

Supporting Information

for

Starazo triple switches – synthesis of unsymmetrical 1,3,5-tris(arylazo)benzenes

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Beilstein J. Org. Chem. 2020, 16, 22-31. doi:10.3762/bjoc.16.4

Additional synthetic procedures, isomerization experiments, and $^{1}\mathrm{H}/^{13}\mathrm{C}$ NMR spectra

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General information

Chemicals were used as purchased from Sigma-Aldrich, Acros Organics, Alfa Aesar, and TCI Europe. Anhydrous solvents were purchased from Acros Organics. Technical grade solvents used during workup and purification were distilled prior to use. Air- and/or moisture-sensitive reactions were carried out under Schlenk conditions or in a nitrogen-filled glovebox. Solids were dried under high vacuum (oil pump, ca. 10⁻³ mbar) at rt, 50 °C, or 60 °C, as necessary. Flash column chromatography and column chromatography were carried out using Silica 60 M (0.04–0.063 mm) from Macherey-Nagel. Thin-layer chromatography was performed on Polygram® SIL G/UV254 from Macherey-Nagel. NMR spectra were recorded on a Bruker Avance II 200 MHz, Avance II 400 MHz, or Avance III 400 MHz HD spectrometer at rt, if not stated otherwise. Chemical shifts are reported in ppm relative to the solvent peak, coupling constants J are reported in Hz. Deuterated solvents were obtained from Deutero GmbH or Eurisotop. For all azobenzenes, the thermodynamically more stable E-isomer is reported, if not noted otherwise. HRESIMS spectra were recorded on a Bruker Daltonics Micro TOF. Melting points were measured on a Krüss M5000 capillary melting point meter with a heating rate of 1 °C/min. Solvents for UV-vis spectroscopy were purchased from Merck (Uvasol® quality). The UV-vis measurements were carried out with a SPECORD® 200 PLUS spectrophotometer equipped with two automatic eight-fold cell changers and a Peltier element thermostat system (0.1 °C accuracy) by Analytik Jena. The system was operated with the ASpect UV software by Analytik Jena. The sample solutions were measured in QS High-precision Cells made of Quartz Suprasil® with a light path of 10 mm by Hellma Analytics.

Irradiation experiments

Irradiation of the NMR or UV-vis samples was conducted in an in-house-built box using high-power LEDs by Lumitronix or Nichia. After the given irradiation time, the samples were immediately placed and measured in the corresponding spectrometer.

Table S1: LEDs used for all irradiation experiments and their specifications.

λ_{max} (nm)	$\Delta\lambda_{FWHM}$ (nm)	luminous flux (mW)	product name
365	0	780	NCSU276AT-
303	9		U365
448	20	520	LXML-PR01-
			0500

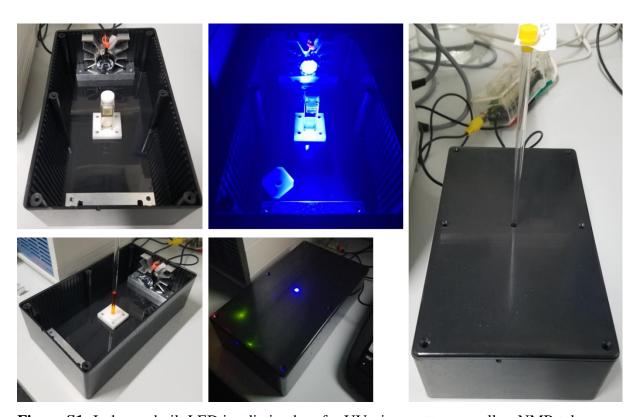


Figure S1: In-house-built LED irradiation box for UV-vis cuvettes as well as NMR tubes.

Synthesis

tert-Butyl 1-(4-methoxyphenyl)hydrazine-1-carboxylate [1]:

4-Iodoanisole (4.02 g, 16.8 mmol, 1.00 equiv), *tert*-butyl carbazate (2.70 g, 20.2 mmol, 1.20 equiv), CuI (3.14 g, 16.4 mmol, 0.975 equiv), and Cs₂CO₃ (8.25 g, 25.1 mmol, 1.49 equiv) were suspended in DMSO (15 mL), and the mixture was stirred at 50 °C for 40 min. Afterwards, the reaction mixture was diluted with EtOAc, filtered through a plug of silica gel (EtOAc wash), and the concentrated filtrate was purified by flash chromatography (SiO₂, cyclohexane/EtOAc, 2:1, v/v) to yield the product as pale yellow oil (2.46 g, 61%). ¹H NMR (400 MHz, CDCl₃) δ 7.30 (d, J = 9.0 Hz, 2H), 6.84 (d, J = 9.0 Hz, 2H), 4.44 (s, 2H), 3.79 (s, 3H), 1.47 (s, 9H). Analytical data corresponded to the literature [1].

tert-Butyl 1-(4-cyanophenyl)hydrazine-1-carboxylate [2]:

An oven-dried Schlenk tube under a nitrogen atmosphere was charged with 4-bromobenzonitrile (1.85 g, 10.1 mmol, 1.00 equiv), tert-butyl carbazate (2.70 g, 20.0 mmol, 1.99 equiv), Cs_2CO_3 (3.27 g, 10.0 mmol, 0.997 equiv), and $Pd(dppf)Cl_2 \cdot CH_2Cl_2$ (85 mg, 0.10 mmol, 1.0 mol %) as well as dry toluene (20 mL). The tube was sealed and the suspension was stirred at 100 °C for 18.5 h. After cooling to rt, the mixture was filtered through a plug of silica gel (EtOAc wash), and concentrated to yield a brown oil, which was purified by flash column chromatography (SiO₂, cyclohexane/EtOAc, 3:1, v/v) to yield an off-white solid (1.83 g, 78%). ¹H NMR (400 MHz, CDCl₃) δ 7.75 (d, J = 8.9 Hz, 1H), 7.57 (d, J = 8.9 Hz, 1H), 4.38 (s, 2H), 1.55 (s, 9H). m.p. 107 °C. Analytical data corresponded to the literature [3].

tert-Butyl 1-phenylhydrazine-1-carboxylate [1]:

Iodobenzene (5.02 g, 24.6 mmol, 1.00 equiv), *tert*-butyl carbazate (3.96 g, 29.4 mmol, 1.19 equiv), CuI (237 mg, 1.24 mmol, 5.03 mol %), and Cs₂CO₃ (12.2 g, 37.0 mmol, 1.51 equiv) were suspended in DMSO (20 mL), and the mixture was stirred at 50 °C for 2.5 h. The mixture was filtered through a plug of silica gel (EtOAc wash), and the solvents were evaporated under reduced pressure. The crude product was purified by flash column chromatography (SiO₂, cyclohexane/EtOAc, 5:1, v/v) to yield the product as a yellow oil (3.45 g, 67%). ¹H NMR (400 MHz, CDCl₃) δ 7.48 – 7.43 (m,

2H), 7.34 – 7.27 (m, 2H), 7.14 – 7.08 (m, 1H), 4.44 (s, 2H), 1.50 (s, 9H). Analytical data corresponded to the literature [3].

N-(3,5-Dinitrophenyl)acetamide (5):

A suspension of 3,5-dinitrophenylaniline (15.9 g, 86.7 mmol, 1.00 equiv) and Ac₂O (17.0 ml, 179 mmol, 2.10 equiv) in AcOH (40 mL) was stirred for a few minutes during which it warmed up and turned pale yellow. When the reaction mixture had cooled back to rt, the precipitate was filtered off, washed with water, and dried in vacuum to yield the product as yellow solid (18.5 g, 95 %). ¹H NMR (400 MHz, DMSO- d_6) δ 10.83 (s, 1H), 8.80 (d, J = 2.1 Hz, 2H), 8.47 (t, J = 2.1 Hz, 1H), 2.14 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 169.7, 148.2, 141.2, 118.1, 112.0, 24.1. HRMS (ESI): m/z, for C₈H₇N₃O₅Na⁺⁺; calcd. 248.0278, found 248.0283. m.p. 191 °C.

N-(3-Amino-5-nitrophenyl)acetamide (6)[4]:

To a refluxing solution of **5** (10.0 g, 44.4 mmol, 1.00 equiv) in EtOH (270 mL), a solution of aqueous (NH₄)₂S (20%, 46.0 mL, 135 mmol, 3.00 equiv) in EtOH (22 mL) was added dropwise over 75 min. The reaction mixture was refluxed for further 30 min and was allowed to cool to rt afterwards. The yellow precipitate formed was filtered off, and the filtrate was concentrated under reduced pressure. The residue was treated with water (220 mL) and conc HCl (27 mL) and was filtered. The filtrate was neutralized with aq NaOH (10%, ca. 140 mL), and the precipitate formed was filtered off, washed with H₂O, and recrystallized from H₂O/EtOH, 10:1, v/v. The obtained orange needles were filtered off, washed with water, and dried in vacuum. The product was obtained as orange needles (5.67 g, 65%). ¹H NMR (200 MHz, DMSO- d_6) δ 10.07 (s, 1H), 7.63 (t, J = 2.0 Hz, 1H), 7.22 (t, J = 2.0 Hz, 1H), 7.07 (t, J = 2.2 Hz, 1H), 5.83 (s, 2H), 2.04 (s, 3H). m.p. 203 °C (decomposition). Analytical data corresponded to the literature [4].

N-(3-Nitroso-5-nitrophenyl)acetamide (7) [5]:

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extracted with EtOAc (170 mL), and the combined organic phases were dried over MgSO₄, filtered, and concentrated to yield a green/brown solid (1.06 g, 99%). The product was used without further purification.

3-Acetamido-5-nitroazobenzene (8):

A suspension of **7** (142 mg, 680 μ mol, 1.50 equiv) in AcOH (1 mL) was degassed with a nitrogen stream for 5 min, and freshly distilled aniline (42 μ L, 0.46 mmol, 1.00 equiv) was added. The reaction mixture was stirred at rt under a nitrogen atmosphere for ca 24 h and water (5 mL) was added afterwards. Following this, the brown

precipitate was filtered and washed with water. The crude product was dissolved in hot EtOH, cooled, and the brown impurity was filtered off. The red filtrate was concentrated under reduced pressure to yield a brown solid (114 mg, 87%). Alternatively, the crude product could be purified by column chromatography (SiO₂, toluene/EtOAc; 1:1). ¹H NMR (400 MHz, DMSO- d_6) δ 10.67 (s, 1H), 8.70 (t, J = 2.1 Hz, 1H), 8.47 (t, J = 1.8 Hz, 1H), 8.25 (t, J = 2.0 Hz, 1H), 8.02 – 7.89 (m, 2H), 7.74 – 7.54 (m, 3H), 2.14 (s, 3H). ¹³C NMR (101 MHz, DMSO- d_6) δ 169.3, 152.3, 151.5, 148.8, 141.2, 132.6, 129.6, 123.0, 118.3, 114.8, 110.7, 24.1. HRMS (ESI): m/z for $C_{14}H_{12}N_4O_3Na^+$; calcd. 307.0801, found 307.0800. m.p. 193 °C.

3,5-Dibromoazobenzene (11) [6]:

To a solution of 3,5-dibromoaniline (2.01 g, 8.01 mmol, 1.00 equiv) in AcOH (10 mL),

nitrosobenzene (1.07 g, 9.69 mmol, 1.21 equiv) was added, and the mixture was stirred at rt for 25 h. The precipitate was filtered off and consecutively washed with AcOH and water, and was dried in vacuum to yield a brown solid (1.83 g, 67%). 1 H NMR (400 MHz, CDCl₃) δ

8.01 (d, J = 1.8 Hz, 2H), 7.95 – 7.89 (m, 2H), 7.76 (t, J = 1.8 Hz, 1H), 7.59 – 7.49 (m, 3H). ¹³C NMR (101 MHz, CDCl₃) δ 154.1, 152.3, 135.9, 132.2, 129.4, 125.0, 123.6, 123.4. HRMS (ESI): m/z for [M+H]⁺; calcd. 338.9127, found 338.9132. m.p. 110 – 114 °C (decomposition). Analytical data corresponded to the literature [6].

3,5-Bis(4-methoxyphenylhydrazo)azobenzene 13:

In a nitrogen-filled glovebox, an oven-dried Schlenk tube was charged with $P(t\text{-Bu})_3$ (5 μ L, 20 μ mol, 5 mol %) and tert-butyl-1-(4-methoxyphenyl)hydrazine-1-carboxylate (209 mg, 877 μ mol, 2.18 equiv) in dry toluene (5 mL). The

following operations were carried out in a fume hood under Schlenk-conditions. **11** (137 mg, 403 μmol, 1.00 equiv), Cs₂CO₃ (200 mg, 608 μmol, 1.51 equiv), and Pd(OAc)₂ (5 mg, 22 μmol, 5 mol %) were added. The mixture was stirred at rt for 30 min and the tube was sealed and heated to 110 °C for 2 h and 15 min afterwards. The cooled mixture was filtered through a plug of silica gel (EtOAc wash), and the concentrated filtrate was purified by flash column chromatography (SiO₂, cyclohexane/EtOAc, 2:1, v/v) to yield a yellow solid (152 mg, 58%). ¹H NMR (400 MHz, CDCl₃) δ 7.91 – 7.77 (m, 2H), 7.53 – 7.38 (m, 7H), 6.99 (d, J = 2.1 Hz, 2H), 6.94 – 6.70 (m, 4H), 6.48 (s, 2H), 6.41 (t, J = 2.1 Hz, 1H), 3.79 (s, 6H), 1.37 (s, 18H). ¹³C NMR (101 MHz, CDCl₃) δ 157.0, 154.6, 154.3, 152.7, 150.2, 136.0, 131.1, 129.2, 124.0, 123.0, 113.9, 101.0, 100.0, 82.3, 55.6, 28.3. HRMS (ESI): m/z for C₃₆H₄₂N₆O₆Na⁺; calcd. 677.3058, found 677.3059. m.p. 93 °C.

Bis-3,5-(4-methoxyphenylazo)azobenzene (14):

An oven-dried Schlenk-tube under a N_2 -atmosphere was charged with 13 (46 mg, 70 μ mol, 1.0 equiv), CuI (55 mg, 29 μ mol, 4.1 equiv), Cs₂CO₃ (95 mg, 29 μ mol, 4.2 equiv), and dry DMF (1 mL). The tube was sealed, and the mixture was heated to 140 °C for 3 d. After cooling to rt,

the mixture was filtered through a plug of silica gel (EtOAc wash), concentrated, and the residue was purified by flash column chromatography (SiO₂, toluene/cyclohexane, 1:1 to 1:0, v/v) to yield the product as orange solid (6 mg, 19%). 1 H NMR (400 MHz, CDCl₃) δ 8.51 (s, 3H), 8.07 -7.97 (m, 6H), 7.61 -7.48 (m, 3H), 7.05 (d, J = 9.0 Hz, 4H), 3.91 (s, 6H). 13 C NMR (101 MHz, CDCl₃) δ 162.7, 154.2, 154.0, 152.7, 147.0, 131.6, 129.3, 125.3, 123.3, 118.6, 118.2, 114.5, 55.8. HRMS (ESI): m/z for C₂₆H₂₂N₆O₂Na⁺; calcd. 473.1696, found 473.1693.

¹H NMR isomerization experiments

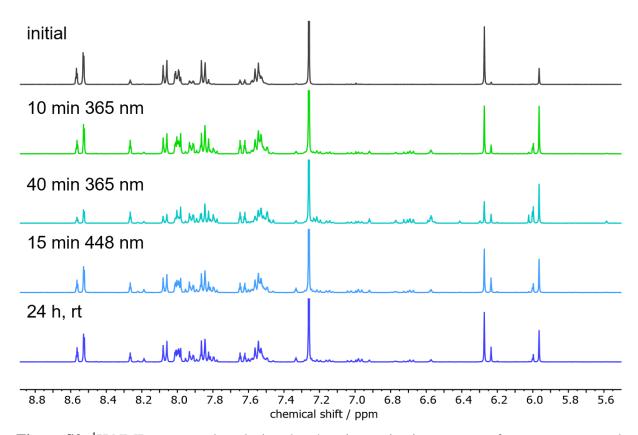


Figure S2: ¹H NMR spectra taken during the photoisomerization process of starazo compound **3a**. Upon irradiation at 365 nm, isomerization to multiple photoisomers could be observed, which could not be unambiguously assigned. When irradiating at 448 nm, the initial spectrum could be roughly restored (light blue spectrum). After standing at rt in the dark overnight, slow thermal back-isomerization could be observed (blue spectrum).

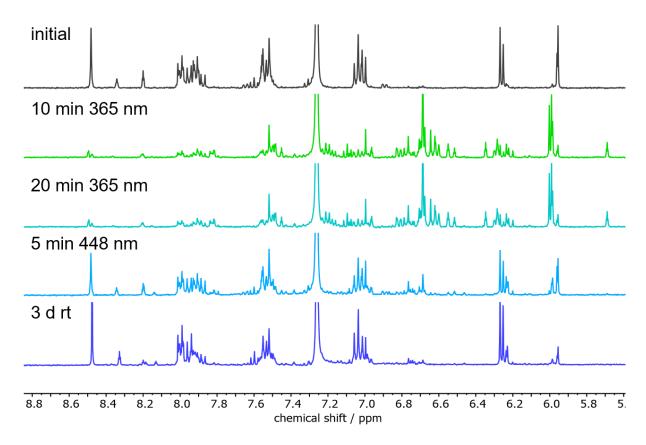
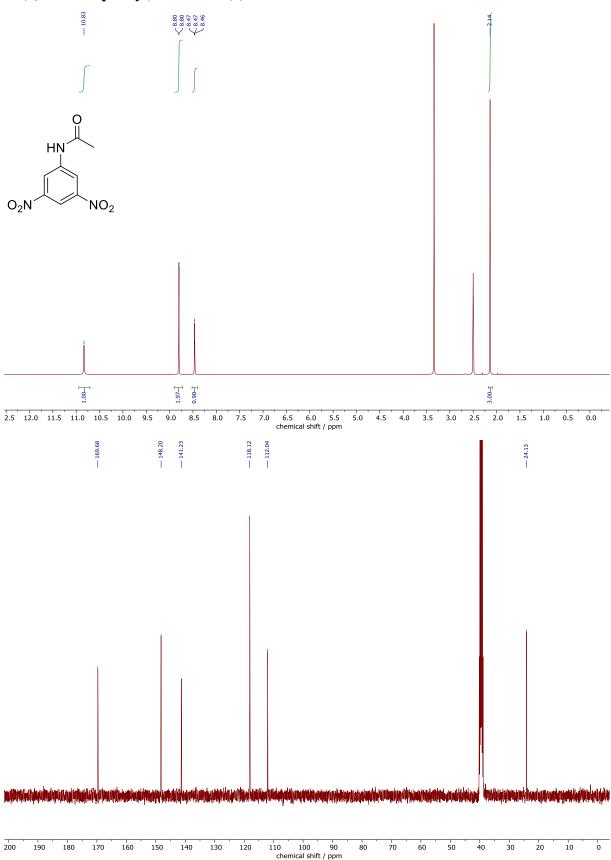
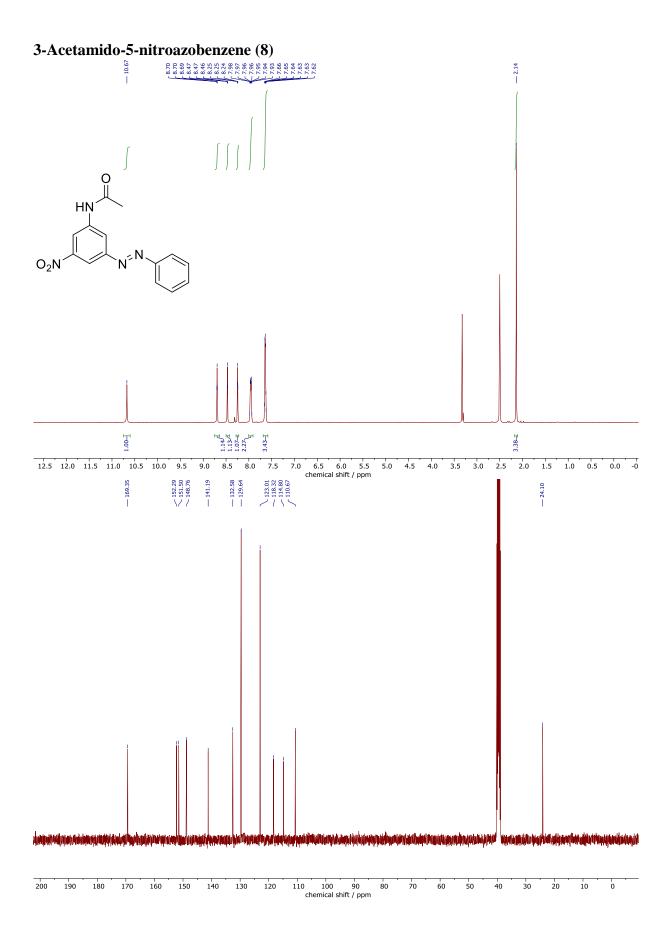


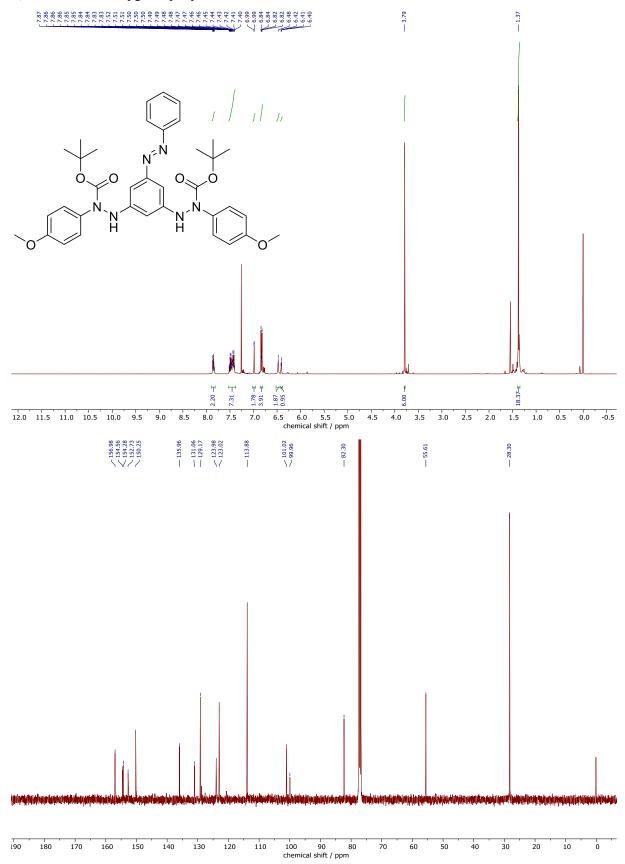
Figure S3: ¹H NMR spectra taken during the photoisomerization process of starazo species **3b**. Upon irradiation at 365 nm, complex spectra of multiple isomerization states were obtained (green and cyan spectra), which could not be assigned unambiguously. When irradiating at 448 nm, the initial spectrum could be roughly restored (light blue spectrum). After standing at rt in the dark for 3 d, the initial spectrum was almost restored (blue spectrum).

1 H and 13 C $\{^{1}$ H $\}$ -NMR spectra N-(3,5-Dinitrophenyl)acetamide (5)

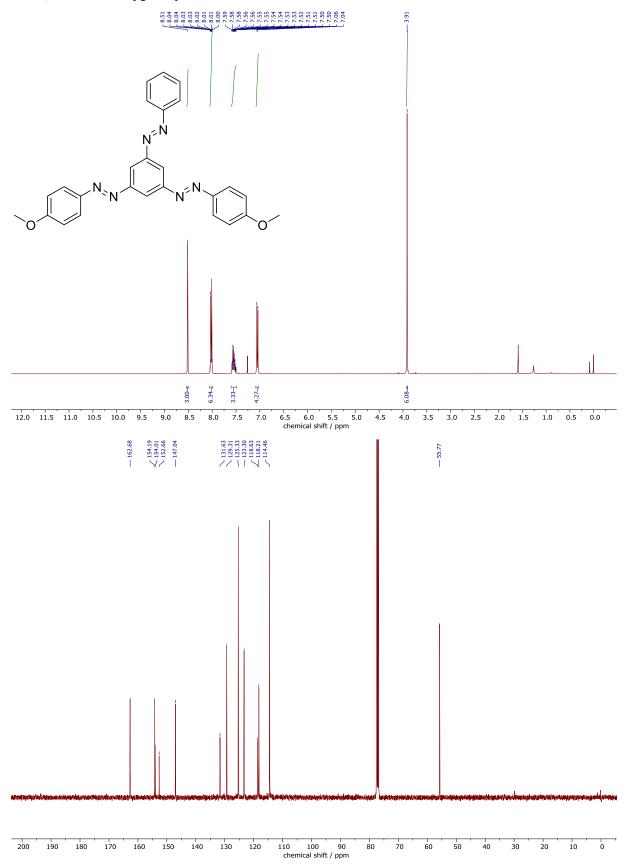


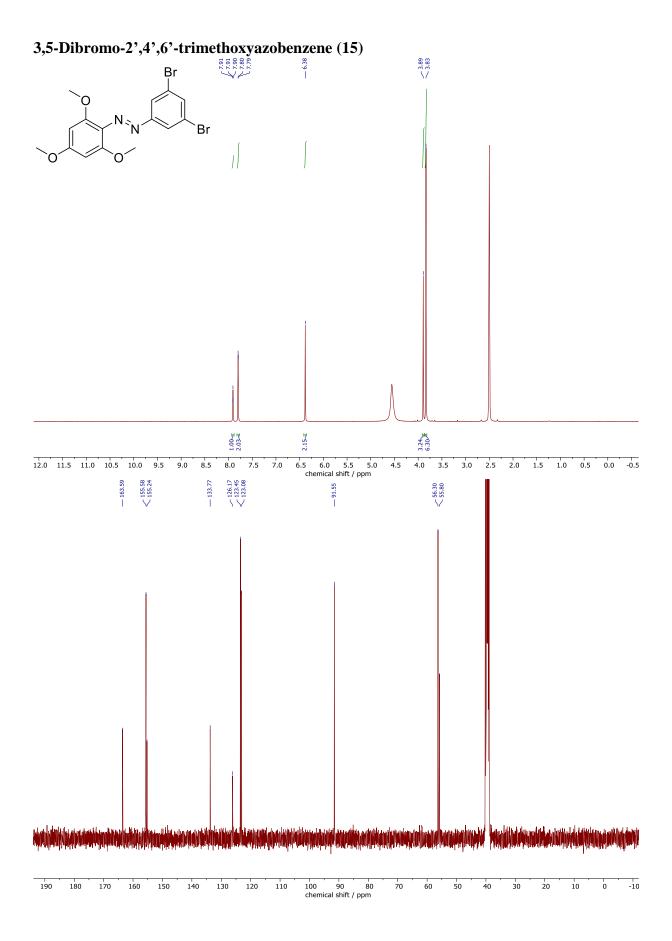


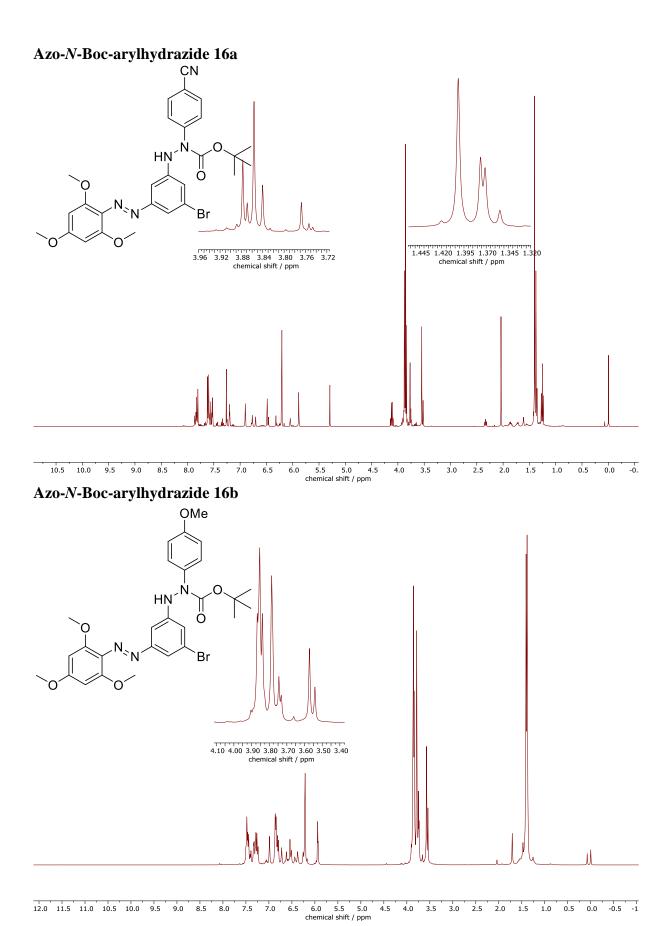
${\it 3,5-Bis} (4-methoxy phenyl hydrazo) azobenzene~13$



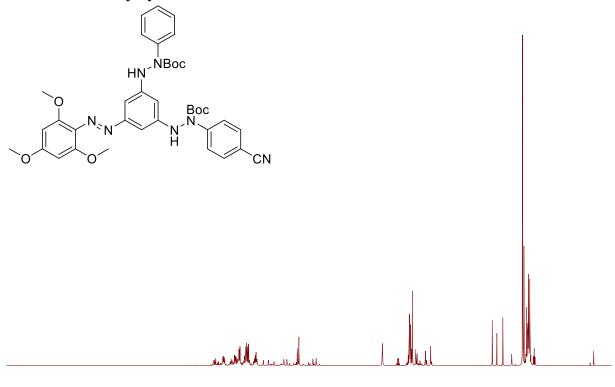
 $Bis-3, 5-(4-methoxyphenylazo) azobenzene\ (14)$

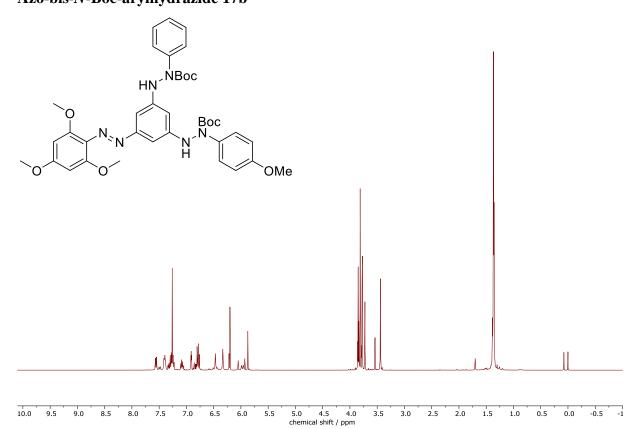


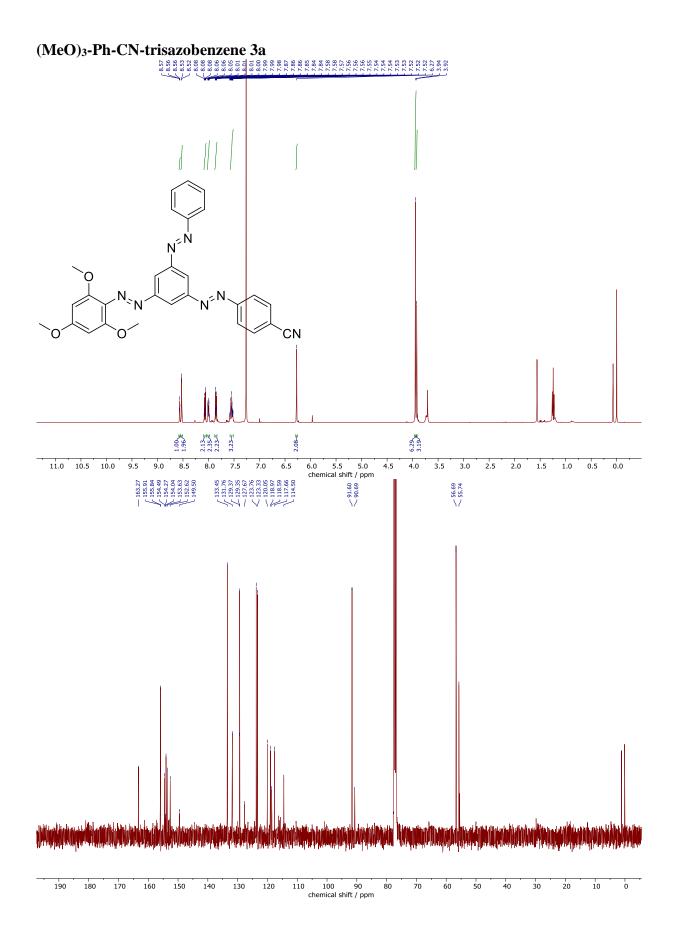


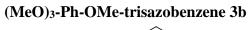


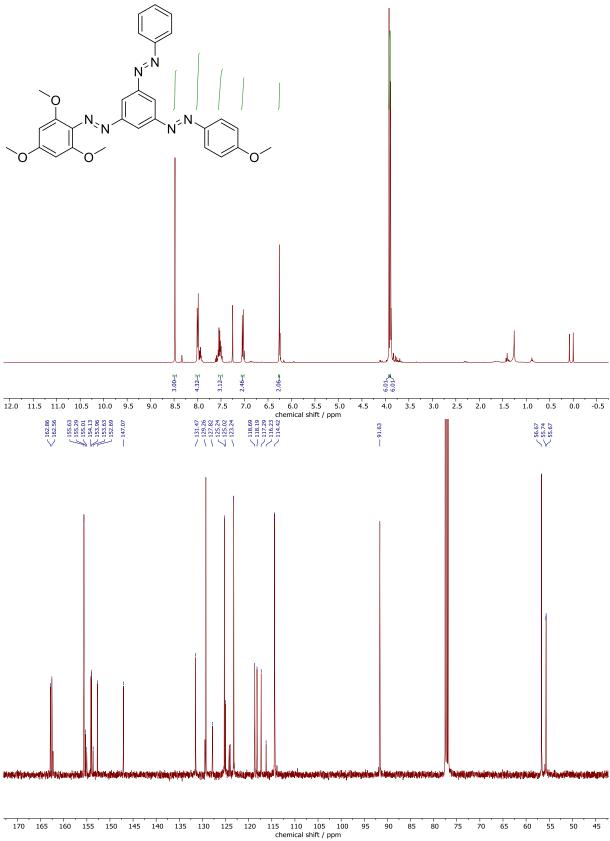
Azo-bis-N-Boc-arylhydrazide 17a











References

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