



Supporting Information

for

Ferrocenoyl-adenines: substituent effects on regioselective acylation

Mateja Toma, Gabrijel Zubčić, Jasmina Lapić, Senka Djaković, Davor Šakić
and Valerije Vrček

Beilstein J. Org. Chem. **2022**, *18*, 1270–1277. [doi:10.3762/bjoc.18.133](https://doi.org/10.3762/bjoc.18.133)

Details on experimental procedures, DFT calculated energies and optimized coordinates for transition state structures and reactants, and the results of in situ ¹H NMR monitoring

Table of contents

1. Experimental part	S2
1.1. Methods and materials	S2
1.2. Preparation of ferrocenoylated adenines	S2
1.3. In situ ¹ H NMR monitoring of the reaction between adenine anions and FcCOCl	S8
1.4. Selected ¹ H NMR spectra of the reaction mixture (regiospecific mode)	S10
2. Computational part	S11
2.1. Computational techniques and methods	S11
2.2. Calculated energies and barriers	S12
2.3. S _N 2-type and alternative reaction mechanisms for <i>N</i> 7-ferrocenylation of adenines	S13
2.4. Calculation of Sterimol and % V _{Bur} parameters	S14
2.5. Optimized coordinates of reactants and transition states	S17
3. References	S31

1. Experimental part

1.1. Methods and materials

The syntheses were carried out under argon atmosphere in dry solvents. The IR spectra were recorded in chloroform with a Cary 630 FTIR Spectrometer (Agilent). The mass spectra were acquired using 4800 MALDI TOF/TOF-MS Analyzer. The ^1H and ^{13}C NMR spectra were recorded at 25 °C on a Varian INOVA 400 spectrometer. The spectrometer operated at 399.6 MHz (^1H) and 100.5 MHz (^{13}C). Chemical shifts in the ^1H NMR and ^{13}C NMR spectra were expressed in parts per million (ppm) and referenced to the residual solvent resonances. Products were purified by column chromatography (Fluka, Silica gel, 90 Å, 70–230 mesh) using the mixtures $\text{CH}_2\text{Cl}_2/(\text{CH}_3)_2\text{CO}$ 10:1.

1.2. Preparation of ferrocenoylated adenines

A detailed protocol for the preparation and characterization of ferrocenoylated purines was described earlier.¹ Some of ferrocenoyl-adenines (**1-N7**, **1-N9**, **5-N7**, and **5-N9**) have been reported earlier,¹ and therefore only ^1H and ^{13}C NMR spectra are included herewith. In short, sodium hydride, NaH (1.5 mmol), was added portionwise to purine bases (1 mmol) suspended in 3 mL DMF. After stirring at room temperature for 30 min, FcCOCl (1 mmol) was added dropwise to the clear solution. The mixture was stirred for 15–30 minutes and then was neutralized with 10% aqueous solution of citric acid and extracted with CH_2Cl_2 . The organic layer was washed with water and evaporated under vacuum. A purification by column chromatography (silica gel, dichloromethane/acetone 10:1) afforded *N7*- or *N9*-ferrocenoylated adenines.

*N*7-Ferrocenoyl-adenine (**1-N7**); NMR spectra in agreement with those reported earlier¹:
¹H NMR (400 MHz, DMSO-*d*₆) δ/ppm: 9.00 (s, 1H), 8.37 (s, 1H), 7.07 (s, 2H), 5.11 (pt, 2H), 4.86 (pt, 2H), 4.34 (s, 5H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ/ppm: 171.09, 161.07, 155.04, 152.63, 146.65, 109.55 74.37, 72.65, 71.35, 69.92.

*N*9-Ferrocenoyl-adenine (**1-N9**); NMR spectra in agreement with those reported earlier¹:
¹H NMR (400 MHz, DMSO-*d*₆) δ/ppm: 8.65 (s, 1H), 8.19 (s, 1H), 7.48 (s, 2H), 5.02 (pt, 2H), 4.78 (pt, 2H), 4.31 (s, 5H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ/ppm: 169.73, 156.82, 154.07, 149.61, 139.88, 119.25, 74.10, 72.63, 71.04, 69.92.

*N*7-Ferrocenoyl-*N*⁶-methyladenine (**2-N7**): ¹H NMR (400 MHz, CDCl₃, 25 °C) δ/ppm: 8.85 (s, 1H, C8-H), 8.64 (s, 1H, C2-H), 7.37 (s, 1H, NH), 4.97 (pt, 2H, C α -H), 4.77 (pt, 2H, C β -H), 4.33 (s, 5H, Cp-H), 3.23 (s, 3H, CH₃). ¹³C{¹H} NMR (100 MHz, CDCl₃, 25 °C) δ/ppm: 171.38 (FcCO), 160.03 (C2), 155.49 (C6), 152.17 (C4), 144.23 (C8), 110.38 (C5), 73.83 (C β), 72.24 (C α), 71.12 (Cp), 69.96 (Ci), 28.07 (CH₃).

*N*9-Ferrocenoyl-*N*⁶-methyladenine (**2-N9**): ¹H NMR (400 MHz, CDCl₃, 25 °C) δ/ppm: 8.49 (s, 1H, C8-H), 8.37 (s, 1H, C2-H), 6.31 (s, 2H, NH₂), 5.07 (pt, 2H, C α -H), 4.68 (pt, 2H, C β -H), 4.26 (s, 5H, Cp-H), 3.32 (s, 3H, CH₃). ¹³C{¹H} NMR (100 MHz, CDCl₃, 25 °C) δ/ppm: 171.16 (FcCO), 169.26 (C2), 155.68 (C6), 154.46 (C4), 138.75 (C8), 120.10 (C5), 73.59 (C β), 72.50 (C α), 70.85 (Cp), 69.92 (Ci), 27.52 (CH₃).

*N*7-Ferrocenoyl-*N*⁶, *N*⁶-dimethyladenine (**3-N7**): red crystals (44.7 mg, 11% isolated yield (1.0 mmol, 163.2 mg of dimethyladenine), mp > 200 °C. ¹H NMR (400 MHz, DMSO-*d*₆, 25 °C)

δ /ppm: 8.90 (s, 1H, C8-H), 8.85 (s, 1H, C2-H), 5.26 (pt, 2H, C α -H), 5.14 (pt, 2H, C β -H), 4.69 (s, 5H, Cp-H), 3.30 (s, 6H, 2xCH₃). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆, 25 °C) δ /ppm: 171.77 (FcCO), 165.97 (C6), 161.22 (C8), 152.24 (C4), 141.51 (C2), 110.71 (C5), 73.21 (C β), 72.55 (C α), 71.63 (Ci), 71.01 (Cp), 37.45 (CH₃). IR (CH₂Cl₂) ν_{\max} /cm⁻¹: 3100 (w C-H), 3068 and 2943 (w C-H and =C-H, Fc), 1677 (s and w C=O), 1599 (s, stretching of purine ring), 1500-1200 (C=C and C=N stretching). UV-vis (CH₂Cl₂, 25 °C) λ /nm: 465. ϵ /dm³ mol⁻¹ cm⁻¹: 6.1 x 10². HRMS (MALDI-TOF/TOF) m/z : [M + H]⁺ calculated for C₁₈H₁₇FeN₅O, 376.0816; found 376.0873.

*N*9-Ferrocenoyl-*N*⁶,*N*⁶-dimethyladenine (**3-N9**): red crystals (223.4 mg, 55% isolated yield (1.0 mmol, 163.2 mg of dimethyladenine), mp > 200 °C. *N*9 isomer was isolated from the reaction mixture and purified twice by column chromatography with the eluent dichloromethane/acetone 7:1.

¹H NMR (400 MHz, DMSO-*d*₆, 25 °C) δ /ppm: 8.46 (s, 1H, C8-H), 8.25 (s, 1H, C2-H), 4.98 (pt, 2H, C α -H), 4.78 (pt, 2H, C β -H), 4.32 (s, 5H, Cp-H), 3.38 (s, 6H, 2xCH₃). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆, 25 °C) δ /ppm: 169.78 (FcCO), 154.87 (C6), 153.25 (C8), 150.54 (C4), 138.50 (C2), 119.70 (C5), 74.22 (C β), 72.56 (C α), 71.83 (Ci), 71.04 (Cp), 38.55 (CH₃). IR (CH₂Cl₂) ν_{\max} /cm⁻¹: 3103 (w C-H), 3047 and 2931 (w C-H and =C-H, Fc), 1702 (s and w C=O), 1599 (s, stretching of purine ring), 1500-1200 (C=C and C=N stretching). UV-vis (CH₂Cl₂, 25 °C) λ /nm: 464. ϵ /dm³ mol⁻¹ cm⁻¹: 5.9 x 10². HRMS (MALDI-TOF/TOF) m/z : [M + H]⁺ calculated for C₁₈H₁₇FeN₅O, 376.0816; found 376.0874.

*N*7-Ferrocenoyl-*N*⁶-(2-*isopentenyl*)adenine (**4-N7**): red crystals (39.3 mg, 9% isolated yield (1.0 mmol, 203.2 mg of 2-*isopentenyl*adenine), mp > 200 °C. *N*7 and *N*9 isomers **4** were

separated by column chromatography with eluent dichloromethane/acetone 10:0.9 whereby *N7* isomer was additionally purified on a column with the eluent dichloromethane/acetone 7:1.

^1H NMR (400 MHz, CDCl_3 , 25 °C) δ /ppm: 8.80 (s, 1H, C8-H), 8.62 (s, 1H, C2-H), 7.18 (t, broad, 1H, NH), 5.40 (t, $J = 7.0$ Hz, 1H, CH=), 4.98 (pt, 2H, $\text{C}\alpha$ -H), 4.77 (pt, 2H, $\text{C}\beta$ -H), 4.32 (s, 5H, Cp-H), 4.19 (dd, $J = 6.5$ and 5.6 Hz, 2H, CH_2), 1.76 (dd, $J = 1.4$ and 0.9 Hz, 6H, $2\times\text{CH}_3$). $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , 25 °C) δ /ppm: 171.23 (FcCO), 169.51, 160.22 (C4), 155.49 (C2), 151.29 (C6), 144.10 (C8), 136.81 (C=), 120.07 (CH=), 110.30 (C5), 73.85 ($\text{C}\beta$), 72.25 ($\text{C}\alpha$), 71.52 (Ci), 71.12 (Cp), 39.09 (CH_2), 25.69 (CH_3), 18.05 (CH_3). IR (CH_2Cl_2) $\nu_{\text{max}}/\text{cm}^{-1}$: 3380-3204 (w NH), 2976 (w C-H and =C-H, Fc), 1681 (s, C=O), 1614 (s, stretching of purine ring), 1500-1200 (C=C and C=N stretching). UV-vis (CH_2Cl_2 , 25 °C) λ/nm : 471.9; 349.0 ($c = 1 \times 10^{-4}$ M). $\epsilon/\text{dm}^3 \text{mol}^{-1} \text{cm}^{-1}$: 7.8×10^2 ; 7.6×10^2 . HRMS (MALDI-TOF/TOF) m/z : $[\text{M} + \text{H}]^+$ calculated for $\text{C}_{21}\text{H}_{26}\text{FeN}_5\text{O}$, 416.1174; found 416,1171.

N9-Ferrocenoyl- N^6 -(2-*isopentenyl*)adenine (**4-N9**): resinous solid (182.1 mg, 44% isolated yield (1.0 mmol, 203.2 mg 2-*isopentenyl*adenine), mp > 200 °C. When isolated from the reaction mixture, *N9* was additionally purified on a column with the eluent dichloromethane/acetone 10:0.9.

^1H NMR (400 MHz, CDCl_3 , 25 °C) δ /ppm: 8.50 (s, 1H, C2-H), 8.37 (s, 1H, C8-H), 6.00 (s, broad, 1H, NH), 5.40 (t, 1H, CH=), 5.09 (pt, 2H, $\text{C}\alpha$ -H), 4.70 (pt, 2H, $\text{C}\beta$ -H), 4.28 (s, 5H, Cp-H), 4.25 (s, 2H, CH_2), 1.76 (s, broad, 6H, $2\times\text{CH}_3$). $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , 25 °C) δ /ppm: 169.23 (FcCO), 154.91 (C6), 154.47 (C2), 148.61 (C4), 138.73 (C8), 137.01 (C=), 120.04 (C5), 119.98 (CH=), 73.51 ($\text{C}\beta$), 72.53 ($\text{C}\alpha$), 70.84 (Ci), 69.97 (Cp), 38.81 (CH_2), 25.69 (CH_3), 18.02 (CH_3). IR (CH_2Cl_2) $\nu_{\text{max}}/\text{cm}^{-1}$: 3424 (w, NH), 3126 and 3048 (w C-H), 2975, 2926 (w C-H and =C-H, Fc), 1700 (s, C=O), 1615 (s, stretching of purine ring), 1500-1200 (C=C and C=N stretching). UV-vis

(CH₂Cl₂, 25 °C) λ /nm: 472; 347. ϵ /dm³ mol⁻¹ cm⁻¹: 8.5×10^2 ; 1.8×10^3 . HRMS (MALDI-TOF/TOF) m/z : [M + H]⁺ calculated for C₂₁H₂₆FeN₅O, 416.1174; found 416.1171.

*N*7-Ferrocenoyl-N⁶-benzylaminopurine (**5-N7**); NMR spectra in agreement with those reported earlier¹: ¹H NMR (400 MHz, CDCl₃) δ /ppm: 8.82 (s, 1H), 8.64 (s, 1H), 7.68 (s, 1H), 7.42 (d, 2H), 7.35 (t, 2H), 7.28 (dd, 1H), 4.96 (pt, 2H), 4.86 (d, 2H), 4.75 (pt, 2H), 4.27 (s, 5H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ /ppm: 171.28, 160.45, 155.50, 151.35, 144.34, 138.39, 128.71, 127.62, 127.41, 110.25, 73.84, 72.24, 71.42, 71.09, 44.94.

*N*9-Ferrocenoyl-N⁶-benzylaminopurine (**5-N9**); NMR spectra in agreement with those reported earlier¹: ¹H NMR (400 MHz, CDCl₃) δ /ppm: 8.52 (s, 1H), 8.31 (s, 1H), 7.41 (t, 2H), 7.35 (t, 2H), 7.28 (dd, C4'-H), 6.45 (s, 1H), 5.08 (pt, 2H), 4.89 (s, 2H), 4.71 (pt, 2H), 4.28 (s, 5H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ /ppm: 169.19, 154.98, 154.47, 148.87, 138.97, 138.25, 128.72, 127.76, 127.59, 120.09, 73.54, 72.51, 71.46, 70.85, 44.63.

*N*7-Ferrocenoyl-6-di-*tert*-butoxycarbonyladenine (**6-N7**) was not isolated from the reaction mixture. ¹H NMR (400 MHz, DMF, 25 °C) δ /ppm: 9.57 (s, 1H, C2-H), 9.45 (s, 1H, C8-H), 5.27 (pt, 2H, C α -H), 5.15 (pt, 2H, C β -H), 4.75 (s, Cp-H), 1.60 (s, 20H).

*N*9-Ferrocenoyl-N⁶-di-*tert*-butoxycarbonyladenine (**6-N9**): red crystals (358.7 mg, 80% isolated yield (1.0 mmol, 336.4 mg of N⁶-di-*tert*-butoxycarbonyladenine), mp > 200 °C. N9 isomer was isolated from the reaction mixture and purified by column chromatography with the eluent dichloromethane/ethyl acetate 10:0.5 – 10:1.

^1H NMR (400 MHz, CDCl_3 , 25 °C) δ /ppm: 8.98 (s, 1H, C2-H), 8.70 (s, 1H, C8-H), 5.08 (pt, 2H, C α -H), 4.75 (pt, 2H, C β -H), 4.31 (s, Cp-H), 1.49 (s, 20H). $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, CDCl_3 , 25 °C) δ /ppm: 168.59 (FcCO), 153.40 (C2), 152.51 (C4), 151.19 (C6), 150.42 (C=O, Boc), 143.35(C8), 129.34 (C5), 84.17 (Cq, Boc), 73.88 (C β), 72.46 (C α), 70.97 (Cp), 70.93 (Ci), 27.82 (CH₃). IR (CH_2Cl_2) $\nu_{\text{max}}/\text{cm}^{-1}$: 2982 and 2931 (w C–H and =C–H, Fc), 1786 and 1757 (s and w C=O, Boc), 1717 (s and w C=O), 1597 (w stretching of purine ring), 1500-1200 (C=C and C=N stretching). UV–vis (CH_2Cl_2 , 25 °C) λ/nm : 476; 351. $\epsilon/\text{dm}^3 \text{mol}^{-1} \text{cm}^{-1}$: 5.9×10^2 ; 1.1×10^3 . HRMS (MALDI-TOF/TOF) m/z : $[\text{M}]^+$ calculated for $\text{C}_{26}\text{H}_{29}\text{FeN}_5\text{O}_5$, 547.1518; found 547.1500.

*N*7-Ferrocenoyl-purine (**purine**): red crystals (139,4 mg, 42% isolated yield (1.0 mmol, 120.2 mg of purine), mp > 200 °C. *N*7 isomer was isolated from the reaction mixture and purified by column chromatography with the eluent dichloromethane/acetone 7:1.

^1H NMR (400 MHz, $\text{DMSO}-d_6$, 25 °C) δ /ppm: 9.45 (s, 1H, C2-H), 9.43 (s, 1H, C6-H), 9.13 (s, 1H, C8-H), 5.15 (pt, 2H, C α -H), 4.83 (pt, 2H, C β -H), 4.37 (t, 5H, Cp-H). $^{13}\text{C}\{^1\text{H}\}$ NMR (100 MHz, $\text{DMSO}-d_6$, 25 °C) δ /ppm: 169.24 (FcCO), 160.93 (C4), 154.70 (C8), 148.78 (C6), 144.76 (C2), 124.45 (C5), 73.89 (C β), 72.11 (C α), 71.09 (Cp), 69,97 (Ci). IR (CH_2Cl_2) $\nu_{\text{max}}/\text{cm}^{-1}$: 3129 and 3054 (s, C–H and =C–H, Fc), 1699 (s, C=O), 1587 (s, –C=N, stretching of purine ring), 1500-1200 (C=C and C=N stretching). UV–vis (CH_2Cl_2 , 25 °C) λ/nm : 472.4; 354.4. $\epsilon/\text{dm}^3 \text{mol}^{-1} \text{cm}^{-1}$: 4.5×10^2 ; 8.3×10^2 . HRMS (MALDI-TOF/TOF) m/z : $[\text{M}]^+$ calculated for $\text{C}_{16}\text{H}_{12}\text{FeN}_4\text{OH}$, 332.0360; found 332.0369.

1.3. In situ ^1H NMR monitoring of the reaction between adenine anions and FcCOCl

The $N9/N7$ product ratio was determined quantitatively by integration of the respective signals in ^1H NMR spectra of the reaction mixture (representative spectra in Figure S1 and S2). ^1H NMR spectra were recorded over 30 minutes. All spectra were recorded in N,N -dimethylformamide- d_7 ($\text{DMF-}d_7$, > 99.5 atom % D, Sigma-Aldrich, Saint Louis, USA) or in DMF. In the latter case, the $\text{DMSO-}d_6$ was used as a deuterated solvent in capillary, and the DMF signal was suppressed, where available, by the presaturation options (PRESAT or WET), as included in the OpenVnmrJ, Version 2.1.A (University of Oregon).

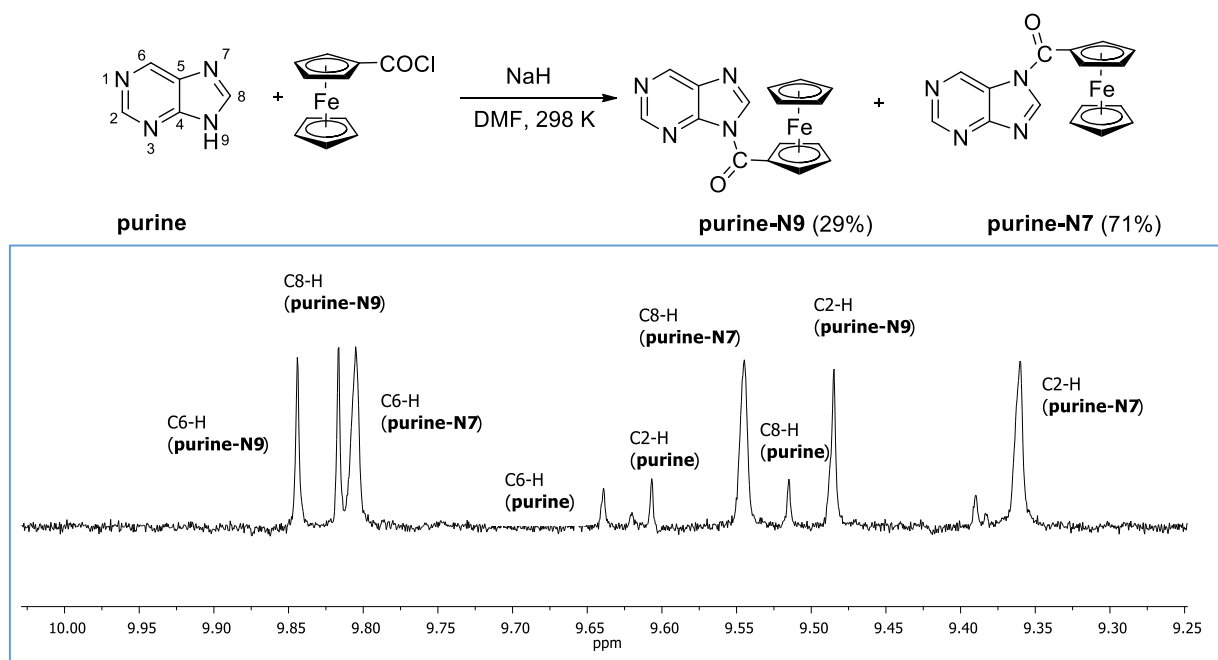


Figure S1: ^1H NMR spectrum (downfield region) of the reaction mixture (purine anion and FcCOCl) in DMF, taken after 20 minutes. The DMF signal was suppressed by the presaturation option (PRESAT), and $\text{DMSO-}d_6$ was used as a deuterated solvent in capillary.

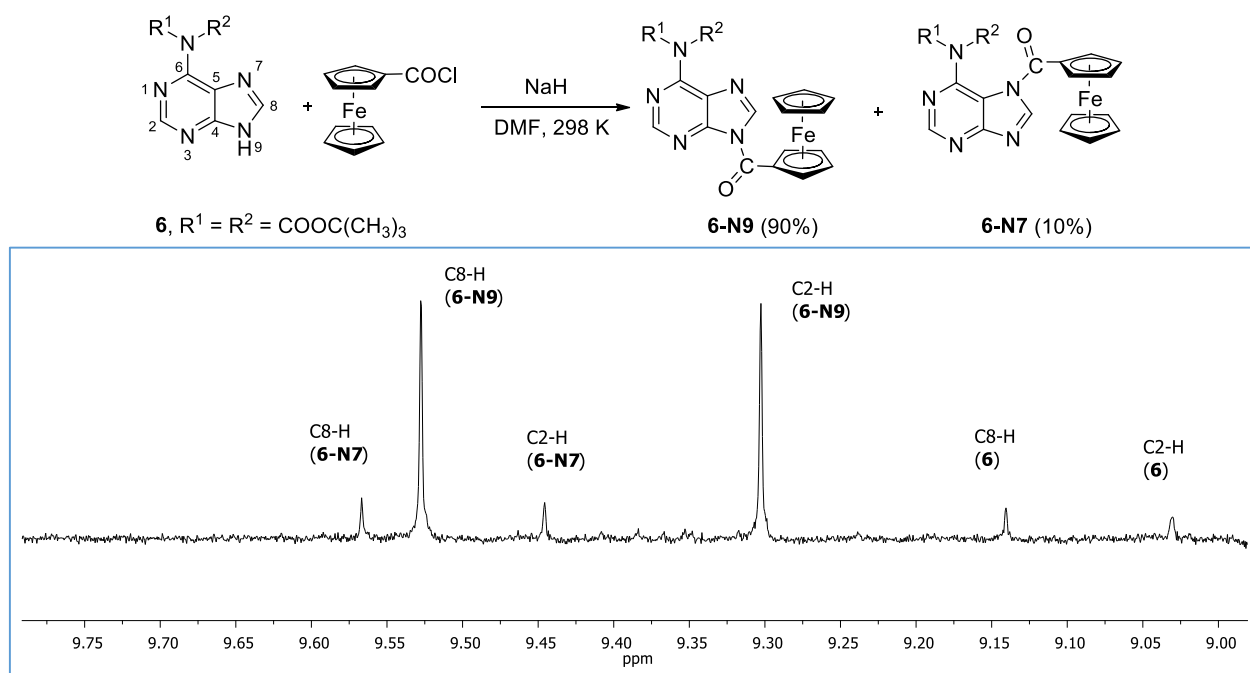


Figure S2: ^1H NMR spectrum (downfield region) of the reaction mixture (adenine anion **6** and FcCOCl) in DMF, taken after 30 minutes. The DMF signal was suppressed by the presaturation option (PRESAT), and DMSO- d_6 was used as a deuterated solvent in capillary.

1.4. Selected ^1H NMR spectra of the reaction mixture (regiospecific mode)

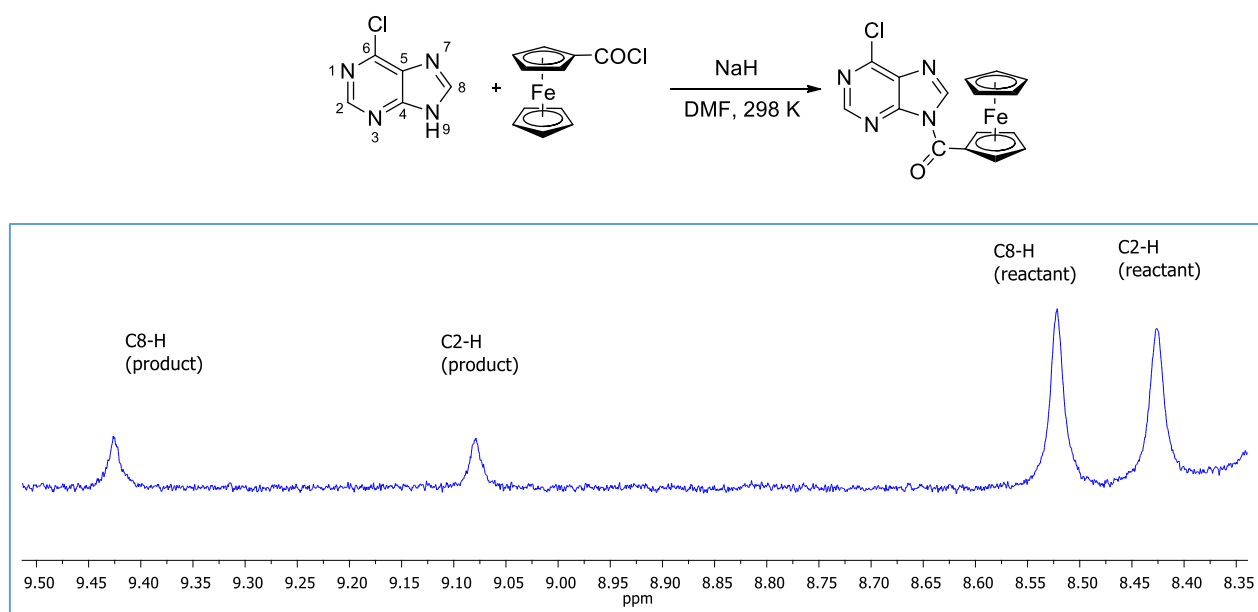


Figure S3: ^1H NMR spectrum (downfield region) of the reaction mixture (6-chloroadenine anion and FcCOCl) in DMF, taken after 10 minutes. The DMF signal was suppressed by the presaturation option (PRESAT), and $\text{DMSO-}d_6$ was used as a deuterated solvent in capillary.

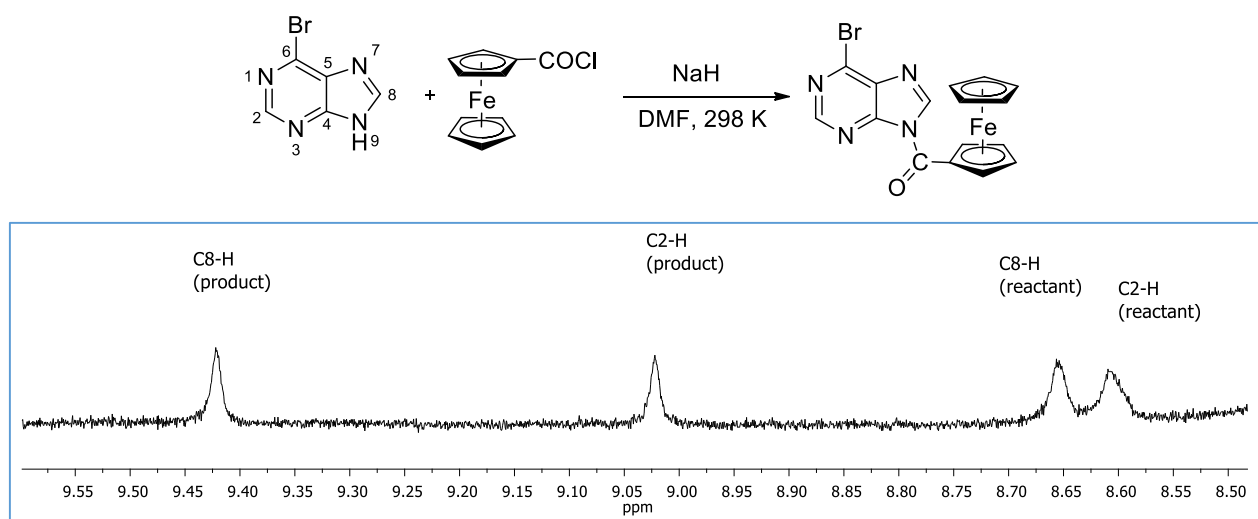


Figure S4: ^1H NMR spectrum (downfield region) of the reaction mixture (6-bromoadenine anion and FcCOCl) in DMF, taken after 10 minutes. The DMF signal was suppressed by the presaturation option (PRESAT), and $\text{DMSO-}d_6$ was used as a deuterated solvent in capillary.

2. Computational part

2.1. Computational techniques and methods

Geometries were fully optimized on the B3LYP level of theory, as implemented in Gaussian 16 software package.² Basis set for optimization was Pople's 6-31+G(d) on non-metal atom centers, while Stuttgart-Dresden (SDD) basis set with effective core potential (ECP) was used for Fe.³ This DFT level has proven successful for organometallic compounds and are well suited for the description of their structures and energies.⁴

Gibbs energies of solvation were determined using the SMD model⁵ at the B3LYP/6-31+G(d)//B3LYP/6-31+G(d) level. The solvent relative permittivity of $\epsilon = 37.22$ (*N,N*-dimethylformamide, DMF) was used.

IRC (intrinsic reaction coordinate) calculations were performed at the corresponding level to identify the minima connected through the transition state. The initial geometries used were that of the corresponding transition structures, and the paths were followed in both directions from that point. This method verified that a given transition state connected the presumed energy minimum structures.

Nucleophilicity (Fukui index f^-) of nitrogen atom A in purine anion (of N electrons) was calculated according to the equation 1.

$$f_A^- = P_A(N) - P_A(N-1) \quad (1)$$

where P stands for the population (or charge) of nitrogen atom A in respective purine anion. Natural population analysis and atomic charge assignments (NPA and CHelpG schemes) were calculated at the (U)B3LYP/6-31+G(d) level using the NBO Version 3.1,⁶ as implemented in Gaussian 16. Conformational search was done using CREST^{7,8}, a part of xtb⁹, using GFN2-xTB method¹⁰.

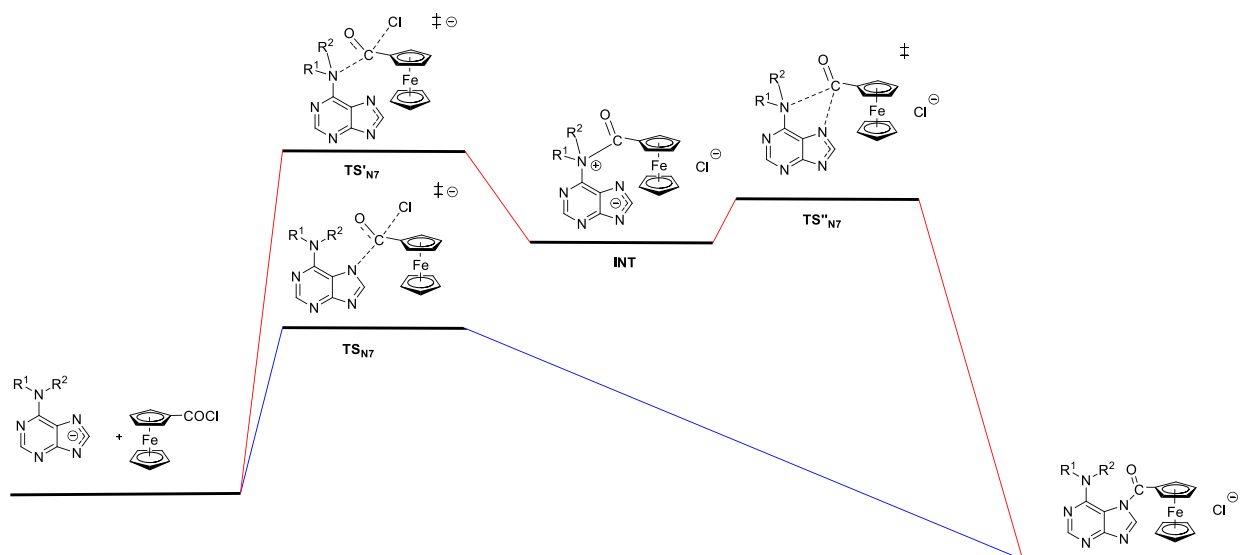
2.2. Calculated energies and barriers

Table S1: B3LYP/6-31+G(d)/SDD calculated Gibbs free energies for reactants and corresponding transition state structures for reaction between purine anions and FcCOCl (*N*7-ferrocenylation). Solvation energies calculated at the SMD-B3LYP/6-31+G(d)/SDD level in DMF ($\epsilon = 37.22$) as a model solvent.

Reactant			Transition state structure			ΔG^\ddagger (kJ/mol) ^b
	G_{298} (Hartree)	ΔG_{solv} (kJ/mol)		G_{298} (Hartree)	ΔG_{solv} (kJ/mol)	
purine^a	-411.374366	-289.9	purine-TS_{N7}	-1495.168160	-248.8	106.4
1 R ¹ = R ² = H	-466.726598	-265.3	1-TS_{N7}	-1550.522334	-261.5	111.0
2 R ¹ = Me, R ² = H	-506.008991	-260.2	2-TS_{N7}	-1589.802595	-263.1	110.0
3 R ¹ = R ² = Me	-545.287581	-252.7	3-TS_{N7}	-1629.067086	-258.2	144.1
4 R ¹ = CH ₂ CHC(CH ₃) ₂ , R ² = H	-661.662098	-264.4	4-TS_{N7}	-1745.752028	-269.5	117.1
5 R ¹ = CH ₂ Ph, R ² = H	-736.998429	-265.7	5-TS_{N7}	-1820.786674	-271.2	121.2
6 R ¹ = R ² = COOC(CH ₃) ₃	-1158.168445	-249.7	6-TS_{N7}	-2241.940467	-257.8	132.3
FcCOCl	-1083.822441	-46.9	-	-	-	-

^aThe anion derived from the parent 9*H*-purine structure. ^b Gibbs free energy of activation calculated as $[(G_{\text{TS}} + \Delta G_{\text{solv}}) - ((G_{\text{reactant}} + \Delta G_{\text{solv}}) + (G_{\text{FcCOCl}} + \Delta G_{\text{solv}}))] - \Delta \Delta G_{\text{solv}}$, where the latter term (≈ 10 kJ/mol) is a correction term to evaluate the effect of the loss of translation degrees of freedom in solution on the Gibbs activation energy in bimolecular reaction, as reported by Ardura et al.¹¹

2.3. S_N2-type and alternative reaction mechanisms for N7-ferrocenylation of adenines



Scheme S1: One step S_N2-type N7-ferrocenylation of adenines via TS_{N7} (blue line) and two step reaction which includes quaternary intermediate INT and carbonyl group shift via TS''_{N7} (red line). The latter alternative mechanism may be applicable for **3** (R¹ = R² = Me); see the main text.

2.4. Calculation of Sterimol and % V_{Bur} parameters

Boltzmann-weighted subparameters B₁, B₅, and L₀ are obtained as follows: first, conformational ensemble is generated, geometry optimization and energy calculation for each conformer is then performed using either semiempirical calculations and/or DFT. Sterimol parameters are computed for each conformer, while the Boltzmann-weighted parameters are obtained by summing multiplication products of conformers Sterimol parameter with weight of each conformer, which is calculated from corresponding energies.

Sterimol was used to describe substituents starting with amino moieties¹², in two different procedures. First approach was to compute Sterimol parameter for the entire substituent, while the other approach was to divide amino substituent to two sub-units, and compute Sterimol parameter for each sub-unit. Better results were obtained when the latter approach was used.

Usage of Sterimol program (wSterimol) is straightforward, but some limitations exist. Namely, only Gaussian input (.com and .gjf) and output (.out and .log) files are expected file formats. Thus, a variant, SteriXYZ was written and implemented as a web-page eliminating installation with more commonly used .xyz format as input¹³. Additional feature involves automatic reading and solving of multiple conformers/optimization steps/molecular dynamic snapshots contained in the same .xyz file. This feature was implemented for automatic processing of CREST files, when conformational sampling was performed by the xtb program package using GFN2-xTB method. Useful visualization of calculated parameters are also present¹⁴. Example of graphical representation of Sterimol parameters can be seen in graphical abstract of this paper.

Additionally, buried volume (% V_{Bur}) is calculated using similar approach as SambVca 2.1.^{15,16} For this, our own implementation was used.^{13,14} For calculation of buried volume only

definition of reaction centre is needed, as compared to SambVca. Ignoring reaction center radius, target radius of 3.5 ångströms, mesh spacing value of 0.1 ångströms, and Bondi atomic radii scaled by 1.17 are used for the calculation. Hydrogen atoms are included in calculation. Boltzmann averaging was performed at 298.15 K on conformers obtained from CREST at GFN2-xTB level of theory.

Table S2: Calculated Sterimol parameters for Boltzmann-averaged purine anion conformers at CREST GFN2-xTB level of theory.

Structure	C6-R			R = R ¹ -N-R ²			R = R ¹ -N-R ²		
	L ₀	B ₁	B ₅	L ₀	B ₁	B ₅	L ₀	B ₁	B ₅
purine	2.173	1.090	1.090	-	-	-	-	-	-
1	2.921	1.550	1.973	2.095	1.090	1.959	2.095	1.090	1.959
2	4.080	1.550	3.139	3.138	1.700	2.125	2.095	1.090	1.959
3	4.072	1.550	3.164	3.138	1.700	2.125	3.138	1.700	2.125
4	5.709	1.550	5.885	5.047	1.700	5.429	2.095	1.090	1.959
5	5.953	1.550	6.627	5.069	1.700	6.102	2.095	1.090	1.959
6	7.028	1.550	6.467	6.148	1.700	4.764	6.148	1.700	4.764
FcCO-N^a				6.045	1.810	6.775			

^a N7-ferrocenoyl purine

Table S3: Calculated % V_{Bur} (using parameters from SambVca 2.1) and minimal distance of reaction center to substituent (C6-R). Boltzmann-averaged purine anion conformers at CREST GFN2-xTB level of theory.

Structure	% V_{Bur}		Distance	
	N7	N9	N7	N9
Purine	43.27	42.38	2.926	4.399
1	45.96	42.36	2.769	4.628
2	48.68	42.37	2.433	4.501
3	51.03	42.39	2.112	4.300
4	51.67	42.49	2.395	4.396
5	54.21	42.51	2.315	4.429
6	60.62	42.94	2.608	4.291
FcCO-N^a	72.71			

^aN7-ferrocenoyl purine

2.5. Optimized coordinates of reactants and transition states

FeCOCl

Sum of electronic and thermal Free Energies= -1083.822441

23

Fe 0.861391 0.152325 -0.040055
C -1.066465 0.858208 0.130542
C -0.198534 1.575249 1.032362
C 0.790696 2.226149 0.250352
C 0.554215 1.915546 -1.126980
C -0.582908 1.069164 -1.209843
C 0.890546 -1.881370 0.399112
C 1.376678 -1.658027 -0.925026
C 2.522729 -0.809026 -0.834651
C 2.741914 -0.507642 0.544159
C 1.731844 -1.170587 1.307939
H -0.299015 1.594356 2.108854
H 1.605031 2.828220 0.632870
H 1.158285 2.244009 -1.963077
H -1.013249 0.650130 -2.108077
H 0.012198 -2.456581 0.664029
H 0.935984 -2.041498 -1.836248
H 3.104711 -0.434491 -1.667233
H 3.519047 0.134763 0.938554
H 1.607730 -1.120759 2.382087
O -2.555534 -0.107419 1.720109
C -2.179148 0.034015 0.592154
Cl -3.063060 -0.847607 -0.754818

purine

Sum of electronic and thermal Free Energies= -411.374366

12

C 0.526812 0.577493 -0.000000
C 0.000000 -0.763444 -0.000000
C -2.093021 0.068372 0.000000
C -0.394849 1.618716 0.000000
H -3.170212 -0.103363 0.000000
H -0.088931 2.666508 0.000000
N -1.718191 1.367556 0.000000

N -1.325382 -1.023439 0.000000
N 1.900946 0.518536 -0.000000
N 1.036713 -1.645530 -0.000000
C 2.108029 -0.812335 -0.000000
H 3.118711 -1.215827 -0.000000

1
Sum of electronic and thermal Free Energies= -466.726598

14

C -0.000000 0.554692 0.000000
C -1.035396 -0.420878 0.000000
C 0.503937 -2.058201 -0.000000
C 1.323321 0.093616 -0.000000
H 0.755224 -3.119946 -0.000000
N 1.576046 -1.227031 -0.000000
N -0.789967 -1.760926 -0.000000
N -0.547315 1.816818 0.000000
N -2.238537 0.221227 0.000000
C -1.864506 1.525107 0.000000
H -2.610741 2.316754 0.000000
N 2.391179 0.962113 -0.000000
H 2.221521 1.955468 -0.000000
H 3.330011 0.596296 -0.000000

2
Sum of electronic and thermal Free Energies= -506.008991

17

N 2.643845 -0.090790 0.076950
C 2.304349 -1.403031 0.042214
H 3.069034 -2.176077 0.075241
N 0.996132 -1.727125 -0.033041
C 0.417122 -0.480439 -0.049874
C -0.918510 -0.055556 -0.102942
N -1.951083 -0.979562 -0.213170
N -1.206022 1.256054 -0.072195
C -0.156862 2.115560 -0.010498
H -0.436198 3.170085 0.006905
N 1.143236 1.852179 0.035591

C 1.425740 0.520365 0.019934
C -3.301996 -0.634248 0.185944
H -3.626788 0.255634 -0.359426
H -3.968655 -1.467678 -0.065331
H -3.397544 -0.414827 1.264149
H -1.641655 -1.918343 0.010838

3

Sum of electronic and thermal Free Energies= -545.287581

20

N 2.813972 -0.398763 -0.021547
C 2.263284 -1.630845 0.072023
H 2.887974 -2.520888 0.106934
N 0.921272 -1.731013 0.120837
C 0.538537 -0.406455 0.047857
C -0.704978 0.263165 0.070643
N -1.941468 -0.377968 0.209830
N -0.719087 1.609470 0.002194
C 0.462849 2.262263 -0.069058
H 0.379860 3.349440 -0.114982
N 1.690475 1.762967 -0.096753
C 1.714856 0.405766 -0.036177
C -2.096090 -1.738837 -0.282772
H -1.169843 -2.295712 -0.134328
H -2.912124 -2.230474 0.264792
H -2.352170 -1.754824 -1.360644
C -3.144432 0.427616 0.110429
H -3.015229 1.363408 0.654943
H -3.398863 0.680699 -0.936477
H -3.979898 -0.140523 0.540159

4

Sum of electronic and thermal Free Energies= -661.962098

27

C 3.301475 -2.061058 -0.082048
N 4.099913 -0.965867 -0.128711
H 3.726523 -3.061062 -0.133364
C 1.889937 -0.509267 0.057620

C 3.195321 0.050327 -0.040770
N 1.968523 -1.881222 0.028219
C 0.807596 0.377210 0.143465
C 2.313970 2.116135 0.028210
H 2.441945 3.199439 0.016747
N 1.024446 1.702088 0.119941
N 3.423368 1.392155 -0.052352
N -0.491932 -0.105301 0.283502
H -0.554650 -1.096171 0.073485
C -1.635383 0.702236 -0.118106
H -1.751732 0.735514 -1.213724
H -1.411074 1.727809 0.196141
C -2.886833 0.218936 0.563588
H -2.793419 0.162235 1.649629
C -4.058781 -0.147907 0.014990
C -4.360383 -0.154437 -1.466196
H -4.737925 -1.139366 -1.778811
H -5.150043 0.573129 -1.708250
H -3.489471 0.083480 -2.081586
C -5.212932 -0.596533 0.884200
H -4.952061 -0.577160 1.947962
H -6.095390 0.045471 0.738748
H -5.526852 -1.620139 0.629109

5

Sum of electronic and thermal Free Energies= -736.998429

27

C -3.188245 0.005507 0.452884
C -3.528760 -1.910090 -0.375941
C -2.099719 -0.368850 -0.383904
H -4.021337 -2.854905 -0.594408
N -4.107167 -1.001156 0.447913
N -2.328238 -1.614685 -0.917140
N -3.230377 1.189915 1.123865
N -1.064556 1.699129 0.160095
C -2.159632 1.947088 0.923973
C -1.034024 0.533514 -0.505594
H -2.141187 2.906468 1.442577
C 1.299149 0.958422 -1.272644
H 1.676937 1.127767 -2.291858
H 1.065144 1.940101 -0.845135
C 2.397990 0.295767 -0.449408

C 3.734202 0.356659 -0.869606
C 2.105641 -0.366460 0.752808
C 4.758124 -0.218016 -0.109453
H 3.975248 0.856986 -1.806854
C 3.124743 -0.942918 1.514445
H 1.073536 -0.429627 1.085571
C 4.456006 -0.870880 1.088645
H 5.787872 -0.162680 -0.457633
H 2.877265 -1.451701 2.443607
H 5.247435 -1.324483 1.681554
N 0.045572 0.240272 -1.329593
H 0.059600 -0.720718 -1.650630

6

Sum of electronic and thermal Free Energies= -1158.168445

44

C -1.625833 2.785953 -1.788129
N -1.248848 1.507455 -1.576497
H -2.182563 3.056750 -2.682354
C -0.629026 2.990050 0.059013
C -0.593412 1.615996 -0.378333
C 0.060748 0.722251 0.461621
C 0.531285 2.439639 1.904674
H 1.002085 2.734775 2.841971
N -1.298124 3.725594 -0.869798
N -0.058950 3.410283 1.206499
N 0.619077 1.126459 1.606436
N 0.193459 -0.683930 0.143082
C 1.474670 -1.260857 0.065740
O 1.732915 -2.419171 0.320535
C -0.935985 -1.497543 -0.052665
O -0.929031 -2.554391 -0.650719
O 2.359037 -0.327741 -0.336837
O -2.006104 -0.923333 0.524905
C 3.804915 -0.559306 -0.262488
C 4.216963 -0.905547 1.173993
C 4.215918 -1.645002 -1.265350
C 4.376340 0.802288 -0.672639
H 3.821471 -0.157038 1.868943
H 3.845551 -1.889036 1.470193
H 5.311952 -0.906114 1.247255
H 3.869647 -1.381821 -2.271536

H 5.310275 -1.727271 -1.289220
H 3.792902 -2.613830 -0.991766
H 5.472421 0.765162 -0.674700
H 4.032138 1.076359 -1.675657
H 4.049481 1.580998 0.023898
C -3.373483 -1.350884 0.193844
C -3.592626 -1.321951 -1.323864
C -3.641128 -2.732993 0.801909
C -4.224258 -0.276151 0.878742
H -3.226132 -0.374876 -1.734142
H -3.069467 -2.142542 -1.820253
H -4.666505 -1.412506 -1.532751
H -3.452166 -2.718593 1.882018
H -4.691588 -3.008037 0.640595
H -3.002626 -3.491461 0.343278
H -5.289596 -0.495864 0.738416
H -4.011803 -0.243865 1.953204
H -4.006844 0.709202 0.455075

purine-TS-N7

Sum of electronic and thermal Free Energies= -1495.168160

Imaginary frequencies: -106.9388

35

Fe 2.467417 -0.485557 -0.192329
C 0.470073 0.054253 -0.321113
C 0.972321 -0.299765 -1.616464
C 1.418128 -1.651269 -1.561228
C 1.198163 -2.134986 -0.233195
C 0.614682 -1.084155 0.531890
C 4.011329 0.804697 -0.705956
C 3.496457 1.140107 0.583959
C 3.660943 0.003844 1.434387
C 4.277579 -1.035867 0.670123
C 4.493887 -0.540969 -0.653988
H 1.006892 0.365363 -2.468404
H 1.870557 -2.205415 -2.374517
H 1.454992 -3.119844 0.136973
H 0.346182 -1.119241 1.577776
H 3.996897 1.444523 -1.579268
H 3.007465 2.067412 0.853466
H 3.331172 -0.067765 2.462946
H 4.508742 -2.033843 1.021989

H 4.918208 -1.098107 -1.480320
C -0.065281 1.404615 -0.018826
C -2.760831 1.416515 -1.390164
N -2.145082 0.843617 -0.336186
Cl -0.208952 1.686848 1.869541
O 0.067870 2.379243 -0.710453
H -2.248472 2.169630 -1.980677
C -4.254420 0.092437 -0.692632
C -3.092987 -0.028536 0.144938
C -3.160557 -0.924451 1.206883
C -5.301437 -1.459358 0.556219
H -6.187945 -2.062870 0.750546
N -4.019293 1.022714 -1.665137
N -5.372272 -0.630897 -0.487692
N -4.274536 -1.649531 1.412809
H -2.338311 -1.069718 1.905895

1-TS-N7

Sum of electronic and thermal Free Energies= -1550.522334
Imaginary frequencies: -112.1076

37

Fe 2.542614 -0.429126 0.364372
C 0.552249 0.123742 0.191514
C 0.665011 -1.305043 0.238212
C 1.275801 -1.655022 1.474996
C 1.547296 -0.448692 2.193844
C 1.101819 0.649012 1.403636
C 3.685255 -1.122570 -1.225284
C 3.542453 0.297018 -1.302562
C 4.096320 0.862694 -0.112999
C 4.582287 -0.207961 0.701248
C 4.327490 -1.435793 0.014001
H 0.326708 -1.979350 -0.536153
H 1.514857 -2.659378 1.802100
H 2.032772 -0.377414 3.159425
H 1.174287 1.697471 1.654346
H 3.328938 -1.836086 -1.957496
H 3.047183 0.844554 -2.094303
H 4.102608 1.914684 0.142340
H 5.032149 -0.109273 1.681605
H 4.550947 -2.429590 0.382198
C -0.023340 0.841709 -0.969505

C -2.489257 1.817110 0.882245
N -2.044534 0.942261 -0.045126
Cl 0.308257 2.709384 -0.872591
O -0.200569 0.378866 -2.069328
H -1.896789 2.683565 1.156836
C -4.077573 0.417291 0.813709
C -3.064319 0.012544 -0.101346
C -3.302681 -1.144358 -0.861776
C -5.334192 -1.343156 0.218615
H -6.248399 -1.928908 0.318943
N -3.691687 1.572812 1.434209
N -5.236164 -0.266883 0.989549
N -4.453751 -1.818326 -0.691450
N -2.387225 -1.661985 -1.766333
H -2.809081 -2.280273 -2.449298
H -1.714153 -0.999477 -2.142747

2-TS-N7

Sum of electronic and thermal Free Energies= -1589.802596
Imaginary frequencies: -107.1879

40

Fe 2.656949 0.229544 -0.527503
C 0.692120 -0.311184 -0.145153
C 0.733403 0.940920 -0.842997
C 1.375650 0.723768 -2.094188
C 1.738774 -0.657340 -2.172988
C 1.318309 -1.296694 -0.972098
C 3.679113 1.626189 0.619625
C 3.634863 0.369006 1.297260
C 4.277794 -0.602257 0.469579
C 4.720011 0.054452 -0.721510
C 4.349609 1.432420 -0.628973
H 0.332258 1.872509 -0.468767
H 1.576599 1.479428 -2.843520
H 2.266656 -1.131641 -2.991106
H 1.456173 -2.335982 -0.711135
H 3.242442 2.551750 0.972871
H 3.145506 0.172997 2.242760
H 4.371485 -1.658024 0.689945
H 5.217493 -0.415495 -1.561057
H 4.517396 2.188724 -1.385930
C 0.108999 -0.451128 1.208667

C -2.210912 -2.291436 -0.105772
N -1.884081 -1.076458 0.387697
Cl 0.514728 -2.130474 1.981141
O -0.133891 0.454139 1.968923
H -1.557560 -3.144923 0.042906
C -3.861179 -1.115237 -0.721754
C -2.947699 -0.291063 -0.010738
C -3.298725 1.061620 0.175392
C -5.239497 0.631861 -1.003208
H -6.166224 1.051786 -1.396123
N -3.374025 -2.392851 -0.772756
N -5.032073 -0.656861 -1.236994
N -4.461398 1.512137 -0.329754
N -2.481462 1.950702 0.842088
H -1.748690 1.523472 1.399849
C -2.981515 3.236263 1.289957
H -3.335747 3.824900 0.438037
H -3.820636 3.146693 1.999085
H -2.160502 3.772954 1.777893

3-TS-N7

Sum of electronic and thermal Free Energies= -1629.067086

Imaginary frequencies: -200.1902

43

Fe 3.085239 0.003104 -0.276006
C 1.053433 -0.362515 -0.446386
C 1.517802 0.399547 -1.569173
C 2.514429 -0.364911 -2.242898
C 2.674879 -1.598681 -1.537117
C 1.774341 -1.598100 -0.430618
C 3.834261 1.818446 0.395729
C 3.350254 1.046246 1.495818
C 4.086101 -0.177635 1.533132
C 5.025073 -0.164061 0.454400
C 4.869140 1.070831 -0.250231
H 1.172742 1.391329 -1.827957
H 3.074515 -0.052162 -3.115852
H 3.379085 -2.385064 -1.780010
H 1.671498 -2.367320 0.321287
H 3.450203 2.780710 0.081364
H 2.520492 1.306088 2.140366
H 3.919595 -0.991073 2.227633

H 5.710090 -0.962190 0.195307
H 5.416350 1.372353 -1.135173
C 0.013638 0.157641 0.489893
C -1.604105 -2.081534 -0.526155
N -1.603216 -0.731697 -0.382694
Cl 0.021482 -0.948880 2.142676
O -0.222473 1.324049 0.667486
H -0.677799 -2.621387 -0.680604
C -3.651511 -1.678276 -0.192483
C -2.946220 -0.439433 -0.168994
C -3.725011 0.733649 -0.029252
C -5.573478 -0.636227 0.274014
H -6.637737 -0.672693 0.508478
N -2.778906 -2.705690 -0.443744
N -4.980364 -1.797565 0.031408
N -5.040628 0.600763 0.234677
N -3.227784 2.028204 -0.141727
C -4.125640 3.129040 0.157124
H -4.686583 2.919642 1.069162
H -4.855637 3.315141 -0.651460
H -3.521065 4.033655 0.298950
C -2.287092 2.318777 -1.214243
H -1.625181 3.139559 -0.917195
H -1.673324 1.445839 -1.426580
H -2.824334 2.610487 -2.137725

4-TS-N7

Sum of electronic and thermal Free Energies= -1745.752028

Imaginary frequencies: -111.7388

50

Fe 2.836459 0.151354 -0.557403
C 1.035433 -0.772383 -0.108353
C 0.806551 0.401574 -0.900362
C 1.484908 0.232568 -2.140261
C 2.139237 -1.038883 -2.118077
C 1.864325 -1.658889 -0.866068
C 3.522626 1.814857 0.478672
C 3.761128 0.628231 1.238065
C 4.603644 -0.229077 0.465569
C 4.886241 0.427294 -0.773317
C 4.217530 1.691163 -0.765410
H 0.221534 1.256702 -0.590559

H 1.520799 0.953821 -2.947359
H 2.759615 -1.449908 -2.904884
H 2.224006 -2.620134 -0.528668
H 2.890464 2.641435 0.777361
H 3.328965 0.392027 2.202030
H 4.931345 -1.219946 0.753289
H 5.474993 0.024172 -1.588192
H 4.210677 2.412651 -1.573197
C 0.489605 -0.928984 1.259867
C -1.232388 -3.359381 0.065539
N -1.271274 -2.096283 0.545759
Cl 1.293643 -2.374738 2.179246
O 0.032774 -0.038110 1.935606
H -0.388774 -4.007839 0.278070
C -3.082805 -2.663169 -0.695091
C -2.471762 -1.622573 0.053101
C -3.177789 -0.408433 0.181602
C -4.831656 -1.333805 -1.143967
H -5.794369 -1.167574 -1.628882
N -2.277455 -3.769574 -0.673620
N -4.287117 -2.530845 -1.311905
N -4.362017 -0.273279 -0.445501
N -2.708142 0.632214 0.946410
C -3.341058 1.939835 1.000604
H -4.417843 1.782677 0.873722
H -3.179458 2.354967 2.000996
C -2.846560 2.875882 -0.079244
H -2.979566 2.477611 -1.086011
C -2.273566 4.085565 0.052832
C -1.836159 4.873084 -1.163245
C -2.008265 4.782703 1.368100
H -2.047629 4.334229 -2.093134
H -2.345553 5.847881 -1.211665
H -0.756799 5.087200 -1.130823
H -0.944894 5.050915 1.455480
H -2.573598 5.724926 1.432481
H -2.270782 4.176754 2.238796
H -1.778470 0.528959 1.340264

5-TS-N7

Sum of electronic and thermal Free Energies= -1820.786674
Imaginary frequencies: -115.6184

Fe 3.493144 0.604516 -0.472537
C 1.777713 -0.433141 0.053031
C 1.688098 0.015903 -1.306120
C 2.764149 -0.567279 -2.032838
C 3.524826 -1.372263 -1.128022
C 2.918626 -1.290087 0.157958
C 3.582614 2.678449 -0.529830
C 3.666370 2.218125 0.820193
C 4.815929 1.376662 0.929675
C 5.443734 1.315684 -0.353869
C 4.680894 2.120358 -1.256693
H 0.928328 0.682469 -1.690014
H 2.982754 -0.406854 -3.081346
H 4.422950 -1.926986 -1.369952
H 3.259373 -1.769504 1.064224
H 2.799841 3.305397 -0.937702
H 2.950313 2.419539 1.606615
H 5.127185 0.844650 1.819609
H 6.322682 0.735254 -0.606307
H 4.881051 2.256396 -2.312426
C 0.821612 -0.001883 1.099832
C -0.121793 -3.057551 1.281639
N -0.437956 -1.810697 0.867478
Cl 1.438803 -0.439809 2.837863
O 0.062073 0.932956 1.011199
H 0.687444 -3.219444 1.985830
C -1.735986 -3.435044 -0.036730
C -1.488264 -2.035941 -0.001159
C -2.331480 -1.205117 -0.766142
C -3.424755 -3.115677 -1.476246
H -4.224308 -3.514321 -2.101355
N -0.853785 -4.072742 0.791419
N -2.717918 -4.000223 -0.787017
N -3.301915 -1.767962 -1.508360
N -2.187375 0.166256 -0.785196
C -3.113760 1.055797 -1.448359
H -3.628741 0.461098 -2.210758
H -2.549854 1.842767 -1.969895
H -1.532329 0.561493 -0.116611
C -4.151721 1.718491 -0.547837
C -4.649604 2.989421 -0.869811
H -4.259192 3.509350 -1.744011
C -5.631340 3.600706 -0.083853
H -5.999306 4.589988 -0.349104
C -6.128889 2.945825 1.046965

H -6.888040 3.419190 1.665901
C -5.634020 1.680350 1.380166
H -6.009645 1.163467 2.260533
C -4.654069 1.072616 0.590414
H -4.269156 0.092895 0.857032

6-TS-N7

Sum of electronic and thermal Free Energies= -2241.940467
Imaginary frequencies: -162.5017

67

Fe -3.815608 0.130824 -0.065703
C -1.925805 -0.143694 0.727283
C -2.024558 1.152581 0.121792
C -3.057930 1.871200 0.788835
C -3.602690 1.022806 1.803772
C -2.905196 -0.219720 1.767273
C -4.204553 -0.099630 -2.092216
C -4.110760 -1.376695 -1.459053
C -5.115281 -1.442596 -0.445397
C -5.831314 -0.204448 -0.450152
C -5.267533 0.626520 -1.468530
H -1.417050 1.494517 -0.704212
H -3.394197 2.872247 0.547622
H -4.424451 1.267852 2.465427
H -3.095462 -1.086291 2.383753
H -3.552502 0.267151 -2.874770
H -3.364973 -2.134308 -1.662085
H -5.269816 -2.269086 0.236442
H -6.636529 0.068994 0.220834
H -5.570545 1.639551 -1.703521
C -0.948544 -1.158137 0.255102
C 0.352886 -1.058649 3.014629
N 0.504548 -0.788842 1.698217
Cl -1.460894 -2.944313 0.967220
O -0.407110 -1.169371 -0.813514
H -0.531769 -1.572660 3.370956
C 2.233608 -0.100793 2.989990
C 1.727564 -0.159202 1.639959
C 2.552792 0.385311 0.657574
C 4.088991 0.941250 2.276588
H 5.052465 1.400723 2.492577
N 1.336906 -0.679754 3.840329

N 3.414350 0.452450 3.315530
N 3.728431 0.932406 0.979980
N 2.232849 0.420400 -0.756108
C 1.988485 1.670751 -1.350054
O 2.152059 1.952991 -2.517356
C 2.438751 -0.722781 -1.560492
O 2.237312 -0.781228 -2.752490
O 1.523882 2.515089 -0.399539
O 2.898399 -1.718049 -0.780278
C 1.451758 3.961731 -0.633754
C 2.831891 4.506174 -1.025088
C 0.384146 4.279815 -1.689052
C 1.031608 4.496483 0.739923
H 3.584102 4.186412 -0.295738
H 3.133373 4.157378 -2.015335
H 2.800662 5.603050 -1.033544
H -0.583262 3.858827 -1.391987
H 0.268651 5.367912 -1.775829
H 0.660286 3.873422 -2.664129
H 0.920472 5.586831 0.700495
H 0.077216 4.056933 1.048573
H 1.782237 4.245908 1.496619
C 3.115000 -3.073111 -1.310461
C 1.815253 -3.645585 -1.887472
C 4.251525 -3.047103 -2.340749
C 3.534807 -3.848607 -0.057032
H 1.003778 -3.565798 -1.158231
H 1.515053 -3.118585 -2.795219
H 1.969299 -4.705983 -2.126714
H 5.151305 -2.596157 -1.905581
H 4.494243 -4.074094 -2.642431
H 3.966034 -2.478385 -3.228596
H 3.748430 -4.892121 -0.317926
H 4.433392 -3.407451 0.388355
H 2.735218 -3.830203 0.690223

3. References

1. Toma, M.; Božičević, L.; Lapić, J.; Djaković, S.; Šakić, D.; Tandarić, T.; Vianello, R.; Vrčec, V. *J. Org. Chem.* **2019**, *84*, 12471–80.
2. Frisch, M.; Trucks, G.; Schlegel, H.; Scuseria, G.; Robb, M.; Cheeseman, J.; Scalmani, G.; Barone, V.; Mennucci, B. *Gaussian 16. Rev. B.01*, Wallingford, CT, 2016.
3. Dolg, M.; Wedig, U.; Stoll, H.; Preuss, H. *J. Chem. Phys.* **1987**, *86*, 866–872.
4. Lapić, J.; Havaić, V.; Šakić, D.; Sanković, K.; Djaković, S.; Vrčec, V. *Eur. J. Org. Chem.* **2015**, *24*, 5424–5431.
5. Marenich, A. V.; Cramer, C. J.; Truhlar, D. G. *J. Phys. Chem. B* **2009**, *113*, 6378–96.
6. Reed, A. E.; Curtiss, L. A.; Weinhold, F. *Chem. Rev.* **1988**, *88*, 899–926.
7. Pracht, P.; Bohle, F.; Grimme, S. *Phys. Chem. Chem. Phys.* **2020**, *22*, 7169–7192.
8. Grimme, S. *J. Chem. Theory Comput.* **2019**, *15*, 2847–2862.
9. Bannwarth, C.; Caldeweyher, E.; Ehlert, S.; Hansen, A.; Pracht, P.; Seibert, J.; Spicher, S.; Grimme, S. *WIREs Comput. Mol. Sci.* **2020**, *11*, e01493.
10. Bannwarth, C.; Ehlert, S.; Grimme, S. *J. Chem. Theory Comput.* **2019**, *15*, 1652–167.
11. Ardura, D.; López, R.; Sordo, T. L. *J. Phys. Chem. B* **2005**, *109*, 23618–23623.
12. Brethome, A. V.; Fletcher, S. P.; Paton, R. S. *ACS Catal.* **2019**, *9*, 2313–2323.
13. Github: <https://github.com/DSakicLab/SteriXYZ>
14. Web-page: <http://sw.pharma.hr/SteriXYZ/>
15. Falivene, L.; Cao, Z.; Petta, A.; Serra, L.; Poater, A.; Oliva, R.; Scarano, V.; Cavallo, L. *Nat. Chem.* **2019**, *11*, 872–879.
16. SambVca 2.1 A web application to characterize catalytic pockets. Available at <https://www.molnac.unisa.it/OMtools/sambvca2.1/index.html> (Accessed on August 15, 2022)