

Supporting Information

to

Ring opening metathesis polymerization-derived block copolymers bearing chelating ligands: synthesis, metal immobilization and use in hydroformylation under micellar conditions

Gajanan M. Pawar¹, Jochen Weckesser², Siegfried Blechert^{*2} and Michael R.
Buchmeiser^{*1}

¹Lehrstuhl für Makromolekulare Stoffe und Faserchemie, Institut für Polymerchemie,
Universität Stuttgart, Pfaffenwaldring 55, D-70550 Stuttgart, Germany and ²Institut für
Chemie, Technische Universität Berlin, Straße des 17. Juni 135, D-10623 Berlin,
Germany

Email: Michael R. Buchmeiser - michael.buchmeiser@ipoc.uni-stuttgart.de; Tel.: +49
(0)711-685-64075; Fax: +49 (0)711-685-64050;
Siegfried Blechert - blechert@chemie.tu-berlin.de

*Corresponding author

Graphical representations of measurements

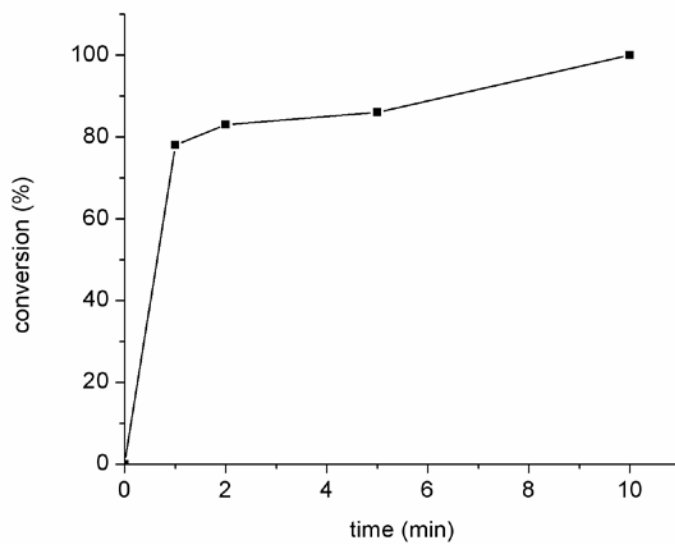


Figure S1: Polymerization kinetics of **M1** by the action of $\text{Mo}(N\text{-}2,6\text{-Me}_2\text{C}_6\text{H}_3)(\text{CHCMe}_2\text{Ph})(\text{OCMe}(\text{CF}_3)_2)_2$.

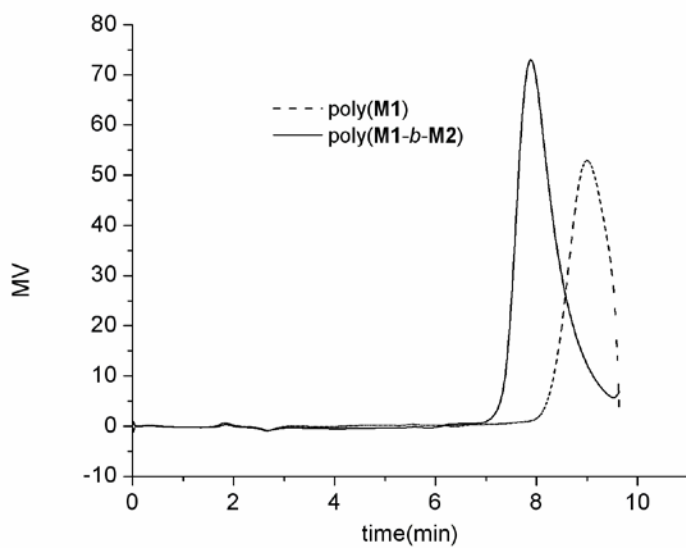


Figure S2: GPC-traces (DMF) of poly(**M1**) and poly(**M1-b-M2**) prepared there from.

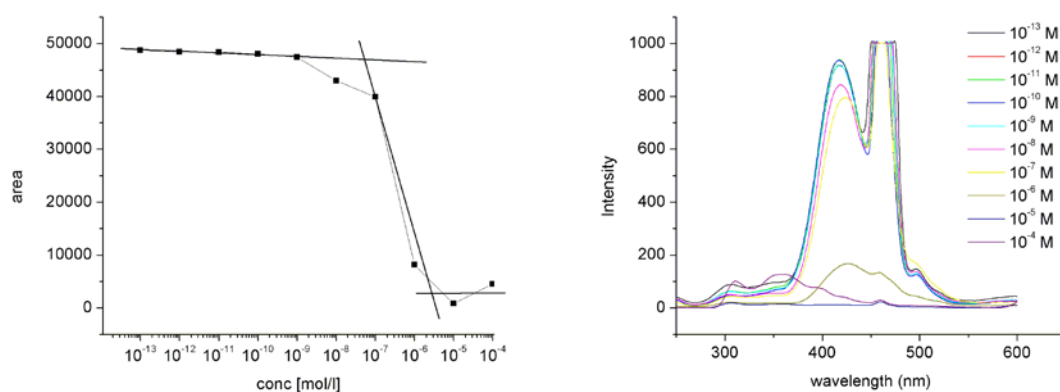


Figure S3: cmc measurements for poly(M1-*b*-M2) in water.

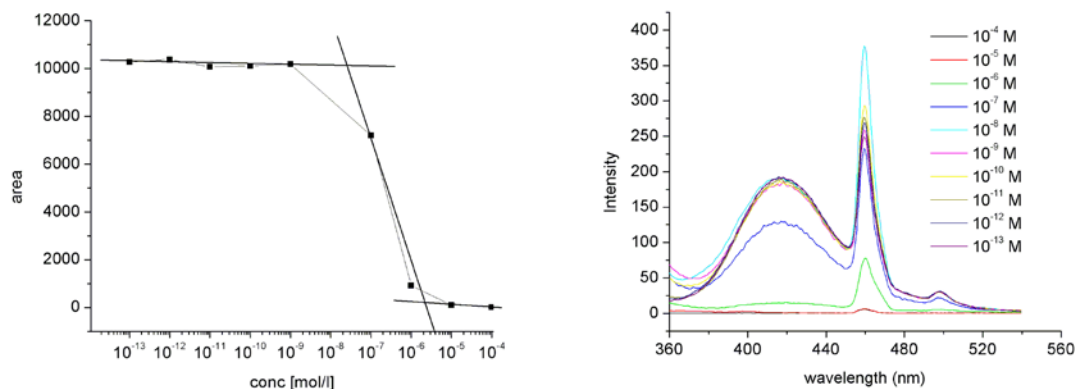


Figure S4: cmc measurements for poly(M1-*b*-M2)-Rh in water.

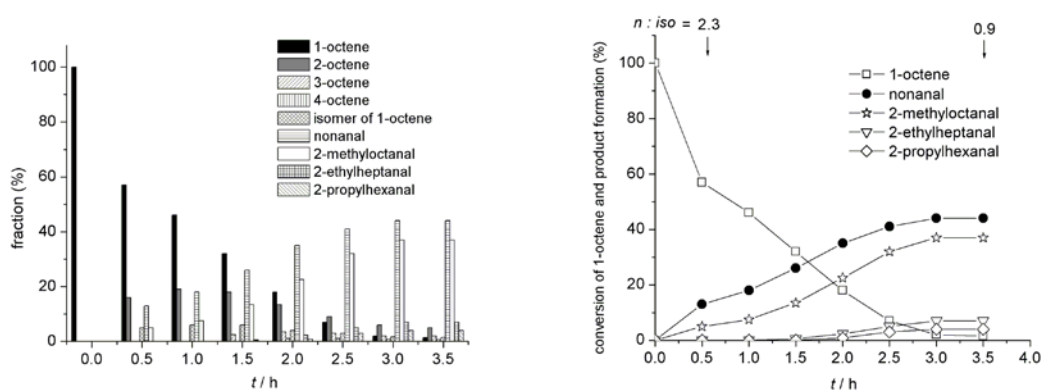


Figure S5: Conversion of 1-octene, product formation, product distribution in the hydroformylation of 1-octene in toluene in the presence of C1[2].

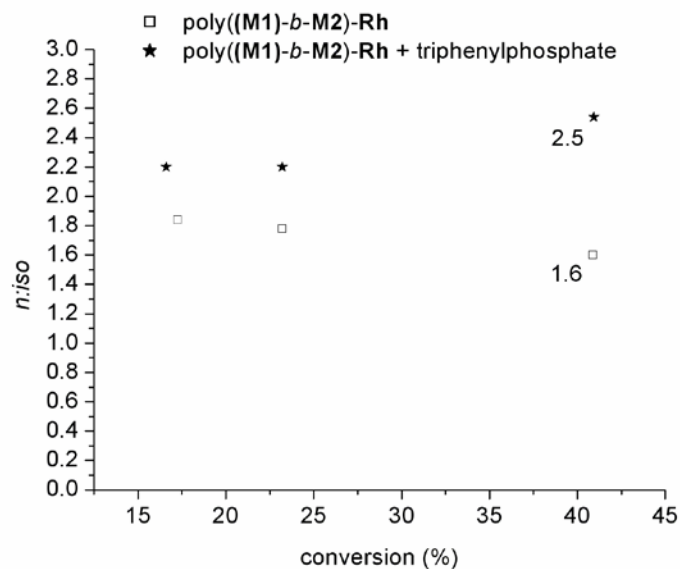


Figure S6: $n:iso$ Selectivities for catalysts poly(**M1**-*b*-**M2**)-Rh (□) and poly(**M1**-*b*-**M2**)-Rh (★) with triphenylphosphite as a function of conversion.

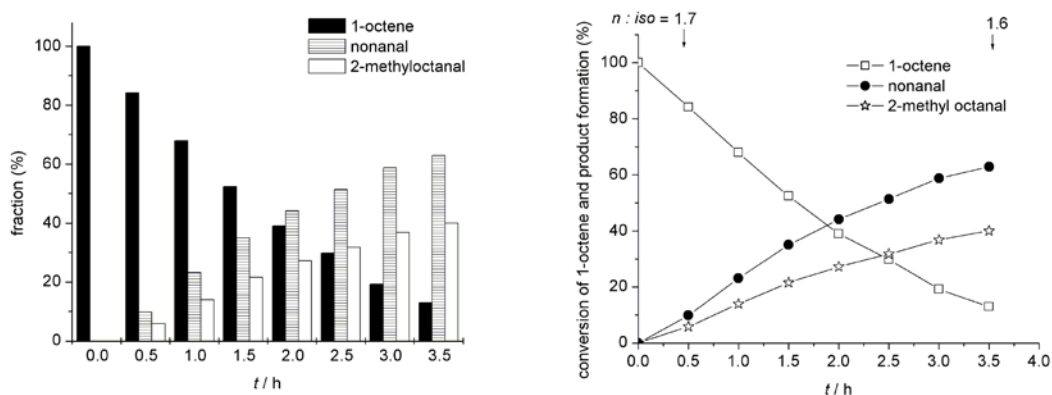


Figure S7: Conversion of 1-octene, product formation, product distribution in the hydroformylation of 1-octene in toluene in the presence of **C1** and triphenylphosphite.