



## Supporting Information

for

### Chemical and biosynthetic potential of *Penicillium shentong* XL-F41

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*Beilstein J. Org. Chem.* **2024**, *20*, 597–606. [doi:10.3762/bjoc.20.52](https://doi.org/10.3762/bjoc.20.52)

### NMR and mass spectra of isolated compounds

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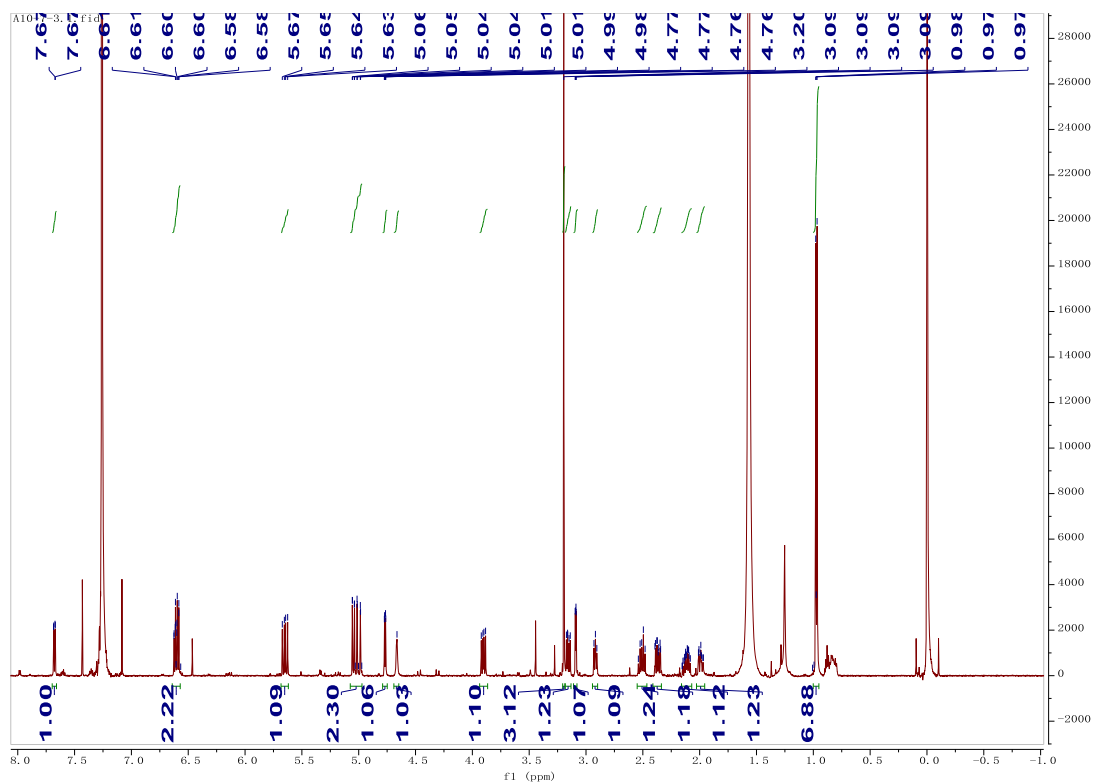


Figure S1. <sup>1</sup>H NMR spectrum (600 MHz, chloroform-d) of **1**

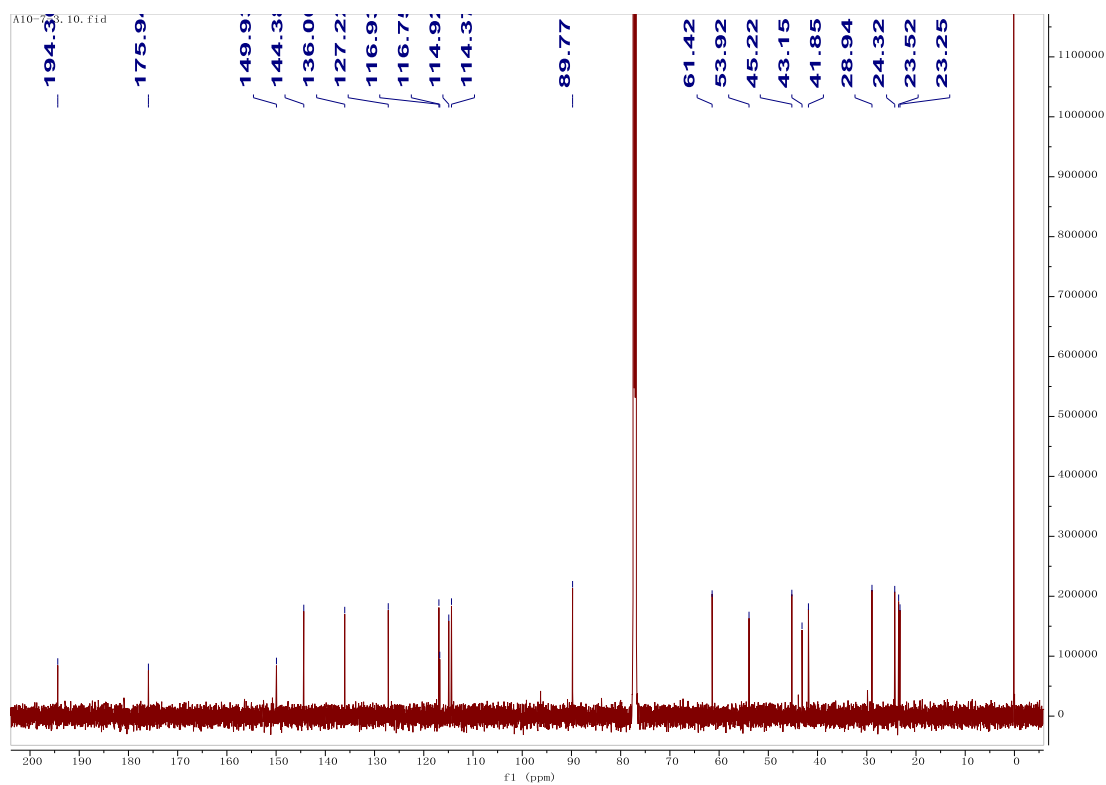


Figure S2. <sup>13</sup>C NMR spectrum (600 MHz, chloroform-d) of **1**

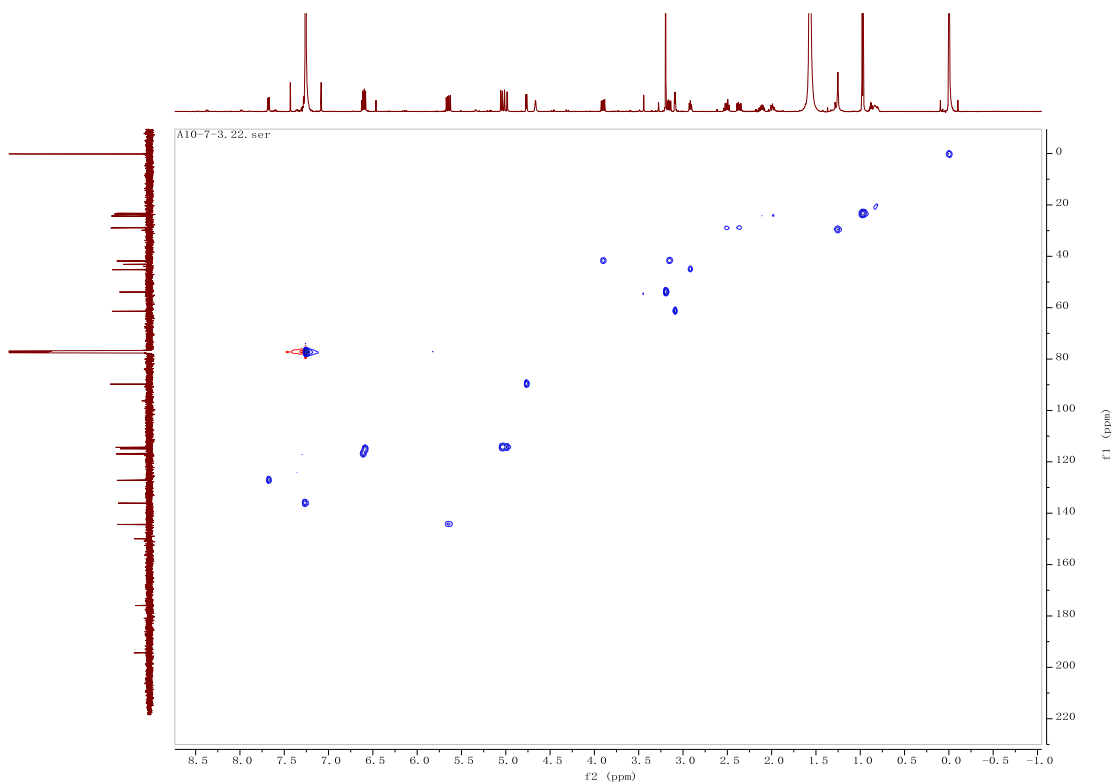


Figure S3. HSQC spectrum (600 MHz, chloroform-d) of **1**

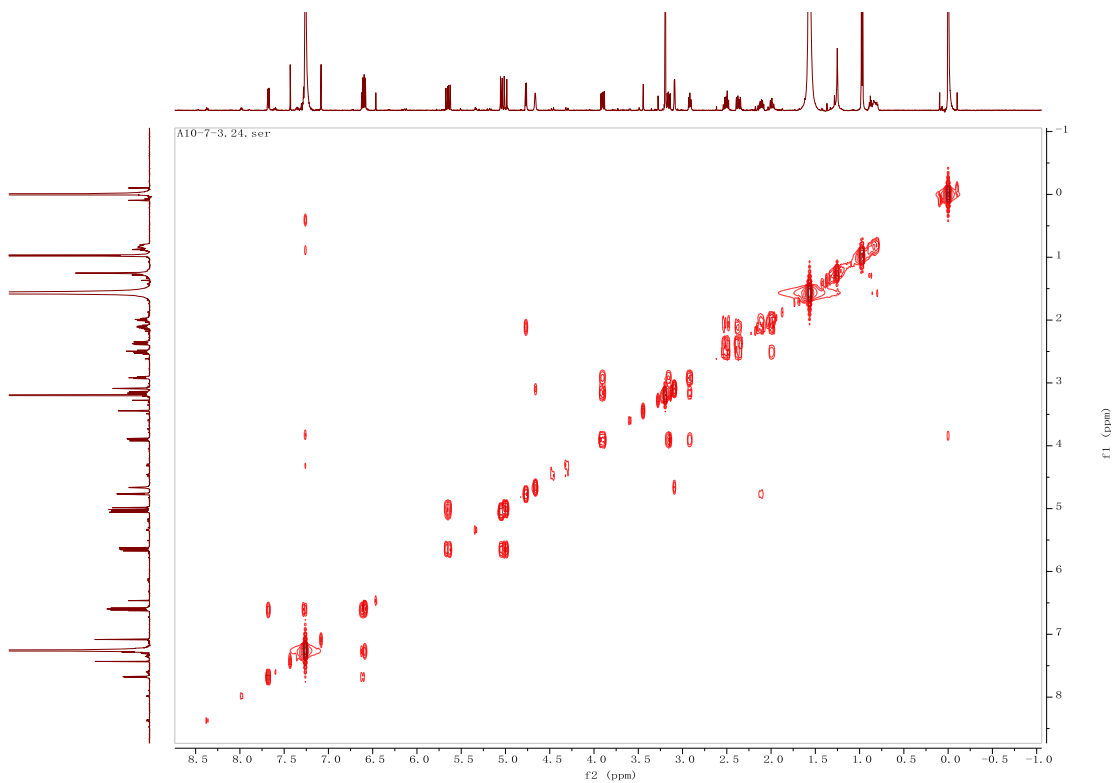


Figure S4.  $^1\text{H}$ - $^1\text{H}$  COSY spectrum (600 MHz, chloroform-d) of **1**

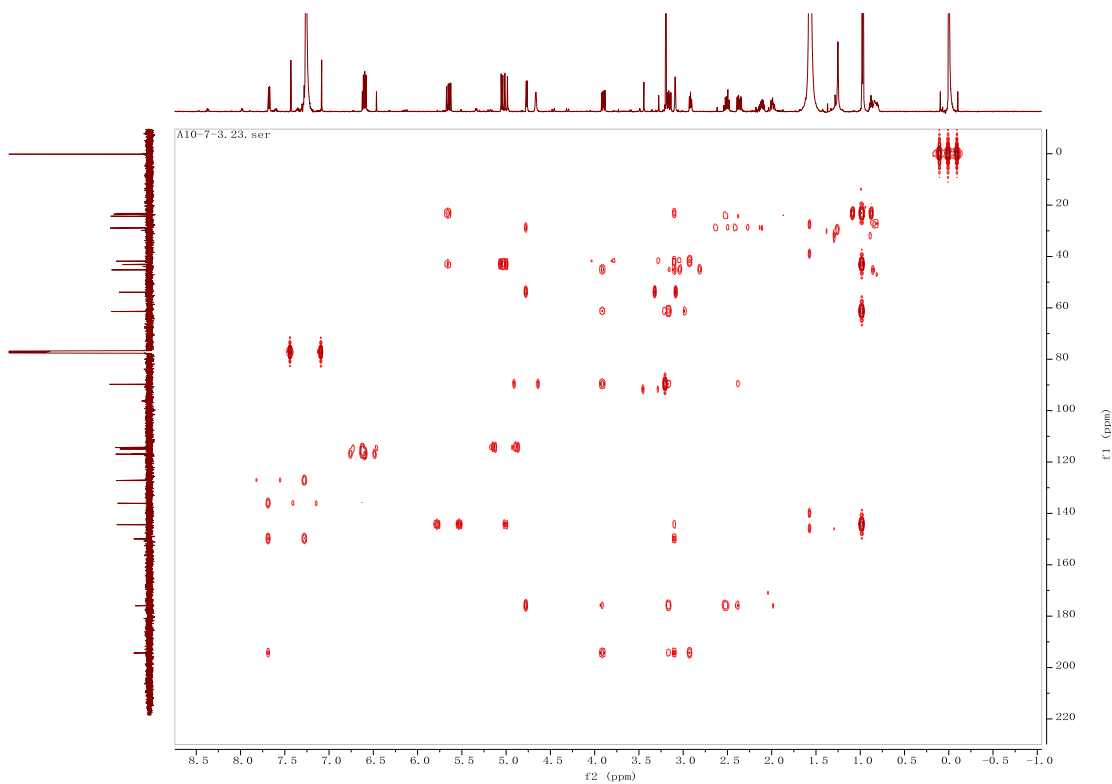


Figure S5. HMBC spectrum (600 MHz, chloroform-d) of **1**

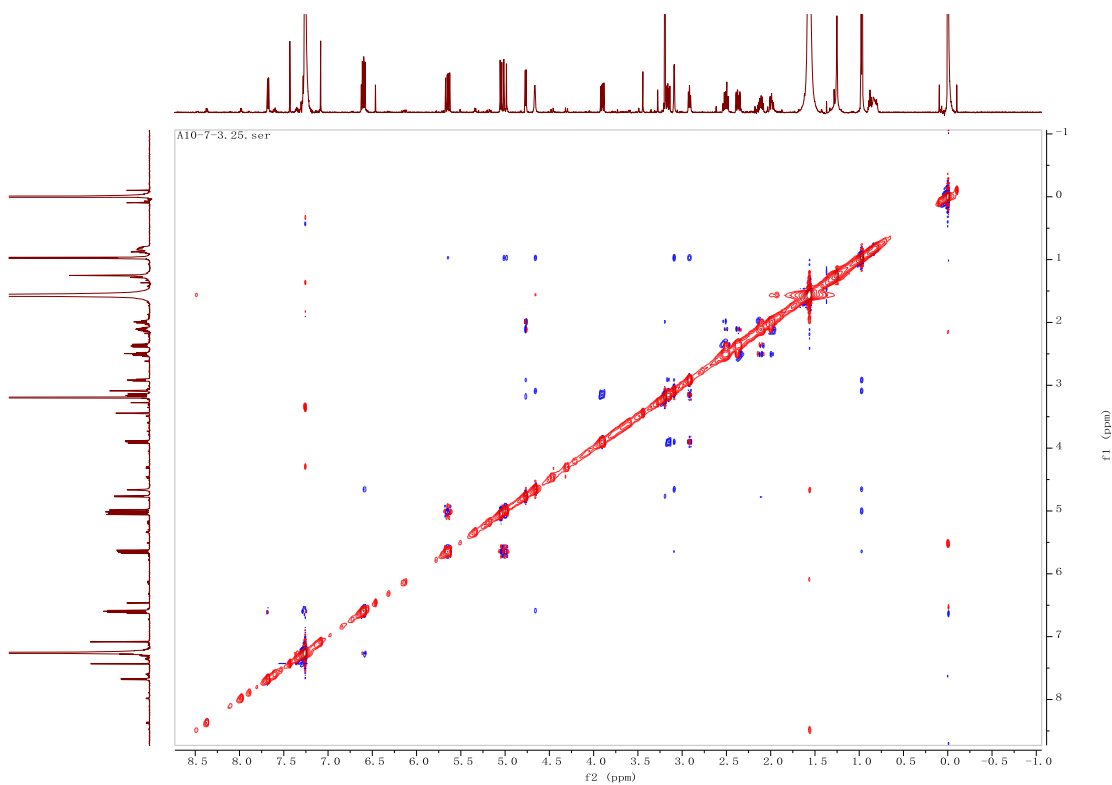


Figure S6. NOESY spectrum (600 MHz, chloroform-d) of **1**

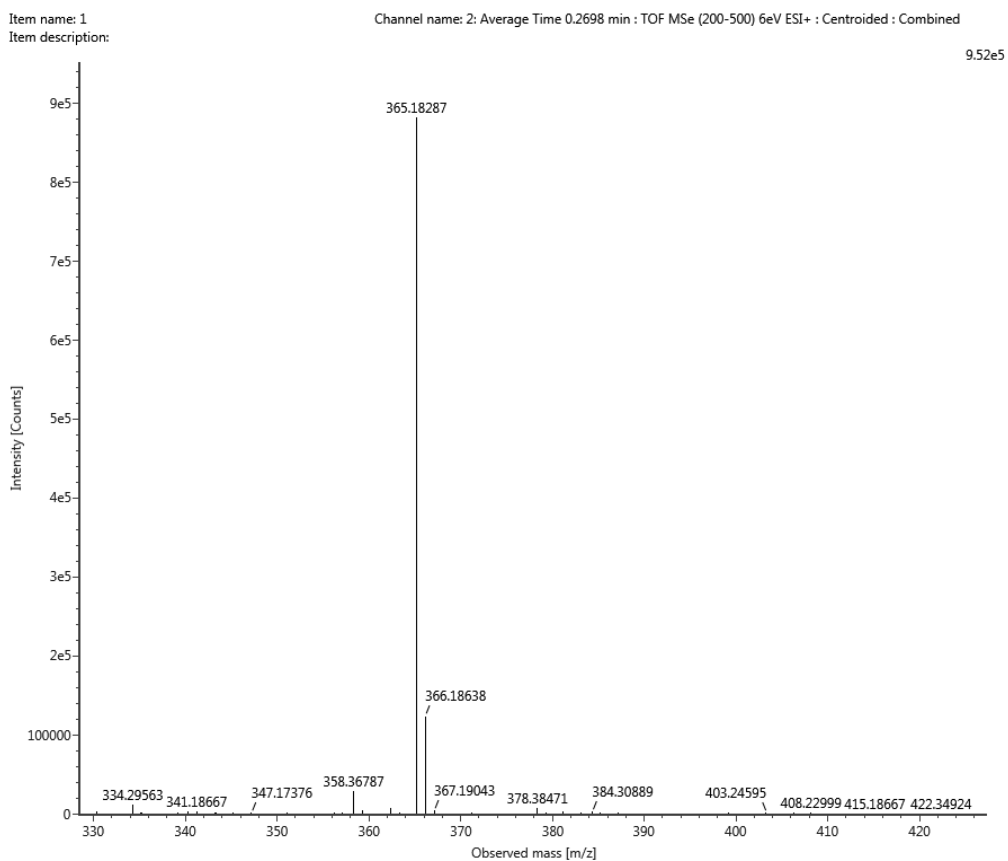


Figure S7. HRESIMS spectrum of **1**

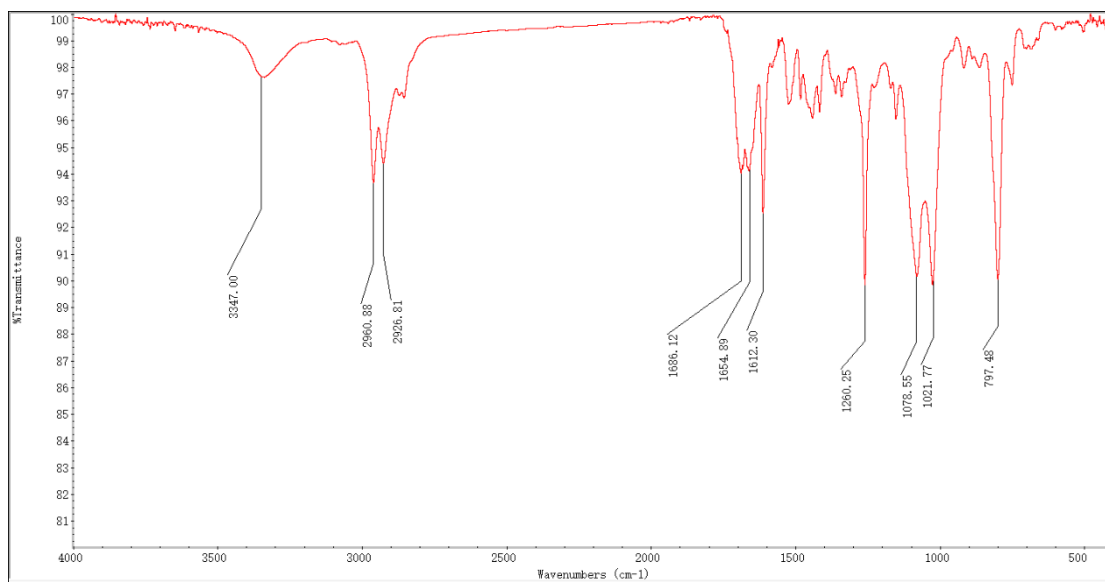


Figure S8. FT-IR spectrum of compound **1**

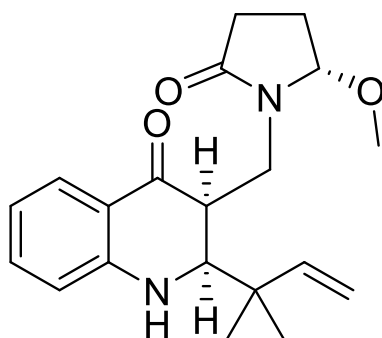


Figure S9. Chemical structure of compound 1



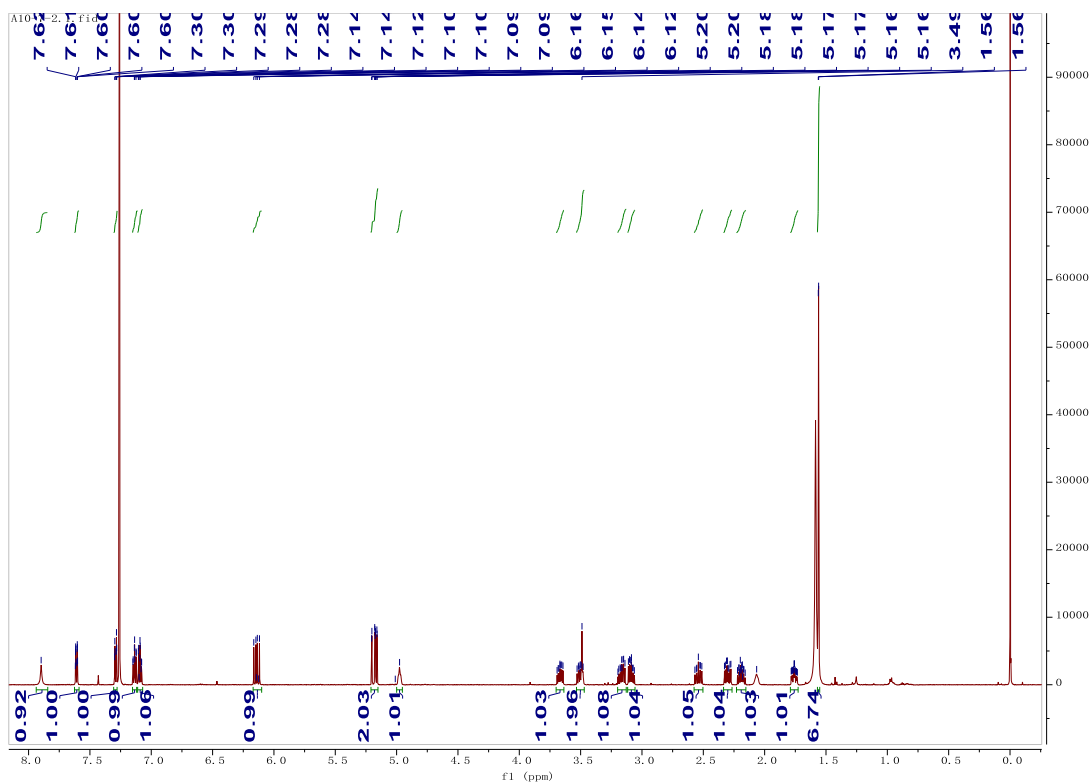


Figure S10.  $^1\text{H}$  NMR spectrum (600 MHz, chloroform-d) of **2**

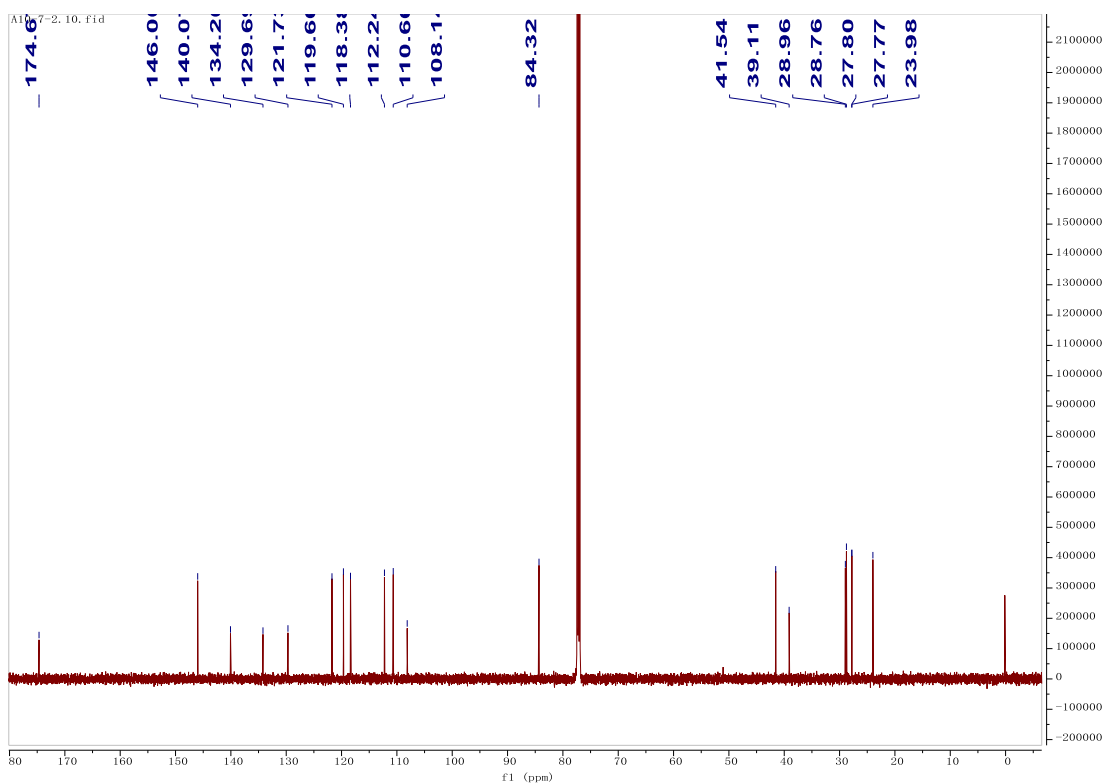


Figure S11.  $^{13}\text{C}$  NMR spectrum (600 MHz, chloroform-d) of **2**

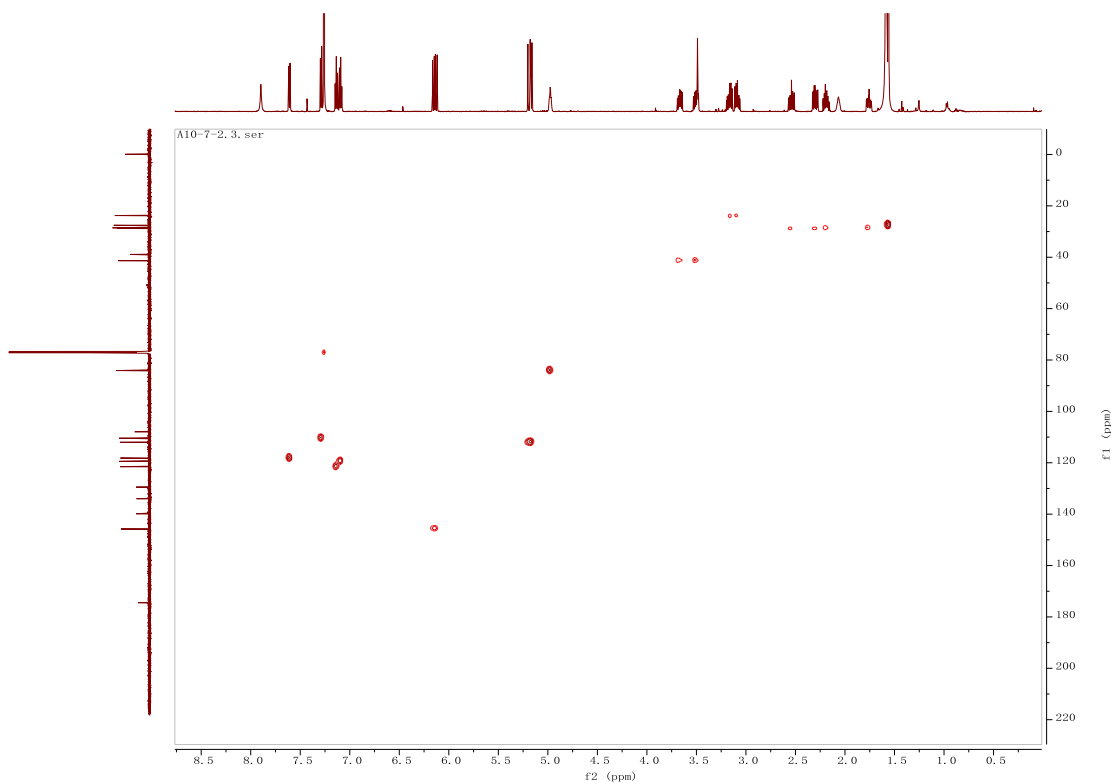


Figure S12. HSQC spectrum (600 MHz, chloroform-d) of **2**

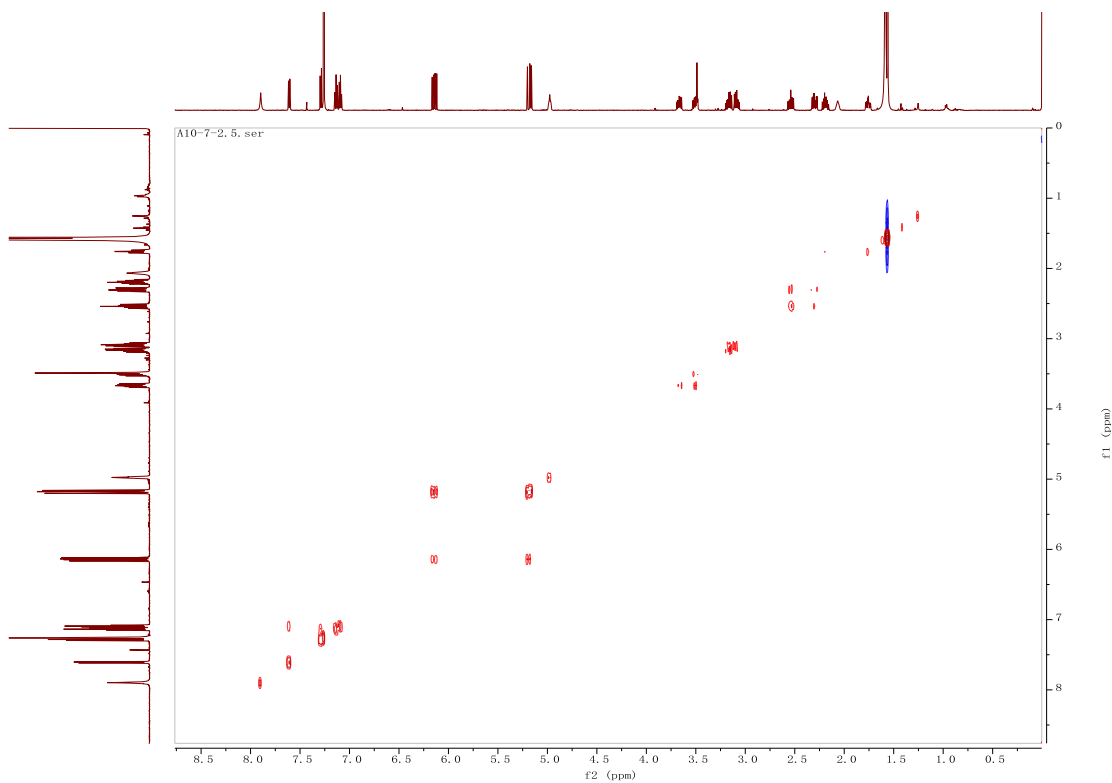


Figure S13.  $^1\text{H}$ - $^1\text{H}$  COSY spectrum (600 MHz, chloroform-d) of **2**

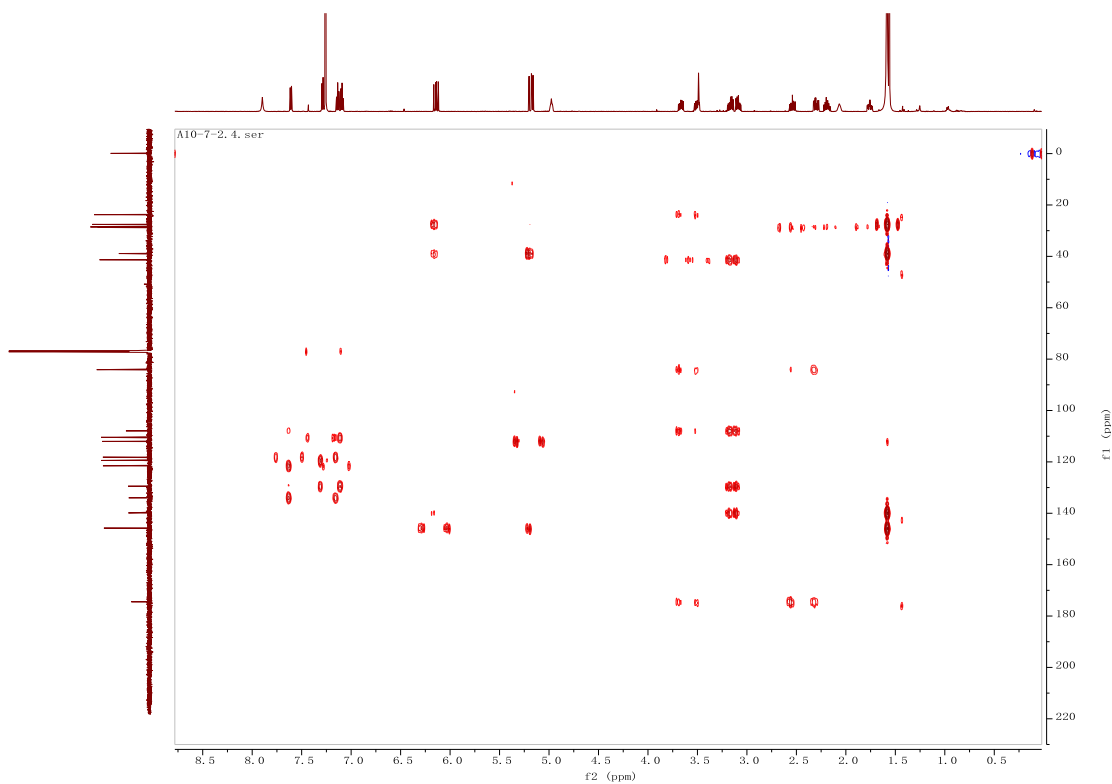


Figure S14. HMBC spectrum (600 MHz, chloroform-d) of **2**

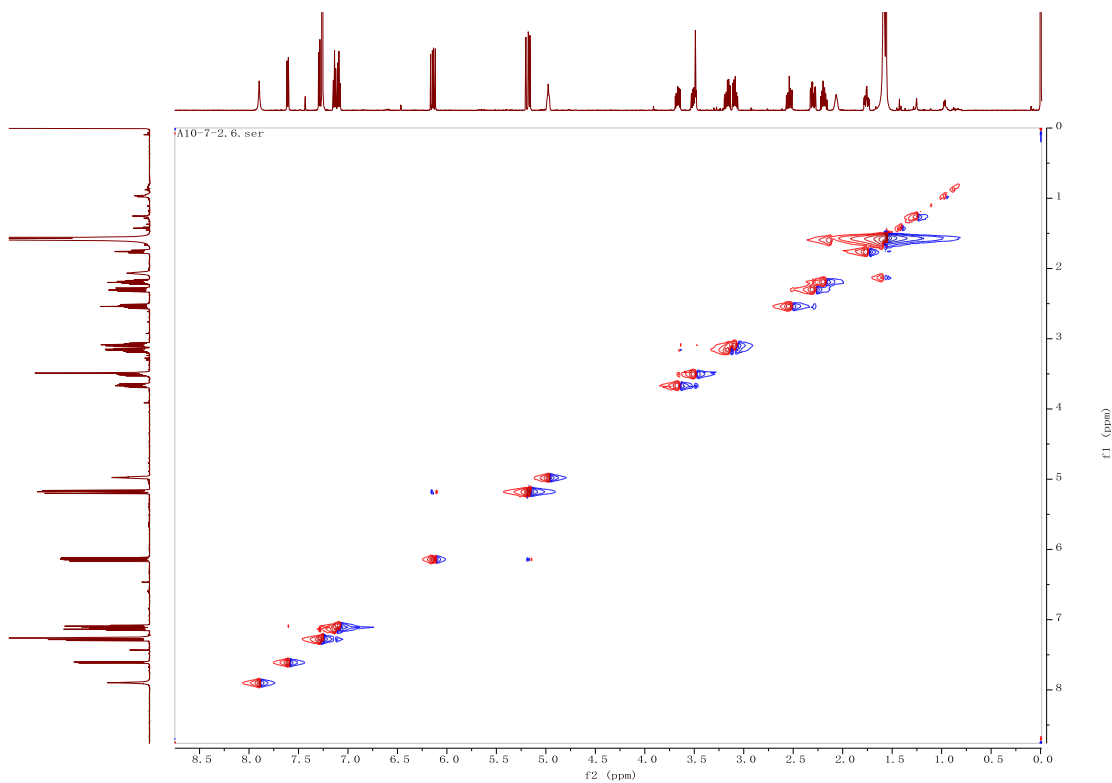


Figure S15. NOESY spectrum (600 MHz, chloroform-d) of **2**

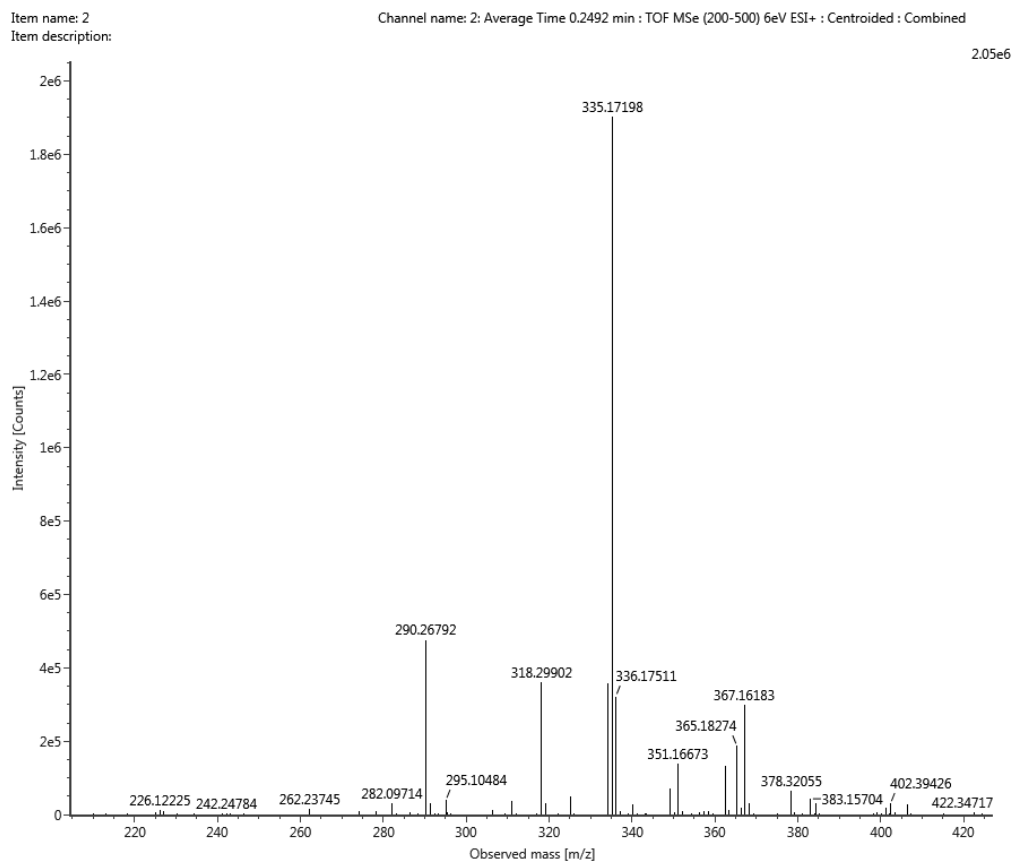


Figure S16. HRESIMS spectrum of **2**

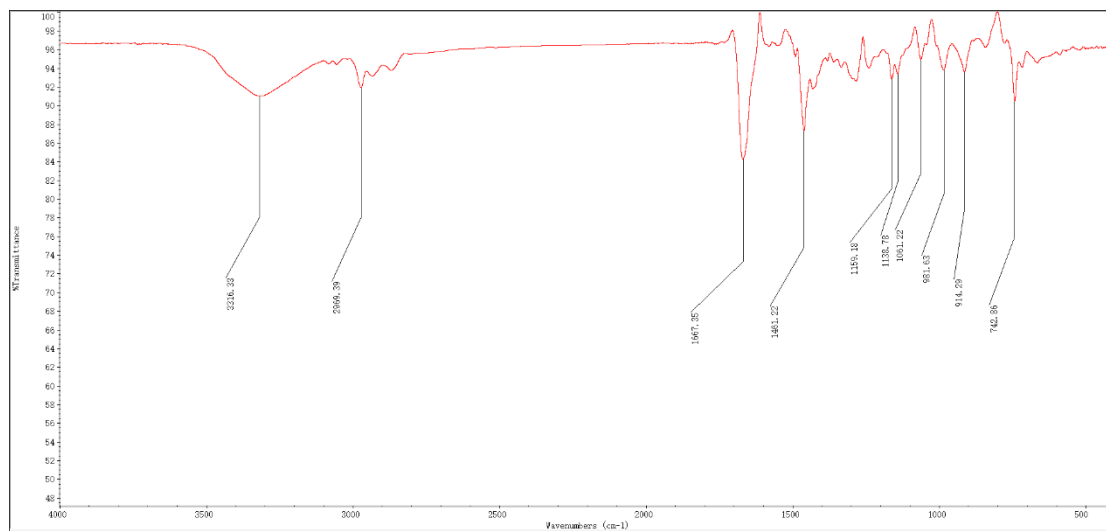


Figure S17. FT-IR spectrum of compound **2**

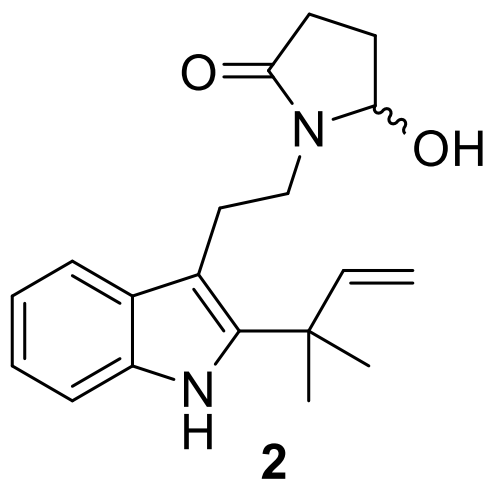


Figure S18. Chemical structure of compound 2



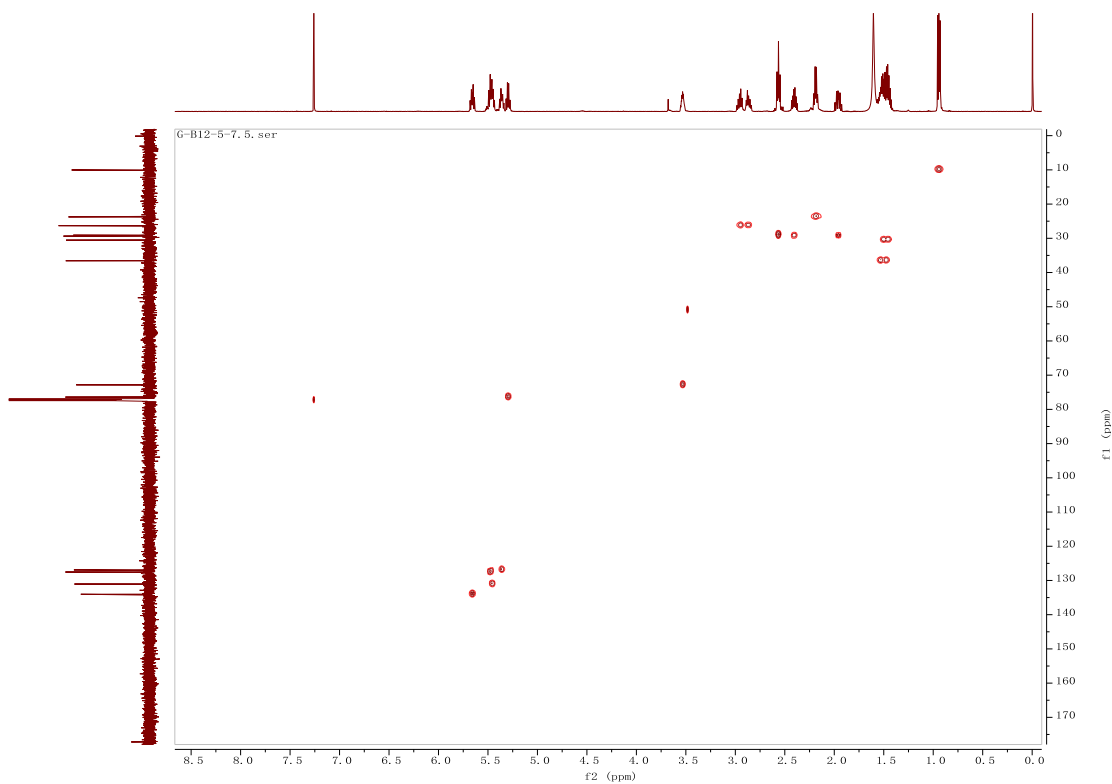


Figure S21. HSQC spectrum (600 MHz, chloroform-d) of **3**

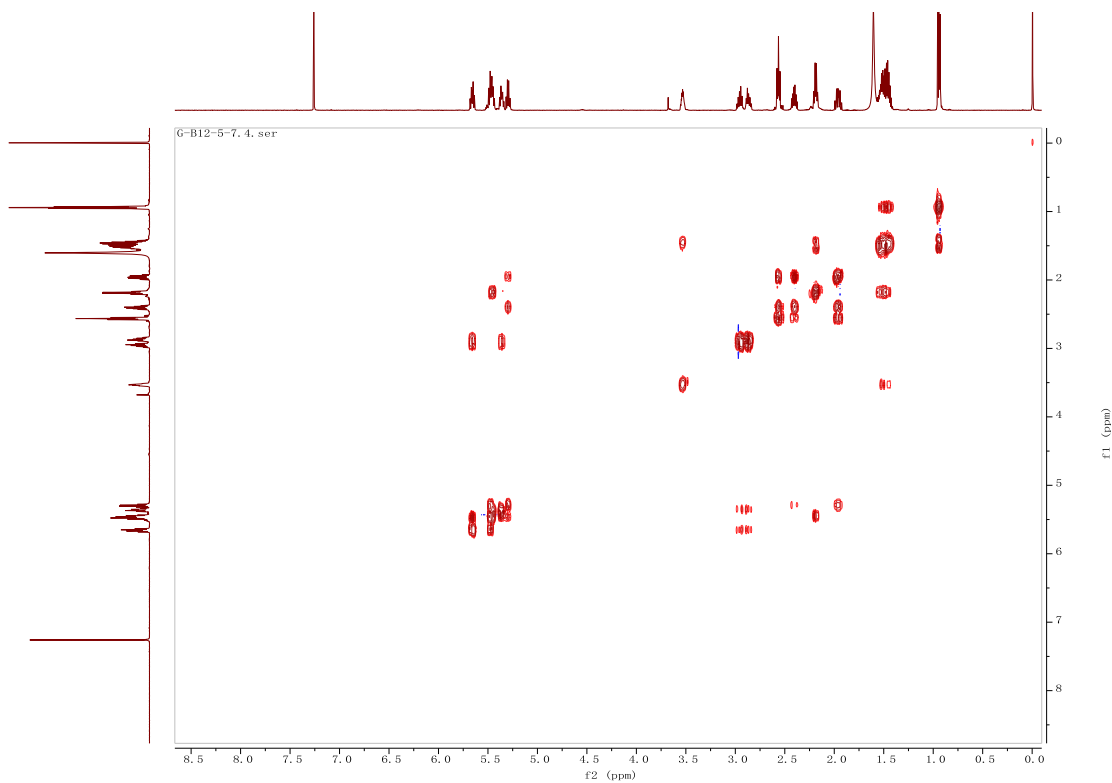


Figure S22.  $^1\text{H}$ - $^1\text{H}$  COSY spectrum (600 MHz, chloroform-d) of **3**

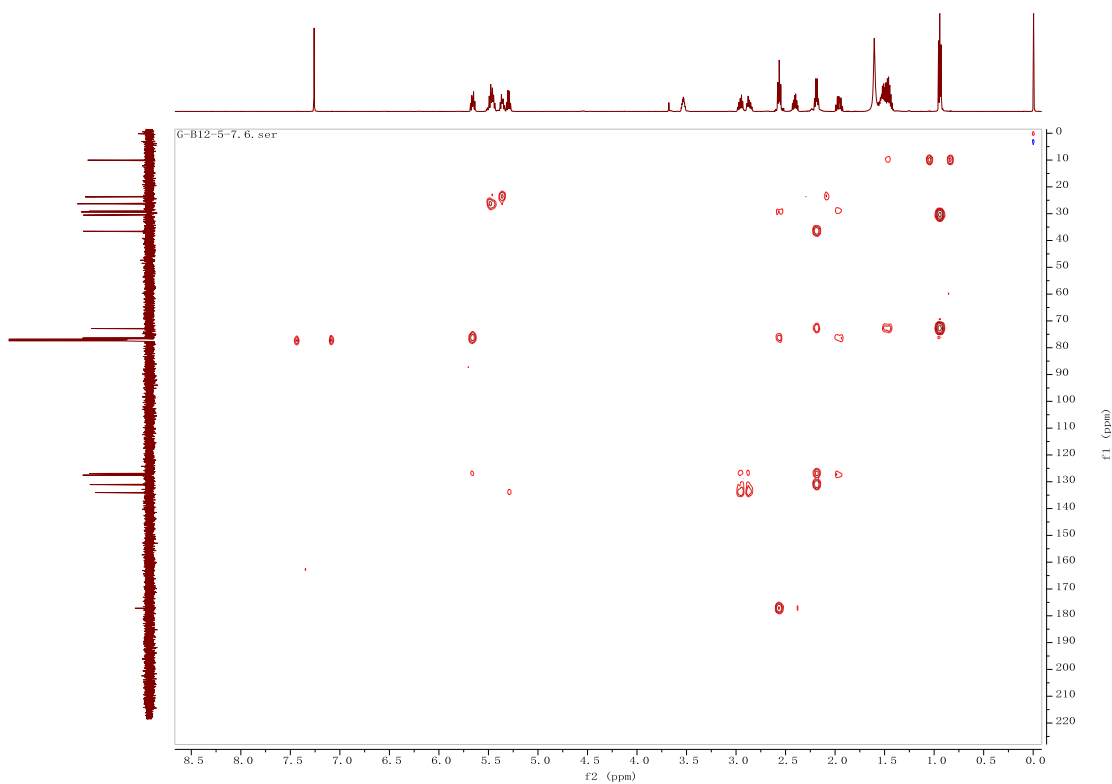


Figure S23. HMBC spectrum (600 MHz, chloroform-d) of **3**

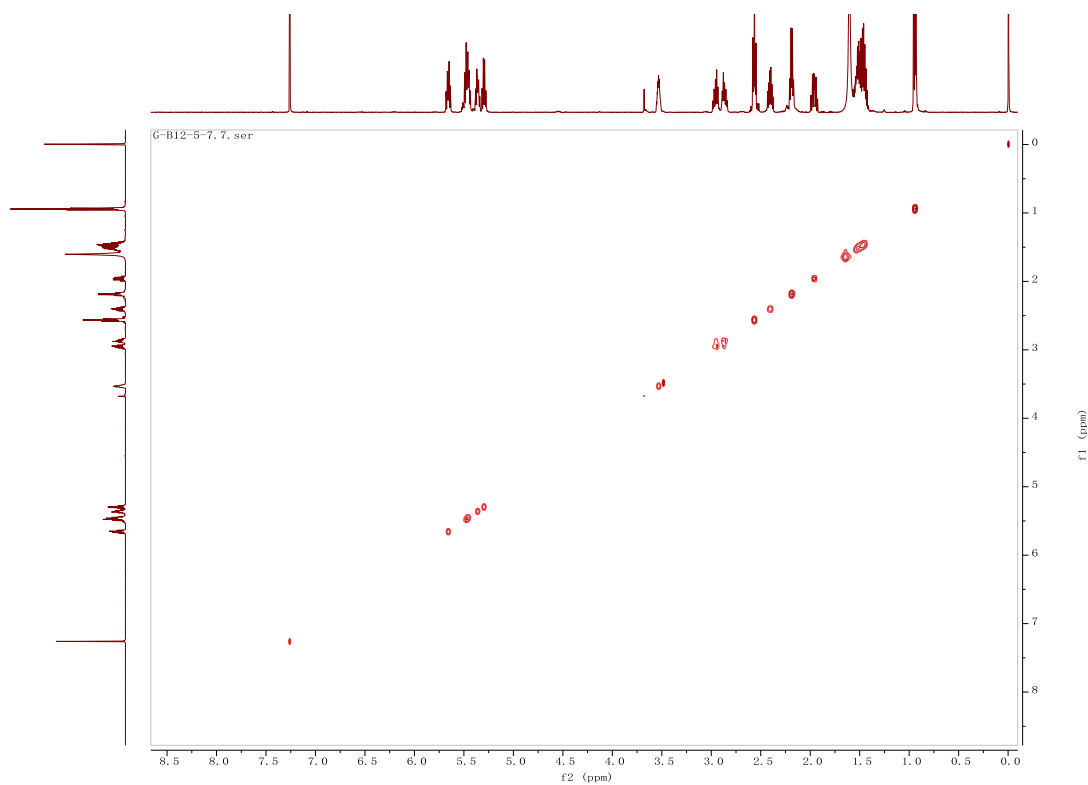


Figure S24. NOESY spectrum (600 MHz, chloroform-d) of **3**



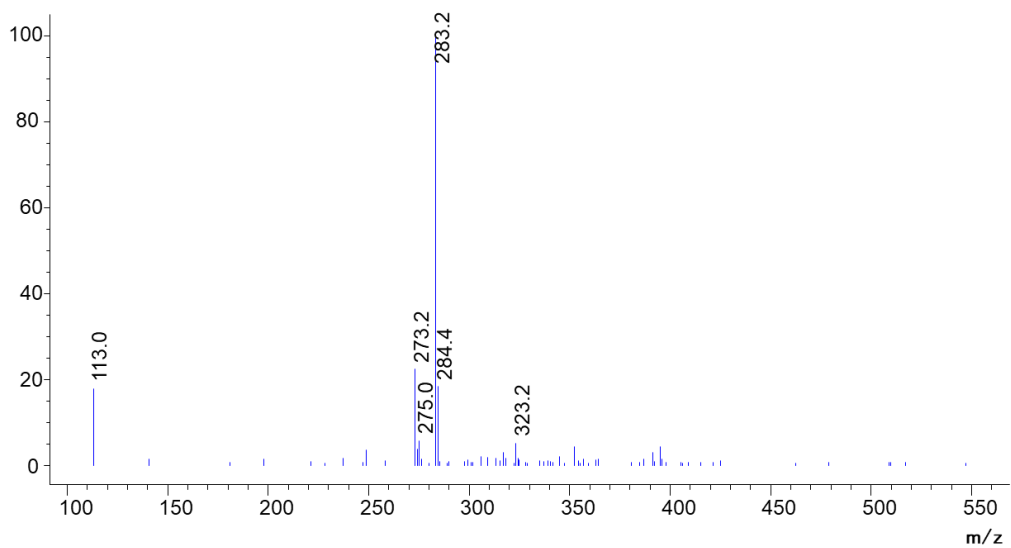


Figure S25. LC-MS spectrum of **3**

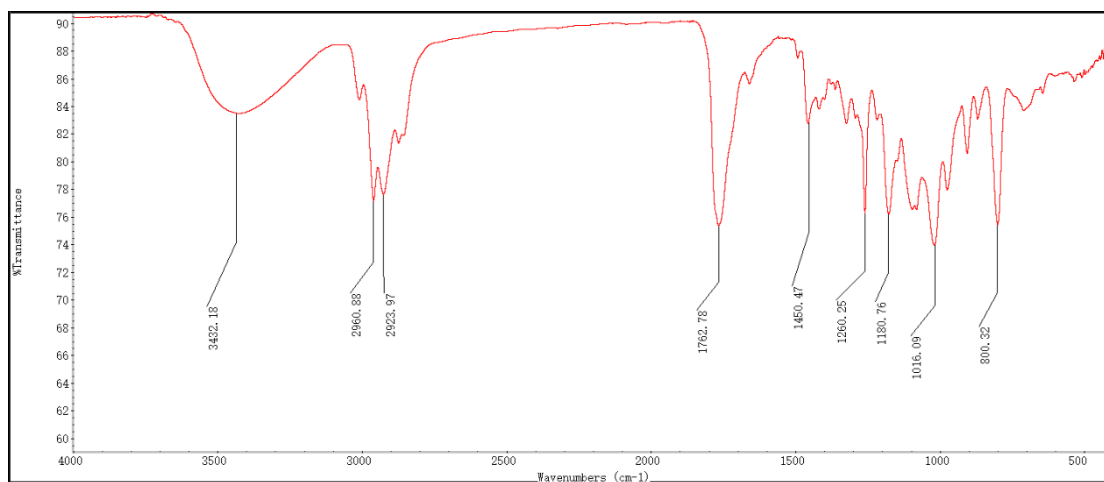


Figure S26. FT-IR spectrum of compound **3**

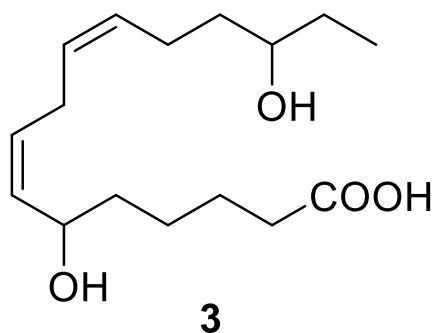


Figure S27. Chemical structure of compounds **3**

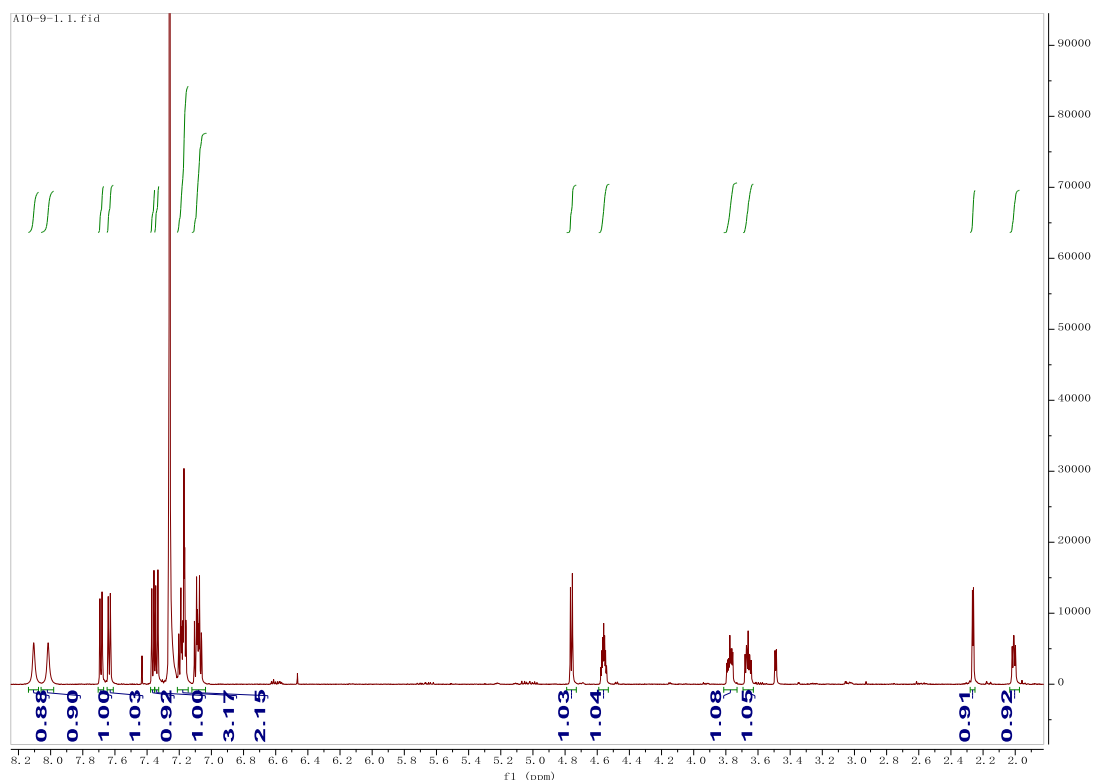


Figure S28. <sup>1</sup>H NMR spectrum (600 MHz, chloroform-d) of **4**

<sup>1</sup>H NMR (600 MHz, chloroform-d)  $\delta$  8.11 (s, 1H), 8.02 (s, 1H), 7.69 (dd,  $J$  = 8.0, 1.0 Hz, 1H), 7.63 (dd,  $J$  = 8.0, 1.0 Hz, 1H), 7.36 (d,  $J$  = 8.2 Hz, 1H), 7.34 (d,  $J$  = 8.1 Hz, 1H), 7.22 – 7.15 (m, 3H), 7.12 – 7.05 (m, 2H), 4.76 (d,  $J$  = 7.2 Hz, 1H), 4.56 (tt,  $J$  = 7.1, 3.6 Hz, 1H), 3.78 (ddd,  $J$  = 10.9, 7.2, 3.5 Hz, 1H), 3.66 (ddd,  $J$  = 11.4, 6.7, 4.9 Hz, 1H), 2.26 (d,  $J$  = 3.9 Hz, 1H), 2.01 (dd,  $J$  = 7.3, 5.1 Hz, 1H). Compound **4** was determined by comparison of its spectroscopic data with fusarindole B in the literature [1].

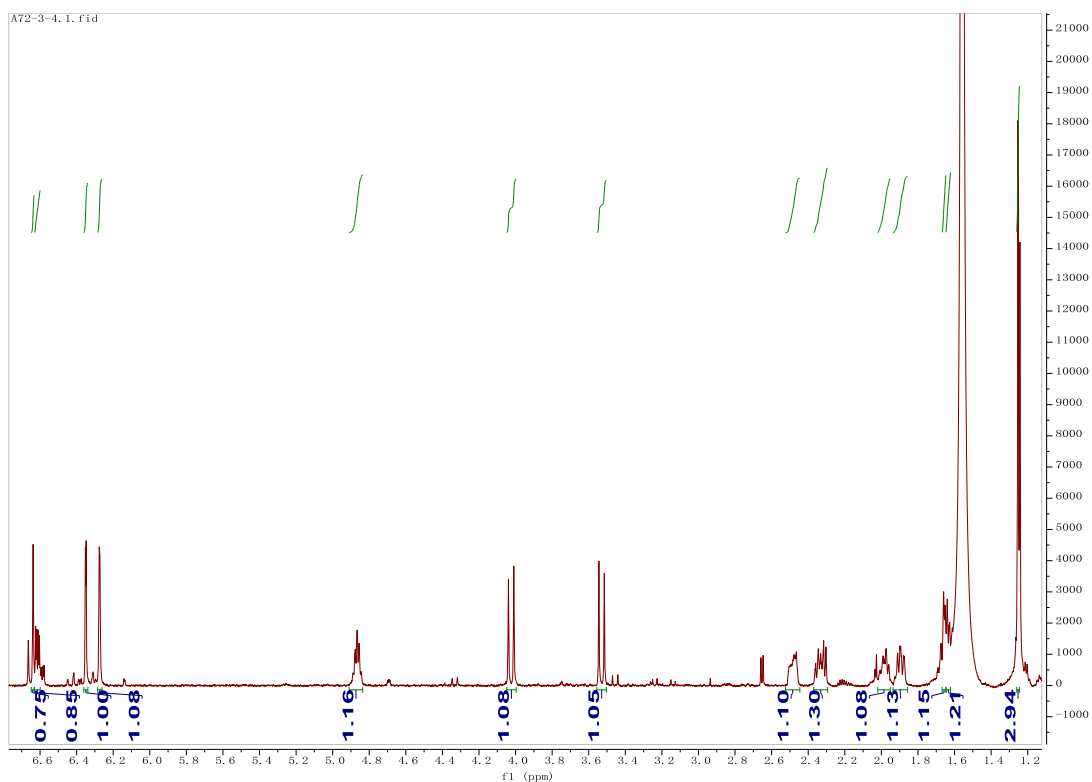


Figure S29.  $^1\text{H}$  NMR spectrum (600 MHz, chloroform-d) of **5**

$^1\text{H}$  NMR (600 MHz, chloroform-d)  $\delta$  6.64 (d,  $J = 1.5$  Hz, 1H), 6.61 (dd,  $J = 8.6, 3.8$  Hz, 1H), 6.35 (d,  $J = 2.5$  Hz, 1H), 6.27 (d,  $J = 2.6$  Hz, 1H), 4.92 – 4.84 (m, 1H), 4.02 (d,  $J = 17.7$  Hz, 1H), 3.53 (d,  $J = 17.7$  Hz, 1H), 2.47 (d,  $J = 8.5$  Hz, 1H), 2.38 – 2.29 (m, 1H), 1.98 (q,  $J = 9.3, 7.6$  Hz, 1H), 1.93 – 1.86 (m, 1H), 1.66 (d,  $J = 4.9$  Hz, 1H), 1.65 – 1.63 (m, 1H), 1.25 (s, 3H). Compound **5** was determined by comparison of its spectroscopic data with dehydrocurvularin in the literature [2].

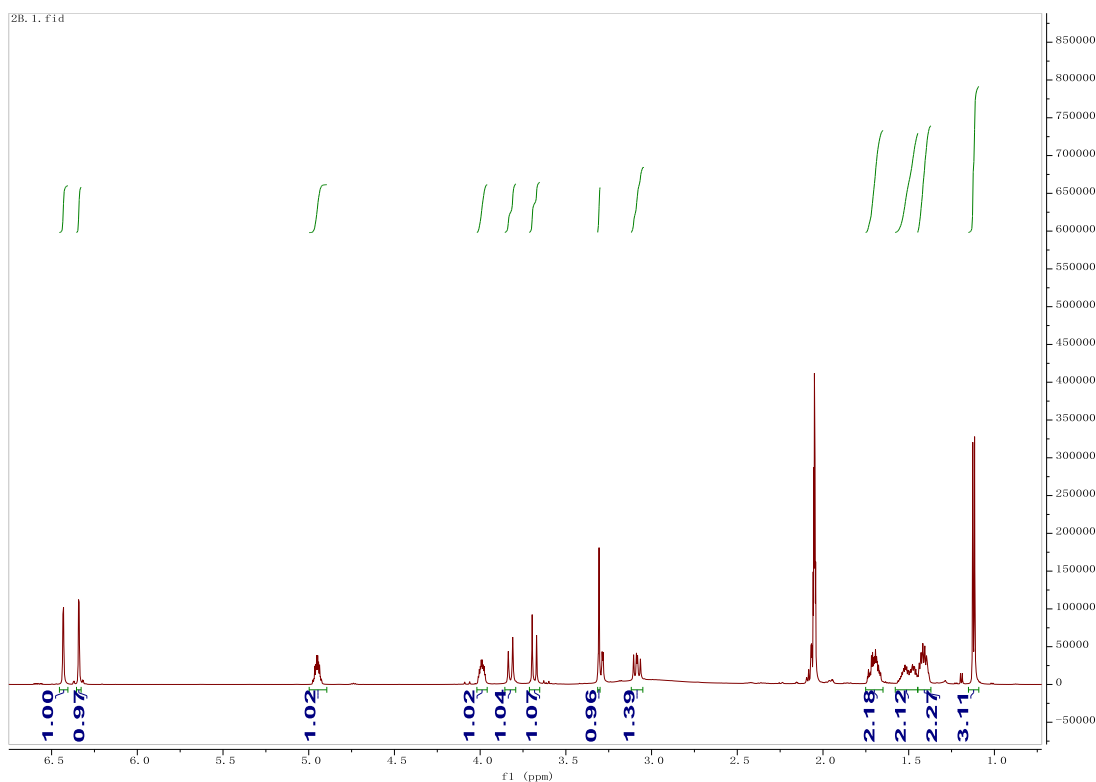


Figure S30.  $^1\text{H}$  NMR spectrum (600 MHz, acetone- $d_6$ ) of **6**

$^1\text{H}$  NMR (600 MHz, acetone- $d_6$ )  $\delta$  6.43 (d,  $J = 2.3$  Hz, 1H), 6.34 (d,  $J = 2.3$  Hz, 1H), 4.95 (td,  $J = 6.6, 3.5$  Hz, 1H), 3.99 (ddt,  $J = 10.4, 7.0, 3.6$  Hz, 1H), 3.82 (d,  $J = 15.4$  Hz, 1H), 3.68 (d,  $J = 15.4$  Hz, 1H), 3.31 (d,  $J = 3.2$  Hz, 1H), 3.08 (dd,  $J = 13.9, 10.0$  Hz, 1H), 1.81 – 1.33 (m, 6H), 1.12 (d,  $J = 6.4$  Hz, 3H). Compound **6** was determined by comparison of its spectroscopic data with hydroxycurvularin in the literature [3].

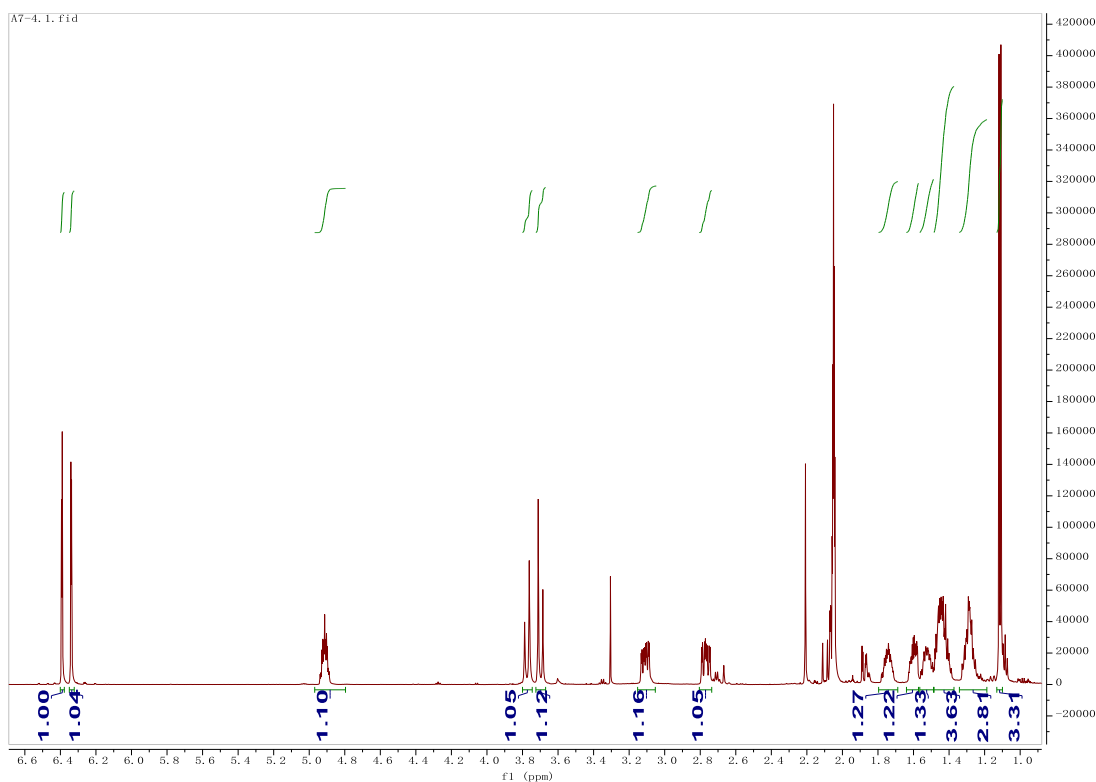


Figure S31.  $^1\text{H}$  NMR spectrum (600 MHz, acetone- $d_6$ ) of **7**

$^1\text{H}$  NMR (600 MHz, acetone- $d_6$ )  $\delta$  6.39 (d,  $J$  = 2.3 Hz, 1H), 6.34 (d,  $J$  = 2.3 Hz, 1H), 4.91 (dq,  $J$  = 9.0, 6.3, 2.8 Hz, 1H), 3.77 (d,  $J$  = 15.7 Hz, 1H), 3.70 (d,  $J$  = 15.6 Hz, 1H), 3.11 (ddd,  $J$  = 15.5, 8.6, 3.0 Hz, 1H), 2.77 (ddd,  $J$  = 15.5, 9.8, 3.0 Hz, 1H), 1.80 – 1.68 (m, 1H), 1.60 (ddt,  $J$  = 14.3, 7.9, 3.2 Hz, 1H), 1.56 – 1.49 (m, 1H), 1.49 – 1.37 (m, 3H), 1.35 – 1.21 (m, 2H), 1.11 (d,  $J$  = 6.3 Hz, 3H). Compound **7** was determined by comparison of its spectroscopic data with curvularin in the literature [4].

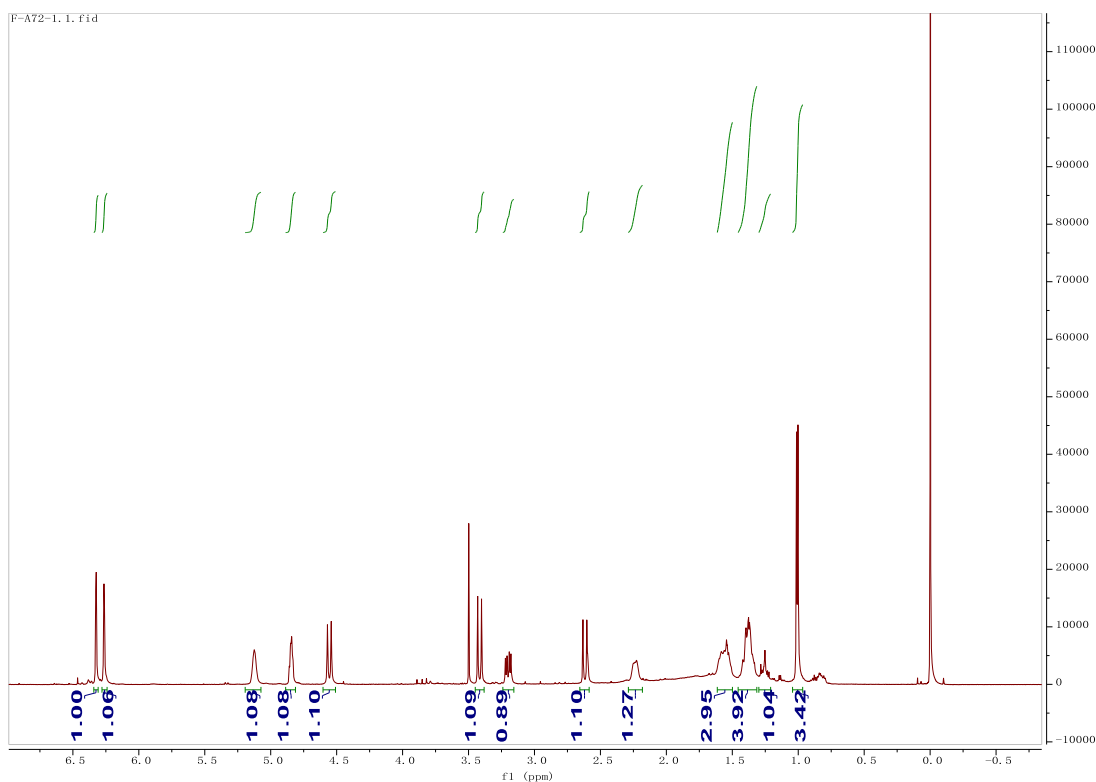


Figure S32. <sup>1</sup>H NMR spectrum (600 MHz, chloroform-d) of **8**

<sup>1</sup>H NMR (600 MHz, chloroform-d)  $\delta$  6.32 (d,  $J = 2.3$  Hz, 1H), 6.28 – 6.24 (m, 1H), 5.12 (s, 1H), 4.85 (q,  $J = 6.1, 5.6$  Hz, 1H), 4.56 (d,  $J = 17.4$  Hz, 1H), 3.42 (d,  $J = 17.4$  Hz, 1H), 3.20 (dd,  $J = 17.7, 7.6$  Hz, 1H), 2.62 (d,  $J = 17.7$  Hz, 1H), 2.23 (s, 1H), 1.66 – 1.49 (m, 2H), 1.44 – 1.31 (m, 3H), 1.30 – 1.22 (m, 1H), 1.01 (d,  $J = 6.6$  Hz, 3H). Compound **8** was determined by comparison of its spectroscopic data with curvulopyran in the literature [5].

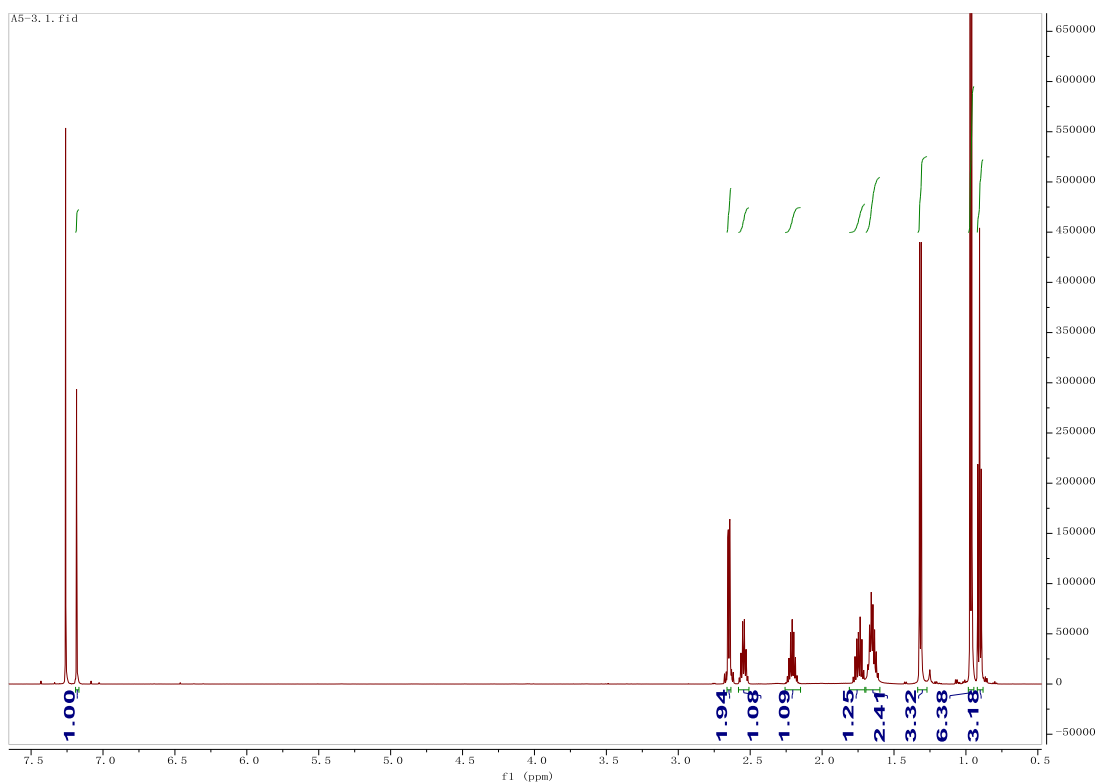


Figure S33.  $^1\text{H}$  NMR spectrum (600 MHz, chloroform-d) of **9**

$^1\text{H}$  NMR (600 MHz, chloroform-d)  $\delta$  7.18 (s, 1H), 2.65 (dd,  $J = 7.1, 1.9$  Hz, 2H), 2.55 (h,  $J = 7.1$  Hz, 1H), 2.21 (dh,  $J = 13.5, 6.7$  Hz, 1H), 1.75 (dt,  $J = 13.7, 7.5$  Hz, 1H), 1.65 (dp,  $J = 14.4, 7.3$  Hz, 2H), 1.32 (d,  $J = 7.0$  Hz, 3H), 0.97 (d,  $J = 6.7$  Hz, 6H), 0.91 (t,  $J = 7.4$  Hz, 3H). Compound **9** was determined by comparison of its spectroscopic data with (S)-6-(sec-butyl)-3-isobutylpyrazin-2(1H)-one in the literature [6].

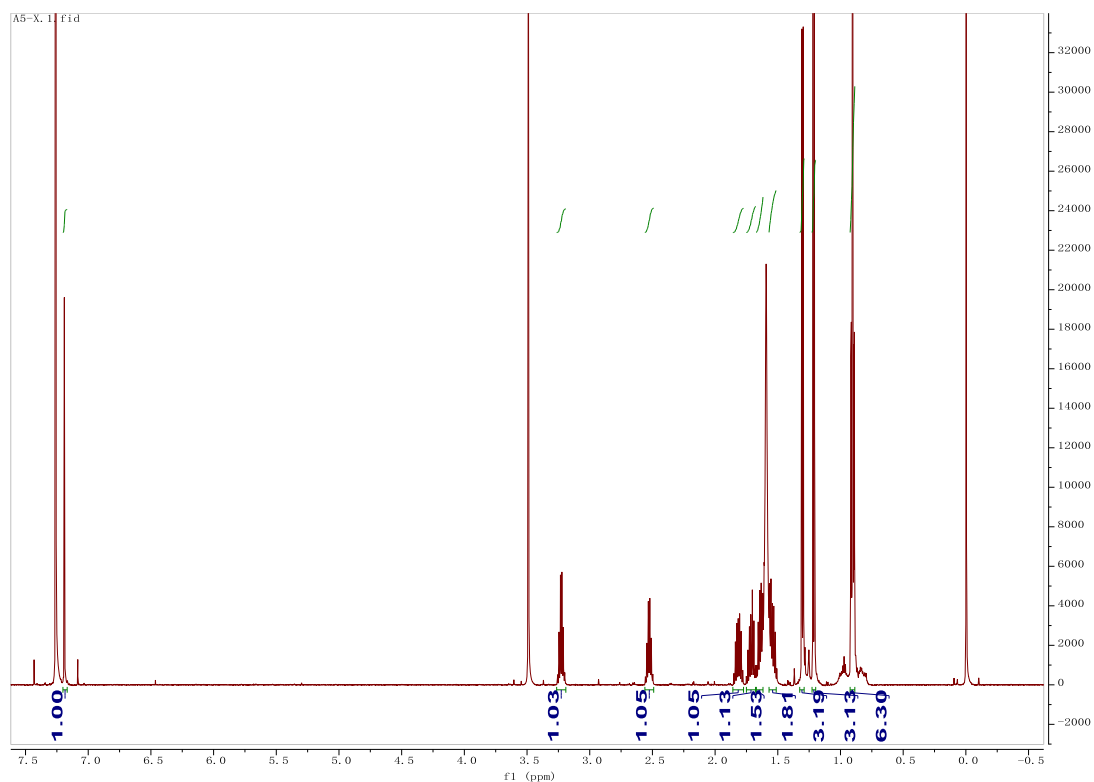


Figure S34.  $^1\text{H}$  NMR spectrum (600 MHz, chloroform-d) of **10**

$^1\text{H}$  NMR (600 MHz, chloroform-d)  $\delta$  7.19 (s, 1H), 3.23 (h,  $J = 6.9$  Hz, 1H), 2.53 (h,  $J = 7.1$  Hz, 1H), 1.87 – 1.77 (m, 1H), 1.71 (dt,  $J = 13.8, 7.5$  Hz, 1H), 1.64 (dt,  $J = 14.1, 7.2$  Hz, 1H), 1.54 (dt,  $J = 13.4, 7.3$  Hz, 1H), 1.30 (d,  $J = 7.0$  Hz, 3H), 1.21 (d,  $J = 6.9$  Hz, 3H), 0.90 (td,  $J = 7.4, 2.9$  Hz, 6H). Compound **10** was determined by comparison of its spectroscopic data with 3,6-di-sec-butyl-2(1H)-pyrazinone in the literature [6].



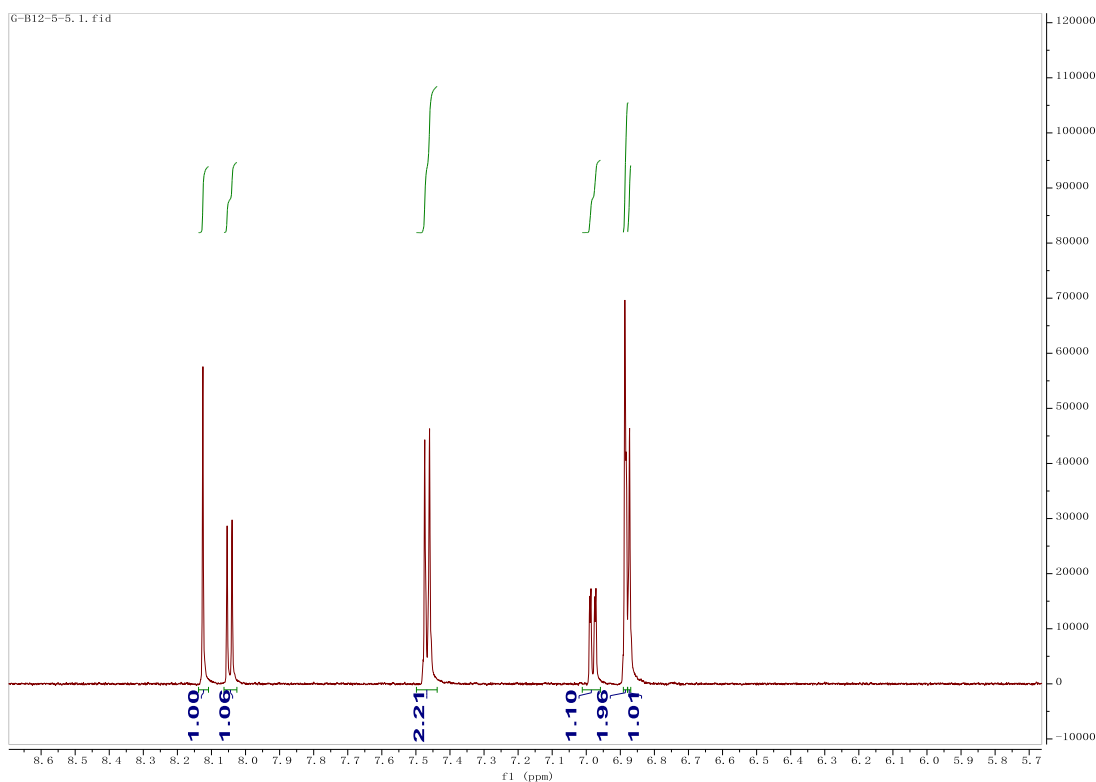


Figure S35.  $^1\text{H}$  NMR spectrum (600 MHz, acetone- $d_6$ ) of **11**

$^1\text{H}$  NMR (600 MHz, acetone- $d_6$ )  $\delta$  8.13 (s, 1H), 8.05 (d,  $J = 8.7$  Hz, 1H), 7.47 (d,  $J = 8.5$  Hz, 2H), 6.98 (dd,  $J = 8.8, 2.3$  Hz, 1H), 6.88 (d,  $J = 2.6$  Hz, 2H), 6.87 (d,  $J = 2.3$  Hz, 1H). Compound **11** was determined by comparison of its spectroscopic data with daidzein in the literature [7].

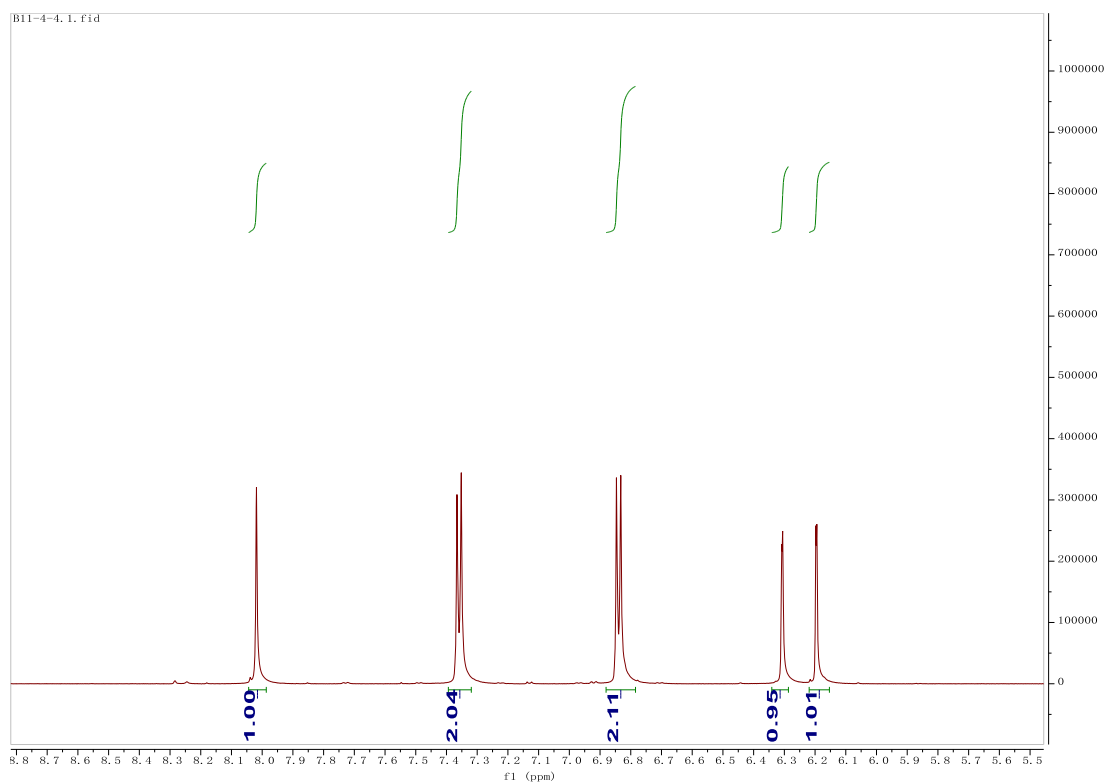


Figure S36.  $^1\text{H}$  NMR spectrum (600 MHz, methanol- $d_4$ ) of **12**

$^1\text{H}$  NMR (600 MHz, methanol- $d_4$ )  $\delta$  8.02 (s, 1H), 7.36 (d,  $J = 8.3$  Hz, 2H), 6.84 (d,  $J = 8.3$  Hz, 2H), 6.31 (d,  $J = 2.1$  Hz, 1H), 6.20 (d,  $J = 2.1$  Hz, 1H). Compound **12** was determined by comparison of its spectroscopic data with genistein in the literature [8].

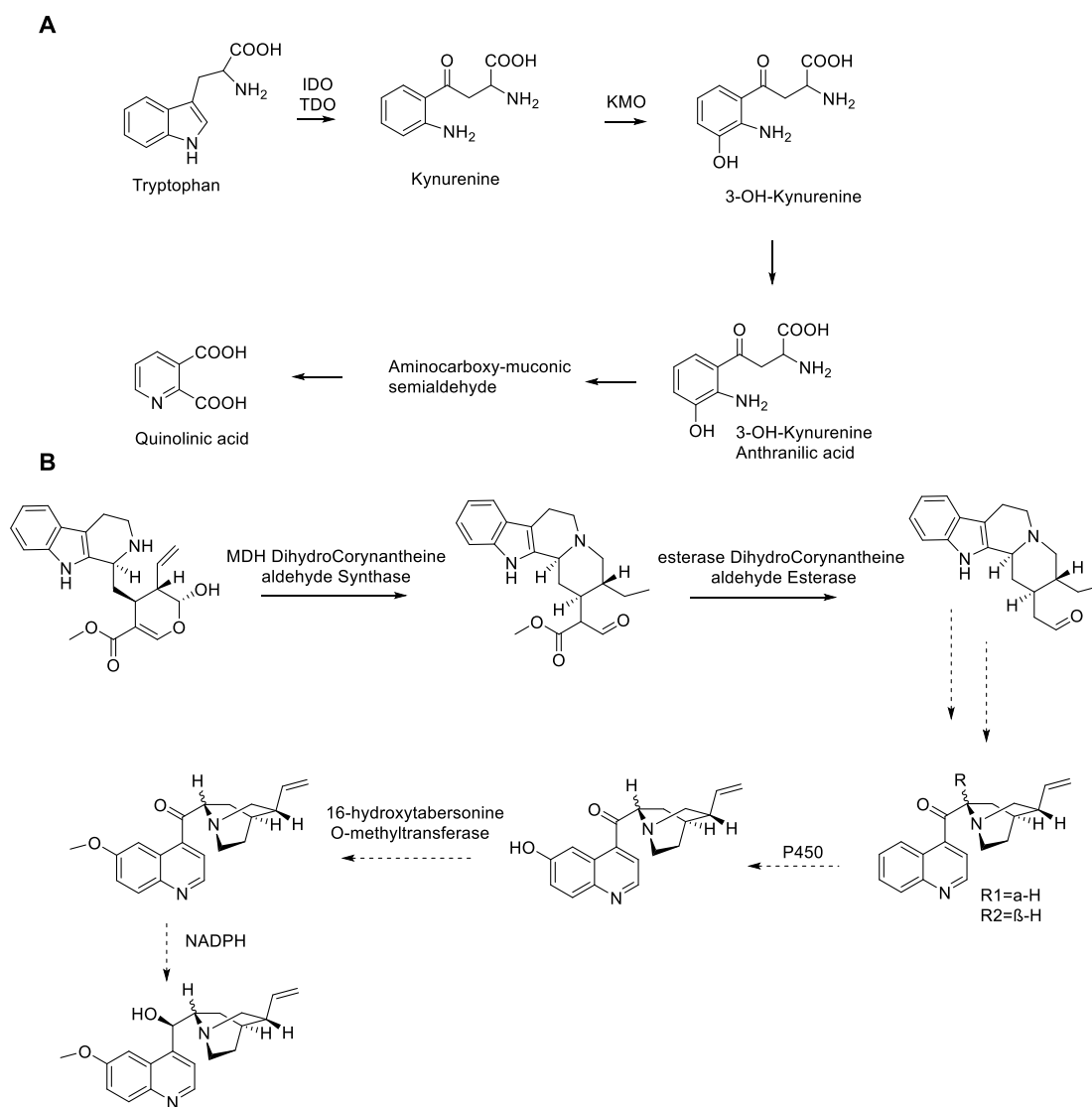


Figure S37. The biosynthetic pathways of the quinoline ring formation

**A:** The formation of primary metabolic pathways of quinoline rings [9] **B:** The formation of secondary metabolic pathways reported on quinoline rings[10]

## Supplementary References

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