [M⁺], 583 (15), 160 (62), 128 (100), 104 (60).

B. X-ray spectral data

X-ray structure determinations

Crystal data are summarized in Table S1. Crystals were mounted in inert oil on glass fibers and transferred to the cold gas stream of an Oxford Diffraction Nova E diffractometer. Intensity measurements were performed using mirror-focussed CuK α radiation ($\lambda = 1.54184$ Å). Absorption corrections were based on multi-scans. Structures were refined anisotropically on F^2 using the program SHELXL-97 [3]. Hydrogen atoms were included using a riding model or rigid methyl groups. Compound **6** was not enantiomerically pure and crystallized only by chance in the chiral (Sohncke) space group $P2_1$; the structure was refined as an enantiomeric twin, with a Flack parameter of 0.440(16).

^{3.} Sheldrick, G. M. Acta Cryst. 2008, A64, 112-122.

Compound	4 a	6
Formula	$C_{21}H_{23}NOS_2$	$C_{19}H_{16}S_3$
M_r	369.52	340.50
Temperature (K)	100	100
Crystal habit	colourless plate	yellow prism
Crystal size (mm)	$0.1 \times 0.1 \times 0.015$	$0.15 \times 0.08 \times 0.04$
Crystal system	orthorhombic	Monoclinic
Space group	Pbca	$P2_1$
Cell dimensions:		
<i>a</i> (Å)	11.8026(7)	7.4723(3)
<i>b</i> (Å)	8.2761(4)	23.1083(8)
<i>c</i> (Å)	37.697(3)	9.2378(4)
α (°)	90	90
β (°)	90	93.560(3)
γ (°)	90	90
Cell volume ($Å^3$)	3682.2	1592.04
Ζ	8	4
D_x (g cm ⁻³)	1.333	1.421
μ (mm ⁻¹)	2.68	4.18
2θ(max) (°)	155.2	151.2
Reflections collected	75990	32381
Independent reflections	3833	6557
R(int)	0.078	0.043
Transmissions	0.607 - 1.000	0.630 - 1.000
Data/restraints/parameters	3833/0/228	6557/1/398
Goodness-of-fit on F^2	1.09	1.07
wR2 (all reflections)	0.116	0.121
$R1 (F > 4\sigma(F))$	0.048	0.045
Max. Δρ (e Å ⁻³)	0.62	0.70

 Table S1: Crystallographic data for 4a and 6



dimethyldithiocarbamate (4a)



Table S2: Crystal data and structure refinement.

Identification code	perc
Empirical formula	$C_{21}H_{23}NOS_2$
Formula weight	369.52
Temperature	100(2) K
Wavelength	1.54184 Å
Crystal system	Orthorhombic
Space group	Pbca
Unit cell dimensions	$a = 11.8026(7) \text{ Å} \qquad \alpha = 90^{\circ}$
	$b = 8.2761(4) \text{ Å} \qquad \beta = 90^{\circ}$
	$c = 37.697(3) \text{ Å}$ $\gamma = 90^{\circ}$
Volume	3682.2(4) Å ³
Ζ	8
Density (calculated)	1.333 Mg/m^{3}
Absorption coefficient	2.677 mm ⁻¹
F(000)	1568
Crystal size	0.10 x 0.10 x 0.02 mm ³
Theta range for data collection	4.42 to 77.61°
Index ranges	-14<=h<=14, -9<=k<=10, -46<=l<=47
Reflections collected	75990
Independent reflections	3833 [R(int) = 0.0781]
Completeness to theta = 75.00°	100.0 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	1.00000 and 0.60742
Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3833 / 0 / 228
Goodness-of-fit on F ²	1.090
Final R indices [I>2sigma(I)]	R1 = 0.0478, $wR2 = 0.1116$
R indices (all data)	R1 = 0.0548, $wR2 = 0.1162$
Largest diff. peak and hole	0.618 and -0.387 e.Å ⁻³

	Х	у	Z	U(eq)	
S(1)	2938.6(5)	814.7(6)	500.2(1)	31.4(1)	
S(2)	4769.9(5)	-1147.2(7)	858.8(2)	37.5(2)	
O(1)	4721.2(14)	3127(2)	684.1(4)	39.3(4)	
N(1)	4089.5(17)	-1500(2)	193.0(5)	33.8(4)	
C(1)	6822.5(19)	3332(3)	1226.2(7)	43.0(6)	
C(2)	6211.5(19)	4933(3)	1110.0(7)	39.1(5)	
C(3)	5075.9(18)	5137(3)	1291.5(6)	30.1(4)	
C(4)	4148.3(17)	4080(2)	1249.4(6)	26.9(4)	
C(5)	3385.8(17)	3887(3)	1532.5(6)	28.5(4)	
C(6)	3515.4(18)	4732(3)	1849.3(6)	31.0(5)	
C(7)	4266.0(19)	6025(3)	1850.2(6)	32.7(5)	
C(8)	5027.4(19)	6224(3)	1575.4(6)	33.1(5)	
C(9)	3066(2)	4042(3)	2192.6(6)	40.5(5)	
C(10)	3967(3)	2986(3)	2398.6(7)	48.1(6)	
C(11)	4980(2)	2549(3)	2173.2(6)	39.7(5)	
C(12)	4931(2)	1399(3)	1903.4(7)	39.2(5)	
C(13)	5654(2)	1498(3)	1613.7(7)	37.7(5)	
C(14)	6446.9(18)	2747(3)	1586.2(7)	36.9(5)	
C(15)	6659(2)	3643(3)	1893.1(7)	41.1(6)	
C(16)	5925(2)	3544(3)	2179.8(7)	41.3(6)	
C(17)	4058.7(18)	3024(3)	931.3(6)	29.6(4)	
C(18)	3105.3(18)	1782(3)	921.6(6)	30.8(4)	
C(19)	3999.6(18)	-700(3)	500.9(6)	30.1(4)	
C(20)	3403(2)	-1130(3)	-118.3(6)	39.7(5)	
C(21)	4837(2)	-2917(3)	167.9(7)	43.7(6)	

Table S3: Atomic coordinates ($x \ 10^4$) and equivalent isotropic displacement parameters (Å²x 10³). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

S(1)-C(19)	1.772(2)	C(5)-C(6)	1.393(3)
S(1)-C(18)	1.790(2)	C(6)-C(7)	1.389(3)
S(2)-C(19)	1.669(2)	C(6)-C(9)	1.511(3)
O(1)-C(17)	1.219(3)	C(7)-C(8)	1.381(3)
N(1)-C(19)	1.341(3)	C(9)-C(10)	1.581(4)
N(1)-C(20)	1.458(3)	C(10)-C(11)	1.511(4)
N(1)-C(21)	1.471(3)	C(11)-C(16)	1.387(4)
C(1)-C(14)	1.508(4)	C(11)-C(12)	1.394(4)
C(1)-C(2)	1.570(4)	C(12)-C(13)	1.388(4)
C(2)-C(3)	1.514(3)	C(13)-C(14)	1.398(3)
C(3)-C(8)	1.399(3)	C(14)-C(15)	1.397(3)
C(3)-C(4)	1.410(3)	C(15)-C(16)	1.387(4)
C(4)-C(5)	1.405(3)	C(17)-C(18)	1.524(3)
C(4)-C(17)	1.488(3)		
C(19)-S(1)-C(18)	103.75(10)	C(11)-C(10)-C(9)	112.9(2)
C(19)-N(1)-C(20)	123.27(19)	C(16)-C(11)-C(12)	116.9(2)
C(19)-N(1)-C(21)	119.8(2)	C(16)-C(11)-C(10)	119.0(2)
C(20)-N(1)-C(21)	116.7(2)	C(12)-C(11)-C(10)	122.8(2)
C(14)-C(1)-C(2)	112.8(2)	C(13)-C(12)-C(11)	120.6(2)
C(3)-C(2)-C(1)	111.99(19)	C(12)-C(13)-C(14)	120.9(2)
C(8)-C(3)-C(4)	116.9(2)	C(15)-C(14)-C(13)	116.8(2)
C(8)-C(3)-C(2)	117.0(2)	C(15)-C(14)-C(1)	121.5(2)
C(4)-C(3)-C(2)	124.5(2)	C(13)-C(14)-C(1)	120.1(2)
C(5)-C(4)-C(3)	118.8(2)	C(16)-C(15)-C(14)	120.2(2)
C(5)-C(4)-C(17)	119.98(19)	C(11)-C(16)-C(15)	121.5(2)
C(3)-C(4)-C(17)	120.68(19)	O(1)-C(17)-C(4)	122.0(2)
C(6)-C(5)-C(4)	121.6(2)	O(1)-C(17)-C(18)	120.1(2)
C(7)-C(6)-C(5)	117.3(2)	C(4)-C(17)-C(18)	117.89(18)
C(7)-C(6)-C(9)	120.9(2)	C(17)-C(18)-S(1)	113.83(15)
C(5)-C(6)-C(9)	120.4(2)	N(1)-C(19)-S(2)	123.19(17)
C(8)-C(7)-C(6)	120.3(2)	N(1)-C(19)-S(1)	113.85(17)
C(7)-C(8)-C(3)	121.6(2)	S(2)-C(19)-S(1)	122.89(13)
C(6)-C(9)-C(10)	113.2(2)		

 Table S4: Bond lengths [Å] and angles [°].

Table S5: Torsion angles [°].

C(14)-C(1)-C(2)-C(3)	-24.8(3)	O(1)-C(17)-C(18)-S(1)	-9.9(3)
C(1)-C(2)-C(3)-C(8)	102.4(2)	C(4)-C(17)-C(18)-S(1)	170.39(15)
C(1)-C(2)-C(3)-C(4)	-62.9(3)	C(19)-S(1)-C(18)-C(17)	80.88(16)
C(8)-C(3)-C(4)-C(5)	-15.5(3)	C(20)-N(1)-C(19)-S(2)	178.90(17)
C(2)-C(3)-C(4)-C(5)	149.8(2)	C(21)-N(1)-C(19)-S(2)	4.4(3)
C(8)-C(3)-C(4)-C(17)	172.52(19)	C(20)-N(1)-C(19)-S(1)	1.8(3)
C(2)-C(3)-C(4)-C(17)	-22.1(3)	C(21)-N(1)-C(19)-S(1)	-172.69(16)
C(3)-C(4)-C(5)-C(6)	1.1(3)	C(18)-S(1)-C(19)-N(1)	-174.77(16)
C(17)-C(4)-C(5)-C(6)	173.13(19)	C(18)-S(1)-C(19)-S(2)	8.12(16)
C(4)-C(5)-C(6)-C(7)	13.8(3)		
C(4)-C(5)-C(6)-C(9)	-152.8(2)		
C(5)-C(6)-C(7)-C(8)	-14.1(3)		
C(9)-C(6)-C(7)-C(8)	152.4(2)		
C(6)-C(7)-C(8)-C(3)	-0.5(3)		
C(4)-C(3)-C(8)-C(7)	15.5(3)		
C(2)-C(3)-C(8)-C(7)	-151.0(2)		
C(7)-C(6)-C(9)-C(10)	-74.9(3)		
C(5)-C(6)-C(9)-C(10)	91.3(3)		
C(6)-C(9)-C(10)-C(11)	-13.9(3)		
C(9)-C(10)-C(11)-C(16)	92.4(3)		
C(9)-C(10)-C(11)-C(12)	-73.6(3)		
C(16)-C(11)-C(12)-C(13)	-14.2(3)		
C(10)-C(11)-C(12)-C(13)	152.0(2)		
C(11)-C(12)-C(13)-C(14)	0.2(3)		
C(12)-C(13)-C(14)-C(15)	14.4(3)		
C(12)-C(13)-C(14)-C(1)	-151.2(2)		
C(2)-C(1)-C(14)-C(15)	-64.6(3)		
C(2)-C(1)-C(14)-C(13)	100.3(3)		
C(13)-C(14)-C(15)-C(16)	-15.0(3)		
C(1)-C(14)-C(15)-C(16)	150.4(2)		
C(12)-C(11)-C(16)-C(15)	13.7(3)		
C(10)-C(11)-C(16)-C(15)	-153.1(2)		
C(14)-C(15)-C(16)-C(11)	0.9(4)		
C(5)-C(4)-C(17)-O(1)	-178.2(2)		
C(3)-C(4)-C(17)-O(1)	-6.3(3)		
C(5)-C(4)-C(17)-C(18)	1.6(3)		
C(3)-C(4)-C(17)-C(18)	173.48(18)		



B2. X-ray data for 4-([2.2]paracyclophan-4-yl)-1,3-dithiol-2-thione (6)

 Table S6: Crystal data and structure refinement.

Identification code	dekel	
Empirical formula	$C_{19}H_{16}S_3$	
Formula weight	340.50	
Temperature	100(2) K	
Wavelength	1.54184 Å	
Crystal system	Monoclinic	
Space group	P21	
Unit cell dimensions	a = 7.4723(3) Å	$\alpha = 90^{\circ}$
	b = 23.1083(8) Å	$\beta = 93.560(3)^{\circ}$
	c = 9.2378(4) Å	$\gamma = 90^{\circ}$
Volume	1592.04(11) Å ³	
Ζ	4	
Density (calculated)	1.421 Mg/m ³	
Absorption coefficient	4.178 mm ⁻¹	
F(000)	712	
Crystal size	0.15 x 0.08 x 0.04 mm ³	
Theta range for data collection	3.83 to 75.60°	
Index ranges	-9<=h<=8, -28<=k<=28, -	·11<=l<=11
Reflections collected	32381	
Independent reflections	6557 [R(int) = 0.0429]	
Completeness to theta = 75.00°	99.8 %	
Absorption correction	Semi-empirical from equi	valents
Max. and min. transmission	1.00000 and 0.62972	
Refinement method	Full-matrix least-squares	$on F^2$
Data / restraints / parameters	6557 / 1 / 398	
Goodness-of-fit on F ²	1.071	
Final R indices [I>2sigma(I)]	R1 = 0.0450, wR2 = 0.120	03
R indices (all data)	R1 = 0.0457, wR2 = 0.120	08
Absolute structure parameter	0.440(16)	
Largest diff. peak and hole	0.696 and -0.410 e.Å ⁻³	

	Х	У	Z	U(eq)	
S (1)	7667.7(12)	4795.7(4)	4645.2(10)	35.8(2)	
S(2)	6709.8(13)	5472.1(4)	7242.1(10)	39.5(2)	
S(3)	4289.7(10)	4581.0(3)	5856.6(8)	28.7(2)	
C(1)	4633(5)	2790.9(18)	6002(4)	36.2(8)	
C(2)	2948(5)	3200.6(16)	6006(4)	33.8(7)	
C(3)	2311(4)	3406.8(14)	4496(3)	26.7(6)	
C(4)	3301(4)	3789.9(14)	3656(4)	25.8(6)	
C(5)	3127(4)	3760.9(14)	2143(4)	27.7(6)	
C(6)	2024(4)	3350.2(14)	1424(4)	28.6(7)	
C(7)	776(4)	3069.5(15)	2253(4)	32.9(7)	
C(8)	929(4)	3097.1(15)	3748(4)	34.3(7)	
C(9)	2421(5)	3140.9(15)	-60(4)	33.1(7)	
C(10)	3755(5)	2604.9(15)	-25(4)	32.4(7)	
C(11)	4497(4)	2456.5(14)	1489(4)	28.9(7)	
C(12)	3556(5)	2086.8(15)	2362(4)	31.0(7)	
C(13)	3777(5)	2133.4(15)	3870(4)	32.0(7)	
C(14)	4932(5)	2544.7(15)	4519(4)	29.9(7)	
C(15)	6131(4)	2813.5(14)	3645(4)	29.3(7)	
C(16)	5916(4)	2769.2(15)	2138(4)	28.4(6)	
C(17)	4698(4)	4178.4(13)	4311(3)	25.8(6)	
C(18)	6288(5)	4291.5(16)	3777(4)	31.8(7)	
C(19)	6253(5)	4978.5(15)	6012(4)	31.2(7)	
S (1')	10343.1(13)	4310.1(4)	8064.9(9)	37.7(2)	
S(2')	6837.6(16)	3879.0(5)	9032.9(12)	49.0(3)	
S(3')	9073.4(11)	4739.7(4)	10733.2(8)	30.3(2)	
C(1')	11040(5)	6443.4(17)	9103(4)	34.4(7)	
C(2')	12889(5)	6114.4(16)	9297(4)	34.9(7)	
C(3')	13214(4)	5842.0(15)	10779(4)	28.4(7)	
C(4')	12181(4)	5384.7(14)	11304(3)	25.9(6)	
C(5')	11944(4)	5348.4(14)	12789(3)	24.8(6)	
C(6')	12735(4)	5746.4(14)	13769(3)	25.6(6)	
C(7')	14066(4)	6095.2(16)	13280(4)	32.6(7)	
C(8')	14296(4)	6141.7(16)	11807(4)	33.2(7)	
C(9')	11908(5)	5869.6(15)	15201(4)	30.8(7)	

Table S7: Atomic coordinates ($x \ 10^4$) and equivalent isotropic displacement parameters (Å²x 10³). U(eq) is defined as one third of the trace of the orthogonalized U_{ij} tensor.

C(10)	10(12(5)	(1110)	15115(4)	29.2(0)
$C(10^{\circ})$	10613(5)	6414.0(18)	15115(4)	38.3(8)
C(11')	10114(4)	6601.4(15)	13577(4)	31.3(7)
C(12')	11084(5)	7030.6(15)	12914(4)	32.9(7)
C(13')	11213(5)	7044.9(15)	11421(4)	33.9(7)
C(14')	10354(4)	6624.9(15)	10547(4)	29.6(7)
C(15')	9043(4)	6284.4(14)	11185(4)	29.4(7)
C(16')	8926(4)	6272.2(14)	12671(4)	30.3(7)
C(17')	11188(4)	4987.3(14)	10300(3)	27.0(6)
C(18')	11758(5)	4775.6(17)	9052(3)	33.3(7)
C(19')	8643(5)	4282.5(16)	9273(4)	35.5(7)

S(1)-C(18)	1.720(4)	S(1')-C(18')	1.728(4)
S(1)-C(19)	1.748(4)	S(1')-C(19')	1.743(4)
S(2)-C(19)	1.631(4)	S(2')-C(19')	1.644(4)
S(3)-C(19)	1.730(4)	S(3')-C(19')	1.728(4)
S(3)-C(17)	1.747(3)	S(3')-C(17')	1.750(3)
C(1)-C(14)	1.512(5)	C(1')-C(14')	1.518(5)
C(1)-C(2)	1.575(5)	C(1')-C(2')	1.577(5)
C(2)-C(3)	1.521(5)	C(2')-C(3')	1.513(5)
C(3)-C(8)	1.403(5)	C(3')-C(8')	1.393(5)
C(3)-C(4)	1.416(4)	C(3')-C(4')	1.412(5)
C(4)-C(5)	1.398(5)	C(4')-C(5')	1.397(4)
C(4)-C(17)	1.478(4)	C(4')-C(17')	1.472(5)
C(5)-C(6)	1.398(4)	C(5')-C(6')	1.396(4)
C(6)-C(7)	1.402(5)	C(6')-C(7')	1.378(5)
C(6)-C(9)	1.501(5)	C(6')-C(9')	1.521(4)
C(7)-C(8)	1.379(5)	C(7')-C(8')	1.386(5)
C(9)-C(10)	1.589(5)	C(9')-C(10')	1.586(5)
C(10)-C(11)	1.511(5)	C(10')-C(11')	1.510(5)
C(11)-C(16)	1.389(5)	C(11')-C(12')	1.393(5)
C(11)-C(12)	1.395(5)	C(11')-C(16')	1.405(5)
C(12)-C(13)	1.398(5)	C(12')-C(13')	1.389(5)
C(13)-C(14)	1.394(5)	C(13')-C(14')	1.394(5)
C(14)-C(15)	1.390(5)	C(14')-C(15')	1.413(5)
C(15)-C(16)	1.395(5)	C(15')-C(16')	1.382(5)
C(17)-C(18)	1.340(5)	C(17')-C(18')	1.346(5)
C(18)-S(1)-C(19)	97.40(17)	C(5)-C(6)-C(7)	116.6(3)
C(19)-S(3)-C(17)	99.21(16)	C(5)-C(6)-C(9)	120.6(3)
C(14)-C(1)-C(2)	113.0(3)	C(7)-C(6)-C(9)	121.7(3)
C(3)-C(2)-C(1)	113.0(3)	C(8)-C(7)-C(6)	120.6(3)
C(8)-C(3)-C(4)	116.2(3)	C(7)-C(8)-C(3)	122.0(3)
C(8)-C(3)-C(2)	118.1(3)	C(6)-C(9)-C(10)	113.1(3)
C(4)-C(3)-C(2)	123.6(3)	C(11)-C(10)-C(9)	112.9(3)
C(5)-C(4)-C(3)	120.0(3)	C(16)-C(11)-C(12)	117.9(3)
C(5)-C(4)-C(17)	117.2(3)	C(16)-C(11)-C(10)	120.5(3)
C(3)-C(4)-C(17)	122.4(3)	C(12)-C(11)-C(10)	120.3(3)
C(4)-C(5)-C(6)	121.5(3)	C(11)-C(12)-C(13)	119.8(3)

 Table S8: Bond lengths [Å] and angles [°].

C(14)-C(13)-C(12)	120.9(3)	C(7')-C(6')-C(5')	117.2(3)
C(15)-C(14)-C(13)	117.3(3)	C(7')-C(6')-C(9')	121.1(3)
C(15)-C(14)-C(1)	119.7(3)	C(5')-C(6')-C(9')	120.6(3)
C(13)-C(14)-C(1)	121.6(3)	C(6')-C(7')-C(8')	120.2(3)
C(14)-C(15)-C(16)	120.6(3)	C(7')-C(8')-C(3')	121.9(3)
C(11)-C(16)-C(15)	120.5(3)	C(6')-C(9')-C(10')	112.6(3)
C(18)-C(17)-C(4)	126.0(3)	C(11')-C(10')-C(9')	112.9(3)
C(18)-C(17)-S(3)	113.6(3)	C(12')-C(11')-C(16')	116.8(3)
C(4)-C(17)-S(3)	120.2(2)	C(12')-C(11')-C(10')	120.8(3)
C(17)-C(18)-S(1)	118.7(3)	C(16')-C(11')-C(10')	121.0(3)
S(2)-C(19)-S(3)	124.7(2)	C(13')-C(12')-C(11')	121.6(3)
S(2)-C(19)-S(1)	124.2(2)	C(12')-C(13')-C(14')	120.1(3)
S(3)-C(19)-S(1)	111.08(19)	C(13')-C(14')-C(15')	116.9(3)
C(18')-S(1')-C(19')	97.54(17)	C(13')-C(14')-C(1')	122.2(3)
C(19')-S(3')-C(17')	98.52(17)	C(15')-C(14')-C(1')	119.6(3)
C(14')-C(1')-C(2')	112.0(3)	C(16')-C(15')-C(14')	121.0(3)
C(3')-C(2')-C(1')	113.2(3)	C(15')-C(16')-C(11')	120.3(3)
C(8')-C(3')-C(4')	116.4(3)	C(18')-C(17')-C(4')	126.3(3)
C(8')-C(3')-C(2')	117.7(3)	C(18')-C(17')-S(3')	114.5(3)
C(4')-C(3')-C(2')	124.1(3)	C(4')-C(17')-S(3')	119.1(2)
C(5')-C(4')-C(3')	119.2(3)	C(17')-C(18')-S(1')	117.6(3)
C(5')-C(4')-C(17')	119.4(3)	S(2')-C(19')-S(3')	124.3(2)
C(3')-C(4')-C(17')	121.0(3)	S(2')-C(19')-S(1')	124.0(2)
C(6')-C(5')-C(4')	121.6(3)	S(3')-C(19')-S(1')	111.7(2)

 Table S9: Torsion angles [°].

C(14)-C(1)-C(2)-C(3)	-17.2(5)	C(19)-S(3)-C(17)-C(18)	0.5(3)
C(1)-C(2)-C(3)-C(8)	95.1(4)	C(19)-S(3)-C(17)-C(4)	-175.4(3)
C(1)-C(2)-C(3)-C(4)	-68.0(4)	C(4)-C(17)-C(18)-S(1)	175.7(3)
C(8)-C(3)-C(4)-C(5)	-12.7(4)	S(3)-C(17)-C(18)-S(1)	0.1(4)
C(2)-C(3)-C(4)-C(5)	150.7(3)	C(19)-S(1)-C(18)-C(17)	-0.6(3)
C(8)-C(3)-C(4)-C(17)	175.1(3)	C(17)-S(3)-C(19)-S(2)	177.6(2)
C(2)-C(3)-C(4)-C(17)	-21.5(5)	C(17)-S(3)-C(19)-S(1)	-0.8(2)
C(3)-C(4)-C(5)-C(6)	-1.6(5)	C(18)-S(1)-C(19)-S(2)	-177.6(2)
C(17)-C(4)-C(5)-C(6)	171.1(3)	C(18)-S(1)-C(19)-S(3)	0.9(2)
C(4)-C(5)-C(6)-C(7)	15.4(4)	C(14')-C(1')-C(2')-C(3')	23.1(4)
C(4)-C(5)-C(6)-C(9)	-152.7(3)	C(1')-C(2')-C(3')-C(8')	-98.6(4)
C(5)-C(6)-C(7)-C(8)	-14.9(5)	C(1')-C(2')-C(3')-C(4')	65.5(4)
C(9)-C(6)-C(7)-C(8)	153.0(3)	C(8')-C(3')-C(4')-C(5')	15.7(5)
C(6)-C(7)-C(8)-C(3)	0.5(5)	C(2')-C(3')-C(4')-C(5')	-148.7(3)
C(4)-C(3)-C(8)-C(7)	13.3(5)	C(8')-C(3')-C(4')-C(17')	-171.3(3)
C(2)-C(3)-C(8)-C(7)	-151.0(3)	C(2')-C(3')-C(4')-C(17')	24.3(5)
C(5)-C(6)-C(9)-C(10)	87.8(4)	C(3')-C(4')-C(5')-C(6')	-1.3(5)
C(7)-C(6)-C(9)-C(10)	-79.7(4)	C(17')-C(4')-C(5')-C(6')	-174.5(3)
C(6)-C(9)-C(10)-C(11)	-6.0(4)	C(4')-C(5')-C(6')-C(7')	-14.1(5)
C(9)-C(10)-C(11)-C(16)	-80.3(4)	C(4')-C(5')-C(6')-C(9')	154.1(3)
C(9)-C(10)-C(11)-C(12)	86.7(4)	C(5')-C(6')-C(7')-C(8')	14.8(5)
C(16)-C(11)-C(12)-C(13)	14.2(5)	C(9')-C(6')-C(7')-C(8')	-153.3(3)
C(10)-C(11)-C(12)-C(13)	-153.1(3)	C(6')-C(7')-C(8')-C(3')	-0.2(5)
C(11)-C(12)-C(13)-C(14)	-0.2(5)	C(4')-C(3')-C(8')-C(7')	-15.2(5)
C(12)-C(13)-C(14)-C(15)	-13.9(5)	C(2')-C(3')-C(8')-C(7')	150.2(3)
C(12)-C(13)-C(14)-C(1)	152.5(3)	C(7')-C(6')-C(9')-C(10')	73.2(4)
C(2)-C(1)-C(14)-C(15)	96.0(4)	C(5')-C(6')-C(9')-C(10')	-94.5(4)
C(2)-C(1)-C(14)-C(13)	-70.0(4)	C(6')-C(9')-C(10')-C(11')	14.3(5)
C(13)-C(14)-C(15)-C(16)	13.9(5)	C(9')-C(10')-C(11')-C(12')	-94.0(4)
C(1)-C(14)-C(15)-C(16)	-152.7(3)	C(9')-C(10')-C(11')-C(16')	72.4(4)
C(12)-C(11)-C(16)-C(15)	-14.3(5)	C(16')-C(11')-C(12')-C(13')	-15.1(5)
C(10)-C(11)-C(16)-C(15)	153.1(3)	C(10')-C(11')-C(12')-C(13')	151.8(3)
C(14)-C(15)-C(16)-C(11)	0.1(5)	C(11')-C(12')-C(13')-C(14')	0.3(5)
C(5)-C(4)-C(17)-C(18)	-33.6(5)	C(12')-C(13')-C(14')-C(15')	14.7(5)
C(3)-C(4)-C(17)-C(18)	138.8(4)	C(12')-C(13')-C(14')-C(1')	-152.1(3)
C(5)-C(4)-C(17)-S(3)	141.7(3)	C(2')-C(1')-C(14')-C(13')	67.6(4)
C(3)-C(4)-C(17)-S(3)	-45.9(4)	C(2')-C(1')-C(14')-C(15')	-98.8(4)

C(13')-C(14')-C(15')-C(16')	-15.0(5)
C(1')-C(14')-C(15')-C(16')	152.2(3)
C(14')-C(15')-C(16')-C(11')	0.2(5)
C(12')-C(11')-C(16')-C(15')	14.7(5)
C(10')-C(11')-C(16')-C(15')	-152.1(3)
C(5')-C(4')-C(17')-C(18')	-148.0(3)
C(3')-C(4')-C(17')-C(18')	39.0(5)
C(5')-C(4')-C(17')-S(3')	30.6(4)
C(3')-C(4')-C(17')-S(3')	-142.4(3)
C(19')-S(3')-C(17')-C(18')	1.6(3)
C(19')-S(3')-C(17')-C(4')	-177.2(3)
C(4')-C(17')-C(18')-S(1')	179.6(3)
S(3')-C(17')-C(18')-S(1')	0.9(4)
C(19')-S(1')-C(18')-C(17')	-2.9(3)
C(17')-S(3')-C(19')-S(2')	177.7(2)
C(17')-S(3')-C(19')-S(1')	-3.4(2)
C(18')-S(1')-C(19')-S(2')	-177.3(2)
C(18')-S(1')-C(19')-S(3')	3.8(2)