

Supporting Information

for

Additive-controlled chemoselective inter-/intramolecular hydroamination via electrochemical PCET process

Kazuhiro Okamoto, Naoki Shida and Mahito Atobe

Beilstein J. Org. Chem. 2024, 20, 264–271. doi:10.3762/bjoc.20.27

Detailed experimental procedures, CV simulation, copies of NMR spectra

License and Terms: This is a supporting information file under the terms of the Creative Commons Attribution License (https://creativecommons.org/ licenses/by/4.0). Please note that the reuse, redistribution and reproduction in particular requires that the author(s) and source are credited and that individual graphics may be subject to special legal provisions.

Table of contents

Ι	Experimental section	S2
II	Simulation for cyclic voltammogram (Figures S1 and S2)	S 8
III	¹ H and ¹³ C NMR spectra of compounds	S10

I. Experimental section

General information

All reactions were performed under an argon atmosphere, unless otherwise noted. ¹H and ¹³C NMR spectra were recorded in CDCl₃ using a Bruker DRX500 spectrometer (¹H 500 MHz, ¹³C 126 MHz). Tetramethylsilane (¹H, 0.00 ppm) and CHCl₃ (¹H, 7.26 ppm, ¹³C, 77.16 ppm) were used as internal standards. Mass spectra were recorded on a JEOL JMS-T100GCV mass spectrometer. Cyclic voltammetry (CV) was performed using a Bio-Logic VSP-3e instrument. The oxidation potential was measured using glassy carbon as the anode (φ 3 mm), platinum as the cathode (φ 3 mm), and Ag/AgCl as the reference electrode. Merck pre-coated silica gel F₂₅₄ plates (thickness: 0.25 mm) were used for thin-layer chromatography (TLC). All materials were obtained from TCI Fine Chemicals, Wako Pure Chemical Industries, Kanto Chemical, and Sigma-Aldrich and were used without purification. Silica gel column chromatography was performed using Kanto Chemical Silica Gel 60N (spherical, neutral, 63-210 µm).

Abbreviations

BOM, benzyloxymethyl; DBU, 1,8-diazabicyclo[5.4.0]undec-7-ene; DMAP, 4-dimethylaminopyridine; DMF, *N,N*-dimethylformamide; HFIP, 1,1,1,3,3,3-hexafluoro-2-propanol; MVK, methyl vinyl ketone; THF, tetrahydrofuran; TIPDS, tetraisopropyldisiloxan-1,3-diyl



3-((Benzyloxy)methyl)-1-((6aR,8R,9R,9aS)-9-hydroxy-2,2,4,4-tetraisopropyltetrahydr o-6H-furo[3,2-f][1,3,5,2,4]trioxadisilocin-8-yl)pyrimidine-2,4(1H,3H)-dione (S1): To a solution of uridine (2.44 g, 10 mmol) in pyridine (50 mL) was added TIPDSCl₂ (3.28 mL, 10.5 mmol). After stirring at rt for 40 h, MeOH (5 mL) was added to the reaction mixture, and the solvent was removed in vacuo. The resulting residue was partitioned between EtOAc (100 mL) and water (200 mL), and the aqueous layer was extracted with EtOAc (3 \times 30 mL). The combined organic layers were dried over anhydrous Na₂SO₄ and the filtrate was concentrated in vacuo. The resulting crude product was dissolved in DMF (50 mL) and DBU (2.99 mL, 20 mmol) and BOM-Cl (2.06 mL, 15 mmol) were added at 0 °C. After stirring for 1 h, MeOH (5 mL) was added to the reaction mixture, which was then partitioned between EtOAc (50 mL) and water (100 mL). The aqueous layer was extracted with EtOAc (3×50 mL), and the combined organic layer was washed with water $(2 \times 200 \text{ mL})$ and brine (200 mL) and dried over anhydrous Na₂SO₄. After the filtrate was concentrated in vacuo, the resulting residue was purified by silica gel column chromatography (Hex/EtOAc = 2:1 to 2:3) to obtain 4.925 g of the title compound as a colorless oil (8.12 mmol, 81% over two steps).

¹**H NMR (500 MHz, CDCl**₃) δ 7.63 (d, *J* = 8.1 Hz, 1H, H-6), 7.39 – 7.29 (m, 4H, Ph), 7.26 (d, *J* = 0.8 Hz, 1H, Ph, overlapped with CHCl₃), 5.73 (s, 1H, H-1'), 5.71 (d, *J* = 8.2 Hz, 1H, H-5), 5.52 – 5.43 (m, 2H, NC<u>H</u>₂O), 4.72 (s, 2H, OC<u>H</u>₂Ph), 4.35 (dd, *J* = 8.8, 4.8 Hz, 1H, H-2'), 4.20 (d, *J* = 12.3 Hz, 1H, H-4'), 4.13 (d, *J* = 4.8 Hz, 1H, H-3'), 4.10 (dt, *J* = 9.0 Hz, 1H, H-5'a), 4.00 (dd, *J* = 13.2, 3.0 Hz, 1H, H-5'b), 1.13 – 0.98 (m, 28H, SiCH(CH₃)₂ x 4); ¹³C NMR (126 MHz, CDCl₃) δ 162.68, 150.78, 138.61, 138.08, 128.32, 127.68, 127.65, 101.57, 91.47, 82.01, 75.29, 72.44, 70.41, 69.21, 60.43, 17.48, 17.42, 17.32, 17.30, 17.11, 17.04, 16.99, 16.91, 13.50, 13.01, 12.98, 12.65; HRMS (ESI) Calcd. for [C₂₉H₄₇N₂O₈Si₂]⁺ 607.2861, found 607.2858.



(6aR,8R,9R,9aR)-8-(3-((Benzyloxy)methyl)-2,4-dioxo-3,4-dihydropyrimidin-1(2H)-yl) -2,2,4,4-tetraisopropyltetrahydro-6H-furo[3,2-f][1,3,5,2,4]trioxadisilocin-9-yl

phenylcarbamate (1): To a solution of S1 (5.11 g, 8.42 mmol) in THF (42 mL) was added DBU (1.26 mL, 8.42 mmol) and PhNCO (912 μ L, 8.42 mmol) and the reaction was stirred at rt for 2 h. the reaction mixture was concentrated in vacuo, and the resulting residue was purified by silica gel column chromatography (Hex/EtOAc = 2:1) to afford 5.32 g of the title compound as a white amorphous solid (7.33 mmol, 87%).

Note: This compound has the same $R_{\rm f}$ value as S1.

¹H NMR (400 MHz, CDCl₃) δ 7.62 (d, J = 8.2 Hz, 1H, H-6), 7.43 – 7.28 (m, 9H, Ph), 7.08 (t, J = 7.5 Hz, 1H, Ph), 6.76 (s, 1H, NH), 5.86 (s, 1H, H-1'), 5.72 (d, J = 8.2 Hz, 1H, H-5), 5.53 – 5.43 (m, 2H, NC<u>H</u>₂O), 5.35 (d, J = 5.0 Hz, 1H, H-2'), 4.71 (d, J = 4.0Hz, 2H, OC<u>H</u>₂Ph), 4.40 (dd, J = 9.2, 5.0 Hz, 1H, H-3'), 4.22 (d, J = 12.0 Hz, 1H, H-4'), 4.06 – 3.96 (m, 2H, H-5'ab), 1.16 – 0.93 (m, 28H, SiCH(CH₃)₂ x 4); ¹³C NMR (126 MHz, CDCl₃) δ 162.73, 162.57, 150.71, 138.10, 137.73, 129.12, 129.08, 128.38, 127.74, 127.70, 127.67, 123.79, 102.02, 101.62, 89.41, 82.43, 76.27, 72.49, 70.59, 68.22, 59.92, 17.52, 17.47, 17.36, 17.34, 17.15, 17.08, 17.02, 17.00, 16.97, 16.95, 16.93, 16.91, 13.91, 13.61, 13.54, 13.32, 13.12, 13.05, 13.03, 13.02, 13.00, 12.80, 12.71, 12.70; HRMS (ESI) Calcd. for [C₃₆H₅₂N₃O₉Si₂]⁺ 726.3237, found 726.3229.



(6aR,8R,9R,9aR)-8-(3-((Benzyloxy)methyl)-2,4-dioxo-3,4-dihydropyrimidin-1(2H)-yl) -2,2,4,4-tetraisopropyltetrahydro-6H-furo[3,2-f][1,3,5,2,4]trioxadisilocin-9-yl

(3-oxobutyl)(phenyl)carbamate (3): Compound 1 (145 mg, 0.2 mmol), Bu₄NPF₆ (387 mg, 1 mmol), CH₂Cl₂ (10 mL), phosphate base (90 mg, 0.2 mmol) and methyl vinyl ketone (32.7 μ L, 0.4 mmol) were added to a test tube, which was then subjected to a constant electrical current of 5 mA (3 F/mol, 57.9 C) through the CF anode (1 × 1 cm) and the Pt cathode (1 × 1 cm). The reaction mixture was concentrated in vacuo and Et₂O (20 mL) was added. The resulting precipitate was removed by filtration through a short silica gel pad under reduced pressure. The filtrate was concentrated in vacuo and the resulting residue was purified by silica gel column chromatography (Hex/EtOAc = 3:1 to 2:1) afforded 78 mg of the title compound as a colorless oil (0.098 mmol, 49%). The analytical yield was determined based on ¹H NMR spectra, using benzaldehyde (20.4 μ L, 0.2 mmol) as an internal standard and the integral of NC<u>H₂</u>CH₂COMe in **3** was compared with that of the reference peak (C<u>H</u>O of benzaldehyde).

¹H NMR (500 MHz, CDCl₃) δ 7.84 (d, J = 8.1 Hz, 1H, H-6), 7.41 – 7.20 (m, 9H, Ph,

overlapped with CHCl₃), 7.07 (m, 1H, Ph), 6.12 (d, *J* = 3.6 Hz, 1H, H-1'), 5.73 (d, *J* = 8.2 Hz, 1H, H-5), 5.47 (s, 2H, NC<u>H</u>₂O), 5.20 – 5.16 (m, 1H, H-2'), 4.68 (s, 2H, OC<u>H</u>₂Ph), 4.53 (t, *J* = 5.5 Hz, 1H, H-3'), 4.14 – 4.05 (m, 3H, H-4', H-5'ab), 2.44 (t, 2H, C<u>H</u>₂COCH₃), 2.13 (s, 3H, COC<u>H</u>₃), 1.56 (q, *J* = 3.5 Hz, 2H, NC<u>H</u>₂CH₂), 1.14 – 0.92 (m, 28H, SiCH(CH₃)₂ x 4); ¹³C NMR (126 MHz, CDCl₃) δ 162.65, 151.28, 138.56, 138.09, 129.27, 128.46, 127.80, 127.78, 124.31, 102.47, 87.83, 84.63, 72.47, 70.62, 69.00, 61.32, 43.56, 30.00, 23.38, 17.49, 17.42, 17.40, 17.35, 13.59, 13.48, 13.15, 13.10, 13.00, 0.13; HRMS (APCI) Calcd. for [C₄₀H₅₈N₃O₁₀Si₂]⁺ 796.3655, found 796.3646.



(7aR,7bR,7'aR,7'bR,13aR,13'aR,14aR,14'aR)-2,2'-Bis((benzyloxy)methyl)-9,9,9',9',11 ,11,11',11'-octaisopropyl-5,5'-diphenyldodecahydro-1H,1'H,13H,13'H-[4,4'-bis[1,3,5, 2,4]trioxadisilocino[6',7':4,5][furo[2,3-f][pyrimido[6,1-d][1,3,5]oxadiazepine]-1,1',3,3', 6,6'(2H,2'H,4H,4'H)hexaone (4): Compound 1 (145 mg, 0.2 mmol), Bu₄NPF₆ (387 mg, 1 mmol), CH₂Cl₂ (10 mL), phosphate base (90 mg, 0.2 mmol), methyl vinyl ketone (32.7 μ L, 0.4 mmol), and HFIP (41.5 μ L, 0.4 mmol) were added to a test tube, which was then subjected to a constant electrical current of 5 mA (3 F/mol, 57.9 C) through the CF anode (1 × 1 cm) and the Pt cathode (1 × 1 cm). The reaction mixture was concentrated in vacuo and Et₂O (20 mL) was added. The resulting precipitate was removed by filtration through a short silica gel pad under reduced pressure. The filtrate was concentrated in vacuo and the resulting residue was passed through a short pad of silica gel (Hex/EtOAc = 1:2, containing 0.5% Et₃N) and further purified by preparative TLC (Hex:EtOAc = 2:3) to afford 39.7 mg of the title compound as a brown oil (0.027 mmol, 27%). The analytical yield was determined based on ¹H NMR spectra, using benzaldehyde (20.4 μ L, 0.2 mmol) as an internal standard and the integral of SiC<u>H(CH₃)₂ in 4</u> was compared with that of the reference peak (C<u>H</u>O of benzaldehyde).

¹**H** NMR (500 MHz, CDCl₃) δ 7.68 (d, J = 7.9 Hz, 1H, Ph), 7.62 (d, J = 8.2 Hz, 1H, Ph), 7.40 – 7.20 (m, 18H, Ph), 5.83 (s, 1H, H-1'), 5.73 – 5.66 (m, 3H, H-1', NCH₂O), 5.52 – 5.37 (m, 6H, NCH₂O, H-2' x 2, H-3' x 2), 4.71 (s, 4H, OCH₂Ph x 2), 4.46 – 3.86 (m, 10H, H-4' x 2, H-5'ab x 2, H-5 x 2, H-6 x 2), 1.64 (m, 4H, SiCH(CH₃)₂ x 4, overlapped with H₂O), 1.35 (m, 4H, SiCH(CH₃)₂ x 4), 1.16 – 0.85 (m, 48H, SiCH(CH₃)₂ x 8); ¹³C NMR (126 MHz, CDCl₃) δ 162.67, 151.85, 150.69, 138.17, 130.24, 129.16, 129.04, 128.57, 128.44, 128.43, 127.75, 127.73, 101.91, 89.39, 89.17, 82.28, 82.06, 76.61, 72.54, 70.62, 69.28, 69.19, 68.20, 68.03, 59.61, 32.32, 32.30, 32.25, 23.38, 18.68, 18.67, 17.57, 17.52, 17.51, 17.41, 17.39, 17.37, 17.21, 17.08, 17.04, 17.00, 16.98, 13.69, 13.64, 13.62, 13.08, 12.81, 12.76; HRMS (APCI) Calcd. for [C₇₂H₁₀₁N₆O₁₈Si₄]⁺ 1449.6244, found 1449.6250.

II. Simulation for cyclic voltammogram (Figures S1 and S2)

CV simulation was performed using DigiElch 8 (Gamry Instruments).

Simulation parameters:

$$1 \stackrel{\checkmark}{\leftarrow} 1^{++} + e^{-} \qquad E^{0} = 2.0 \text{ V}; \ k_{s(1)} = 1 \text{ cm s}^{-1}$$

$$1 + \text{phosphate base} \stackrel{\checkmark}{\leftarrow} 1 - \text{phosphate base} + e^{-} \quad E^{0} = 1.8 \text{ V}; \ k_{s(1+\text{phosphate base})} = 0.1 \text{ cm s}^{-1}$$

$$\text{cm s}^{-1}$$

1 + phosphate base + HFIP \implies 1-phosphate base-HFIP + e⁻ $E^0 = 1.8$ V; $k_{s(1+phosphate base+HFIP)} = 0.01$ cm s⁻¹



Figure S1. CV simulation for 1 and 1-phosphate base complex.



Figure S2. CV simulation for 1-phosphate base-HFIP complex.

We hypothesized that the two- or three-component hydrogen bond complexes (1-phosphate base and 1-phosphate base-HFIP) would have a diffusion coefficient smaller than that of 1. Although a decrease in the diffusion coefficient decreased the current value, similar to our experimental results, the reported diffusion coefficient of the hydrogen-bond complex between the amide and phosphate base is only twice as small as that of the sole amide molecule (Gschwind et al., *J. Am. Chem. Soc.*, **2021**, *143*, 724-735.), while our simulated diffusion coefficient was unrealistically smaller than previously reported values. Therefore, we concluded that the diffusion coefficient is not a major factor affecting the CV behavior in the present study.



III. ¹H and ¹³C NMR spectra of compounds





S17