## Supporting Information

## for

# Experimental and theoretical investigations into the stability of cyclic aminals 

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Detailed synthetic procedures, spectral data, stability analyses and computational investigations

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## Experimental section

## Chemistry

## General

Common reagents and solvents were obtained from commercial suppliers and used without any further purification. Tetrahydrofuran (THF) was distilled from sodium/benzophenone under argon atmosphere. Reaction progress was monitored using analytical thin layer chromatography (TLC) on precoated silica gel GF254 plates (Macherey-Nagel GmbH \& Co. KG, Düren, Germany) and spots were detected under UV light ( 254 nm ) or by staining with iodine. Nuclear magnetic resonance spectra were performed with a Bruker AV-400 NMR instrument (Bruker, Karlsruhe, Germany) in DMSO- $d_{6}$ or $\mathrm{CDCl}_{3}$. Chemical shifts are expressed in ppm relative to $\mathrm{CDCl}_{3}$ or DMSO- $d_{6}$ (7.26/2.50 and 77.16/39.52 ppm for ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR, respectively). Melting points were determined in open capillaries on a Büchi B-540 without any further correction. For purity and reaction analyses, analytical HPLC was performed on a system from Shimadzu equipped with a DGU-20A3R controller, LC20AB liquid chromatograph, and a SPD-20A UV/Vis detector. The stationary phase was a Synergi 4 U fusion-RP $(150 \times 4.6 \mathrm{~mm})$ column (Phenomenex, Aschaffenburg, Germany). As mobile phase, water (phase A) and MeOH (phase B) were used with $1 \mathrm{~mL} / \mathrm{min}$. (conc. B: $5 \% \rightarrow 90 \%$ from 0 to $8 \mathrm{~min} ; 90 \%$ from 8 to $13 \mathrm{~min} ; 90 \% \rightarrow 5 \%$ from 13 to $15 \mathrm{~min} ; 5 \%$ from 15 to 18 min .) ESI mass spectral data were acquired on a Shimadzu LCMS-2020.

## General reaction procedures

General amide formation Procedure (GP1): Isatoic anhydride derivatives were dissolved in dry DMF ( 30 mL ) and treated with the corresponding amine ( 5 equiv) or a mixture of the amine hydrochloride ( 5 equiv) and triethylamine ( 5 equiv). The mixture was heated to $40-$ $120^{\circ} \mathrm{C}$ (depending on the amine) for 3-6 h . For workup, the mixture was poured into water $(100 \mathrm{~mL})$ and the product was extracted with EtOAc $(5 \times 100 \mathrm{~mL})$. The combined organic layers were washed with brine ( 50 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and evaporated to dryness. The crude product was purified by column chromatography to obtain the benzamide derivatives.

General cyclization procedure (GP2): Benzamide derivatives were dissolved in glacial acetic acid ( 20 mL ). The mixture was treated with the corresponding aldehyde (1.2 equiv) and
heated to $70^{\circ} \mathrm{C}$ for $1-4 \mathrm{~h}$. Then the mixture was poured onto ice water ( 20 mL ), basified with a NaOH solution $(2 \mathrm{M})$ and the pH was adjusted to 9 with sat. $\mathrm{NH}_{4} \mathrm{Cl}$ solution. The product was extracted with $\mathrm{EtOAc}(3 \times 40 \mathrm{~mL})$, the combined organic layers were washed with brine ( 30 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was evaporated under reduced pressure. The crude product was either crystallised or purified by column chromatography to obtain dihydroquinazolinones.

General reduction procedure (GP3): Dihydroquinazolinones were dissolved in dry THF ( 30 mL ) at $0{ }^{\circ} \mathrm{C}$ and $\mathrm{LiAlH}_{4}$ (4 equiv) was added. The mixture was allowed to reach rt and was then heated to reflux temperature for $1-3 \mathrm{~h}$. After cooling to rt , the mixture was poured into ice water ( 50 mL ) followed by the addition of saturated $\mathrm{NH}_{4} \mathrm{Cl}$ solution until pH 9 . The aqueous phase was then extracted with EtOAc $(3 \times 80 \mathrm{~mL})$, the combined organic layers were washed with brine ( 50 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was evaporated under reduced pressure. The crude product was purified by column chromatography to obtain the corresponding tetrahydroquinazolines.

## Synthesis and experimental data

1-Methyl-2H-benzo[d][ 1,3$]$ oxazine-2,4(1H)-dione (4): A solution of isatoic anhydride ( $7.60 \mathrm{~g}, 46.6 \mathrm{mmol}, 1$ equiv) in DMAc ( 100 mL ) was treated with DIPEA ( 16.0 mL , $93.2 \mathrm{mmol}, 2$ equiv) and methyl iodide ( $11.6 \mathrm{~mL}, 186 \mathrm{mmol}, 4$ equiv). After stirring for 24 h at $40^{\circ} \mathrm{C}$, ice-cold water $(100 \mathrm{~mL})$ was added. The formed suspension was stirred for further 40 min . The precipitated solid was filtered off, washed with water $(2 \times 15 \mathrm{~mL})$, cyclohexane $(3 \times 15 \mathrm{~mL})$, and dried in vacuo to obtain 1-methyl- 2 H -benzo $[d][1,3]$ oxazine-2,4(1H)-dione $(4,7.35 \mathrm{~g}, 89 \%)$ as yellow brown solid; $\mathbf{m p} 174-177{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}$ NMR (DMSO-d $\left.{ }_{6}, 400 \mathrm{MHz}\right): \delta=$ 8.01 (dd, $J=7.8,1.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.88-7.84(\mathrm{~m}, 1 \mathrm{H}), 7.44(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.36-7.32(\mathrm{~m}$, $1 \mathrm{H}), 3.47(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (DMSO-d ${ }_{6}, 101 \mathrm{MHz}$ ): $\delta=159.0,147.7,142.2,137.1$, 129.3, 123.5, 114.8, 111.5, 31.6 ppm .

N-Methyl-2-(methylamino)benzamide (5a): According to GP1, 1-methyl-2Hbenzo $[d][1,3]$ oxazine- $2,4(1 H)$-dione $(4,1.50 \mathrm{~g}, 8.47 \mathrm{mmol}, 1$ equiv), methylamine hydrochloride ( $2.86 \mathrm{~g}, 42.3 \mathrm{mmol}, 5$ equiv) and triethylamine ( $5.86 \mathrm{~mL}, 42.3 \mathrm{mmol}, 5$ equiv) were used to obtain $N$-methyl-2-(methylamino)benzamide (5a, $1.28 \mathrm{~g}, 92 \%$ ) after column
chromatography (petroleum ether/EtOAc 1:1) as yellow oil; ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=$ 7.41 (br, NH), 7.31-7.27(m, 2H), 6.65-6.63 (d, $J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.57-6.52(\mathrm{~m}, 1 \mathrm{H}), 6.04$ (br, NH), 2.92 (d, $J=4.8 \mathrm{~Hz}, 3 \mathrm{H}$ ), $2.83(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=170.8$, 150.6, 132.9, 127.3, 115.5, 114.7, 111.3, 29.9, 26.7 ppm .

2-(Methylamino)- N -phenylbenzamide (5b): According to GP1, 1-methyl-2H-benzo[d][1,3]oxazine-2,4(1H)-dione (4, $700 \mathrm{mg}, 3.95 \mathrm{mmol}$, 1 equiv), and aniline ( 1.80 mL , 19.8 mmol , 5 equiv) were used to obtain 2 -(methylamino)- N -phenylbenzamide ( $\mathbf{5 b}, 625 \mathrm{mg}$, $70 \%$ ) after column chromatography (petroleum ether/EtOAc 2:1) as pale red solid; mp $125-127{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathbf{H}$ NMR (DMSO- $\mathrm{d}_{6}, 400 \mathrm{MHz}$ ): $\delta=10.05(\mathrm{~s}, \mathrm{NH}), 7.72-7.70(\mathrm{~m}, 2 \mathrm{H}), 7.67$ (dd, $J=7.8,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 7.37-7.29(\mathrm{~m}, 3 \mathrm{H}+\mathrm{NH}), 7.10-7.06(\mathrm{~m}, 1 \mathrm{H}), 6.68(\mathrm{~d}, J=7.4 \mathrm{~Hz}$, $1 \mathrm{H}), 6.64(\mathrm{t}, J=7.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.80(\mathrm{~d}, J=5.0 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (DMSO-d ${ }_{6}, 101 \mathrm{MHz}$ ): $\delta=168.0,150.0,139.1,132.7,128.8,128.4$ (2C), 123.4, 120.6 (2C), 115.7, 114.0, 110.6, 29.4 ppm.

N-Isopropyl-2-(methylamino)benzamide (5c): According to GP1, 1-methyl-2H-benzo[d][1,3]oxazine-2,4(1H)-dione ( $4,700 \mathrm{mg}, 3.95 \mathrm{mmol}, 1$ equiv) and isopropylamine ( $1.69 \mathrm{~mL}, 19.8 \mathrm{mmol}, 5$ equiv) were used to obtain $N$-isopropyl-2-(methylamino)benzamide ( 5 c, $682 \mathrm{mg}, 90 \%$ ) after column chromatography (petroleum ether/EtOAc 2:1) as white solid; mp 119-121 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}$ NMR ( DMSO- $_{6}, 400 \mathrm{MHz}$ ): $\delta=8.00(\mathrm{~d}, J=7.6 \mathrm{~Hz}, \mathrm{NH}), 7.57(\mathrm{q}, J=$ $4.9 \mathrm{~Hz}, \mathrm{NH}), 7.51(\mathrm{dd}, J=7.8,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.28-7.25(\mathrm{~m}, 1 \mathrm{H}), 6.60(\mathrm{dd}, J=8.4,0.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.56-6.52(\mathrm{~m}, 1 \mathrm{H}), 4.12-4.00(\mathrm{~m}, 1 \mathrm{H}), 2.76(\mathrm{~d}, J=5.1 \mathrm{~Hz}, 3 \mathrm{H}), 1.14(\mathrm{~d}, J=6.6 \mathrm{~Hz}$, $6 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (DMSO- $\left.\mathrm{d}_{6}, 101 \mathrm{MHz}\right): \delta=168.2,149.9,132.0,128.3,115.6,113.8$, 110.3, 40.4, 29.2, 22.3 (2C) ppm.

2-(Methylamino)- N -propylbenzamide (5d): According to GP1, 1-methyl-2 H benzo $[d][1,3]$ oxazine- $2,4(1 \mathrm{H})$-dione ( $4,700 \mathrm{mg}, 3.95 \mathrm{mmol}, 1$ equiv) and $n$-propylamine ( $973 \mu \mathrm{~L}, 11.9 \mathrm{mmol}, 3$ equiv) were used to obtain 2-(methylamino)- $N$-propylbenzamide ( $\mathbf{5 d}$, $759 \mathrm{mg}, 97 \%$ ) after column chromatography (petroleum ether/EtOAc 1:2) as pale yellow solid; mp 57-59 ${ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.40(\mathrm{br}, \mathrm{NH}), 7.31-7.29(\mathrm{~m}, 1 \mathrm{H})$, 7.29-7.27 (m, 1H), 6.64 (dd, $J=8.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.58-6.54 (m, 1H), 6.01 (br, NH), 3.37 $3.32(\mathrm{~m}, 2 \mathrm{H}), 2.83(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.60(\operatorname{sextet}, J=7.2 \mathrm{~Hz}, 2 \mathrm{H}), 0.96(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$
ppm. ${ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=170.1,150.7,132.9,127.2,115.7,114.6,111.3,41.6$, 29.9, 23.2, 11.7 ppm .

1,3-Dimethyl-2-phenyl-2,3-dihydroquinazolin-4(1H)-one (6a): According to GP2, $N$-methyl-2-(methylamino)benzamide ( $\mathbf{5 a}, 565 \mathrm{mg}, 3.44 \mathrm{mmol}, 1$ equiv) and benzaldehyde ( $417 \mu \mathrm{~L}$, $4.13 \mathrm{mmol}, 1.2$ equiv) were used to obtain 1,3-dimethyl-2-phenyl-2,3-dihydroquinazolin$4(1 \mathrm{H})$-one ( $\mathbf{6 a}, 628 \mathrm{mg}, 73 \%$ ) after column chromatography (petroleum ether/EtOAc 2:1) as white solid; mp $139-142{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}$ NMR (DMSO- ${ }_{6}, 400 \mathrm{MHz}$ ): $\delta=7.77(\mathrm{dd}, J=7.6,1.4 \mathrm{~Hz}$, $1 \mathrm{H}), 7.37-7.34(\mathrm{~m}, 1 \mathrm{H}), 7.33-7.31(\mathrm{~m}, 3 \mathrm{H}), 7.23-7.18(\mathrm{~m}, 2 \mathrm{H}), 6.78(\mathrm{dt}, J=7.6,0.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.59(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.78(\mathrm{~s}, 1 \mathrm{H}), 2.88(\mathrm{~s}, 3 \mathrm{H}), 2.80(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (DMSO- $\mathrm{d}_{6}, 101 \mathrm{MHz}$ ): $\delta=161.6,146.2,137.0,133.7,128.9,128.7$ (2C), 127.5, 126.0 (2C), 117.3, 115.7, 111.8, 78.3, 34.9, 31.8 ppm . ESI-MS: $m / z$ calcd: 252.1 , found: $253.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: 98\%.

2-(4-(tert-Butyl)phenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{6 b}$ ): According to GP2, $N$-methyl-2-(methylamino)benzamide (5a, $200 \mathrm{mg}, 1.22 \mathrm{mmol}, 1$ equiv) and 4 -(tertbutyl)benzaldehyde ( $245 \mu \mathrm{~L}, 1.45 \mathrm{mmol}$, 1.2equiv) were used to obtain 2-(4-(tert-butyl)phenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{6 b}, 329 \mathrm{mg}, 88 \%$ ) after column chromatography (petroleum ether/EtOAc 3:1) as white solid; mp $126-129{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ): $\delta=7.76$ (dd, $J=7.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.41-7.24(\mathrm{~m}, 3 \mathrm{H}), 7.13(\mathrm{~d}$, $J=8.3 \mathrm{~Hz}, 2 \mathrm{H}), 6.78(\mathrm{t}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.60(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.75(\mathrm{~s}, 1 \mathrm{H}), 2.88(\mathrm{~s}, 3 \mathrm{H})$, $2.81(\mathrm{~s}, 3 \mathrm{H}), 1.23(\mathrm{~s}, 9 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{DMSO}$ ): $\delta=162.1,151.7,146.8,134.6$, $134.2,127.9,126.3$ (2C), 126.0 (2C), 117.8, 116.3, 112.4, 78.5, 35.5, 34.8, 32.4, 31.5 (3C) ppm.

1,3-Dimethyl-2-(p-tolyl)-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{6 c}$ ): According to GP2, N -methyl-2-(methylamino)benzamide (5a, $200 \mathrm{mg}, \quad 1.22 \mathrm{mmol}, \quad 1$ equiv) and 4methylbenzaldehyde ( $175 \mu \mathrm{~L}, 1.45 \mathrm{mmol}, 1.2$ equiv) were used to obtain 1,3-dimethyl-2-( $p$ -tolyl)-2,3-dihydroquinazolin-4( 1 H )-one ( $\mathbf{6 c}, 287 \mathrm{mg}, 88 \%$ ) after column chromatography (petroleum ether/EtOAc 3:1) as white solid; mp 133-136 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO-d $\mathrm{d}_{6}$ ): $\delta=7.76(\mathrm{dd}, J=7.6,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.37-7.28(\mathrm{~m}, 1 \mathrm{H}), 7.16-7.05(\mathrm{~m}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 6.78$ (t, $J=7.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.58(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.73(\mathrm{~s}, 1 \mathrm{H}), 2.87(\mathrm{~s}, 3 \mathrm{H}), 2.78(\mathrm{~s}, 3 \mathrm{H}), 2.25(\mathrm{~s}$,
$3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 101 MHz, DMSO- $\mathrm{d}_{6}$ ): $\delta=162.1,146.7,138.8,134.5,134.2,129.8$ (2C), $127.9,126.5$ (2C), 117.8, 116.2, 112.3, 78.7, 35.3, 32.3, 21.2 ppm .

2-(4-Fluorophenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one (6d): According to GP2, $N$-methyl-2-(methylamino)benzamide (5a, $200 \mathrm{mg}, \quad 1.22 \mathrm{mmol}, 1$ equiv) and 4fluorobenzaldehyde ( $157 \mu \mathrm{~L}, 1.45 \mathrm{mmol}, 1.2$ equiv) were used to obtain 2-(4-fluorophenyl)-1,3-dimethyl-2,3-dihydroquinazolin- $4(1 \mathrm{H}$ )-one ( $\mathbf{6 d}, 209 \mathrm{mg}, 64 \%$ ) after crystallization from a mixture of petroleum ether/DCM as white solid; mp $134-138{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO-d $\mathrm{d}_{6}$ : $\delta=7.77$ (dd, $J=7.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.36 (ddd, $\left.J=8.3,7.3,1.7 \mathrm{~Hz}, 1 \mathrm{H}\right), 7.28-7.21$ (m, 2H), $7.21-7.13(\mathrm{~m}, 2 \mathrm{H}), 6.79(\mathrm{td}, J=7.5,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.60(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.81(\mathrm{~s}$, $1 \mathrm{H}), 2.87(\mathrm{~s}, 3 \mathrm{H}), 2.79(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{DMSO}_{\mathrm{d}}^{6}$ ): $\delta=162.8(\mathrm{~d}, J=$ 244.9 Hz ), 162.0, 146.6, 134.3, 133.8 ( $\mathrm{d}, J=3.1 \mathrm{~Hz}$ ), 128.7 (d, $J=8.4 \mathrm{~Hz}, 2 \mathrm{C}), 128.0,118.0$, $116.1(\mathrm{~d}, J=21.6 \mathrm{~Hz}, 2 \mathrm{C}), 116.1,112.5,78.0,35.3,32.3 \mathrm{ppm}$.

## 1,3-Dimethyl-2-(4-(trifluoromethyl)phenyl)-2,3-dihydroquinazolin-4(1 H )-one ( $\mathbf{6 e}$ ):

According to GP2, $N$-methyl-2-(methylamino)benzamide (5a, $200 \mathrm{mg}, 1.22 \mathrm{mmol}, 1$ equiv) and 4-(trifluoromethyl)benzaldehyde ( $200 \mu \mathrm{~L}, 1.46 \mathrm{mmol}, 1.2$ equiv) were used to obtain 1,3-dimethyl-2-(4-(trifluoromethyl)phenyl)-2,3-dihydroquinazolin-4( $1 H$ )-one ( $\mathbf{6 e}, 354 \mathrm{mg}, 91 \%$ ) after column chromatography (petroleum ether/EtOAc 1:1) as white solid; mp $130-132{ }^{\circ} \mathrm{C}$. ${ }^{1}$ H NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=8.01-7.99(\mathrm{~m}, 1 \mathrm{H}), 7.54(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.36-7.30$ $(\mathrm{m}, 3 \mathrm{H}), 6.86(\mathrm{dt}, J=7.6,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.48(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.45(\mathrm{~s}, 1 \mathrm{H}), 2.99(\mathrm{~s}, 3 \mathrm{H})$, $2.81(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=162.6,146.0,140.9,134.2,131.7(\mathrm{q}, J=$ $32.8 \mathrm{~Hz}), 128.9,126.9(2 \mathrm{C}), 126.2(\mathrm{q}, J=3.8 \mathrm{~Hz}, 2 \mathrm{C}), 124.0(\mathrm{q}, J=273.2 \mathrm{~Hz}), 118.8,116.3$, 112.0, 80.5, 35.8, 32.7 ppm.

2-(2,6-Dichlorophenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{6 f}$ ): According to GP2, $N$-methyl-2-(methylamino)benzamide (5a, $400 \mathrm{mg}, 2.44 \mathrm{mmol}, 1$ equiv) and 2,6dichlorobenzaldehyde ( $512 \mathrm{mg}, 2.93 \mathrm{mmol}, 1.2$ equiv) were used to obtain $2-(2,6-$ dichlorophenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one ( $6 \mathbf{f}, 571 \mathrm{mg}, 73 \%$ ) after column chromatography (petroleum ether/EtOAc 2:1) as white solid; mp $195-201{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.92(\mathrm{dd}, J=7.7,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.28(\mathrm{~m}, 2 \mathrm{H}), 7.26$ (ddd, $J=8.4,7.3,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.21-7.14(\mathrm{~m}, 1 \mathrm{H}), 6.78(\mathrm{~s}, 1 \mathrm{H}), 6.67(\mathrm{td}, J=7.6,0.9 \mathrm{~Hz}$, $1 \mathrm{H}), 6.43(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.72(\mathrm{~s}, 3 \mathrm{H}), 2.62(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ):
$\delta=161.9,146.5,136.1$ (2C), 133.8, 132.3, 130.4, 128.3, 116.7, 113.6, 109.0, 76.0, 33.7, 30.7 ppm.

1-Methyl-2-phenyl-3-propyl-2,3-dihydroquinazolin-4(1H)-one (7a): According to GP2, 2-(methylamino)- $N$-propylbenzamide ( $\mathbf{5 d}, 672 \mathrm{mg}, 3.50 \mathrm{mmol}, 1$ equiv) and benzaldehyde ( $424 \mu \mathrm{~L}, 4.19 \mathrm{mmol}, ~ 1.2 \mathrm{equiv}$ ) were used to obtain 1-methyl-2-phenyl-3-propyl-2,3-dihydroquinazolin- $4(1 \mathrm{H})$-one ( $\mathbf{7 a}, 649 \mathrm{mg}, 66 \%$ ) after column chromatography (petroleum ether/EtOAc 3:1) as white solid; mp 176-180 ${ }^{\circ} \mathrm{C} . \mathbf{~}^{\mathbf{1}} \mathbf{H}$ NMR (DMSO-d ${ }_{6}, 400 \mathrm{MHz}$ ): $\delta=7.77$ (dd, $J=7.6,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.35-7.29$ (m, 4H), $7.25-7.21$ (m, 2H), 6.79 (dt, $J=7.5,0.9 \mathrm{~Hz}$, $1 \mathrm{H}), 6.57(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.78(\mathrm{~s}, 1 \mathrm{H}), 3.79-3.72(\mathrm{~m}, 1 \mathrm{H}), 2.82(\mathrm{~s}, 3 \mathrm{H}), 2.79-2.72(\mathrm{~m}$, $1 \mathrm{H}), 1.67-1.54(\mathrm{~m}, 1 \mathrm{H}), 1.54-1.41(\mathrm{~m}, 1 \mathrm{H}), 0.84(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (DMSO- $\mathrm{d}_{6}, 101 \mathrm{MHz}$ ): $\delta=161.3,146.1,137.7,133.6,128.8,128.6$ (2C), 127.5, 126.2 (2C), $117.5,116.4,112.1,76.5,45.9,35.0,20.5,11.2 \mathrm{ppm}$.

3-Isopropyl-1-methyl-2-phenyl-2,3-dihydroquinazolin-4(1H)-one (7b): According to GP2, N -isopropyl-2-(methylamino)benzamide ( $\mathbf{5 c}, 657 \mathrm{mg}, 3.42 \mathrm{mmol}, 1$ equiv) and benzaldehyde ( $414 \mu \mathrm{~L}, 4.1 \mathrm{mmol}, 1.2$ equiv) were used to obtain 3-isopropyl-1-methyl-2-phenyl-2,3-dihydroquinazolin- $4(1 \mathrm{H}$ )-one ( $\mathbf{7 b}, 820 \mathrm{mg}, 86 \%$ ) after column chromatography (petroleum ether/EtOAc 2:1) as white solid; mp 153-155 ${ }^{\circ} \mathrm{C}$. ${ }^{\mathbf{1}} \mathbf{H}$ NMR (DMSO-d ${ }_{6}, 400 \mathrm{MHz}$ ): $\delta=7.96$ (dd, $J=7.6,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.49-7.45(\mathrm{~m}, 1 \mathrm{H}), 7.45-7.40(\mathrm{~m}, 5 \mathrm{H}), 6.97(\mathrm{dt}, J=7.5,0.8 \mathrm{~Hz}$, $1 \mathrm{H}), 6.67(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.00(\mathrm{~s}, 1 \mathrm{H}), 4.84($ septet, $J=6.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.01(\mathrm{~s}, 3 \mathrm{H}), 1.42(\mathrm{~d}$, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.03(\mathrm{~d}, J=6.9 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{DMSO}_{6}, 101 \mathrm{MHz}\right): \delta=161.1$, 145.7, 138.8, 133.4, 128.4, 128.3 (2C), 127.5, 126.0 (2C), 117.8, 117.7, 112.5, 72.6, 45.2, 35.0, 20.12, 20.11 ppm .

1-Methyl-2,3-diphenyl-2,3-dihydroquinazolin-4(1H)-one (7c): According to GP2, 2-(methylamino)- $N$-phenylbenzamide ( $\mathbf{5 b}, 604 \mathrm{mg}, 2.67 \mathrm{mmol}, 1$ equiv) and benzaldehyde (324 $\mu \mathrm{L}, 3.20 \mathrm{mmol}, 1.2$ equiv) were used to obtain 1-methyl-2,3-diphenyl-2,3-dihydroquinazolin- $4(1 \mathrm{H})$-one ( $7 \mathrm{c}, 772 \mathrm{mg}, 93 \%$ ) after column chromatography (petroleum ether/EtOAc 3:1) as white solid; mp $138-142{ }^{\circ} \mathbf{C} .{ }^{1} \mathbf{H}$ NMR (DMSO-d ${ }_{6}, 400 \mathrm{MHz}$ ): $\delta=7.85$ (dd, $J=7.7,1.6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.43-7.39 (m, 1H), 7.39-7.34 (m, 2H), 7.31-7.28 (m, 3H), 7.27$7.22(\mathrm{~m}, 5 \mathrm{H}), 6.85(\mathrm{dt}, J=7.6,0.7 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.22(\mathrm{~s}, 1 \mathrm{H}), 2.97(\mathrm{~s}$, 3H) ppm. ${ }^{13} \mathbf{C}$ NMR (DMSO- $\mathrm{d}_{6}, 101 \mathrm{MHz}$ ): $\delta=161.2,146.3,140.5,137.5,134.2,128.8$ (2C),
128.7, 128.6 (2C), 128.1, 126.53 (2C), 126.48, 126.2 (2C), 117.9, 116.7, 112.8, 79.0, 35.5 ppm.

1,3-Dimethyl-2-phenyl-1,2,3,4-tetrahydroquinazoline (8a): According to GP3, starting from 1,3-dimethyl-2-phenyl-2,3-dihydroquinazolin- 4 ( 1 H )-one ( $\mathbf{6 a}, 250 \mathrm{mg}, 0.99 \mathrm{mmol}, 1$ equiv) the title compound 1,3-dimethyl-2-phenyl-1,2,3,4-tetrahydroquinazoline ( $\mathbf{8 a}, 208 \mathrm{mg}, 89 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 2:1) as brown solid; mp 94-96 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.30-7.19(\mathrm{~m}, 3 \mathrm{H}), 7.19-7.10(\mathrm{~m}, 3 \mathrm{H}), 6.80(\mathrm{~d}$, $J=7.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.61-6.55(\mathrm{~m}, 2 \mathrm{H}), 4.79(\mathrm{~s}, 1 \mathrm{H}), 3.71(\mathrm{~d}, J=16.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.30(\mathrm{~d}, J=16.0$ $\mathrm{Hz}, 1 \mathrm{H}), 2.88(\mathrm{~s}, 3 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=143.8,141.0$, 128.6 (2C), 128.02, 127.95, 127.3 (3C), 118.2, 116.0, 109.0, 82.3, 49.5, 42.5, 36.9 ppm. ESIMS: $m / z$ calcd: 238.2, found: $239.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.

2-(4-(tert-Butyl)phenyl)-1,3-dimethyl-1,2,3,4-tetrahydroquinazoline (8b): According to GP3, starting from 2-(4-(tert-butyl)phenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{6 b}$ ) ( $309 \mathrm{mg}, 1.00 \mathrm{mmol}, 1$ equiv) the title compound 2-(4-(tert-butyl)phenyl)-1,3-dimethyl-1,2,3,4-tetrahydroquinazoline ( $\mathbf{8 b}, 222 \mathrm{mg}, 75 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 4:1) as white solid; mp $96-99{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $\mathrm{d}_{6}$ ): $\delta=7.41-7.34(\mathrm{~m}, 2 \mathrm{H}), 7.20-7.07(\mathrm{~m}, 3 \mathrm{H}), 6.87(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.70(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 1 \mathrm{H})$, $6.59(\mathrm{td}, J=7.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.98(\mathrm{~s}, 1 \mathrm{H}), 3.60(\mathrm{~d}, J=16.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.35(\mathrm{~d}, J=14.8 \mathrm{~Hz}$, $1 \mathrm{H}), 2.96(\mathrm{~s}, 3 \mathrm{H}), 2.44(\mathrm{~s}, 3 \mathrm{H}), 1.30(\mathrm{~s}, 9 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( 101 MHz, DMSO- $\mathrm{d}_{6}$ ): $\delta=150.2$, 144.0, 138.5, 128.1, 127.1, 126.9 (2C), 125.4 (2C), 118.0, 115.7, 109.2, 80.9, 49.1, 42.0, 37.0, 34.7, 31.6 (3C) ppm. ESI-MS: $m / z$ calcd: 294.2, found: $295.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $100 \%$.

1,3-Dimethyl-2-(p-tolyl)-1,2,3,4-tetrahydroquinazoline (8c): According to GP3, starting from 1,3-dimethyl-2-( $p$-tolyl)-2,3-dihydroquinazolin-4( 1 H )-one ( $\mathbf{6 c}, 269 \mathrm{mg}, 1.01 \mathrm{mmol}, 1$ equiv) the title compound 1,3-dimethyl-2-(p-tolyl)-1,2,3,4-tetrahydroquinazoline ( $\mathbf{8 c}, 203 \mathrm{mg}, 80 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 4:1) as yellow solid; mp $34-36{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}_{\mathrm{d}}^{6}$ ): $\delta=7.15-7.06(\mathrm{~m}, 3 \mathrm{H}), 7.05-7.00(\mathrm{~m}, J=8.0$ $\mathrm{Hz}, 2 \mathrm{H}), 6.82(\mathrm{~d}, J=7.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.65(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 1 \mathrm{H}), 6.54(\mathrm{td}, J=7.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.91$ $(\mathrm{s}, 1 \mathrm{H}), 3.55(\mathrm{~d}, J=16.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.31(\mathrm{~d}, J=16.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.88(\mathrm{~s}, 3 \mathrm{H}), 2.37(\mathrm{~s}, 3 \mathrm{H}), 2.26$ (s, 3H) ppm. ${ }^{13} \mathrm{C}$ NMR (101 MHz, DMSO-d ${ }_{6}$ ): $\delta=144.0$, 138.4, 137.1, 129.2 (2C), 128.1,
127.2 (2C), 127.1, 118.1, 115.7, 109.2, 81.0, 49.1, 42.0, 36.8, 21.1 ppm . ESI-MS: $m / z$ calcd: 252.2, found: $253.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.

2-(4-Fluorophenyl)-1,3-dimethyl-1,2,3,4-tetrahydroquinazoline (8d): According to GP3, starting from 2-(4-fluorophenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{8 c}, 200 \mathrm{mg}$, $0.74 \mathrm{mmol}, 1$ equiv) the title compound 2-(4-fluorophenyl)-1,3-dimethyl-1,2,3,4tetrahydroquinazoline ( $\mathbf{8 d}, 94 \mathrm{mg}, 50 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 2:1) as yellow oil; ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $\mathrm{d}_{6}$ ): $\delta=7.21-7.03$ (m, 5H), 6.84 (d, $J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 6.56(\mathrm{td}, J=7.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.98$ (s, 1H), $3.53(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.33(\mathrm{~d}, J=12.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.91(\mathrm{~s}, 3 \mathrm{H}), 2.38(\mathrm{~s}, 3 \mathrm{H})$ ppm. ${ }^{13} \mathrm{C}$ NMR ( 101 MHz , DMSO-d ${ }_{6}$ ): $\delta=161.9(\mathrm{~d}, J=243.1 \mathrm{~Hz}$ ), 143.7, $137.6(\mathrm{~d}, J=2.9$ $\mathrm{Hz}), 129.2(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 2 \mathrm{C}), 128.1,127.2,118.0,115.5(\mathrm{~d}, J=21.3 \mathrm{~Hz}, 2 \mathrm{C}), 115.4,109.3$, 80.4, 48.9, 42.0, 36.9 ppm. ESI-MS: $m / z$ calcd: 256.1, found: $257.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $96 \%$.

1,3-Dimethyl-2-(4-(trifluoromethyl)phenyl)-1,2,3,4-tetrahydroquinazoline (8e): According to GP3, starting from 1,3-dimethyl-2-(4-(trifluoromethyl)phenyl)-2,3-dihydroquinazolin-4(1 H )one ( $\mathbf{6 e}, 337 \mathrm{mg}, 1.05 \mathrm{mmol}, 1$ equiv) the title compound 1,3-dimethyl-2-(4-(trifluoromethyl)phenyl)-1,2,3,4-tetrahydroquinazoline ( $\mathbf{8 e}, 224 \mathrm{mg}, 70 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 4:1) as clear oil; ${ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}): \delta=7.53,(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 2 \mathrm{H}), 7.35-7.33(\mathrm{~m}, 2 \mathrm{H}), 7.19(\mathrm{~m}, 1 \mathrm{H}), 6.88-6.86(\mathrm{~m}, 1 \mathrm{H})$, $6.68(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.66(\mathrm{dt}, J=7.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.88(\mathrm{~s}, 1 \mathrm{H}), 3.70(\mathrm{~d}, 16.2 \mathrm{~Hz}, 1 \mathrm{H})$, 3.39-3.35(m, 1H), $2.96(\mathrm{~s}, 3 \mathrm{H}), 2.51(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=145.2$, $143.4,130.2$ (q, $J=32.4 \mathrm{~Hz}, 2 \mathrm{C}), 128.2,127.7$ (2C), $127.4,125.6(\mathrm{q}, J=3.8 \mathrm{~Hz}), 124.4$ (q, $J$ $=273.7 \mathrm{~Hz}), 118.0,116.5,109.3,81.9,49.4,42.6,37.1 \mathrm{ppm}$. ESI-MS: $m / z$ calcd: 306.1, found: $307.0[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $97 \%$.

2-(2,6-Dichlorophenyl)-1,3-dimethyl-1,2,3,4-tetrahydroquinazoline (8f): According to GP3, starting from 2-(2,6-dichlorophenyl)-1,3-dimethyl-2,3-dihydroquinazolin-4(1H)-one (6f) ( $520 \mathrm{mg}, 1.63 \mathrm{mmol}, 1$ equiv) the title compound 2 -(2,6-dichlorophenyl)-1,3-dimethyl-1,2,3,4-tetrahydroquinazoline ( $\mathbf{8 f}, 386 \mathrm{mg}, 78 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 5:1) as white solid; mp $78-79{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=$ 7.24 (br, 2H), $7.14-7.04(\mathrm{~m}, 2 \mathrm{H}), 6.89(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.57(\mathrm{td}, J=7.3,1.0 \mathrm{~Hz}, 1 \mathrm{H})$, $6.49(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 1 \mathrm{H}), 5.40(\mathrm{~d}, J=0.9 \mathrm{~Hz}, 1 \mathrm{H}), 3.83(\mathrm{~d}, J=15.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.41(\mathrm{~d}, J=15.0$
$\mathrm{Hz}, 1 \mathrm{H}$ ), $2.68(\mathrm{~s}, 3 \mathrm{H}), 2.36(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=144.2,135.2$, 129.0, 128.0, 126.5, 119.3, 115.7, 109.0, 79.9, 52.6, 42.2, 34.7 ppm. ESI-MS: $m / z$ calcd: 306.1, found: $307.0[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.

1-Methyl-2-phenyl-3-propyl-1,2,3,4-tetrahydroquinazoline (9a): According to GP3, starting from 1-methyl-2-phenyl-3-propyl-2,3-dihydroquinazolin-4(1H)-one (7a, $350 \mathrm{mg}, 1.25 \mathrm{mmol}$, 1 equiv) the title compound 1-methyl-2-phenyl-3-propyl-1,2,3,4-tetrahydroquinazoline ( $9 \mathbf{a}$, $277 \mathrm{mg}, 84 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 3:1) as beige solid; mp: 46-48 ${ }^{\circ} \mathbf{C}^{\mathbf{1}} \mathbf{H} \mathbf{~ N M R ~}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.24-7.20(\mathrm{~m}, 2 \mathrm{H}), 7.19-7.13$ $(\mathrm{m}, 3 \mathrm{H}), 7.13-7.09(\mathrm{~m}, 1 \mathrm{H}), 6.79(\mathrm{~d}, J=7.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.59-6.53(\mathrm{~m}, 2 \mathrm{H}), 4.90(\mathrm{~s}, 1 \mathrm{H}), 3.68$ $(\mathrm{d}, J=16.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.37-3.32(\mathrm{~m}, 1 \mathrm{H}), 2.90(\mathrm{~s}, 3 \mathrm{H}), 2.57-2.42(\mathrm{~m}, 2 \mathrm{H}), 1.56($ sextet, $J=$ $7.3 \mathrm{~Hz}, 2 \mathrm{H}), 0.89(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=144.2,141.6$, 128.6 (2C), 127.9, 127.7, 127.24 (2C), 127.16, 118.6, 115.7, 108.7, 80.4, 55.6, 47.6, 37.0, 21.6, 12.1 ppm . ESI-MS: $m / z$ calcd: 266.2, found: $267.2[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $100 \%$.

3-Isopropyl-1-methyl-2-phenyl-1,2,3,4-tetrahydroquinazoline (9b): According to GP3, starting from 3-isopropyl-1-methyl-2-phenyl-2,3-dihydroquinazolin-4( 1 H )-one ( $\mathbf{7 b}, 300 \mathrm{mg}$, $1.07 \mathrm{mmol}, 1$ equiv) the title compound 3-isopropyl-1-methyl-2-phenyl-1,2,3,4tetrahydroquinazoline ( $\mathbf{9 b}, 247 \mathrm{mg}, 87 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 4:1) as yellow solid; $\mathbf{m p}: 40-42^{\circ} \mathrm{C} .{ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta$ $=7.31-7.28(\mathrm{~m}, 1 \mathrm{H}), 7.27-7.26(\mathrm{~m}, 1 \mathrm{H}), 7.24-7.20(\mathrm{~m}, 3 \mathrm{H}), 7.18-7.13(\mathrm{~m}, 1 \mathrm{H}), 6.85-$ $6.83(\mathrm{~m}, 1 \mathrm{H}), 6.62-6.58(\mathrm{~m}, 2 \mathrm{H}), 5.19(\mathrm{~s}, 1 \mathrm{H}), 3.73(\mathrm{~d}, J=16.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.62(\mathrm{~d}, J=16.6$, $1 \mathrm{H}), 2.94(\mathrm{~s}, 3 \mathrm{H}), 2.87$ (septet, $J=6.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.20(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.23(\mathrm{~d}, J=6.4 \mathrm{~Hz}$, 3H) ppm. ${ }^{13} \mathbf{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=144.9,142.3,128.6$ (2C), 127.9, 127.5, 127.2 (2C), 126.6, 119.5, 115.6, 108.5, 77.6, 50.0, 44.7, 36.6, 22.1, 21.7 ppm . ESI-MS: $m / z$ calcd: 266.2, found: $267.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.

1-Methyl-2,3-diphenyl-1,2,3,4-tetrahydroquinazoline ( $\mathbf{9 c}$ ): According to GP3, starting from 1-methyl-2,3-diphenyl-2,3-dihydroquinazolin-4( 1 H )-one ( $\mathbf{7 c}, 310 \mathrm{mg}, 0.99 \mathrm{mmol}, 1$ equiv) the title compound 1-methyl-2,3-diphenyl-1,2,3,4-tetrahydroquinazoline ( $\mathbf{9 c}, 178 \mathrm{mg}, 61 \%$ ) was obtained after crystallization from petroleum ether/DCM as pale yellow solid; mp 132$134{ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.25-7.22(\mathrm{~m}, 1 \mathrm{H}), 7.22-7.17(\mathrm{~m}, 4 \mathrm{H}), 7.17-$ $7.14(\mathrm{~m}, 2 \mathrm{H}), 7.10-7.06(\mathrm{~m}, 1 \mathrm{H}), 7.00-6.97(\mathrm{~m}, 2 \mathrm{H}), 6.86-6.84(\mathrm{~m}, 1 \mathrm{H}), 6.80(\mathrm{tt}, J=7.3$, $1.0 \mathrm{~Hz}, 1 \mathrm{H}), 6.60-6.56(\mathrm{~m}, 2 \mathrm{H}), 5.77(\mathrm{~s}, 1 \mathrm{H}), 4.16(\mathrm{~d}, J=16.4 \mathrm{~Hz}, 1 \mathrm{H}), 4.10(\mathrm{~d}, J=16.2 \mathrm{~Hz}$,
$1 \mathrm{H}), 2.96$ (s, 3H) ppm. ${ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=150.5,144.4,140.5,129.4$ (2C), 128.7 (2C), 128.1, 128.0, 127.1 (2C), 126.4, 120.8, 120.0, 118.6 (2C), 116.6, 110.2, 78.8, 46.7, 37.2 ppm . ESI-MS: $m / z$ calcd: 300.2, found: $301.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.

2-Amino- $N$-methylbenzamide (10): According to GP1, isatoic anhydride ( $1.00 \mathrm{~g}, 6.13 \mathrm{mmol}$, 1 equiv), methylamine hydrochloride ( $2.07 \mathrm{~g}, 30.7 \mathrm{mmol}, 5$ equiv) and triethylamine ( $4.25 \mathrm{~mL}, 30.7 \mathrm{mmol}, 5$ equiv) were used to obtain 2-amino- $N$-methylbenzamide ( $\mathbf{1 0}, 862 \mathrm{mg}$, $94 \%$ ) after column chromatography (petroleum ether/EtOAc 1:1) as white solid; mp: 80$82{ }^{\circ}{ }^{\circ}{ }^{1}{ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.27(\mathrm{dd}, J=7.8,1.48 \mathrm{~Hz}, 1 \mathrm{H}), 7.20-7.16(\mathrm{~m}, 1 \mathrm{H})$, $6.66(\mathrm{dd}, J=8.2,0.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.64-6.60(\mathrm{~m}, 1 \mathrm{H}), 6.02(\mathrm{br}, \mathrm{NH}), 5.47\left(\mathrm{br}, \mathrm{NH}_{2}\right), 2.95(\mathrm{~d}, J=$ $4.8 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=170.2,148.8,132.4,127.2,117.5,116.8$, 116.5, 26.7 ppm.

3-Methyl-2-phenyl-2,3-dihydroquinazolin-4(1H)-one (11): According to GP2, 2-amino- $N$ methylbenzamide ( $\mathbf{1 0}, 529 \mathrm{mg}, 3.52 \mathrm{mmol}, 1$ equiv) and benzaldehyde ( $436 \mu \mathrm{~L}, 4.23 \mathrm{mmol}$, 1.2 equiv) were used to obtain 3-methyl-2-phenyl-2,3-dihydroquinazolin-4(1H)-one (11, $628 \mathrm{mg}, 75 \%$ ) after crystallization from a mixture of petroleum ether/DCM as white solid; $\mathbf{m p}$ $165-167{ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}$ NMR (DMSO-d ${ }_{6}, 400 \mathrm{MHz}$ ): $\delta=7.65(\mathrm{dd}, J=7.6,1.4 \mathrm{~Hz}, \mathrm{NH}), 7.39-7.35$ $(\mathrm{m}, 2 \mathrm{H}), 7.34-7.30(\mathrm{~m}, 4 \mathrm{H}), 7.22-7.18(\mathrm{~m}, 1 \mathrm{H}), 6.67-6.62(\mathrm{~m}, 2 \mathrm{H}), 5.82(\mathrm{~d}, J=2.3 \mathrm{~Hz}$, $1 \mathrm{H}), 2.85(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (DMSO-d ${ }_{6}, 101 \mathrm{MHz}$ ): $\delta=162.5,146.4,140.6,133.2$, 128.6 (2C), 128.5, 127.3, 126.1 (2C), 116.9, 114.3, 114.0, 72.0, 32.0 ppm .

3-Methyl-2-phenyl-1-propyl-2,3-dihydroquinazolin-4(1H)-one (12a): Starting from 3-methyl-2-phenyl-2,3-dihydroquinazolin-4( $1 H$ )-one ( $\mathbf{1 1}, 236 \mathbf{m g}, 0.99 \mathrm{mmol}, 1$ equiv) dissolved in dry DMF ( 10 mL ), $n$-propyl bromide ( $452 \mu \mathrm{~L}, 4.96 \mathrm{mmol}, 5$ equiv) and $t$-BuOK ( 557 mg , $4.96 \mathrm{mmol}, 5$ equiv) were added successively. The reaction mixture was stirred for 16 h at $110{ }^{\circ} \mathrm{C}$. Ice-cold water ( 30 mL ) was then added and the mixture was stirred for further 30 min . The product was extracted with EtOAc $(3 \times 30 \mathrm{~mL})$, the combined organic layers were washed with brine ( 30 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was evaporated under reduced pressure to yield 3-methyl-2-phenyl-1-propyl-2,3-dihydroquinazolin-4(1H)-one (12a, 200 mg , $72 \%$ ) after column chromatography (petroleum ether/EtOAc 1:1) as white solid; mp 116-119 ${ }^{\circ}{ }^{\circ}$ C. ${ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.97(\mathrm{dd}, J=7.7,1.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.32-7.24(\mathrm{~m}, 4 \mathrm{H}), 7.23$ $-7.20(\mathrm{~m}, 2 \mathrm{H}), 6.80-6.76(\mathrm{~m}, 1 \mathrm{H}), 6.57(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 5.43(\mathrm{~s}, 1 \mathrm{H}), 3.35-3.28(\mathrm{~m}$,
$1 \mathrm{H}), 3.05-2.97(\mathrm{~m}, 4 \mathrm{H}), 1.72-1.59(\mathrm{~m}, 1 \mathrm{H}), 1.57-1.46(\mathrm{~m}, 1 \mathrm{H}), 0.92(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H})$ ppm. ${ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=162.8,146.1,139.0,133.7,129.2,129.0$ (2C), 128.9, 126.5 (2C), 118.0, 116.7, 112.6, 79.6, 51.4, 32.9, 20.7, 11.6 ppm.

2-(Isopropylamino)benzoic acid (14): To a solution of 2-aminobenzoic acid ( $1 \mathrm{~g}, 7.3 \mathrm{mmol}$, 1 equiv) in methanol ( 20 mL ) was added acetone ( $643 \mu \mathrm{~L}, 8.76 \mathrm{mmol}, 1.2$ equiv). The mixture was heated to reflux temperature and stirred for 5 h . After cooling to room temperature, $\mathrm{NaBH}_{4}$ ( $333 \mathrm{mg}, 8.76 \mathrm{mmol}, 1.2$ equiv) was added portion wise and stirring was continued for further 3 h . The solvent was then evaporated to dryness and suspended in water. The product was extracted with $\mathrm{DCM}(3 \times 150 \mathrm{~mL})$, the organic layer was dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed. The crude product was purified by column chromatography (petroleum ether/EtOAc 4:1) to yield 2-(isopropylamino)benzoic acid (14, $794 \mathrm{mg}, 61 \%$ ) as white solid; mp $110-112{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H}$ NMR ( 400 MHz, DMSO- $\mathrm{d}_{6}$ ): $\delta=12.52(\mathrm{br}, \mathrm{COOH}), 7.78$ (dd, $J=8.0,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.35(\mathrm{ddd}, J=8.7,7.1,1.8 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 6.52$ (ddd, $J=8.0,7.1,1.1 \mathrm{~Hz}, 1 \mathrm{H}), 3.72($ hept $, J=6.2 \mathrm{~Hz}, 1 \mathrm{H}), 1.19(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 6 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR(101 MHz, DMSO-d $\mathrm{d}_{6}$ : $\delta=170.5,150.5,134.9,132.3,114.3,112.1,110.2,43.0,23.1$ (2C) ppm.

2-(Isopropylamino)- $N$-methylbenzamide (15a): A mixture of 2-(isopropylamino)benzoic acid ( $\mathbf{1 4}, 770 \mathrm{mg}, 4.3 \mathrm{mmol}$, 1 equiv), methylamine hydrochloride ( $1.45 \mathrm{~g}, 21.5 \mathrm{mmol}, 5$ equiv), triethylamine ( $2.98 \mathrm{~mL}, 21.5 \mathrm{mmol}, 5$ equiv), EDCI hydrochloride ( $991 \mathrm{mg}, 5.16 \mathrm{mmol}$, 1.2 equiv) and HOBT ( $697 \mathrm{mg}, 5.16 \mathrm{mmol}, 1.2$ equiv) in DMF ( 50 mL ) were heated to $70{ }^{\circ} \mathrm{C}$ for 14 h . The mixture was then poured into water $(150 \mathrm{~mL})$ and extracted with EtOAc ( $4 \times$ 100 mL ). The combined organic layers were washed with brine ( 100 mL ), dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was removed under reduced pressure. The crude product was purified by column chromatography (petroleum ether/EtOAc 4:1) to yield 2-(isopropylamino)- N methylbenzamide (15a, $627 \mathrm{mg}, 76 \%$ ) as clear oil; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.66$ (br, NH ), $7.35-7.25(\mathrm{~m}, 2 \mathrm{H}), 6.73(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 1 \mathrm{H}), 6.62-6.47(\mathrm{~m}, 1 \mathrm{H}), 6.12(\mathrm{br}, \mathrm{NH}), 3.68$ (hept, $J=6.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), $2.97(\mathrm{~d}, J=4.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.27(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 6 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (101 MHz, $\mathrm{CDCl}_{3}$ ): $\delta=170.6,148.7,132.6,127.4,115.0,114.3,112.3,43.5,26.5,22.7$ (2C) ppm.

N-Methyl-2-(phenylamino)benzamide (15b): To a solution of 2-(phenylamino)benzoic acid ( $1.00 \mathrm{~g}, 4.69 \mathrm{mmol}$, 1 equiv) in dry DMF ( 40 mL ), $1-\mathrm{HOBt}(761 \mathrm{mg}, 5.63 \mathrm{mmol}, 1.2$ equiv), EDCI hydrochloride ( $1.08 \mathrm{~g}, 5.63 \mathrm{mmol}, 1.2$ equiv), methylamine hydrochloride ( 1.59 g , $23.5 \mathrm{mmol}, 4$ equiv), and triethylamine ( $3.25 \mathrm{~mL}, 23.5 \mathrm{mmol}, 5$ equiv) were added successively. After stirring for 8 h at ambient temperature, the reaction mixture was poured onto water ( 60 mL ) and extracted with EtOAc $(3 \times 30 \mathrm{~mL})$. The combined organic layers were washed with a $0.1 \mathrm{M} \mathrm{HCl}(2 \times 20 \mathrm{~mL})$, brine $(30 \mathrm{~mL})$, and dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. The solvent was evaporated under reduced pressure and the crude product was purified by column chromatography (petroleum ether/EtOAc 2:1) to obtain the title compound 15b ( 584 mg , $56 \%$ ) as white solid; $\mathbf{m p} 85-87^{\circ} \mathrm{C} .{ }^{1} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=9.22(\mathrm{br}, \mathrm{NH}), 7.32(\mathrm{dd}$, $J=7.8,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 7.29-7.24(\mathrm{~m}, 1 \mathrm{H}), 7.23-7.16(\mathrm{~m}, 3 \mathrm{H}), 7.13-7.10(\mathrm{~m}, 2 \mathrm{H}), 6.92(\mathrm{tt}, J$ $=7.3,1.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.70-6.66(\mathrm{~m}, 1 \mathrm{H}), 6.09(\mathrm{br}, \mathrm{NH}), 2.91(\mathrm{~d}, J=4.9 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm}$. ${ }^{13} \mathbf{C}$ NMR ( $\left.\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=170.4,145.5,141.8,132.3,129.5$ (2C), 127.6, 122.5, 120.9 (2C), 118.8, 118.1, 115.7, 26.8 ppm .

1-Isopropyl-3-methyl-2-phenyl-2,3-dihydroquinazolin-4(1H)-one (12b): According to GP2, 2-(isopropylamino)- $N$-methylbenzamide (15a, $600 \mathrm{mg}, \quad 3.13 \mathrm{mmol}, \quad 1$ equiv) and benzaldehyde ( $379 \mu \mathrm{~L}, 3.75 \mathrm{mmol}, 1.2$ equiv), were used to obtain 1-isopropyl-3-methyl-2-phenyl-2,3-dihydroquinazolin-4( $1 H$ )-one ( $\mathbf{1 2 b}, 648 \mathrm{mg}, 74 \%$ ) after column chromatography (petroleum ether/EtOAc 3:1) as white solid; mp 107-109 ${ }^{\circ} \mathrm{C} . \mathbf{}^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta$ $=7.92-7.84(\mathrm{~m}, 1 \mathrm{H}), 7.23(\mathrm{ddd}, J=8.3,7.3,1.7 \mathrm{~Hz}, 1 \mathrm{H}), 7.18-7.08(\mathrm{~m}, 5 \mathrm{H}), 6.87-6.79$ (m, 2H), $5.46(\mathrm{~s}, 1 \mathrm{H}), 3.87$ (hept, $J=6.7 \mathrm{~Hz}, 1 \mathrm{H}), 3.10(\mathrm{~s}, 3 \mathrm{H}), 1.29(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.14$ $(\mathrm{d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR ( $101 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=163.3,146.6,140.2,132.9,128.5$ (2C), 128.1, 127.9, 126.3 (2C), 121.8, 120.8, 119.6, 71.9, 55.0, 33.1, 21.8, 20.3 ppm .

3-Methyl-1,2-diphenyl-2,3-dihydroquinazolin-4(1H)-one (12c): According to GP2, N -methyl-2-(phenylamino)benzamide ( $\mathbf{1 5 b}, 454 \mathrm{mg}, 2.01 \mathrm{mmol}, 1$ equiv) and benzaldehyde ( $244 \mu \mathrm{~L}$, $2.41 \mathrm{mmol}, 1.2$ equiv) were used to obtain 3-methyl-1,2-diphenyl-2,3-dihydroquinazolin$4(1 \mathrm{H})$-one ( $\mathbf{1 2 c}, 488 \mathrm{mg}, 78 \%$ ) after crystallization from a mixture of petroleum ether/DCM as white solid; mp $158-160{ }^{\circ} \mathrm{C}$. ${ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=8.00-7.99(\mathrm{~m}, 1 \mathrm{H}), 7.35-$ $7.32(\mathrm{~m}, 2 \mathrm{H}), 7.31-7.26(\mathrm{~m}, 5 \mathrm{H}), 7.26-7.22(\mathrm{~m}, 1 \mathrm{H}), 7.17-7.14(\mathrm{~m}, 2 \mathrm{H}), 7.14-7.12(\mathrm{~m}$, $1 \mathrm{H}), 6.96-6.92(\mathrm{~m}, 1 \mathrm{H}), 6.84-6.81(\mathrm{~m}, 1 \mathrm{H}), 5.93(\mathrm{~s}, 1 \mathrm{H}), 3.16(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR
$\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=162.8,146.3,144.1,139.5,133.1,129.9$ (2C), 129.0 (2C), 128.9, 128.7, 126.6 (2C), 125.0, 123.5 (2C), 121.4, 120.6, 118.9, 80.2, 34.3 ppm.

3-Methyl-2-phenyl-1-propyl-1,2,3,4-tetrahydroquinazoline (13a): According to GP3, starting from 3-methyl-2-phenyl-1-propyl-2,3-dihydroquinazolin-4(1H)-one (12a, $180 \mathrm{mg}, 0.64$ mmol, 1 equiv) the title compound 3-methyl-2-phenyl-1-propyl-1,2,3,4-tetrahydroquinazoline (13a, $140 \mathrm{mg}, 82 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 3:1) as grey solid; $\mathbf{m p} 45-46{ }^{\circ} \mathrm{C} .{ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.31-7.21(\mathrm{~m}, 5 \mathrm{H}), 7.16-7.12$ $(\mathrm{m}, 1 \mathrm{H}), 6.85-6.83(\mathrm{~m}, 1 \mathrm{H}), 6.67(\mathrm{~d}, J=8.2 \mathrm{~Hz}, 1 \mathrm{H}), 6.59(\mathrm{dt}, J=7.3,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 4.91(\mathrm{~s}$, $1 \mathrm{H}), 3.76(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.52-3.41(\mathrm{~m}, 1 \mathrm{H}), 3.30(\mathrm{dd}, J=16.2,1.0 \mathrm{~Hz}, 1 \mathrm{H}), 2.97-$ $2.85(\mathrm{~m}, 1 \mathrm{H}), 2.51(\mathrm{~s}, 3 \mathrm{H}), 1.74-1.54(\mathrm{~m}, 2 \mathrm{H}), 0.92(\mathrm{t}, J=7.4 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=142.9,141.6,128.5$ (2C), 127.89, 127.86, 127.7, 127.4 (2C), 117.6, 115.5, 108.9, 80.9, 50.9, 49.2, 42.5, 20.7, 11.6 ppm. ESI-MS: $m / z$ calcd: 266.2, found: 267.1 $[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.

1-Isopropyl-3-methyl-2-phenyl-1,2,3,4-tetrahydroquinazoline (13b): According to GP3, starting from 1-isopropyl-3-methyl-2-phenyl-2,3-dihydroquinazolin-4(1H)-one ( $\mathbf{1 2 b}, 300 \mathrm{mg}, 1.07 \mathrm{mmol}$, 1 equiv) the title compound 1 -isopropyl-3-methyl-2-phenyl-1,2,3,4-tetrahydroquinazoline (13b, 174 $\mathrm{mg}, 61 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 3:1) as clear oil; ${ }^{1} \mathbf{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=7.25-7.13(\mathrm{~m}, 5 \mathrm{H}), 7.13-7.07(\mathrm{~m}, 1 \mathrm{H}), 6.83-6.73(\mathrm{~m}, 2 \mathrm{H}), 6.55(\mathrm{td}, J$ $=7.3,0.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.04(\mathrm{~s}, 1 \mathrm{H}), 4.18(\mathrm{hept}, J=6.6 \mathrm{~Hz}, 1 \mathrm{H}), 3.71(\mathrm{~d}, J=16.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.25(\mathrm{~d}, J=$ $16.1 \mathrm{~Hz}, 1 \mathrm{H}), 2.40(\mathrm{~s}, 3 \mathrm{H}), 1.20(\mathrm{~d}, J=6.5 \mathrm{~Hz}, 3 \mathrm{H}), 1.00(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR (101 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ : $\delta=142.9,142.3,128.2$ (2C), 127.9, 127.7, 127.5, 127.3 (2C), 117.4, 115.2, 109.3, 74.2, 49.0, 46.6, 42.1, 20.7, 19.9 ppm . ESI-MS: $m / z$ calcd: 266.2, found: $267.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $94 \%$.

3-Methyl-1,2-diphenyl-1,2,3,4-tetrahydroquinazoline (13c): According to GP3, starting from 3-methyl-1,2-diphenyl-2,3-dihydroquinazolin- $4(1 \mathrm{H})$ - one ( $\mathbf{1 2 c}, 200 \mathrm{mg}, 0.64 \mathrm{mmol}$, 1 equiv) the title compound 3-methyl-1,2-diphenyl-1,2,3,4-tetrahydroquinazoline ( $\mathbf{1 3 c}, 153 \mathrm{mg}, 81 \%$ ) was obtained after column chromatography (petroleum ether/EtOAc 3:1) as pale yellow solid; mp 107-109 ${ }^{\circ} \mathrm{C} .{ }^{1} \mathbf{H}$ NMR $\left(\mathrm{CDCl}_{3}, 400 \mathrm{MHz}\right): \delta=7.47-7.44(\mathrm{~m}, 2 \mathrm{H}), 7.33-7.29(\mathrm{~m}, 2 \mathrm{H})$, 7.28-7.25 (m, 3H), 7.25-7.22(m, 2H), 7.10-7.06(m, 1H), 7.06-7.01 (m, 2H), 6.87 (dd, $J$ $=7.5 \mathrm{~Hz}, 0.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.73(\mathrm{dt}, J=7.2,1.3 \mathrm{~Hz}, 1 \mathrm{H}), 5.32(\mathrm{~s}, 1 \mathrm{H}), 3.89(\mathrm{~d}, J=16.4 \mathrm{~Hz}, 1 \mathrm{H})$, $3.47(\mathrm{~d}, J=16.7 \mathrm{~Hz}, 1 \mathrm{H}), 2.63(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=147.7,141.4$,
141.1, 129.5 (2C), 128.7 (2C), 127.8 (2C), 127.5 (2C), 127.0, 123.7, 123.4 (2C), 119.7, 118.9, 115.2, 82.4, 49.3, 42.4 ppm. ESI-MS: $m / z$ calcd: 300.2, found: $301.1[\mathrm{M}+\mathrm{H}]^{+}$. HPLC: $99 \%$.
$N$-Methyl-2-((methylamino)methyl)aniline (16): $N$-methyl-2-(methylamino)benzamide (5a, $300 \mathrm{mg}, 1.83 \mathrm{mmol}, 1$ equiv) was dissolved in dry THF ( 20 mL ) and treated with $\mathrm{LiAlH}_{4}$ ( $278 \mathrm{mg}, 7.32 \mathrm{mmol}, 4$ equiv) under ice cooling. The mixture was heated for 24 h to reflux temperature under argon atmosphere, poured into water ( 30 mL ) and extracted with DCM (3 $\times 30 \mathrm{~mL})$. The combined organic layers were extracted with a 1 M HCl solution $(3 \times 30 \mathrm{~mL})$. The aqueous phase was washed with $\mathrm{DCM}(1 \times 30 \mathrm{~mL})$, basified with concentrated $\mathrm{NH}_{3}$ solution and extracted with DCM $(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and the solvent was evaporated under reduced pressure to obtain N -methyl-2((methylamino)methyl)aniline (16, $247 \mathrm{mg}, 90 \%$ ) as a yellow liquid; . ${ }^{\mathbf{1}} \mathbf{H} \mathbf{N M R}\left(\mathrm{CDCl}_{3}, 400\right.$ $\mathrm{MHz}): \delta=7.13(\mathrm{td}, J=7.7,1.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.97-6.92(\mathrm{~m}, 1 \mathrm{H}), 6.58-6.52(\mathrm{~m}, 2 \mathrm{H}), 3.65(\mathrm{~s}$, $2 \mathrm{H}), 2.76(\mathrm{~s}, 3 \mathrm{H}), 2.33(\mathrm{~s}, 3 \mathrm{H}) \mathrm{ppm} .{ }^{13} \mathbf{C}$ NMR $\left(\mathrm{CDCl}_{3}, 101 \mathrm{MHz}\right): \delta=149.41,129.53$, 128.64, 123.70, 115.99, 109.60, 55.60, 36.07, 30.21 ppm. ESI-MS: $m / z$ calcd: 150.1, found: $151.2[\mathrm{M}+\mathrm{H}]^{+}$.

## Stability tests

## Preparation of phosphate-buffered systems

For the preparation of phosphate-buffered solutions with defined pH values, $0.1 \mathrm{M} \mathrm{H}_{3} \mathrm{PO}_{4}$ was titrated with a 0.3 M NaOH solution while monitoring the pH value on a Metohm 744 pH Meter (Metrohm GmbH \& Co. KG, Filderstadt, Germany). Using this procedure, phosphatebuffered systems were obtained with $\mathrm{pH} 2,3,4,5,6,7,8$ and 12 .

## Hydrolysis experiments

A calibration curve for each test compound and their respective aldehydes as hydrolysis products was recorded. For stability analyses, $300 \mu \mathrm{~L}$ of a solution of the test compound (333 $\mu \mathrm{g} / \mathrm{mL}$ ) in acetonitrile was thoroughly mixed with $700 \mu \mathrm{~L}$ of a phosphate buffered solution with defined pH value to give a solution with a final test concentration of $100 \mu \mathrm{~g} / \mathrm{mL}$ (a certain amount of acetonitrile is necessary to keep the compound in solution). The reaction mixture was incubated for 1 h before it was directly analysed with HPLC by injecting a volume of $20 \mu \mathrm{~L}$ of the test solution. Analytical HPLC was performed on a system from Shimadzu equipped with a DGU-20A3R controller, LC20AB liquid chromatograph, and a SPD-20A UV/Vis detector. Stationary phase was a Synergi 4 U fusion-RP ( $150 \times 4.6 \mathrm{~mm}$ ) column (Phenomenex, Aschaffenburg, Germany). As mobile phase, water (phase A) and MeOH (phase B) were used with $1 \mathrm{~mL} / \mathrm{min}$. (conc. B: $5 \% \rightarrow 90 \%$ from 0 to $8 \mathrm{~min} ; 90 \%$ from 8 to $13 \mathrm{~min} ; 90 \% \rightarrow 5 \%$ from 13 to $15 \mathrm{~min} ; 5 \%$ from 15 to 18 min .). Using the beforehand recorded calibration curves, it was possible to calculate the ration between the intact tetrahydroquinazoline as well as the respective aldehyde as its hydrolysis product. Data were subjected to one way ANOVA followed by Dunnett's multiple comparison post test using GraphPad Prism 4 Software (levels of significance * p < 0.05; ** p < 0.01; *** p < 0.001). Each experiment was performed three times with three independently prepared test solutions.

Kinetic monitoring of the hydrolysis of compound $\mathbf{8 a}$ at different pH values (3, 4 and 5) was followed the above mentioned procedure after $t=0,18.5,37,55.5,74,92,5111,129.5,148$, $166.5,185,203.5,222,240.5,259$ and 277.5 min . Due to the fast hydrolysis at pH 2 , the time interval was shortened to $t=0,1,2,3,4,6,18.5,37$ and 55.5 min . Calculation of $k_{2}$ was performed under assumption of a pseudo-first-order kinetics using
$\mathrm{Y}=\left(\mathrm{Y}_{0}\right) * \exp \left(-k_{2} * t\right)$

With $\mathrm{Y}=$ amount of non-hydrolysed test compound (\%) in dependency of time and $\mathrm{Y}_{0}=$ amount of non-hydrolysed test compound (\%) at $t=0 \mathrm{~min}$. All data were analysed using GraphPad Prism 4 Software.

## Aminal condensation experiments:

To proof that acid-mediated hydrolysis of the described tetrahydroquinazolines can be reversed by changing the reaction conditions to basic ones, the synthesis of tetrahydroquinazoline 8a was performed using $N$-methyl-2-((methylamino)methyl)aniline (16) and benzaldehyde:


Therefore, $N$-methyl-2-((methylamino)methyl)aniline (16, $100 \mathrm{mg}, 0.67 \mathrm{mmol}, 1$ equiv) and benzaldehyde ( 1.2 equiv) were mixed at room temperature in either a) acetonitrile ( 1 mL ) as reference for the condensation, $b$ ) water $(1 \mathrm{~mL})$ and c$)$ water/acetonitrile ( $1: 1,1 \mathrm{~mL}$ ). After 24 h the reaction mixtures were directly analysed by LC-MS (cf. general part).

Chromatogram for the reaction in water:


Chromatogram for the reaction in acetonitrile:


Chromatogram for the reaction in acetonitrile/water:


Interestingly, all reactions showed the same result with complete consumption of N -methyl-2((methylamino)methyl)aniline $\mathbf{1 6}$ at $t_{\mathrm{R}} \approx 5 \mathrm{~min}$ and benzyldehyde at $t_{\mathrm{R}}=2.2 \mathrm{~min}$ while forming the tetrahydroquinazoline $\mathbf{8 a}\left(t_{\mathrm{R}} \approx 7 \mathrm{~min}\right)$. It is remarkable, that there is another peak at $t_{\mathrm{R}} \approx 8 \mathrm{~min}$ for all cases which could not be identified by mass spectrometry. This peak might be a byproduct from the reaction of benzaldehyde due to the fact, that benzaldehyde is completely consumed although being used in excess. However, these results clearly indicated, that the condensation of diamines and aldehydes is possible in neutral aqueous media, while the hydrolysis is promoted in acidic aqueous media (see main manuscript).

## Computational section

## Materials and methods

Conformational search: All compounds prior to the conformational search were minimised in MOE ${ }^{[1]}$ using the MMFF94x forcefield ${ }^{[2]}$ in gas phase $(\varepsilon=1)$ and an RMS-gradient of $0.0001 \mathrm{kcal} / \mathrm{mol}$ Å.

The systematic search was conducted in Macromodel 10.3 (Schrodinger 2014-1 platform) using the Systematic Pseudo Monte Carlo (SPMC) method ${ }^{[3]}$. The conformational search was conducted in MMFF94s forcefield ${ }^{[4]}$ and the potential treatment was set according to the Macromodel built-in module for water $(\varepsilon=7.8000 \mathrm{E}+1)$. The convergence threshold for minimization was set at $0.0001 \mathrm{kcal} / \mathrm{mol} \AA$ and the maximum iteration limit was set at 25000 . Miscellaneous technical characteristics: Maximum number of steps 100000; steps per rotational bond 10000 ; energy window $478.01 \mathrm{kcal} / \mathrm{mol}$; maximum atom deviation cut-off: 0.01 A

MM relative ring strain comparison: The transformation of the structures and the potential energy calculation was conducted in the Schrodinger 2014-1 platform environment, using the MMFF94s forcefield. The potential treatment for calculations in water environment was set according to the Macromodel built-in module for water as previously described.

QM single point energy calculations: All tetrahydroquinazoline structures used were results of the aforementioned MM systematic conformational search. All other structures were designed and minimised in Maestro (Schrodinger platform) using the MMFF94s forcefield. The Brønsted acid was simulated by an ammonium cation to facilitate the calculations.

The single point calculations were performed with DFT using the functional B3LYP-D3 ${ }^{[5-9]}$ (grid m4 ${ }^{[10]}$ ) in Turbomole (dscf ${ }^{[11,12]}$ ). The basis set used was cc-pVTZ ${ }^{[13,14]}$.

Calculations were performed in water environment $(\varepsilon=78.5)$ and in gas phase. The Conductor-Like Screening Model (COSMO) ${ }^{[15]}$ was used to form the implicit water continuum solvation model. The studied structures were optimised in water and gas phase, respectively.
$\underline{\text { MoKa }^{[16]} \text { calculations }}$

| Structure | $\mathbf{N} 1 \mathbf{p} \boldsymbol{K}_{\mathbf{a}}$ | $\mathbf{S D}$ | $\mathbf{Q u a l i t y}$ parameter | $\mathbf{N} \mathbf{~ p} \boldsymbol{K}_{\mathbf{a}}$ | $\mathbf{S D}$ | Quality parameter |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| $\mathbf{8 a}$ | $-0,86$ | 0,51 | $-0,40$ | 6,31 | 0,60 | 0,00 |
| $\mathbf{8 c}$ | $-0,86$ | 0,51 | $-0,40$ | 6,31 | 0,60 | 0,00 |
| $\mathbf{8 d}$ | $-1,00$ | 0,51 | $-0,40$ | 6,06 | 0,60 | 0,00 |
| $\mathbf{8 b}$ | $-0,92$ | 0,51 | $-0,40$ | 6,25 | 0,60 | 0,00 |
| $\mathbf{8 e}$ | $-0,51$ | 0,51 | $-0,40$ | 6,66 | 0,60 | 0,00 |
| $\mathbf{8 f}$ | $-1,00$ | 0,51 | $-0,40$ | 4,84 | 0,60 | 0,00 |
| $\mathbf{9 a}$ | $-0,82$ | 0,51 | $-0,40$ | 6,35 | 0,60 | 0,00 |
| $\mathbf{9 b}$ | $-0,56$ | 0,51 | $-0,40$ | 6,61 | 0,60 | 0,00 |
| $\mathbf{9 c}$ | 8,85 | 0,51 | 0,00 | 2,41 | 0,51 | 0,00 |
| $\mathbf{1 3 a}$ | $-0,72$ | 0,51 | $-0,40$ | 6,45 | 0,60 | 0,00 |
| $\mathbf{1 3 b}$ | $-0,69$ | 0,51 | $-0,40$ | 6,48 | 0,60 | 0,00 |
| $\mathbf{1 3 c}$ | $-1,00$ | 0,51 | $-0,40$ | 6,00 | 0,60 | 0,00 |
| $\mathbf{6 a}$ | 1,98 | 0,51 | $-0,04$ | NA | NA | NA |

Quality parameter (QP): The quality parameter can be used to check whether the compound contains features that MoKa cannot take into account when it predicts the specific $\mathrm{p} K_{\mathrm{a}}$.
$\mathrm{QP}=0$ indicates that all the chemical groups in the compound have appropriate parameters to calculate the specific $\mathrm{p} K_{\mathrm{a}}$.
$\mathrm{QP}<$ or $>$ indicates that the effect of one or more chemical groups on the specific $\mathrm{p} K_{\mathrm{a}}$ is unknown. The higher the absolute value of the QP, the higher is the expected effect of these groups on $\mathrm{p} K_{\mathrm{a}}$. The +/- sign before QP specifies whether these groups are expected to shift $\mathrm{p} K_{\mathrm{a}}$ upwards or downwards.

## QM calculations

## Absolute energies of all reactants

Table S1: Absolute energy values.

|  | B3LYP-D3 water <br> (kcal/mol) | B3LYP-D3 gas <br> (kcal/mol) |
| :--- | ---: | ---: |
| $\mathrm{H}_{2} \mathbf{O}$ | $-47962,92$ | $-47955,99$ |
| $\mathbf{N H}_{4}{ }^{+}$ | $-35780,86$ | $-35698,10$ |
| $\mathbf{N H}_{3}$ | $-35489,63$ | $-35484,54$ |
| PhCHO | $-216805,79$ | $-216799,60$ |
| 8a_neutral | $-457969,49$ | $-457963,11$ |
| 8a_N3 | $-458256,03$ | $-458205,26$ |
| 8a_N1 | $-458247,24$ | $-458200,30$ |
| 8a_N1_N3 | $-458515,90$ | $-458347,50$ |
| N-methyl-2-[(methylamino)methyl]aniline_neutral | $-289123,51$ | $-289116,71$ |
| N-Methyl-2-[(methylamino)methyl]aniline_N1 | $-289416,51$ | $-289365,03$ |
| N-Methyl-2-[(methylamino)methyl]aniline_N3 | $-289419,85$ | $-289366,17$ |
| N-Methyl-2-[(methylamino)methyl]aniline_N1_N3 | $-289693,03$ | $-289505,43$ |
| 13b_neutral | $-507287,54$ | $-507281,09$ |
| 13b_N3 | $-507574,42$ | $-507524,90$ |
| 13b_N1 | $-507564,37$ | $-507520,02$ |
| 13b_N1_N3 | $-507832,45$ | $-507672,01$ |
| N-Isopropyl-2-[[methylamino)methyl]aniline_neutral | $-338449,55$ | $-338444,20$ |
| N-Isopropyl-2-[(methylamino)methyl]aniline_N1 | $-338738,70$ | $-338690,85$ |
| N-Isopropyl-2-[(methylamino)methyl]aniline_N3 | $-338742,54$ | $-338690,41$ |
| N-Isopropyl-2-[(methylamino)methyl]aniline_N1_N3 | $-339015,83$ | $-338836,56$ |

## Determining the protonation pattern:

Energy differences between the different protonation patterns for compounds 8a and 13b in water revealed that protonation is most likely at the $N-3$ nitrogen.

U


U


$+2 \mathrm{NH}_{3}$
=

$$
23,2 \mathrm{kcal} / \mathrm{mol}
$$

## QM summary

The scheme below depicts a reaction pathway. The energy difference ( $\Delta U=U_{2}-U_{1}$ ) between different reaction states $(\mathbf{A}$ to $\mathbf{H})$ for compounds $\mathbf{8 a}$ and 13b is presented in Table 2. $\Delta U<0$ indicates that the respective reaction is spontaneous (exothermic) while $\Delta U>0$ indicates that it is not (endothermic).The energy of each state is calculated from the absolute energies of the molecules that comprise it, which are presented in Table S1.


Table S2: Energy differences between states along the reaction pathway.

|  | Compound 8a:R = R' = Me |  | Compound 13b: $\mathrm{R}=\mathrm{Me}, \mathrm{R}{ }^{\prime}=\mathrm{iPr}$ |  |
| :---: | :---: | :---: | :---: | :---: |
| $\Delta \mathrm{U}$ | B3LYP-D3 water (kcal/mol) | B3LYP-D3 gas (kcal/mol) | B3LYP-D3 water (kcal/mol) | B3LYP-D3 gas (kcal/mol) |
| B - A | 4,68 | -28,58 | 4,34 | -30,24 |
| C-A | 13,47 | -23,63 | 14,39 | -25,36 |
| D - B | 31,36 | 71,33 | 33,19 | 66,46 |
| D-C | 22,56 | 66,37 | 23,15 | 61,58 |
| E-A | 3,11 | 2,78 | -4,88 | -6,72 |
| F-E | -5,12 | -35,89 | -1,76 | -32,64 |
| G - E | -1,78 | -34,75 | 2,08 | -33,08 |
| H-F | 18,05 | 74,31 | 17,93 | 67,42 |
| H-G | 14,71 | 73,17 | 14,09 | 67,85 |
| H-D | -20,00 | -1,54 | -26,25 | -8,16 |
| F-B | -6,69 | -4,53 | -10,99 | -9,12 |
| G - C | -12,14 | -8,34 | -17,20 | -14,43 |

It is worth mentioning that in all calculations for both compounds, the N1 protonated openring diamine forms an intramolecular H bond, while the N 3 protonated open-ring diamine of both compounds formed an intramolecular H bond only in the DFT optimisation in gas phase.

## MM conformational search:


C


D


A: anti-axial conformation of minimal energy for neutral compounds (8a-c,e, 9a, 13a-c) B: anti-axial conformation of minimal energy forN3 protonated compounds ( $\mathbf{8 a} \mathbf{a}^{\prime} \mathbf{- e}, \mathbf{9} \mathbf{9}^{\prime}$, 13a'-c')

C: equatorial conformation of minimal energy for neutral compounds (8d,f, 9b)
D: equatorial conformation of minimal energy for N 3 protonated compounds ( $\mathbf{8 f}, \mathbf{9} \mathbf{9}$ )

| Compounds | Number of Conformers | RMSD ( ${ }_{\text {( }}{ }^{\text {a }}{ }^{a}$ | Motif of minimum energy conformation $^{b}$ |
| :---: | :---: | :---: | :---: |
| 8a | 2 | 0.0311 | i (A) |
| 8c | 4 | 0.0207 | i (A) |
| 8d | 2 | 0.0314 | ii (C) |
| 8b | 4 | 0.0206 | i (A) |
| 8 e | 6 | 0.0187 | i (A) |
| 8 f | 5 | 0.0256 | ii (C) |
| 9a | 74 | 0.0230 | i (A) |
| 9b | 24 | 0.0331 | ii (C) |
| 9c | 32 | 0.0208 | i (E) |
| 13a | 69 | 0.0198 | i (A) |
| 13b | 32 | 0.0283 | i (A) |
| 13c | 27 | 0.0333 | i (A) |
| 8a' | 4 | 0.0185 | $\mathrm{i}(\mathbf{B})$ |
| $8 \mathrm{c}^{\prime}$ | 9 | 0.0140 | i (B) |
| 8d' | 4 | 0.0188 | $\mathrm{i}(\mathbf{B})$ |
| 8b' | 9 | 0.0157 | $\mathrm{i}(\mathrm{B})$ |
| $8 \mathrm{e}^{\prime}$ | 9 | 0.0158 | i (B) |
| 8f' | 3 | 0.0188 | ii (D) |
| 9a' | 40 | 0.0115 | i (B) |
| 9b' | 11 | 0.0178 | ii (D) |
| 9c N1 prot | 22 | 0.0368 | (F) |
| 13a' | 35 | 0.0156 | i (B) |
| 13b' | 23 | 0.0216 | i (B) |
| 13c ${ }^{\text {' }}$ | 14 | 0.0298 | i (B) |
| 8a N1 prot | 6 | 0.0258 | (G) |
| 13b N1 prot | 20 | 0.0464 | (G) |
| 8a diprot (anti) | 3 | 0.0494 | (H) |
| 13b diprot (anti) | 13 | 0.0580 | (H) |

$\boldsymbol{a}$ : Mean value of the RMS for every conformer individually compared to the conformer of minimum energy after superposition.
$\boldsymbol{b}: \mathrm{i}=$ anti-axial motif as described in the manuscript, $\mathrm{ii}=$ equatorial motif as described in the manuscript, ()$=$ letter corresponding to relative figure as presented in the Supporting Information.

Both the anti-axial and the equatorial conformers are found in all compounds, usually as next optimal conformation after the one of minimal energy. The energy difference between the respective other conformation is $2-5 \mathrm{kcal} / \mathrm{mol}$.

In neutral compound $\mathbf{9 c}$ the structure of minimal energy follows the anti-axial conformational motif, although N3 is more planar due to the presence of the phenyl group. The N1 protonated form of compound $9 \mathbf{c}$ could not be superimposed with the rest of the compounds, as the protonation occurs in different atoms.

Conformers of minimal energy for $\mathbf{9 c}$ neutral ( $\mathbf{E}$ ) and N1 protonated ( $\mathbf{F}$ )



The systematic conformational search of the diprotonated compounds retained the protonated N stereoisomerism, therefore the syn- and anti- form of the $\mathrm{N}^{+}-\mathrm{H}$ were both submitted for a conformational search and their minimal energy conformers were compared. The anticonformer was of lower energy than the syn-, thus it was selected for the QM calculations.

Conformers of minimal energy for 8a and 13b N1 protonated (G) and anti-diprotonated (H)

H


It is noteworthy to highlight that both the reference compound 8a and the least stable compound 13b share the same conformational motif in all cases - neutral from, N3 protonation, N 1 protonation, and diprotonation.

## MM ring strain comparison in gas phase

The energy values presented are calculated by: $\Delta U=U$ ( $\mathbf{C p d}$ _mod or $\mathbf{C p d} \mathbf{- m o d}$ ' $)-U$ (8a or 8a' $)$.

## $\Delta U[\mathrm{kcal} / \mathrm{mol}]$

gas

|  | 8b_mod | 8f_mod | 9b_mod | 13b_mod | 13c_mod |
| :---: | :---: | :---: | :---: | :---: | :---: |
| neutral | 0.71 | 3.61 | 3.74 | 3.09 | 4.90 |
| protonated | 0.81 | 0.48 | 0.74 | 3.6 | 5.84 |


|  | $U(\mathrm{kcal} / \mathrm{mol})$ <br> gas | $U(\mathrm{kcal} / \mathrm{mol})$ <br> water |
| :---: | :---: | :---: |
| 8a | -4.46 | -11.09 |
| 8a' | -58.85 | -106.01 |
| 8b | 13.72 | 8.99 |
| 8b_mod | -3.75 | -10.35 |
| 8b' | -41.02 | -86.19 |
| 8b'_mod | -58.04 | -105.28 |
| 8 f | -16.45 | -23.88 |
| 8f_mod | -0.85 | -9.62 |
| 8f' | -84.20 | -127.68 |
| 8f'_mod | -58.37 | -104.40 |
| 9b | -1.65 | -8.52 |
| 9b_mod | -0.72 | -9.00 |


| 9b' | -54.64 | -97.10 |
| :---: | :---: | :---: |
| 9b'_mod | -58.11 | -103.55 |
| 13b | 1.28 | -4.04 |
| 13b_mod | -1.37 | -8.04 |
| 13b' | -52.27 | -97.41 |
| 13b'_mod | -55.21 | -102.51 |
| 13c | 38.41 | 33.76 |
| 13c_mod | 0.44 | -5.82 |
| 13c' | -13.02 | -60.89 |
| 13c'_mod | -53.01 | -100.79 |

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