Supporting Information

for

Correction: Mechanochemical enzymatic resolution of N-benzylated- β^3 -amino esters

Mario Pérez-Venegas¹, Gloria Reyes-Rangel¹, Adrián Neri², Jaime Escalante^{*,2} and Eusebio Juaristi.^{*,1,3}

Address: ¹Departamento de Química, Centro de Investigación y de Estudios Avanzados, Avenida I.P.N. 2508, Ciudad de México, 07360, Mexico, ²Centro de Investigaciones Químicas, Universidad Autónoma del Estado de Morelos, Av. Universidad 1001, Cuernavaca, Morelos, 62210, Mexico and ³El Colegio Nacional, Luis Gonzáles Obregón 23, Centro Histórico, Ciudad de México, 06020, Mexico

Email: <u>Jaime Escalante* - jaime@uaem.mx;</u> Eusebio Juaristi* - <u>juaristi@relaq.mx</u> *Corresponding author

Experimental section, NMR spectra, chromatograms and X-ray diffraction data

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1) General methods

NMR spectra were recorded on a BRUKER DP300 (300 MHz) and a Jeol Eclipse (400 MHz). High resolution mass spectra were recorded on a HPLC 1100 coupled to a MSD-TOF Agilent series HR-MSTOF model 1969 A. Chromatograms were acquired in a Dionex HPLC Ultimate 3000 with a UV/Visible detector, with diode array, at 210 and 254 nm. HSBM reactions were carry out in a Retsch, Mixer Mill (MM200). *N*-benzylated-β³-amino esters were synthetized according to the literature. All reagents for the synthesis were purchased from Sigma-Aldrich. Immobilized CALB was purchased from Novozymes, Novozym 453 (Immobilized on acrylic resine, U/q >100000)

2) General method for the synthesis of N-benzylated- β^3 -amino esters.

The substrates *rac-1a* to *rac-1f* (Table 2 in article) for the enzymatic resolution were synthetized according to previously reported methodologies [1-3], the synthesis for *rac-1a* starts from Methyl crotonate. For substrates *rac-1g* to *rac-1i* we used the methodology describe by Escalante [4].

Br OMe
$$Ph_3P$$
 OMe Ph_3P OM

Scheme 1. Route for the synthesis of substrates rac-1.

Synthesis of Methyl (triphenylphosphoranyldiene) acetate (**TPA**): Triphenylphosphine (7.86 g, 0.03 mmol) was placed in a round flask equipped with a magnetic bar and was dissolved in Toluene (100 mL), methyl bromoacetate (5.01 g, 0.03 mmol) was added and the reaction was stirred overnight. The reaction product was filtered and the resulting solid was dissolved in water (200 mL), basified till pH 9 using KOH (2M) and phenolphthalein as indicator and was extracted with CH₂Cl₂ (150 mL x 2), the organic fractions were collect, dried with anhydride

sodium sulfate and concentrate to obtain 9.3 g (93% yield) of a white solid (**TPA**). MS-TOF: calculated for $C_{21}H_{20}0_2P$ 335.1156, found 335.1197. ¹H NMR (CDCl₃, 300MHz): δ = 7.53 (m, 15H), 3.51 (s, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 28.7, 30.5, 50.0, 128.8, 128.9, 132.1, 133.0, 133.1. ³¹P{¹H} NMR (CDCl₃, 300MHz): δ 18.9, 23.1, 30.3.

Representative example for the synthesis of Methyl *trans*-2 pentenoate (**M***t***E-b**): In a round flask equipped with a magnetic bar **TPA** (10.02 g 0.03 mmol) was dissolved in anhydride THF (150 mL), propionaldehyde (1.32 g, 0.03 mmol) was added to the solution and the reaction mixture was refluxed for 4 hrs. The reaction was filtered and evaporated for his purification. The product was purified by flash chromatography on silica gel (Haxanes/Ethyl acetate) to give 5.2 g of a colorless liquid (**M**(*E*)**E-b**) (86% yield).

Methyl *trans*-2 pentenoate (*M t*E-b) [5], from propionaldehyde, colorless liquid, 86% yield. ¹H NMR (CDCl₃, 300MHz): δ 7.01 (dt, *J*=6.3 Hz, *J_{trans}*=15.6 Hz, 1H), 5.80 (d, *J*=15.6 Hz, 1H), 3.71 (s, 3H), 2.21 (quint, *J*=6.7 Hz, 2H), 1.05 (t, *J*=7.2 Hz, 3H), 5.28 (CH₂Cl₂). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 12.3, 25.5, 51.5, 120.1, 151.2, 167.4.

Methyl *trans*-2 hexenoate (**M tE-c**) [6], from butyraldehyde, colorless liquid, 70% yield.
 ¹H NMR (CDCl₃, 300MHz): δ 6.96 (dt, *J*=6.9 Hz, *J_{trans}*=15.6 Hz, 1H), 5.81 (d, *J*=15.6 Hz, 1H), 3.71 (s, 3H), 2.21 (q, *J*=6.8 Hz, 2H), 1.47 (m, *J*=7.5 Hz, 2H), 0.92 (t, *J*=7.3 Hz, 3H), 5.29 (CH₂Cl₂). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 13.7, 21.4, 34.6, 51.4, 121.1, 149.8, 167.6.

Methyl *trans*-2 heptenoate (**M tE-d**) [7], from valeraldehyde, colorless liquid, 83% yield.
 ¹H NMR (CDCl₃, 300MHz): δ 6.95 (dt, *J*=6.9 Hz, *J_{trans}*=15.6 Hz, 1H), 5.79 (d, *J*=15.6 Hz, 1H), 3.69 (s, 3H), 2.17 (q, *J*=6.9 Hz, 2H), 1.35 (m, 2H), 0.88 (t, *J*=7.2 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 13.7, 22.4, 30.2, 32.3, 51.6, 120.9, 149.7, 167.2.

Methyl *trans*-2 octenoate (**M tE-e**) [8], from hexanal, colorless liquid, 80% yield. ¹H NMR (CDCl₃, 300MHz): δ 6.82 (dt, *J*=7.2 Hz, *J_{trans}*=15.6 Hz, 1H), 5.80 (d, *J*=15.6 Hz, 1H), 3.71 (s, 3H), 2.18 (q, *J*=7.2 Hz, 2H), 1.44 (quint, *J*=16.5 Hz, 2H), 1.29 (m, 4H), 0.87 (t, *J*=6.3 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 41.1, 22.6, 27.8, 31.4, 32.2, 120.8, 150.0, 167.4.

Methyl *trans*-2 nonenoate (**M tE-f**) [9], from heptanal, colorless liquid, 75% yield. ¹H NMR (CDCl₃, 300MHz): δ 6.96 (dt, *J*=6.9 Hz, *J*_{trans}=15.6 Hz, 1H), 5.85 (d, *J*=15.9 Hz, 1H), 3.71 (s, 3H), 2.18 (q, *J*=7.2 Hz, 2H), 1.43 (quint, *J*=5.1 Hz, 2H), 1.27 (m, 4H), 0.87 (t, *J*=6.6 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 14.2, 22.7, 28.1, 29.0, 31.7, 32.4, 51.6, 120.9, 150.1, 167.5.

Representative example for the synthesis of Methyl 3-(benzylamino) butanoate (rac-1a): Bi(NO₃)₃·5H₂O (0.8 g) was added to a round flask containing Methyl crotonate (3.18 mL, 0.03 mmol) equipped with a stir bar, the reaction was stirred by 5 min at 0 °C, after that benzylamine (3.27 mL, 0.03 mmol) was added. The reaction was stirred overnight, filtered, extracted with CH₂Cl₂, dried with anhydride sodium sulfate and concentrated for his purification, obtaining 4 g (63% yield) of an amber liquid, rac-1a. Compounds rac-1g to rac-1i were synthetized as describe in literature [4].

rac-Methyl 3-(benzylamino) butanoate (rac-1a) [10], from methyl crotonate, amber liquid, 63% yield. MS-TOF: calculated for C₁₂H₁₈NO₂ 208.1293, found 208.1331. ¹H NMR (CDCl₃, 300MHz): δ 7.32 (d, *J*=4.2 Hz, 4H), 7.26 (m, 1H), 3.79 (q, *J*=12.3 Hz, 2H), 3.67 (s, 3H), 3.16 (m, *J*=6.3 Hz, 1H), 2.51 (dd, *J*= 6.9, 15.3, 1H), 2.39 (dd, *J*= 6, 15, 1H), 1.83 (br s, 1H), 1.43 (d, *J*=6.3 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 20.58, 41.5, 49.8, 51.3, 51.7, 127.1, 128.3, 128.6, 140.4, 173.0.

rac-Methyl 3-(benzylamino) pentanoate (rac-1b) [11], from MtE-b, amber liquid, 67% yield. MS-TOF: calculated for C₁₃H₂₀NO₂ 222.1449, found 222.1489. ¹H NMR (CDCl₃, 300MHz): δ 7.32 (m, 5H), 3.78 (s, 2H), 3.67 (s, 3H), 2.97 (quint, J= 6 Hz, 1H), 2.46 (m, 2H), 1.53 (m, 2H) + (br, 1H), 0.92 (t, J= 7.2 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 10.0, 26.9, 38.8, 51.0, 51.6, 55.6, 127.0, 128.3, 128.5, 140.7, 173.2.

rac-Methyl 3-(benzylamino) hexanoate (rac-1c) [12], from MtE-c, amber liquid, 57% yield. MS-TOF: calculated for C₁₄H₂₂NO₂ 236.1606, found 236.1647. ¹H NMR (CDCl₃, 300MHz): δ 7.27 (m, 5H), 3.78 (s, 2H), 3.67 (s, 3H), 3.03 (quint, J= 6.3 Hz, 1H), 2.46 (d, J= 6.3 Hz, 2H), 1.43 (m, 4H) + (br, 1H), 0.91 (t, J= 6.9 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 14.3, 19.1, 36.7, 39.2, 51.1, 51.6, 54.1, 127.0, 128.2, 128.5, 140.7, 173.2.

rac-Methyl 3-(benzylamino) heptanoate (rac-1d), from MtE-d, amber liquid, 79% yield. MS-TOF: calculated for C₁₅H₂₄NO₂ 250.1762, found 250.1803. ¹H NMR (CDCl₃, 300MHz): δ 7.28 (m, 5H), 3.78 (s, 2H), 3.67 (s, 3H), 3.01 (quint, J= 6.3 Hz, 1H), 2.46 (d, J= 6.0 Hz, 2H), 1.42 (m, 6H) + (br, 1H), 0.89 (t, J= 6.6 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 14.2, 22.9, 28.0, 34.2, 39.2, 51.1, 51.7, 54.4, 127.0, 128.3, 128.5, 140.7, 173.2.

• rac-Methyl 3-(benzylamino) octanoate (rac-1e), from MtE-e, amber liquid, 60% yield. MS-TOF: calculated for $C_{16}H_{24}NO_2$ 264.1919, found 264.1956. ¹H NMR (CDCl₃, 300MHz): δ 7.27 (m, 5H), 3.78 (s, 2H), 3.67 (s, 3H), 3.02 (quint, J= 6 Hz, 1H), 2.46 (d, J= 6 Hz, 2H), 1.46 (m, 8H) + (br, 1H), 0.89 (t, J= 6 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 14.1, 22.7, 25.5, 32.0, 34.4, 39.2, 51.1, 51.6, 54.3, 127.0, 128.2, 128.5, 140.7, 173.2.

• rac-Methyl 3-(benzylamino) nonanoate (rac-1f), from MtE-f, amber liquid, 50% yield. MS-TOF: calculated for C₁₇H₂₈NO₂ 278.2075, found 278.1156. ¹H NMR (CDCl₃, 300MHz): δ 7.26 (m, 5H), 3.78 (s, 2H), 3.67 (s, 3H), 3.01 (quint, J= 6 Hz, 1H), 2.46 (d, J= 6.3 Hz, 2H), 1.46 (m, 10H) + (br, 1H), 0.88 (t, J= 5.7 Hz, 3H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 14.2, 22.7, 25.8, 29.5, 31.9, 34.5, 39.2, 51.1, 51.6, 54.4, 127.0, 128.3, 128.5, 140.7, 173.2.

• rac-Methyl 3-(benzylamino)-3-phenylpropanoate (rac-1 \mathbf{g}) [4], amber liquid. MS-TOF: calculated for $C_{17}H_{20}NO_2$ 270.1494, found 270.1501. 79% yield. ¹H NMR (CDCl₃, 400MHz): δ 7.39-7.25 (m, 10H), 4.15 (dd, J= 5.2, 8.6 Hz, 1H), 3.71-3.55 (m, 5H), 2.77 (dd, J= 8.8, 15.6 Hz, 1H), 2.55 (dd, J= 5.2, 15.6 Hz, 1H). ¹³C{¹H} NMR (CDCl₃, 400MHz): δ 42.8, 51.3, 58.8, 65.3, 126.9, 127.1, 127.5, 128.1, 128.3, 128.6, 140.2, 142.4, 172.2.

rac-Methyl 3-(benzylamino)-3-(p-metoxy)phenylpropanoate (rac-1h) [13], amber liquid, 50% yield (0.03 mmol). ¹H NMR (CDCl₃, 300MHz): δ 7.33-7.23 (m, 7H), 6.85 (d, J=9 Hz, 2H), 4.06 (q, J= 5.4 Hz, 1H), 3.81 (s, 3H), 3.67-3.50 (m, 5H), 2.76-2.57 (m, 2H), 1.97 (br s, 1H). ¹³C{¹H} NMR (CDCl₃, 300MHz): δ 43.1, 51.4, 51.8, 55.4, 58.2, 114.1, 127.0, 128.3, 128.5, 134.6, 140.4, 159.0, 172.6.

• rac-Methyl 3-(benzylamino)-4,4-dimethylpentanoate (rac-1i) [4], amber liquid. HRMS(ES⁺): calculated for C₁₅H₂₄NO₂ 250.3616, found 250.1812. ¹H NMR (CDCl₃, 400MHz): δ 7.32-7.23 (m, 5H), 3.90-3.68 (m, 5H), 2.80 (q, J= 4.2 Hz, 1H), 2.57 (dd, J=4.2, 14.7 Hz, 1H), 2.29 (dd, J= 8.4, 14.7 Hz, 1H), 1.29 (br s, 1H), 0.92 (s, 9H). I¹³C{¹H} NMR (CDCl₃, 100MHz): δ 26.7, 35.6, 37.0, 51.8, 54.3, 64.1, 127.0, 128.4, 141.2, 174.3.

3) General method for the synthesis of N-benzylated- β^3 -amino acids

Representative example for the synthesis of *N-benzylated*- β^3 -amino acids⁵: The appropriate *N*-benzylated- β^3 -amino acid methyl ester is placed in a round flask equipped with a stir bar and 10 mL of water, 1 equivalent of NaOH is added when required (substrates **2c** to **2j**), the reaction was refluxed for 4 hrs. The solution was neutralized, the solvent was evaporated under vacuum and the product was washed with CH₂Cl₂, filtered and decanted with MeOH to obtain a withe off solid.

rac-3-(N-benzylamino) butanoic acid (rac-2a) [4], white off solid, 86% yield. mp 170-171 °C (170-171°C [4]). MS-TOF: calculated for C₁₁H₁₅NO₂ 194.1136, found 194.1175. ¹H NMR (D₂O, 400MHz): δ 7.42 (s, 5H), 4.18 (q, *J*= 16.8 Hz, 2H), 3.52 (m, *J*= 6.4 Hz, 1H), 3.28 (br s, 1H), 2.59 (dd, *J*=4.8, 13 Hz, 1H), 2.54 (dd, *J*=4.5, 12.8 Hz, 1H), 1.32 (d, *J*= 6.4 Hz, 3H). ¹³C{¹H} NMR (D₂O, 400MHz): δ 15.9, 38.2, 48.1, 51.2, 129.3, 129.6, 130.9, 176.4.

rac-3-(N-benzylamino) pentanoic acid (rac-2b) [14], 89% yield. white solid. mp 167-169°C (169-170°C [14]). MS-TOF: calculated for C₁₁H₁₅NO₂ 208.1293, found 208.1331.
 ¹H NMR (D₂O, 400MHz): δ 7.40 (s, 5H), 4.17 (q, *J*= 9, 1 Hz, 2H), 3.31 (m, 1H), 2.56 (dd, *J*=3.6, 12.9 Hz, 1H), 2.40 (dd, *J*=5.4, 12.7 Hz, 1H), 1.78 (m, 1H), 1.59 (m, 1H),

0.88 (t, J= 7.6Hz, 3H). ¹³C{¹H} NMR (D₂O, 400MHz): δ 8.8, 23.3, 35.3, 47.9, 56.7, 129.3, 129.5, 131.2, 178.2.

• rac-3-(N-benzylamino) hexanoic acid (rac-2c) [14], 97% yield. white solid. mp 156-159°C (160-162°C [14]). MS-TOF: calculated for C₁₃H₂₀NO₂ 222.1449, found 222.1488. ¹H NMR (D₂O, 400MHz): δ 7.40 (s, 5H), 4.18 (q, J= 10.2 Hz, 2H), 3.55 (m, 1H), 2.56 (dd, J= 3.6, 12.6 Hz, 1H), 2.40 (dd, J= 5.7, 12.6 Hz, 1H), 1.70 (m, 1H), 1.55 (m, 1H), 1.30 (m, 2H), 0.83 (t, J= 5.7Hz, 3H). 13 C{ 1 H} NMR (D₂O, 400MHz): δ 12.9, 18.0, 32.3, 35.9, 47.8, 55.1, 129.3, 129.4, 129.5, 131.4, 178.3.

• rac-3-(N-benzylamino) heptanoic acid (rac-2d) [15], 84% yield. white solid. mp 152-155°C. MS-TOF: calculated for C₁₄H₂₂NO₂ 236.1606, found 236.1644. ¹H NMR (D₂O, 400MHz): δ 7.26 (s, 5H), 4.0 (t, J= 10.2 Hz, 2H), 3.32 (m, 1H), 2.68 (dd, J= 4.8, 17.8 Hz, 1H), 2.28 (dd, J= 6.8, 18 Hz, 1H), 1.60 (m, 1H), 1.47 (m, 1H), 1.08 (m, 4H), 0.63 (t, J= 6.4 Hz, 3H). ¹³C{¹H} NMR (D₂O, 400MHz): δ 12.9, 21.4, 26.4, 29.5, 34.0, 48.5, 54.1, 129.1, 129.5, 129.7, 130.3, 137.9.

rac-3-(N-benzylamino) octanoic acid (rac-2e), 80% yield. white solid. mp 150-152°C.
 MS-TOF: calculated for C₁₅H₂₄NO₂ 250.1762, found 250.1803. ¹H NMR (D₂O, 400MHz):
 δ 7.25 (s, 5H), 4.03 (s, 2H), 3.32 (m, 1H), 2.67 (dd, J= 4.8, 18 Hz, 1H), 2.28 (dd, J= 6.8,

17.6 Hz, 1H), 1.58 (m, 1H), 1.46 (m, 1H), 1.03 (m, 6H), 0.60 (t, J= 6.4 Hz, 3H). ¹³C{¹H} NMR (D₂O, 400MHz): δ 13.0, 21.4, 23.9, 29.7, 30.3, 34.0, 48.5, 54.1, 129.1, 129.5, 129.7, 130.2, 137.9.

• rac-3-(N-benzylamino) nonanoic acid (rac-**2f**), 83% yield. white solid. mp 151-153°C MS-TOF: calculated for $C_{16}H_{26}N0_2$ 264.1919, found 264.2005. ¹H NMR (D_2O , 400MHz): δ 7.36 (s, 5H), 4.18 (s, 2H), 3.41 (m, 1H), 2.76 (dd, J= 4.8, 17.8 Hz, 1H), 2.67 (dd, J= 6.8, 17.6 Hz, 1H), 1.69 (m, 1H), 1.56 (m, 1H), 1.16 (m, 8H), 0.72 (t, J= 6 Hz, 3H). $^{13}C\{^{1}H\}$ NMR (D_2O , 400MHz): δ 13.2, 21.7, 24.2, 27.8, 29.7, 30.6, 34.1, 48.6, 54.0, 129.2, 129.7, 129.8, 130.3, 174.0.

rac-3-(N-benzylamino)-3-phenylpropanoic acid (rac-2g) [4], 80% yield. white solid. mp 149-151°C (148-151°C [4]). H NMR (CD₃OD, drops NH₄OH, 200MHz): δ 7.42-7.22 (m, 10H), 4.95 (dd, *J*= 4.4, 10.4, 1H), 3.33 (s, 2H), 2.63 (dd, *J*= 10.4, 16.8, 1H), 2.48 (dd, *J*= 4.4, 17.2, 1H), H NMR (CD₃OD+NH₄OH, 50MHz): δ 45.3, 50.9, 60.3, 126.6, 126.9, 127.3, 128.0, 128.1, 139.3, 142.6, 178.3.

rac-3-(N-benzylamino)-3-(p-methoxy)phenylpropanoic acid (rac-2h), 54% yield. white solid. mp 172-174°C. MS-TOF: calculated for C₁₆H₂₅NO₂ 263.1885, found 263.2005. ¹H NMR (CD₃OD, 400MHz): δ 7.26 (d, J= 9.6 Hz, 7H), 7.04 (d, J= 8 Hz, 2H), 4.45 (dd, J=

3.2, 10.2 Hz 1H), 2.76 (dd, J= 4.8, 17.8 Hz, 1H), 4.00 (dd, J= 13.2, 25.2 Hz, 2H), 3.85 (s, 3H), 2.86 (dd, J= 10.8, 16.6 Hz, 1H), 2.65 (dd, J= 3.6, 16.8 Hz, 2H). ¹³C{¹H} NMR (CD₃OD, 400MHz): δ 38.5, 48.1, 54.4, 59.3, 14.4, 126.4, 128.9, 128.9, 129.0, 129.1, 131.8, 160.7, 176.0.

rac-3-(N-benzylamino)-4,4-dimethylpentanoic acid (rac-2i) [4], 88% yield. white solid. mp 152-153°C (150-153°C [4]). HRMS (ES⁺): calculated for C₁₄H₂₂NO₂ 236.1651, found 236.1638. ¹H NMR (D₂O) δ 7.27 (s, 5H), 4.28 (m, 1H), 4.04 (m, 1H), 3.14 (m, 1H), 2.69 (m, 1H), 2.53 (m, 1H), 0.75 (m, 9H) ¹³C{¹H} NMR (CD₃OD, 100MHz): δ 24.6, 32.0, 33.3, 50.6, 62.6, 129.1, 129.8, 130.2, 174.3.

4) General method for the enzymatic resolution using CALB.

$$\begin{array}{c}
\text{Bn} \\
\text{NH} & \text{O} \\
\text{OMe} &
\begin{array}{c}
\text{H}_2\text{O} \\
\text{CALB} \\
\text{LAG, HSBM}
\end{array}$$

$$\begin{array}{c}
\text{Bn} \\
\text{NH} & \text{O} \\
\text{CR)-2a}
\end{array}$$

$$\begin{array}{c}
\text{Bn} \\
\text{NH} & \text{O} \\
\text{OMe}
\end{array}$$

$$\begin{array}{c}
\text{OMe} \\
\text{(S)-1a}
\end{array}$$

Representative example for the enzymatic resolution using CALB and rac-1a: 82 mg of substrate rac-1a (0.4 mmol), 3.6 μ L of water (0.2 mmol) and 0.2 mL of 2M2B were placed in an Agate jar (12 mm of Diameter) with an Agate ball (6 mm of Diameter, 480mg of weight), 40 mg of enzyme (CALB) were aggregated and the reactor was closed and placed in a Mixer Mill MM200 programed to carry out the reaction at 25 Hz during 30 min. Once the reaction is finished the content was extracted with methanol and transferred to a Falcon tube (50 mL), the solution was centrifuged at 3500 rpm for 5 min twice; the supernatant was collected and concentrated for his purification with silica gel, a mixture of Hexanes and ethyl acetate was used as mobile phase for recovering the raw material and Methanol and CH_2CI_2 for the purification of the product. Methyl 3-(benzylamino) butanoic acid (R)-2a was obtain as a white solid (49% yield). In order to avoid contamination by the wear of the stainless-steel reactors we decided to use Agate reactors.

Table S1. Search of the best parameters for the enzymatic hydrolysis resolution under HSBM.

Bn NH O
$$H_2O$$
 OMe O

	a LAG	frequency (Hz)	time	(S		(<i>R</i>)- 2 a	c^{f}				
entry ^a			(h)	recovered ^b (%)	ee ^c	$[\alpha]_D^{25d}$	yield ^b (%)	ee ^c (%)	$[\alpha]_D^{25e}$	(%)	E ^g
				,	, ,		, ,	,			
1	2M2B	25	0.5	51	99	20.4	49	80	-31.7	55	46
2	2M2B	15	0.5	70	89	22.0	30	77	-33.1	54	23
3	2M2B	15	1	51	99	22.5	49	95	-30.0	51	>200
4	Hexane	15	1	40	97	23.3	60	86	-20.4	53	55
5	AcOEt	15	1	86	69	10.3	13	95	-30.1	42	81
6	DIPE	15	1	76	92	17.3	<mark>19</mark>	88	-25.6	51	51
7	Toluene	15	1	75	72	12.6	<mark>25</mark>	93	-29.5	44	60
8	Dioxane	15	1	<mark>53</mark>	79	16.7	47	95	-30.3	45	94
9	IPA	15	1	80	48	7.4	20	95	-29.8	34	63
10	CH₃CN	15	1	65	65	9.8	29	95	-30.0	41	77
11	-	15	1	58	95	16.6	41	92	-28.2	51	89
12	-	25	1	58	93	16.5	42	86	-31.2	52	45
13 ^h	-	15	1	68	74	13.5	31	80	-27.0	48	20
14 ¹	2M2B	15	1	-	-	-	92	rac	-	-	-
15 ^J	2M2B	15	1	89	rac	-	-	-	-	-	-

^aReactions were carried out using the general method. ^bDeterminated after purification by flash chromatography. ^cDetermined by HPLC with chiral stationary phase. ^dUsing CH₃Cl and c=0.33, ^eUsing MeOH and c=0.33, ^fCalculated from $c = ee_s/(ee_s + ee_p)$. ^gE= ln[1-c(1+ ee_p)]/ln[1-c(1- ee_p)]. ^h0.25 equivalents of water were used. ⁱ1 equivalent of water were used. ^j water-free conditions.

Table S2. Substrate scope for the enzymatic resolution of N-benzylated- β^3 -amino esters.

		R	Subs		2		c^d				
entry ^a	rac-1		recovered ^D	ee ^c	ee ^c [α] _D ²⁵ (%)	yield⁵	ee ^c	$[\alpha]_D^{25}$	(%)	<i>E</i> ^e	a.c. ^f
			(%)	(%)		(%)	(%)				
1	1b	CH ₃ -(CH ₂)-	51	91	4.5	49	97	-36.5	48	>200	R
2	1c	CH ₃ -(CH ₂) ₂ -	53	84	2.1	43	98	-45.2	46	>200	R
3	1d	CH ₃ -(CH ₂) ₃ -	68	23	2.0	29	94	-35.3	20	40	R
4 ^g	1d	CH ₃ -(CH ₂) ₃ -	66	57	0.2	25	94	-33.3	38	58	R
5 ⁿ	1d	CH ₃ -(CH ₂) ₃ -	51	30	2.7	41	85	-40.5	26	17	R
6	1e	CH ₃ -(CH ₂) ₄ -	74	16	1.8	24	94	-40.0	15	38	R
7 ⁹	1e	CH ₃ -(CH ₂) ₄ -	76	27	0.8	22	82	-30.2	25	13	R
8 ^h	1e	CH ₃ -(CH ₂) ₄ -	82	27	2.7	10	76	-13.0	26	10	R
9	1f	CH ₃ -(CH ₂) ₅ -	79	13	0.8	18	91	-39.7	13	24	R
10 ^g	1f	CH ₃ -(CH ₂) ₅ -	85	21	1.1	11	82	-28.5	20	12	R
11 ^h	1f	CH ₃ -(CH ₂) ₅ -	84	18	0.7	8	83	-28.9	18	13	R
12 ¹	1g	Ph	90	18	3.4	10	83	-35.0	18	13	S
13 ^{i, h}	1g	Ph	85	18	3.0	15	68	-44.0	21	6	S
14 ^J	1g	Ph	90	16	3.2	9	37	-18.1	30	3	S
15 ^{i, k}	1g	Ph	80	5	2.5	20	48	-21.0	9	3	S
16	1h	<i>p</i> -MeO-Ph	89	1	-0.5	10	80	-31.7	1	9	S
17	1i	<i>t</i> -Bu	89	4	-0.6	4	94	12.8	4	34	S
18 ^h	1i	<i>t</i> -Bu	87	5	-0.7	6	89	16.6	5	18	S

^aReactions were carried out using the general method. ^bDetermined after purification by flash chromatography. ^cDetermined by HPLC with chiral stationary phase. ^dCalculated from $c = ee_s/(ee_s + ee_p)$. ^e $E = ln[1-c(1+ee_p)]/ln[1-c(1-ee_p)]$. **a.c**. Absolute Configuration of product **2**. ^f Assigned by chemical comparison and HPLC with chiral stationary phase. ^g Carried out with 15 Hz during 1 h. ^h Carried out with 25 Hz during 1 h. ^l 0.75 equiv. of water were used. ^l 1 equiv. of water were used. ^k 2 equiv. of enzyme were used.

Table S3. Recycling capacity of immobilized CALB under HSBM conditions.

Bn NH O
$$H_2O$$
 OMe $CALB$ OH H_2O OH H_2O OH H_2O OH H_2O OMe H_2O OH H_2O

	rCALB (cycle)	(<i>S</i>)-1a				(<i>R</i>)-2a	c^{d}		
entry ^a		recovered ^b	eec	$[\alpha]_D^{25}$	yield ^b	ee ^c	$[\alpha]_D^{25}$	(%)	E ^e
		(%)	(%)	ι1 <i>D</i>	(%)	(%)	£1 <i>D</i>		
1	-	51	>99	22.0	49	95	-31.7	51	>200
2	1	65	35	2.8	<mark>35</mark>	88	-33.3	<mark>28</mark>	22
3	2	80	6	0.5	20	80	-53.9	<mark>7</mark>	10
4	3	<mark>100</mark>	0	0.0	0	0	-	-	-
5 [†]	1	<mark>60</mark>	67	15.6	38	94	-29.2	42	65
6 ⁹	1	<mark>62</mark>	83	15.9	38	91	-26.8	48	55

^aReactions were carried out using the general method. ^bDetermined after purification by flash chromatography. ^cDetermined by HPLC with chiral stationary phase. ^dCalculated from $c = ee_s/(ee_s + ee_p)$. ^e $E = ln[1-c(1+ee_p)]/ln[1-c(1-ee_p)]$. ^fEnzyme with a pretreatment (Milling for 1h at 15 Hz). ^gPretreatment was carried out using 0.2mL of LAG.

Table S4. Scaling-up for the enzymatic hydrolysis using the substrate *rac*-1a.

	substrate equivalents	(<i>S</i>)-1a				(<i>R</i>)-2	c^d		
entry ^a		recovered ^b	ee ^c	$[\alpha]_D^{25}$	yield ^b	ee ^c	$[\alpha]_D^{25}$	(%)	E ^e
		(%)	(%)	[4][(%)	(%)	[#][(70)	
1	1	51	>99	22.0	49	95	-31.7	51	>200
2	3	<mark>52</mark>	62	8.0	<mark>48</mark>	93	-33.0	40	52
3	6	61	53	7.8	<mark>38</mark>	93	-33.9	36	47
4	9	59	49	7.0	40	94	-33.0	34	53

^aReactions were carried out using the general method but in absence of LAG. ^bDetermined after purification by flash chromatography. ^cDetermined by HPLC with chiral stationary phase. ^dCalculated from $c = ee_s/(ee_s + ee_p)$. ^e $E = ln[1-c(1+ee_p)]/ln[1-c(1-ee_p)]$.

5) ¹H and ¹³C NMR spectra for TPA, M*t*E-b to M*t*E-d, *rac*-1a to *rac*-1j and products 2a to 2j.

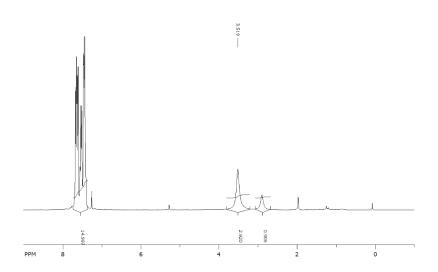


Figure S1. ¹H spectrum (300 MHz, CDCl₃) of TPAc

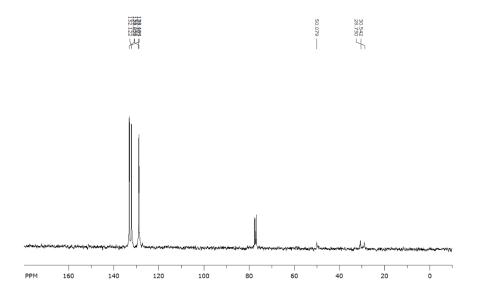


Figure S2. ¹³C spectrum (75 MHz, CDCl₃) of TPA

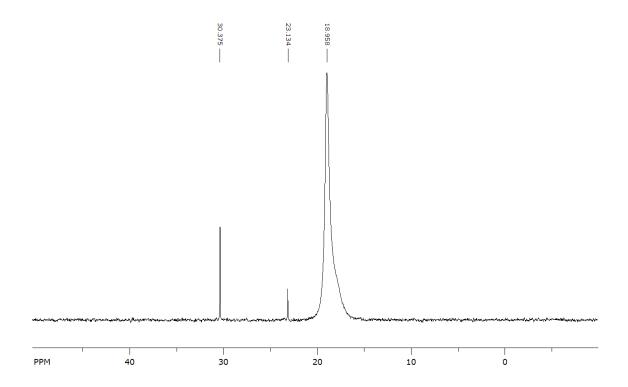


Figure S3. ³¹P spectrum (120 MHz, CDCl₃) of TPA

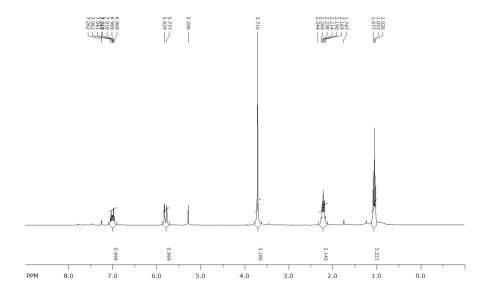


Figure S4. ¹H spectrum (300 MHz, CDCl₃) of MtE-b

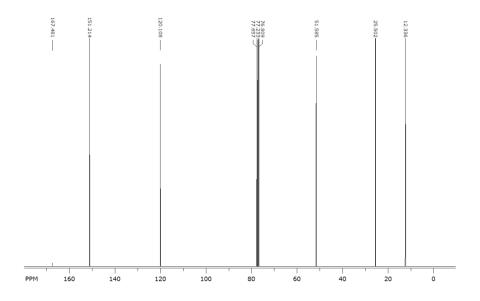
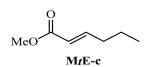


Figure S5. ¹³C spectrum (75 MHz, CDCl₃) of M*t*E-b



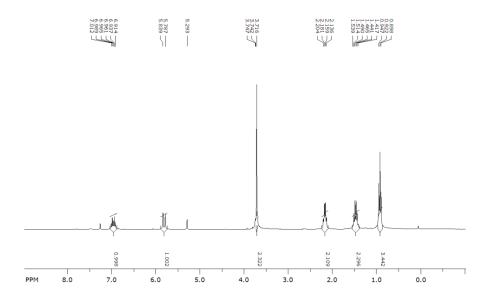


Figure S6. ¹H spectrum (300 MHz, CDCl₃) of MtE-c

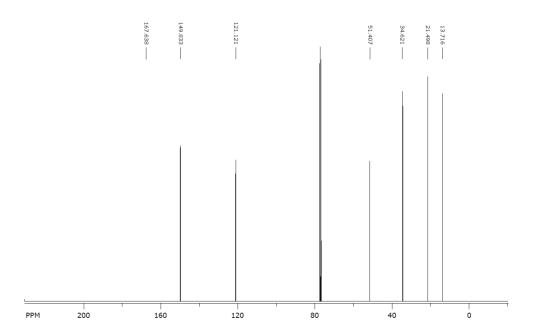
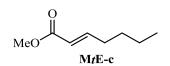


Figure S7. ¹³C spectrum (75 MHz, CDCl₃) of M*t*E-c



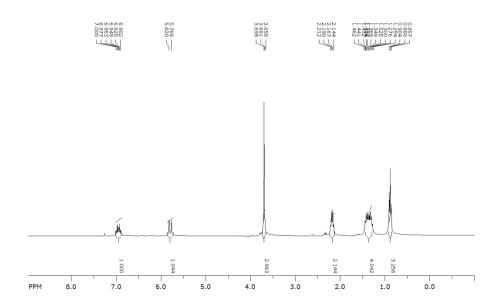


Figure S8. ¹H spectrum (300 MHz, CDCl₃) of MtE-d



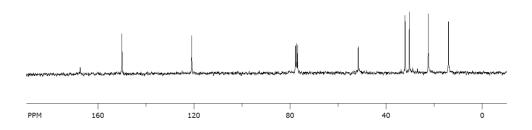
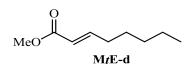


Figure S9. ¹³C spectrum (75 MHz, CDCl₃) of M*t*E-d



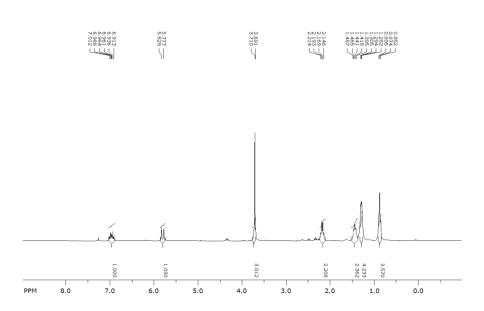


Figure S10. ¹H spectrum (300 MHz, CDCI₃) of MtE-e

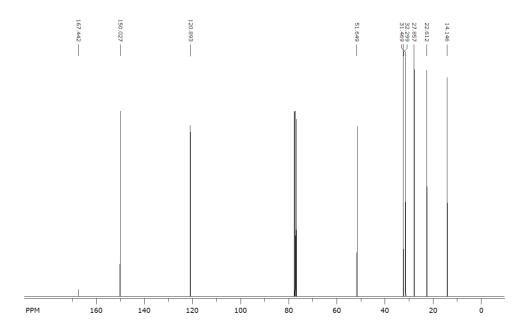
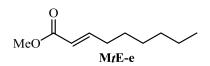


Figure S11. ¹³C spectrum (75 MHz, CDCl₃) of MtE-e



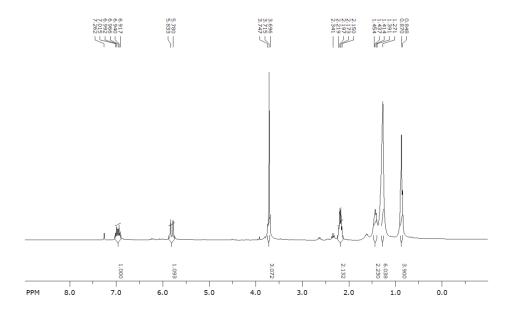


Figure S12. ¹H spectrum (300 MHz, CDCl₃) of MtE-f

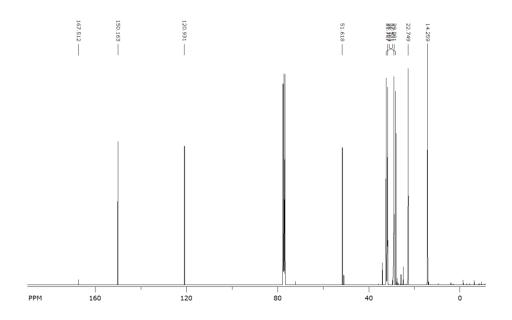
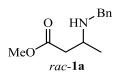


Figure S13. ¹³C spectrum (75 MHz, CDCl₃) of MtE-f



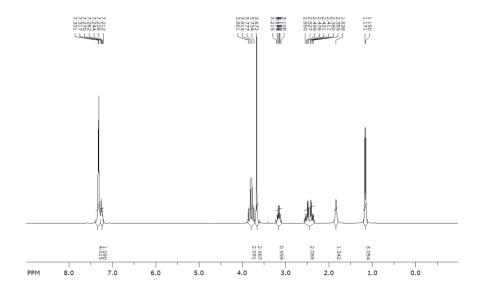


Figure S14. ¹H spectrum (300 MHz, CDCI₃) of rac-1a

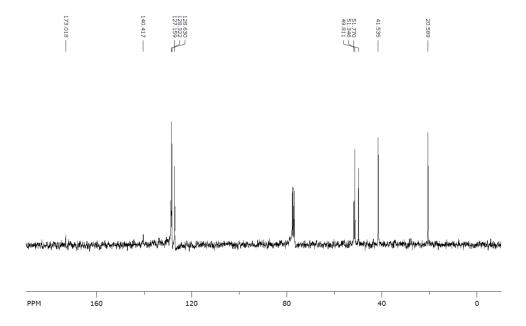


Figure S15. 13 C spectrum (75 MHz, CDCl₃) of *rac-1a*

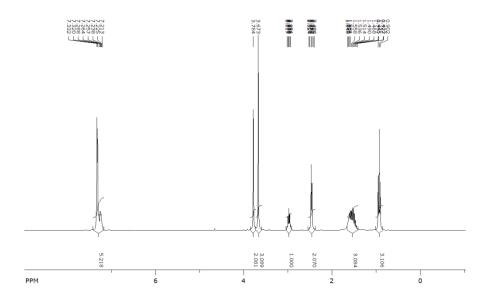


Figure S16. ¹H spectrum (300 MHz, CDCI₃) of *rac-1b*

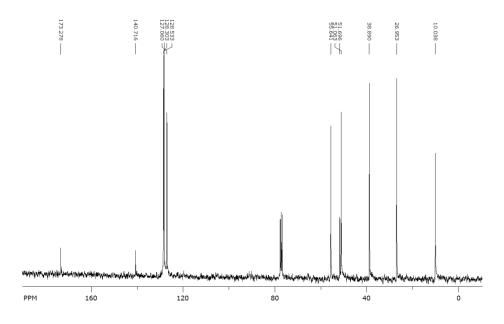


Figure S17. ¹³C spectrum (75 MHz, CDCl₃) of *rac-1b*

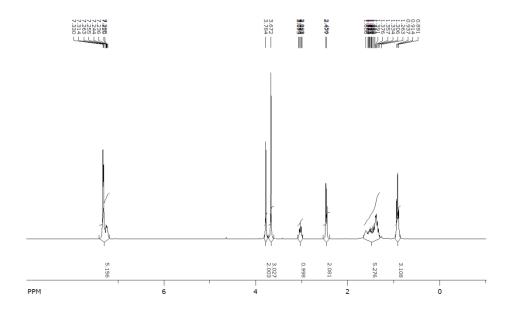


Figure S18. ¹H spectrum (300 MHz, CDCI₃) of *rac-1c*

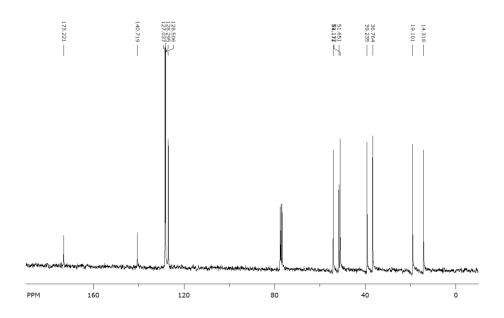


Figure S19. 13 C spectrum (75 MHz, CDCl₃) of *rac-1c*

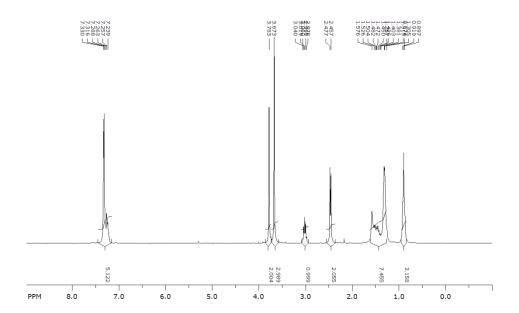


Figure S20. ¹H spectrum (300 MHz, CDCI₃) of *rac-1d*

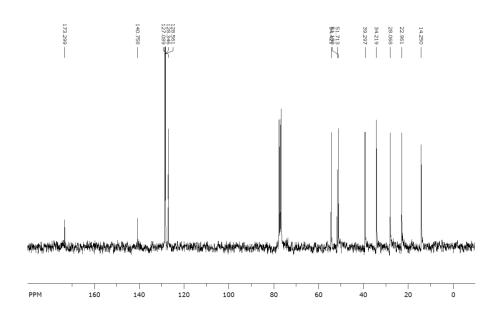


Figure S21. ¹³C spectrum (75 MHz, CDCl₃) of *rac-1d*

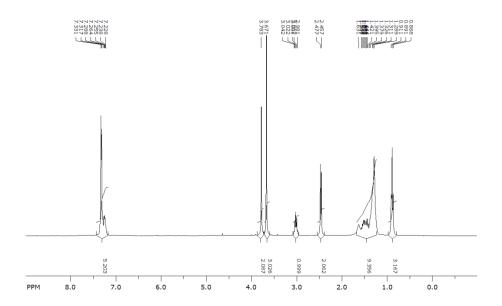


Figure S22. ¹H spectrum (300 MHz, CDCI₃) of rac-1e

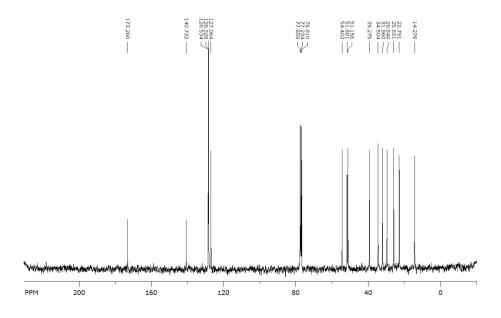


Figure S23. 13 C spectrum (75 MHz, CDCl₃) of *rac-1e*

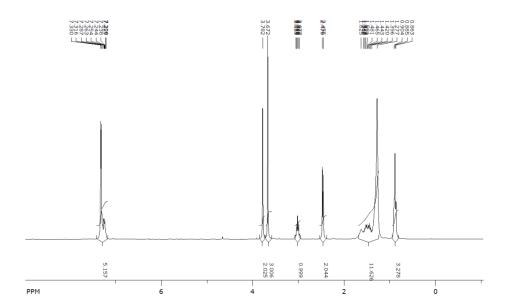


Figure S24. ¹H spectrum (300 MHz, CDCl₃) of rac-1f

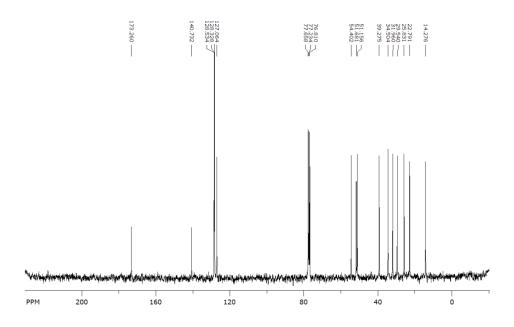


Figure S25. ¹³C spectrum (75 MHz, CDCl₃) of *rac-1f*

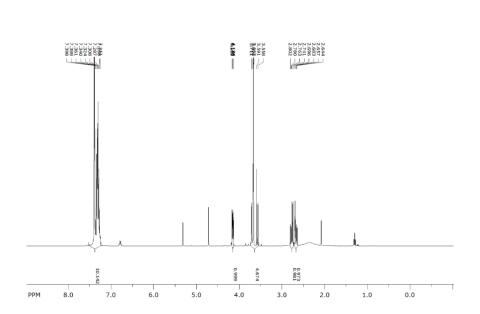


Figure S26. ¹H spectrum (300 MHz, CDCl₃) of *rac-1g*.

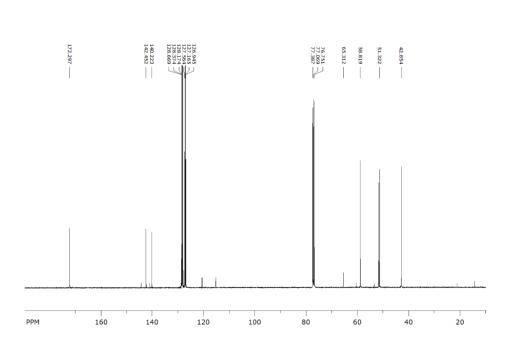


Figure S27. ¹³C spectrum (75 MHz, CDCl₃) of *rac-1g*.

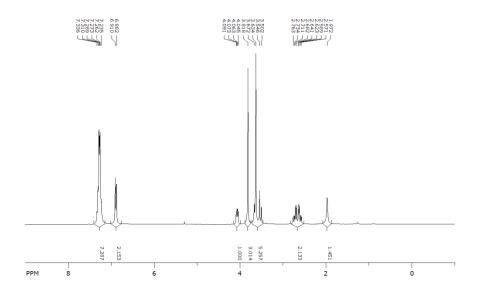


Figure S28. ¹H spectrum (300 MHz, CDCl₃) of *rac-1h*.

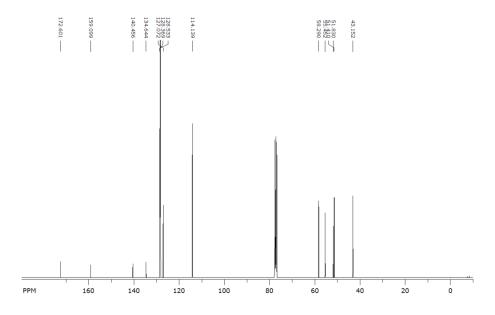


Figure S29. ¹³C spectrum (75 MHz, CDCl₃) of *rac-1h*.

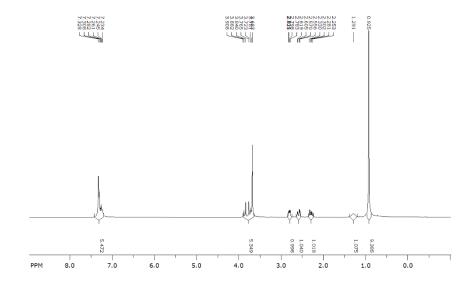


Figure S30. ¹H spectrum (300 MHz, CDCl₃) of *rac-1h*.

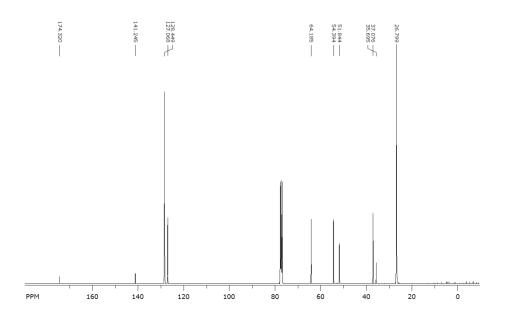
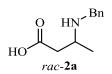


Figure S31. ¹³C spectrum (75 MHz, CDCl₃) of *rac-1h*.



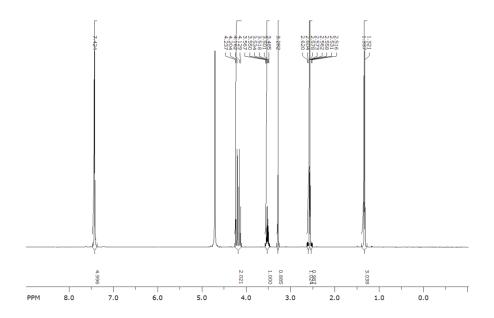


Figure S32. ¹H spectrum (400 MHz, D₂O) of 2a

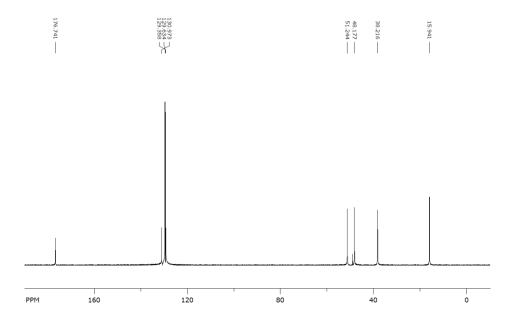


Figure S33. 13 C spectrum (100 MHz, D_2 O) of 2a

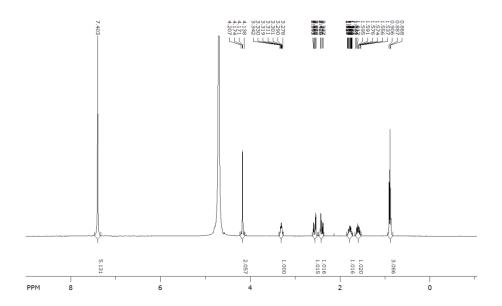


Figure S34. ¹³C spectrum (400 MHz, D₂O) of 2b

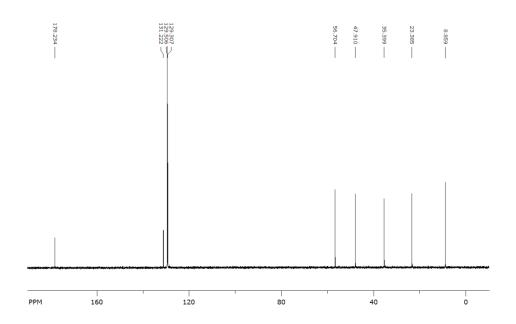


Figure S35. ¹³C spectrum (100 MHz, D₂O) of 2b

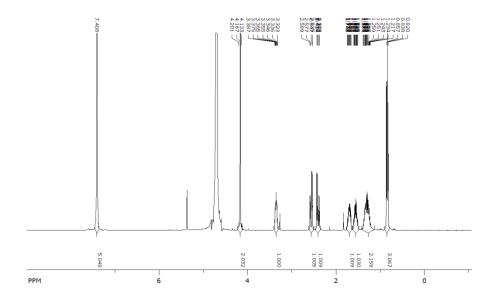


Figure S36. 1 H spectrum (400 MHz, D_2O) of 2c

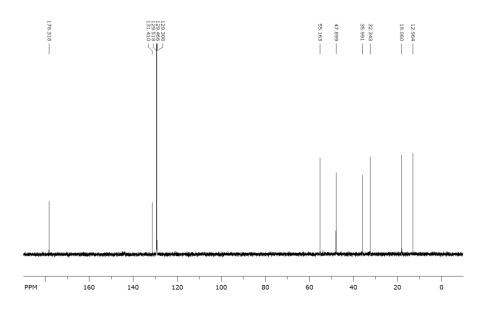


Figure S37. 13 C spectrum (100 MHz, D_2 O) of 2c

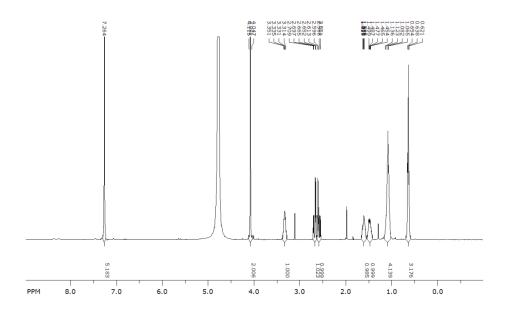


Figure S38. ¹H spectrum (400 MHz, D₂O) of 2d

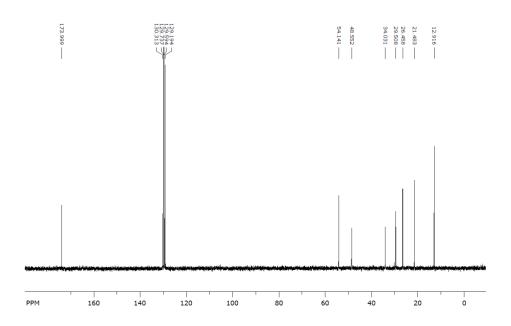


Figure S39. ¹³C spectrum (100 MHz, D₂O) of 2d

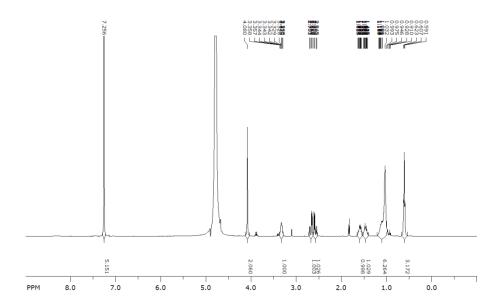


Figure S40. ¹H spectrum (400 MHz, D₂O) of 2e

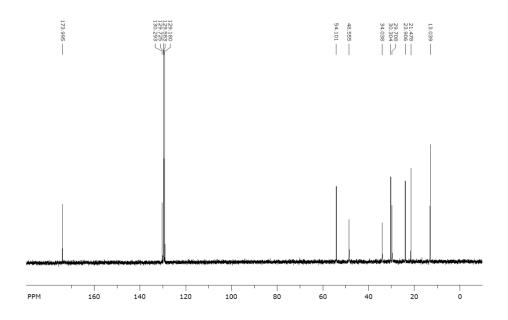


Figure S41. 13 C spectrum (100 MHz, D_2 O) of 2e

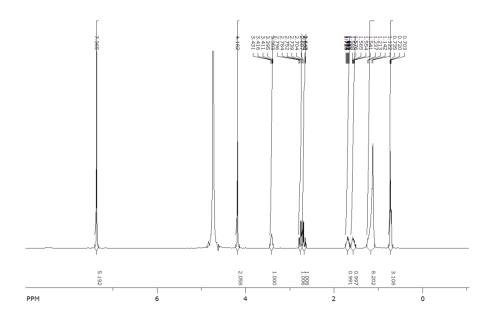


Figure S42. ¹H spectrum (400 MHz, D₂O) of 2f

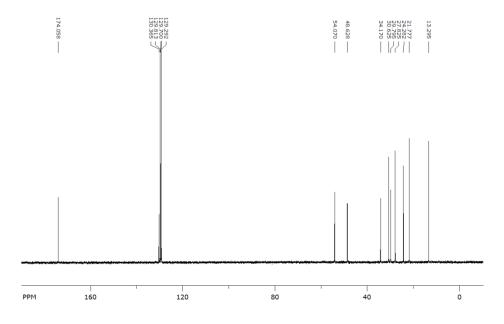


Figure S43. 13 C spectrum (100 MHz, D_2 O) of 2f

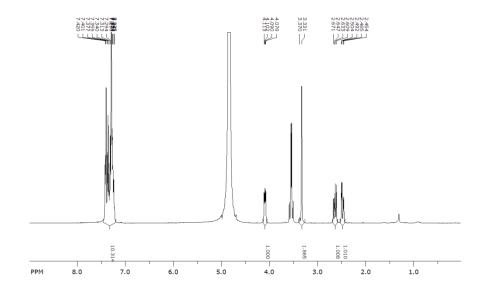


Figure S44. ¹H spectrum (400 MHz, CD₃OD) of 2g

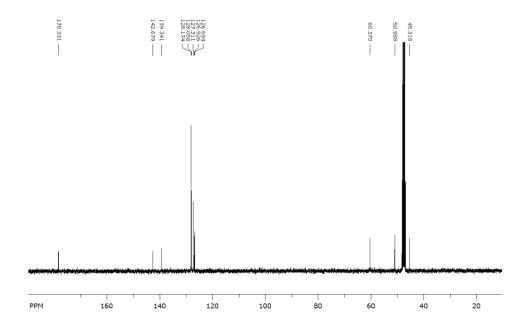


Figure S45. 13 C spectrum (100 MHz, CD $_{3}$ OD) of 2g

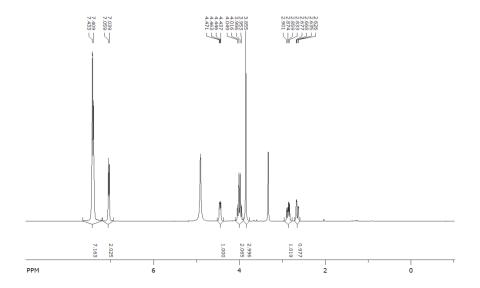


Figure S46. ¹H spectrum (400 MHz, CD₃OD) of 2h

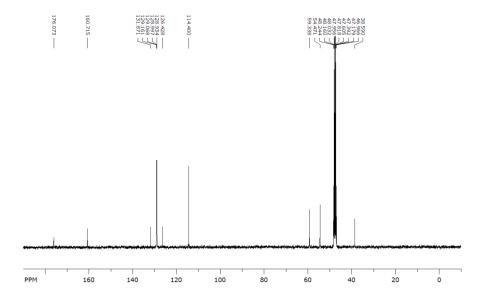


Figure S47. ¹³C spectrum (100 MHz, CD₃OD) of 2h

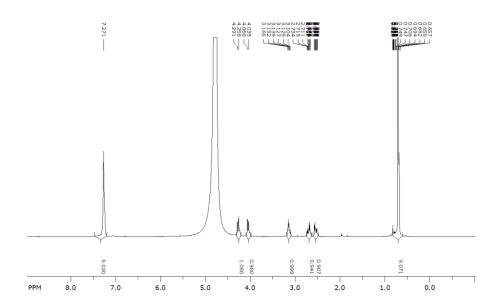


Figure S48. ¹H spectrum (400 MHz, D₂O) of 2i

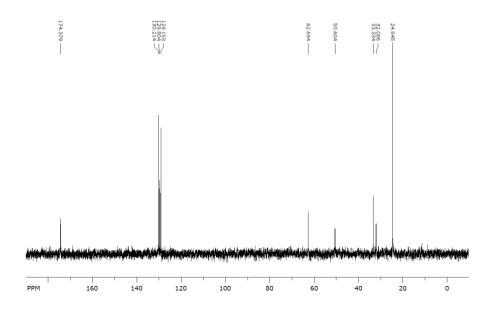
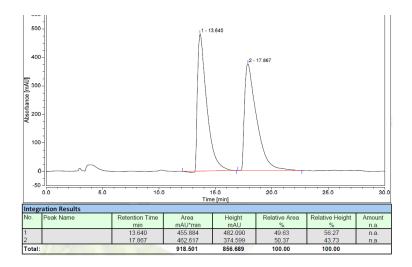
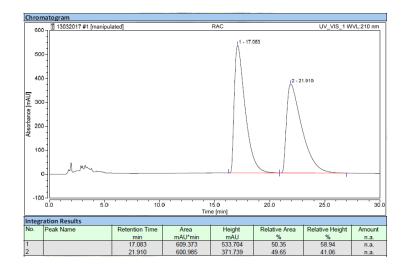


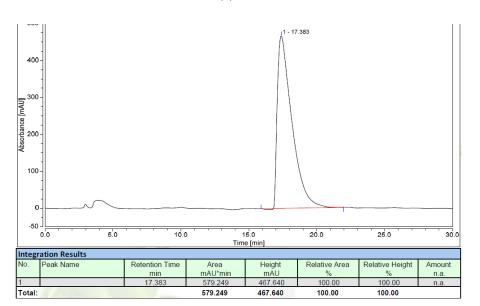
Figure S49. 13 C spectrum (100 MHz, D_2 O) of 2i

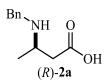
- 6) Chromatograms for the determination of enantiomeric excess of raw material and hydrolysis products.
 - Chromatograms for the search of best conditions for the enzymatic hydrolysis resolution under HSBM (Table S1).

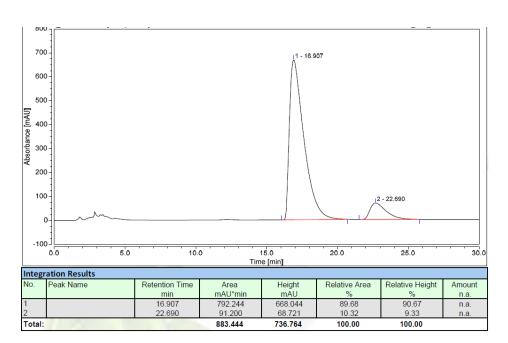




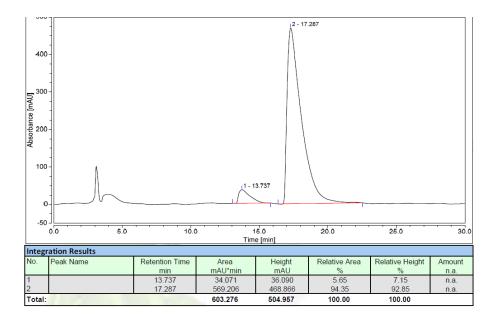
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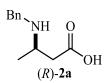


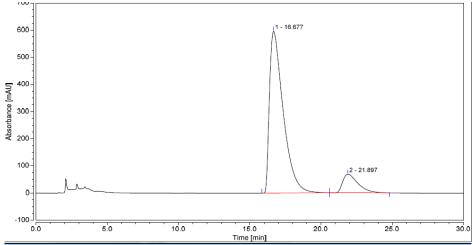




For entry 2.

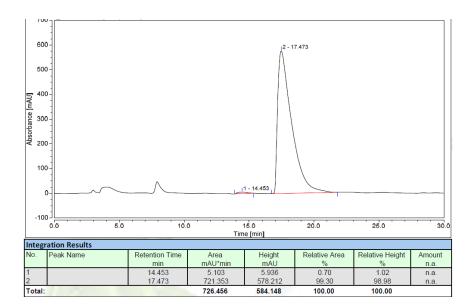


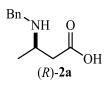


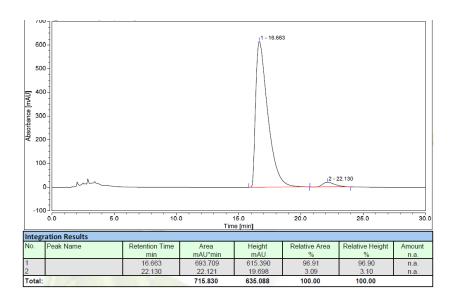


Integration Results												
No.	Peak Name	Retention Time	Area	Height	Relative Area	Relative Height	Amount					
		min	mAU*min	mAU	%	%	n.a.					
1		16.677	664.848	596.004	88.71	89.70	n.a.					
2		21.897	84.597	68.427	11.29	10.30	n.a.					
Total:			749.446	664.430	100.00	100.00						

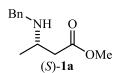
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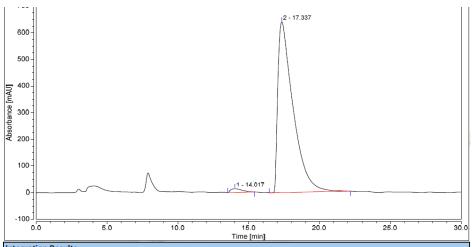




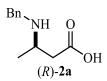


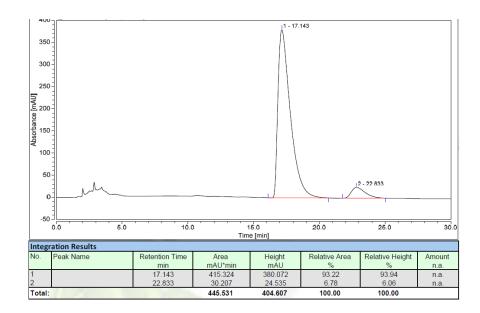
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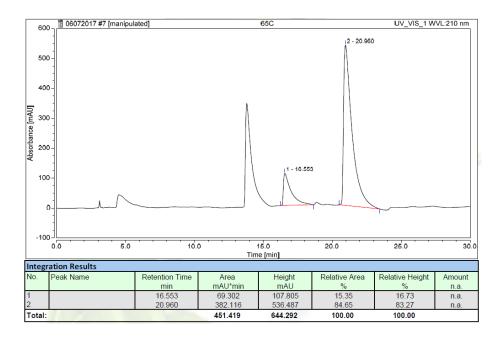


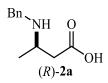
Integration Results											
No.	Peak Name	Retention Time	Area	Height	Relative Area	Relative Height	Amount				
		min	mAU*min	mAU	%	%	n.a.				
1		14.017	12.412	13.802	1.53	2.11	n.a.				
2		17.337	799.088	641.507	98.47	97.89	n.a.				
Total:			811.499	655.309	100.00	100.00					

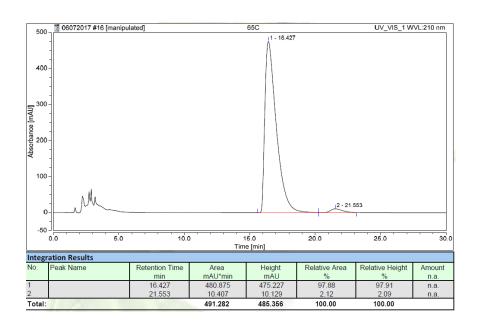




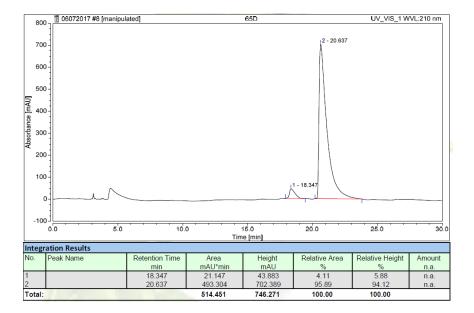
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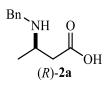


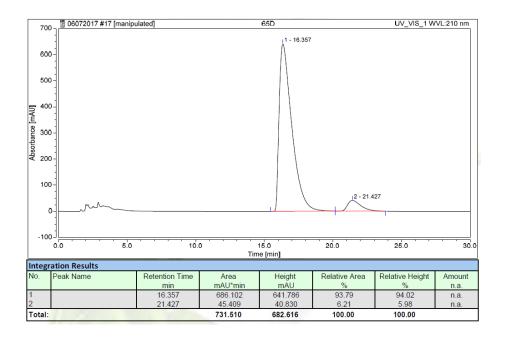




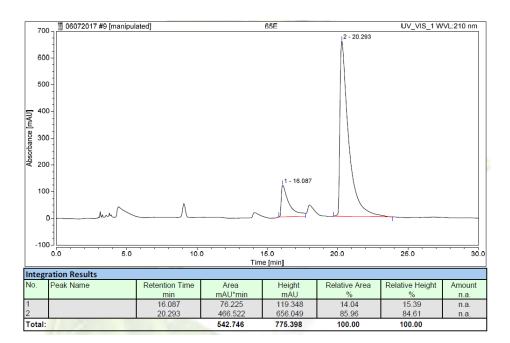
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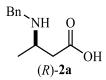


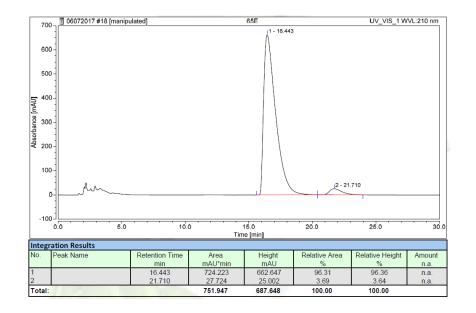




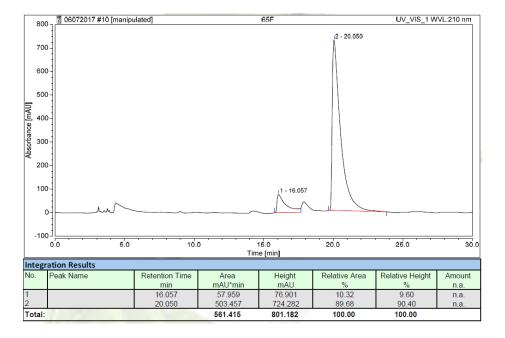
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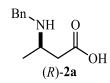


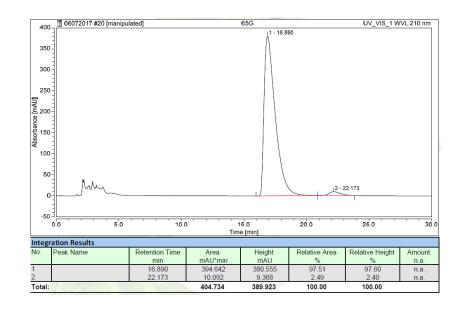




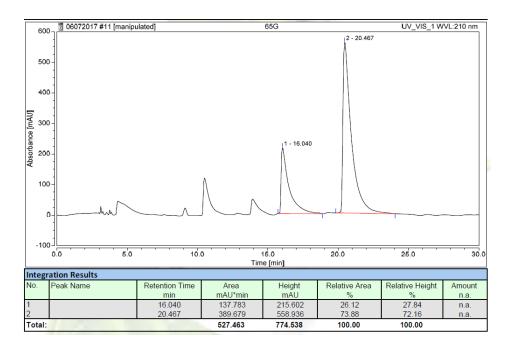
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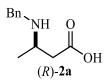


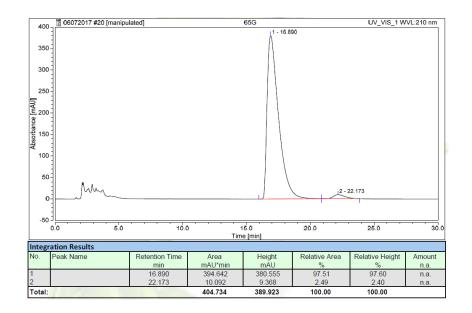




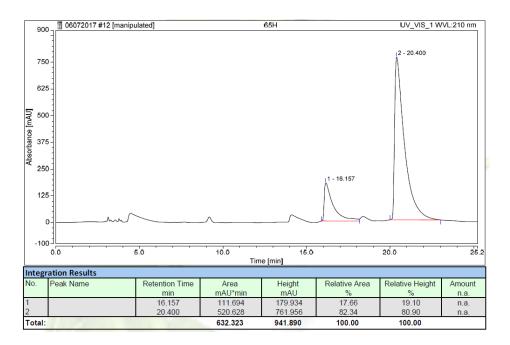
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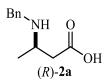


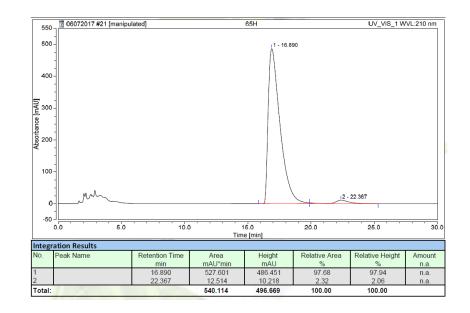




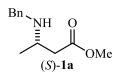
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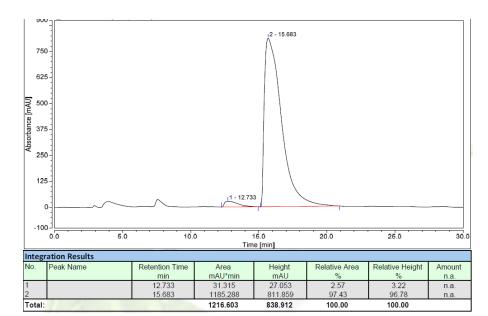


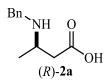


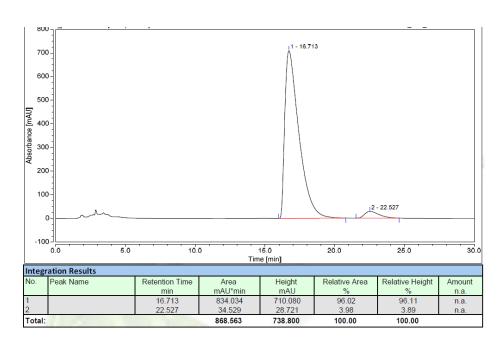


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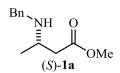


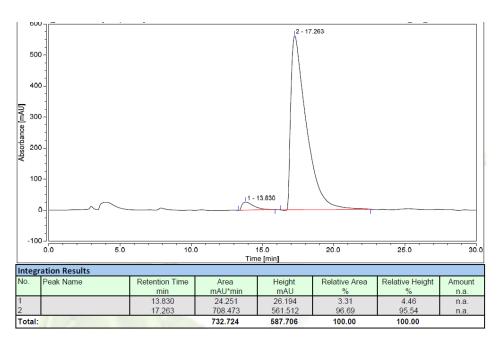


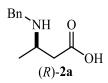


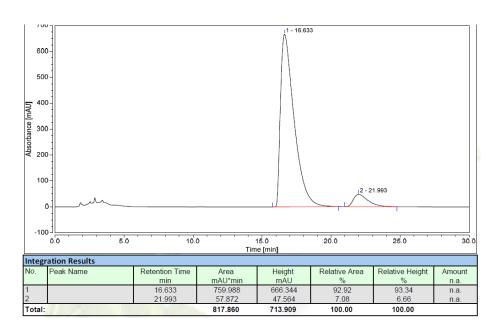


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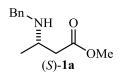


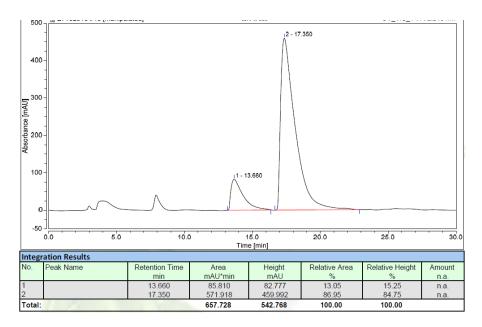


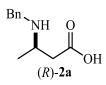


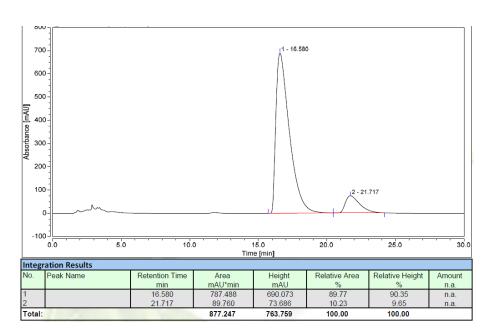


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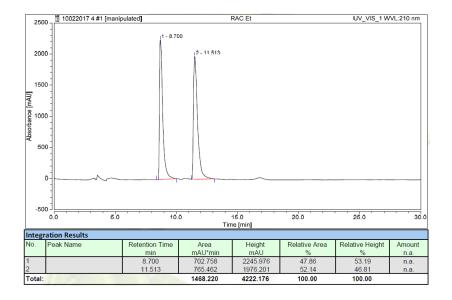


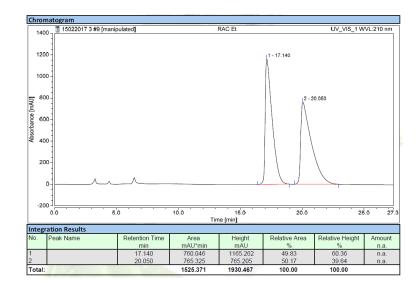




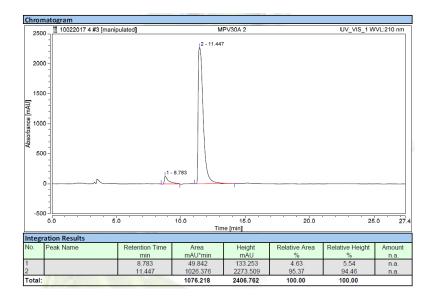


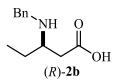
• Chromatograms for substrate scope for the enzymatic resolution of N-benzylated- β^3 -amino acid methyl esters (Table S2)

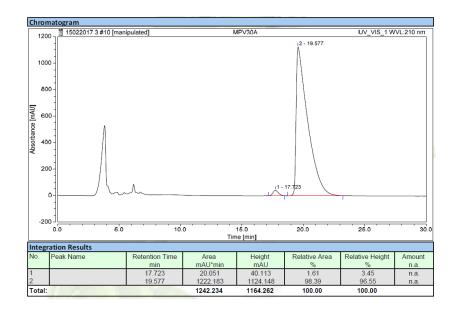


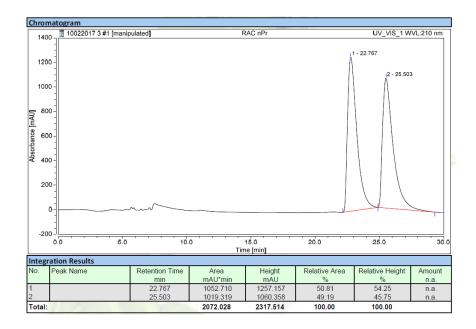


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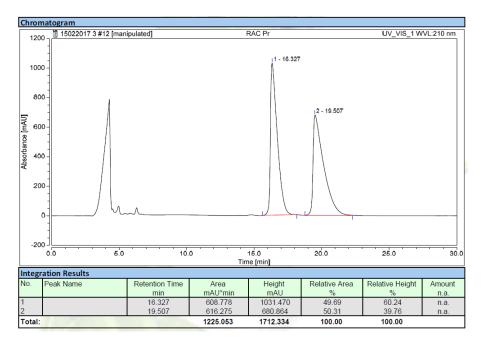




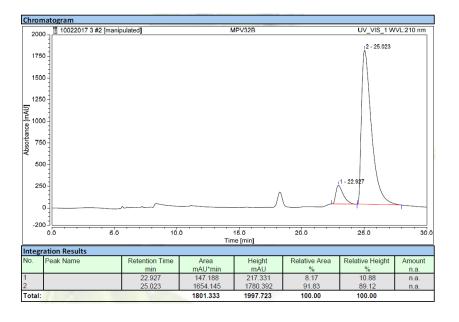


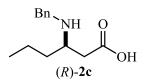


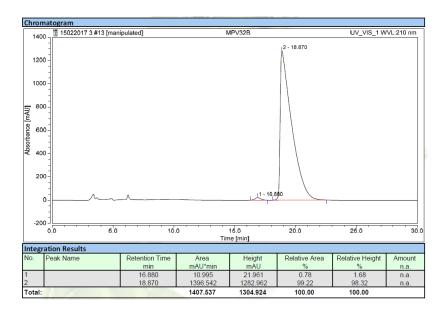


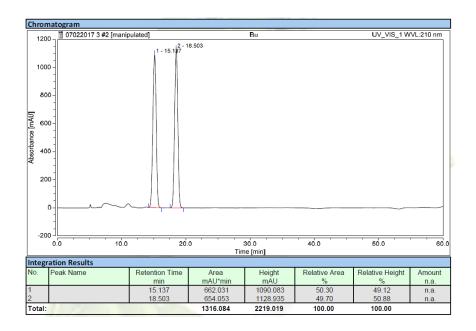


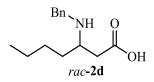
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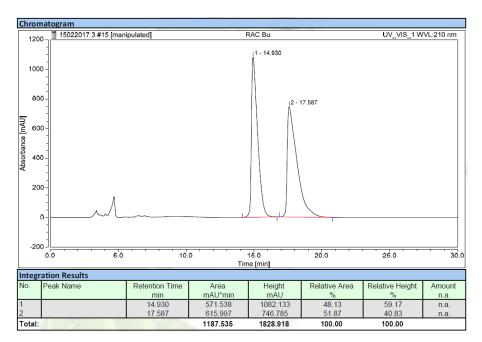




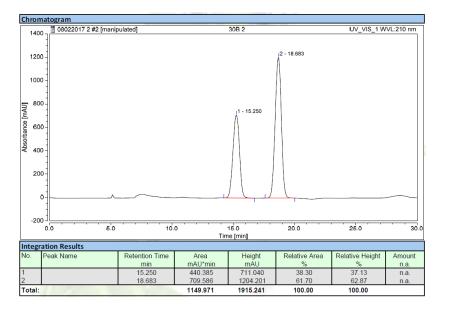


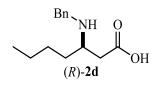


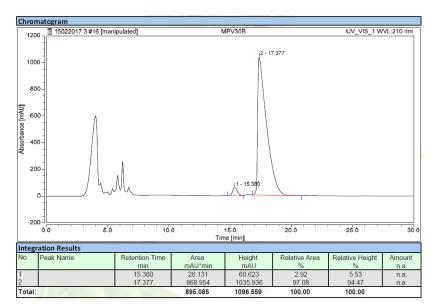




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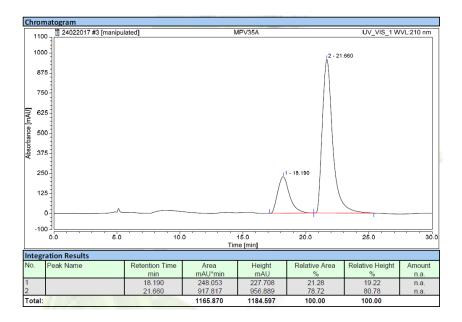


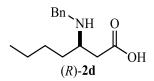


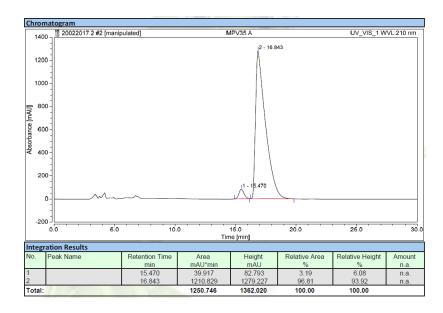


For entry 4.

$$\begin{array}{c}
\text{Bn} \\
\text{NH} \\
\text{O} \\
\text{OMe}
\end{array}$$



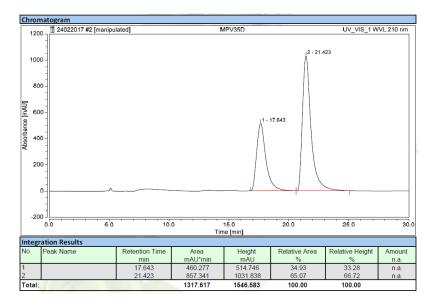


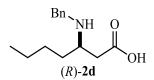


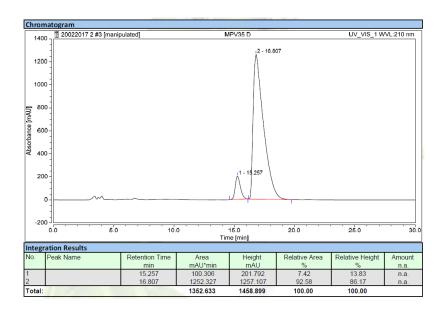
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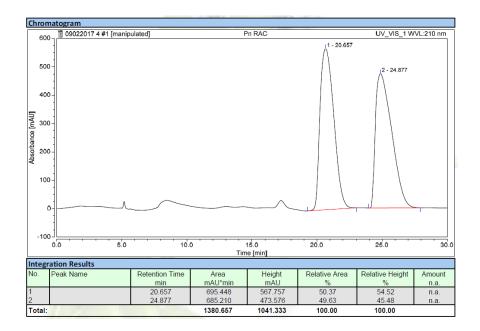
$$\begin{array}{c}
\text{Bn} \\
\text{NH} \\
\text{O}
\end{array}$$

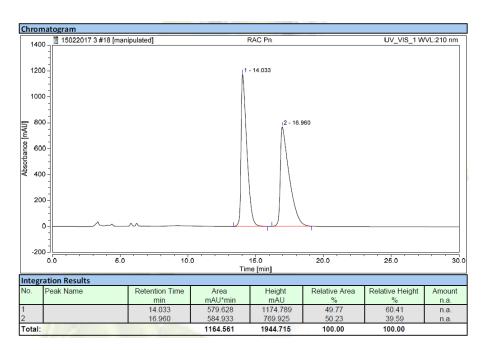
$$\begin{array}{c}
\text{OMe}
\end{array}$$



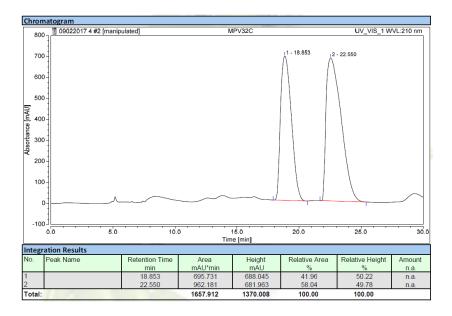




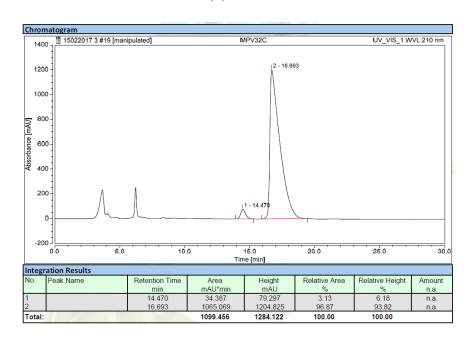




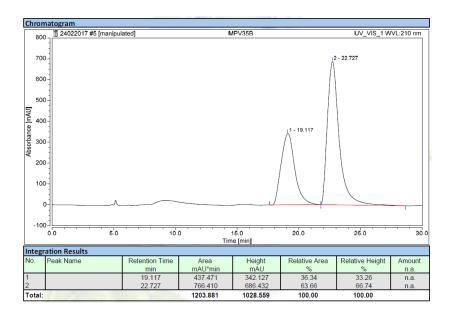
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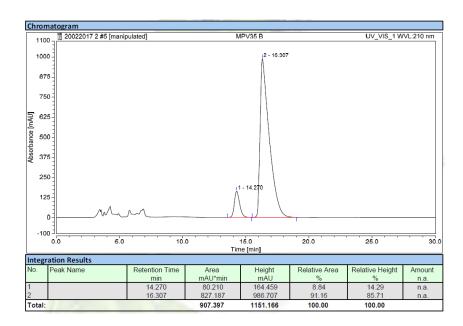
$$\begin{array}{c}
\text{Bn} \\
\text{NH} \\
\text{OH}
\end{array}$$



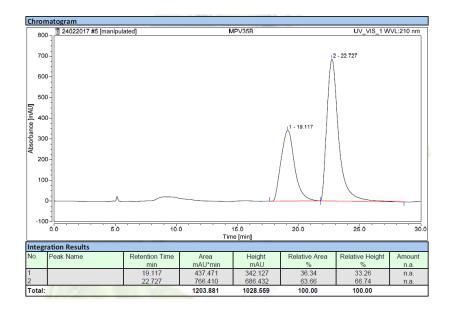
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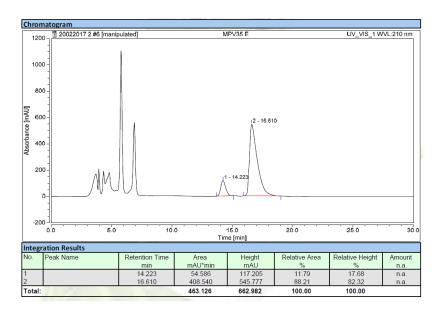
$$\begin{array}{c|c}
& \text{Bn} \\
& \text{NH} & \text{O} \\
& (R)\text{-2e}
\end{array}$$

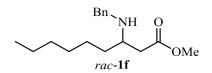


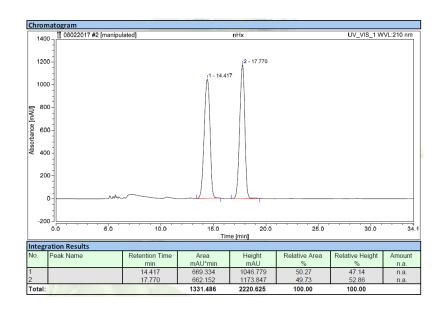
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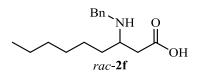


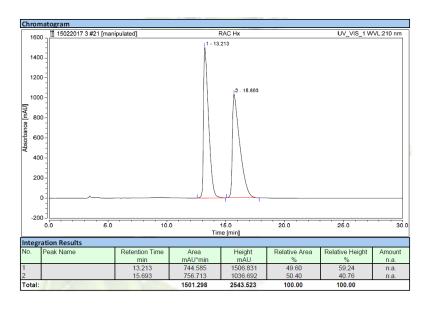






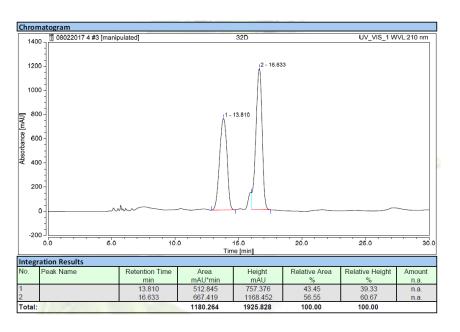




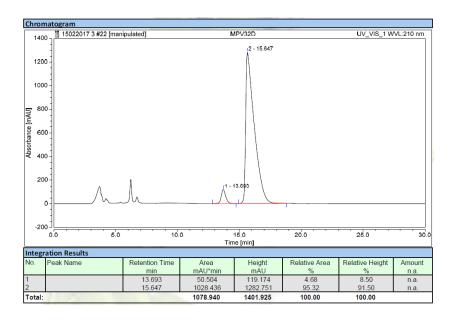


For entry 9.

$$\begin{array}{c|c}
& \text{Bn} & \text{NH} & \text{O} \\
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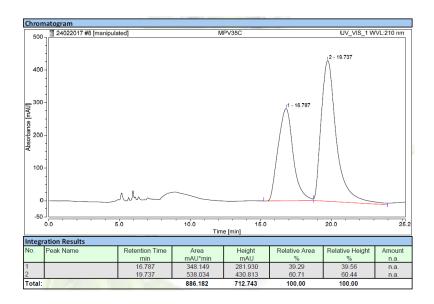




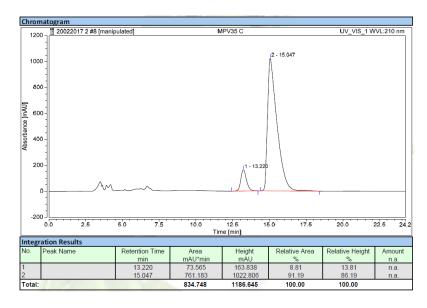


For entry 10.

$$\begin{array}{c|c}
& \text{Bn} \\
& \text{NH} & \text{O} \\
& & \\
\hline
& \text{OMe}
\end{array}$$

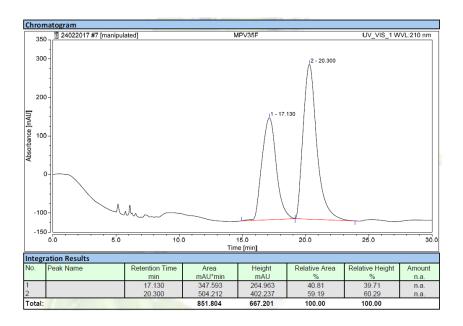


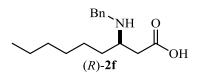


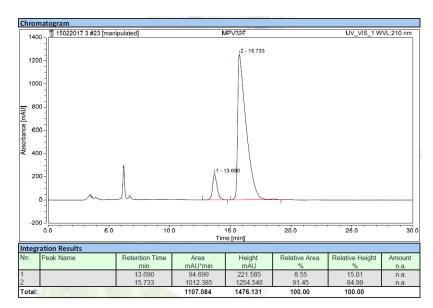


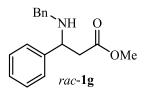
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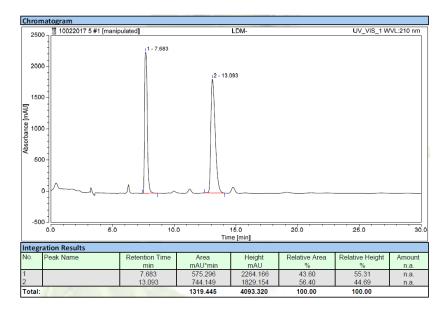
$$\begin{array}{c|c}
& \text{Bn} \\
& \text{NH} & \text{O} \\
& & \\
\hline
& \text{OMe}
\end{array}$$

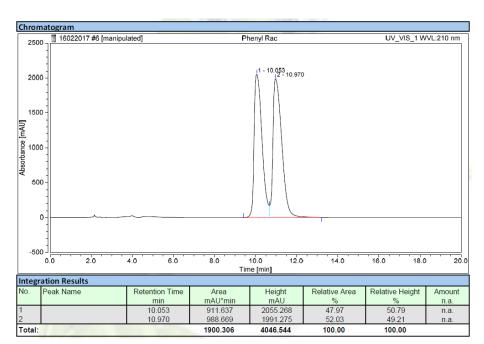




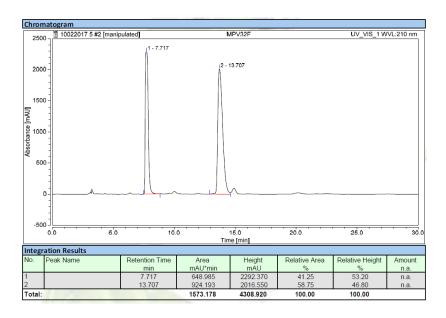


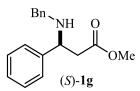


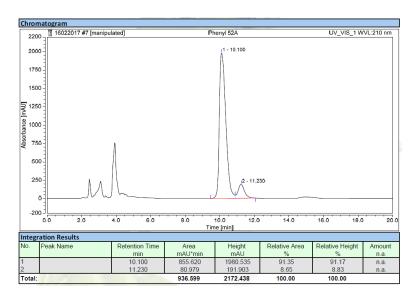




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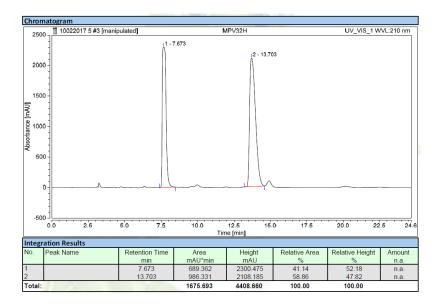


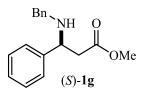


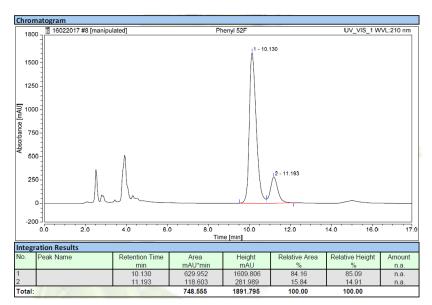


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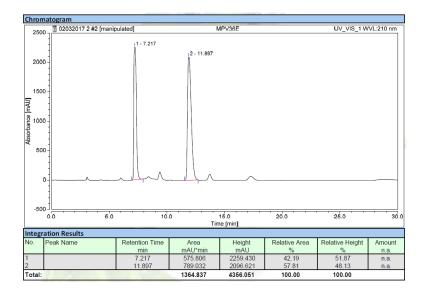
$$\begin{array}{c|c}
& \text{Bn} \\
& \text{NH} & \text{O} \\
& & \\
\hline
& \text{OMe}
\end{array}$$

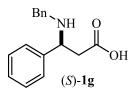


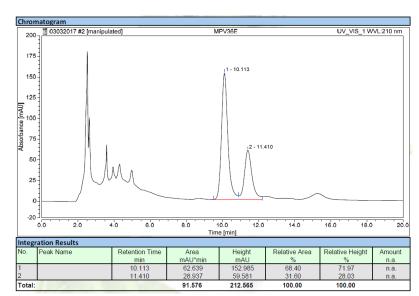




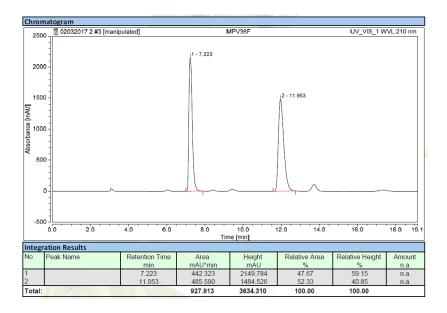
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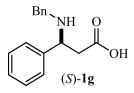


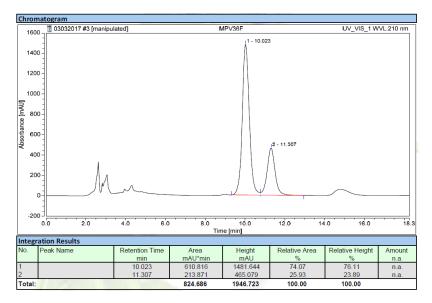


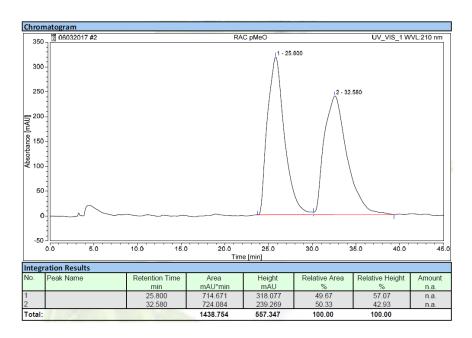


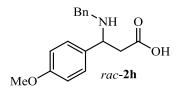
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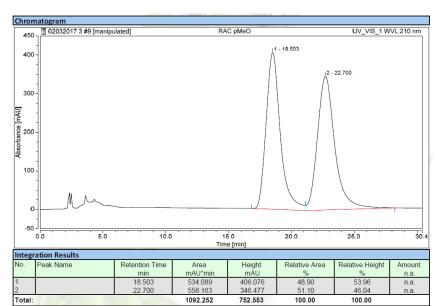




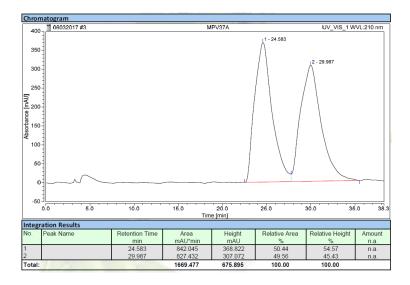


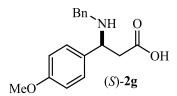


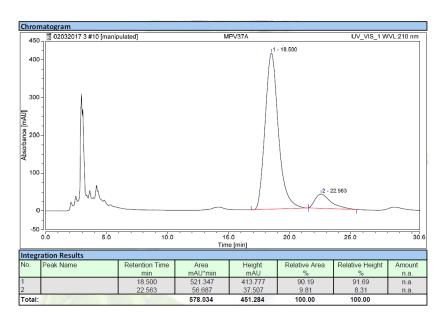


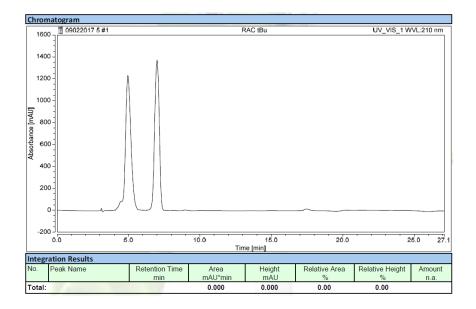


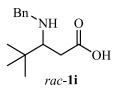
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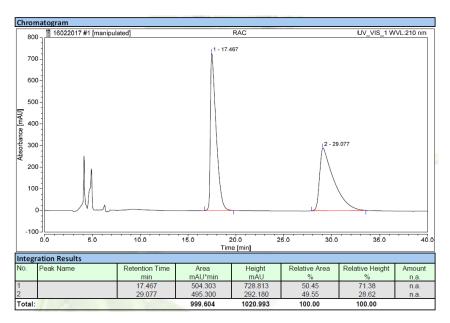




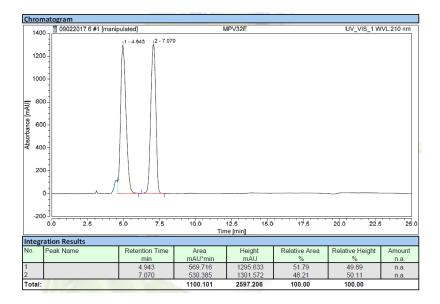


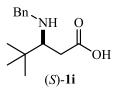


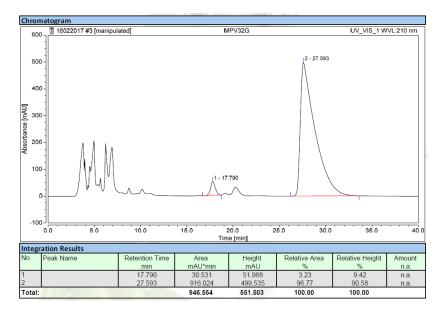




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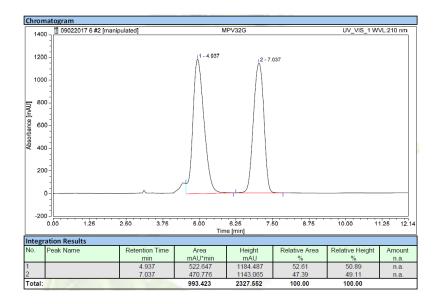


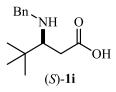


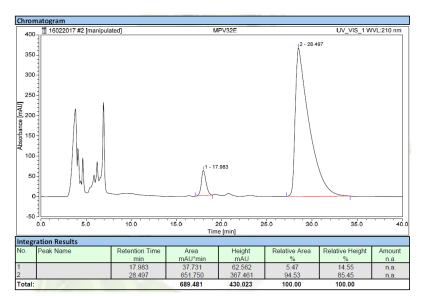


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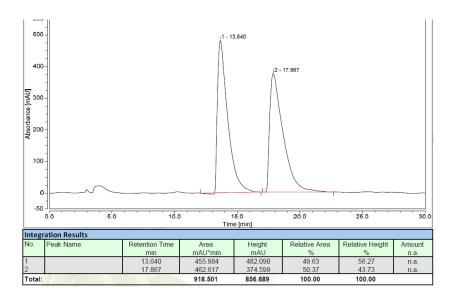
$$\begin{array}{c|c}
\text{Bn} & \text{NH} & \text{O} \\
& & \\
\hline
(R)-1i & \text{OMe}
\end{array}$$

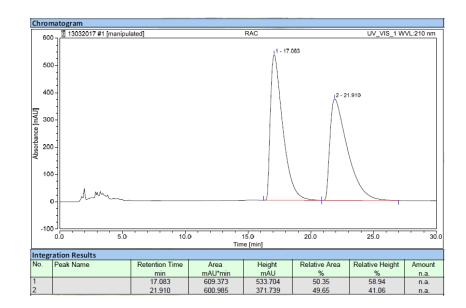




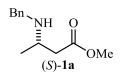


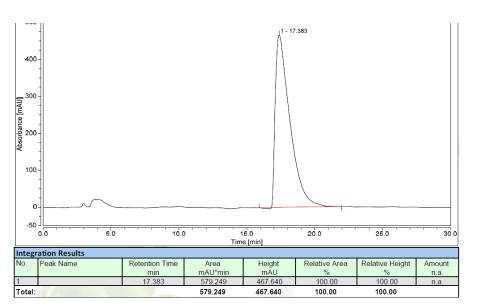
 Chromatograms for the evaluation of recycling capacity of immobilized CALB (Table S3).



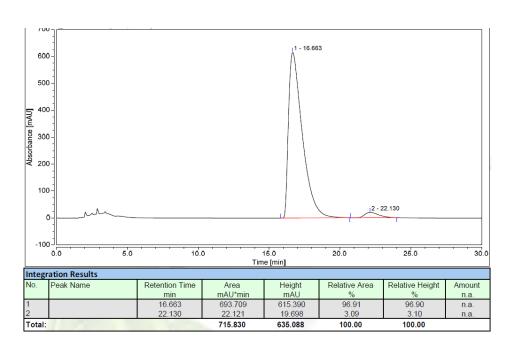


For entry 2.

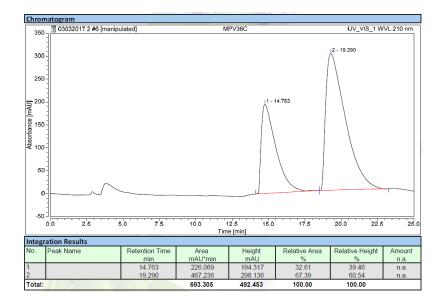


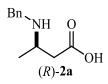


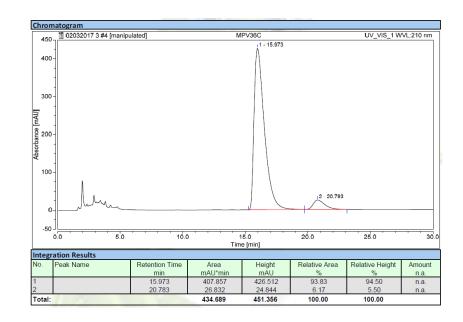
$$\begin{array}{c}
\text{Bn} \\
\text{NH} \\
\text{OH}
\end{array}$$



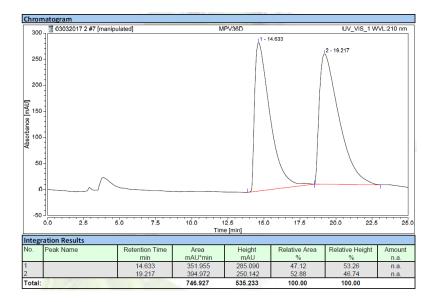
For entry 3.

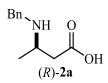


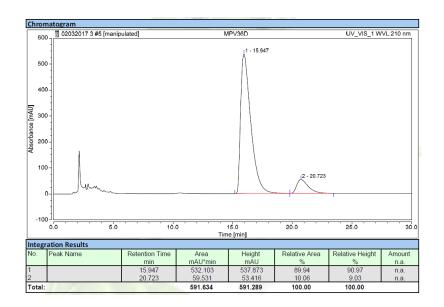




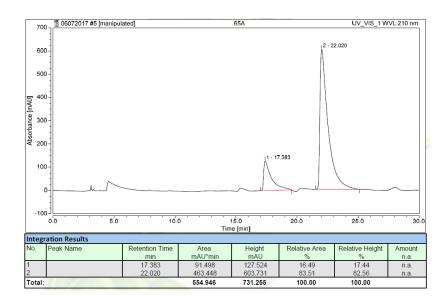
For entry 4.

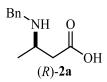


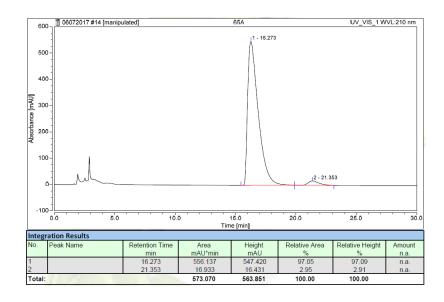




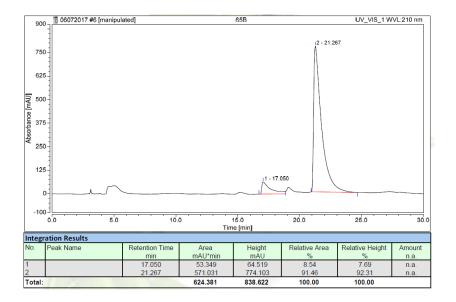
For entry 5.

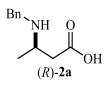


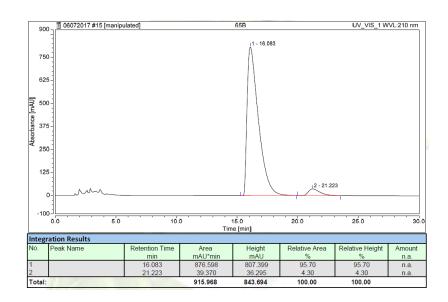




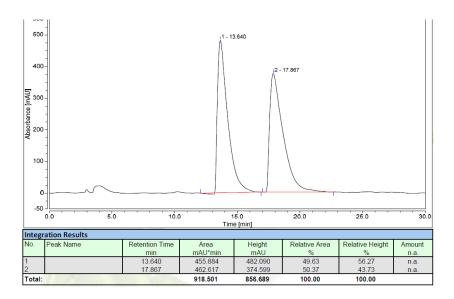
For entry 6.

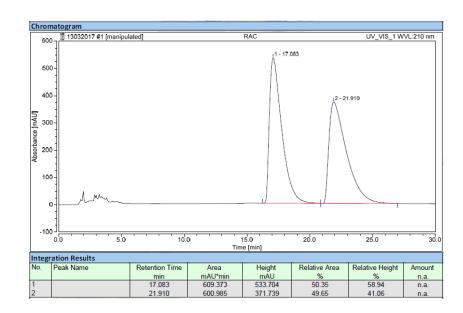




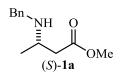


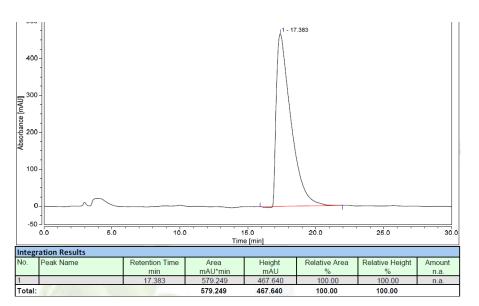
Chromatograms for scaling-up for the enzymatic hydrolysis using the substrate
 1a (Table S4).



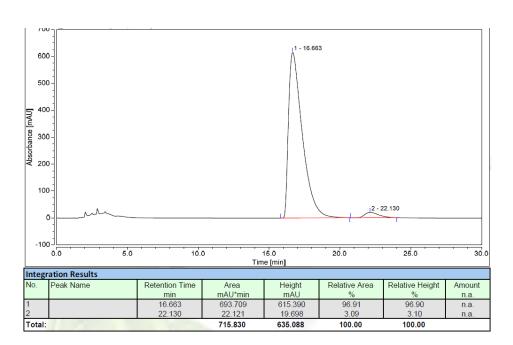


For entry 1.

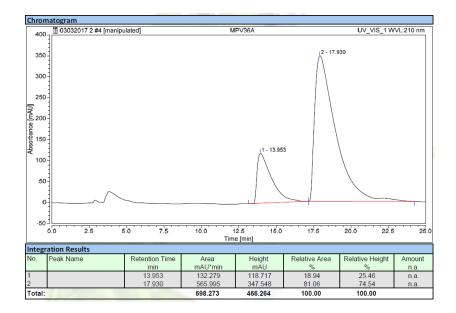


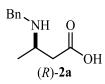


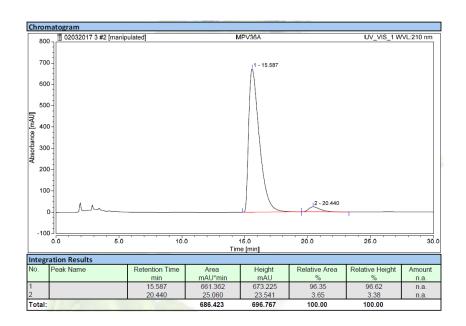
$$\begin{array}{c|c}
\text{Bn} & \text{OH} \\
\hline
(R)-2a & \text{OH}
\end{array}$$



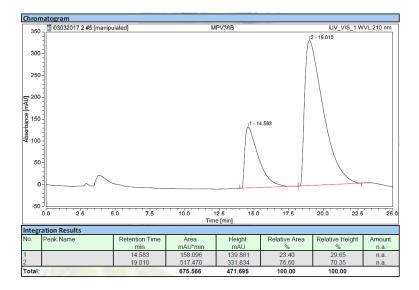
For entry 2.



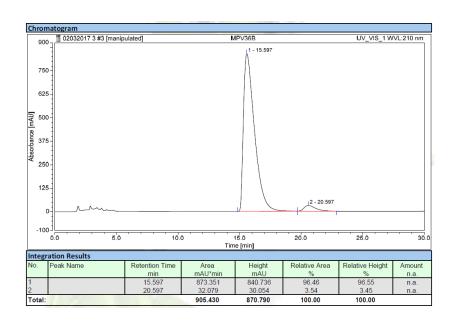




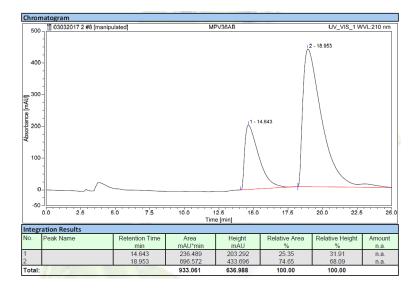
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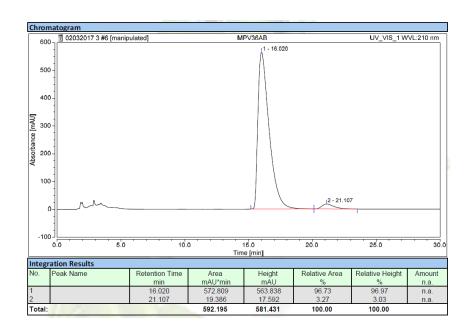


$$\begin{array}{c}
\text{Bn} \\
\text{NH} \\
\text{OH}
\end{array}$$



For entry 4.





7) HPLC conditions and retention times for the substrates *rac*-1a to *rac*-1j and hydrolysis products 2a to 2j.

Substrates			Products		
Conditions			Conditions		
	Column, Eluent, Flow(mL/min), Time(min)		Column, Eluent, Flow(mL/min), Time(min)		
1a	Chiralpack OD , Hex/IPA (99:1), 1, 30	2a	Chirobiotic TAG , MeOH, 1, 30		
1b	Chiralpack OD-H , Hex/IPA (98:2), 1, 30	2b	Chirobiotic \mathbf{T} , EtOH/H ₂ O (70:30), 0.5, 30		
1c	Chiralpack OD-H , Hex/IPA (99:1), 0.6, 30	2c	Chirobiotic T, EtOH/ H_2O (70:30), 0.5, 30		
1d	Chiralpack AD-H , Hex/IPA (99:1), 0.6, 30	2d	Chirobiotic T, EtOH/ H_2O (70:30), 0.5, 30		
1e	Chiralpack AD-H , Hex/IPA (99:1), 0.6, 30	2e	Chirobiotic T , EtOH/H ₂ O (70:30), 0.5, 30		
1f	Chiralpack AD-H , Hex/IPA (99:1), 0.6, 30	2f	Chirobiotic T , EtOH/H ₂ O (70:30), 0.5, 30		
1g	Chiralpack OD-H, Hex/IPA (95:5), 1, 30	2g	Chirobiotic T, MeOH/ H_2O (95:5), 0.8, 20		
1h	Chiralpack OD , Hex/IPA (99:1), 0.9, 45	2h	Chirobiotic TAG , MeOH, 0.8, 30		
1i	Chiralpack AD-H, Hex/IPA (99:1), 1, 15	2i	Chirobiotic T, EtOH/ H_2O (70:30), 0.5, 40		

Substrates				Products		
Retention time (min)			Retention time (min)			
	t _R	t _S		t _R	t _S	
1a	13.64	17.86	2a	17.08	21.91	
1b	8.70	11.51	2b	17.14	20.50	
1c	22.76	25.50	2c	16.31	19.50	
1d	15.13	18.50	2d	14.93	17.58	
1e	20.65	24.87	2e	14.03	16.96	
1f	14.41	17.70	2f	13.21	15.69	
1g	13.09	7.68	2g	10.97	10.05	
1h	24.58	32.58	2h	7.96	7.13	
1i	4.93	7.03	2i	17.46	29.07	

8) Determinations of the absolute configuration for the products 2a to 2j.

For the determination of the absolute configuration of all products we used different methodologies. A crystallographic report for product (*R*)-2a was obtained and the ORTEP diagram are show below, as well, a CIF report is attached to this archive, the absolute configuration for this product was determined as *R*, furthermore, the specific optical rotation was compared with the reported by Escalante [4] (Table S2, entry 1).

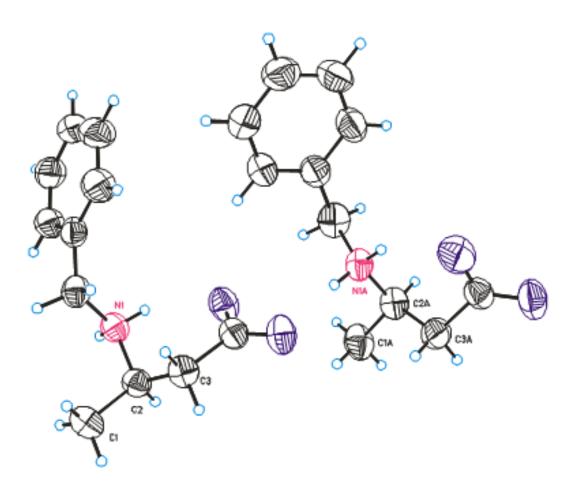
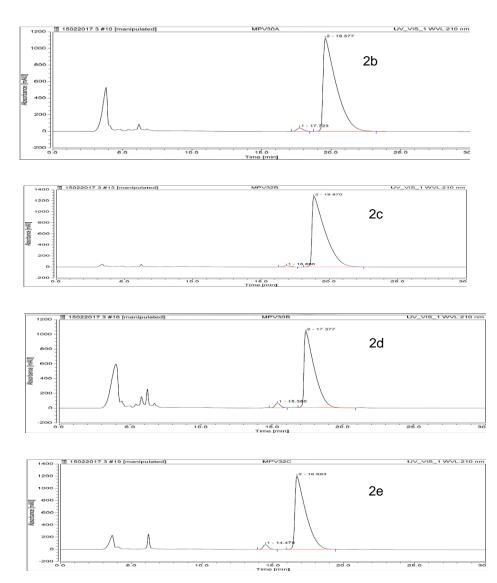
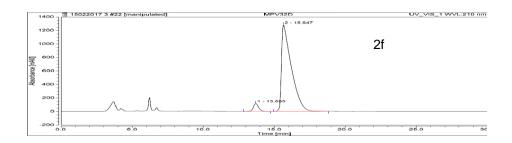


Figure S50. ORTEP diagram for product (R)-2a.

The absolute configuration of product **2b** was assigned by chemical comparison with the reported data [17], **reference** (*S*)-**2b** $[\alpha]_D^{25} = +32.4$ and ee>99%, **product** (*R*)-**2b** $[\alpha]_D^{25} = -36.5$ and ee=97%, obtaining an *R* configuration. The products **2c** to **2f** were assigned followed an elution sequence general rule already describe [18] using the same column that we used (Chirobiotic T). With the certainty that product **2b** has a well know absolute configuration and that the elution sequence for the majority product is the same for product **2b** to **2f** we assigned the same configuration for all them (*R*).





The absolute configuration for products **2g** and **2i** was assigned by chemical comparison [4], **reference** (*S*)-**2g**⁴ [α]_D²⁰ = -51.9 and *ee*>99%, **product** (*S*)-**2g** [α]_D²⁵ = -35.0 and *ee*= 83%, **reference** (*S*)-**2i**⁴ [α]_D²⁰ = +26 and *ee*>99%, **product** (*S*)-**2i** [α]_D²⁵ = +12.8 and *ee*= 94%, and for product **2h** an elution sequence general rule was used, compare product **2a** and **2h**, obtaining the *S* configuration for product **2h**.

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