



Supporting Information

for

Development of fluorinated benzils and bisbenzils as room-temperature phosphorescent molecules

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Experimental precedures, NMR spectra and Cartesian coordinates

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1. Experimental procedure

1-1. General

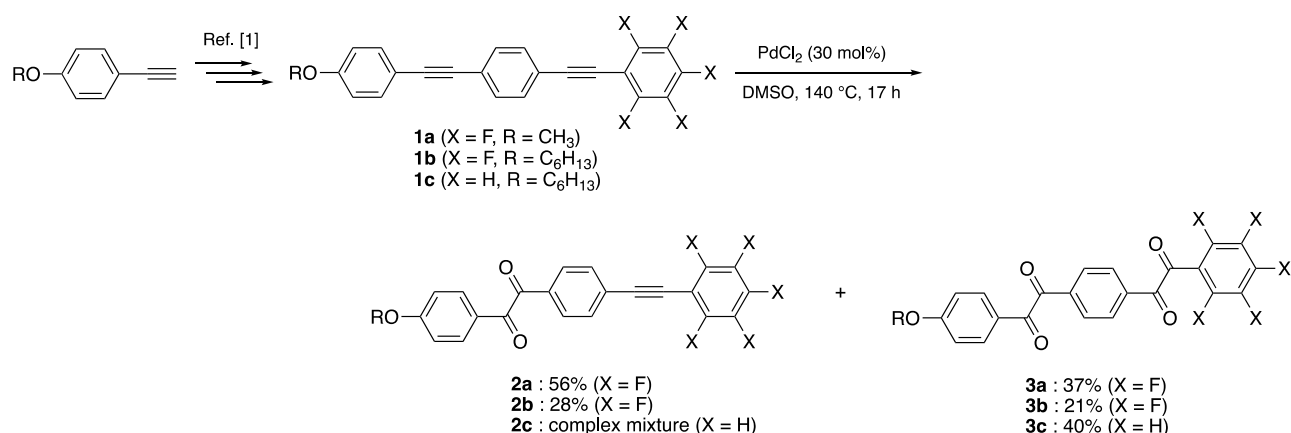
^1H and ^{13}C NMR spectra were obtained with a Bruker AVANCE III 400 NMR spectrometer (^1H : 400 MHz and ^{13}C : 100 MHz) in chloroform-*d* (CDCl_3) solution and the chemical shifts are reported in parts per million (ppm) using the residual proton in the NMR solvent. ^{19}F NMR (376 MHz) spectra were obtained with a Bruker AVANCE III 400 NMR spectrometer in CDCl_3 solution with CFCl_3 ($\delta_{\text{F}} = 0$ ppm) as an internal standard. Infrared spectra (IR) were recorded in a KBr method with a JASCO FT/IR-4100 type A spectrometer; all spectra were reported in wavenumber (cm^{-1}). High-resolution mass spectra (HRMS) were recorded on a JEOL JMS700MS spectrometer using fast atom bombardment (FAB) methods. All chemicals including solvent were of reagent grade and where necessary were purified in the usual manner prior to use. Column chromatography was carried out on silica gel (Wakogel® 60N, 38–100 μm) and thin-layer chromatography (TLC) analysis was performed on silica gel TLC plates (Merck, Silica gel 60F₂₅₄).

1-2. Photophysical properties

UV–vis absorption spectra were recorded on a JASCO V-500 absorption spectrometer. Samples for the absorption measurements were prepared by dissolving the pristine powder solid sample of **2** and **3** in toluene to a concentration of 1.0×10^{-5} M, and the solution was transferred into quartz cuvettes with an optical path length 1.0 cm. The steady-state PL spectra and quantum yields in solution, and pristine powder solid states were acquired using a JASCO FP-6600 fluorescence spectrometer and an absolute PL quantum yield measurement system (Hamamatsu Photonics, C11347-01). A solution-phase sample with a concentration of 1.0×10^{-3} mol L^{-1} was used for PL measurements using quartz cuvettes (1.0 cm path length). The excitation wavelength (λ_{ex}) corresponded to the maximum absorption wavelength.

1-3. Synthesis

Pentafluorinated benzils **2a** and **2b** and bisbenzils **3a** and **3b** were synthesized from the corresponding pentafluorinated bistolanes **1a** and **1b**, which were prepared according to the reported procedure [1] (Scheme S1). The non-fluorinated bisbenzil **3c** was also prepared from the corresponding **1c** in a similar manner.



Scheme S1. Synthetic pathway for benzils **2a**, **2b** and bisbenzils **3a–c**.

Typical procedure for PdCl₂-catalyzed DMSO-oxidation of bistolane **1**

In a two-necked round-bottomed flask equipped with a teflon[®]-coated magnetic stirrer bar was placed pentafluorinated bistolane (**1a**, 0.47 g, 1.2 mmol), freshly prepared according to our previous reports [1], in dimethyl sulfoxide (DMSO, 12 mL). To the solution was added PdCl₂ (63 mg, 1.2 mmol) at room temperature, and the whole was heated at 140 °C for 17 h. The resultant was poured into H₂O (30 mL), and extracted with diethyl ether (30 mL, three times). The combined extracts were washed with H₂O (20 mL), brine (20 mL) and dried over anhydrous sodium sulfate (Na₂SO₄), filtered and concentrated under reduced pressure using a rotary evaporator. The residue was purified by silica gel column chromatography (eluent: hexane/EtOAc = 10:1 → 5:1), followed by recrystallization from hexane, providing the corresponding benzil (**2a**, 0.29 g, 0.67 mmol) in 56% yield as a white solid and bisbenil (**3a**, 0.20 g, 0.44 mmol) in 37% yield as a white solid.

1-(4-Methoxyphenyl)-2-[4-{2-(2,3,4,5,6-pentafluorophenyl)ethyn-1-yl}phenyl]-1,2-ethanedione (**2a**)

Yield: 56% (White solid); R_f = 0.42 (hexane/EtOAc = 5/1); M.p.: 138.2–139.4 °C; ¹H NMR (CDCl₃): δ 3.90 (s, 3H), 6.99 (d, *J* = 9.2 Hz, 2H), 7.69 (d, *J* = 8.8 Hz, 2H), 7.95 (d, *J* = 9.2 Hz, 2H), 8.00 (d, *J* = 8.8 Hz, 2H); ¹³C NMR (CDCl₃): δ 55.7, 77.0–77.5 (one carbon was overlapped with CDCl₃), 99.6 (td, *J* = 18.9, 2.9 Hz), 100.0 (d, *J* = 2.9 Hz), 114.4, 125.9, 127.7, 129.8, 132.3, 132.5, 133.4, 136.0–139.2 (dm, *J* = 252.5 Hz), 140.4–143.8 (dm, *J* = 263.5 Hz), 145.8–148.7 (dm, *J* = 254.7 Hz), 165.2, 192.4, 193.6; ¹⁹F NMR (CDCl₃): δ –135.6 to –135.9 (m, 2F), –151.53 (t, *J* = 20.3 Hz, 1F), –161.6 to –161.9 (m, 2F); IR

[1] (a) Yamada, S.; Miyano, K.; Konno, T.; Agou, T.; Kubota, T.; Hosokai, T. *Org. Biomol. Chem.* **2017**, *15*, 5949; (b) Yamada, S.; Morita, M.; Konno, T. *J. Fluorine Chem.* **2017**, *202*, 54; (c) Yamada, S.; Morita, M.; Agou, T.; Kubota, T.; Ichikawa, T.; Konno, T. *Org. Biomol. Chem.* **2018**, *16*, 5609; (d) Yamada, S.; Miyano, K.; Agou, T.; Kubota, T.; Konno, T. *Crystals* **2019**, *9*, 195; (e) Morita, M.; Yamada, S.; Agou, T.; Kubota, T.; Konno, T. *Appl. Sci.* **2019**, *9*, 1905.

(KBr): ν 3075, 3016, 2971, 2845, 2360, 2345, 2225, 1678, 1600, 1522, 1504, 1266, 1168, 993 cm^{-1} ; HRMS (FAB+) m/z $[M+H]^+$ calcd for $\text{C}_{23}\text{H}_{12}\text{F}_5\text{O}_3$: 431.0707, found 431.0709.

1-(4-Hexyloxyphenyl)-2-[4-{2-(2,3,4,5,6-pentafluorophenyl)ethyn-1-yl}phenyl]-1,2-ethanedione (2b)

Yield: 28% (Yellow solid); R_f = 0.57 (hexane/EtOAc = 5/1); M.p.: 114.0–114.5 $^{\circ}\text{C}$; ^1H NMR (CDCl_3): δ 0.90 (t, J = 6.8 Hz, 3H), 1.30–1.38 (m, 4H), 1.46 (quin., J = 8.0 Hz, 2H), 1.81 (quin., J = 6.8 Hz, 2H), 4.04 (t, J = 6.8 Hz, 2H), 6.97 (d, J = 8.8 Hz, 2H), 7.69 (d, J = 8.8 Hz, 2H), 7.94 (d, J = 8.8 Hz, 2H), 7.99 (d, J = 8.8 Hz, 2H); ^{13}C NMR (CDCl_3): δ 14.0, 22.5, 25.6, 28.9, 31.5, 68.6, 77.0–77.5 (one carbon was overlapped with CDCl_3), 99.6 (td, J = 17.6, 2.9 Hz), 100.0 (d, J = 2.9 Hz), 114.9, 125.6, 127.7, 129.8, 132.3, 132.4, 133.4, 136.2–139.3 (dm, J = 250.1 Hz), 140.4–143.7 (dm, J = 262.6 Hz), 145.7–148.8 (dm, J = 254.5 Hz), 164.8, 192.4, 193.6; ^{19}F NMR (CDCl_3): δ –135.6 to –135.9 (m, 2F), –151.56 (t, J = 20.3 Hz, 1F), –161.6 to –161.9 (m, 2F); IR (KBr): ν 3073, 2937, 2857, 2364, 2329, 2217, 1670, 1600, 1522, 1502, 1261, 1170, 992 cm^{-1} ; HRMS (FAB+) m/z $[M+H]^+$ calcd for $\text{C}_{28}\text{H}_{22}\text{F}_5\text{O}_3$: 501.1490, found 501.1491.

1,1'-(1,4-Phenylene)-2-(4-methoxyphenyl)-2'-(2,3,4,5,6-pentafluorophenyl)bis(1,2-ethanedione) (3a)

Yield: 37% (Yellow solid); R_f = 0.42 (hexane/EtOAc = 10/1); M.p.: 113.7–114.8 $^{\circ}\text{C}$; ^1H NMR (CDCl_3): δ 3.91 (s, 3H), 7.01 (d, J = 8.8 Hz, 2H), 7.97 (d, J = 8.8 Hz, 2H), 8.15 (ABq, J = 8.8 Hz, 2H), 8.17 (ABq, J = 8.8 Hz, 2H); ^{13}C NMR (CDCl_3): δ 55.7, 114.5, 125.6, 128.8, 130.3, 130.7, 132.5, 135.0, 136.5–139.8 (dm, J = 247.2 Hz), 137.8, 143.2–146.3 (dm, J = 271.3 Hz), 144.5–148.0 (dm, J = 254.4 Hz), 165.4, 184.3, 187.8, 191.7, 193.3; ^{19}F NMR (CDCl_3): δ –137.8 to –138.0 (m, 2F), –143.97 (t, J = 20.7 Hz, 1F), –159.4 to –159.7 (m, 2F); IR (KBr): ν 3056, 2980, 2851, 1735, 1700, 1686, 1602, 1498, 1425, 1257, 1172, 1001, 888 cm^{-1} ; HRMS (FAB+) m/z $[M+H]^+$ calcd for $\text{C}_{28}\text{H}_{22}\text{F}_5\text{O}_5$: 533.1388, found 533.1380.

1,1'-(1,4-Phenylene)-2-(4-hexyloxyphenyl)-2'-(2,3,4,5,6-pentafluorophenyl)bis(1,2-ethanedione) (3b)

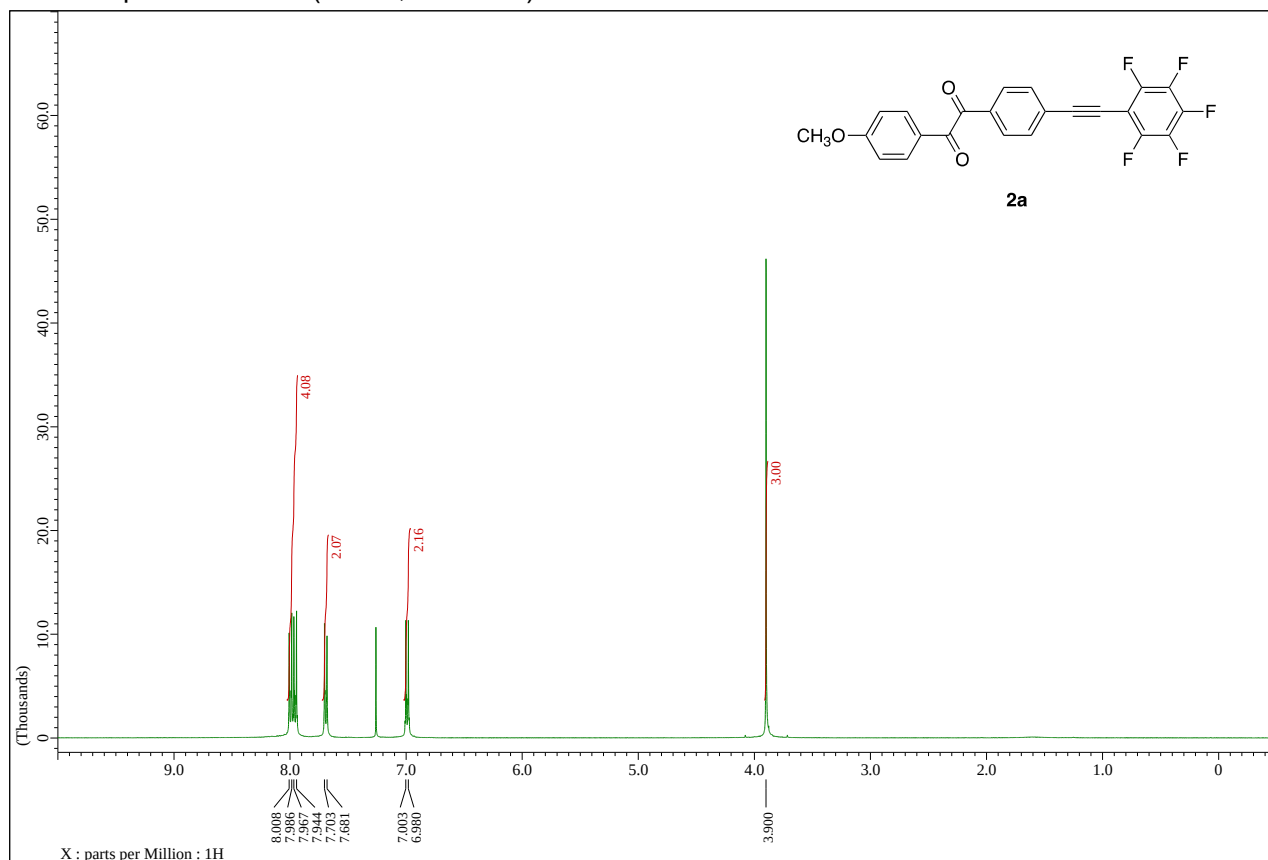
Yield: 21% (Yellow solid); R_f = 0.34 (hexane/EtOAc = 10/1); M.p.: 94.3–95.9 $^{\circ}\text{C}$; ^1H NMR (CDCl_3): δ 0.91 (t, J = 6.4 Hz, 3H), 1.30–1.52 (m, 6H), 1.81 (quin., J = 6.8 Hz, 2H), 4.05 (t, J = 6.8 Hz, 2H), 6.98 (d, J = 8.8 Hz, 2H), 7.94 (d, J = 8.8 Hz, 2H), 8.14 (ABq, J = 8.8 Hz, 2H), 8.17 (ABq, J = 8.8 Hz, 2H); ^{13}C NMR (CDCl_3): δ 14.0, 22.5, 25.6, 28.9, 31.5, 68.6, 115.0, 125.4, 128.8, 130.3, 130.7, 132.5, 134.9, 136.2–139.3 (dm, J = 250.1 Hz), 137.8, 140.4–143.7 (dm, J = 262.6 Hz), 145.7–148.8 (dm, J = 254.5 Hz), 165.0, 184.4 (d, J = 11.1 Hz), 187.8, 191.8, 193.4, three sp^2 carbons attached with fluorine atom cannot be detected due to low solubility; ^{19}F NMR (CDCl_3): δ –137.8 to –138.0 (m, 2F), –143.97 (tt, J = 20.7, 5.6 Hz, 1F), –159.4 to –159.7 (m, 2F); IR (KBr): ν 3098, 2939, 2857, 1684, 1603, 1498, 1255, 1172, 991, 896 cm^{-1} ; HRMS (FAB+) m/z $[M+H]^+$ calcd for $\text{C}_{28}\text{H}_{22}\text{F}_5\text{O}_3$: 501.1490, found 501.1491.

1,1'-(1,4-Phenylene)-2-(4-hexyloxyphenyl)-2'-phenylbis(1,2-ethanedione) (3c)

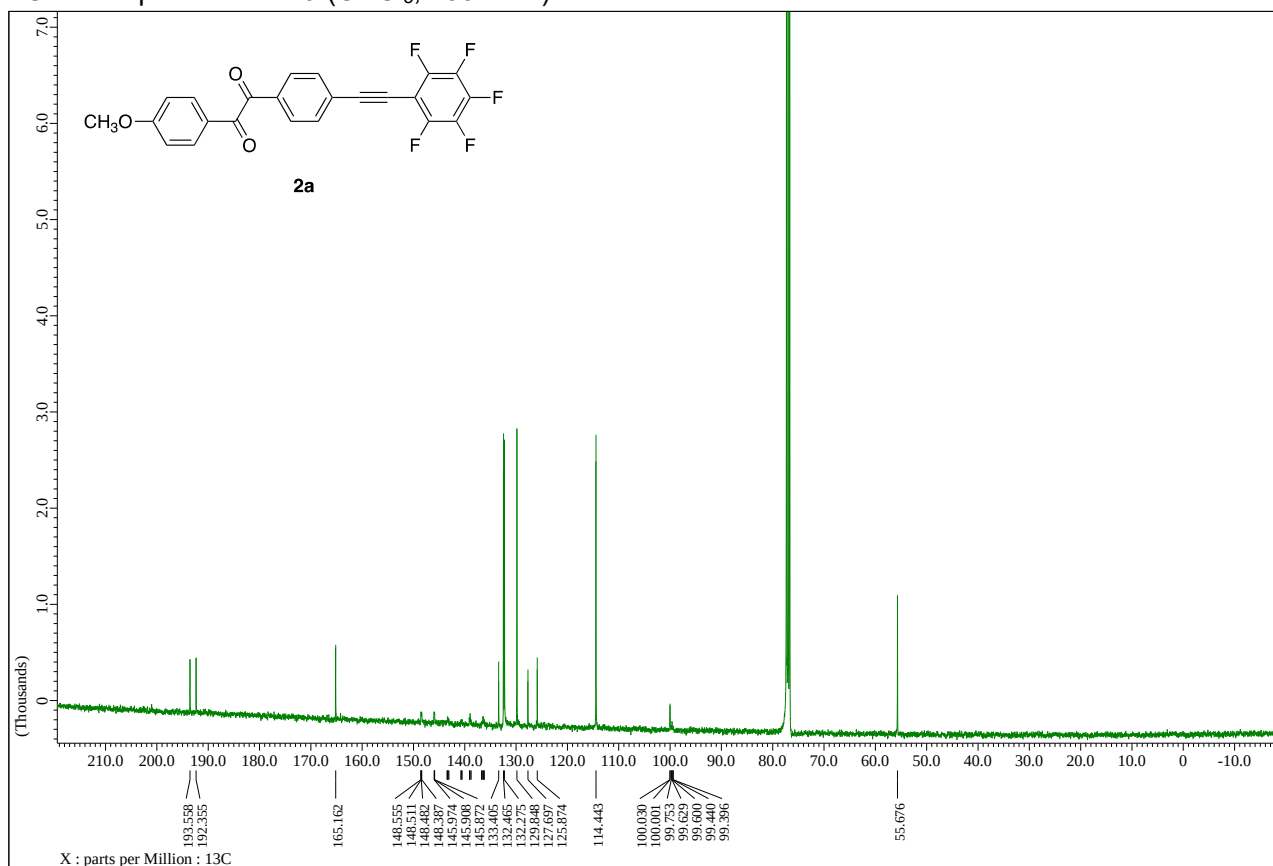
Yield: 40% (Yellow solid); R_f = 0.45 (hexane/EtOAc = 5/1); M.p.: 115.6–116.8 °C; ^1H NMR (CDCl_3): δ 0.90 (t, J = 7.6 Hz, 3H), 1.30–1.37 (m, 4H), 1.46 (quin., J = 8.0 Hz, 2H), 1.80 (quin., J = 7.6 Hz, 2H), 4.04 (t, J = 6.4 Hz, 2H), 6.97 (d, J = 9.2 Hz, 2H), 7.53 (t, J = 8.0 Hz, 2H), 7.68 (tt, J = 7.2, 1.6 Hz, 1H), 7.92 (d, J = 8.8 Hz, 2H), 7.96 (dd, J = 8.4 Hz, 1.2 Hz, 2H), 8.10 (s, 4H); ^{13}C NMR (CDCl_3): δ 14.0, 22.5, 25.6, 28.9, 31.4, 68.6, 114.9, 125.3, 129.1, 129.9, 130.16, 130.22, 132.5, 132.6, 135.2, 136.9, 137.4, 164.9, 191.9, 193.3, 193.5, 193.6; IR (KBr): ν 3069, 2951, 2857, 2361, 2332, 1675, 1663, 1601, 1572, 1263, 1210, 1169, 887 cm^{-1} ; HRMS (FAB+) m/z $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{28}\text{H}_{27}\text{O}_5$: 443.1859, found 443.1869.

2. NMR spectra

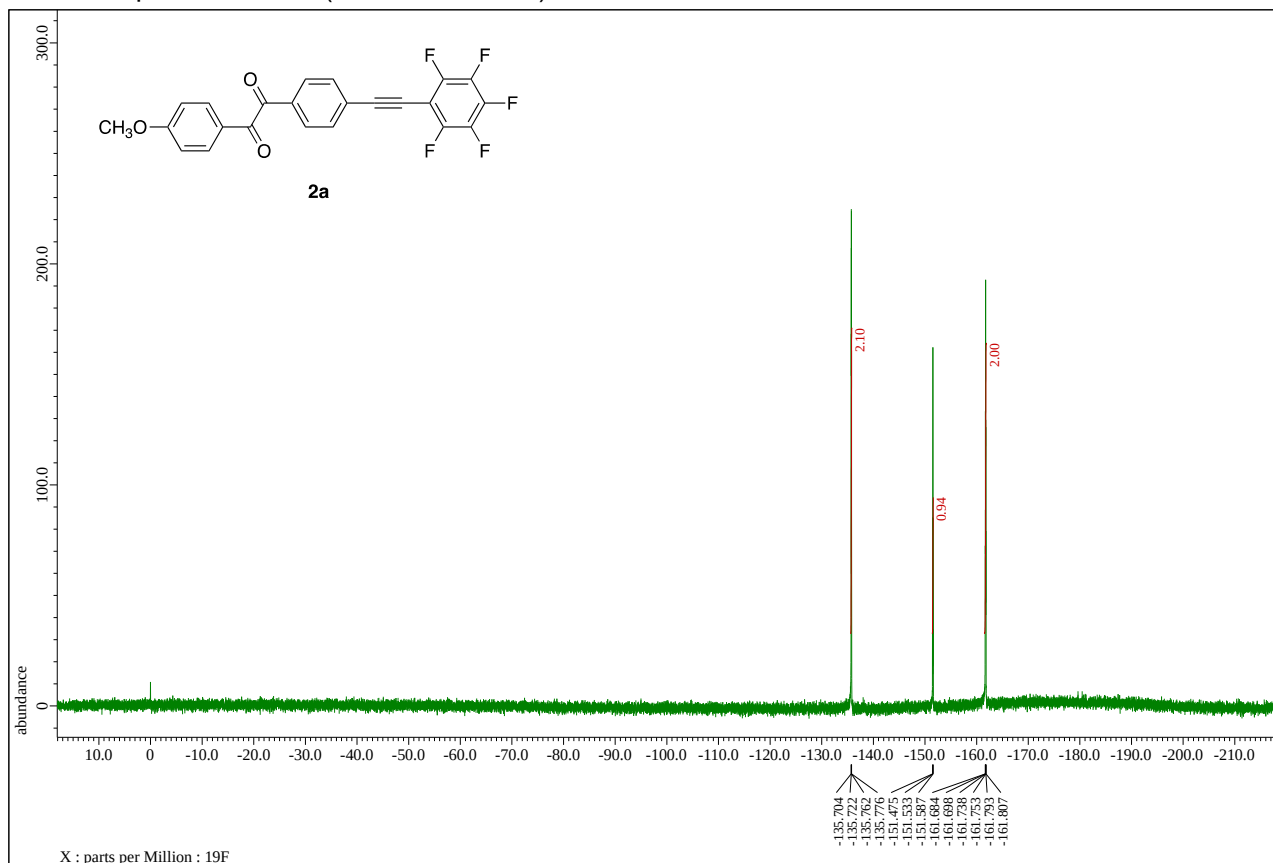
¹H NMR spectrum for **2a** (CDCl₃, 400 MHz)



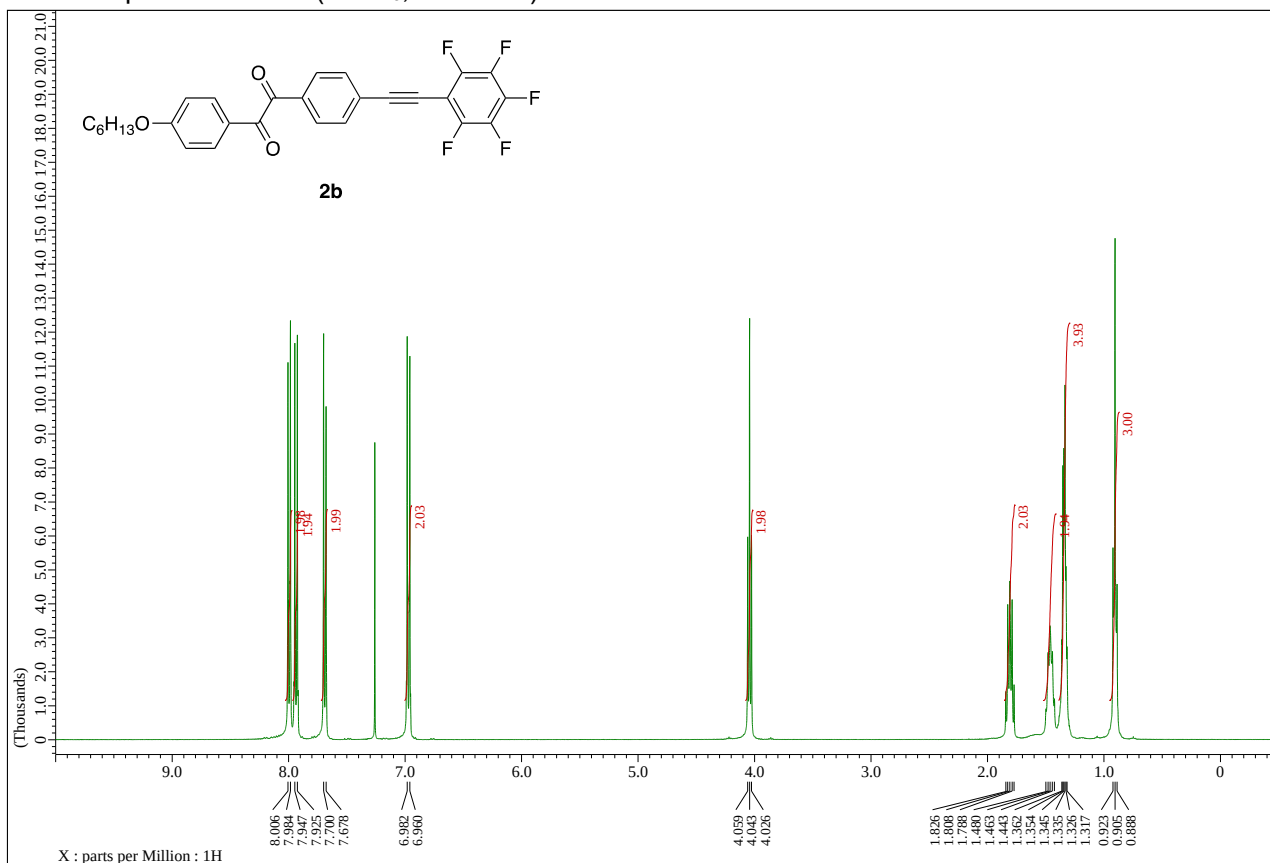
¹³C NMR spectrum for **2a** (CDCl₃, 100 MHz)



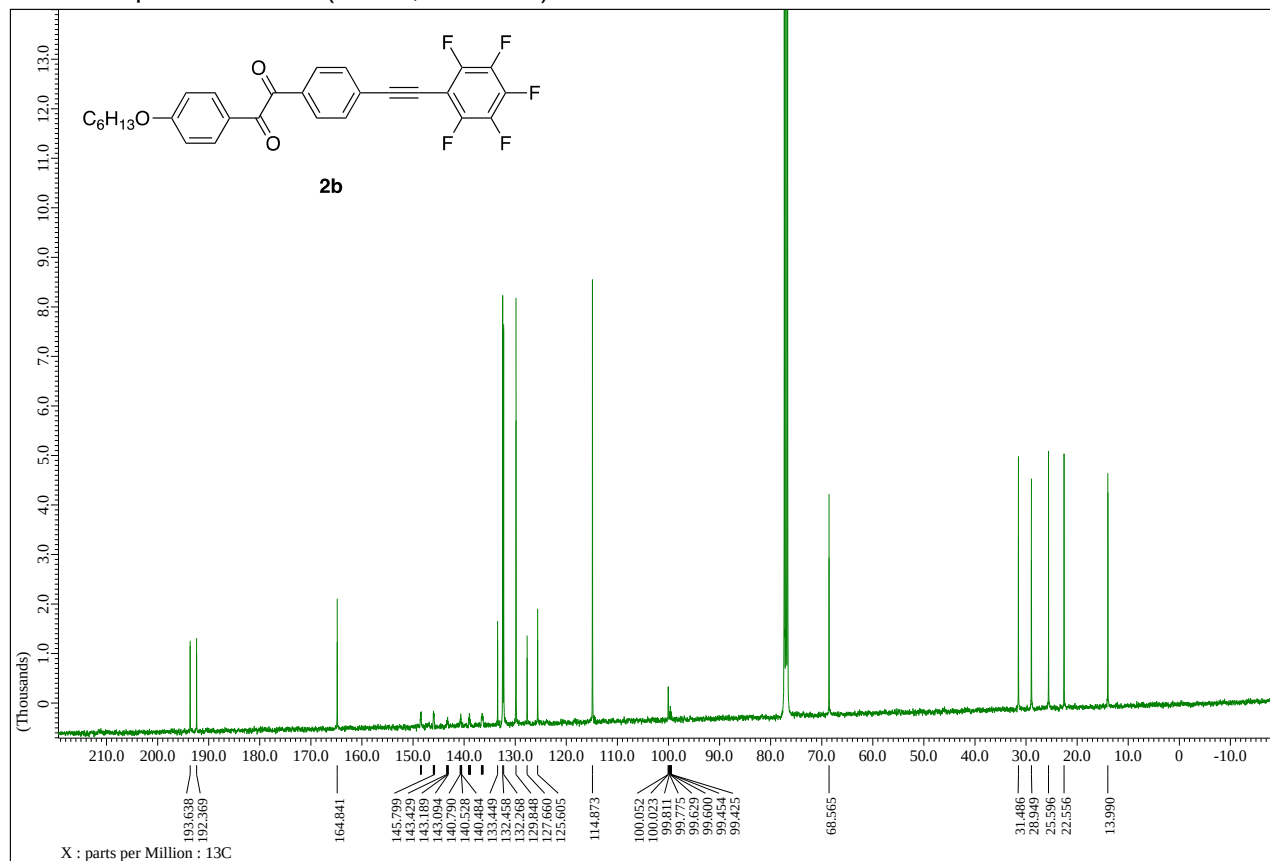
^{19}F NMR spectrum for **2a** (CDCl_3 , 376 MHz)



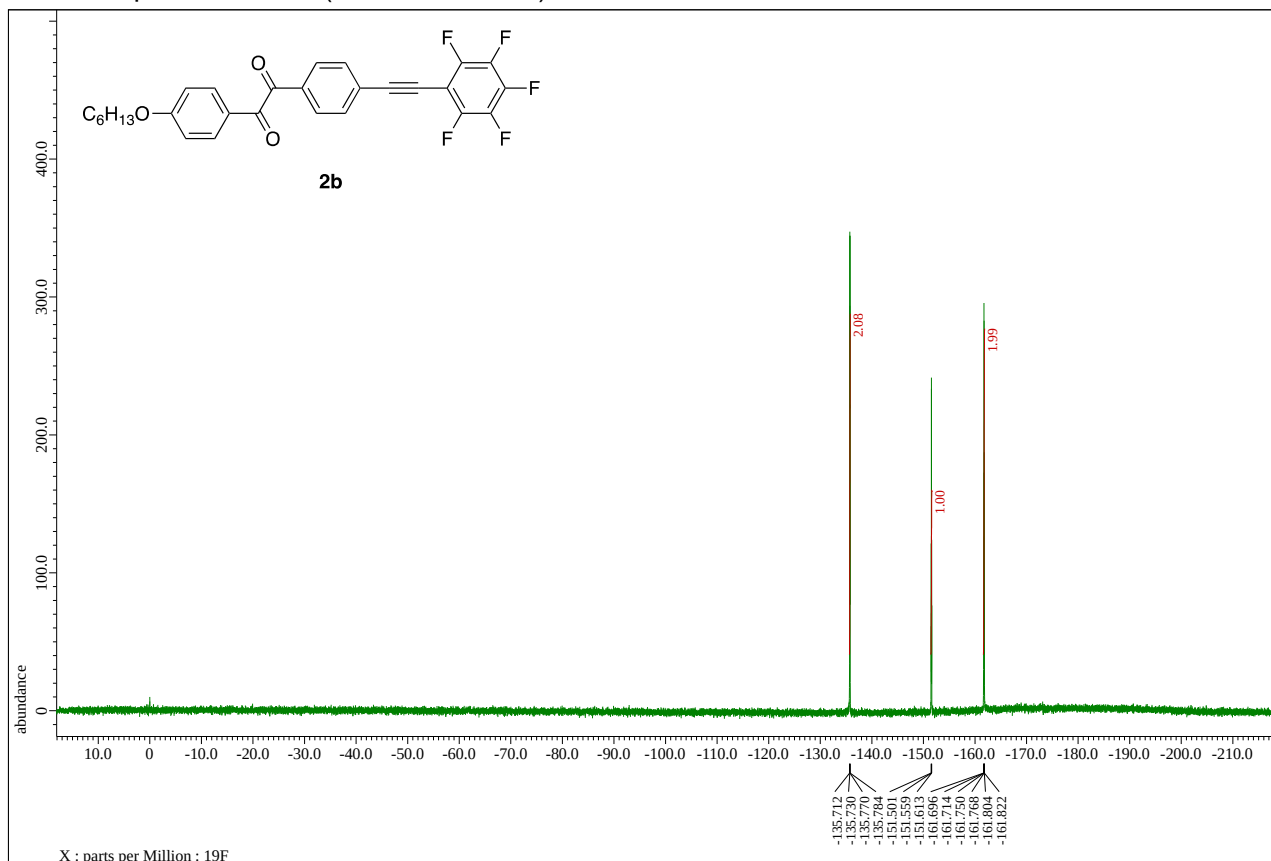
^1H NMR spectrum for **2b** (CDCl_3 , 400 MHz)



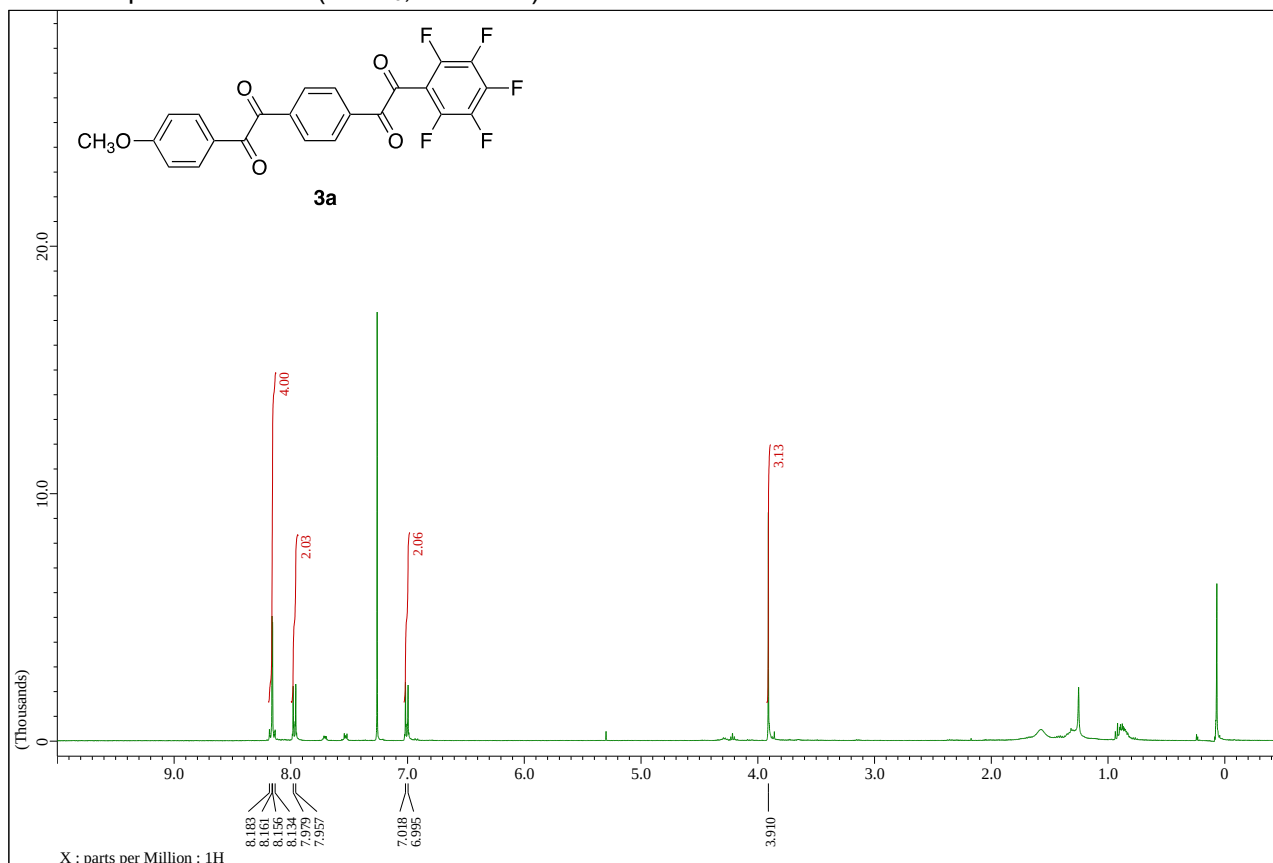
^{13}C NMR spectrum for **2b** (CDCl_3 , 100 MHz)



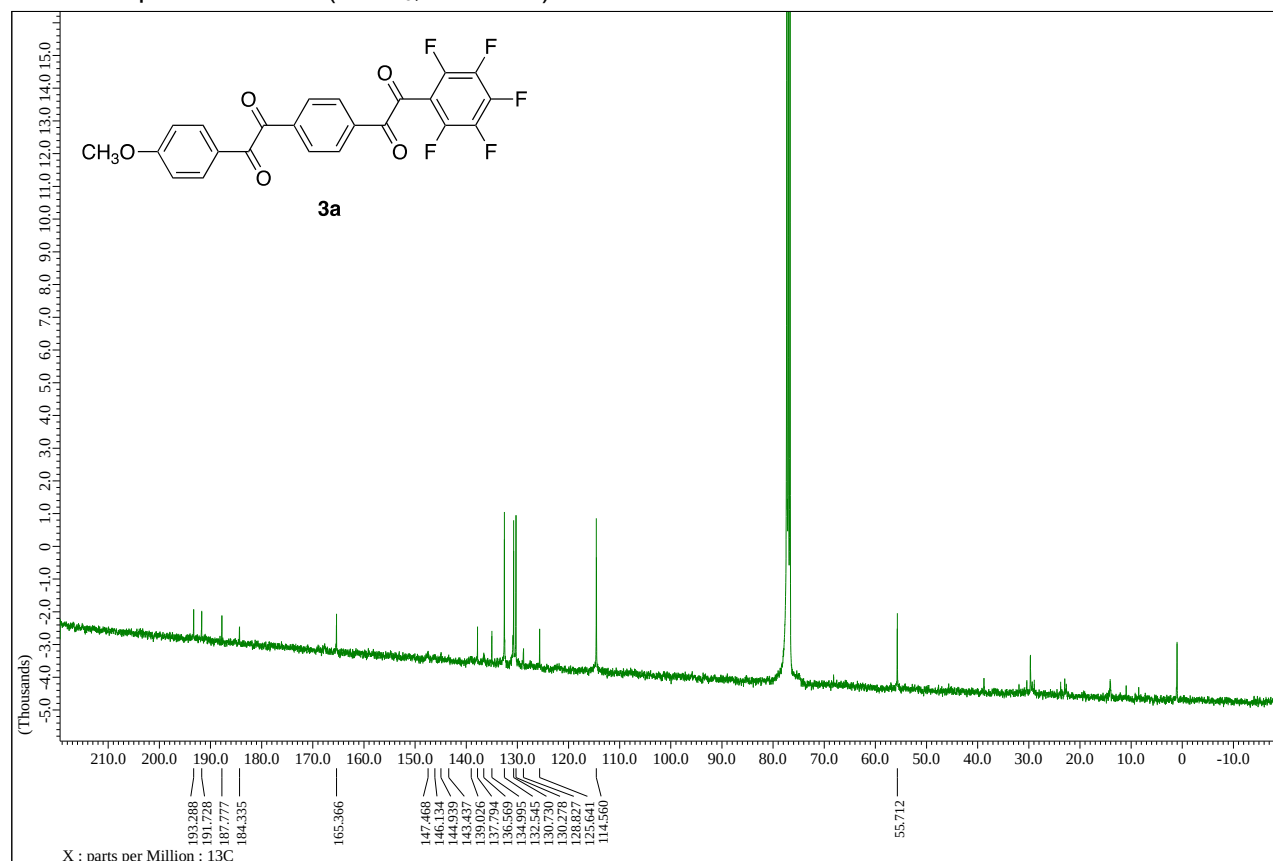
^{19}F NMR spectrum for **2b** (CDCl_3 , 376 MHz)



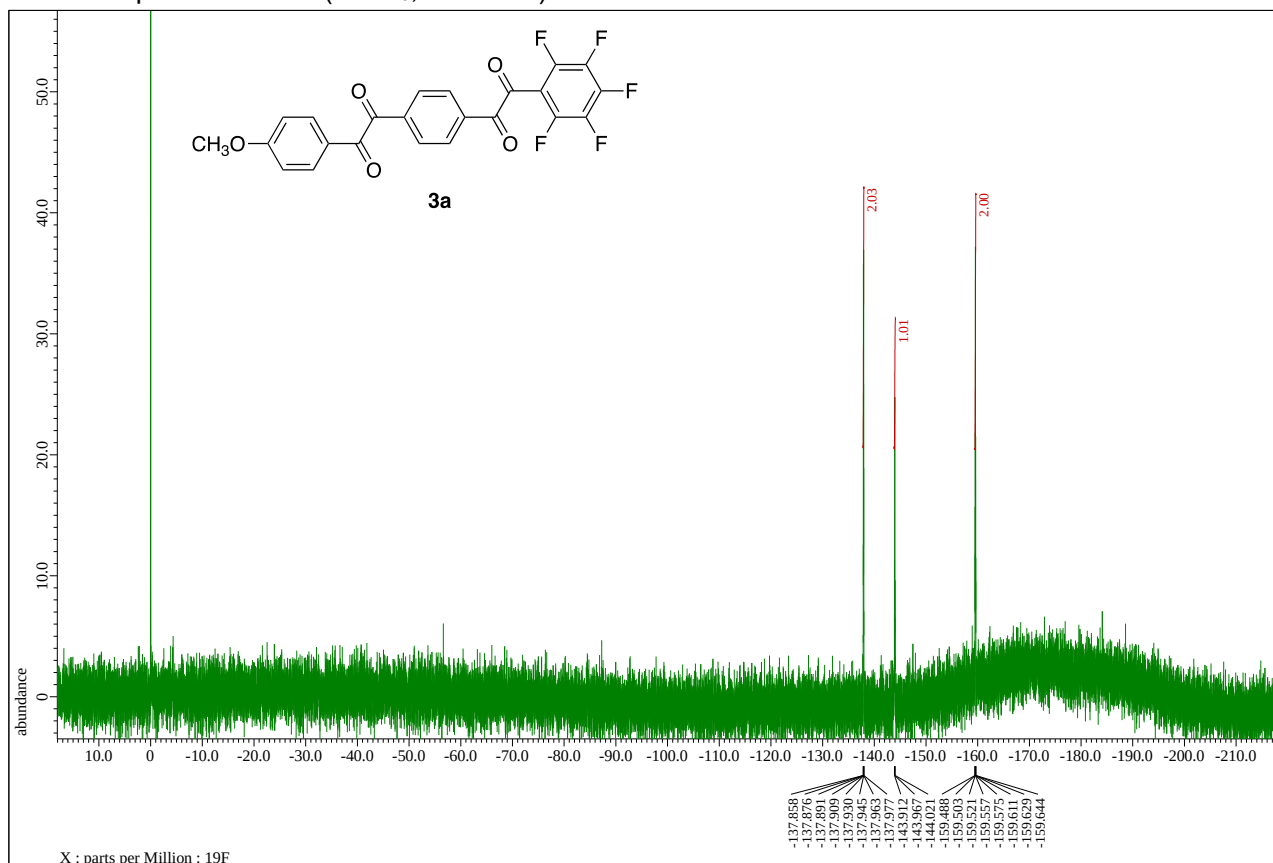
¹H NMR spectrum for **3a** (CDCl₃, 400 MHz)



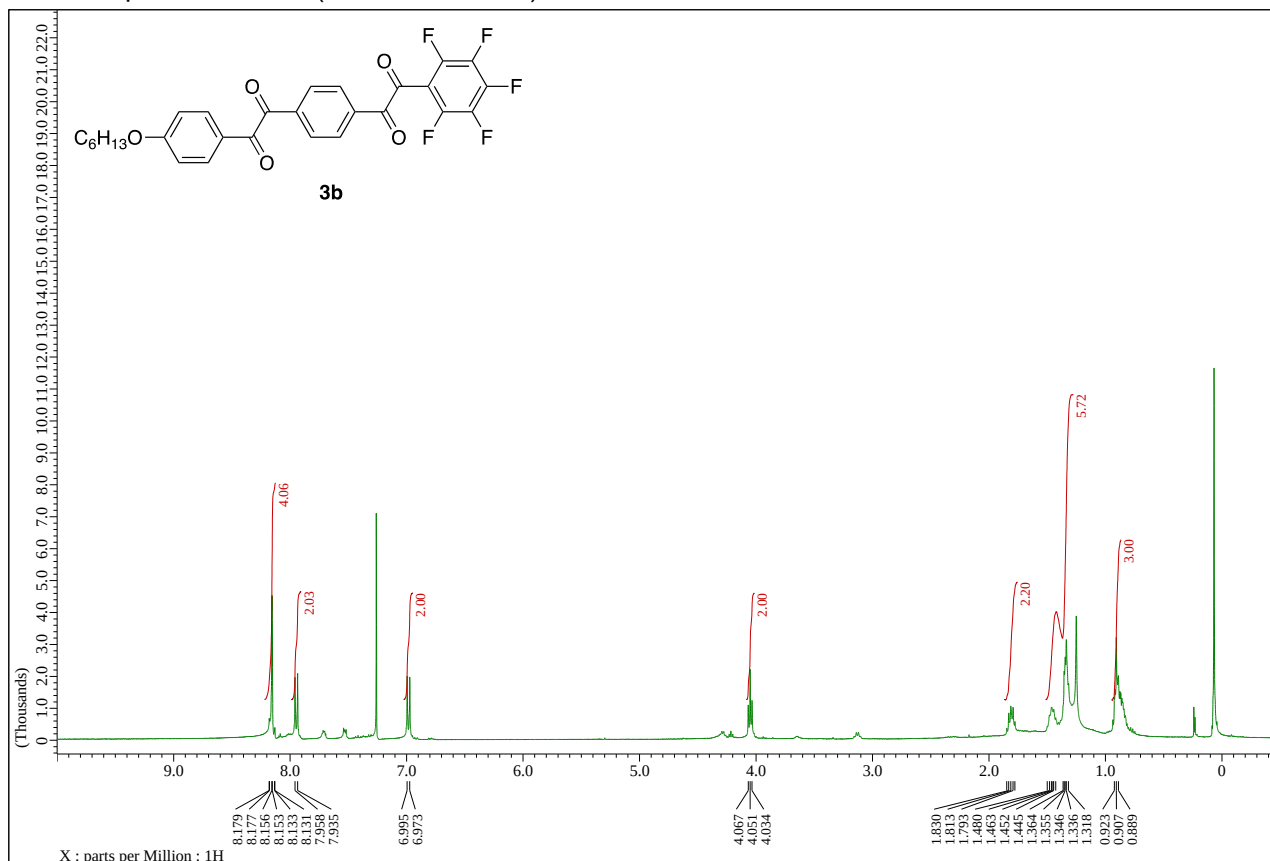
¹³C NMR spectrum for **3a** (CDCl₃, 100 MHz)



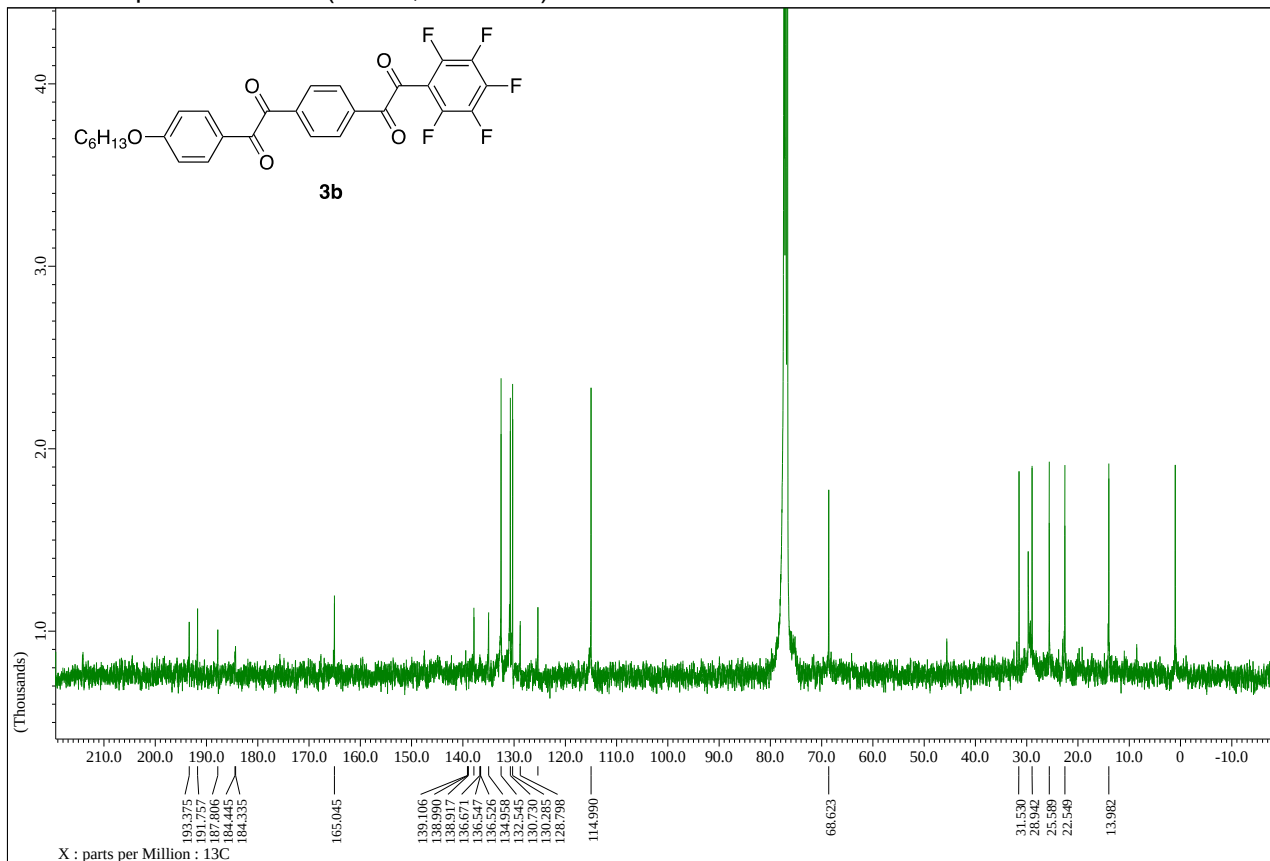
^{19}F NMR spectrum for **3a** (CDCl_3 , 376 MHz)



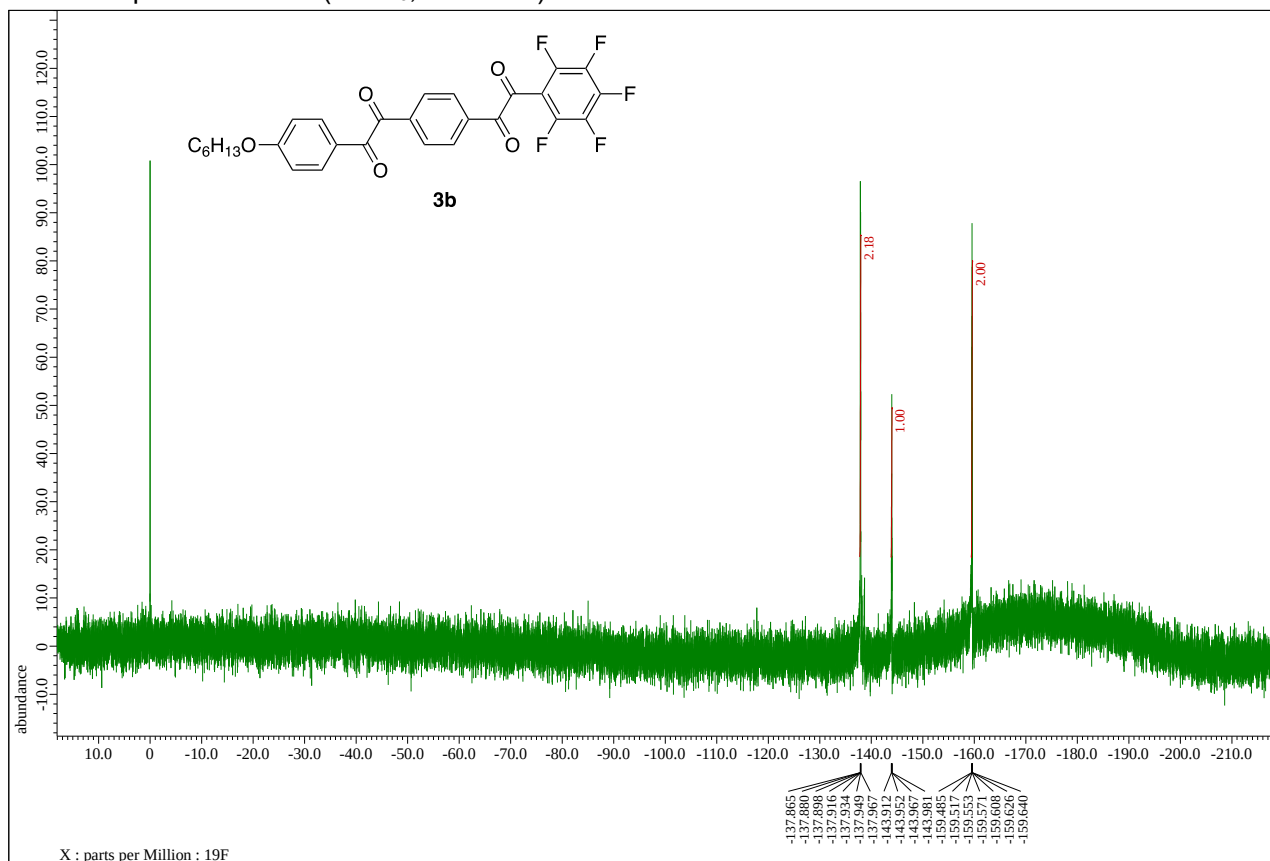
^1H NMR spectrum for **3b** (CDCl_3 , 400 MHz)



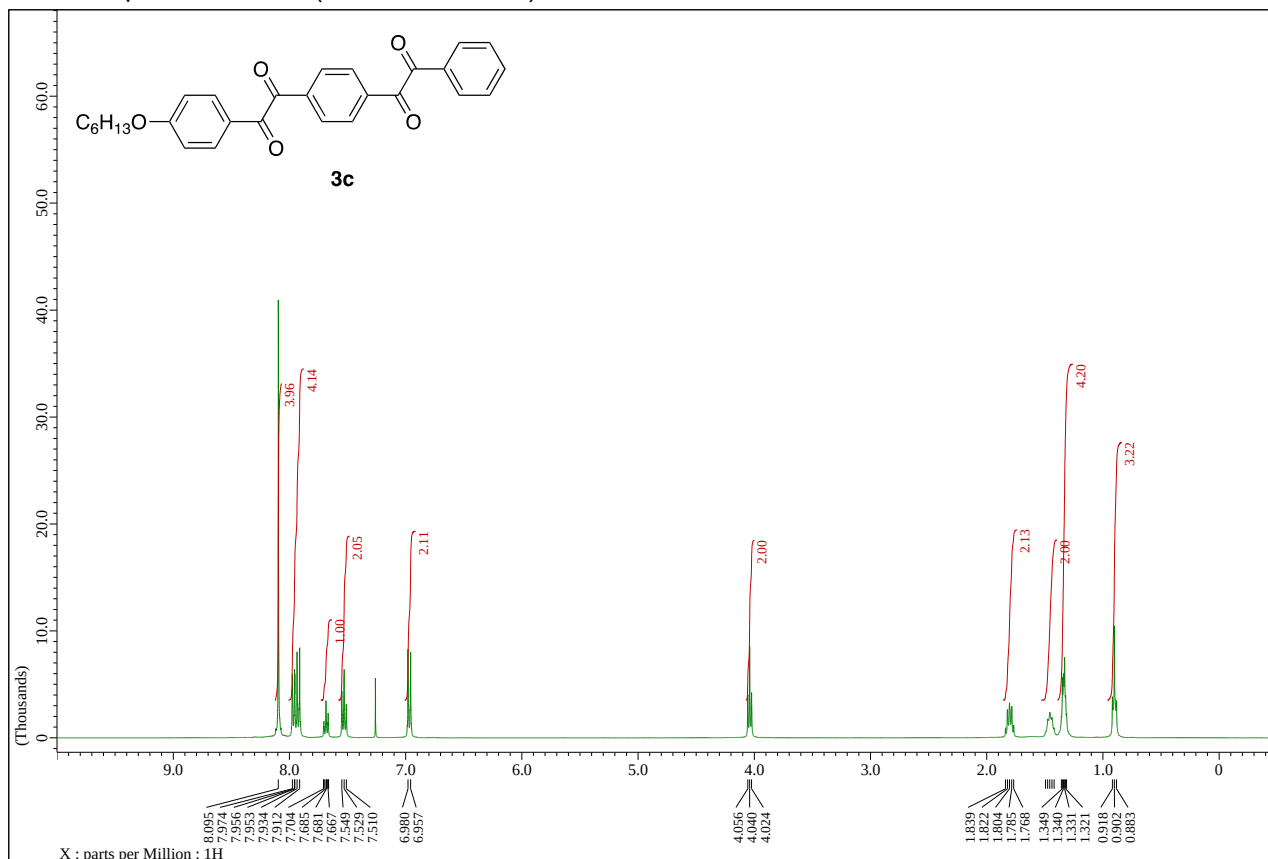
^{13}C NMR spectrum for **3b** (CDCl_3 , 100 MHz)



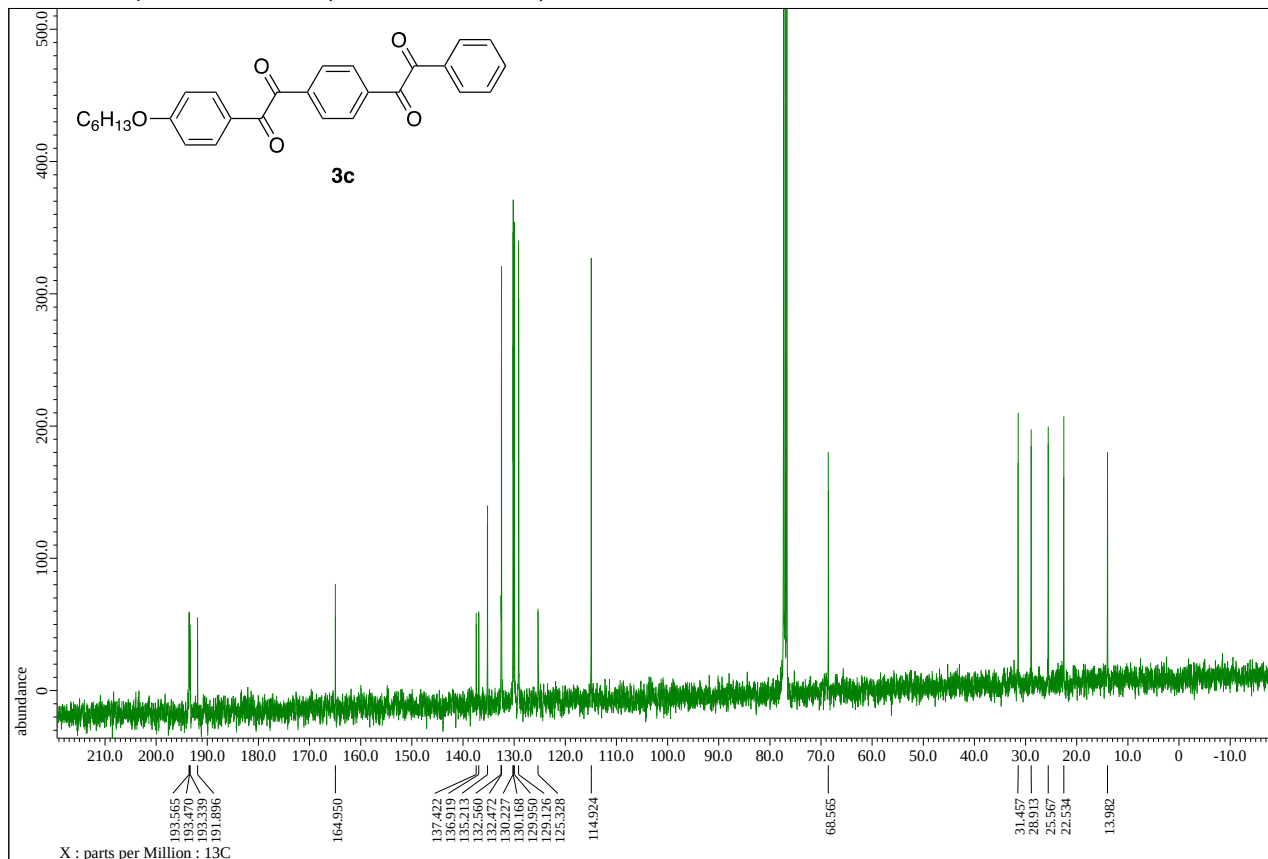
^{19}F NMR spectrum for **3b** (CDCl_3 , 376 MHz)



^1H NMR spectrum for **3c** (CDCl_3 , 400 MHz)



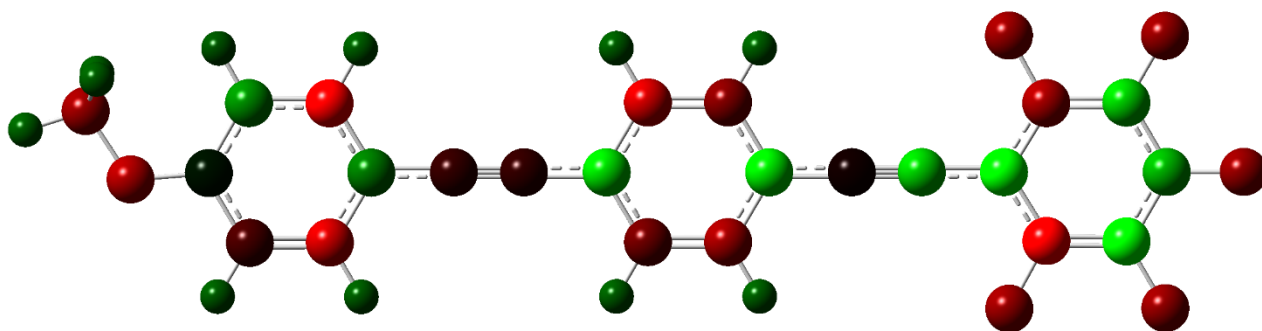
^{13}C NMR spectrum for **3c** (CDCl_3 , 100 MHz)



3. Computation

All computations were performed by a density functional theory (DFT) using the Gaussian 16 (Rev. B.01) suite of programs. Geometry optimizations were executed at the CAM-B3LYP/6-31G(d) level of theory with the implicit solvation model, namely, the conductor-like polarizable continuum model (CPCM), for heptane. The vertical electronic transitions were calculated using a time-dependent (TD)-DFT method at the same level of theory.

3-1. Optimized structure (charge distribution) of 1a



SCF Done: E (RCAM-B3LYP) = -1456.76520762 hartree

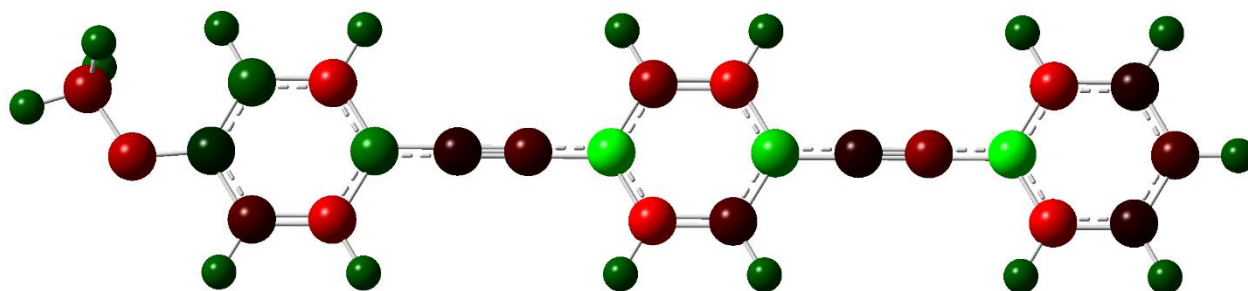
Dipole moment (field-independent basis, Debye-Ang):

X = -4.9568, Y = 1.2690, Z = 0.0001, Tot = 5.1167

3-2. Cartesian coordinates

No.	Atom	Type	Coordinates (Angstroms)								
	No.		X	y	z						
1	6	0	-11.09	0.873484	0.000009	20	6	0	0.037781	-1.25508	-0.000108
2	1	0	-10.9045	1.47744	0.895991	21	1	0	0.594707	-2.1863	-0.000217
3	1	0	-12.1259	0.534657	-0.000058	22	6	0	0.73295	-0.037224	0.000029
4	1	0	-10.9044	1.47761	-0.895852	23	6	0	0.006753	1.1624	0.000186
5	8	0	-10.299	-0.299563	-0.000073	24	1	0	0.539664	2.10758	0.000301
6	6	0	-8.94753	-0.173342	-0.000049	25	6	0	-1.37767	1.14371	0.000192
7	6	0	-8.22376	-1.37095	-0.000137	26	1	0	-1.93413	2.07501	0.000314
8	1	0	-8.77109	-2.30767	-0.000225	27	6	0	-2.07656	-0.073287	0.000041
9	6	0	-6.84285	-1.34824	-0.000111	28	6	0	2.15988	-0.018659	0.000017
10	1	0	-6.28671	-2.28009	-0.000185	29	6	0	3.36832	-0.002205	0.000013
11	6	0	-6.14231	-0.129029	0.000003	30	6	0	4.78789	0.017019	0.000002
12	6	0	-6.8783	1.05909	0.0001	31	6	0	5.53557	-1.16517	0.000052
13	1	0	-6.35331	2.00894	0.000194	32	6	0	6.92003	-1.15512	0.000042
14	6	0	-8.26954	1.04603	0.000075	33	6	0	7.59898	0.05526	-0.000027
15	1	0	-8.80717	1.98632	0.000159	34	6	0	6.88737	1.24673	-0.000074
16	6	0	-4.71431	-0.107056	0.000031	35	6	0	5.50315	1.2191	-0.000064
17	6	0	-3.50376	-0.091643	0.000049	36	9	0	4.90939	-2.34385	0.000118
18	6	0	-1.3467	-1.27198	-0.000102	37	9	0	7.60473	-2.30023	0.000093
19	1	0	-1.87907	-2.21724	-0.000211	38	9	0	8.93022	0.073351	-0.000037
						39	9	0	7.5407	2.41005	-0.000135
						40	9	0	4.84515	2.38037	-0.000112

3-3. Optimized structure (charge distribution) of 1c



SCF Done: E (RCAM-B3LYP) = -960.681573201 hartree

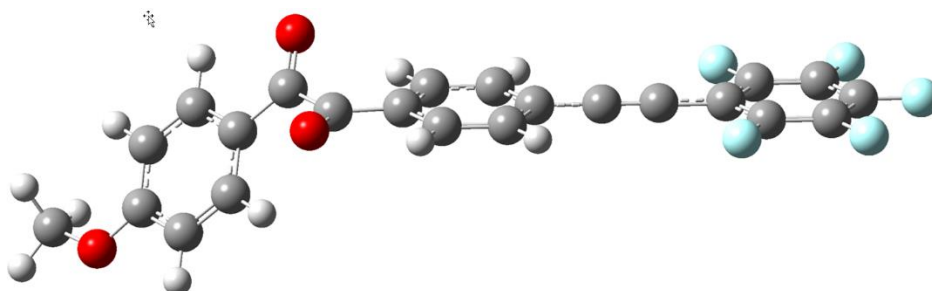
Dipole moment (field-independent basis, Debye-Ang):

X = 1.2195, Y = -1.3069, Z = 0.0007, Tot = 1.7875

3-4. Cartesian coordinates

No.	Atom		Coordinates (Angstroms)								
	No.	Type	X	y	z						
1	6	0	9.42305	-0.861537	0.000486	20	6	0	-1.71362	1.237	-0.000185
2	1	0	9.23864	-1.46564	0.896729	21	1	0	-2.27185	2.1675	-0.000296
3	1	0	10.4586	-0.521509	0.00005	22	6	0	-2.41004	0.019411	-0.000021
4	1	0	9.23827	-1.46682	-0.894881	23	6	0	-1.67617	-1.17598	0.000119
5	8	0	8.63057	0.309972	-0.000127	24	1	0	-2.20529	-2.12334	0.000252
6	6	0	7.27837	0.180508	-0.000069	25	6	0	-0.291357	-1.15448	0.0001
7	6	0	6.55168	1.376	-0.001034	26	1	0	0.266841	-2.08498	0.000207
8	1	0	7.09649	2.31424	-0.001769	27	6	0	0.405539	0.063136	-0.000065
9	6	0	5.17054	1.34986	-0.001039	28	6	0	-3.83837	-0.002775	-0.000003
10	1	0	4.6122	2.2804	-0.001799	29	6	0	-5.04852	-0.02163	0.000011
11	6	0	4.47232	0.129399	-0.000084	30	6	0	-6.4785	-0.043963	0.000028
12	6	0	5.21173	-1.0565	0.000869	31	6	0	-7.20843	1.15328	0.001269
13	1	0	4.6891	-2.00766	0.001617	32	6	0	-8.5972	1.12791	0.001289
14	6	0	6.60324	-1.04012	0.000887	33	6	0	-9.27725	-0.087711	0.000071
15	1	0	7.14291	-1.97929	0.001664	34	6	0	-8.55955	-1.28148	-0.001169
16	6	0	3.04395	0.104069	-0.000089	35	6	0	-7.17066	-1.26342	-0.001195
17	6	0	1.83344	0.085372	-0.000087	36	1	0	-6.67571	2.09878	0.002219
18	6	0	-0.328792	1.25857	-0.000206	37	1	0	-9.15149	2.06177	0.002261
19	1	0	0.200247	2.20592	-0.000333	38	1	0	-10.363	-0.104668	0.000089
						39	1	0	-9.08437	-2.23221	-0.002122
						40	1	0	-6.60867	-2.19182	-0.002169

3-5. Optimized geometry of 2a



SCF Done: E (RCAM-B3LYP) = -1607.22472657 hartree

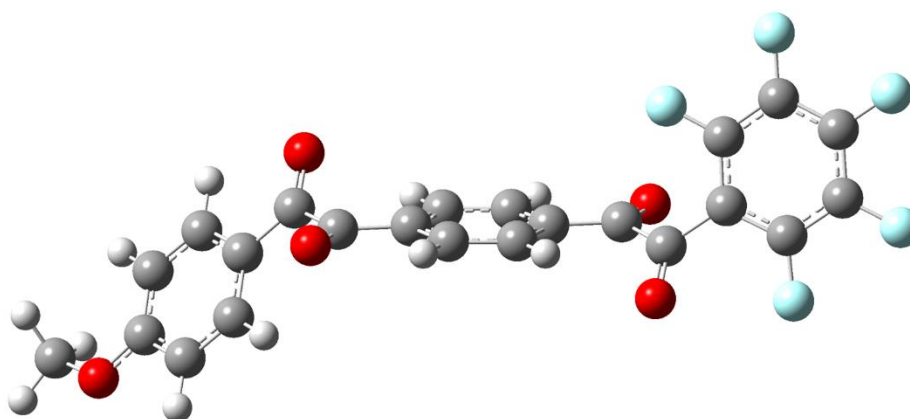
Dipole moment (field-independent basis, Debye-Ang):

X = -3.5454, Y = 2.5640, Z = -1.8632, Tot = 4.7556

3-6. Cartesian coordinates of 2a

No.	Atom		Coordinates (Angstroms)								
	No.	Type	X	y	z						
1	6	0	-7.79207	0.664165	-0.994869	21	1	0	0.983298	-2.73583	-0.179698
2	6	0	-6.90737	-0.291934	-1.50785	22	1	0	0.345439	1.36266	0.940614
3	6	0	-5.78984	-0.660464	-0.784976	23	8	0	-4.0912	0.107549	2.34942
4	6	0	-5.5279	-0.081481	0.467166	24	8	0	-3.80291	-2.59381	0.53562
5	6	0	-6.41794	0.872884	0.966864	25	8	0	-8.85667	0.957871	-1.77382
6	6	0	-7.54476	1.24949	0.251881	26	6	0	-9.79951	1.91235	-1.31661
7	1	0	-7.12504	-0.733508	-2.47431	27	1	0	-9.33225	2.89364	-1.17722
8	1	0	-5.12542	-1.41617	-1.18747	28	1	0	-10.5567	1.97812	-2.0974
9	1	0	-6.21086	1.31903	1.93411	29	1	0	-10.2673	1.58831	-0.380287
10	1	0	-8.21739	1.9904	0.666081	30	6	0	2.21457	-0.395005	0.205293
11	6	0	-4.33521	-0.407147	1.26976	31	6	0	3.39485	-0.173679	0.074391
12	6	0	-3.36131	-1.48477	0.775139	32	6	0	4.78163	0.092074	-0.077609
13	6	0	-1.91294	-1.16366	0.666698	33	6	0	5.30998	1.36543	0.158716
14	6	0	-1.04407	-2.18769	0.27329	34	6	0	5.67901	-0.906616	-0.469528
15	6	0	-1.40528	0.116985	0.90568	35	6	0	6.66031	1.63423	0.013795
16	6	0	0.308851	-1.94175	0.122795	36	6	0	7.03212	-0.65435	-0.619128
17	1	0	-1.45256	-3.17572	0.089802	37	6	0	7.52435	0.620533	-0.376503
18	6	0	-0.050233	0.370268	0.752844	38	9	0	4.50128	2.35751	0.534676
19	1	0	-2.06331	0.91447	1.2293	39	9	0	7.13536	2.8585	0.246275
20	6	0	0.819222	-0.655787	0.360874	40	9	0	8.82336	0.870681	-0.517928
						41	9	0	7.86441	-1.62652	-0.994148
						42	9	0	5.23221	-2.14048	-0.709095

3-7. Optimized geometry of 3a



SCF Done: E (RCAM-B3LYP) = -1757.67288894 hartree

Dipole moment (field-independent basis, Debye-Ang):

X = -4.5783, Y = -0.3793, Z = -1.3769, Tot = 4.7959

3-8. Cartesian coordinates of 3a

No.	Atom Type		Coordinates (Angstroms)								
	No.		x	y	z						
1	6	0	-8.1024	-0.641623	-0.29745	22	1	0	0.019385	0.652454	-2.11013
2	6	0	-7.12931	-1.33762	0.429992	23	6	0	2.08139	-0.331576	-0.796345
3	6	0	-5.90044	-0.757427	0.67426	24	6	0	3.15365	-1.07878	0.015177
4	6	0	-5.61212	0.530561	0.194632	25	6	0	4.54114	-0.523154	-0.029653
5	6	0	-6.59058	1.21297	-0.533514	26	6	0	5.65414	-1.35575	-0.146371
6	6	0	-7.82969	0.642788	-0.781878	27	6	0	4.77192	0.847111	0.063814
7	1	0	-7.36778	-2.33001	0.796913	28	6	0	6.94097	-0.840206	-0.182485
8	1	0	-5.16553	-1.30207	1.25561	29	6	0	6.04774	1.38207	0.043413
9	1	0	-6.36287	2.20704	-0.904306	30	6	0	7.13633	0.530059	-0.083177
10	1	0	-8.56937	1.19672	-1.34654	31	8	0	-4.03344	2.29066	-0.047694
11	6	0	-4.30934	1.18695	0.394667	32	8	0	-3.52549	0.180274	2.39061
12	6	0	-3.23841	0.501852	1.2536	33	8	0	2.89217	-2.13207	0.548802
13	6	0	-1.86361	0.318343	0.700827	34	8	0	2.40816	0.011586	-1.91504
14	6	0	-0.893374	-0.228165	1.54437	35	9	0	3.73844	1.68348	0.214222
15	6	0	-1.53696	0.634569	-0.622263	36	9	0	6.23839	2.69582	0.154553
16	6	0	0.391148	-0.459862	1.07781	37	9	0	8.36723	1.02785	-0.105861
17	1	0	-1.16568	-0.470715	2.56578	38	9	0	7.99014	-1.65078	-0.313828
18	6	0	-0.252552	0.404761	-1.08986	39	9	0	5.51022	-2.67207	-0.268834
19	1	0	-2.27685	1.07672	-1.27853	40	8	0	-9.2733	-1.28958	-0.481151
20	6	0	0.717376	-0.143974	-0.245947	41	6	0	-10.3126	-0.642756	-1.19633
21	1	0	1.1301	-0.896578	1.73862	42	1	0	-10.0064	-0.422147	-2.22499
						43	1	0	-11.1454	-1.34526	-1.20794
						44	1	0	-10.6206	0.281333	-0.694747