Supporting Information

for

Cationic cobalt-catalyzed [1,3]-rearrangement of *N*-alkoxycarbonyloxyanilines

Itaru Nakamura^{1*}, Mao Owada², Takeru Jo² and Masahiro Terada^{1,2}

Address: ¹Research and Analytical Center for Giant Molecule, Graduate School of Science, Tohoku University, 6-3 Aramaki Aza Aoba, Aoba-ku, Sendai 980-8578 Japan and ²Department of Chemistry, Graduate School of Science, Tohoku University, 6-3 Aramaki Aza Aoba, Aoba-ku, Sendai 980-8578 Japan Email: Itaru Nakamura - itaru-n@tohoku.ac.jp

General procedure and analytic data for obtained products

General information	S2
2. Thermally induced reaction of 1y'	S2
Crossover experiment (analyzed by HRMS)	S3
Intramolecular competitive experiments	S5
5. ¹⁸ O-Labeling experiments	S7
6. Analytical data of 1	S18
7. Analytical data of 2	S21
8. Analytical data of 3	S23
9. ¹ H and ¹³ C NMR chart of 1 , 2 , 3 , and 4	S24
10. Reference	S37

^{*} Corresponding author

1. General Information

 1 H and 13 C NMR spectra were recorded on a JEOL JNM-ECS400 (400 MHz for 1 H and 100 MHz for 13 C) spectrometer and JEOL JNM-ECA600 (150 MHz for 13 C)) spectrometer. Chemical shifts are reported in ppm relative to CHCl₃ (for 1 H, δ 7.26), and CDCl₃ (for 13 C, δ 77.00). 1 H NMR data are reported as follows: chemical shift, integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, sext = sextet, sept = septet, dd = double doublet, dt = double triplet, dq = double quartet, ddt = double double triplet, br = broad, m = multiplet) and coupling constants (Hz). Infrared spectra were recorded on a JASCO FT/IR-4100 spectrometer. High-resolution mass spectra analysis was performed on a Bruker Daltonics solariX FT-ICR-MS spectrometer at the Research and Analytical Center for Giant Molecules, Graduate School of Science, Tohoku University. Flash column chromatography was performed with Kanto Chemical silica gel 60N (spherical, neutral, 40–50 μm). Analytical thin layer chromatography (TLC) was performed on Merck precoated TLC plates (silica gel 60 F₂₅₄). All reactions were carried out under argon atmosphere.

Preparative procedure for the starting material 1, general procedures for the cobalt-catalyzed reaction of 1, and spectroscopic data for starting materials 1a-d, 1f-i, 1k, 1m-v, 1y-ab, and products 2a-i, 2k, 2o-y' are described in Supporting Information of our previous report [S1].

Materials

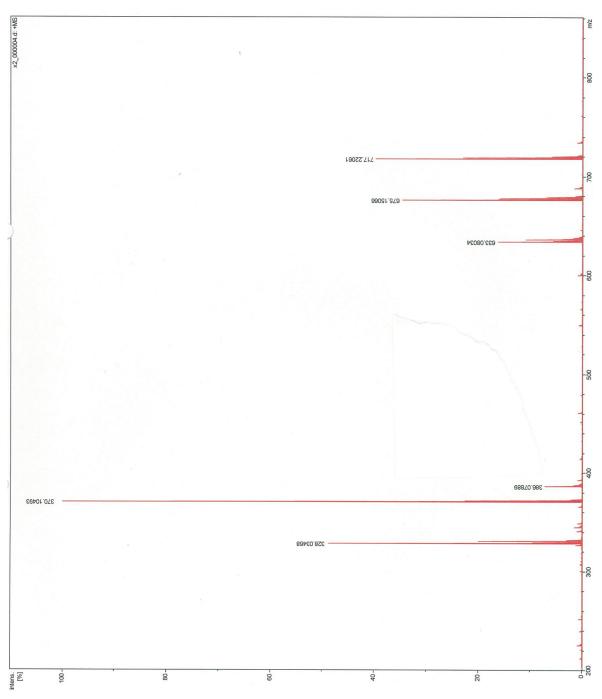
CoCl₂ was purchased from Wako and used as received.

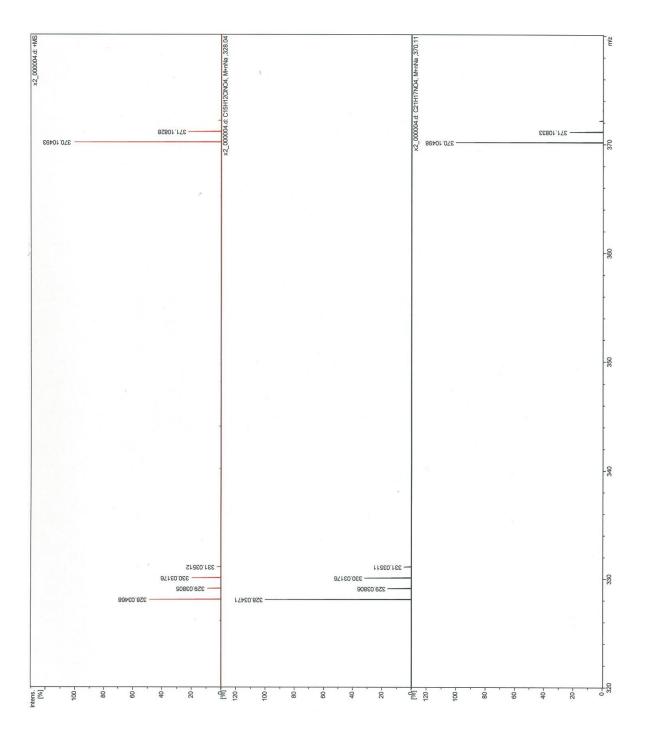
AgSbF₆ was purchased from Aldrich and used as received.

Water-18 (98% ¹⁸O) was purchased from ISOTEC and used as received.

2. Thermally induced reaction of 1y'

3. Crossover experiment analyzed by HRMS [S1]



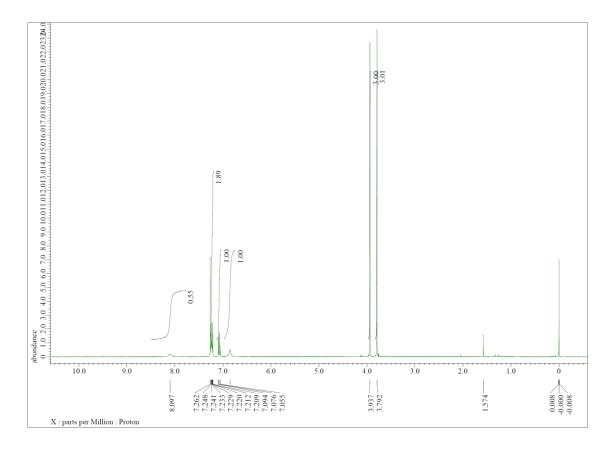


4. Intramolecular competitive experiments [S1]

Synthesis of nitrobenzene-d₁

To a solution of **S5** (3.0 g, 15 mmol) in THF (50 mL) under argon atmosphere was added dropwise a 1.6 M butyl ether solution of phenyllithium (9.4 mL, 15 mmol) at -78 °C. After stirring the solution at -78 °C for 5 hours, the reaction was quenched by saturated aqueous sodium chloride. Then, the mixture was extracted with ether and the organic layer was washed by brine and dried over sodium sulfate. After the solvent was removed in vacuo, the residue was purified by distillation to obtain **S1**- d_1 (89% D, 0.9 mmol, 6% yield).

2a-d



5. ¹⁸O-Labeling experiments [S1]

a) Preparation of nitrobenzene-¹⁸O

¹⁸O-labelled sodium nitrate [S2] and nitrobenzene-¹⁸O [S3] were prepared according to a reported procedure. ¹⁸O content was determined by intensity of mass spectra.

$$NaN^{18}O_3$$
 + CF_3CO_2H $S1-^{18}O_2$

To water-¹⁸O was (0.3 mL, 16.6 mmol) in a 10 mL flask was added fumed nitric acid (d = 1.5, 0.1 mL, 2.4 mmol) and the solution in the tightly sealed flask was stirred at room temperature for 4 days. Then, to the solution was added sodium hydroxide (382 mg) in water-¹⁸O (0.344 mL) and stirred for 24 hours. After water-¹⁸O was distilled, fumed nitric acid (0.016 mL) was added to the residue. The mixture was stirring at 70 °C for 24 hours. After confirming that the solution was not acidic by pH test paper, water-¹⁸O was distilled out. To the residue was added benzene (0.14 mL, 1.53 mmol) and trifluoroacetic acid (7.7 mL). The solution was stirred at room temperature for 24 hours. The reaction was quenched by 3 N NaOH in water-¹⁸O until the solution was neutralized. The mixture was extracted three times with ether. Ether of the organic layer was distilled out. The obtained nitrobenzene-¹⁸O (¹⁸O 56%) was utilized without further purification.

$$N^{18}O_2$$
 Zn NH_4CI NH_4

To a solution of nitrobenzene-¹⁸O in THF (3.6 mL) were added water-¹⁸O (1.8 mL) and ammonium chloride (52.96 mg) at 0 °C under argon atmosphere. Then, the solution was added portion-wise zinc power (91.53 mg). After the solution was stirred at 0 °C for 2.5 hours, The reaction mixture was passed through a short pad of silica gel with ethyl acetate. After solvents were removed in vacuo, the crude product was purified by flash silica gel column chromatography using hexane/ethyl acetate (3/1) as eluents. After column chromatography, **S2**-¹⁸O was not completely concentrated in order to avoid decomposition.

To a solution of **\$2**-¹⁸O in THF (8 mL) was added pyridine at 0 °C under argon atmosphere. Benzoyl chloride was added dropwise to the solution. The solution was stirred at 0 °C with monitoring by TLC. After complete consumption of **\$2**-¹⁸O for 1 hour, the solution was quenched by ammonium chloride. The reaction mixture was passed through a short pad of silica gel with dehydrated dichloromethane. After removing solvents in vacuo, the crude product was purified by silica gel column chromatography using hexane/ethyl acetate (3/1 to 2/1) as eluent to obtain **\$3**-¹⁸O (72% ¹⁸O, 0.4 mmol, 16% yield based on fumed nitric acid in the first step).

To a solution of **S3**- 18 O (94 mg, 0.396 mmol) in THF (8.8 mL) was added pyridine (36 μ L, 0.44 mmol) at 0 °C under argon atmosphere. Then, benzyl chloroformate (64 μ L, 0.45 mmol) was added dropwise to the solution. After 1 hour, the reaction was quenched by water and the reaction mixture was extracted with ether three times. The organic layer was washed with brine and dried over sodium sulfate. The crude product was purified by silica gel column chromatography to obtain **1h**- 18 O (62% 18 O, 0.37 mmol, 84% yield).

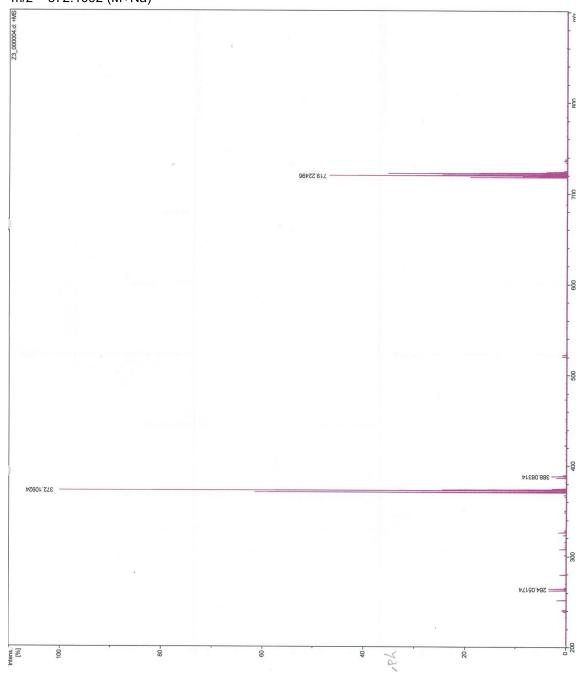
To a solution of **2h**-¹⁸O (34.8 mg, 0.10 mmol) in methanol (8.3 mL) in a 20 mL round-bottom flask was added 5% Pd/C under nitrogen atmosphere. Nitrogen gas was evacuated until

the solvent began to bubble, and then the flask was filled with hydrogen gas. The suspension was stirred at room temperature for 24 hours. The reaction mixture was filtered through celite and the filtrate was concentrated in vacuo. The residue was passed through a short pad of silica gel with ethyl acetate. The solution was washed with water and the water layer was extracted with ether and ethyl acetate. The combined organic layer was concentrated in vacuo to obtain **4h**-¹⁸O (64% ¹⁸O, 0.053 mmol, 53% yield) in an analytically pure form.

To $1h^{-18}O$ (17.4 mg, 0.05 mmol) were added potassium carbonate (0.40 mg, 2.89 mmol) and methanol (0.3 mL). Then, the suspension was stirred for 6 hours. The mixture was purified by silica gel column chromatography using hexane/ethyl acetate (2.5:1) as eluent to obtain $$3^{-18}O$$ (65% ^{18}O) in 38% yield.

This result clearly indicates that oxygen-18 existed at the hydroxylamine moiety in 1h-18O.



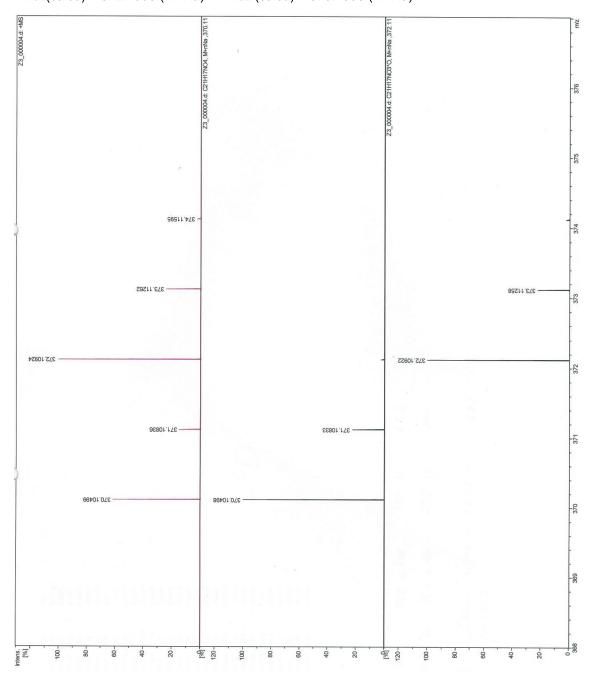


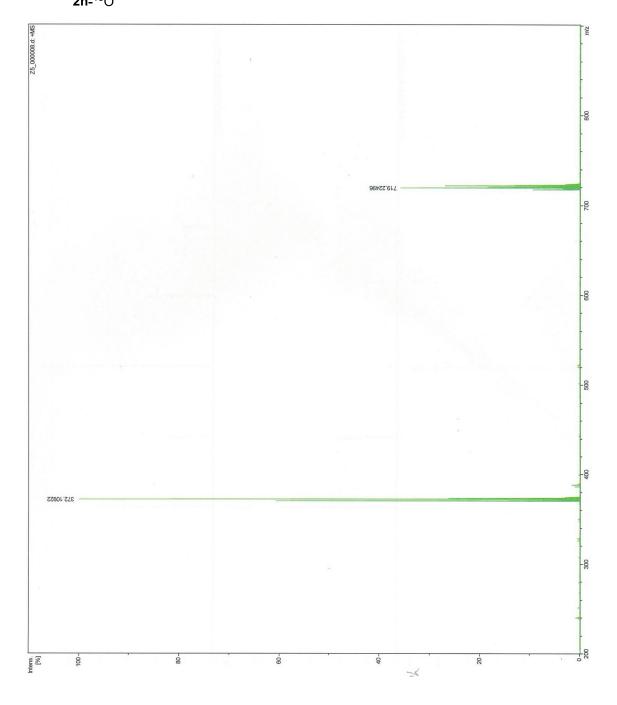
1h-¹⁸O

 $m/z(calcd) = 372.1098 (M+Na)^{+}$

1h

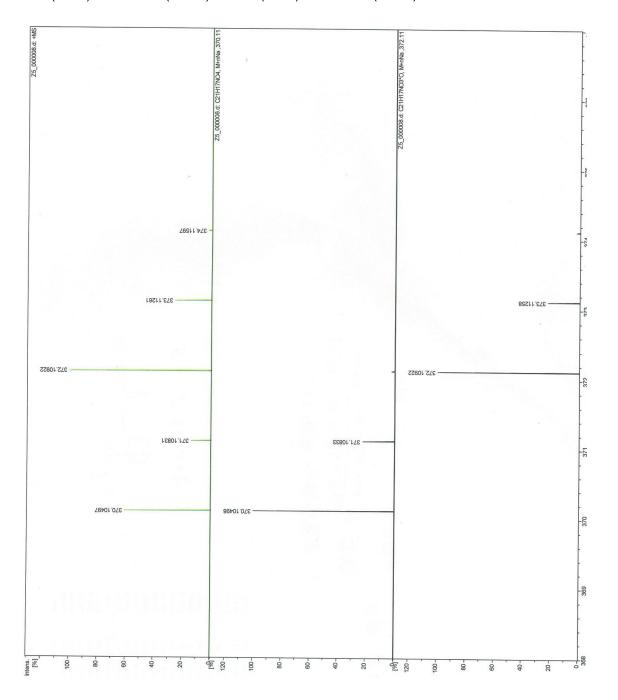
m/z (calcd) = 370.1055 $(M+Na)^+$





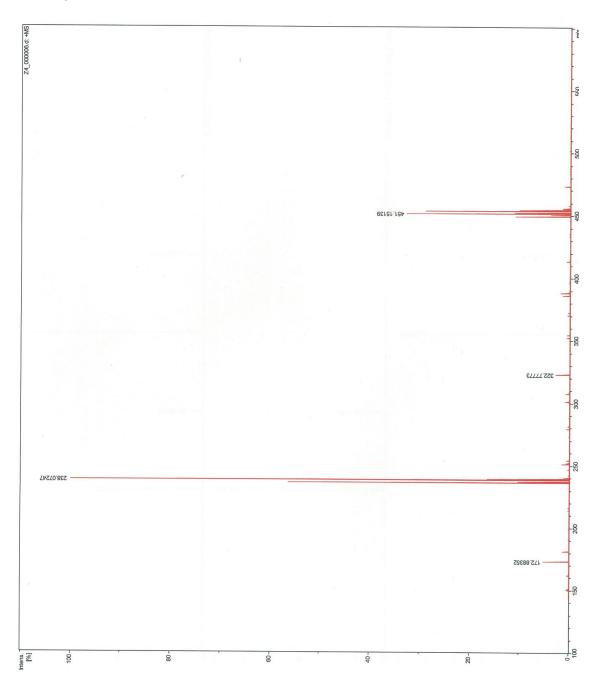
 $m/z(calcd) = 372.1092 (M+Na)^+$

m/z (calcd) = 370.1055 $(M+Na)^+$

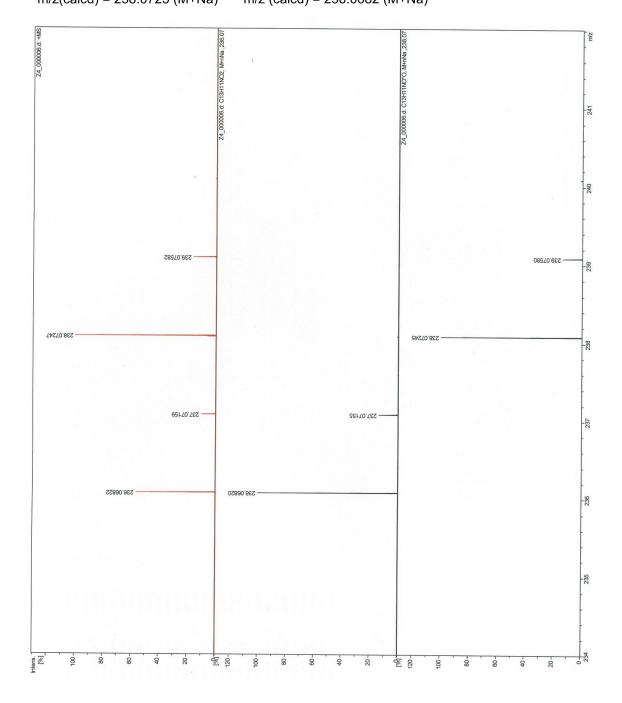


 ${
m 4h^{-18}O}$ obtained by the Co-catalyzed reaction of ${
m 1h^{-18}O}$ followed by deprotection.

4h-¹⁸O

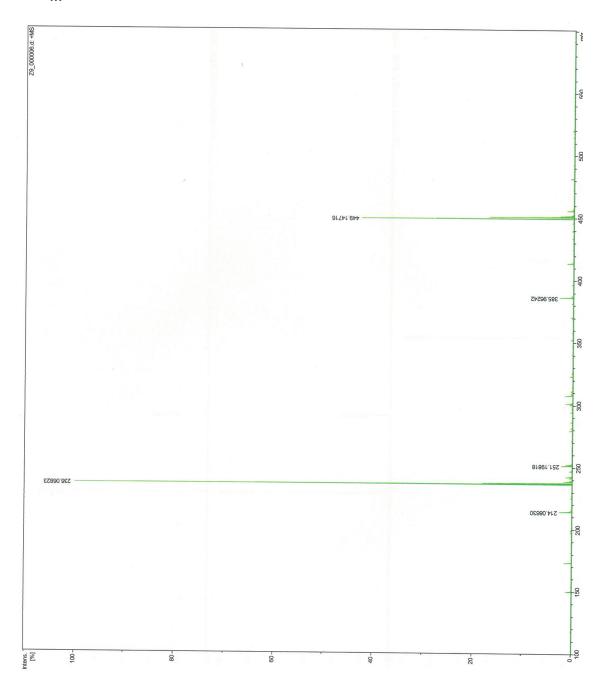


Ph O Ph O NH OH NH OH MH MINING
$$M/Z(calcd) = 238.0725 (M+Na)^{+}$$
 m/z (calcd) = 236.0682 (M+Na)⁺

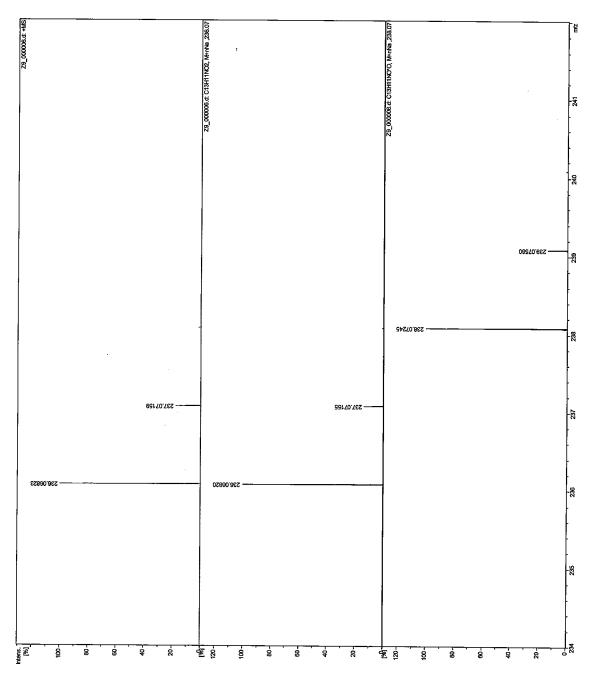


4h obtained by the thermal reaction of **1h**-¹⁸O followed by deprotection

4h



m/z(calcd) = 238.0725 (M+Na)⁺ m/z (calcd) = 236.0682 (M+Na)⁺



6. Analytical data of 1

1e

N-phenyl-*N*-(2,2,2-trichloroethylcarboxyl)-*O*-(methoxycarbonyl)hydroxylamine (1e). 1 H NMR (400 MHz, CDCl₃) δ 4.82 (s, 2H), 5.28 (s, 2H), 7.32 (t, J = 7.3 Hz, 1H), 7.37 (s, 5H), 7.41 (t, J = 7.3 Hz, 2H), 7.51 (d, J = 7.8 Hz, 2H). 13 C NMR (100 MHz, CDCl₃) δ 71.84, 75.54, 94.42, 124.29, 128.18, 128.55, 128.71, 129.06, 133.82, 138.42, 151.52, 153.79. IR (neat) 3067, 3035, 2958, 2893, 1794, 1739, 1596, 1493, 1456, 1385, 1341, 1308, 1216, 1159, 1118, 1051, 1030, 954, 901, 827, 795, 772, 752, 735, 717, 694 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 439.9830, found. 439.9830.

1j

N-Phenyl-*N***-(4-toluenesulfonyl)-***O***-(methoxycarbonyl)hydroxylamine (1j).** ¹H NMR (400 MHz, CDCl₃) δ 2.41 (s, 3H), 5.25 (s, 2H), 7.13 (m, 2H), 7.21 (d, J = 8.2 Hz, 2H), 7.25-7.29 (m, 3H), 7.36 (s, 5H), 7.45 (d, J = 8.2 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 21.69, 71.48, 124.00, 128.51, 128.59, 128.65, 128.89, 129.33, 129.43, 129.57, 129.74, 134.05, 139.85, 145.53, 153.03. IR (neat) 3067, 3035, 2959, 1596, 1487, 1455, 1089, 1029, 948, 910, 814, 795, 772, 757, 708, 658 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 420.0876, found. 420.0876.

1

N-Benzoyl-*N*-phenyl-*O*-(2-chloroethoxycarbonyl)hydroxylamine (11). 1 H NMR (400 MHz, CDCl₃) δ 3.70 (t, J = 5.9 Hz, 2H), 4.48 (t, J = 5.9 Hz, 2H), 7.25-7.34 (m, 7H), 7.38 (t, J = 7.5 Hz, 1H), 7.53 (d, J = 7.3 Hz, 2H). 13 C NMR (100 MHz, CDCl₃) δ 40.65, 68.81, 127.06, 128.17, 128.84, 128.88, 129.28, 131.38, 132.59, 139.97, 153.50, 166.94. IR (neat) 3063, 3027, 2963, 1788, 1678, 1593, 1489, 1448, 1385, 1336, 1306, 1219, 1141, 1075, 993, 915, 846, 768, 694, 661, 611 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 342.0503, found. 342.0503.

1m

N-Benzoyl-*N*-phenyl-*O*-(*tert*-butoxycarbonyl)hydroxylamine (1m). ¹H NMR (400 MHz, CDCl₃) δ 1.48 (s, 9H), 7.28-7.41 (m, 8H), 7.57 (d, J = 7.3 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 27.48, 85.89, 125.99, 128.05, 128.14, 128.67, 129.09, 131.03, 133.29, 140.14, 151.62, 167.30. IR (neat) 3453, 3058, 2981, 2932, 1785, 1681, 1592, 1578, 1489, 1452, 1369, 1342, 1296, 1269, 1245, 1141, 1050, 1015, 929, 850, 795, 700 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 336.1206, found. 336.1206.

N-Benzoyl-*N*-(4-acetylphenyl)-*O*-(benzyloxycarbonyl)hydroxylamine (1w). H NMR (400 MHz, CDCl₃) δ 2.58 (s, 3H), 5.24 (s, 2H), 7.28-7.38 (m, 7H), 7.42-7.46 (m, 3H), 7.55-7.57 (m, 2H), 7.92 (d, J = 9.2 Hz, 2H). C NMR (100 MHz, CDCl₃) δ 26.59, 71.86, 123.92, 128.32, 128.36, 128.56, 128.69, 129.00, 129.25, 131.67, 132.62, 133.71, 135.53, 143.70, 153.64, 167.23, 196.74.IR (neat) 3350, 3065, 3031, 3011, 2965, 2921, 1790, 1682, 1599, 1504, 1448, 1416, 1358, 1326, 1297, 1266, 1218, 1147, 1054, 958, 899, 836, 768, 697, 658 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 412.1155, found. 412.1155.

N-Benzoyl-*N*-(4-cyanophenyl)-*O*-(benzyloxycarbonyl)hydroxylamine (1x). H NMR (400 MHz, CDCl₃) δ 5.21 (s, 2H), 7.24-7.26 (m, 2H), 7.32-7.40 (m, 5H), 7.45-7.57 (m, 5H), 7.64 (d, J = 8.7 Hz, 2H). The sum of the su

4-(N-(((Benzyloxy)carbonyl)oxy)benzamido)phenethyl benzoate (1z). ¹H NMR (400 MHz, CDCl₃) δ 3.05 (t, J = 6.9 Hz, 2H), 4.50 (t, J = 6.9 Hz, 2H), 5.25 (s, 2H), 7.59-7.21 (m, 17H), 7.97 (d, J = 7.3 Hz, 2H). ¹³C NMR (100 MHz, CDCl₃) δ 34.79, 64.88, 71.51, 127.01, 128.12, 128.34, 128.40, 128.68, 128.77, 128.86, 129.53, 129.62, 130.05, 131.24, 132.89, 133.04, 134.08, 138.64, 138.85, 153.86, 166.41, 167.16. IR (neat) 3034, 2957, 2896, 2359, 2341, 1788, 1716, 1682, 1602, 1509, 1450, 1379, 1336, 1314, 1178, 1115, 1070, 1026, 968, 901, 843, 789, 739, 713, 698, 615 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 518.1574, found. 518.1574.

N-Benzoyl-*N*-[4-{2-(*tert*-butyldimethylsilyloxy)ethyl}phenyl]-*O*-(benzyloxycarbonyl)hy droxylamine (1aa). ¹H NMR (400 MHz, CDCl₃) δ -0.08 (s, 6H), 0.83 (s, 9H), 2.77 (t, 2H), 3.76 (t, 2H), 5.26 (s, 2H), 7.15 (d, 2H), 7.24-7.27 (m, 6H), 7.35-7.38 (m, 6H), 7.53 (d, 2H). ¹³C NMR (100 MHz, CDCl₃) -5.49, 18.22, 25.83, 39.00, 63.81, 71.36, 126.88, 128.09, 128.28, 128.62, 128.78, 130.00, 131.12, 132.99, 134.12, 138.20, 140.49, 153.81, 167.12. δ IR (neat) 3416, 3370, 3064, 3033, 2952, 2928, 2886, 2856, 1788, 1682, 1602, 1579, 1508, 1471, 1462, 1448, 1379, 1334, 1253, 1215, 1179, 1094, 1040, 1004, 966, 916, 901, 832, 774, 736, 695 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 528.2177, found. 528.2177.

N-Benzoyl-*N*-[4-{2-(methoxymethoxy)ethyl}phenyl]-*O*-(benzyloxycarbonyl)hydroxyla mine (1ab). (a mixture with 2ab (8.3 : 1) 1 H NMR (400 MHz, CDCl₃) δ 2.87 (t, J = 6.9 Hz, 2H), 3.22 (s, 3H), 3.72 (t, J = 6.9 Hz, 2H), 4.57 (s, 2H), 5.25 (s, 2H), 7.18-7.20 (m, 2H), 7.23-7.28 (m, 4H), 7.32-7.38 (m, 6H), 7.52-7.55 (m, 2H). 13 C NMR (100 MHz, CDCl₃) δ 35.83, 55.17, 67.80, 71.44, 96.30, 127.02, 128.10, 128.33, 128.56, 128.66, 128.78, 128.84, 129.72, 131.17, 133.00, 134.10, 138.29, 140.11, 153.85. IR (neat) 3062, 3034, 2931, 2883, 2823, 1787, 1681, 1509, 1448, 1379, 1335, 1218, 1147, 1108, 1068, 1030, 965, 915, 845, 795, 770, 740, 698 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 458.1574, found. 458.1574.

7. Analytical data of 2

2-[(Benzoyl)amino]phenyl 2-chloroethyl carbonate (2l). ¹H NMR (400 MHz, CDCl₃) δ 3.79 (t, J = 5.5 Hz, 2H), 4.55 (t, J = 5.5 Hz, 2H), 7.18 (t, J = 7.8 Hz, 1H), 7.31-7.34 (m, 2H), 7.51 (t, J = 7.3 Hz, 2H), 7.58 (t, J = 7.3 Hz, 1H), 7.89 (d, J = 7.3 Hz, 2H), 8.12 (br s, 1H), 8.42 (d, J = 8.2 Hz, 1H). ¹³C NMR (100 MHz, CDCl₃) δ 41.17, 68.25, 121.40, 122.32, 124.60, 126.90, 127.08, 128.84, 129.73, 132.10, 134.37, 140.62, 152.48, 165.31. IR (neat) 3317, 3065, 2959, 1767, 1667, 1608, 1580, 1523, 1496, 1454, 1386, 1309, 1235, 1190, 1125, 1082, 1033, 998, 766, 740, 709 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 342.0503, found. 342.0503.

4-Benzamido-3-(((benzyloxy)carbonyl)oxy)phenethyl benzoate (2z). Included with inseparable byproducts 1 H NMR (400 MHz, CDCl₃) δ 3.07 (t, J = 6.9 Hz, 2H), 4.52 (t, J = 6.9 Hz, 2H), 5.28 (s, 2H), 7.20-8.03 (m, 17H), 8.32 (d, J = 8.7 Hz, 1H). Reaction mixture was not purified by flash chromatography (Silica gel, hexanes:EtOAc = 3:1). Isolation of **2z** was unsuccessful due to containination by inseparable byproducts; further purification by using gel permeation chromatography resulted in decomposition of the product.

2-[(Benzoyl)amino]-4-[2-(*tert*-butyldimethylsiloxy)ethyl]phenyl **2-benzyl** carbonate **(2aa).** ¹H NMR (400 MHz, CDCl₃) -0.01 (s, 6H), 0.90 (s, 9H), 2.83 (t, 2H), 3.82 (t, 2H), 5.29 (s, 2H), 7.15 (d, 2H), 7.35-7.37 (m, 3H), 7.42-7.45 (m, 4H), 7.78 (d, 2H), 8.04 (s, 1H), 8.26 (d, 1H). ¹³C NMR (100 MHz, CDCl₃) -5.45, 18.28, 25.89, 38.86, 64.02, 70.82, 122.02, 122.34, 127.00, 127.45, 127.79, 128.51, 128.73, 128.76, 128.94, 131.86, 134.43, 134.56, 136.43, 140.89, 152.84, 165.19. IR (neat) 3444, 3314, 3065, 3033, 2952, 2928, 2895, 2856, 2737,

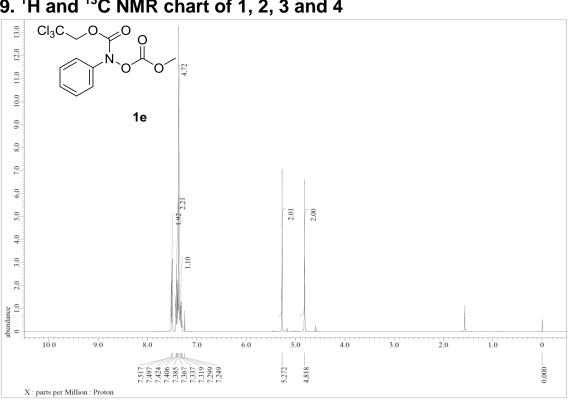
1766, 1679, 1601, 1520, 1481, 1421, 1379, 1312, 1208, 1130, 1093, 1045, 1028, 1005, 930, 908, 831, 774, 695 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 528.2177, found. 528.2177.

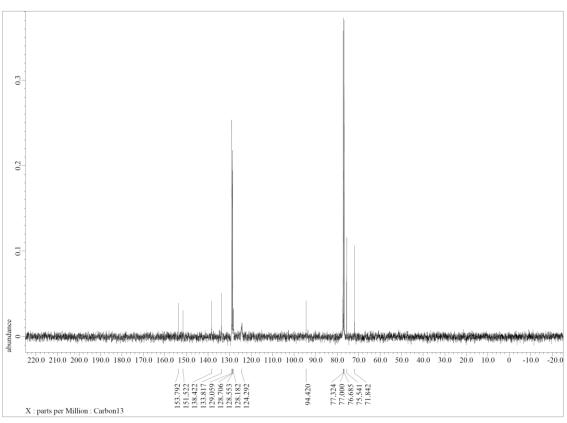
2-[(Benzoyl)amino]-4-[2-(methoxymethoxy)ethyl]phenyl 2-benzyl carbonate (2ab). 1 H NMR (400 MHz, CDCl₃) δ 2.90 (t, J = 6.9 Hz, 2H), 3.29 (s, 3H), 3.75 (t, J = 6.9 Hz, 2H), 4.63 (s, 2H), 5.28 (s, 2H), 7.17 (d, J = 9.6 Hz, 2H), 7.35-7.45 (m, 8H), 7.54 (t, J = 7.3 Hz, 1H), 7.77 (d, J = 7.3 Hz, 2H), 8.02 (br s, 1H), 8.26 (d, J = 8.2 Hz, 1H). 13 C NMR (100 MHz, CDCl₃) δ 35.59, 55.17, 67.94, 70.82, 96.31, 121.79, 122.40, 126.98, 127.15, 128.87, 128.51, 128.71, 128.75, 128.93, 131.34, 134.47, 136.03, 140.89, 152.80. IR (neat) 3323, 3062, 3033, 2934, 2884, 2824, 1766, 1675, 1602, 1581, 1521, 1483, 1421, 1380, 1311, 1234, 1212, 1147, 1108, 1070, 1029, 914, 795, 746, 697 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 458.1574, found. 458.1574.

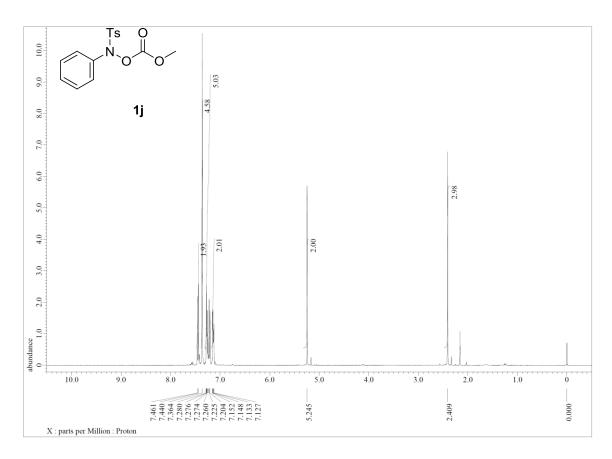
8. Analytical data of 3

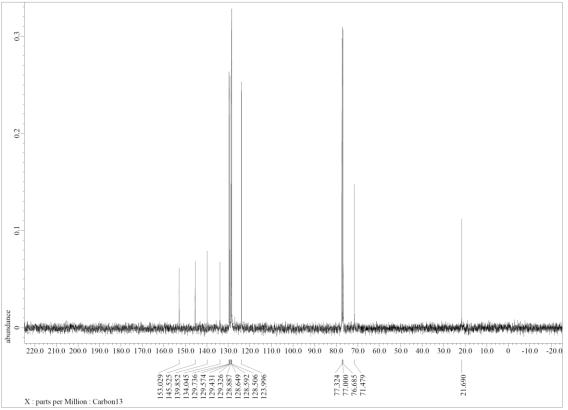
4-[(Methoxycarbonyl)amino]phenyl methyl carbonate (3a) 1 H NMR (400 MHz, CDCl₃) 5 3.77 (s, 3H), 3.90 (s, 3H), 6.79 (br s, 1H), 7.09-7.12 (m, 2H), 7.39 (d, J = 8.2 Hz, 2H). 13 C NMR (100 MHz, CDCl₃) 5 52.40, 55.41, 119.38, 121.53, 135.73, 146.59, 153.94, 154.43. IR (neat) 3628, 3540, 3367, 3348, 3134, 3011, 2956, 2851, 1764, 1734, 1713, 1610, 1540, 1512, 1440, 1411, 1265, 1209, 1113, 1071, 1016, 948, 930, 839, 806, 771 cm⁻¹. HRMS (ESI) calcd. for (M+Na)⁺ 248.0529, found. 248.0529.

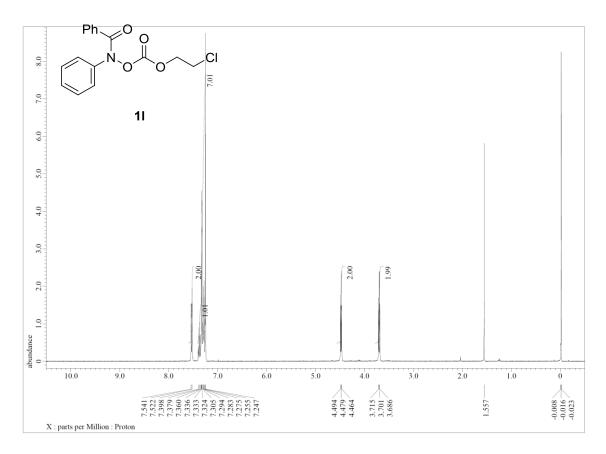
9. ^{1}H and ^{13}C NMR chart of 1, 2, 3 and 4

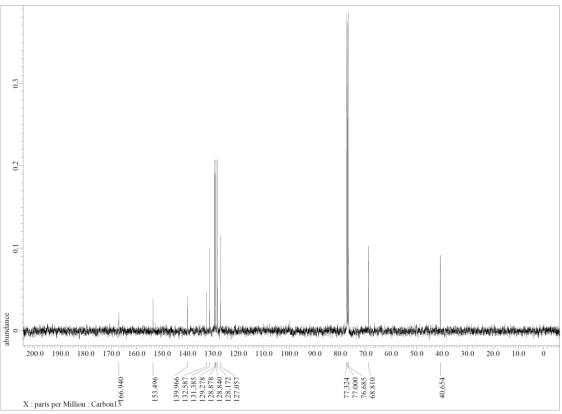


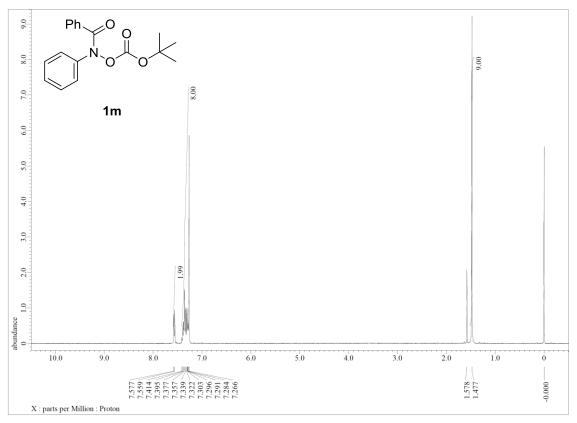


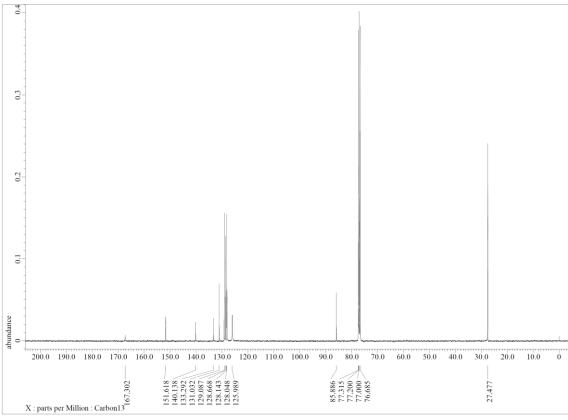


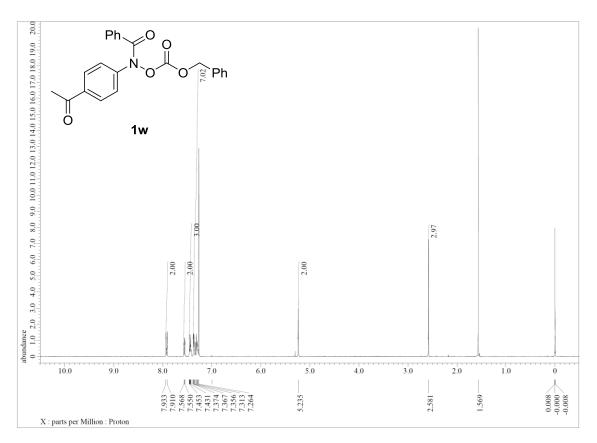


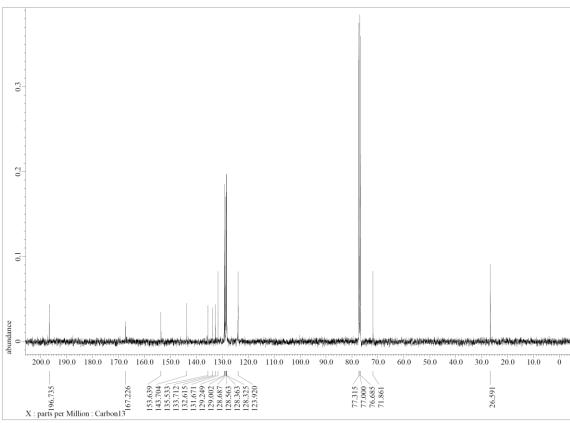


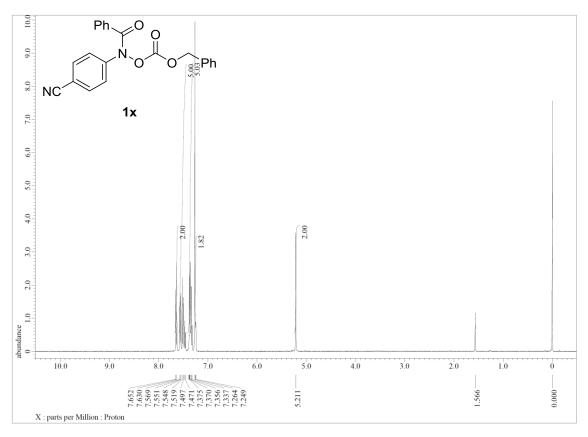


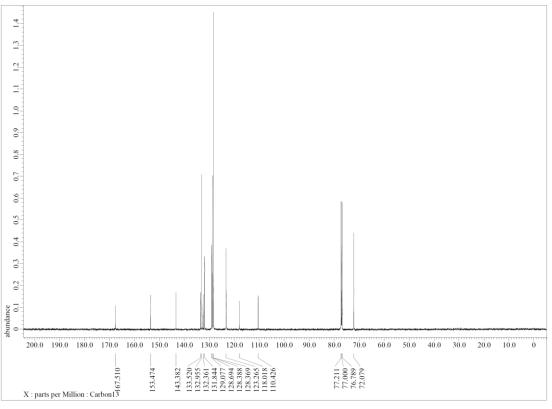


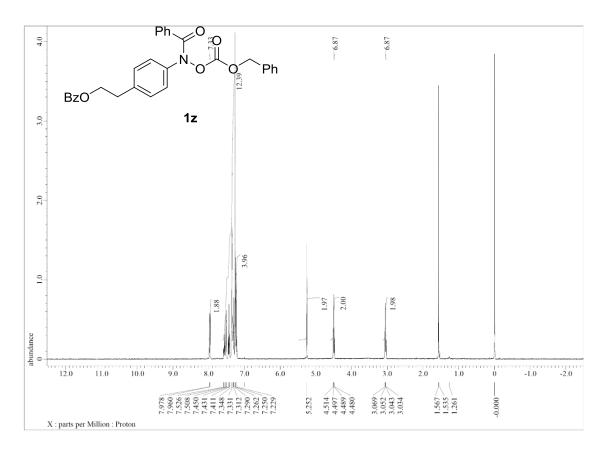


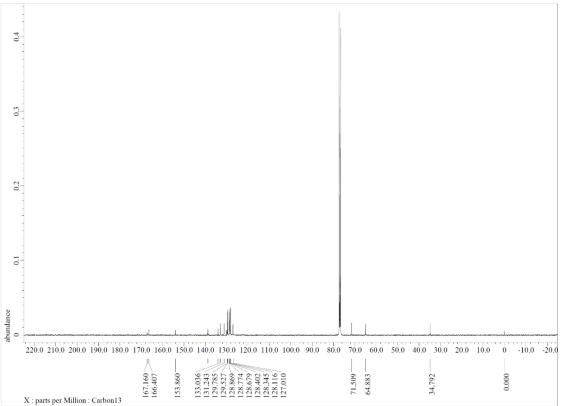


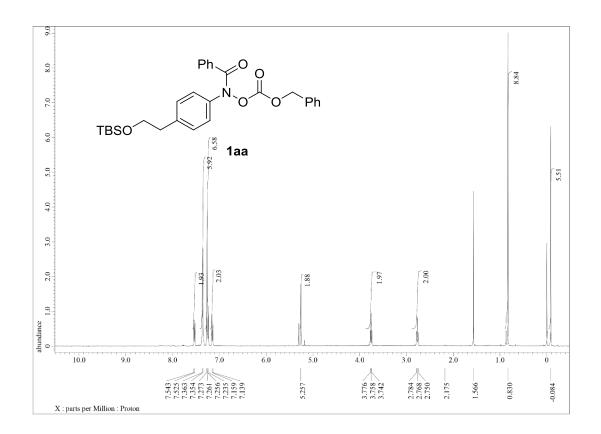


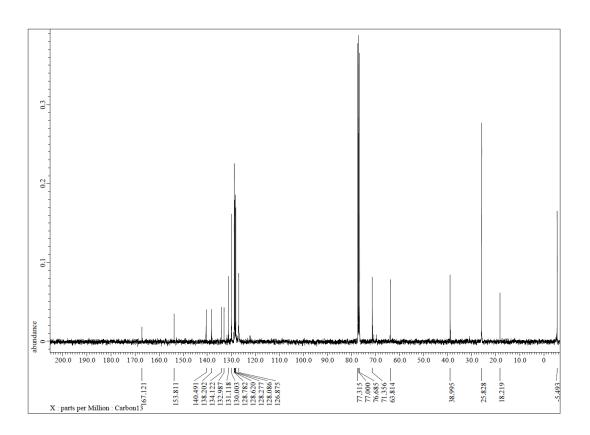


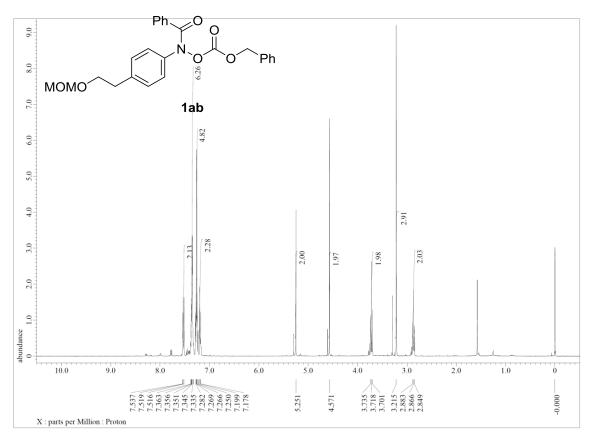


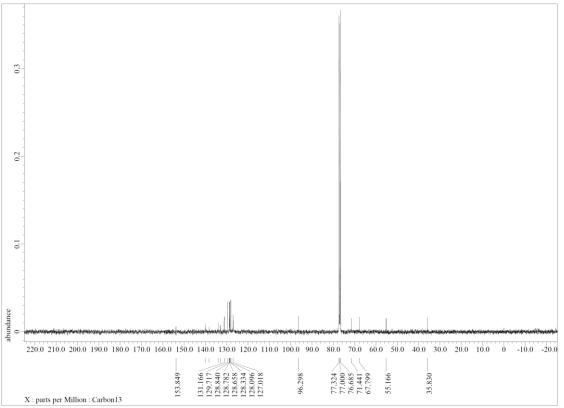


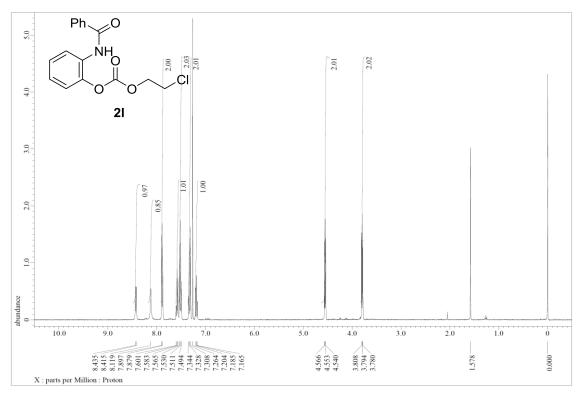


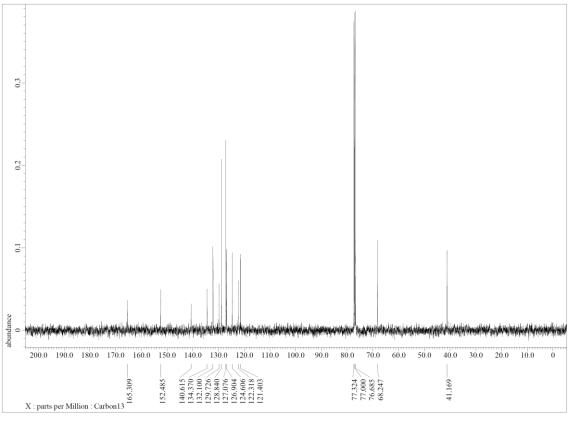


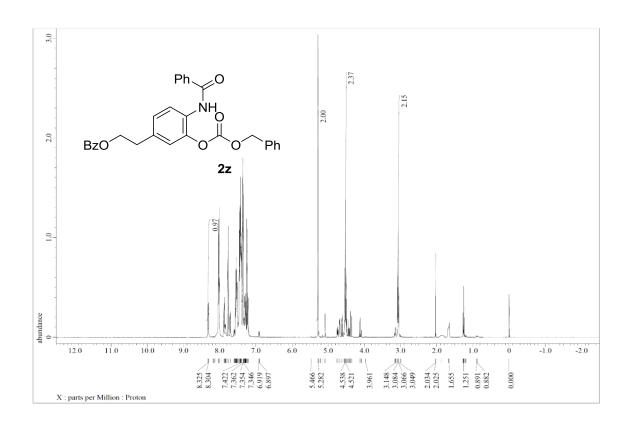


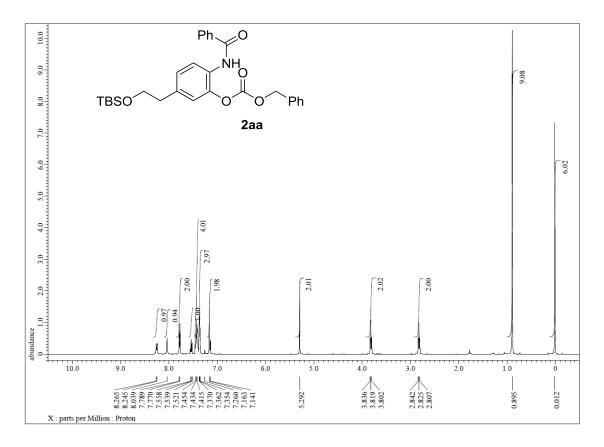


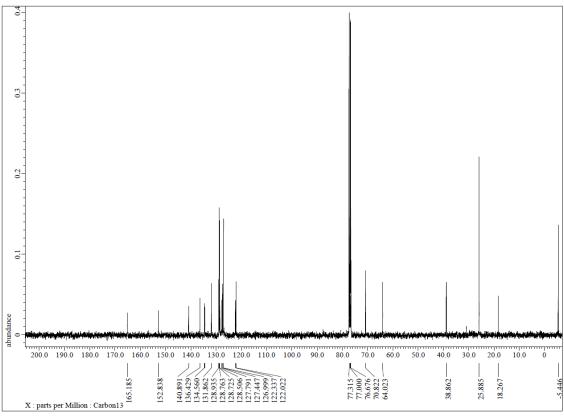


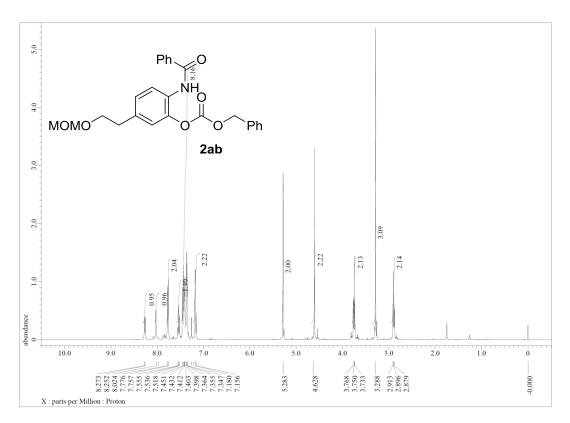


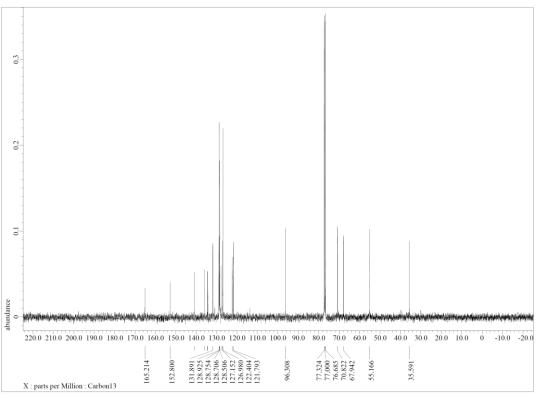


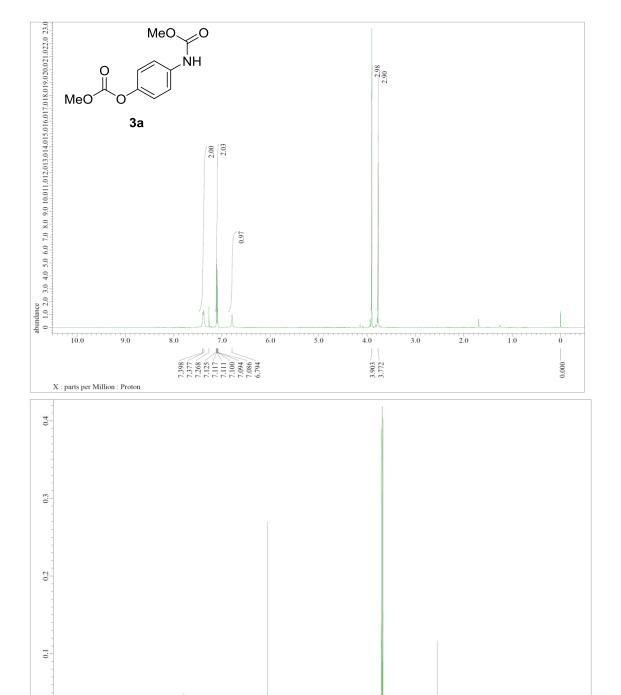












10. References

X : parts per Million : Carbon13

200.0 190.0 180.0 170.0 160.0 150.0 140.0 130.0

154.430 153.944 146.593

[S1] Nakamura, I.; Owada, M.; Jo, T.; Terada, M. Org. Lett. 2017, 19, 2194.

121.526

[S2] Rajendran, G.; Van Etten, R. L. Inorg. Chem. 1986, 25, 876.

135.733

[S3] Rajendran, G.; Santini, R. E.; Van Etten, R. L. J. Am. Chem. Soc. 1987, 109, 4357.

120.0 110.0 100.0 90.0

80.0

77.315

60.0 50.0

55.414

20.0