## Supporting Information

for

# Design and synthesis of a bis-macrocyclic host and guests as building blocks for small molecular knots 

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Beilstein J. Org. Chem. 2020, 16, 2314-2321. doi:10.3762/bjoc.16.192

Conformations of host 1, TLC knot-forming scheme, experimental procedures and copies of ${ }^{1} \mathrm{H}$ NMR spectra

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## Conformations of host 1

Host 1 has two rigid, electron-rich macrocycles. CPK models suggest that the macrocycles will remain in a relatively open conformation which should aid threading of the electron-poor guest molecules. The phenyl ring that has the propargyloxy substituent is free to rotate such that the two alkyne groups in $\mathbf{1}$ can adopt an anti or syn-conformation as shown in Figure $\mathrm{S}_{1}$ for the meta isomer of $\mathbf{1}$. The same conformations are possible for the ortho isomer of $\mathbf{1}$ as well.


Figure $\mathbf{S 1}_{1}$ : The anti and syn conformations of the meta isomer of host $\mathbf{1}$.

## Proposed second-generation TLC knot-forming scheme

Figure 3 in the manuscript shows a schematic representation of the thread-link-cut (TLC) sequence of electron-rich bis-macrocyclic host 1 and electron-poor guest 2 and Scheme S1 below shows a molecular representation of the sequence. The proposed threading step is promoted by electrostatic attraction of the ammonium groups with the electron-rich macrocycles as well as a potential hydrogen bond between the ammonium NH and a nitrogen or oxygen atom on the host. The linking step is a copper-catalyzed click cycloaddition reaction between the two alkynes on the host and the two azides on the guest. CPK models suggest the cycloaddition can occur on the alkyne of the threaded macrocycle (right side in Scheme $\mathrm{S}_{1}$ ) or on the alkyne of the other macrocycle (left side in Scheme $\mathrm{S}_{1}$ ). Finally, the cutting step involves cleavage of the ester linkage by saponification to release either the unknotted macrocycle (right) or the trefoil knot (left), depending upon which click reaction occurred in the linking step.





Scheme S1: Molecular representation of the second-generation TLC approach to small trefoil knots, see text for details. Click reactions between the alkyne $\mathbf{1}$ and azide $\mathbf{2}$ and alkyne $\mathbf{2}$ and azide $\mathbf{1}$ leads to the unknot macrocycle after cutting whereas click reactions between the alkyne $\mathbf{1}$ and azide $\mathbf{1}$ and alkyne $\mathbf{2}$ and azide $\mathbf{2}$ leads to the trefoil knot after cutting.

Figure $\mathrm{S}_{2}$ shows the proposed knot shown in three different ways to better illustrate the topology of the molecule. Figure S3 shows the backbone numbering for the unknot isomer that results when the TLC approach is successful for host $\mathbf{1}$ reacting with guest $\mathbf{2}$ (Figure S3a) or with
guest 3 (Figure S3b). The topologically isomeric trefoil knot would have the same backbone numbering.
a

b

C


Figure S2: (a) Proposed knot with normal bond lengths. (b) Proposed knot with bond lengths exaggerated to make the structure more clear. (c) Proposed knot with bond lengths exaggerated and reduced to a minimum crossing diagram.


Unknot 73


Unknot 75

Figure S3: Backbone numbering for the unknots resulting from (a) the reaction of $\mathbf{1}$ and $\mathbf{2}$ or (b) $\mathbf{1}$ and 3, see text for details.

## Experimental

## General

All reagents were ACS reagent quality, purchased from Aldrich, Alfa-Aesar, or TCI and used without further purification. 2-Azidoethyl methanesulfonate was synthesized by the literature method of Tookmanian et al. ${ }^{[1]}$ 1,5-Bis-(2-hydroxyethoxy)naphthalene (7) was prepared by a modification of Dyer and Scott's ${ }^{[2]}$ method, wherein potassium hydroxide, 18 -crown-6, and aliquat 336 were used in place of benzyltrimethylammonium hydroxide. 3,5- $\mathrm{Di}(2-$ propynyloxy)benzyl alcohol (10) was prepared by the method of Antoni et al. ${ }^{[3]}$ Azide 17 was prepared by the method of Yao et al. ${ }^{[4]}$ 4-Azidobutylamine (18) was prepared by the method of Cunningham et al. ${ }^{[5]}$ and matched the spectroscopic data reported by Carboni et al. ${ }^{[6]}$ Dialdehyde 19 was prepared by the method of Yeager and Schissel. ${ }^{[7]}$ Dibromide 23 was prepared by the method of Breyholz et al. ${ }^{[8]}$
All reaction mixtures were stirred with a magnetic stirring bar and conducted under a dry argon atmosphere unless done in an aqueous solvent system. Analytical thin layer chromatography (TLC) was performed on 0.2 mm silica plastic-coated sheets with $\mathrm{F}_{254}$ indicator. Preparative TLC was performed on 1 mm thick silica gel with $\mathrm{F}_{254}$ indicator. Flash chromatography was performed using 230-400 mesh silica gel. NMR spectra were obtained at the following frequencies: ${ }^{1} \mathrm{H}$ ( 500 MHz ) and ${ }^{13} \mathrm{C}(125 \mathrm{MHz})$. Spectra were obtained in chloroform-d $\left(\mathrm{CDCl}_{3}\right)$ unless noted otherwise. Chemical shifts are reported in parts per million (ppm) and coupling constants are reported in hertz $(\mathrm{Hz})$. ${ }^{1} \mathrm{H}$ NMR spectra in $\mathrm{CDCl}_{3}$, $\mathrm{DMSO}-d_{6}$, and acetone- $d_{6}$ were referenced to
tetramethylsilane (TMS $=0.0 \mathrm{ppm}$ ) as an internal standard. ${ }^{13} \mathrm{C}$ NMR spectra in $\mathrm{CDCl}_{3}$ were referenced to the solvent peak at $77.0 \mathrm{ppm} .{ }^{13} \mathrm{C} \mathrm{NMR} \mathrm{spectra} \mathrm{in} \mathrm{DMSO}_{6}$ were referenced to the residual solvent peak at $39.52 \mathrm{ppm} .{ }^{13} \mathrm{C}$ NMR spectra in $\mathrm{CD}_{3} \mathrm{OD}$ were referenced to the residual solvent peak at 49.00 ppm . IR spectra were obtained as a thin film (for solid samples) or neat (for oils and liquids) on an ATR instrument and the absorption peak frequencies are reported $\mathrm{in}_{\mathrm{cm}}{ }^{-1}$ and intensities as very strong (vs), strong (s), medium (m), or weak (w). Mass spectrometry was performed at Franklin \& Marshall College or at the University of Illinois School of Chemical Sciences, Urbana, IL and used electrospray ionization (ESI) in the positive ion mode unless noted otherwise. Mass spectral $\mathrm{m} / \mathrm{z}$ peak data are reported in Daltons with a relative intensity to a base peak (base $=100$ ). Melting points were measured on a Mel-Temp melting point apparatus and are uncorrected.
Abbrevations: ATR (attenuated total reflectance); brine (saturated aqueous sodium chloride); DMAP ( 4 - $\mathrm{N}, \mathrm{N}$-dimethylaminopyridine), DMF (dimethylformamide); DMSO (dimethyl sulfoxide); El (electron ionization); ESI (electrospray ionization); EtOAc (ethyl acetate); $\mathrm{Et}_{2} \mathrm{O}$ (diethyl ether); FC (flash column chromatography using silica gel); MALDI (matrix-assisted laser desorption ionization); MeOH (methanol); PE (low boiling petroleum ether); TOF (time of flight); water (deionized water).

## Procedures



1-(2-Azidoethoxy)-5-hydroxynaphthalene (5). A mixture of 1,5-naphthalenediol (4, 5.00 g, 31.3 mmol ) and acetone ( 100 mL ) was degassed by argon bubbling for 8 min . Potassium carbonate ( $4.31 \mathrm{~g}, 31.2 \mathrm{mmol}$ ) was added and the mixture was stirred for 10 min . 2-Azidoethyl methanesulfonate ( $2.6 \mathrm{~g} \mathrm{~g}, 16.3 \mathrm{mmol}$ ) was then added and the mixture was heated at reflux for 17 h . Additional 2-azidoethyl methanesulfonate ( $2.70 \mathrm{~g}, 16.3 \mathrm{mmol}$ ) was added and the heating was continued for 23.5 h . The solvent was removed under reduced pressure and hot $2.5 \%$ $\mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}(300 \mathrm{~mL})$ was added to the residue. The salts were removed by vacuum filtration and silica gel was added to the filtrate, which was concentrated under reduced pressure. Purification on a large silica gel plug ( $0 \rightarrow 60 \% \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{PE}$ ) gave $1.915 \mathrm{~g}(27 \%)$ of 5 as pale yellow powder; mp 136-137 ${ }^{\circ} \mathrm{C}$; IR v 3274.4 (br, w), $2933.8(\mathrm{w}), 2878.0(\mathrm{w}), 2100.8(\mathrm{~m}), 2086.2(\mathrm{~m}), 2062.6$ $(\mathrm{m}), 1600.3(\mathrm{~m}), 1522.8(\mathrm{w}), 1414.3(\mathrm{~m}), 1381.1(\mathrm{~m}), 1283.0(\mathrm{~m}), 1269.4(\mathrm{~m}), 1058.8(\mathrm{~m}), 965.2(\mathrm{~m})$, $956.8(\mathrm{~m}), 773.0(\mathrm{vs}) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 7.87(\mathrm{~d}, J=9,1 \mathrm{H}), 7.79(\mathrm{~d}, J=9,1 \mathrm{H}), 7.38(\mathrm{~m}, 1 \mathrm{H}), 7.32(\mathrm{~m}$, $\left.{ }_{1} \mathrm{H}\right), 6.85(\mathrm{~m}, 2 \mathrm{H}), 4.31(\mathrm{t}, J=5,2 \mathrm{H}), 3.72(\mathrm{t}, \mathrm{J}=5,2 \mathrm{H}) ;{ }^{13} \mathrm{CNMR}\left(\sim 15 \% \mathrm{CD}_{3} \mathrm{OD}\right.$ in $\left.\mathrm{CDCl}_{3}\right) \delta 153.73$, $152.35,126.65,125.79,125.54,124.36,114.95,113.29,108.88,105.01,67.04,50.36$; MS (ESI) $230.1(\mathrm{M}+1,12), 202.1\left(\mathrm{M}+1-\mathrm{N}_{2}, 100\right), 174.1\left(\mathrm{M}+1-\mathrm{N}_{3} \mathrm{CH}_{2}, 46\right), 172.1$ (50); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{~N}_{3} \mathrm{O}_{2}[\mathrm{M}+1]^{+}$: 230.0930 ; found: 230.0931.


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1,5-Bis[(2-methanesulfonoxy)ethoxy]naphthalene (8). A solution of 7 ( $8.83 \mathrm{~g}, 35.6 \mathrm{mmol}$ ) in DMF ( 90 mL ) and triethylamine ( 15.0 mL ) was cooled in an ice-water bath and methanesulfonyl chloride ( $7.1 \mathrm{~mL}, 91 \mathrm{mmol}$ ) was added dropwise over a 25 min period. The mixture was allowed to warm to room temperature and stirring was continued for 116 h . Methanol ( 4 mL ) was added and stirring continued for 25 min . Then the mixture was poured into ice water ( 500 mL ) and the tan precipitate was collected by vacuum filtration and washed with water. The pasty tan material was dried in an oven at $50^{\circ} \mathrm{C}$ for 48 h to give 13.245 g ( $92 \%$ ) of 8 as beige powder. mp $168-170{ }^{\circ} \mathrm{C}$ (dec.); IR v 3034.5 (w), 2979.6 (w), 2946.1 (w), 1595.2 (w), 1514.2 (w), 1419.1 (w), $1401.5(\mathrm{~m}), 1354.5(\mathrm{~s}), 1330.1(\mathrm{~m}), 1270.2(\mathrm{w}), 1181.4(\mathrm{~s}), 1087.1(\mathrm{~m}), 1075.9(\mathrm{w}), 1023.3(\mathrm{~m}), 978.0$ (m), 966.3 (m), $931.0(\mathrm{~s}), 837.5$ (w), 813.3 (m), 775.1 ( s$), 742.3$ (w); ${ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ) $\delta 7.79$ (d, J $=8,2 \mathrm{H}), 7.42(\mathrm{t}, J=8,2 \mathrm{H}), 7.03(\mathrm{~d}, J=8,2 \mathrm{H}), 4.67(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 4.42(\mathrm{br} \mathrm{s}, 4 \mathrm{H}), 3.27(\mathrm{~s}, 6 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (DMSO-d ${ }_{6}$ ) $\delta 153.32,125.85,125.59,114.18,106.22,68.82,66.16,36.86 ;$ MS (ESI-TOF) 405.1 (M+1, 100), 382.4 (17); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{16} \mathrm{H}_{21} \mathrm{O}_{8} \mathrm{~S}_{2}[\mathrm{M}+1]^{+}: 405.0678$; found: 405.0667 .


1-(2-Azidoethoxy)-5-(2-methanesulfonylethoxy)naphthalene (9). A heterogeneous mixture of $8(10.0 \mathrm{~g}, 24.7 \mathrm{mmol})$ in DMSO ( 150 mL ) was stirred and heated at $55^{\circ} \mathrm{C}$ for 7 min until homogenous. Sodium azide ( $1.625 \mathrm{~g}, 24.99 \mathrm{mmol}$ ) was dissolved in water ( 4 mL ) by sonication and then added to the homogenous DMSO solution. The mixture was heated at $60{ }^{\circ} \mathrm{C}$ and stirred for 18.5 h . The mixture was allowed to cool to room temperature. The product was extracted with water two times ( 150 mL ), ethyl acetate two times ( 150 mL ), and brine two times $(150 \mathrm{~mL})$. The combined organic layers were dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give 6.7814 g of a dark solid. The compound was dissolved in acetone/methanol/ $/ \mathrm{CH}_{2} \mathrm{Cl}_{2}$ 2:1:1 and preloaded onto silica gel. The column was eluted with $10 \%$ ethyl acetate/petroleum ether ( 1 L ), $20 \%$ ethyl acetate/petroleum ether ( 500 mL ), $30 \%$ ethyl acetate/petroleum ether ( 500 mL ), $33 \%$ ethyl acetate/petroleum ether ( 1 L ), and $45 \%$ ethyl acetate/petroleum ether ( 1.5 L ). A beige powder was isolated and recrystallized from toluene to give $1.991 \mathrm{~g}(46 \%)$ of 9 as beige powder: mp 111.0-112.5 ${ }^{\circ} \mathrm{C}$; IR v 3061.3 (w), 3035.8 (w), 2982.8 (w), 2947.1 (w), 2877.1 (w), 2104.6 (s), 2071.0 (m), $1594.4(\mathrm{~m}), 1513.4(\mathrm{~m}), 1459.3$ ( w$), 1416.7$ (m), 1403.1 (m), 1388.3 (w), 1355.8 ( s$)$, 1330.0 (s), 1301.2 (w), 1270.3 ( s$), 1242.3$ (m), 1183.5 (s), 1120.4 (w), 1086.0 (s), 1073.2 (m), 1022.2 (m), 977.8 (m), 965.7 (m), 931.9 (s), 839.3 (w), 819.3 (m), 811.2 (m), 771.8 (s), 746.5 (w), $632.4(\mathrm{w})$;
${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 7.92(\mathrm{~d}, \mathrm{~J}=9,1 \mathrm{H}), 7.88(\mathrm{~d}, J=9,1 \mathrm{H}), 7.39(\mathrm{~m}, 2 \mathrm{H}), 6.85(\mathrm{~d}, J=7,2 \mathrm{H}), 4.71(\mathrm{~m}$, $2 \mathrm{H}), 4.42(\mathrm{t}, J=3,2 \mathrm{H}), 4.32(\mathrm{t}, J=5,2 \mathrm{H}), 3.73(\mathrm{t}, J=5,2 \mathrm{H}), 3.08(\mathrm{~s}, 3 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(\mathrm{CDCl}_{3}\right) \delta 153.86$, $153.45,126.58,126.54,125.35,125.24,115.31,114.73,105.94,105.61,67.88,67.21,66.11,50.44$, $37.84 ; \mathrm{MS}$ (ESI-TOF) $374.1\left(\mathrm{M}+\mathrm{Na}^{+}, 23\right), 352.1(\mathrm{M}+1,35), 324.1\left(\mathrm{M}+1-\mathrm{N}_{2}, 100\right), 228.1\left(\mathrm{M}-\mathrm{C}_{2} \mathrm{H}_{4} \mathrm{OMS}\right.$, 52); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{~S}[\mathrm{M}+1]^{+}: 352.0967$; found: 352.0950 .


1-(2-Azidoethoxy)-5-(2-bromoethoxy)naphthalene (6).
Method A from 5: A mixture of 5 ( $1.92 \mathrm{~g}, 8.38 \mathrm{mmol}$ ), 1,2-dibromoethane ( $2.5 \mathrm{~mL}, 29 \mathrm{mmol}$ ), potassium carbonate ( $1.18 \mathrm{~g}, 8.55 \mathrm{mmol}$ ), 18 -crown- $6(24 \mathrm{mg}, 0.091 \mathrm{mmol})$, and acetone ( 30 mL ) was heated at reflux for 18 h . Additional 1,2-dibromoethane ( $2.0 \mathrm{~mL}, 23 \mathrm{mmol}$ ), potassium carbonate ( $1.43 \mathrm{~g}, 10.4 \mathrm{mmol}$ ), and acetone ( 10 mL ) were added and heating continued for 7 h . A final dose of 1,2-dibromoethane ( $2.0 \mathrm{~mL}, 23 \mathrm{mmol}$ ), potassium carbonate ( $1.00 \mathrm{~g}, 7.25 \mathrm{mmol}$ ), and acetone ( 10 mL ) were added and heating continued for an additional 47 h . The mixture was concentrated under reduced pressure and the residue was partitioned between methylene chloride and water. The aqueous layer was extracted with methylene chloride and the combined organic layers were washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated under reduced pressure. The crude product was recrystallized from $\mathrm{MeOH} /$ toluene $4: 1$, followed by recrystallization from acetonitrile to give $1.695 \mathrm{~g}(60 \%)$ of 6 as yellow crystals; mp 107.5$108.0^{\circ} \mathrm{C}$; IR v 3066.3 (w), 2937.7 (w), 2873.6 ( w ), 2105.0 ( s$)$, 2071.2 (m), 1959.9 (m), 1511.9 (m), 1413.6 (s), 1384.2 (m), 1270.8 (s), 1090.2 (m), 1037.7 (m), 1019.1 (m), $964.5(\mathrm{w}), 948.6$ (w), 773.8 (vs); ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 7.93(\mathrm{~d}, J=9,1 \mathrm{H}), 7.90(\mathrm{~d}, J=9,1 \mathrm{H}), 7.38$ (apparent t, $\left.J=8,2 \mathrm{H}\right), 6.84(\mathrm{~m}$, $2 \mathrm{H}), 4.46(\mathrm{t}, J=6,2 \mathrm{H}), 4.31(\mathrm{t}, J=5,2 \mathrm{H}), 3.77(\mathrm{t}, J=6,2 \mathrm{H}), 3.72(\mathrm{t}, J=5,2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta$ 153.80, 153.59, 126.69, 126.56, 125.27, 125.16, 115.04, 115.00, 106.01, 105.54, 68.02, 67.17, 50.43, 29.26; MS (ESI) $358.4\left(\mathrm{M}+\mathrm{Na}^{+}\right)$; HRMS (ESI-TOF) calcd. for $\mathrm{C}_{14} \mathrm{H}_{15} \mathrm{~N}_{3} \mathrm{O}_{2} \mathrm{Br}[\mathrm{M}+1]^{+}: 336.0348 ;$ found: 336.0340 .
Method B from 9: To a solution of mesylate $9(0.636 \mathrm{~g}, 1.81 \mathrm{mmol})$ in DMF ( 10 mL ) was added $\mathrm{KBr}(0.502 \mathrm{~g}, 4.22 \mathrm{mmol})$ and 18 -crown-6 ( $\approx 5 \mathrm{mg}, 0.02 \mathrm{mmol}$ ). The mixture was stirred at $45^{\circ} \mathrm{C}$ for 23.5 h . The mixture was diluted with EtOAc and water and the organic layer was washed with water $(3 \times)$, brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to give $0.568 \mathrm{~g}(94 \%)$ of 6 as yellow solid which was identical to the material prepared by Method A above.


11
4-[3-(Hydroxymethyl)-5-(2-propyn-1-yloxy)phenoxy]methyl]-1-[5-(2-bromoethoxy)-1-naphthyloxy)ethoxy]-1H-1,2,3-triazole (11). A solution of 10 ( $3.220 \mathrm{~g}, 14.91 \mathrm{mmol}$ ), methylene chloride ( 12 mL ), and acetonitrile ( 3 mL ) was degassed by argon bubbling and then $\left[\mathrm{CH}_{3} \mathrm{CN}\right]_{4} \mathrm{CuPF}_{6}$ ( $427 \mathrm{mg}, 1.14 \mathrm{mmol}$ ), ascorbic acid ( 2 mL of o.2 M aqueous solution, 0.4 mmol ), and the tris(2-benzimidazolylmethyl)amine ligand ( $18 \mathrm{mg}, 0.044 \mathrm{mmol}$ ) were added. This solution was added to a second degassed solution of $6(1.25 \mathrm{~g}, 3.72 \mathrm{mmol})$ in methylene chloride $(20 \mathrm{~mL})$, and stirred at room temperature for 3 h . The solution was concentrated and purified by flash chromatography ( $50 \%$ ethyl acetate/petroleum ether, $67 \%$ ethyl acetate/petroleum ether, $80 \%$ ethyl acetate/petroleum ether, $100 \%$ ethyl acetate) to give $866 \mathrm{mg}(42 \%)$ of 11 as off-white flaky solid. mp 116-118 ${ }^{\circ} \mathrm{C}$; IR v 3394.9 (br w), 3287.5 (w), 2929.4 (w), 2871.1 (w), 2121.1 (w), 1685.4 (w), 1594.9 (vs), 1509.2 (m), 1454.7 (m), 1414.5 (s), 1380.6 (m), 1289.3 (m), 1265.8 (s), $1165.3(\mathrm{~s}), 1090.0(\mathrm{~m}), 1052.1(\mathrm{~s}), 971.7(\mathrm{~m}), 834.6(\mathrm{~m}), 777.0(\mathrm{~s}) ;{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ) $\delta 8.42(\mathrm{~s}, 1 \mathrm{H})$, $7.80(\mathrm{~d}, J=9,1 \mathrm{H}), 7.65(\mathrm{~d}, J=9,1 \mathrm{H}), 7.41$ (apparent $\mathrm{t}, J=8,1 \mathrm{H}), 7.37$ (apparent $\mathrm{t}, J=8,1 \mathrm{H}$ ), 7.03 $(\mathrm{d}, J=8,1 \mathrm{H}), 6.99(\mathrm{~d}, J=8,1 \mathrm{H}), 6.63(\mathrm{~s}, 1 \mathrm{H}), 6.56(\mathrm{~s}, 1 \mathrm{H}), 6.53(\mathrm{t}, J=2,1 \mathrm{H}), 5.22(\mathrm{t}, J=6,1 \mathrm{H}), 5.12$ $(\mathrm{s}, 2 \mathrm{H}), 4.95(\mathrm{t}, J=5,4 \mathrm{H}), 4.75(\mathrm{~d}, J=2,2 \mathrm{H}), 4.56(\mathrm{t}, J=5,2 \mathrm{H}), 4.47(\mathrm{t}, J=5,2 \mathrm{H}), 4.44(\mathrm{~d}, J=6,2 \mathrm{H})$, $3.93(\mathrm{t}, \mathrm{J}=5,2 \mathrm{H}), 3.54(\mathrm{t}, \mathrm{J}=2,1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR (DMSO- $\left.\mathrm{d}_{6}\right) \delta 159.08,158.30,153.28,145.31,142.75$, 125.93, 125.84, 125.63, 125.55, 125.21, 114.28, 114.14, 106.41, 106.24, 105.37, 100.06, 79.37, $78.17,68.06,66.66,62.79,61.17,55.44,49.25,31.72$; MS (ESI-TOF) 552.1 ( $\mathrm{M}+1,100$ ), 534.1 (M$\mathrm{H}_{2} \mathrm{O}, 86$ ), 205.2 (7); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{27} \mathrm{H}_{27} \mathrm{~N}_{3} \mathrm{O}_{5} \mathrm{Br}[\mathrm{M}+1]^{+}$: 552.1134 ; found: 552.1145 .


13
Diethyl 2,5-bis[2-[5-[4-[3-(hydroxymethyl)-5-(2-propyn-1-yloxy)phenoxy]methyl]-1H-1,2,3-triazol-1-yl]-1-naphthyloxy]ethoxy]-1,4-benzenedicarboxylate (13).
Method A from 11. A mixture of 11 ( $553 \mathrm{mg}, 1.00 \mathrm{mmol}$ ), 18 -crown-6 ( $20.4 \mathrm{mg}, 0.077 \mathrm{mmol}$ ), in DMF ( 8.5 mL ) was added 12 ( $41.4 \mathrm{mg}, 0.164 \mathrm{mmol}$ ), and potassium carbonate ( 232 mg , $1.68 \mathrm{mmol})$ and the mixture was heated at $85^{\circ} \mathrm{C}$ for 200 min . Additional 11 ( $81 \mathrm{mg}, 0.32 \mathrm{mmol}$ ) was added and the mixture was heated at $85^{\circ} \mathrm{C}$ for 18 h . The mixture was cooled and diluted
with EtOAc ( 25 mL ) and water ( 25 mL ). The organic layer was separated and the aqueous layer was extracted with $\mathrm{EtOAc}(20 \mathrm{~mL}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$, and $\mathrm{EtOAc}(20 \mathrm{~mL})$. The combined organic layers were washed with water $(1 \times 30 \mathrm{~mL})$, brine $(2 \times 30 \mathrm{~mL})$, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to give 527 mg of a brown oil. The oil was dissolved in acetone, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$, and methanol; silica gel was added, and the solvents were removed under reduced pressure. The silica gel with the crude product absorbed onto it was added to the top of a silica gel column and the product was purified by flash chromatography (1:1 $\mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$, 1:25:25 $\mathrm{MeOH} / \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}, 1: 20: 20 \mathrm{MeOH} / \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give 105.2 mg (18\%) of 13 as white solid identical to the material prepared by Method B below.
Method B from 14. A mixture of 14 ( $298 \mathrm{mg}, 0.390 \mathrm{mmol}$ ), methylene chloride ( 10 mL ), and methanol ( 4 mL ) was heated at $\approx 30^{\circ} \mathrm{C}$ until homogeneous. Ascorbic acid ( $86 \mathrm{mg}, 0.48 \mathrm{mmol}$ ) was added and the mixture was sonicated until homogenous. Benzyl alcohol $\mathbf{1 0}$ ( 796 mg , 3.68 mmol ) was added, the solution was degassed by bubbling with argon, and $\left[\mathrm{CH}_{3} \mathrm{CN}\right]_{4} \mathrm{CuPF} 6$ $(76 \mathrm{mg}, 0.20 \mathrm{mmol})$ was added. The mixture was capped tightly and stirred at room temperature for 240 h . The solution was decanted away from a yellow precipitate and a solid-liquid extraction done with $20 \%$ methanol/methylene chloride ( 10 mL ) and acetone ( 10 mL ). The combined solutions were concentrated to a yellow oil which was purified by flash chromatography (1:1 EtOAc/PE, 1:20:20 $\quad \mathrm{MeOH} / \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}, \quad 1: 15: 15 \quad \mathrm{MeOH} / E t O A c / \mathrm{CH}_{2} \mathrm{Cl}_{2}, \quad 1: 10: 10$ $\mathrm{MeOH} / \mathrm{EtOAc} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$, and 1:5:5 $\mathrm{MeOH} / \mathrm{EtOAc}_{2} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to provide $419.1 \mathrm{mg}(87 \%)$ of 13 as pale yellow solid. mp $156.5-158.0^{\circ} \mathrm{C}$; IR v 3286.6 ( w ), 2925.5 ( w ), 1697.3 ( m ), 1595.1 ( m ), 1509.8 (m), 1454.7 (w), 1414.5 (s), 1338.3 (w), 1291.2 (w), 1266.6 (s), 1225.9 (m), 1210.9 (m), 1151.0 (s), 1089.9 (s), 1048.9 (m), 1035.5 (m), $835.1(\mathrm{w}), 767.0(\mathrm{~s}) ;{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ) $\delta 8.40(\mathrm{~s}, 2 \mathrm{H}), 7.65(\mathrm{~d}, \mathrm{~J}=9$, $2 \mathrm{H}), 7.62(\mathrm{~d}, J=9,2 \mathrm{H}), 7.48(\mathrm{~s}, 2 \mathrm{H}), 7.37(\mathrm{~m}, 2 \mathrm{H}), 7.31(\mathrm{~m}, 2 \mathrm{H}), 7.03(\mathrm{~d}, J=8,2 \mathrm{H}), 6.97(\mathrm{~d}, J=8,2 \mathrm{H})$, $6.61(\mathrm{~s}, 2 \mathrm{H}), 6.55(\mathrm{~s}, 2 \mathrm{H}), 6.51(\mathrm{~s}, 2 \mathrm{H}), 5.18(\mathrm{t}, \mathrm{J}=6,2 \mathrm{H}), 5.11(\mathrm{~s}, 4 \mathrm{H}), 4.94(\mathrm{t}, \mathrm{J}=5,4 \mathrm{H}), 4.74(\mathrm{~d}, J=$ $2,4 \mathrm{H}), 4.54(\mathrm{t}, J=5,4 \mathrm{H}), 4.52(\mathrm{~m}, 4 \mathrm{H}), 4.44(\mathrm{~m}, 4 \mathrm{H}), 4.42(\mathrm{~d}, J=6,4 \mathrm{H}), 4.16(\mathrm{q}, \mathrm{J}=7,4 \mathrm{H}), 3.54$ (apparent s, 2H), 1.11 (t, J = 7, 6H); MS (MALDI) 1219.69 (M+Na+, 100), 626.75 (93), 507.63 (97); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{66} \mathrm{H}_{65} \mathrm{~N}_{6} \mathrm{O}_{16}[\mathrm{M}+1]^{+}: 1197.4457$; found: 1197.4445 .


14
Diethyl 2,5-bis[[5-(2-azidoethoxy)-1-naphthyloxy]ethoxy]-1,4-benzene-dicarboxylate (14). A mixture of 6 ( $212 \mathrm{mg}, 0.633 \mathrm{mmol}$ ), 12 ( $73 \mathrm{mg}, 0.29 \mathrm{mmol}$ ), cesium carbonate ( 215 mg , 0.660 mmol ), and acetone ( 4 mL ) were heated at reflux for 23 h . The mixture was cooled to $5^{\circ} \mathrm{C}$ and the yellow precipitate was collected by vacuum filtration, washed with water and diethyl ether, and dried to give $180 \mathrm{mg}(82 \%)$ of 14 as yellow powder: mp $155-158^{\circ} \mathrm{C}$; IR $v 3320(\mathrm{w})$, 2974.0 (w), 2934.1 (w), 2914.3 (w), 2161.2 (w), 2143.2 (w), 2104.3 (m), 2072.8 (m), 1720.7 (m), 1702.1 (m), 1592.7 (m), 1509.5 (m), 1459.6 (w), 1430.2 (w), $1420.1(\mathrm{~s}), 1409.9(\mathrm{~s}), 1372.2(\mathrm{~m})$, $1295.0(\mathrm{~m}), 1270.3(\mathrm{~s}), 1239.6$ (m), 1229.2 (m), 1212.3 ( s$), 1062.1(\mathrm{~s}), 1015.8(\mathrm{w}), 969.2(\mathrm{~m}), 951.1$ (m), 879.6 (w), 775.6 ( s$), 758.2(\mathrm{~m}), 737.2(\mathrm{w}) ;{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ) $\delta 7.76(\mathrm{~d}, \mathrm{~J}=8,2 \mathrm{H}), 7.67(\mathrm{~d}, \mathrm{~J}=$

8, 2H), $7.50(\mathrm{~s}, 2 \mathrm{H}), 7.42(\mathrm{t}, J=8,2 \mathrm{H}), 7.33(\mathrm{t}, J=8,2 \mathrm{H}), 7.06(\mathrm{~d}, J=8,2 \mathrm{H}), 6.97(\mathrm{~d}, J=8,2 \mathrm{H}), 4.53$ $(\mathrm{m}, 4 \mathrm{H}), 4.47(\mathrm{~m}, 4 \mathrm{H}), 4.33(\mathrm{t}, J=5,4 \mathrm{H}), 4.18(\mathrm{q}, J=7,4 \mathrm{H}), 3.78(\mathrm{t}, J=5,4 \mathrm{H}), 1.13(\mathrm{t}, J=7,6 \mathrm{H}) ; \mathrm{MS}$ (ESI) 765.3 ( $\mathrm{M}+1,100$ ), 338.3 (32); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{40} \mathrm{H}_{41} \mathrm{~N}_{6} \mathrm{O}_{10}[\mathrm{M}+1]^{+}: 765.2884$; found: 765.2853 .


15
2,5-Bis[2-[5-[4-[3-(hydroxymethyl)-5-(2-propyn-1-yloxy)phenoxy]methyl]-1H-1,2,3-triazol-1-yl]-1-naphthyloxy]ethoxy]-1,4-benzenedicarboxylic acid (15). To a solution of KOH ( 8 omg , $1.4 \mathrm{mmol})$ in water ( 2.0 mL ) was added $13(43 \mathrm{mg}, 36 \mathrm{mmol})$ in THF ( 2.0 mL ) and ethanol $(2.0 \mathrm{~mL})$. The heterogeneous mixture was stirred at $95^{\circ} \mathrm{C}$ for 95 h . Upon cooling to room temperature, 1 M hydrochloric acid ( 1 mL ) was added to the solution. The mixture was partitioned between saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ and methylene chloride. The aqueous layer was extracted with methylene chloride ( $3 \times 2 \mathrm{~mL}$ ). The combined organic layers were diluted with methanol ( 2 mL ) and washed with brine, dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated to give 32 mg ( $78 \%$ ) of 15 as off-white solid. mp 190-191 ${ }^{\circ} \mathrm{C}$; IR v 3161.0 (w), 3054.4 (w), 2931.1 (w), 1723.0 (m), 1594.0 (s), 1508.6 (m), 1453.9 (m), 1413.4 (s), 1339.7 (w), 1291.7 (w), 1264.2 (s), 1210.9 (w), 1157.5 (m), $1089.4(\mathrm{~m}), 1049.5(\mathrm{~m}), 963.7(\mathrm{w}), 832.0(\mathrm{w}), 777.0(\mathrm{~m}){ }^{1}{ }^{1} \mathrm{H}$ NMR (DMSO-d $\mathrm{d}_{6} \delta 12.99$ (br s, 2 H ), 8.40 $(\mathrm{s}, 2 \mathrm{H}), 7.71(\mathrm{~d}, J=9,2 \mathrm{H}), 7.62(\mathrm{~d}, J=9,2 \mathrm{H}), 7.51(\mathrm{~s}, 2 \mathrm{H}), 7.37(\mathrm{~m}, 2 \mathrm{H}), 7.33(\mathrm{~m}, 2 \mathrm{H}), 7.04(\mathrm{~d}, J=8$, $2 \mathrm{H}), 6.98(\mathrm{~d}, J=7,2 \mathrm{H}), 6.61(\mathrm{~d}, J=1,2 \mathrm{H}), 6.54(\mathrm{~s}, 2 \mathrm{H}), 6.51(\mathrm{~m}, 2 \mathrm{H}), 5.18(\mathrm{t}, J=5,2 \mathrm{H}), 5.11(\mathrm{~s}, 4 \mathrm{H})$, $4.94(\mathrm{t}, J=5,4 \mathrm{H}), 4.74(\mathrm{~d}, J=2,4 \mathrm{H}), 4.54(\mathrm{t}, J=5,4 \mathrm{H}), 4.52(\mathrm{~m}, 4 \mathrm{H}), 4.44(\mathrm{~m}, 4 \mathrm{H}), 4.42(\mathrm{~d}, J=5$, $4 \mathrm{H}), 3.54(\mathrm{t}, \mathrm{J}=2,2 \mathrm{H}) ; \mathrm{MS}(E S I-T O F) 1141.4(\mathrm{M}+1)$; HRMS (ESI-TOF) calcd. for $\mathrm{C}_{62} \mathrm{H}_{57} \mathrm{~N}_{6} \mathrm{O}_{16}$ $[\mathrm{M}+1]^{+}$: 1141.3831 ; found: 1141.3870 .


Bismacrocyclic host (1). To a solution of 2-methyl-6-nitrobenzoic anhydride ( 4.2 mg , 0.13 mmol ), and 4 -dimethylaminopyridine ( $31 \mathrm{mg}, 0.25 \mathrm{mmol}$ ) in acetone ( 20 mL ) was added a solution of $14(45 \mathrm{mg}, 0.039 \mathrm{mmol})$ in acetone $(180 \mathrm{~mL})$ dropwise over 2 h . After the addition was complete, the mixture was stirred at room temperature for 24 h . The solution was concentrated to $\approx 5 \mathrm{~mL}$ and cooled in a refrigerator at $5{ }^{\circ} \mathrm{C}$ overnight. The solid precipitate was
collected on a fine porosity glass frit and washed well with methanol ( $2 \times 4 \mathrm{~mL}$ ) and diethyl ether $(1 \times 4 \mathrm{~mL})$. The solid was then suspended in a boiling solvent system of 1:1:1 acetone/methanol $/ \mathrm{CHCl}_{3}(30 \mathrm{~mL})$. The insoluble precipitate was collected on a fine porosity glass frit and washed with acetone, warm methanol, $20 \%$ methanol/methylene chloride, and diethyl ether to give 12.1 mg (28\%) of 1 as gray-white solid. mp $216^{\circ} \mathrm{C}$ (dec.); IR v 2912 (w), 1719.8 (w), 1595.4 (s), 1508.2 (m), 1458.9 (w), 1413.7 (s), 1363.9 (w), 1292.2 (w), 1264.4 (s), 1230.2 (m), $1211.2(\mathrm{~m}), 1196.0(\mathrm{w}), 1166.3(\mathrm{~m}), 1154.8(\mathrm{~m}), 1109.9(\mathrm{w}), 1089.0(\mathrm{~s}), 1055.8(\mathrm{~s}), 1026.1(\mathrm{w})$, $963.4(\mathrm{w}), 775.7(\mathrm{~m}) ;{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}$ ) $\delta 8.15(\mathrm{~s}, 2 \mathrm{H}), 7.41(\mathrm{~s}, 2 \mathrm{H}), 7.41(\mathrm{~d}, J=8,2 \mathrm{H}), 7.34(\mathrm{~d}, J=$ $9,2 \mathrm{H}$ ), 7.15 (apparent $\mathrm{t}, \mathrm{J}=8,2 \mathrm{H}), 7.06$ (apparent $\mathrm{t}, J=8,2 \mathrm{H}), 7.02(\mathrm{~d}, J=8,2 \mathrm{H}), 6.72$ (d, $J=7$, $2 \mathrm{H}), 6.51(\mathrm{~s}, 2 \mathrm{H}), 6.38(\mathrm{~s}, 2 \mathrm{H}), 6.00(\mathrm{~s}, 2 \mathrm{H}), 5.11(\mathrm{~s}, 4 \mathrm{H}), 4.90(\mathrm{~m}, 4 \mathrm{H}), 4.71(\mathrm{~d}, \mathrm{~J}=2,4 \mathrm{H}), 4.59(\mathrm{~s}$, $4 \mathrm{H}), 4.54(\mathrm{~s}, 4 \mathrm{H}), 4.48(\mathrm{br} \mathrm{s}, 12 \mathrm{H}), 3.59(\mathrm{~s}, 2 \mathrm{H}) ; \mathrm{MS}(\mathrm{MALDI}) 1105.8(\mathrm{M}+1,100), 1127.9\left(\mathrm{M}+\mathrm{Na}^{+}\right.$, 67), 1143.9 (M+K+, 34); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{62} \mathrm{H}_{53} \mathrm{~N}_{6} \mathrm{O}_{14}[\mathrm{M}+1]^{+}: 1105.3620$; found: 1105.3595 .


1-Phthalimidyl-4-azidobutane. To a solution of $17(1.54 \mathrm{~g}, 11.6 \mathrm{mmol})$ in DMF ( 30 mL ) was added potassium phthalimide ( $2.22 \mathrm{~g}, 12.0 \mathrm{mmol}$ ) and $\mathrm{KI}(0.210 \mathrm{~g}, 1.27 \mathrm{mmol})$. The mixture was stirred for 4 days at $70^{\circ} \mathrm{C}$. On the fifth day the mixture was worked up by adding ether ( 40 mL ) and washing with water $(2 \times 40 \mathrm{~mL})$ and brine $(1 \times 40 \mathrm{~mL})$. The ethereal phase was isolated, dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$, and concentrated under reduced pressure to give 1.68 g (59\%) of 1-phthalimidyl4 -azidobutane as yellow oil. All spectroscopic data matched the values reported in literature. ${ }^{[9]}$


Bis[4-(N-4-azidobutyl)aminomethylphenyl] ether (20). A solution of 19 ( $0.182 \mathrm{~g}, 0.807 \mathrm{mmol}$ ) and 18 ( $0.317 \mathrm{~g}, 2.83 \mathrm{mmol}$ ) in toluene ( 40 mL ) was heated at reflux under a Dean-Stark trap. Imine formation was monitored by IR spectroscopy and the reaction was complete after 24 h . The mixture was cooled to rt, and $\mathrm{NaBH}_{4}(0.153 \mathrm{~g}, 4.04 \mathrm{mmol}$ ) in $\mathrm{EtOH}(5 \mathrm{~mL})$ was added. The mixture was stirred at ambient temperature and then heated at reflux. After 5 days the solution was concentrated under reduced pressure. The crude brown residue was partitioned between aqueous $\mathrm{NaOH}(1.5 \mathrm{M}, 30 \mathrm{~mL})$ and $\mathrm{CH}_{2} \mathrm{Cl}_{2}(30 \mathrm{~mL})$. The aqueous layer was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (2X). The organic layers were combined and concentrated, giving a crude brown oil. The product was purified by flash chromatography ( $7 \% \mathrm{MeOH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to give $48.6 \mathrm{mg}(14 \%)$ of 20 as an oil. IR v 2933 (m), 2863 ( w ), 2093 ( s$), 1601$ ( m$), 1500$ ( s$), 1237$ ( s$), 1167$ (m), 1115 (m), 1015 (w), 874 (m), $834(\mathrm{~m}) ;{ }^{1} \mathrm{H}$ NMR (acetone- $\left.\mathrm{d}_{6}\right) 7.356(\mathrm{~d}, J=9,4 \mathrm{H}), 6.933(\mathrm{~d}, J=9,4 \mathrm{H}), 3.754(\mathrm{~s}, 4 \mathrm{H}), 3.346(\mathrm{t}$, $J=7,4 \mathrm{H}), 2.639(\mathrm{t}, J=7,4 \mathrm{H}), 1.680(\mathrm{~m}, 4 \mathrm{H}), 1.592(\mathrm{~m}, 4 \mathrm{H}) ; \mathrm{MS}(\mathrm{ESI}) 423.3(\mathrm{M}+1,32), 212.1$ ([M+2]/2, 70); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{22} \mathrm{H}_{31} \mathrm{~N}_{8} \mathrm{O}[\mathrm{M}+1]^{+}: 423.2621$; found: 423.2636 .


2
Bis[4-( N -4-azidobutyl)ammoniummethylphenyl] ether bis(hexafluorophosphate) (2). Diamine 20 ( $0.153 \mathrm{~g}, 0.362 \mathrm{mmol}$ ) was dissolved in a minimum volume of $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The mixture was stirred while HCl in dioxane ( $0.40 \mathrm{~mL}, 1.6 \mathrm{mmol}$ ) was added dropwise. The chloride salt precipitated and the solvent was removed under reduced pressure. This material was dissolved a minimum volume of $1: 1 \mathrm{EtOH} /$ water and a solution of $\mathrm{NH}_{4} \mathrm{PF}_{6}(0.725 \mathrm{~g}, 4.45 \mathrm{mmol}$ ) in a minimum volume of water was added with stirring. The mixture was concentrated to a white solid, which was collected by vacuum filtration and washed with cold water to give 186 mg (72\%) of 2 white solid. $\mathrm{mp} 213-215^{\circ} \mathrm{C}$; IR v 3264 (m), 3240 (w), 2938 (w), 2099 ( s$), 1508$ (m), 1605 (w), 1583 (m), 1421 (m), 1257 (s), 1174 (w), 894 (m), 845 (vs), 767 (s); ${ }^{1 H}$ NMR (acetone- $d_{6}$ ) 88.194 (br $\mathrm{s}, 4 \mathrm{H}), 7.656(\mathrm{~d}, J=9,4 \mathrm{H}), 7.118(\mathrm{~d}, J=9,4 \mathrm{H}), 4.613(\mathrm{~s}, 4 \mathrm{H}), 3.498(\mathrm{t}, J=8,4 \mathrm{H}), 3.441(\mathrm{t}, J=7,4 \mathrm{H})$, $2.167(\mathrm{~m}, 4 \mathrm{H}), 1.759(\mathrm{~m}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (CD $\left.{ }_{3} \mathrm{OD}\right) \delta 159.23,133.03,127.77,120.49,51.74,51.71$, 48.04, 26.8, 24.52; MS (ESI-TOF) 569.2 ( M - $\mathrm{PF}_{6}{ }^{-}$, 39), 238.1 (76), 212.1 ( $\mathrm{M}-\left[\mathrm{PF}_{6}{ }^{-}\right]_{2,}$ 100); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{22} \mathrm{H}_{32} \mathrm{~N}_{8} \mathrm{OF}_{6} \mathrm{P}\left[\mathrm{M}-\mathrm{PF}_{6}{ }^{-}\right]^{+}: 569.2341$; found: 569.2325 .


22
3-(3-Azidopropyl)pyridine (22). A stirred solution of 21 ( $5.192 \mathrm{~g}, 37.90 \mathrm{mmol}$ ) and triethyl amine ( $6.8 \mathrm{~mL}, 49 \mathrm{mmol}$ ) in DMF ( 25 mL ) was cooled in an ice-water bath while methanesulfonyl chloride ( $3.8 \mathrm{~mL}, 49 \mathrm{mmol}$ ) was added dropwise over 15 min . The mixture was warmed to ambient temperature and stirred for 135 min . Methanol ( 1 mL ) was added and, after stirring for 25 min , sodium azide ( $3.33 \mathrm{~g}, 51.2 \mathrm{mmol}$ ) was added. The resulting mixture was stirred for an additional 2.5 d . The mixture was partitioned between diethyl ether ( 50 mL ) and water ( 75 mL ). The aqueous layer was extracted with diethyl ether $(2 \times 25)$ and the combined organic layers were washed with water, dilute aqueous ammonium chloride, water, and brine ( $2 \times$ ), dried $\left(\mathrm{Na}_{2} \mathrm{SO}_{4}\right)$, and concentrated under reduced pressure to give $5.042 \mathrm{~g}(82 \%)$ of 22 as coppercolored liquid that was pure by TLC and NMR: IR v 3375.6(br w), 2936.6 (w), 2866.6(w), 2091.7 (s), 1576.4 (w), 1479.0 (w), 1452.0 (w), 1423.3 (m), 1348.3 (w), 1257.2 (m), 1191.3 (w), 1044.9 (w), $1027.2(\mathrm{w}), 825.2(\mathrm{w}), 791.3(\mathrm{~m}), 712.2(\mathrm{~s}), 629.7(\mathrm{~m}) ;{ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 8.47(\mathrm{~m}, 2 \mathrm{H}), 7.51(\mathrm{dt}, J=$ $8, J=2.0,1 \mathrm{H}), 7.23$ (ddd, $J=8, J=5, J=1,1 \mathrm{H}), 3.32(\mathrm{t}, J=7,2 \mathrm{H}), 2.72(\mathrm{t}, J=8,2 \mathrm{H}), 1.92(\mathrm{~m}, 2 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(\mathrm{CDCl}_{3}\right) \delta 149.69,147.49,135.95,135.65,123.21,50.23,29.92,29.68$; MS (ESI) 162.7 $(\mathrm{M}+1)$. All spectroscopic data matched the values reported in literature. ${ }^{[10]}$


Bis[2-[3-(3-Azidopropyl)pyridinium]ethoxy]ethyl] ether bis(hexafluorophosphate) (3). A solution of dibromide 23 ( $0.6737 \mathrm{~g}, 2.105 \mathrm{mmol}$ ) and 22 ( $0.692 \mathrm{~g}, 4.27 \mathrm{mmol}$ ) in DMF ( 5 mL ) was
stirred at ambient temperature for 14 d . The solution was concentrated under reduced pressure to give a brown oil. To the oil was added a solution of $\mathrm{NH}_{4} \mathrm{PF}_{6}(4.940 \mathrm{~g}, 30.3 \mathrm{mmol})$ in a minimal volume of water. Methylene chloride was then added and the organic layer was separated and concentrated under reduced pressure. The resulting brown oil was purified on a reversed-phase preparatory TLC plate (1:5:94 ammonium hydroxide/ $\mathrm{CH}_{3} \mathrm{OH} / \mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). The UV-active band with the highest $R_{\mathrm{f}}$ value was isolated to give $205.4 \mathrm{mg}(21 \%)$ of 3 as dark yellow oil. IR $v 2926.0$ (s), $2855.1(\mathrm{~m}), 2102.2(\mathrm{~s}), 1505.8(\mathrm{~m}), 1464.8(\mathrm{~m}), 1353.0(\mathrm{w}), 1256.1(\mathrm{~m}), 1118.2(\mathrm{~m}), 834.9(\mathrm{vs}), 68 \mathrm{~g} .6$ (s); ${ }^{1} \mathrm{H}$ NMR (Acetone- $\mathrm{d}_{6}$ ) $\delta 9.03(\mathrm{~s}, 2 \mathrm{H}), 8.95(\mathrm{~d}, J=6,2 \mathrm{H}), 8.65(\mathrm{~d}, J=8,2 \mathrm{H}), 8.28(\mathrm{~m}, 2 \mathrm{H}), 4.95$ $(\mathrm{t}, J=5,4 \mathrm{H}), 4.10(\mathrm{t}, J=5,4 \mathrm{H}), 3.64(\mathrm{~m}, 4 \mathrm{H}), 3.53(\mathrm{~m}, 4 \mathrm{H}), 3.48(\mathrm{t}, J=7,4 \mathrm{H}), 3.05(\mathrm{t}, \mathrm{J}=8,4 \mathrm{H}), 2.06$ ( $\mathrm{m}, 4 \mathrm{H}$ ); MS (ESI) $629.3\left(\mathrm{M}-\mathrm{PF}_{6}{ }^{-}, 100\right), 214.1\left(\mathrm{M}-\left[\mathrm{PF}_{6}{ }^{-}\right]_{2}, 83\right.$ ); HRMS (ESI-TOF) calcd. for $\mathrm{C}_{24} \mathrm{H}_{36} \mathrm{~N}_{8} \mathrm{O}_{3} \mathrm{~F}_{6} \mathrm{P}\left[\mathrm{M}-\mathrm{PF}_{6}{ }^{-}\right]^{+}: 629.2552$; found: 629.2535 .


TLC reaction sequence using host $\mathbf{1}$ and guest $\mathbf{2}$. A mixture of $\mathbf{1}(2.6 \mathrm{mg}, 2.4 \mu \mathrm{~mol}), \mathbf{2}(1.7 \mathrm{mg}$, $2.4 \mu \mathrm{~mol}),\left[\mathrm{CH}_{3} \mathrm{CN}_{4} \mathrm{CuPF}_{6}(1.4 \mathrm{mg}, 3.8 \mu \mathrm{~mol}), \mathrm{CH}_{3} \mathrm{CN}(1 \mathrm{~mL}), \mathrm{CH}_{2} \mathrm{Cl}_{2}(1 \mathrm{~mL})\right.$, and $\mathrm{CH}_{3} \mathrm{OH}(0.1 \mathrm{~mL})$ was stirred at ambient temperature for 20 h . Ascorbic acid ( $5.3 \mathrm{mg}, 30 . \mu \mathrm{mol}$ ) was added and the mixture was stirred for 5 more days. The solvent was then removed under reduced pressure. The solid was dissolved in a solvent mixture of $\mathrm{H}_{2} \mathrm{O}(0.2 \mathrm{~mL})$, THF ( 0.3 mL ), and $\mathrm{EtOH}(0.3 \mathrm{~mL})$ and $\mathrm{KOH}(10 \mathrm{mg}, 179 \mu \mathrm{~mol})$ was added. The mixture was heated at reflux overnight and then concentrated to a small volume. The pH of the solution was adjusted to neutral by the addition of $1 \mathrm{M} \mathrm{HCl}(0.15 \mathrm{~mL})$ and the precipitate was collected and dried to give $5.6 \mathrm{mg}(124 \%)$ of the crude product as a tan solid. The excess mass was assumed to be from cation sequestration and $\mathrm{PF}_{6}{ }^{-}$anions which was consistent with the IR and mass spectral data. IR v 2957.5 (m), 2918.3 ( s ), 2849.5 (M), 1769.9 (w), 1733.5 (m), 1716.0 (m), 1698.6 (m), 1683.5 (m), 1596.1 (w), 1560.1 (w), 1456.1 (w), 1415.7 (m), $1394.5(\mathrm{w}), 1362.6$ (w), 1265.8 (m), 1245.3 (m), 1210.1 (m), $1152.1(\mathrm{~m})$, 1089.8 (m), 1066.5 (m), 1053.8 (m), 1021.2 (w), 844.6 (vs), 713.8 (w), 669.3 (w), 619.0 (w), 560.0 $(\mathrm{m}) ;{ }^{1} \mathrm{H}$ NMR (DMSO- $\mathrm{d}_{6}, 45^{\circ} \mathrm{C}$ ) $\delta 8.85$ (br s, 4H), $8.38(\mathrm{~s}, 2 \mathrm{H}), 8.19(\mathrm{~s}, 2 \mathrm{H}), 7.71(\operatorname{app} \mathrm{~s}, 2 \mathrm{H}), 7.60$
(app s, 2H), $7.50(\mathrm{apps}, 6 \mathrm{H}), 7.34$ (app s, 4H), $7.03(\mathrm{~m}, 10 \mathrm{H}), 6.57$ (app s, 6H), $5.10(\mathrm{app} \mathrm{s}, 6 \mathrm{H}), 4.92$ (app s, 4H), 4.55-4.40 (m, 16H), 4.09 (app s, 4H), $2.95(a p p s, 4 H), 1.90(a p p ~ s, 4 H), 1.61(a p p ~ s$, 4H), 1.22 (app s, 4H), 0.85 (app s, 4H); MS (MALDI) $1563.9\left(\mathrm{M}+\mathrm{H}^{+}, 88\right), 1584.9\left(\mathrm{M}^{+} \mathrm{Na}^{+}, 28\right), 1601.9$ $\left(\mathrm{M}+\mathrm{K}^{+}, 52\right), 1626.8\left(\mathrm{M}+\mathrm{Cu}^{+}, 42\right), 425.1(100)$.

## References

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Automation directory:
Pulse Sequence: s2pul





EEF1001
Automation directory: Pulse Sequence: s2pul


15








acetone- $d_{5}$


File: xp
Pulse Sequence: s2pul


${ }^{-}$


autoont ion sirectory: Room temperature $\left(\sim 24^{\circ} \mathrm{C}\right)$ spectrum Pulse sequence: s2pul



## $45^{\circ} \mathrm{C}$ spectrum

```
EEF1087
un at 45 deg. C
Automation directory:
Pulse Sequence: s2pul
```



## Product from the TLC sequence of 1 and 2 .

## Applied Biosystems Voyager System 4289




