## Supporting Information

for

# Chiral terpene auxiliaries V: Synthesis of new chiral $\gamma$-hydroxyphosphine oxides derived from $\alpha$-pinene 

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# General information, experimental procedures and characterization data of the following compounds: 2, 3, 5-8, 10, 11, and 14-18 

## 1. General

## Analytical methods

Experiments with air- and moisture-sensitive materials were carried out under an argon atmosphere using Schlenk techniques. Glassware was oven dried at $120{ }^{\circ} \mathrm{C}$, assembled hot, and cooled in a steam of argon. ${ }^{1} \mathrm{H},{ }^{13} \mathrm{C}$ and ${ }^{31} \mathrm{P}$ NMR spectra were recorded on a Bruker AMX 300 MHz , Avance III 400 MHz , and Avance 700 MHz instruments. Optical rotations were measured on an Optical Activity PoIAAr 3000 automatic polarimeter. GC analyses were performed on a Perkin-Elmer Auto System XL chromatograph and HPLC analyses on a Shimadzu LC-10AT chromatograph. Melting points were determined in open glass capillaries and are uncorrected. Elemental analyses were performed on Vario MACRO CHN, ELEMENTAR Analysensysteme GmbH instrument.

## Materials

Silica Gel 60, Merck, 230-400 mesh, was used for preparative column chromatography. Macherey-Nagel Polygram Sil G/UV254 0.2 mm plates were used for analytical TLC. Tetrahydrofuran was freshly distilled from sodium benzophenone ketyl. Lead(IV) acetate, diphenyl diselenide, $\alpha$-pinene, $\beta$-pinene, platinum(IV) oxide, cerium(III) chloride heptahydrate, borane-dimethyl sulfide adduct, metachloroperoxybenzoic acid, sodium methoxide, diisopropyl azadicarboxylate were purchased from commercial sources. (+)-Nopinone was prepared according to the literature [1].

## 2. Synthetic procedures and analytical data

## 2.1. (1R,5R)-(+)-Verbenone (2)


$(+)-\alpha$-Pinene $\left([\alpha]_{D}^{24}+47.7\left(c 2.0, \mathrm{CHCl}_{3}\right), 93 \% \mathrm{ee}\right)(13.62 \mathrm{~g}, 0.1 \mathrm{~mol})$ and toluene (200 mL ) were placed under nitrogen atmosphere in a 500 mL three necked flask equipped with reflux condenser, magnetic stirrer, and thermometer. The solution was heated in an oil bath to $65^{\circ} \mathrm{C}$. Lead(IV) acetate ( $47.44 \mathrm{~g}, 0.107 \mathrm{~mol}$ ) was added over 25 min and the stirring was continued for 75 min . The solution was cooled to room temperature,
filtered through a pad of celite and the precipitate was washed with toluene ( $4 \times 25$ $\mathrm{mL})$. Water ( 150 mL ) was added to the filtrate and the mixture was stirred for 1 h . The mixture was filtered through a pad of celite to remove a brown precipitate. The filtrate was transferred to a separatory funnel, layers were separated and the aqueous layer was extracted with diethyl ether ( $3 \times 50 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate. After filtration, the solvent was removed on rotary evaporator under reduced pressure and then a water-methanol solution of $\mathrm{KOH}(10 \%$, 75 mL ( 15 mL of water and 60 mL of methanol)) was added to the residue. The solution was stirred for 20 hours, water ( 100 mL ) was added and the mixture was extracted with diethyl ether ( $4 \times 75 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate. After filtration, the solvent was removed on rotary evaporator under reduced pressure. The resulting yellow-orange oil and ether ( 150 mL ) were transferred into a two-necked flask equipped with a thermometer, dropping funnel, and a magnetic stirring bar. The solution was cooled to $0{ }^{\circ} \mathrm{C}$ and sodium dichromate dihydrate ( $13.80 \mathrm{~g}, 46.3 \mathrm{mmol}$ ) dissolved in water ( 53 mL ) with concentrated sulfuric acid ( 6 mL ) was added dropwise over 30 minutes. The brown solution was stirred for 1 h at $0{ }^{\circ} \mathrm{C}$, followed by warming to an ambient temperature stirring was continued for 20 h . Water ( 100 mL ) was added and the mixture was transferred to a separatory funnel. Layers were separated, the aqueous layer was extracted with diethyl ether (3 $\times 100 \mathrm{~mL})$. The combined organic layers were washed with brine ( 100 mL ) and dried over anhydrous magnesium sulfate. After filtration, the solvent was removed on a rotary evaporator. The product was distilled under reduced pressure to yield (+)verbenone ( $8.71 \mathrm{~g}, 58 \%$ ) as a colorless liquid, bp $75-78{ }^{\circ} \mathrm{C} / 5.5 \mathrm{mmHg},[\alpha]_{D}^{22}=$ +231.5 (c 2.0, $\mathrm{CHCl}_{3}$ ), 93\% ee.
${ }^{1} \mathrm{H}$ NMR $\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=1.01\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.50\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.04(\mathrm{~d}, J=1.5$ Hz, 3H, CH3), 2.08 (d, $J=9.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.40-2.43 (m, 1H), 2.63-2.66 (td, $J=5.9,1.7$ $\mathrm{Hz}, 1 \mathrm{H}$ ), 2.78-2.83 (dt, $J=9.2,5.6 \mathrm{~Hz}, 1 \mathrm{H}), 5.73(\mathrm{sxt}, J=1.6 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (176 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=21.99,23.50,26.54,40.78,49.68,53.97,57.56,121.10,170.17$, 203.91.

## 2.2. ( $1 R, 2 S, 5 R$ )-(+)-Verbanone (3)



The autoclave was charged with (+)-verbenone ( $7.51 \mathrm{~g}, 50 \mathrm{mmol}$ ), cyclohexane ( 9 mL ) and platinum(IV) oxide ( 0.020 g ). The autoclave was closed and the air was removed by charging and releasing the hydrogen three times. Then, the autoclave was filled with hydrogen to a pressure of about 10 atm. The mixture was stirred overnight at rt. The reaction mixture was filtered through a pad of celite and cyclohexane was evaporated on a rotary evaporator. The crude product was distilled under reduced pressure to yield (+)-verbanone ( $7.08 \mathrm{~g}, 93 \%$ ) as a colorless liquid, bp $56-58^{\circ} \mathrm{C} / 0.5$ $\mathrm{mmHg},[\alpha]_{D}^{22}=+59.2\left(c 2.0, \mathrm{CHCl}_{3}\right), 93 \%$ ee.
${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=1.01\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.17\left(\mathrm{~d}, J=7.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.34$ (s, 3H, CH3), $1.40(\mathrm{~d}, J=10.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.13-2.14(\mathrm{~m}, 1 \mathrm{H}), 2.16$ (dd, $J=20.0,4.8 \mathrm{~Hz}$, 1 H ), 2.33-2.24 (m, 1H), $2.55(\mathrm{t}, \mathrm{J}=5.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.57-2.63(\mathrm{~m}, 1 \mathrm{H}), 2.86(\mathrm{dd}, J=20.0$, $10.6 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=20.9,24.4,26.8,28.3,30.9,40.1,41.3$, 47.27, 57.84, 214.2. Anal. calcd for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}: \mathrm{C}, 78.90 ; \mathrm{H}, 10.59$; found: $\mathrm{C}, 78.34 ; \mathrm{H}$, 10.37.

## 2.3. (1R,2R,5R)-(+)-3-Methyleneverbanone (5)


(+)-Verbanone (3) ( $6.09 \mathrm{~g}, 40 \mathrm{mmol}$ ) was added dropwise to a suspension of sodium methoxide ( $4.32 \mathrm{~g}, 80 \mathrm{mmol}$ ) and ethyl formate ( $5.93 \mathrm{~g}, 80 \mathrm{mmol}$ ) in toluene ( 200 mL ), and the mixture was stirred at room temperature for 24 h . Hydrochloric acid solution ( $2 \mathrm{M}, 50 \mathrm{~mL}$ ) was added and the resulting mixture was extracted with diethyl ether (3 $\times 50 \mathrm{~mL})$. The combined organic layers were washed with brine $(50 \mathrm{~mL})$, dried over anhydrous magnesium sulfate, and the solvent was removed on rotary evaporator to give keto aldehyde 4 ( $5.23 \mathrm{~g}, 29 \mathrm{mmol}$ ).
Keto aldehyde (4) (5.23 g, 29 mmol ), sodium carbonate ( 15.9 g ), and formaldehyde ( $37 \%$ solution in water, 16 mL ) were vigorously stirred in diethyl ether $(125 \mathrm{~mL})$ at room temperature for 3 h . Water ( 150 mL ) was then added and the product was extracted with ethyl acetate $(2 \times 75 \mathrm{~mL})$. The combined organic layers were washed with brine
( 50 mL ), dried over anhydrous magnesium sulfate, and the solvent was removed on rotary evaporator. The crude product was purified by flash chromatography on silica gel (hexane/ethyl acetate $=90: 10)$. Enone $5(3.71 \mathrm{~g}, 78 \%)$ was obtained as a colorless oil, $[\alpha]_{D}^{23}+25.2\left(c 1,4 ; \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.95(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH} 3), 1.30-1.32(\mathrm{~d}, \mathrm{~J}=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.34$ (s, 3H, CH3), 1.38-1.40 (d, $J=10.3 \mathrm{~Hz}, 1 \mathrm{H}), 2.08-2.12(\mathrm{td}, J=5.9,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.57-$ $2.65(\mathrm{~m}, 2 \mathrm{H}), 2.85-2.92(\mathrm{~m}, 1 \mathrm{H}), 5.45-5.46(\mathrm{~m}, 1 \mathrm{H}), 6.41-6.42(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=20.42,23.68,27.08,29.24,37.81,40.30,46.36,56.06,122.05$, 147.01, 202.20. Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}: \mathrm{C}, 80.44$; H, 9.82; found: C, 80.11; H, 9.49.
2.4. (1R,2R,4S,5R)-3-Methyleneverbanol (6)


Enone 5 ( $3.28 \mathrm{~g}, 20 \mathrm{mmol}$ ) and cerium(III) chloride heptahydrate ( $7.45 \mathrm{~g}, 20 \mathrm{mmol}$ ) were dissolved in methanol ( 50 mL ). The solution was cooled in an ice-water bath and then sodium borohydride ( $1.14 \mathrm{~g}, 30 \mathrm{mmol}$ ) was added in small portions. The reaction mixture was stirred overnight. Water ( 100 mL ) and dichloromethane ( 50 mL ) were added to the flask and pH of the mixture was adjusted to $5-6$ with concentrated hydrochloric acid. Layers were separated and the aqueous layer was extracted with dichloromethane ( $2 \times 50 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate filtered, and the solvent was removed by rotary evaporation. The crude product was purified by flash chromatography on silica gel (hexane/ethyl acetate $=90: 10$ ). Allylic alcohol 6 ( $2.93 \mathrm{~g}, 88 \%$ ) was obtained as a colorless solid, $\mathrm{mp} 72-74$ ${ }^{\circ} \mathrm{C},[\alpha]_{D}^{25}+59.4\left(c 1.4 ; \mathrm{CHCl}_{3}\right)$.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.82(\mathrm{~d}, J=10.5 \mathrm{~Hz}, 1 \mathrm{H}), 1.05\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.23(\mathrm{~s}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.25\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.80(\mathrm{t}, J=5.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.06-2.10(\mathrm{~m}, 1 \mathrm{H}), 2.34-2.40(\mathrm{~m}$, $1 \mathrm{H}), 2.63-2.67(\mathrm{~m}, 1 \mathrm{H}), 4.50-4.51(\mathrm{~m}, 1 \mathrm{H}), 5.23(\mathrm{t}, J=2.4 \mathrm{~Hz}, 1 \mathrm{H}), 5.48(\mathrm{t}, J=2.6 \mathrm{~Hz}$, $1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=20.22,22.86,28.05,31.26,38.62,40.41,47.34$, 47.66, 75.52, 111.40, 154.96.

Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}: \mathrm{C}, 79.46 ; \mathrm{H}, 10.91$; found: $\mathrm{C}, 79.89 ; \mathrm{H}, 10.31$.

## 2.5. (-)-(1R,2R,3R,5S)-Isopinocampheol (7)


(+)- $\alpha$-Pinene ( $17.68 \mathrm{~g}, 130 \mathrm{mmol}$ ) was dissolved in THF ( 15 mL ) under nitrogen atmosphere. The solution was cooled to $0^{\circ} \mathrm{C}$ and 10 M borane-dimethyl sulfide adduct (BMS, $5 \mathrm{~mL}, 50 \mathrm{mmol}$ ) was added with stirring. The mixture was allowed to crystallize at $0^{\circ} \mathrm{C}$ for 20 h . The supernatant liquid was removed using a syringe with a long needle. The crystalline product was washed with dry hexane ( $3 \times 10 \mathrm{~mL}$ ) and dried under vacuum for 1 h to give pure $\mathrm{lpc}_{2} \mathrm{BH}(12.03 \mathrm{~g}, 84 \%)$.
$\mathrm{lpc}_{2} \mathrm{BH}$ was suspended in THF ( 40 mL ), then water ( 5 mL ) was carefully added followed by the addition of 3 M sodium hydroxide solution ( $16.8 \mathrm{~mL}, 50.4 \mathrm{mmol}$ ). After cooling the mixture to $10^{\circ} \mathrm{C}, 30 \%$ hydrogen peroxide solution ( $10 \mathrm{~mL}, 100 \mathrm{mmol}$ ) was added dropwise. After addition, the reaction mixture was stirred at room temperature for 30 min and at $50^{\circ} \mathrm{C}$ for 1 h . The solution was saturated with sodium chloride and extracted with diethyl ether ( $3 \times 50 \mathrm{~mL}$ ). The combined organic layers were washed with brine and dried over anhydrous magnesium sulfate. After filtration, the solvent was removed on rotary evaporator to give isopinocampheol 7 (10.11 g, 78\%) as a white solid, mp $54-55^{\circ} \mathrm{C},[\alpha]_{D}^{25}-34.6$ (c 2.0; EtOH).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.93\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.04(\mathrm{~d}, J=9.8 \mathrm{~Hz}, 1 \mathrm{H}), 1.14(\mathrm{~d}, J$ $=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), $1.23\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.55(\mathrm{br} \mathrm{s}, 1 \mathrm{H}), 1.71$ (ddd, $J=13.9,4.7,2.6 \mathrm{~Hz}$, $1 \mathrm{H}), 1.81(\mathrm{td}, J=5.9,1.9 \mathrm{~Hz}, 1 \mathrm{H}), 1.91-1.97(\mathrm{~m}, 2 \mathrm{H}), 2.35-2.40(\mathrm{~m}, 1 \mathrm{H}), 2.49-2.55$ $(\mathrm{m}, 1 \mathrm{H}), 4.07(\mathrm{dt}, J=9.5,4.9 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C} \operatorname{NMR}\left(176 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): \delta=20.73,23.68$, 27.67, 34.35, 38.15, 38.99, 41.76, 47.67, 47.82, 71.57. Anal. calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{O}: \mathrm{C}$, 77.87; H, 11.76; found: C, 77.26; H, 11.19.

## 2.6. (1R,2R,5S)-Isopinocamphone (8)


(-)-Isopinocampheol ( $9.26 \mathrm{~g}, 60 \mathrm{mmol}$ ) was dissolved in diethyl ether ( 30 mL ). The solution of sodium dichromate ( $5.50 \mathrm{~g}, 21 \mathrm{mmol}$ ) in water ( 30 mL ) and concentrated $\mathrm{H}_{2} \mathrm{SO}_{4}(4.7 \mathrm{~mL})$ was added with vigorous stirring keeping the temperature below 30 ${ }^{\circ} \mathrm{C}$. The solution was stirred for 2 h , then extracted with diethyl ether ( $3 \times 20 \mathrm{~mL}$ ). The
combined organic layers were washed with water ( 15 mL ), $5 \%$ sodium bicarbonate solution ( 15 mL ), brine ( 20 mL ), and dried over anhydrous magnesium sulfate. After filtration, the solvent was removed on rotary evaporator. The crude product was distilled under reduced pressure to give $8(7.12 \mathrm{~g}, 78 \%)$ as a colorless liquid, bp 64$66^{\circ} \mathrm{C} / 4.6 \mathrm{~mm} \mathrm{Hg},[\alpha]_{D}^{22}-19.2$ (c 1.0, EtOH).
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.87\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.19-1.23(\mathrm{~m}, 4 \mathrm{H}), 1.31(\mathrm{~s}, 3 \mathrm{H}$, $\mathrm{CH}_{3}$ ), 2.03-2.08 (m, 1H), 2.09-2.15 (m, 1H), $2.47(\mathrm{~m}, 1 \mathrm{H}), 2.53(\mathrm{dd}, J=2.3,1.1 \mathrm{~Hz}$, 1H), 2.60-2.64 (m, 2H). ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=16.68,21.77,26.90,34.25$, 38.84, 39.05, 44.59, 44.87, 51.14, 214.69. Anal. calcd for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{O}: \mathrm{C}, 78.90 ; \mathrm{H}, 10.59$; found: C, 78.37; H, 10.71.

## 2.7. (1R,2R,5R)-4-Methyleneisopinocamphone (10)



The synthesis of enone 10 was carried out according to procedure described for compound 5. From 8 ( $6.09 \mathrm{~g}, 40 \mathrm{mmol}$ ), sodium methoxide ( $4.32 \mathrm{~g}, 80 \mathrm{mmol}$ ), ethyl formate ( $5.93 \mathrm{~g}, 80 \mathrm{mmol}$ ), and formaldehyde ( $37 \%$ solution in water, 16 mL ), enone 10 ( $5.45 \mathrm{~g}, 83 \%$ ) was obtained as a yellow oil.
${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.81\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.13\left(\mathrm{~d}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.30$ (d, $J=10.8 \mathrm{~Hz}, 1 \mathrm{H}$ ), $1.35\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.98(\mathrm{td}, J=6.2,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.53-2.58(\mathrm{~m}$, 1 H ), 2.66 (qt, $J=7.3,2.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), $2.73(\mathrm{~m}, 1 \mathrm{H}), 4.97(\mathrm{~d}, J=1.7 \mathrm{~Hz}, 1 \mathrm{H}), 5.94(\mathrm{~d}, J=$ $1.7 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (176 MHz, $\left.\mathrm{CDCl}_{3}\right): \delta=15.26,21.10,26.15,29.11,41.08,44.38$, 45.00, 48.59, 117.22, 148.96, 203.40.

Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{16} \mathrm{O}: \mathrm{C}, 80.44 ; \mathrm{H}, 9.82$; found: C, $79.97 ; \mathrm{H}, 9.61$.

## 2.8. (1R,2R,3R,5R)-4-Methyleneneoisopinocampheol (11)



The synthesis of alcohol 11 was carried out according to procedure described for compound 6. Enone 10 ( $1.643 \mathrm{~g}, 10 \mathrm{mmol}$ ) was reduced with sodium borohydride
( $0.567 \mathrm{~g}, 15 \mathrm{mmol}$ ) in the presence of cerium chloride heptahydrate ( $3.726 \mathrm{~g}, 10 \mathrm{mmol}$ ). Allylic alcohol 11 ( $1.396 \mathrm{~g}, 84 \%$ ) was obtained as a colorless oil.
${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta=0.76\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.13\left(\mathrm{~d}, \mathrm{~J}=6.7 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.29$ (s, 3H, CH3), $1.44(\mathrm{~d}, \mathrm{~J}=10.3 \mathrm{~Hz}, 1 \mathrm{H}), 1.75-1.78(\mathrm{~m}, 2 \mathrm{H}), 2.18-2.23(\mathrm{~m}, 1 \mathrm{H}), 2.56-$ $2.60(\mathrm{~m}, 1 \mathrm{H}), 4.03(\mathrm{~m}, 1 \mathrm{H}), 4.80(\mathrm{~m}, 1 \mathrm{H}), 5.09(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): $\delta$ $=18.73,21.34,23.90,25.93,39.01,41.72,47.42,51.15,74.02,105.77,154.16$. Anal. calcd for $\mathrm{C}_{11} \mathrm{H}_{18} \mathrm{O}: \mathrm{C}, 79.46$; $\mathrm{H}, 10.91$; found: $\mathrm{C}, 79.27 ; \mathrm{H}, 10.58$.

## 2.9. (1R,5R)-3-PhenylselanyInopinone (14)


(+)-Nopinone ( $5.53 \mathrm{~g}, 40 \mathrm{mmol}$ ), selenium dioxide ( $5.33 \mathrm{~g}, 48 \mathrm{mmol}$ ), diphenyl diselenide ( $6.87 \mathrm{~g}, 22 \mathrm{mmol}$ ), and methanol ( 80 mL ) were placed in a round-bottom flask. A few drops of sulfuric acid were added to the solution and stirring was continued at room temperature for 5 h . The reaction mixture was diluted with saturated sodium bicarbonate $(100 \mathrm{~mL})$ and extracted with ethyl acetate $(3 \times 100 \mathrm{~mL})$. The combined organic layers were dried with anhydrous magnesium sulfate. The solvent was evaporated using a rotary evaporator under reduced pressure. The crude product was purified by column chromatography on silica gel to remove selenium impurities using hexane/ethyl acetate ( $90: 10$ ) as eluent and then hexane/ethyl acetate ( $80: 20$ ). Phenylselenide $14(9.033 \mathrm{~g})$ was obtained in $77 \%$ yield as a light yellow oil.
${ }^{1} \mathrm{H} \operatorname{NMR}\left(700 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 0.85\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.36\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.88(\mathrm{~d}, \mathrm{~J}=$ $10.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.20-2.25(\mathrm{~m}, 2 \mathrm{H}), 2.53-2.61(\mathrm{~m}, 2 \mathrm{H}), 2.70(\mathrm{t}, J=5.3 \mathrm{~Hz}, 1 \mathrm{H}), 3.86$ (dd, J = $9.5 \mathrm{~Hz}, 2.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $7.29-7.33$ (m, 3H), $7.65-7.69(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta(\mathrm{ppm}): 21.97,25.62,25.79,31.53,40.34,40.69,42.07,57.72,128.15$, 129.05 (2xCH), 130.46 (C), 134.62 ( $2 \times \mathrm{CH}$ ), 209.81.

### 2.10. (1R,5R)-Apoverbenone (15)



Phenyl selenide 14 ( $5.866 \mathrm{~g}, 20 \mathrm{mmol}$ ) dissolved in dichloromethane ( 100 mL ) was placed in a round-bottom flask. Pyridine ( $4.84 \mathrm{~mL}, 60 \mathrm{mmol}$ ) was added to the solution followed by $30 \% \mathrm{H}_{2} \mathrm{O}_{2}(4.0 \mathrm{~mL}, 40 \mathrm{mmol})$. The mixture was stirred for 30 minutes at room temperature. The reaction mixture was diluted with saturated sodium bicarbonate solution ( 150 mL ). The two-phase mixture was transferred to a separatory funnel and separated. The aqueous layer was extracted with dichloromethane ( $3 \times 100 \mathrm{~mL}$ ). Organic layers were combined and dried with anhydrous magnesium sulfate. The solvent was evaporated on a rotary evaporator. The product was purified on silica gel by column chromatography (eluent: hexane: ethyl acetate $90: 10$ ) to obtain 1.770 g of a light yellow oil ( $65 \%$ yield), $[\alpha]_{D}^{23}+284.5$ ( $c 1.1, \mathrm{CHCl}_{3}$ ).
${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm})$ : 1.04 (s, 3H, CH3), 1.52 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), 2.14 (d, $J=9.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.60(\mathrm{~m}, 1 \mathrm{H}), 2.72(\mathrm{td}, J=6.0 \mathrm{~Hz}, 1.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.84$ (dtd, $J=9.2 \mathrm{~Hz}$, $5.5 \mathrm{~Hz}, 0.9 \mathrm{~Hz}, 1 \mathrm{H}), 5.95(\mathrm{~m}, 1 \mathrm{H}), 7.52$ (dd, $J=8.6 \mathrm{~Hz}, 6.7 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 176 $\mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 22.20,26.44,41.86,43.75,54.86,58.57,125.57,156.94$, 203.93.

### 2.11. (1R,4R,5R)-(-)-Apopinenol (16)



The reduction of enone 15 was carried out according to the procedure described for compound 6. Enone 15 ( $1.362 \mathrm{~g}, 10 \mathrm{mmol}$ ) was reduced with sodium borohydride ( $0.416 \mathrm{~g}, 11 \mathrm{mmol}$ ) in the presence of cerium chloride heptahydrate ( $5.589 \mathrm{~g}, 15 \mathrm{mmol}$ ). Allylic alcohol 16 ( $1.313 \mathrm{~g}, 95 \%$ ) was obtained as a crystalline product (m.p. $79-82^{\circ} \mathrm{C}$ ). GC analysis has shown the presence of two diastereoisomeric alcohols: $(1 R, 4 R, 5 R)$ 16: $\mathrm{t}_{\mathrm{r}}=8,387$ (92,01\%); (1R,4S,5R)-18: $\mathrm{tr}=8,622$ (7,99\%).
Purification of an analytical sample of this mixture gave pure 16 with $[\alpha]_{D}^{21}-21.2\left(c 1.0, C H C l_{3}\right)$. Lit. [2]: $[\alpha]_{D}^{25}-22.5\left(c 1.0, C_{6} H_{6}\right)$.
${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 1.10(\mathrm{~s}, 3 \mathrm{H}), 1.34(\mathrm{~s}, 3 \mathrm{H}), 1.37(\mathrm{~d}, \mathrm{~J}=9.03 \mathrm{~Hz}$, 1 H ), 1.76 ( $\mathrm{s}, 1 \mathrm{H}, \mathrm{OH}$ ), $2.15(\mathrm{q}, J=5.9 \mathrm{~Hz}, 1 \mathrm{H}), 2.32$ (tdd, $J=5.9 \mathrm{~Hz}, 3.5 \mathrm{~Hz}, 2.2 \mathrm{~Hz}$, 1 H ), 2.48 (dddd, $J=9.1 \mathrm{~Hz}, 6.2 \mathrm{~Hz}, 5.4 \mathrm{~Hz}, 0.6 \mathrm{~Hz}, 1 \mathrm{H}), 4.50(\mathrm{~m}, 1 \mathrm{H}), 5.69$ (m, 1H), 6.31 (ddt, $J=8.7 \mathrm{~Hz}, 6.5 \mathrm{~Hz}, 1.0 \mathrm{~Hz}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $176 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta$ (ppm): 22.67; 26.52; 36.35; 38.38; 42.00; 48.41; 73.66; 126.04; 137.88.

### 2.12. (1R,4S,5R)-(+)-Apopinenol (18)



Allylic alcohol 16 ( $0.691 \mathrm{~g}, 5 \mathrm{mmol}$ ), p-nitrobenzoic acid ( $3.359 \mathrm{~g}, 20.1 \mathrm{mmol}$ ), triphenylphosphine ( $5.232 \mathrm{~g}, 19.95 \mathrm{mmol}$ ) and tetrahydrofuran $(40 \mathrm{~mL})$ were added to the flask under an inert gas atmosphere. The flask was placed in an ice-water bath and diisopropyl azadicarboxylate ( $4.054 \mathrm{~g}, 3.95 \mathrm{~mL}, 20.05 \mathrm{mmol}$ ) was added dropwise at a rate such that the temperature of the solution did not exceed $10^{\circ} \mathrm{C}$. The ice-water bath was removed and the solution was stirred overnight at room temperature followed by a further 3 hours at $40^{\circ} \mathrm{C}$. The reaction mixture was cooled to room temperature, diethyl ether ( 40 mL ) was added, and washed with sodium hydrogen carbonate solution ( $2 \times 25 \mathrm{~mL}$ ). The aqueous layers were combined and washed with diethyl ether ( 25 mL ). The combined organic layers were dried over anhydrous magnesium sulfate. Solvents were evaporated under reduced pressure on a rotary evaporator. The residue was purified by flash column chromatography on silica gel using hexane/ethyl acetate (90:10) as an eluent to obtain p-nitro-benzoate ester 17 ( $1.092 \mathrm{~g}, 76 \%$ ) as a mixture of $4 R: 4 S$ diastereoisomers in the ratio of $79: 21$ from ${ }^{1} \mathrm{H}$ NMR spectra. It was not possible to separate these diastereoisomers by TLC analysis using different elution systems.



4S-ester

17: ${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 1.03$ (s, 3H, 4R), 1.18 (s, 3H, 4S), 1.40 (s, 3H, $4 R+4 S), 1.57(\mathrm{~d}, 1 \mathrm{H}, \mathrm{J}=9.25 \mathrm{~Hz}, 4 \mathrm{~S}), 1.60-1.63(\mathrm{~m}, 1 \mathrm{H}, 4 R), 2.26-2.30(\mathrm{~m}, 1 \mathrm{H}, 4 \mathrm{~S})$, 2.30-2.34 (m, 1H, 4R), 2.40-2.43 (m, 2H, 4R), 2.48-2.52 (m, 1H, 4S), 2.59-2.63 (m, 1H, $4 S), 5.69(\mathrm{~m}, 1 \mathrm{H}, 4 R), 5.74-5.77(\mathrm{~m}, 1 \mathrm{H}, 4 R+4 S), 5.86(\mathrm{~m}, 1 \mathrm{H}, 4 \mathrm{~S}), 6.52(\mathrm{~m}, 1 \mathrm{H}, 4 S)$, 6.56 (dd, J=8.60, $6.20 \mathrm{~Hz}, 1 \mathrm{H}, 4 R$ ), 8.18-8.22 (m, 2H, 4R + 4S), 8.26-8.30 (m, 2H, 4R $+4 S)$.

The obtained mixture of esters was dissolved in THF ( 30 mL ) and $5 \% \mathrm{NaOH}$ aqueous solution ( 20 mL ) was added. The solution was stirred at room temperature to disappear the starting material on TLC analysis $(\approx 5 h)$. Diethyl ether ( 40 mL ) was added, the organic layer was separated, and the aqueous layer was extracted with diethyl ether ( $2 \times 15 \mathrm{~mL}$ ). The organic layers were combined, washed with brine ( 15 mL ), and dried with anhydrous magnesium sulfate. Solvents were evaporated on a rotary evaporator under reduced pressure to obtain 0.431 g ( $82 \%$ ) of the product. GC analysis has shown the existence of two diastereomeric alcohols: $(1 R, 4 S, 5 R)-18: \mathrm{tr}_{\mathrm{r}}=9,950$ (79,45\%) and $(1 R, 4 R, 5 R)-16: t_{r}=9,692(20,54 \%)$.

Purification of an analytical sample of this mixture by column chromatography on silica gel with hexane/diethyl ether 70/30 as an eluent gave pure 18 as a colorless oil with $[\alpha]_{D}^{20}+37.2\left(c 1.3, \mathrm{CHCl}_{3}\right)$. Lit. [2]: $[\alpha]_{D}^{25}+14.7\left(c 1.0, \mathrm{C}_{6} \mathrm{H}_{6}\right)$.
${ }^{1} \mathrm{H}$ NMR ( $700 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta(\mathrm{ppm}): 0.91(\mathrm{~s}, 3 \mathrm{H}), 1.33(\mathrm{~d}, \mathrm{~J}=9.03 \mathrm{~Hz}, 1 \mathrm{H}), 1.34(\mathrm{~s}$, 3 H ), $1.53(\mathrm{~s}, 1 \mathrm{H}, \mathrm{OH}), 2.19-2.23(\mathrm{~m}, 2 \mathrm{H}), 2.29(\mathrm{dtd}, J=9.1 \mathrm{~Hz}, 5.5 \mathrm{~Hz}, 0.6 \mathrm{~Hz}, 1 \mathrm{H})$, $4.33(\mathrm{~m}, 1 \mathrm{H}), 5.65-5.67(\mathrm{~m}, 1 \mathrm{H}), 6.35-6.37(\mathrm{~m}, 1 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ (ppm): 20.90; 26.69; 29.28, 42.71;46.77; 47.56; 70.75; 125.94; 139.45.
3. Literature
[1] Binder, C. M.; Bautista, A.; Zaidlewicz, M.; Krzemiński, M. P.; Oliver, A.; Singaram, B. J. Org. Chem. 2009, 74, 2337-2343.
[2] Nishino, C.; Takayanagi, H. Agric. Biol. Chern. 1980, 44, 1649-1652.

Chromatograms of (-)-(1R,4R,5R)-apopinenol (16) and (+)-(1R,4S,5R)-apopinenol (18)

GC analysis: column: Phenomenex ZB-WAX, $15 \mathrm{~m} \times 0.53 \mathrm{~mm} \times 1.00 \mu \mathrm{~m}$
Initial temperature $60{ }^{\circ} \mathrm{C}$
Initial time 2 min
Rate $10{ }^{\circ} \mathrm{C} / \mathrm{min}$
Final temperature $240{ }^{\circ} \mathrm{C}$


(1R,5R)-(+)-Verbenone (2)










( $1 R, 2 R, 5 S$ )-Isopinocamphone (8) (contains $12 \%$ of unoxidezed isopinocampheol 7 )




(1R,2R,3R,5R)-4-Methyleneneoisopinocampheol (11)



## (1R,5R)-3- PhenyIselanylnopinone (14)



(1R,5R)-Apoverbenone (15)















Diphenyl (((1R,2R,5S)- $\delta$-pinen-3-yl)methyl)phosphine oxide (21)



(( $(1 R, 2 R, 3 R, 4 R, 5 R)$-4-Hydroxypinan-3-yl)methyl)diphenylphosphine oxide (22)




## (1R,2R,3R,4R,5R)-3-((Diphenylphosphanyl)methyl)isoverbanol (23)





Diphenyl (((1R,2S,5R)- $\delta$-pinen-4-yl)methyl)phosphine oxide (26)



(((1R,2R,3S,4S,5S)-3-Hydroxypinan-4-yl)methyl)diphenylphosphine oxide (27)




