



## Supporting Information

for

### Extending mechanochemical porphyrin synthesis to bulkier aromatics: tetramesitylporphyrin

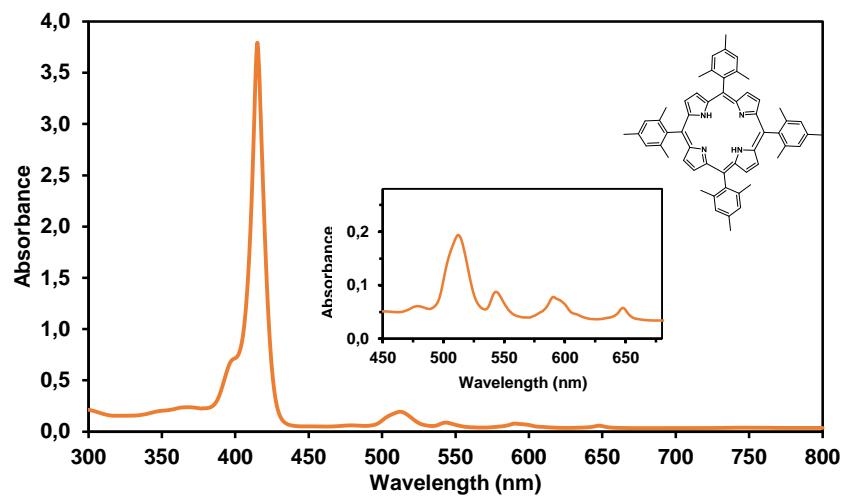
Qiwen Su and Tamara D. Hamilton

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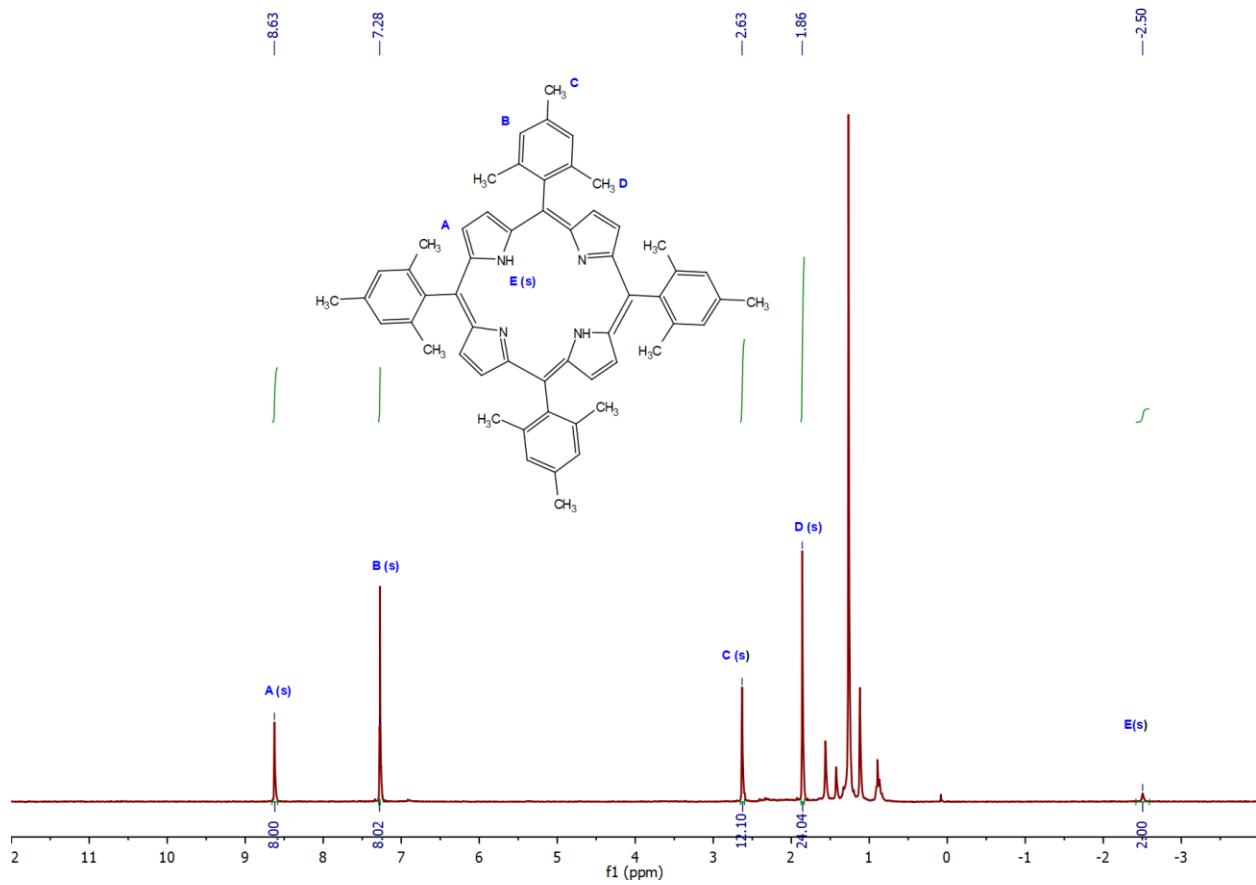
### Spectroscopic characterization of *meso*-tetrakis[2,4,6-(trimethyl)phenyl]porphyrin (TMP)

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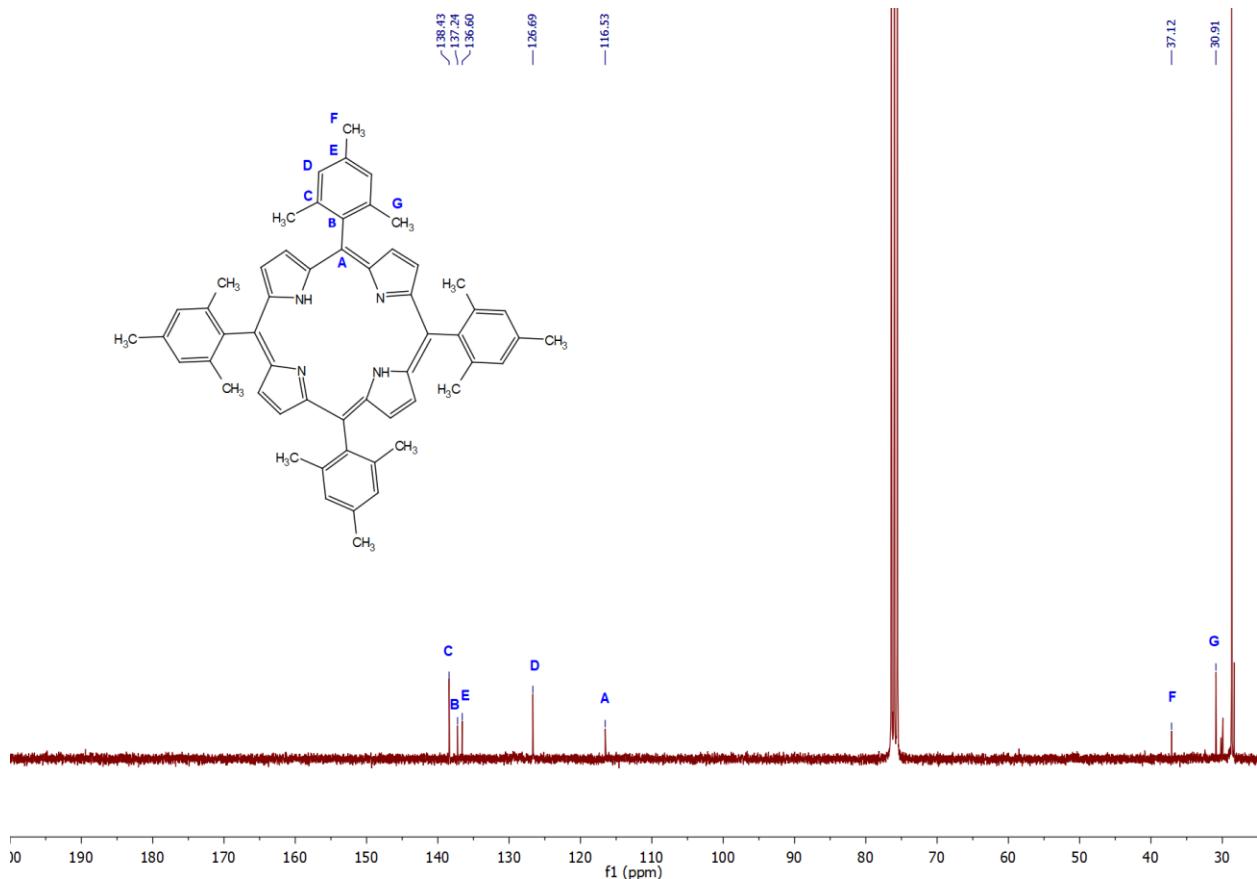
A.	UV-vis spectrum of TMP.....	S2
B.	$^1\text{H}$ NMR spectrum of TMP.....	S3
C.	$^{13}\text{C}$ NMR spectrum of TMP.....	S4



**Figure S1:** UV–vis spectrum of *meso*-tetrakis[2,4,6-(trimethyl)phenyl]porphyrin.  $\lambda = 414$  (Soret band);  $\lambda = 513, 543, 590, 648$  (Q-bands).



**Figure S2:**  $^1\text{H}$  NMR spectrum of *meso*-tetrakis[2,4,6-(trimethyl)phenyl]porphyrin.  
 (300 MHz,  $\text{CDCl}_3$ ):  $\delta$  = 8.63 (s, 8H), 7.28 (s, 8H), 2.63 (s, 12H), 1.86 (s, 24H), -2.50 (s, 2H).



**Figure S3:** <sup>13</sup>C NMR spectrum of *meso*-tetrakis[2,4,6-(trimethyl)phenyl]porphyrin (300 MHz, CDCl<sub>3</sub>): δ = 138.4, 137.2, 136.6, 126.7, 116.5, 37.1, 30.9.

Note: In <sup>13</sup>C NMR spectra of free base porphyrins, the α and β carbons on the pyrrole rings are often indistinguishable from baseline noise or too broad to be observable.<sup>1</sup>

### Reference:

- 1) Mohajer, D.; Zakavi, S.; Rayati, S.; Zahedi, M.; Safari, N.; Khavasi, H. R.; Shahbazian, S. *New J. Chem.* **2004**, 28, 1600-1607.