

Supporting Information

for

Co-solvation effect on the binding mode of the α -mangostin/ β -cyclodextrin inclusion complex

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Additional data

1. Parameters of α -mangostin

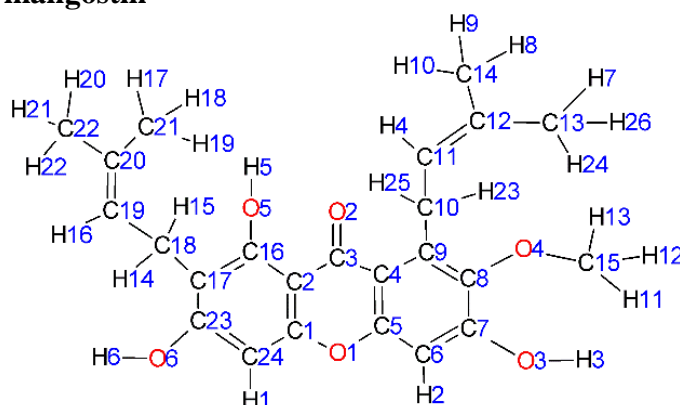


Table S1: Atom types and partial charges of α -mangostin (C₂₄H₂₆O₆)

Name	Type	q	Name	Type	q	Name	Type	q
C13	CT	-0.2454	O5	OH	-0.5789	H5	CA	-0.7066
H24	HC	0.0774	H5	HO	0.4428	H1	HA	0.2645
H26	HC	0.0774	C17	CA	-0.3465	O6	OH	-0.6283
H7	HC	0.0774	C18	CT	0.1081	H6	HO	0.4754

Name	Type	q	Name	Type	q	Name	Type	q
C12	CM	0.1632	C19	CM	-0.429	C1	CA	0.5328
C14	CT	-0.2454	C20	CM	0.1938	O1	OS	-0.3212
H8	HC	0.0774	C21	CT	-0.3281	C5	CA	0.4579
H9	HC	0.0774	H17	HC	0.0989	C6	CA	-0.612
H10	HC	0.0774	H18	HC	0.0989	H2	HA	0.2511
C11	CM	-0.4282	H19	HC	0.0989	C7	CA	0.4545
H4	HA	0.1947	C22	CT	-0.3281	O3	OH	-0.6202
C10	CT	0.0641	H20	HC	0.0989	H3	HO	0.4765
H23	HC	0.0763	H21	HC	0.0989	C8	CA	0.0344
H25	HC	0.0763	H22	HC	0.0989	O4	OS	-0.3791
C9	CA	0.0394	H16	HA	0.2054	C15	CT	0.2301
C4	CA	-0.4221	H14	HC	0.068	H11	HC	0.0188
C3	C	0.7764	H15	HC	0.068	H12	HC	0.0188
O2	O	-0.5921	C23	CA	0.548	H13	HC	0.0188
C2	CA	-0.5643	C16	CA	0.4603			

2. Inclusion complex of α -MGS/ β -CD in a water solvation system

To explore the preferential orientation of α -MGS inside the β -CD cavity, two conformations of inclusion complexes in water (complexes I and II in **Figure S1**) were generated and subjected to MD simulation for 20 ns. For inclusion complex I, the A ring of α -MGS was placed close to the narrow rim of β -CD (O^6 region). On the other hand, the α -MGS in complex II was oriented in the opposite direction.

2.1 System stability

The RMSDs of the α -MGS and β -CD, relative to their initial coordinates, were monitored along the 20 ns of MD simulation and are shown in **Figure S1 (a,b)**. During the first 15 ns, the α -MGS in complex I was positioned closer to the starting structure (0.79 ± 0.31 Å) than that of complex II (1.23 ± 0.28 Å). Meanwhile, the RMSD of β -CD in both complexes was steady at ≈ 1.3 Å. Since two systems apparently reached their equilibration states after 15 ns, the trajectories from the last 5 ns were extracted for structural and binding energy analysis.

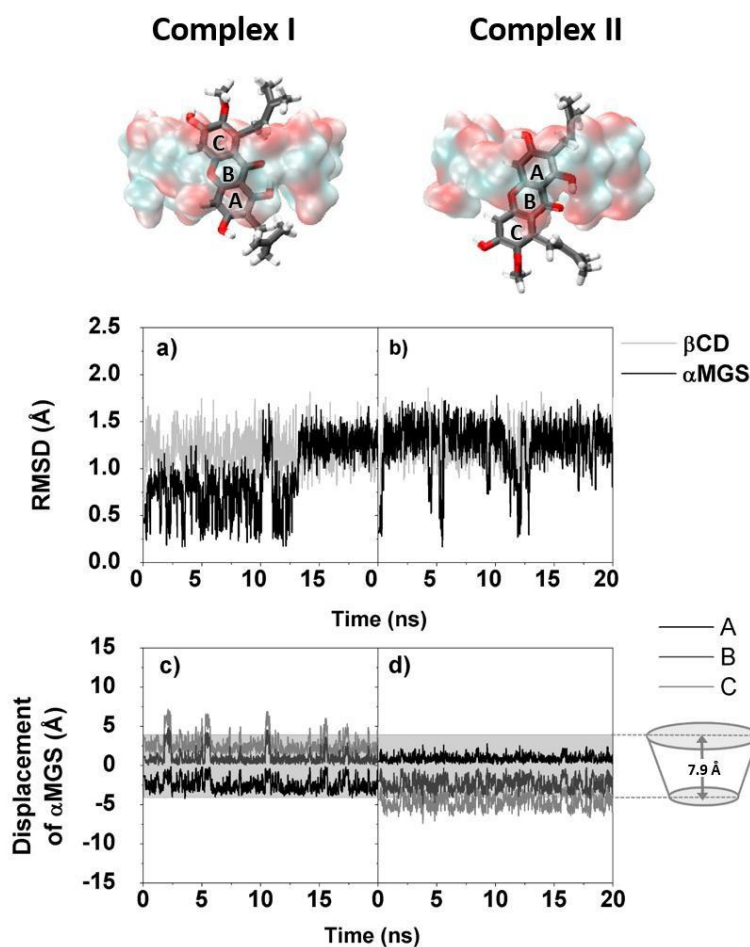


Figure S1: (a,b) RMSD plots of α -MGS and β -CD with respect to their initial structures, and (c,d) displacement plots of the α -MGS rings A–C versus simulation time for the inclusion complexes I and II. The last snapshot structures are depicted above the graphs, where the light blue shade represents the β -CD cavity.

2.2 Displacement of α -mangostin

From the displacement analysis in **Figure S1 (c-d)**, the α -MGS in complex I fluctuated inside the β -CD cavity while its inclusion in complex II is more stable with only slight oscillation. The α -MGS in complex II is also located deeper in the cavity and more towards the narrow rim of β -CD, as seen by the displacements of the A- and C-rings (approximately at 0.9 and -4.7 Å, respectively).

2.3 Binding energy analysis

MM-PBSA binding free energies of the α -MGS/ β -CD complexes I and II are presented in **Table S1**. It can be seen that the binding affinities of the two forms are likely comparable (ΔG_{bind} of ≈ 9 kcal/mol).

Table S2: MM-PBSA binding free energy and the energy components (kcal/mol) for the α -MGS/ β -CD complexes I and II

	Complex I	Complex II
ΔE_{ele}	-5.33 ± 3.34	-4.61 ± 2.67
ΔE_{vdW}	-38.96 ± 2.86	-37.04 ± 1.93
ΔE_{MM} (1)	-44.29 ± 4.16	-41.65 ± 3.22
ΔG_{nsolv}	-4.68 ± 0.19	-4.53 ± 0.17
ΔG_{psolv}	27.55 ± 4.26	23.83 ± 3.80
ΔG_{solv} (2)	22.87 ± 4.22	19.30 ± 3.72
$\Delta G_{\text{psolv}} + E_{\text{ele}}$	18.94 ± 2.14	19.22 ± 3.00
$\Delta G_{\text{nsolv}} + E_{\text{vdW}}$	-43.64 ± 1.53	-41.57 ± 2.10
$-T\Delta S$ (3)	12.56 ± 2.72	13.29 ± 2.72
ΔG_{bind} (1)+(2)+(3)	-8.86 ± 3.25	-9.06 ± 2.87

3. Solvation accessibility

3.1 Solvation accessibility in pure water or ethanol

The RDF plots of the oxygen atoms of α -MGS complexed with β -CD in pure water, and ethanol, are presented in **Figure S2**. The first solvation shell in water solvation system appears around 2 Å from oxygen atoms (O^2 , O^3 , O^4 and O^6) of the encapsulated α -MGS. Differentially, a much lower accessibility of ethanol is observed in the same solvation shell around the O^3 and O^2 atoms.

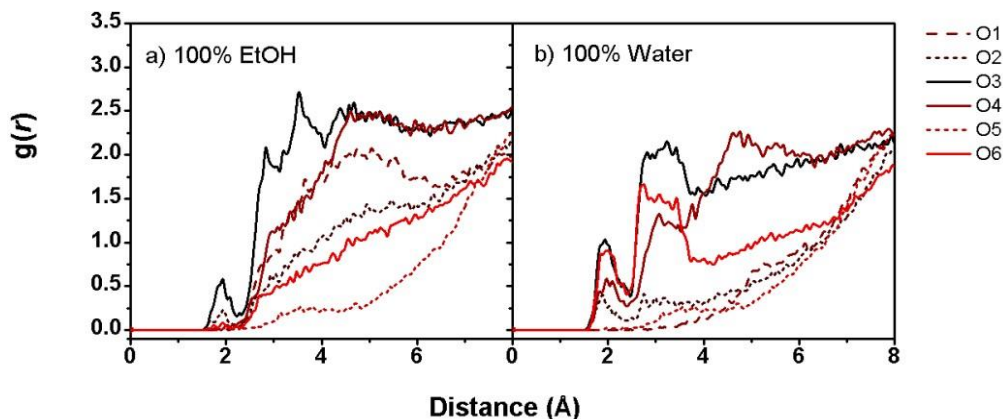


Figure S2: Radial distribution function (RDF, $g(r)$) of (a) ethanol, and (b) water oxygens towards the ligand oxygens (O^1 – O^6) of α -MGS/ β -CD complex.

3.2 Solvation accessibility in co-solvent

Table S3: Integration number, $n(r)$, of the first (2.2 Å) and second solvation shells (3.1 Å) around the heteroatoms of α -MGS in each system

System	$n(r)$ of water molecules		$n(r)$ of ethanol molecules	
	1 st solvation shell	2 nd solvation shell	1 st solvation shell	2 nd solvation shell
Water				
O^1	0.0	0.0	-	-
O^2	0.2	0.9	-	-
O^3	0.7	4.9	-	-
O^4	0.4	2.6	-	-
O^5	0.0	0.2	-	-
O^6	0.7	4.1	-	-
5 %v/v Ethanol				
O^1	0.0	0.0	0.0	0.0
O^2	0.2	1.0	0.0	0.0
O^3	0.7	4.6	0.0	0.2
O^4	0.4	2.4	0.0	0.2
O^5	0.0	0.2	0.0	0.0
O^6	0.6	3.5	0.0	0.4
15 %v/v Ethanol				
O^1	0.0	0.0	0.0	0.1
O^2	0.1	0.4	0.0	0.2
O^3	0.6	3.7	0.0	1.0
O^4	0.2	1.7	0.0	0.6
O^5	0.0	0.2	0.0	0.1
O^6	0.4	2.7	0.0	0.7
30 %v/v Ethanol				

System	$n(r)$ of water molecules		$n(r)$ of ethanol molecules	
	1 st solvation shell	2 nd solvation shell	1 st solvation shell	2 nd solvation shell
O ¹	0.0	0.3	0.0	1.5
O ²	0.0	0.1	0.0	0.2
O ³	0.6	3.2	0.1	1.8
O ⁴	0.2	1.4	0.0	1.0
O ⁵	0.0	0.1	0.0	0.1
O ⁶	0.0	0.6	0.0	0.6
60 %v/v Ethanol				
O ¹	0.0	0.3	0.0	1.5
O ²	0.0	0.2	0.0	0.3
O ³	0.4	2.5	0.2	2.2
O ⁴	0.1	0.8	0.0	1.0
O ⁵	0.0	0.0	0.0	0.1
O ⁶	0.0	0.6	0.0	0.3
Ethanol				
O ¹	-	-	0.0	1.7
O ²	-	-	0.1	1.2
O ³	-	-	0.3	4.1
O ⁴	-	-	0.0	2.0
O ⁵	-	-	0.0	0.7
O ⁶	-	-	0.0	0.9