



Supporting Information

for

Synthesis of new fluorescent molecules having an aggregation-induced emission property derived from 4-fluoroisoxazoles

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Beilstein J. Org. Chem. **2020**, *16*, 1411–1417. doi:10.3762/bjoc.16.117

General procedures and analytical data, including copies of ^1H NMR, ^{13}C NMR and ^{19}F NMR spectra

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General Information:

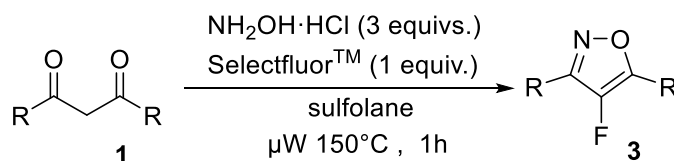
^1H NMR and ^{13}C NMR spectra were recorded on JEOL JNM-ECZS-400 and ECA-600SN spectrometers. ^{19}F NMR spectra were recorded on JEOL JNM-ECZS-400 spectrometer. Chemical shifts of ^1H NMR are reported in ppm from tetramethylsilane (TMS: 0 ppm) as an internal standard. Chemical shifts of ^{13}C NMR are reported in ppm from tetramethylsilane (TMS: 0 ppm) or from the solvent (DMSO- d_6 : 39.520 ppm) as an internal standard. Chemical shifts of ^{19}F NMR are reported in ppm from trichlorofluoromethane (CFCl_3 : 0 ppm) as an internal standard. All data are reported as follows: chemical shifts, relative integration value, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, br = broad, m = multiplet), coupling constants (Hz). Mass spectra were obtained on JEOL JMS-700T spectrometers. Melting points were measured on Yanagimoto micro melting point apparatus MP-S3. UV-vis absorption spectra were measured on a Jasco V-650 spectrometer. Fluorescence spectra were measured on a Jasco FP-8600 spectrofluorometer and appropriate blank spectra were subtracted. Microwave reactions were performed in microwave tubes with clip lids using Biotage Initiator+ microwave reactor.

Materials:

Solvents (sulfolane, CH_3CN and EtOH) were distilled under inert argon atmosphere or under reduced pressure before use. Tetrahydrofuran (THF) was purchased from Kanto Chemical Co. Inc. as "Dehydrated". All commercially available reagents were used without further purification. All experiments were carried out under argon atmosphere unless otherwise noted.

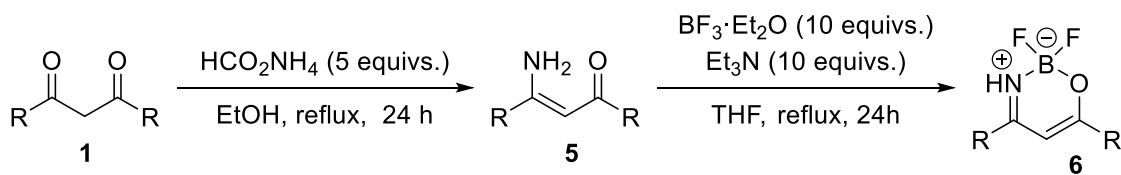
Experimental section:

General procedure for the synthesis of 4-fluoroisoxazoles **3** [1].



1,3-Diketone (**1**; 1 mmol), hydroxylamine hydrochloride (3 mmol) and SelectfluorTM (1 mmol) were added to a microwave vial and suspended in sulfolane (4 mL). The vial was sealed and heated by microwave irradiation for 1 hour at 150 °C. The resulting mixture was quenched with NaHCO_3 and extracted with AcOEt. The AcOEt layer was washed with sat. NaCl and dried over MgSO_4 . The solvent was removed in vacuo and the residue purified by column chromatography to give the 4-fluoroisoxazoles **3**.

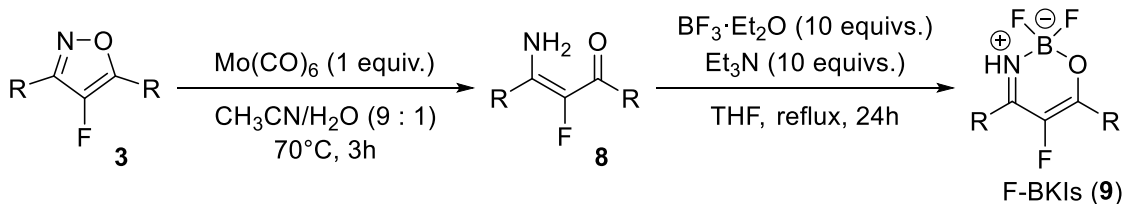
General procedure for the synthesis of boron ketoiminates **6**.



The mixture of 1,3-diketone (**3**, 1 mmol) and ammonium formate (5 mmol) in ethanol (5.5 mL) was stirred under reflux condition for 24 hours, and then the ethanol was distilled off under reduced pressure. To the residue was added AcOEt and water, and extracted. The AcOEt layer was dried over MgSO_4 and the solvent was removed in vacuo. The residue was passed through a silica gel column to give the enaminoketone (**5**), the formation of which was checked by NMR.

To a solution of the enaminoketone (**5**, 1 mmol) in THF (14 mL) was added $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (10 mmol) and triethylamine (10 mmol), and the mixture was heated under reflux for 24 hours. The resulting mixture was quenched with H_2O and extracted with AcOEt. The AcOEt layer was washed with sat. NaCl and dried over MgSO_4 . The solvent was removed in vacuo and the residue was purified by column chromatography to give the boron ketoiminates (BKIs, **6**).

General procedure for the synthesis of α -fluorinated boron ketoiminates (F-BKIs, **9**).

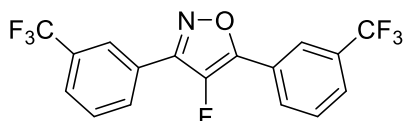


The solution of the appropriate 4-fluoroisoxazole **3** (1 mmol) in $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ 9:1 (27 mL) was stirred for 15 min at room temperature. Then, $\text{Mo}(\text{CO})_6$ (1.1 mmol) was added into the solution, and the mixture was heated at 70°C for 3 hours. The resulting mixture was quenched with sat. NaCl and extracted with AcOEt. The AcOEt layer was dried over MgSO_4 and the solvent was removed in vacuo. The residue was passed through a silica gel column for separation of the corresponding defluorinated enaminoketone **5** and the α -monofluoro-1,3-diketone to give the desired α -fluorinated enaminoketone **8**, which was checked by NMR for the formation.

To a solution of the α -fluorinated enaminoketone **8** (1 mmol) in THF (14 mL) was added $\text{BF}_3 \cdot \text{Et}_2\text{O}$ (10 mmol) and triethylamine (10 mmol), and the mixture was heated under reflux condition for 24 hours. The resulting mixture was quenched with H_2O and extracted with AcOEt. The AcOEt layer was washed with sat. NaCl and dried over MgSO_4 . The solvent was removed in vacuo and the residue was purified by column chromatography to give the α -fluorinated boron ketoiminate (F-BKIs, **9**).

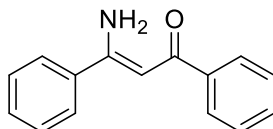
Spectroscopic data:

4-Fluoro-3,5-bis(3-(trifluoromethyl)phenyl)isoxazole (3d)



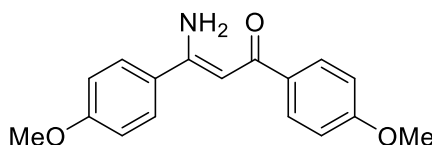
Colorless solid; M.p. 96.5–97.5 °C; ^1H NMR (400 MHz, CDCl_3) δ : 7.66–7.81 (4H, m), 8.06–8.23 (4H, m); ^{13}C NMR (100 MHz, CDCl_3) δ : 122.11 (m), 123.59 (q, $J = 272.6$ Hz), 123.69 (q, $J = 272.6$ Hz), 123.92 (m), 126.25 (d, $J = 5.2$ Hz), 126.96 (m), 127.03 (d, $J = 4.0$ Hz), 127.39 (m), 129.33 (d, $J = 4.9$ Hz), 129.74, 129.87, 130.20, 131.69 (q, $J = 32.8$ Hz), 131.83 (q, $J = 33.0$ Hz), 141.92 (d, $J = 261.0$ Hz), 152.28, (d, $J = 10.5$ Hz), 152.70 (d, $J = 19.1$ Hz); ^{19}F NMR (376 MHz, CDCl_3) δ : -62.78 (3F, s), -62.89 (3F, s), -175.83 (1F, s); MS m/z : 375 (M^+); HRMS Calcd for $\text{C}_{17}\text{H}_8\text{F}_7\text{NO}$: 375.0494 (M^+), Found: 375.0497.

3-Amino-1,3-diphenylprop-2-en-1-one (5a) [2]



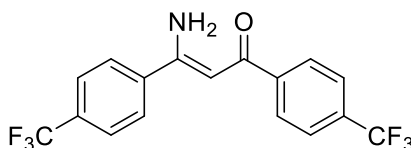
^1H NMR (400 MHz, CDCl_3) δ : 5.45 (1H, bs), 6.15 (1H, s), 7.41–7.53 (6H, m), 7.63–7.66 (2H, m), 7.94–7.97 (2H, m), 10.42 (1H, bs).

3-Amino-1,3-bis(4-methoxyphenyl)prop-2-en-1-one (5b) [2]



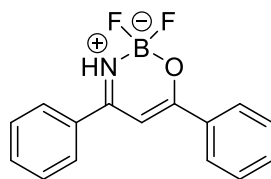
^1H NMR (400 MHz, CDCl_3) δ : 3.86 (3H, s), 3.87 (3H, s), 5.33 (1H, bs), 6.10 (1H, s), 6.92–7.00 (4H, m), 7.58–7.61 (2H, m), 7.92–7.96 (2H, m), 10.41 (1H, bs).

3-Amino-1,3-bis(4-(trifluoromethyl)phenyl)prop-2-en-1-one (5c) [2]



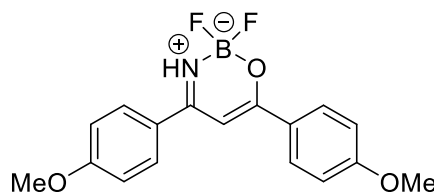
^1H NMR (400 MHz, CDCl_3) δ : 5.55 (1H, bs), 6.11 (1H, s), 7.69–7.71 (2H, m), 7.76 (4H, s), 8.01–8.03 (2H, m), 10.42 (1H, bs); ^{19}F NMR (376 MHz, CDCl_3) δ : -62.70 (3F, s), -62.80 (3F, s).

2,2-Difluoro-4,6-diphenyl-2H-1,3,2λ⁴-oxazaborinine (6a) [3]



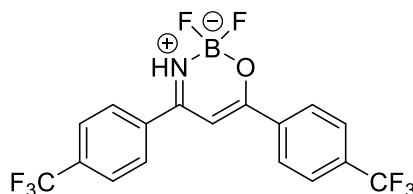
Colorless solid; M.p. 204.5–206.5 °C; ¹H NMR (400 MHz, DMSO-d₆) δ: 6.93 (1H, d, *J* = 1.8 Hz), 7.53–7.73 (6H, m), 8.04–8.14 (4H, m), 10.33 (1H, bs); ¹³C NMR (100 MHz, DMSO-d₆) δ: 92.08, 127.17, 127.92, 128.73, 128.93, 132.30, 132.77, 132.91, 133.33, 169.40, 170.70; ¹⁹F NMR (376 MHz, DMSO-d₆) δ: -128.03–128.15 (2F, m); MS *m/z*: 271 (M⁺); HRMS Calcd for C₁₅H₁₂BF₂NO: 271.0980 (M⁺), Found: 271.0982.

2,2-Difluoro-4,6-bis(4-methoxyphenyl)-2H-1,3,2λ⁴-oxazaborinine (6b) [4]



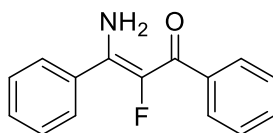
Pale yellow solid; M.p. 220.0–223.0 °C; ¹H NMR (400 MHz, DMSO-d₆) δ: 3.86 (3H, s), 3.88 (3H, s), 6.82 (1H, d, *J* = 1.3 Hz), 7.07–7.14 (4H, m), 8.08–8.10 (4H, m), 9.85 (1H, bs); ¹³C NMR (100 MHz, DMSO-d₆) δ: 55.42, 55.55, 90.20, 114.11, 114.31, 124.71, 125.78, 129.18, 129.97, 162.55, 163.05, 167.56, 170.02; ¹⁹F NMR (376 MHz, DMSO-d₆) δ: -128.87–128.95 (2F, m); MS *m/z*: 331 (M⁺); HRMS Calcd for C₁₇H₁₆BF₂NO₃: 331.1191 (M⁺), Found: 331.1193.

2,2-Difluoro-4,6-bis(4-(trifluoromethyl)phenyl)-2H-1,3,2λ⁴-oxazaborinine (6c)



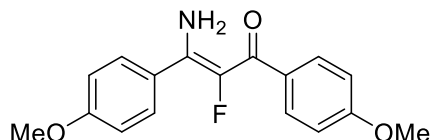
Pale yellow solid; M.p. 265.0–270.0 °C; ¹H NMR (400 MHz, DMSO-d₆) δ: 7.10 (1H, d, *J* = 1.4 Hz), 7.91–8.00 (4H, m), 8.23–8.35 (4H, m), 10.82 (1H, bs); ¹³C NMR (100 MHz, DMSO-d₆) δ: 93.86, 123.58 (q, *J* = 272.6 Hz), 123.72 (q, *J* = 272.6 Hz), 125.67 (q, *J* = 4.1 Hz), 125.75 (q, *J* = 4.1 Hz), 128.05, 129.12, 132.06 (q, *J* = 31.9 Hz), 132.38 (q, *J* = 31.9 Hz), 136.46, 136.94, 169.01, 169.26; ¹⁹F NMR (376 MHz, DMSO-d₆) δ: -61.44 (3F, s), -61.55 (3F, s), -127.35–127.42 (2F, m); MS *m/z*: 407 (M⁺); HRMS Calcd for C₁₇H₁₀BF₈NO: 407.0728 (M⁺), Found: 407.0724.

3-Amino-2-fluoro-1,3-diphenylprop-2-en-1-one (8a)



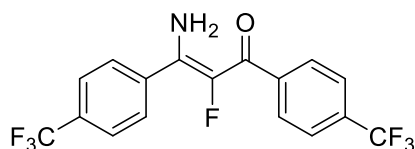
^1H NMR (400 MHz, CDCl_3) δ : 7.42–7.51 (6H, m), 7.63–7.65 (2H, m), 7.89–7.92 (2H, m); ^{19}F NMR (376 MHz, CDCl_3) δ : -170.44 (1F, s).

3-Amino-2-fluoro-1,3-bis(4-methoxyphenyl)prop-2-en-1-one (8b)



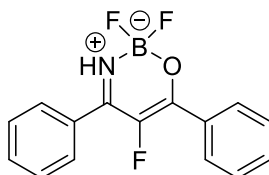
^1H NMR (400 MHz, CDCl_3) δ : 3.86 (3H, s), 3.86 (3H, s), 6.92–7.01 (4H, m), 7.58–7.62 (2H, m), 7.93–7.97 (2H, m); ^{19}F NMR (376 MHz, CDCl_3) δ : -169.97 (1F, s).

3-Amino-2-fluoro-1,3-bis(4-(trifluoromethyl)phenyl)prop-2-en-1-one (8c)



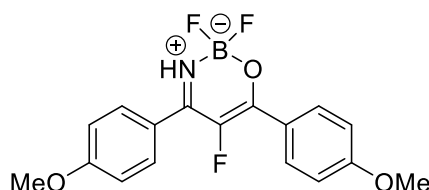
^1H NMR (400 MHz, CDCl_3) δ : 7.70–7.73 (2H, m), 7.77 (4H, s), 7.99–8.01 (2H, m); ^{19}F NMR (376 MHz, CDCl_3) δ : -62.83 (3F, s), -62.86 (3F, s), -170.60 (1F, s).

2,2,5-Trifluoro-4,6-diphenyl-2H-1,3,2λ⁴-oxazaborinine (9a)



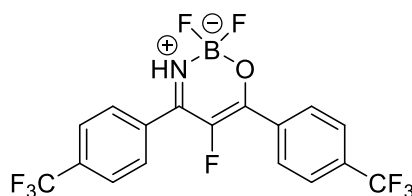
Pale yellow solid; M.p. 172.5–174.0 °C; ^1H NMR (400 MHz, DMSO-d_6) δ : 7.57–7.73 (6H, m), 7.79–7.81 (2H, m), 7.96–7.98 (2H, m), 10.80 (1H, bs); ^{13}C NMR (100 MHz, DMSO-d_6) δ : 128.49 (d, J = 8.3 Hz), 128.63, 128.75, 129.05 (d, J = 3.8 Hz), 129.34, 130.79 (d, J = 5.7 Hz), 132.16, 132.61, 139.59 (d, J = 221.5 Hz), 157.0 (d, J = 20.8 Hz), 165.24 (d, J = 26.1 Hz); ^{19}F NMR (376 MHz, DMSO-d_6) δ : -129.38–-129.50 (2F, m), -171.41 (1F, s); MS m/z : 289 (M^+); HRMS Calcd for $\text{C}_{15}\text{H}_{11}\text{BF}_3\text{NO}$: 289.0886 (M^+), Found: 289.0888.

2,2,5-Trifluoro-4,6-bis(4-methoxyphenyl)-2*H*-1,3,2λ⁴-oxazaborinine (9b)



Yellow solid; M.p. 188.0–189.0 °C; ^1H NMR (400 MHz, DMSO- d_6) δ : 3.87 (3H, s), 3.88 (3H, s), 7.11–7.17 (4H, m), 7.78–7.81 (2H, m), 7.95–7.97 (2H, m), 10.28 (1H, bs); ^{13}C NMR (100 MHz, DMSO- d_6) δ : 55.43, 55.54, 114.15, 114.26, 121.50, 123.12 (d, $J = 5.8$ Hz), 130.66 (d, $J = 9.1$ Hz), 131.38 (d, $J = 4.7$ Hz), 139.28 (d, $J = 220.9$ Hz), 156.68 (d, $J = 20.8$ Hz), 162.28, 162.81, 163.42 (d, $J = 25.1$ Hz); ^{19}F NMR (376 MHz, DMSO- d_6) δ : -130.66–130.73 (2F, m), -171.94 (1F, s); MS m/z : 349 (M^+); HRMS Calcd for $\text{C}_{17}\text{H}_{15}\text{BF}_3\text{NO}_3$: 349.1097 (M^+), Found: 349.1102.

2,2,5-Trifluoro-4,6-bis(4-(trifluoromethyl)phenyl)-2*H*-1,3,2λ⁴-oxazaborinine (9c)



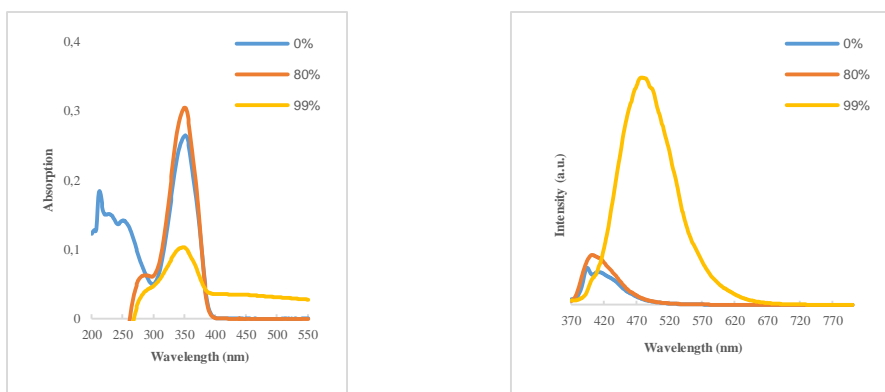
Pale yellow solid; M.p. 224.0–226.0 °C; ^1H NMR (400 MHz, DMSO- d_6) δ : 7.95–8.05 (6H, m), 8.15–8.17 (2H, m), 11.29 (1H, bs); ^{13}C NMR (100 MHz, DMSO- d_6) δ : 123.52 (q, $J = 272.8$ Hz), 123.61 (q, $J = 272.5$ Hz), 125.49 (q, $J = 3.8$ Hz), 125.69 (d, $J = 3.8$ Hz), 129.31 (d, $J = 8.3$ Hz), 130.16 (d, $J = 3.4$ Hz), 131.64 (q, $J = 32.6$ Hz), 132.28 (q, $J = 32.4$ Hz), 132.91, 134.43 (d, $J = 5.3$ Hz), 140.08 (d, $J = 225.4$ Hz), 155.22 (d, $J = 20.7$ Hz), 164.82 (d, $J = 26.9$ Hz); ^{19}F NMR (376 MHz, DMSO- d_6) δ : -61.60 (3F, s), -61.63 (3F, s), -128.26–128.34 (2F, m), -170.12 (1F, s); MS m/z : 425 (M^+); HRMS Calcd for $\text{C}_{17}\text{H}_9\text{BF}_9\text{NO}$: 425.0633 (M^+), Found: 425.0632.

References:

- [1] Sato, K.; Sandford, G.; Shimizu, K.; Akiyama, S.; Lancashire, M. J.; Yufit, D. S.; Tarui, A.; Omote, M.; Kumadaki, I.; Harusawa, S.; Ando, A. *Tetrahedron* **2016**, 72, 1690–1698.
- [2] H. S. P. Rao, N. Muthanna, *Eur. J. Org. Chem.* **2015**, 1525–1532.
- [3] I. Bally, E. Ciornei, A. Vasilescu, A. T. Balaban, *Tetrahedron* **1973**, 29, 3185–3187.
- [4] R. Yoshii, A. Nagai, K. Tanaka, Y. Chujo, *Chem. -Eur. J.* **2013**, 19, 4506–4512.

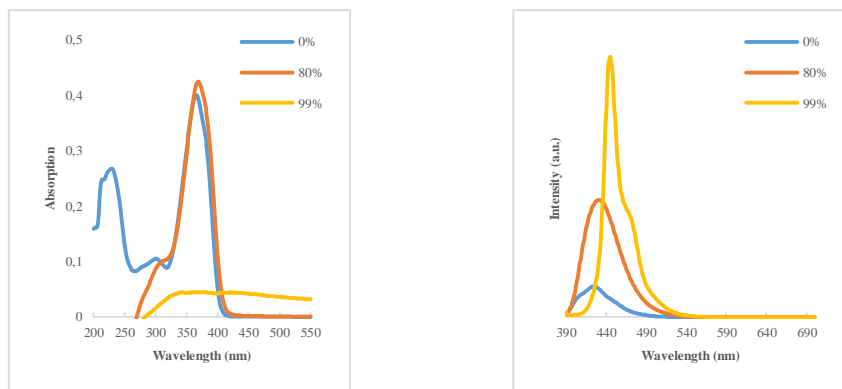
UV-vis and FL spectra:

2,2-Difluoro-4,6-diphenyl-2*H*-1,3,2λ⁴-oxazaborinine (**6a**)



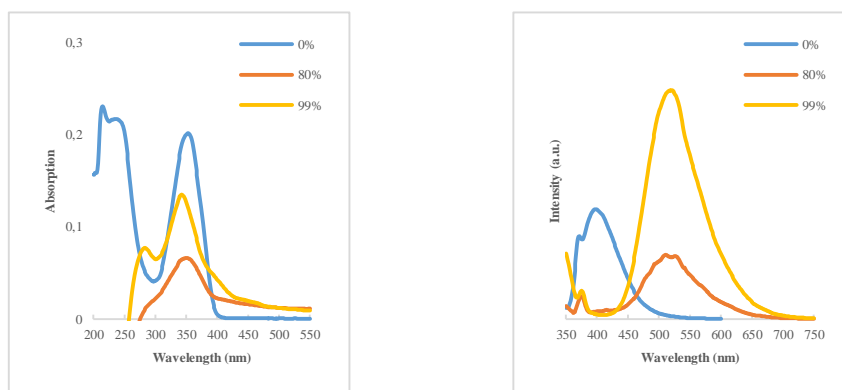
UV/Vis and FL spectra of **6a** with solvent compositions of the THF/H₂O mixture upon excitation at 352 nm (1.0×10^{-5} M).

2,2-Difluoro-4,6-bis(4-methoxyphenyl)-2*H*-1,3,2λ⁴-oxazaborinine (**6b**)



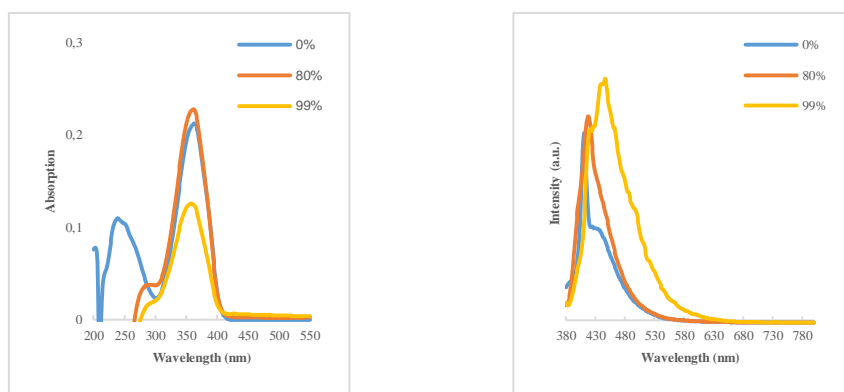
UV/Vis and FL spectra of **6b** with solvent compositions of the THF/H₂O mixture upon excitation at 365 nm (1.0×10^{-5} M).

2,2-Difluoro-4,6-bis(4-(trifluoromethyl)phenyl)-2*H*-1,3,2λ⁴-oxazaborinine (**6c**)



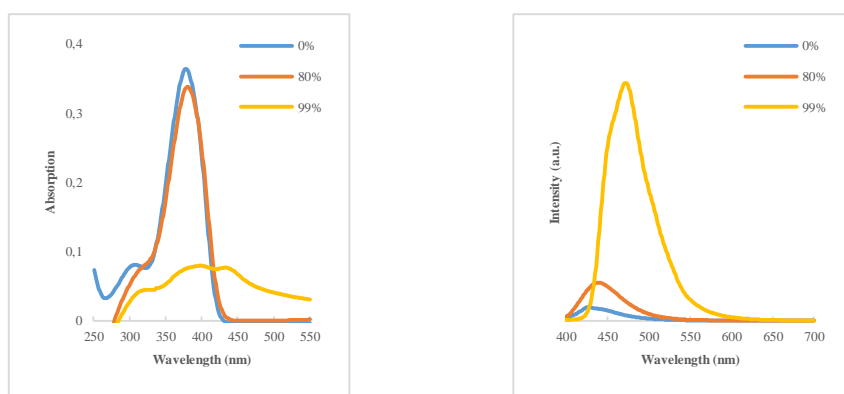
UV/Vis and FL spectra of **6c** with solvent compositions of the THF/H₂O mixture upon excitation at 332 nm (1.0×10^{-5} M).

2,2,5-Trifluoro-4,6-diphenyl-2*H*-1,3,2λ⁴-oxazaborinine (**9a**)



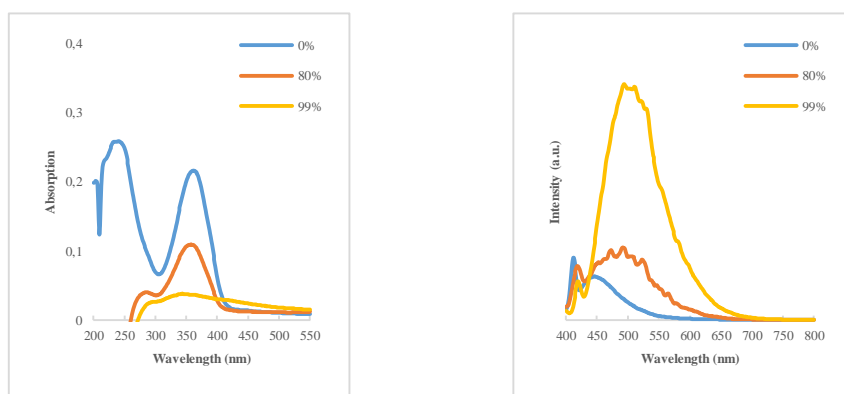
UV/Vis and FL spectra of **9a** with solvent compositions of the THF/H₂O mixture upon excitation at 365 nm (1.0×10^{-5} M).

2,2,5-Trifluoro-4,6-bis(4-methoxyphenyl)-2*H*-1,3,2λ⁴-oxazaborinine (**9b**)



UV/Vis and FL spectra of **9b** with solvent compositions of the THF/H₂O mixture upon excitation at 380 nm (1.0×10^{-5} M).

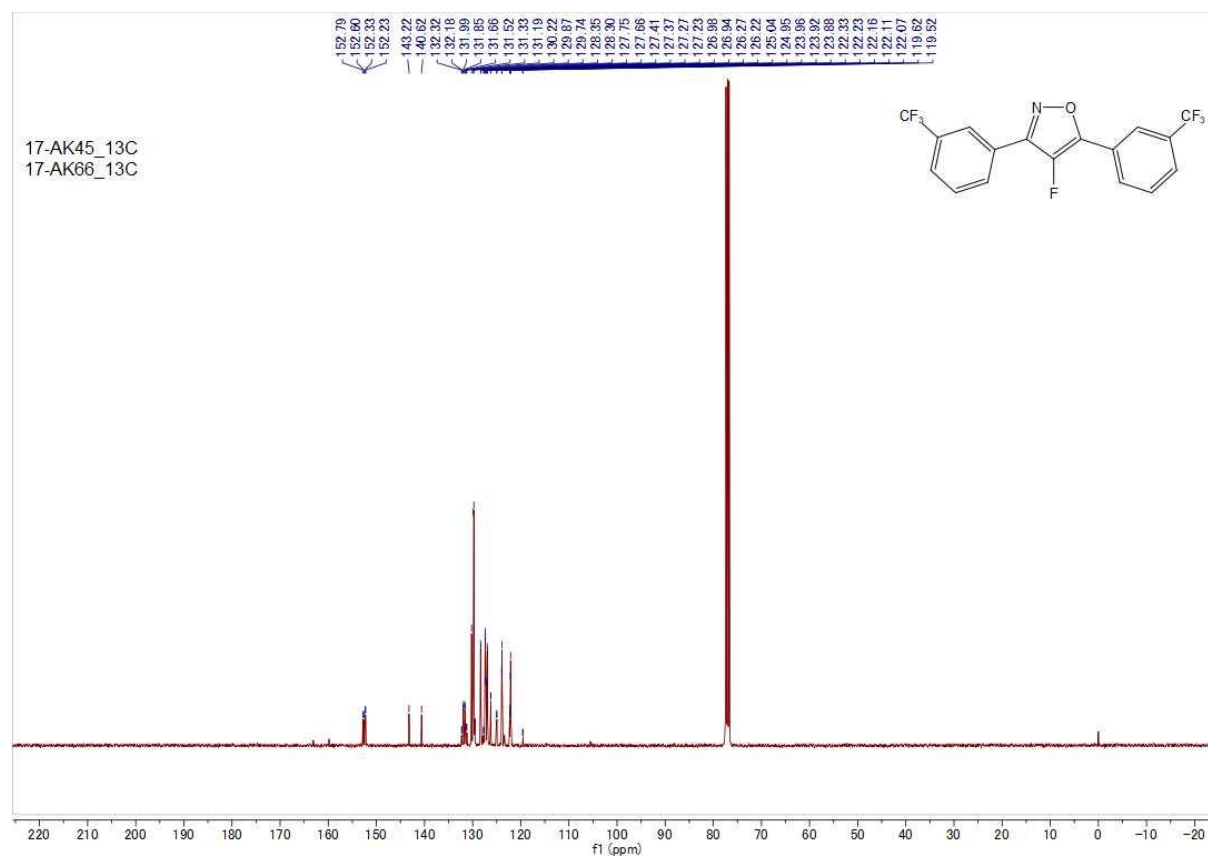
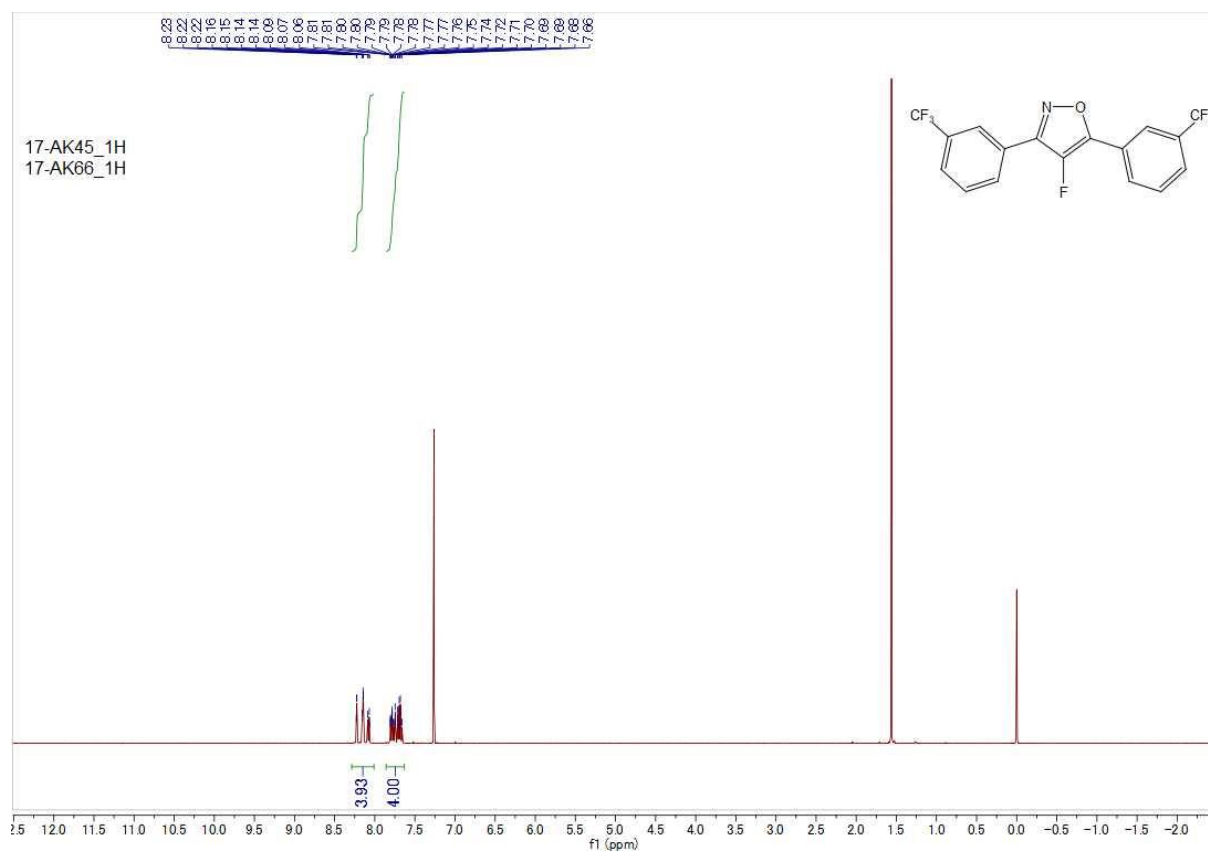
2,2,5-Trifluoro-4,6-bis(4-(trifluoromethyl)phenyl)-2*H*-1,3,2λ⁴-oxazaborinine (**9c**)

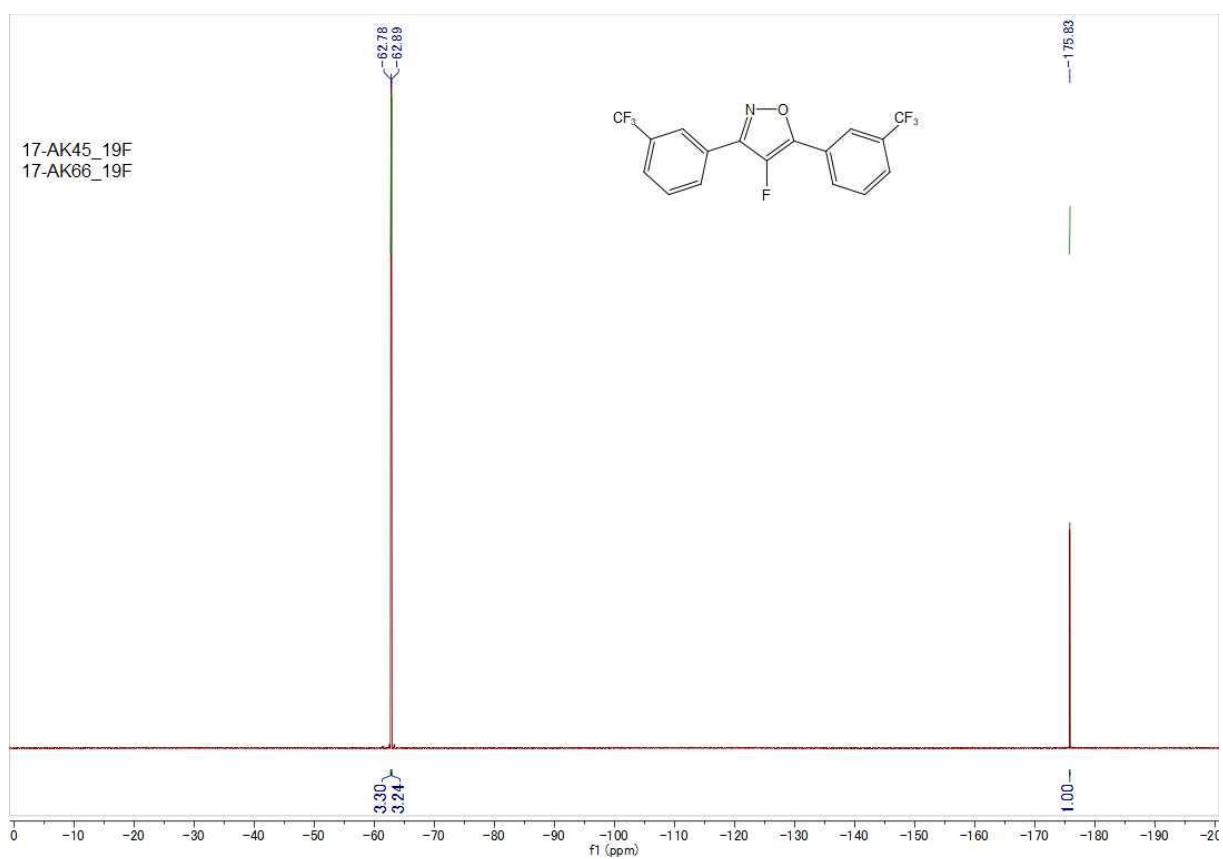


UV/Vis and FL spectra of **9c** with solvent compositions of the THF/H₂O mixture upon excitation at 366 nm (1.0×10^{-5} M).

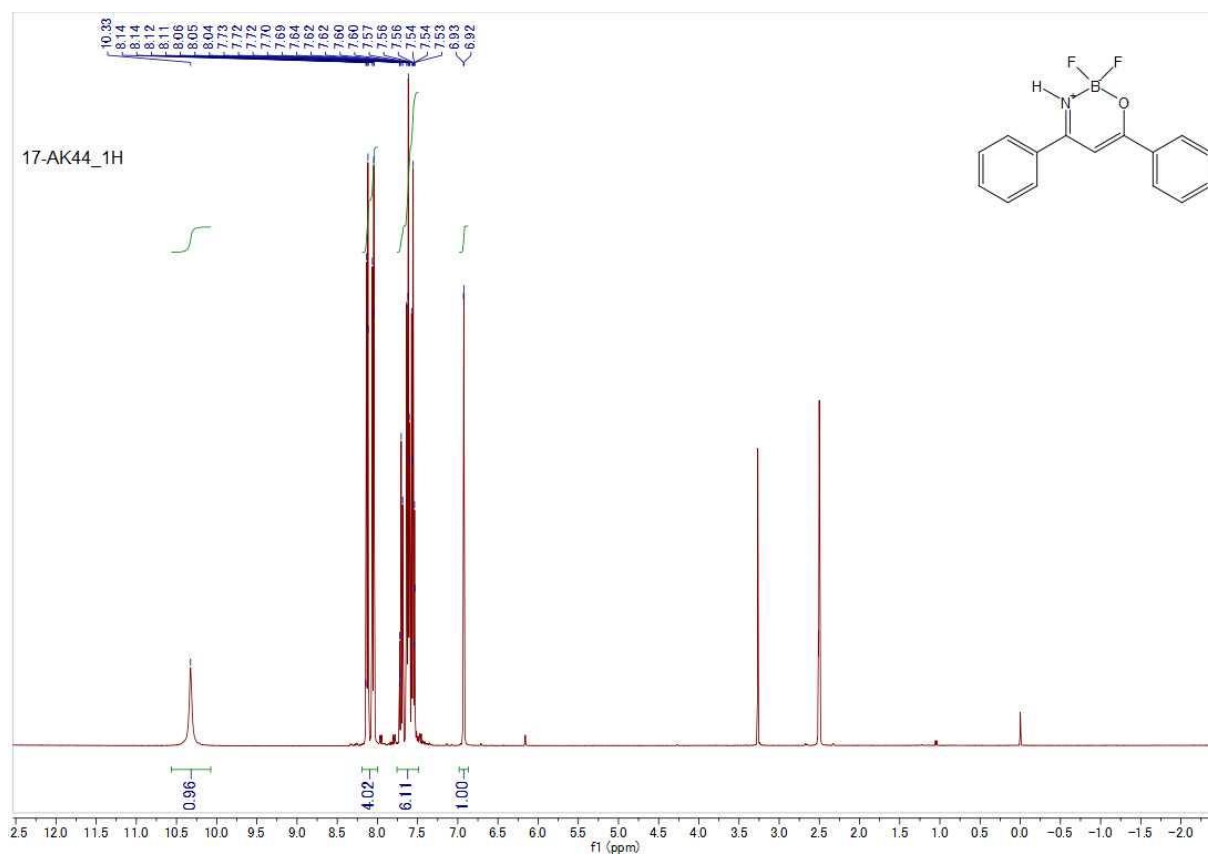
NMR spectra:

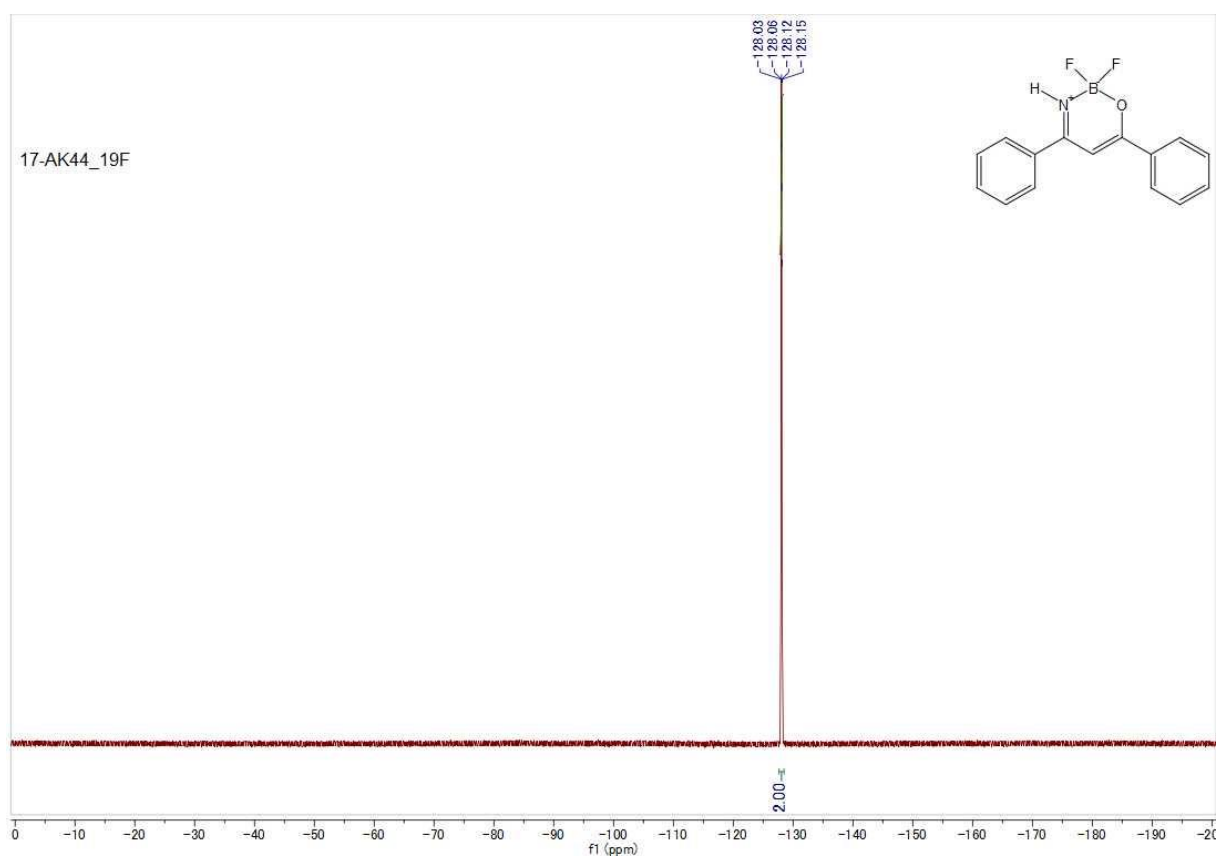
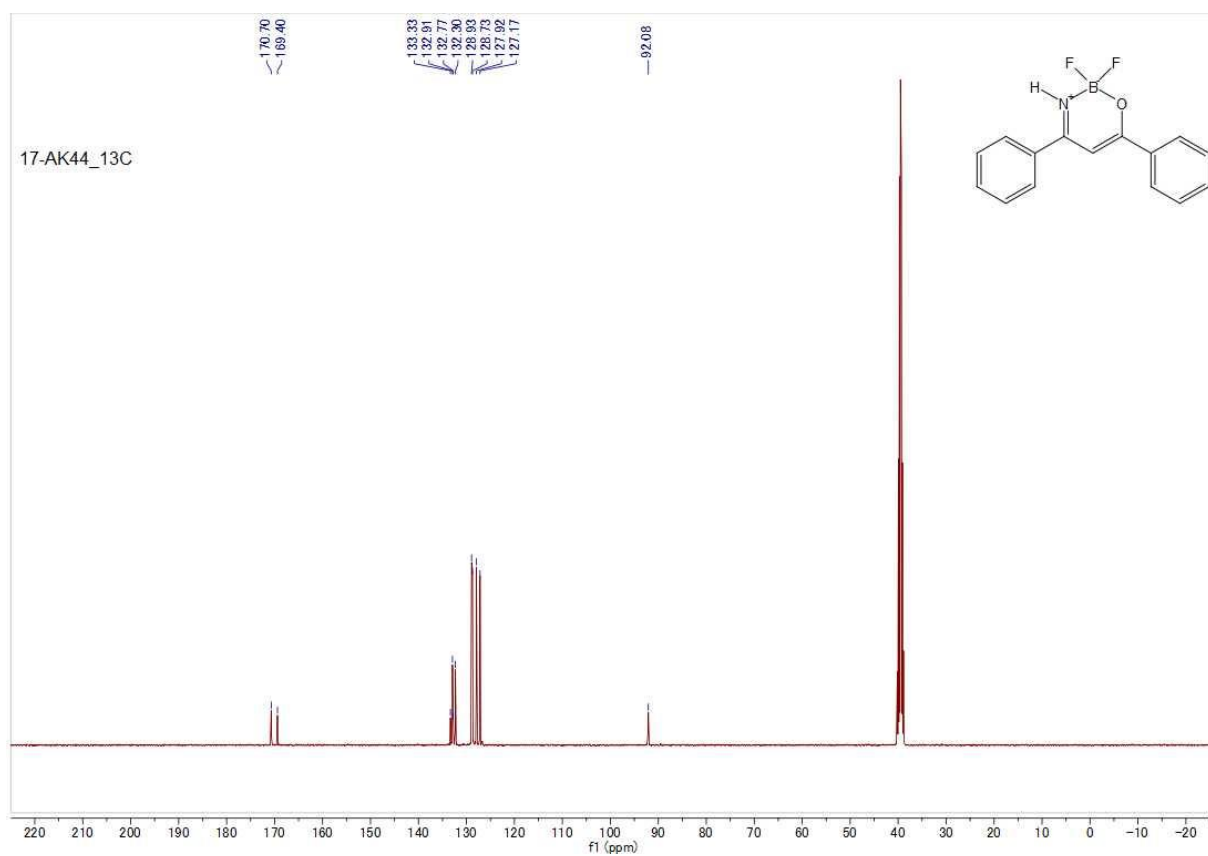
4-Fluoro-3,5-bis(3-(trifluoromethyl)phenyl)isoxazole (3d)



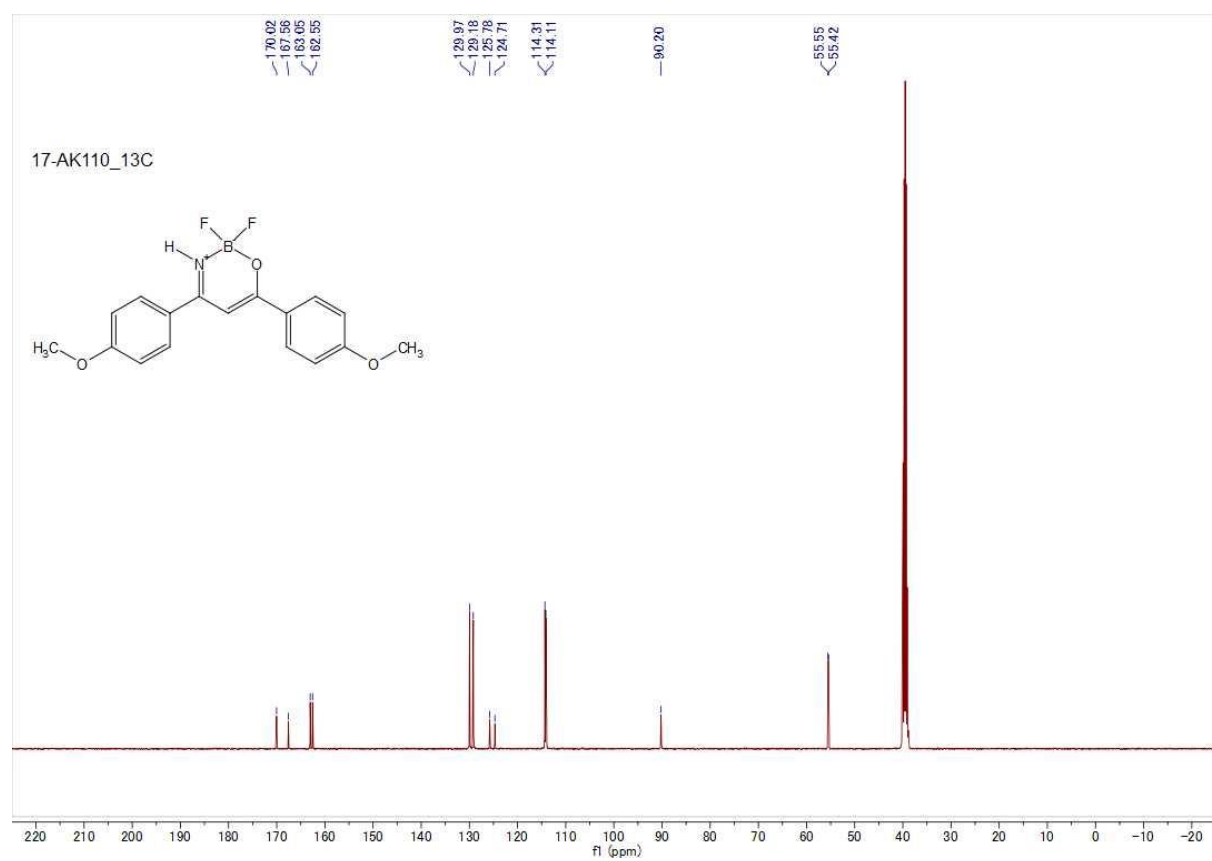
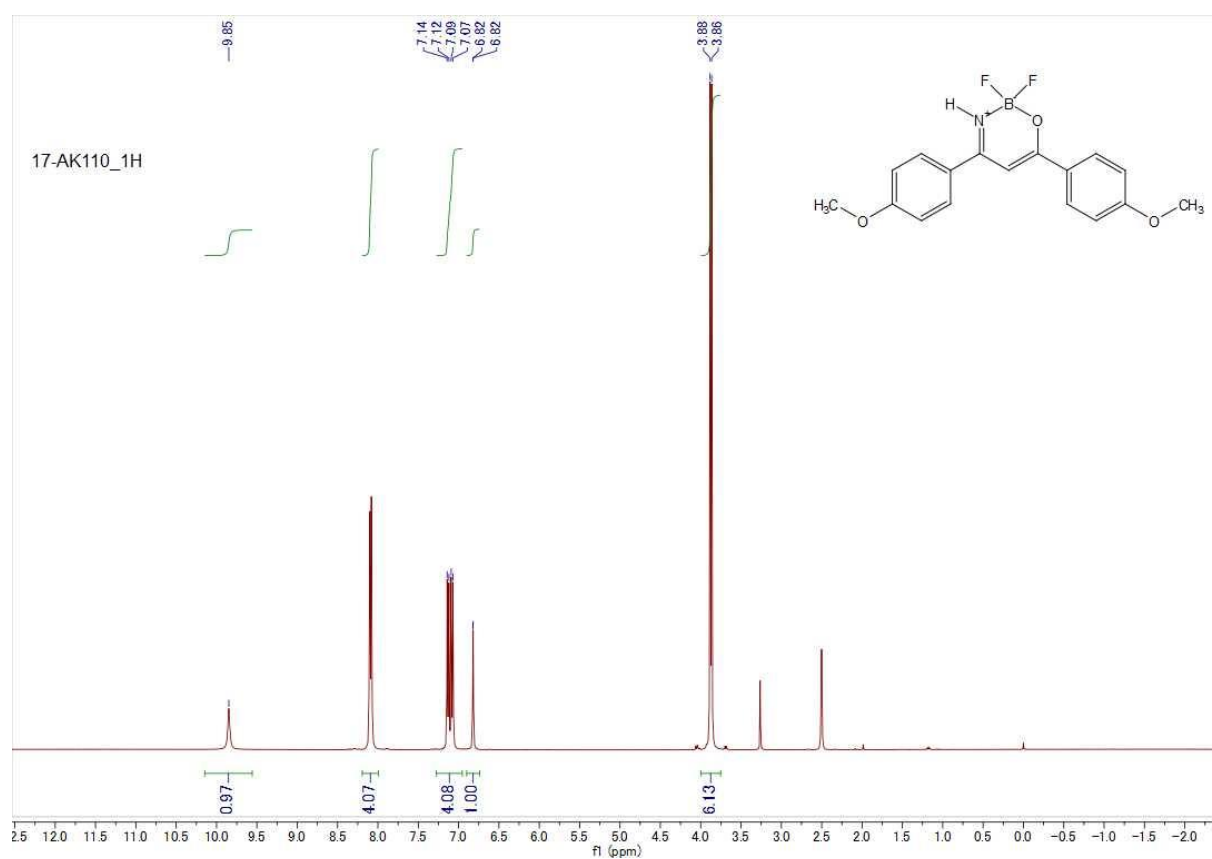


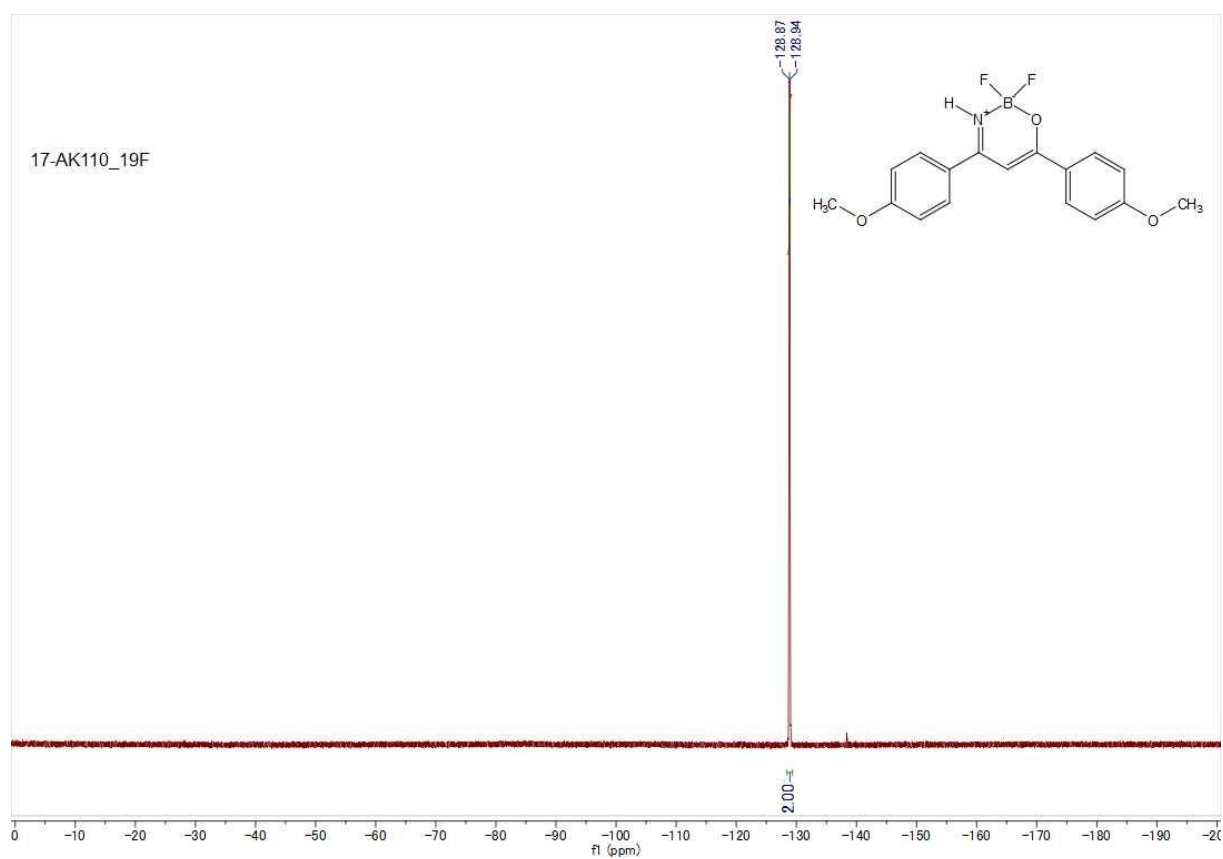
2,2-Difluoro-4,6-diphenyl-2H-1,3,2λ⁴-oxazaborinine (6a)



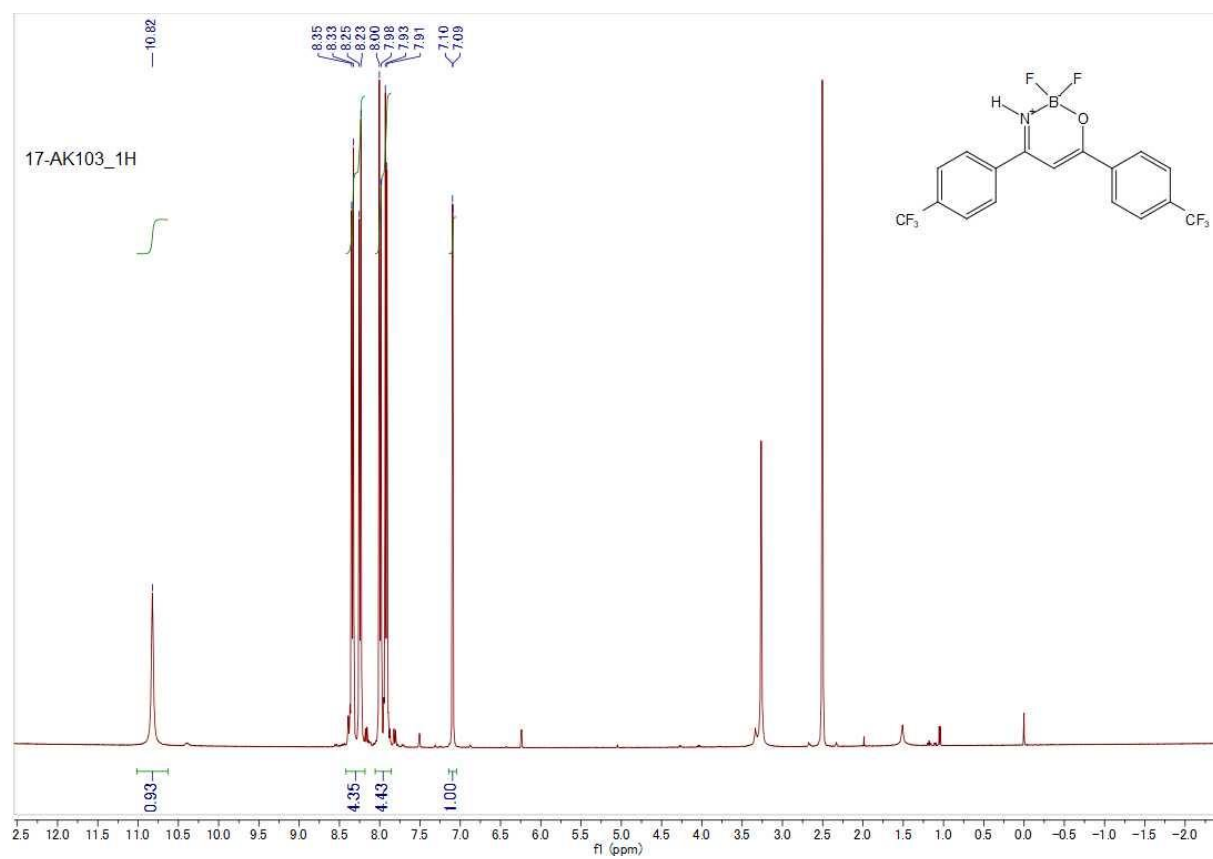


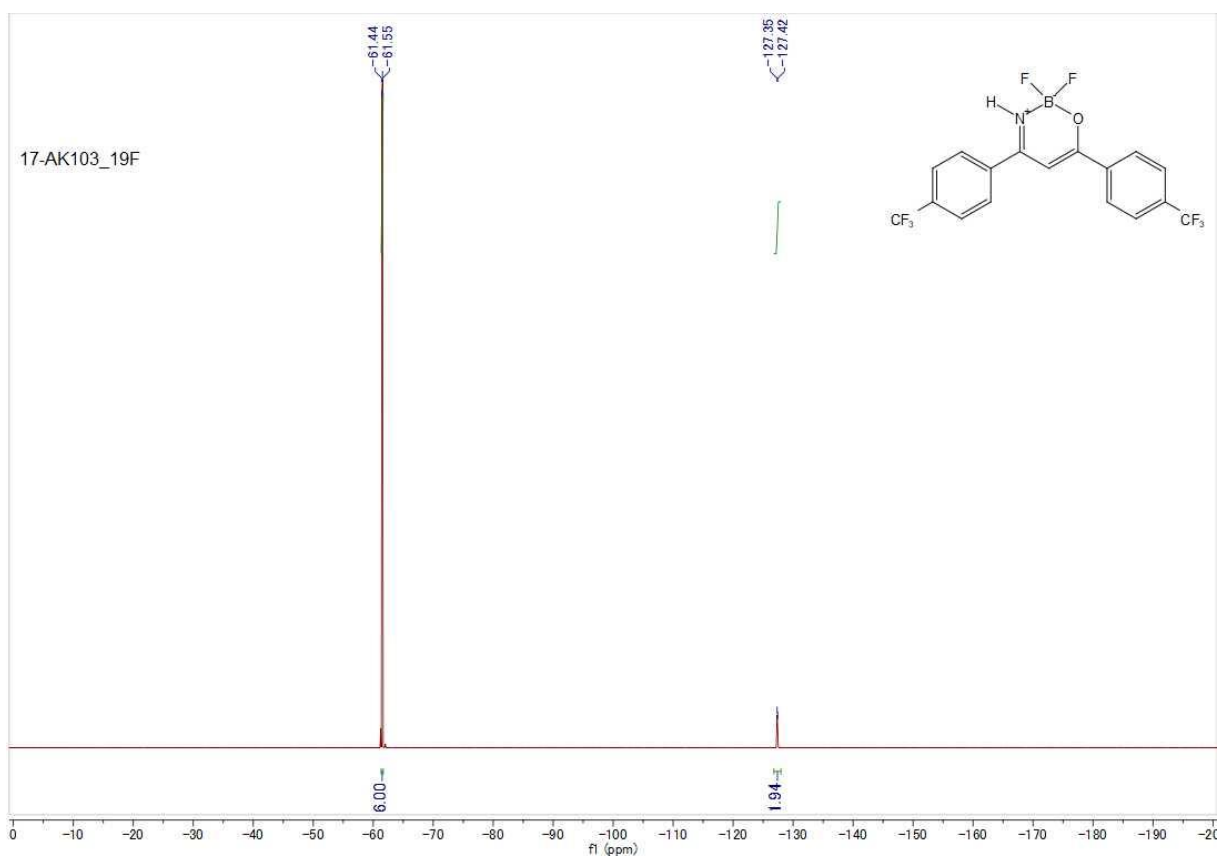
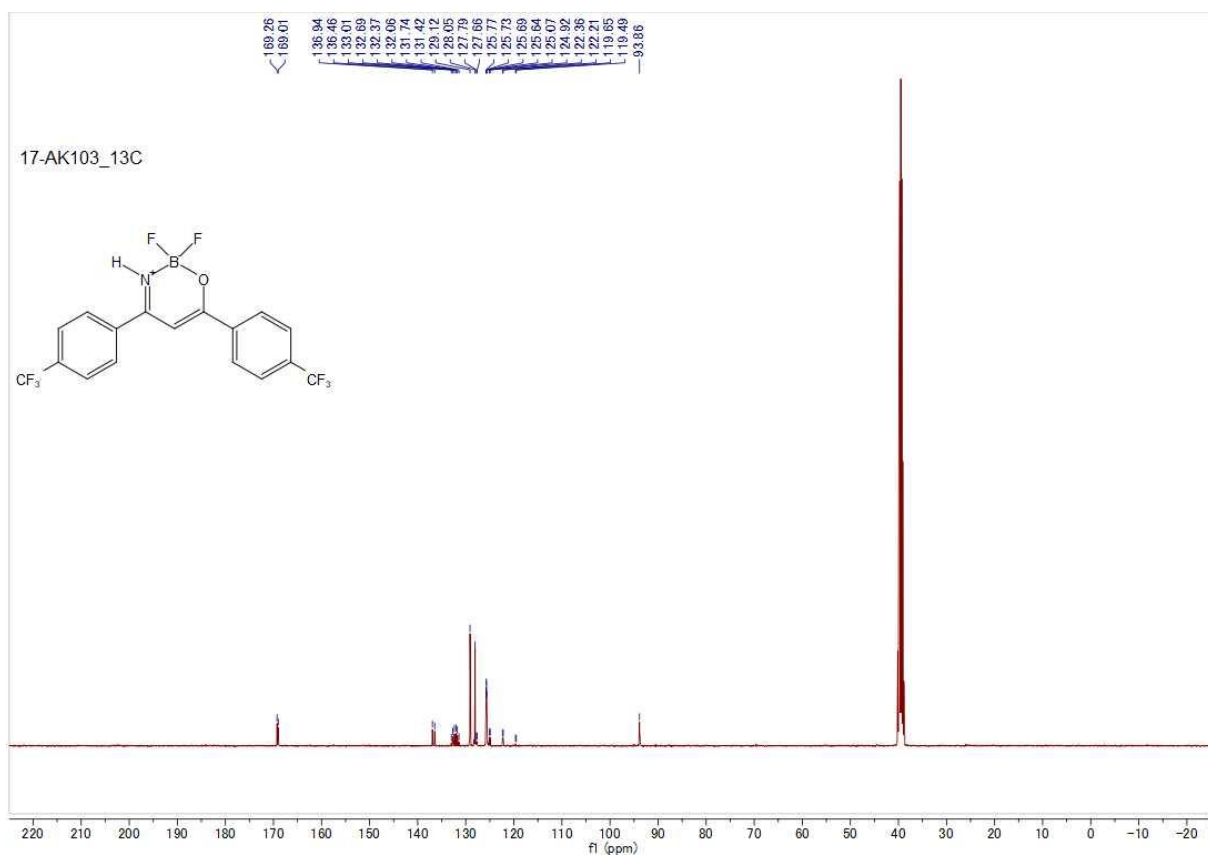
2,2-Difluoro-4,6-bis(4-methoxyphenyl)-2*H*-1,3,2λ⁴-oxazaborinine (**6b**)



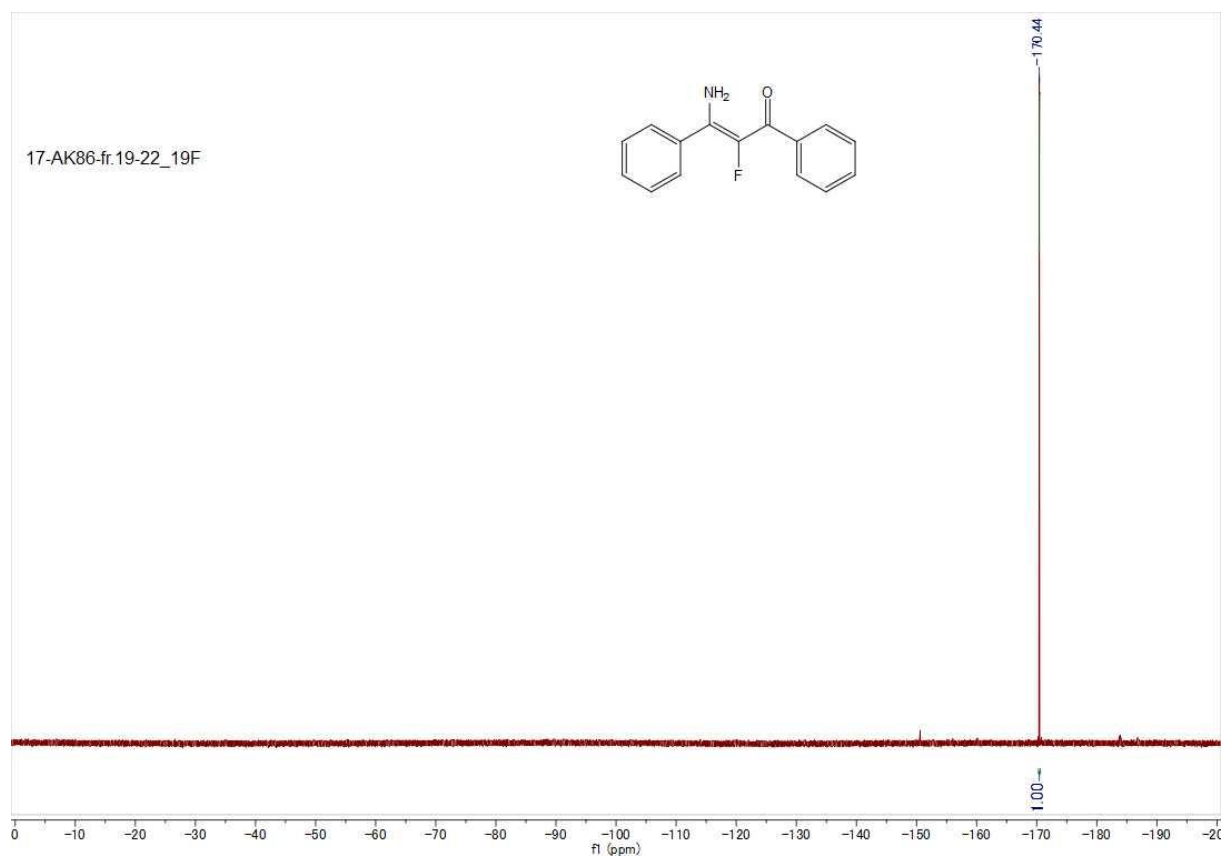
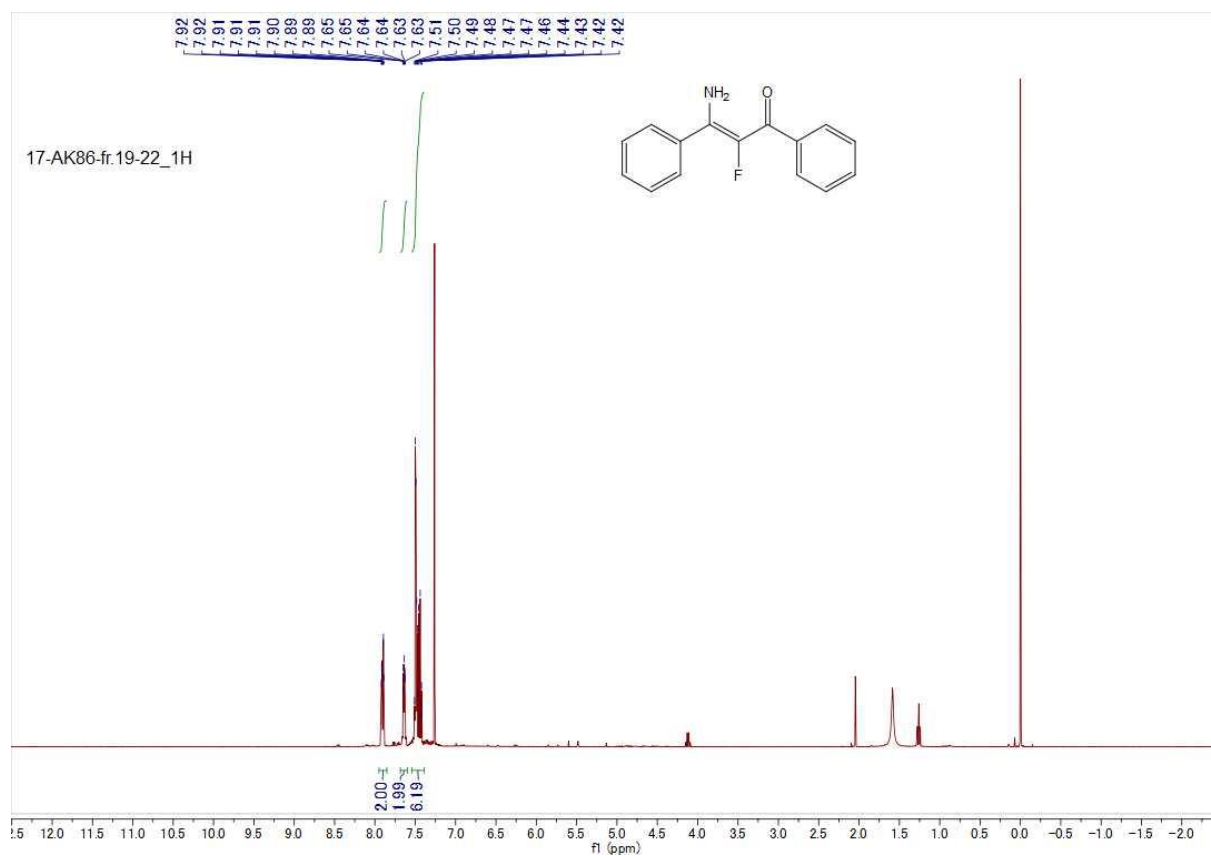


2,2-Difluoro-4,6-bis(4-(trifluoromethyl)phenyl)-2*H*-1,3,2 λ^4 -oxazaborinine (**6c**)



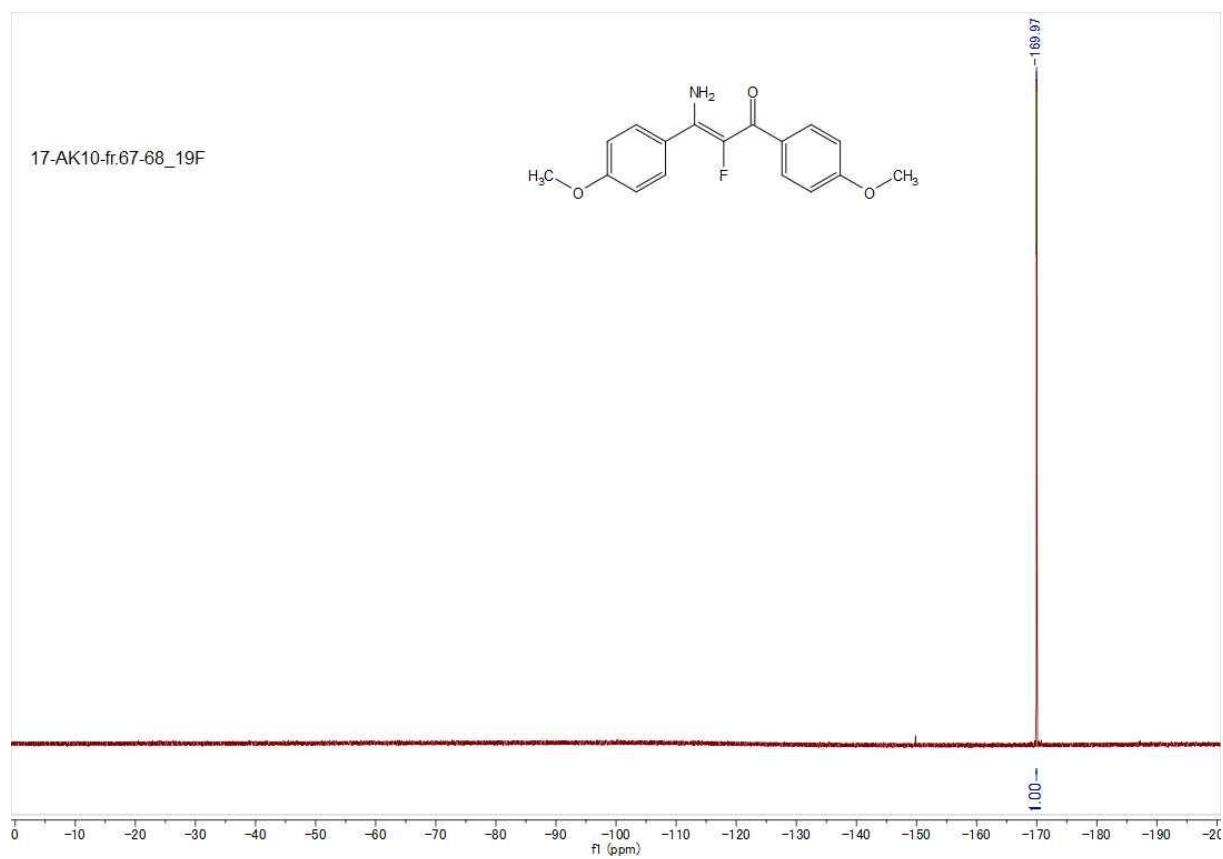


3-Amino-2-fluoro-1,3-diphenylprop-2-en-1-one (**8a**)



17-AK10-fr.67-68_1H

Chemical structure of 17-AK10-fr.67-68_1H is shown, featuring a benzene ring substituted with a methoxy group (H₃C-O-) and a fluorine atom (-F). The ring is connected via a carbonyl group (C=O) to a side chain containing an amino group (-NH₂) and a fluorine atom (-F).

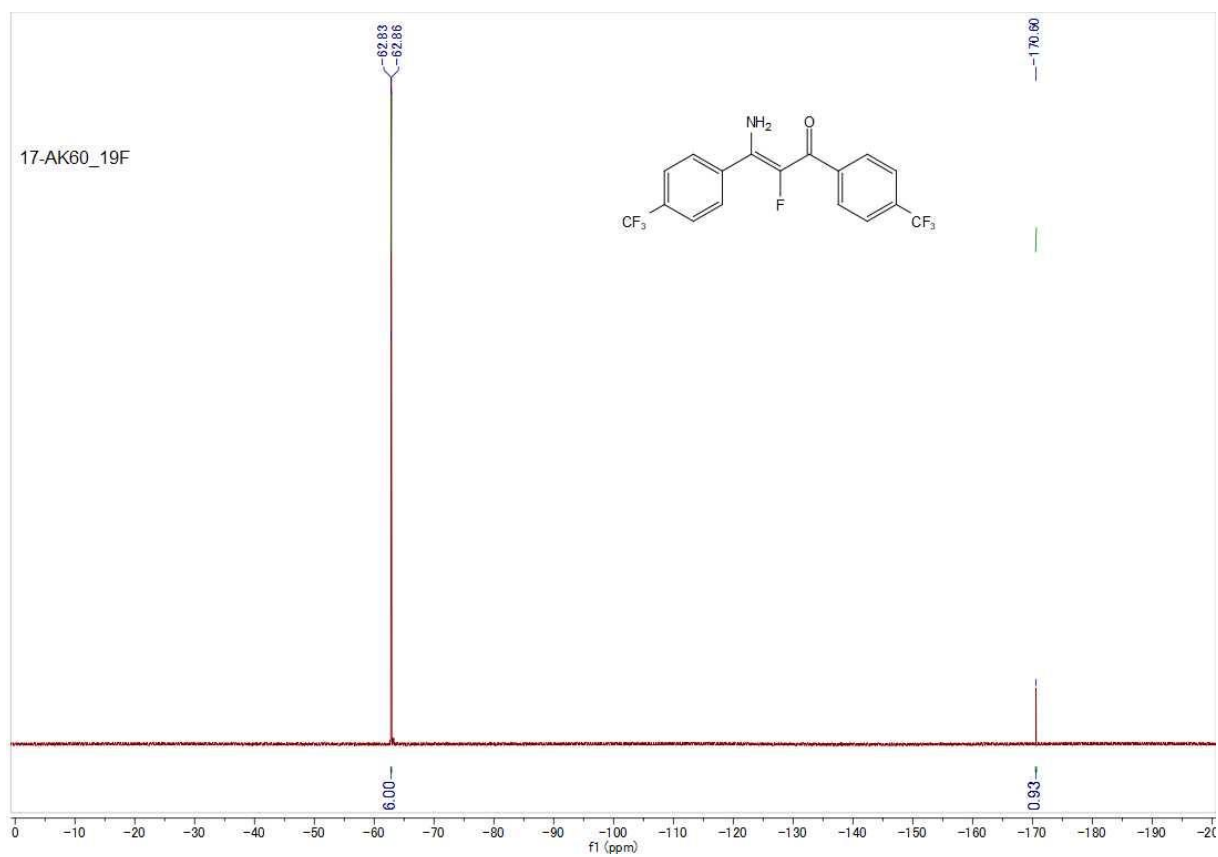


17-AK60_1H

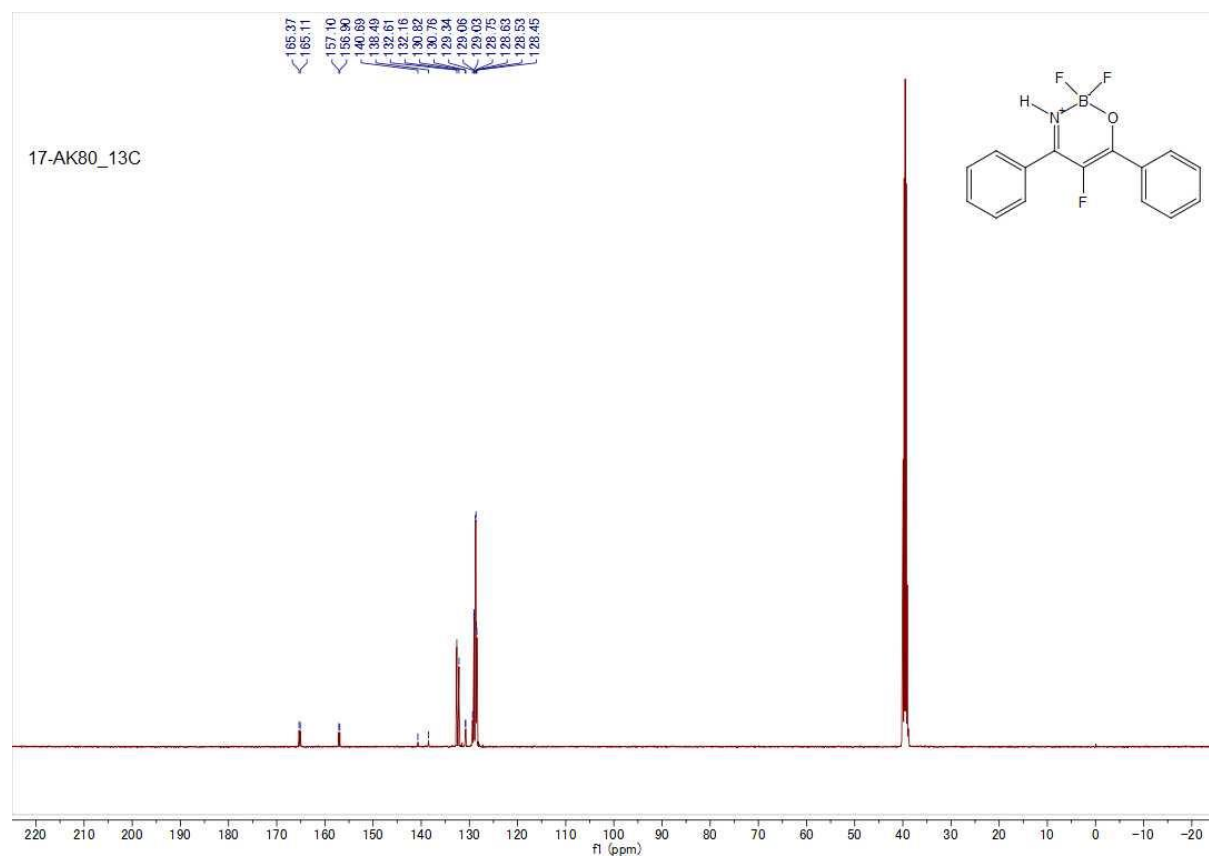
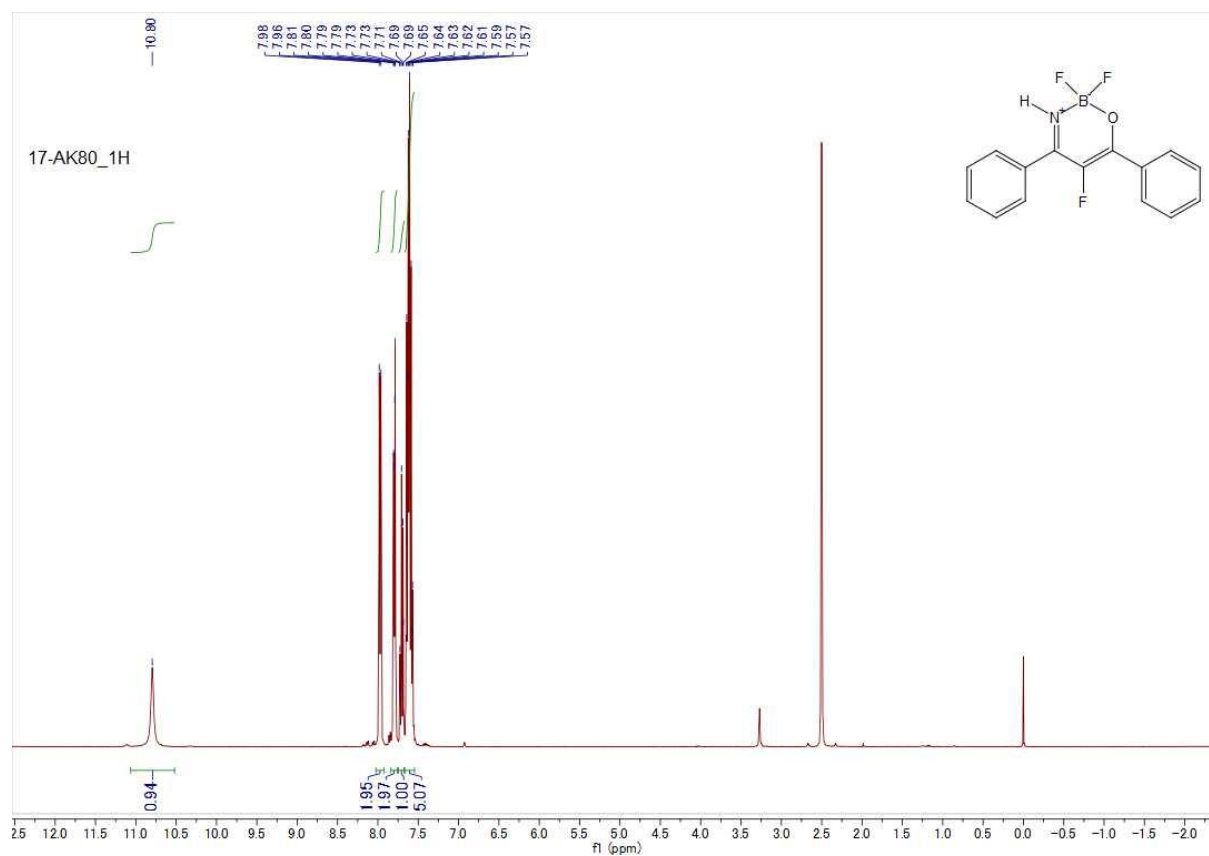
Chemical structure: N/C(=C(F)C(=O)c1ccc(C(F)(F)F)cc1)c2ccc(C(F)(F)F)cc2

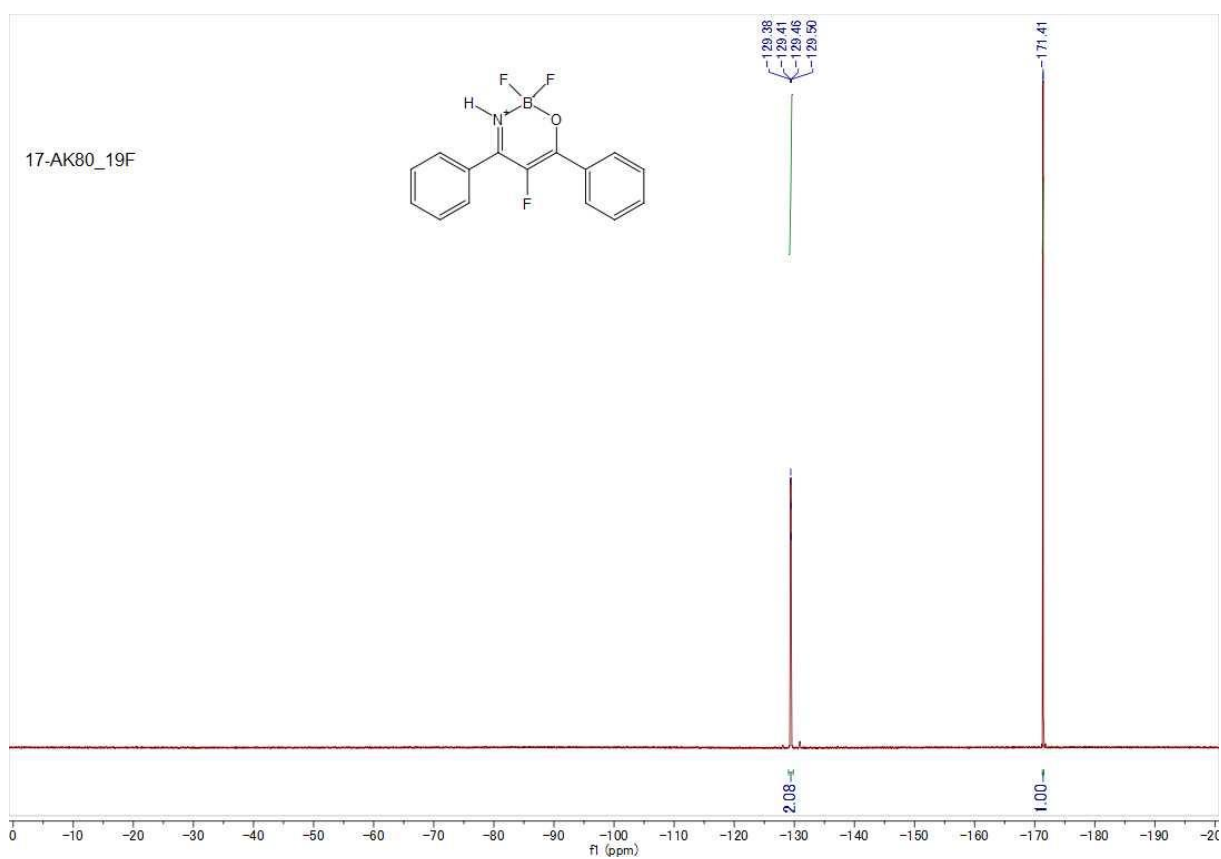
Chemical shift (ppm): 8.01, 7.99, 7.98, 7.97, 7.73, 7.72, 7.71, 7.70, 1.55, 0.00

Integration: 2.00, 3.91, 2.05

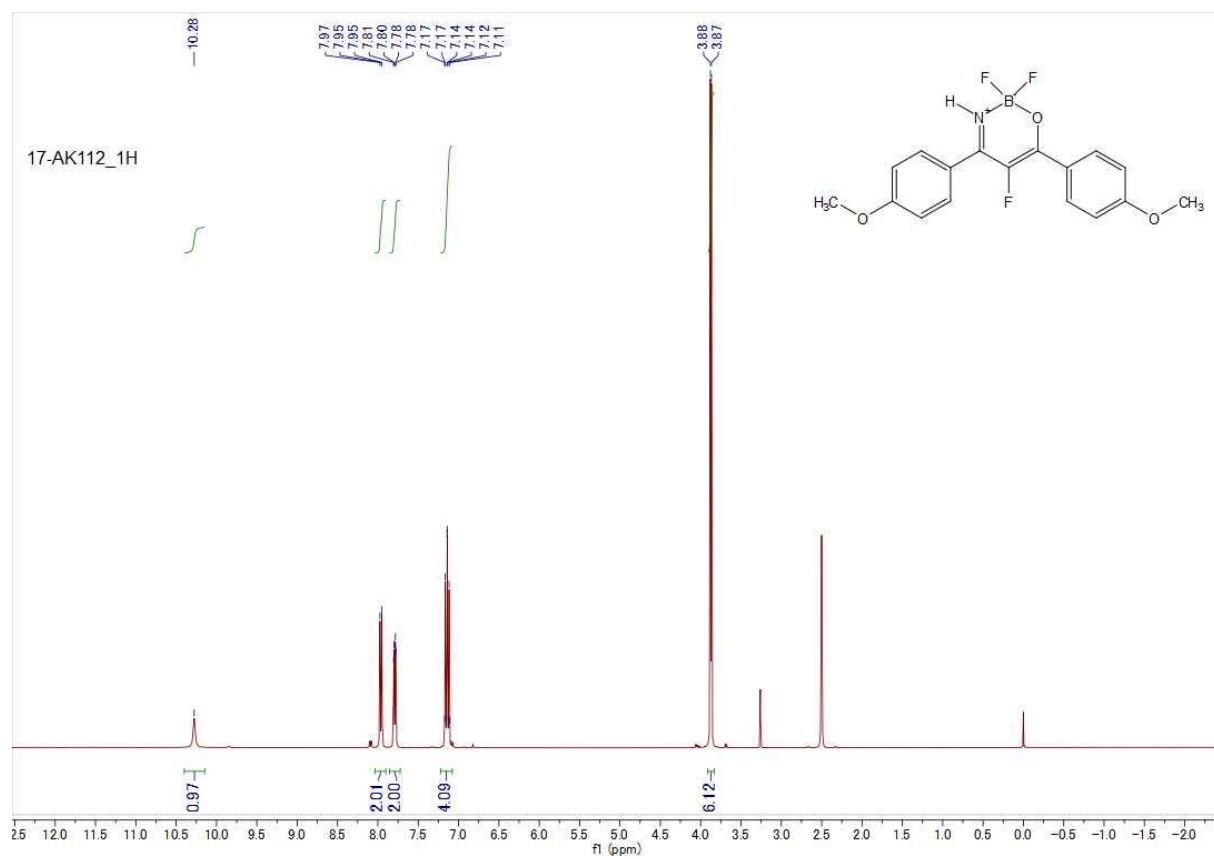


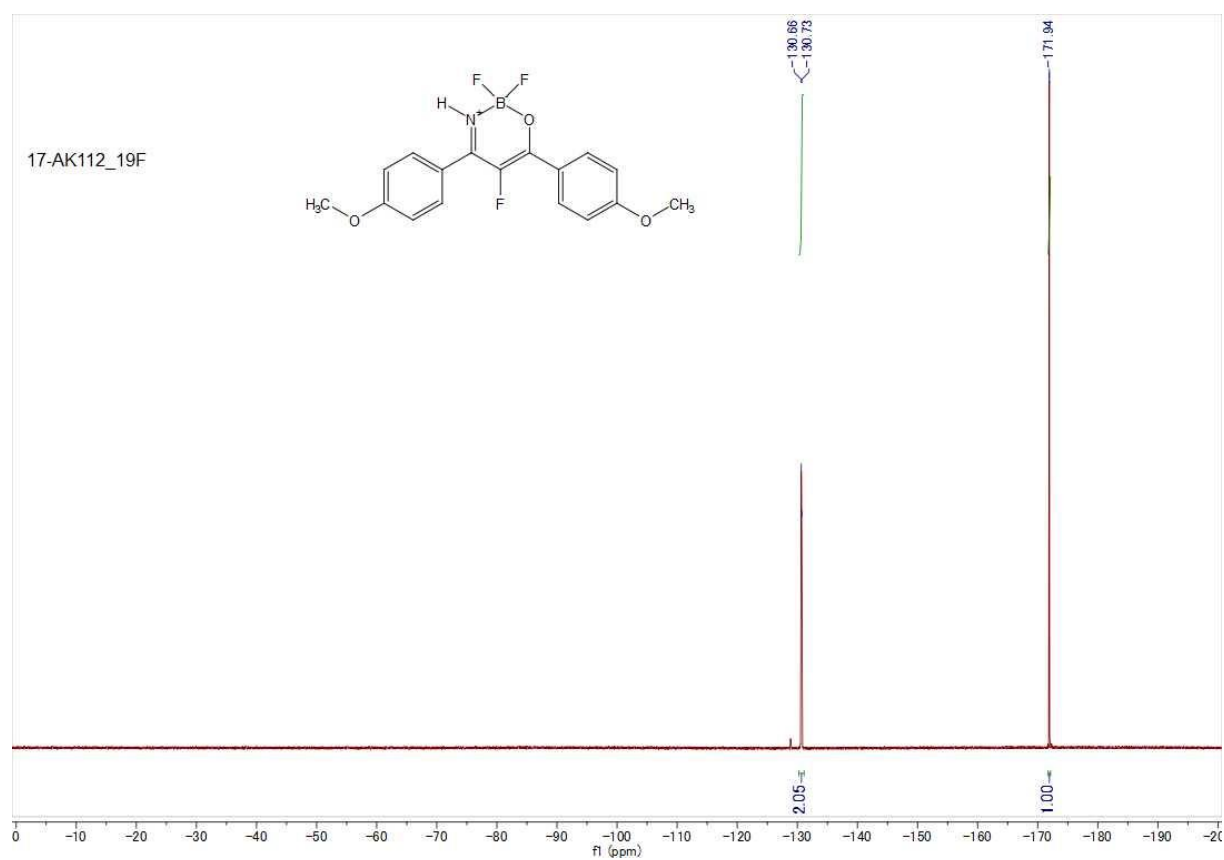
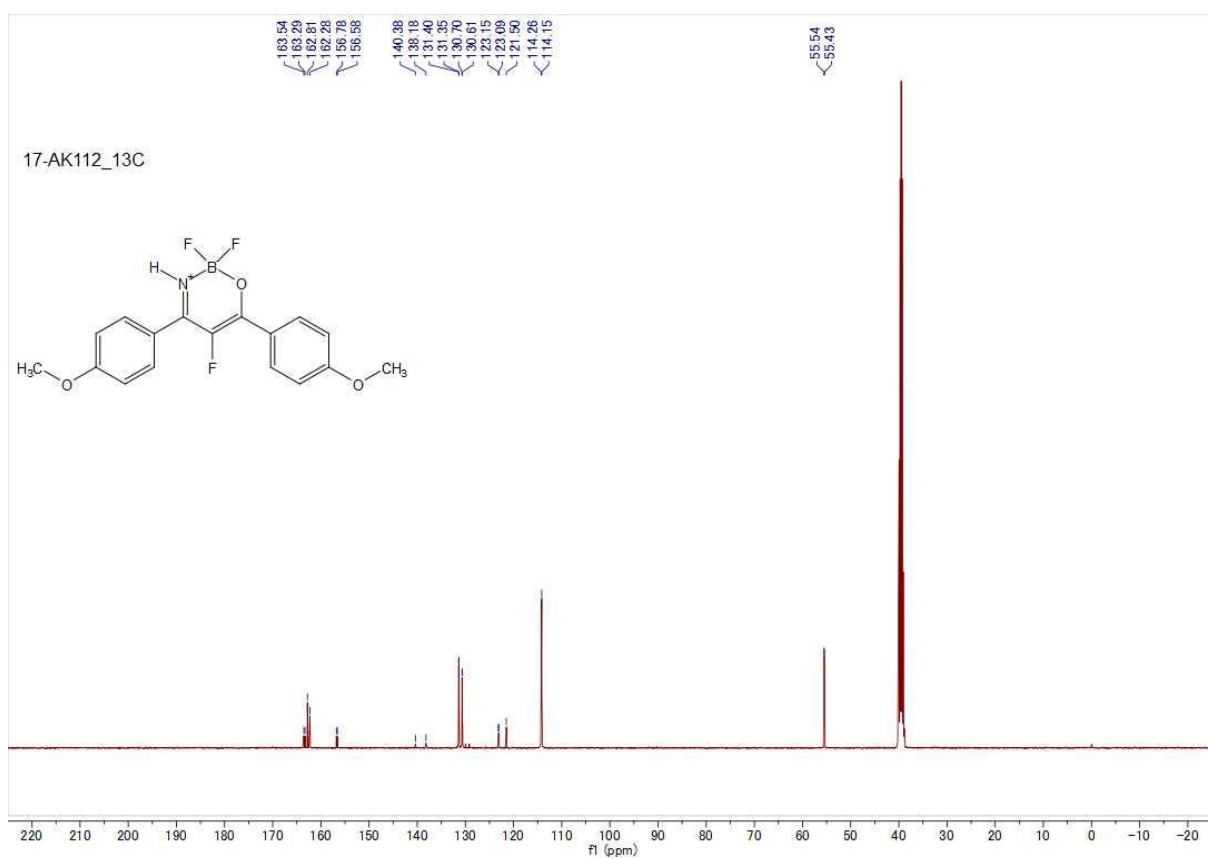
2,2,5-Trifluoro-4,6-diphenyl-2*H*-1,3,2λ⁴-oxazaborinine (**9a**)





2,2,5-Trifluoro-4,6-bis(4-methoxyphenyl)-2H-1,3,2λ⁴-oxazaborinine (**9b**)





2,2,5-Trifluoro-4,6-bis(4-(trifluoromethyl)phenyl)-2*H*-1,3,2λ⁴-oxazaborinine (**9c**)

