

Supporting Information
for
The preferred conformation of
***erythro-* and *threo*-1,2-difluorocyclododecanes**

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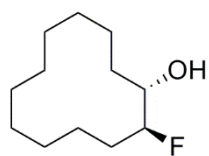
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Additional data

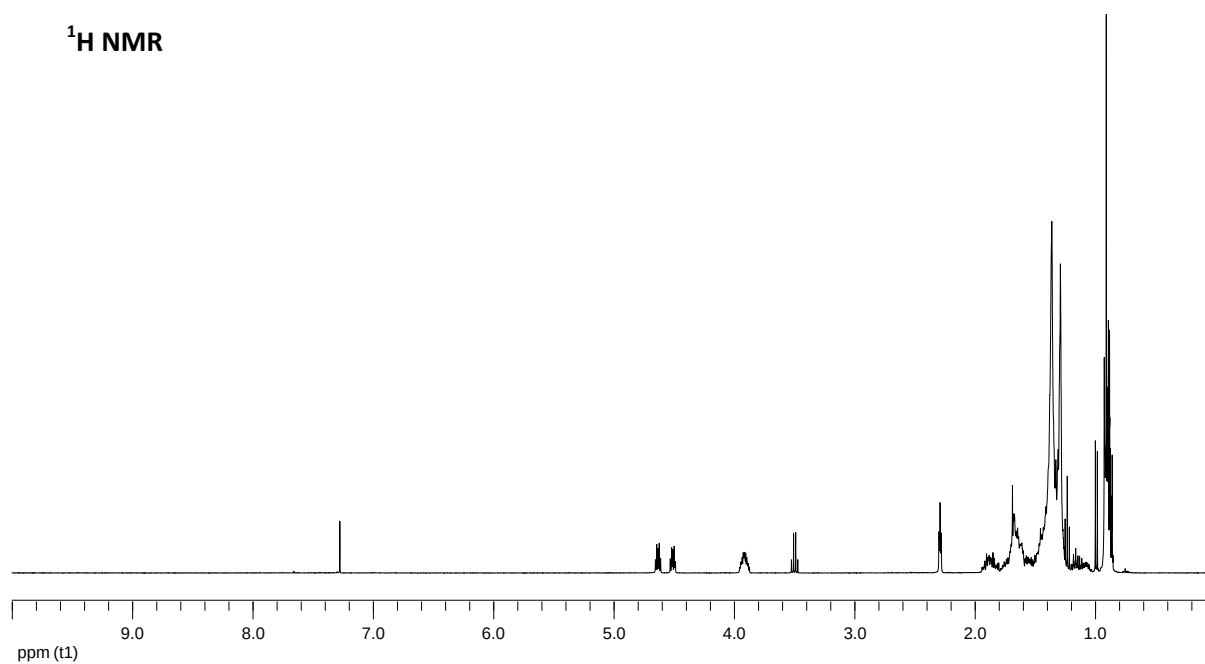
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Section 1 NMR spectra

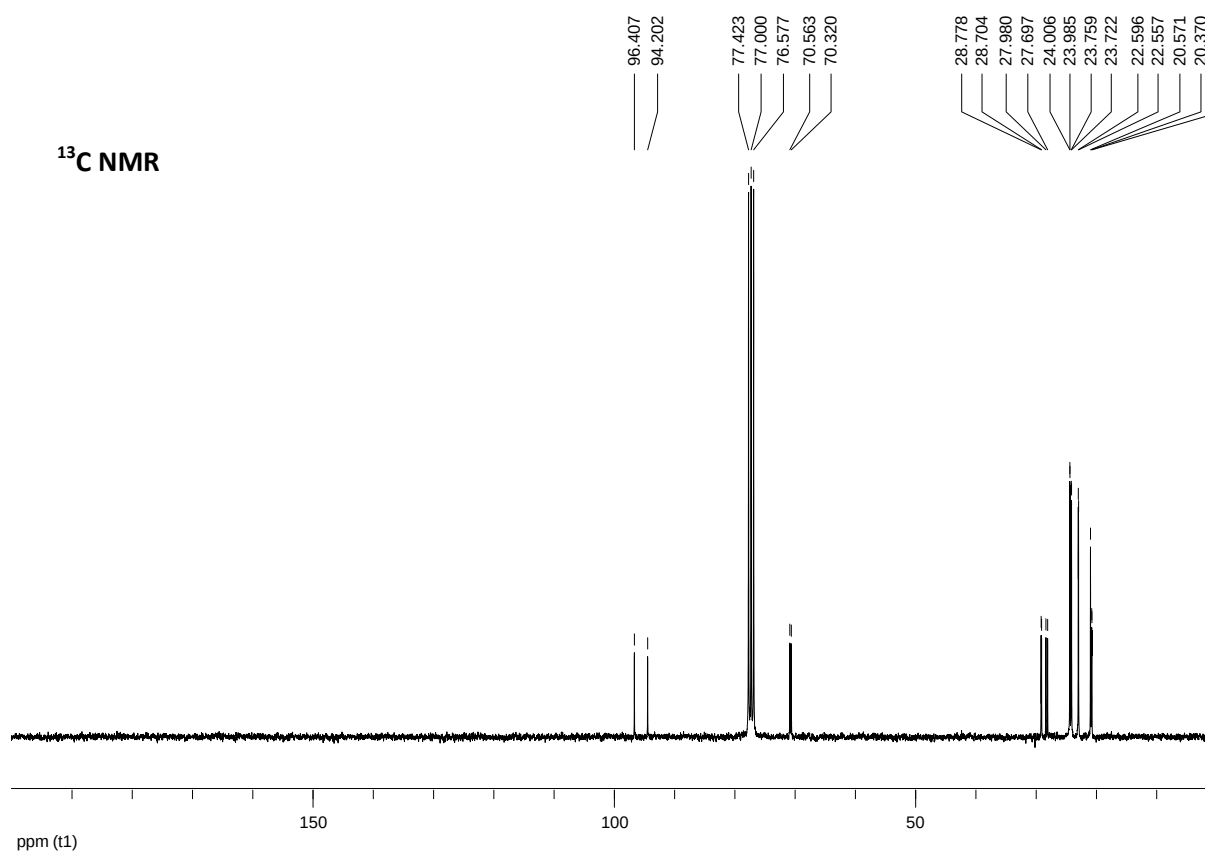


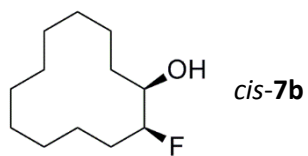
trans-7a

^1H NMR

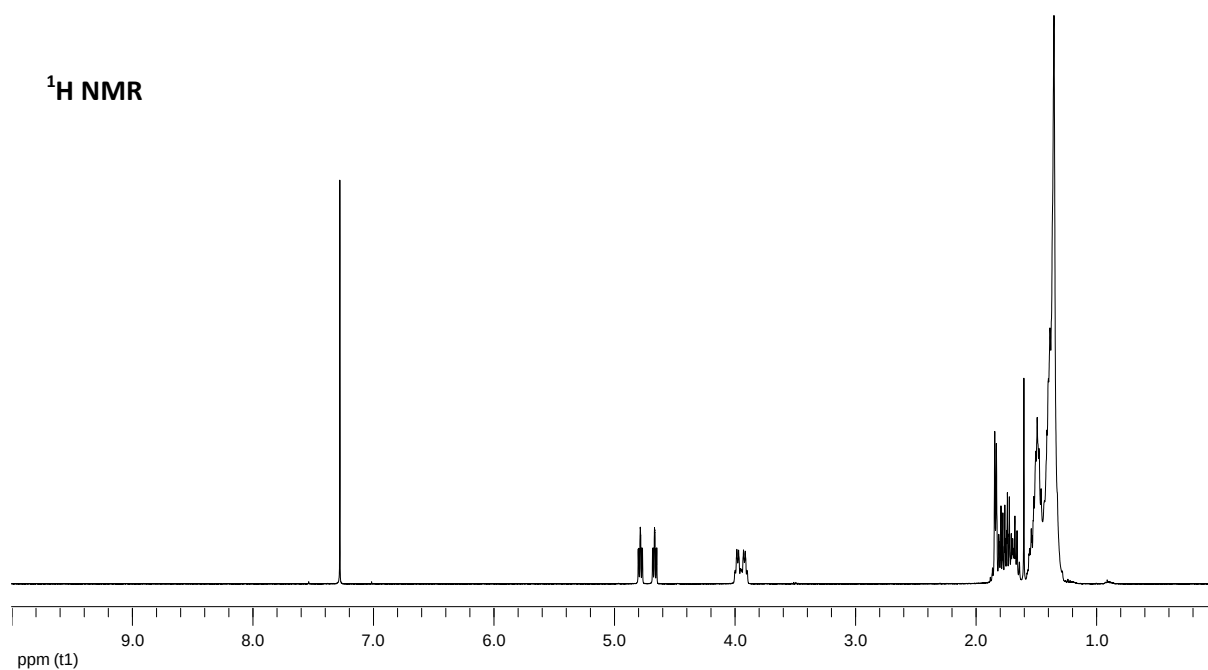


^{13}C NMR

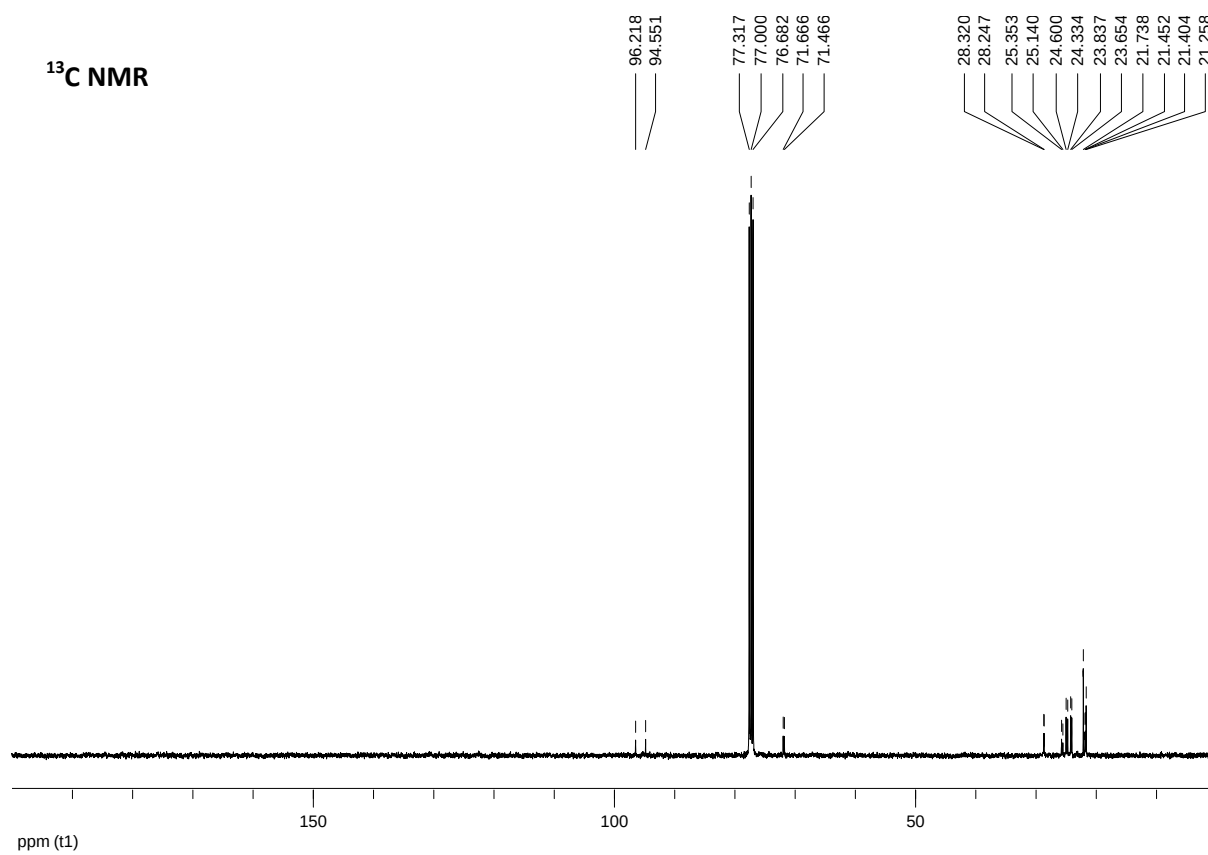


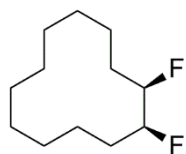


¹H NMR



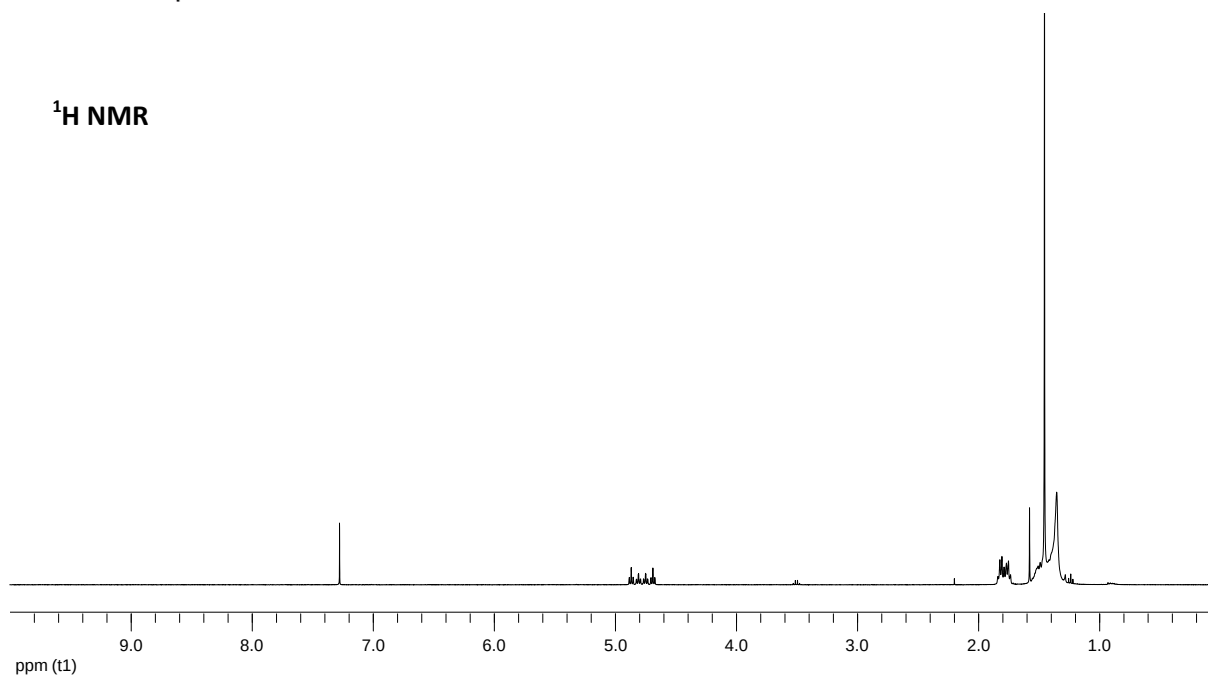
¹³C NMR



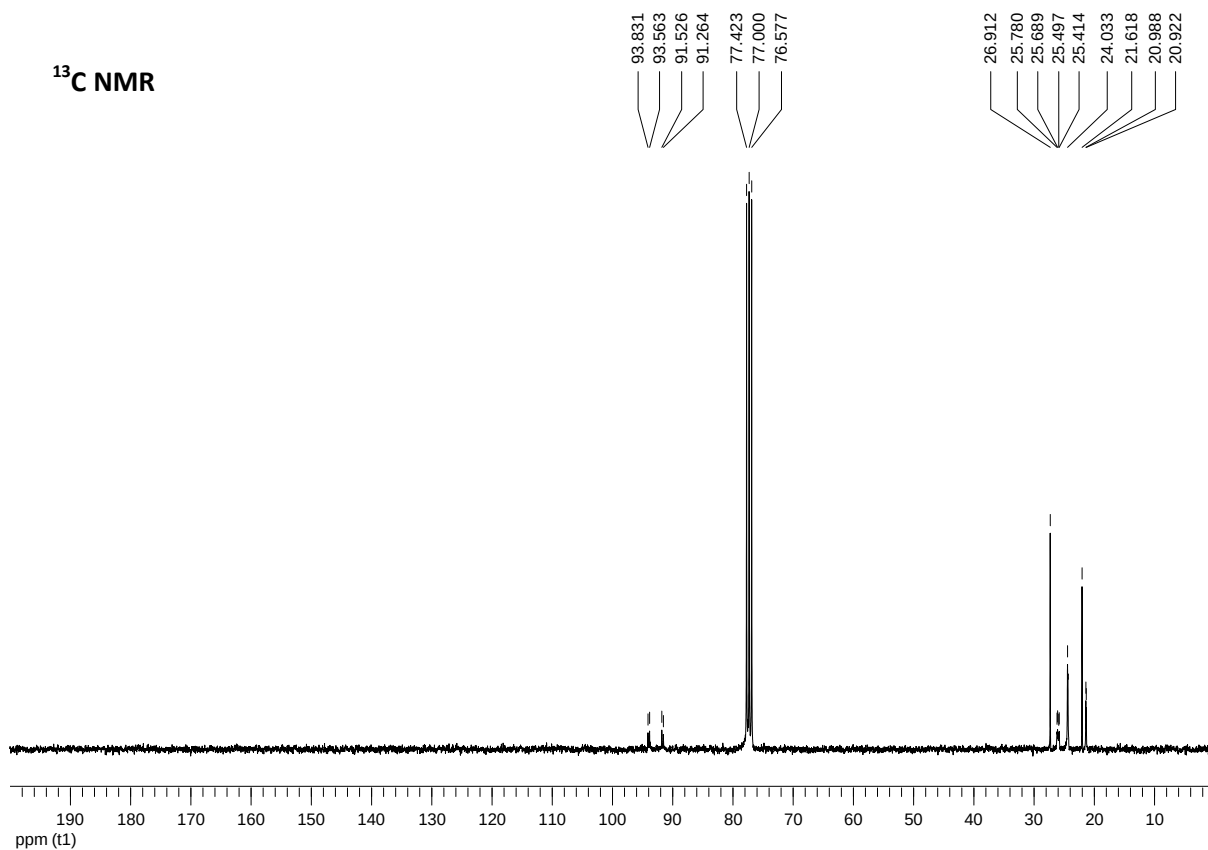


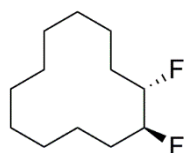
Erythro-5a

^1H NMR



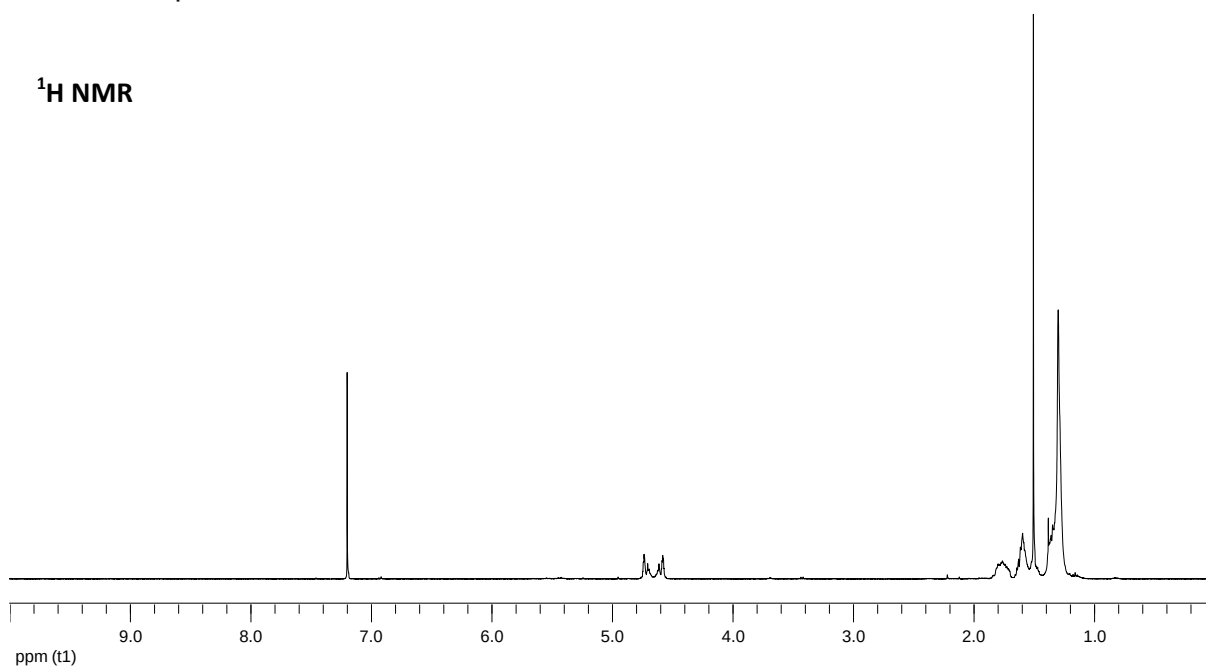
^{13}C NMR



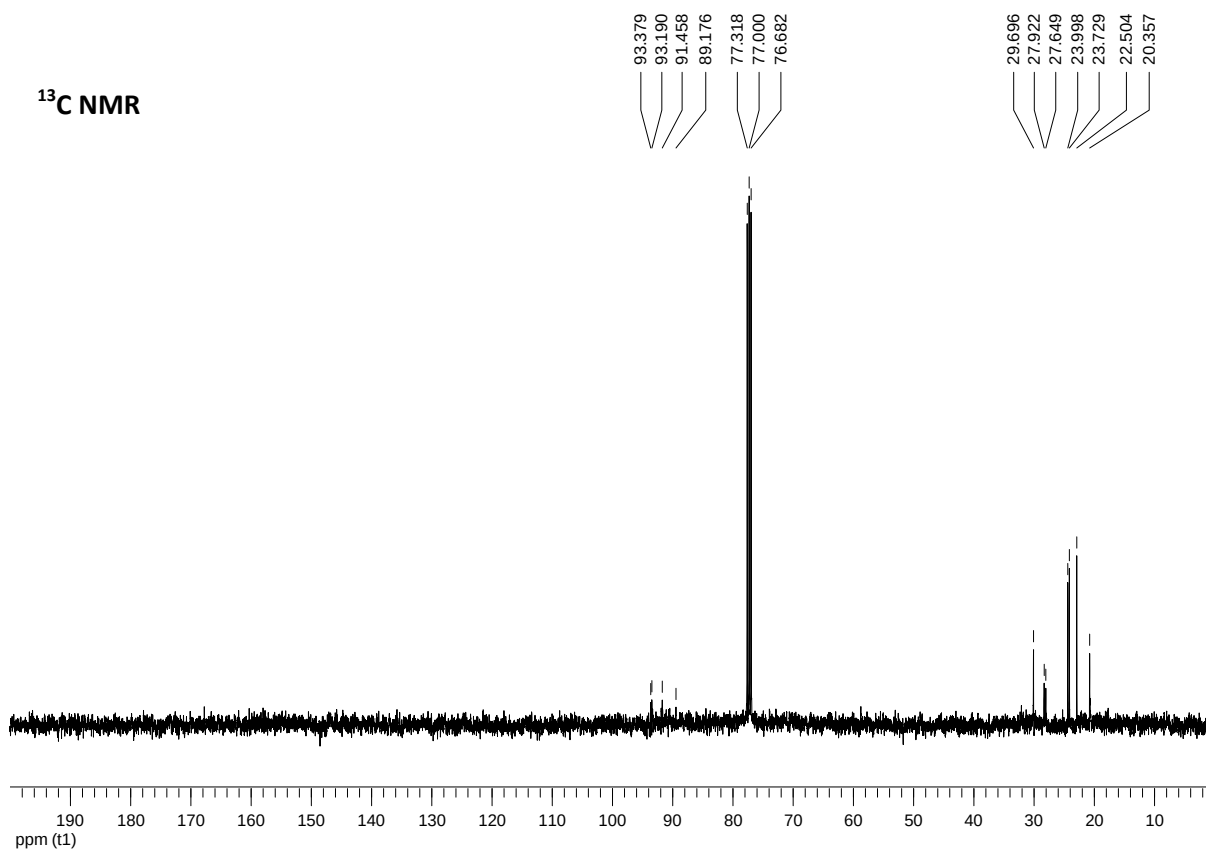


Threo-5b

¹H NMR

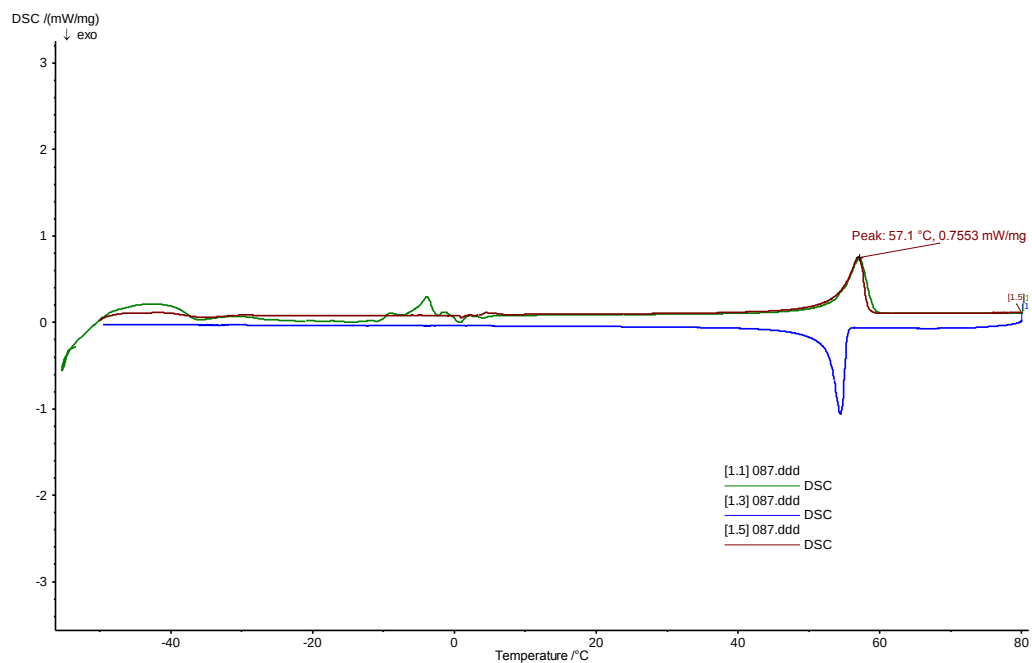


¹³C NMR

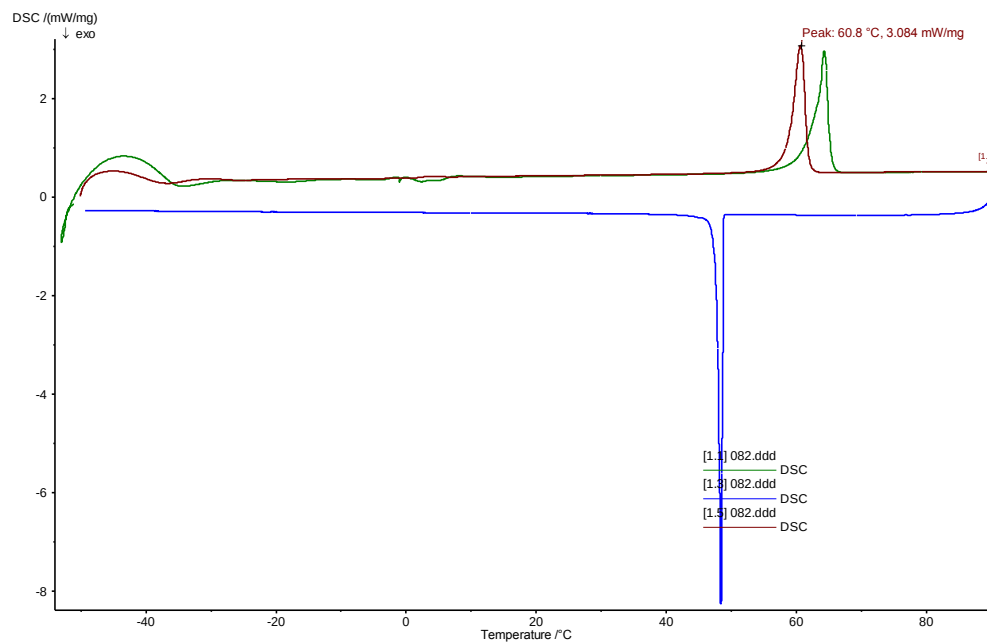


Section 2 Differential scanning calorimetry (DSC)

Measured on Netzsch DSC 204 Instrument with temperature range of –150 to 400 °C



DSC of *erythro* cyclododecane 5a



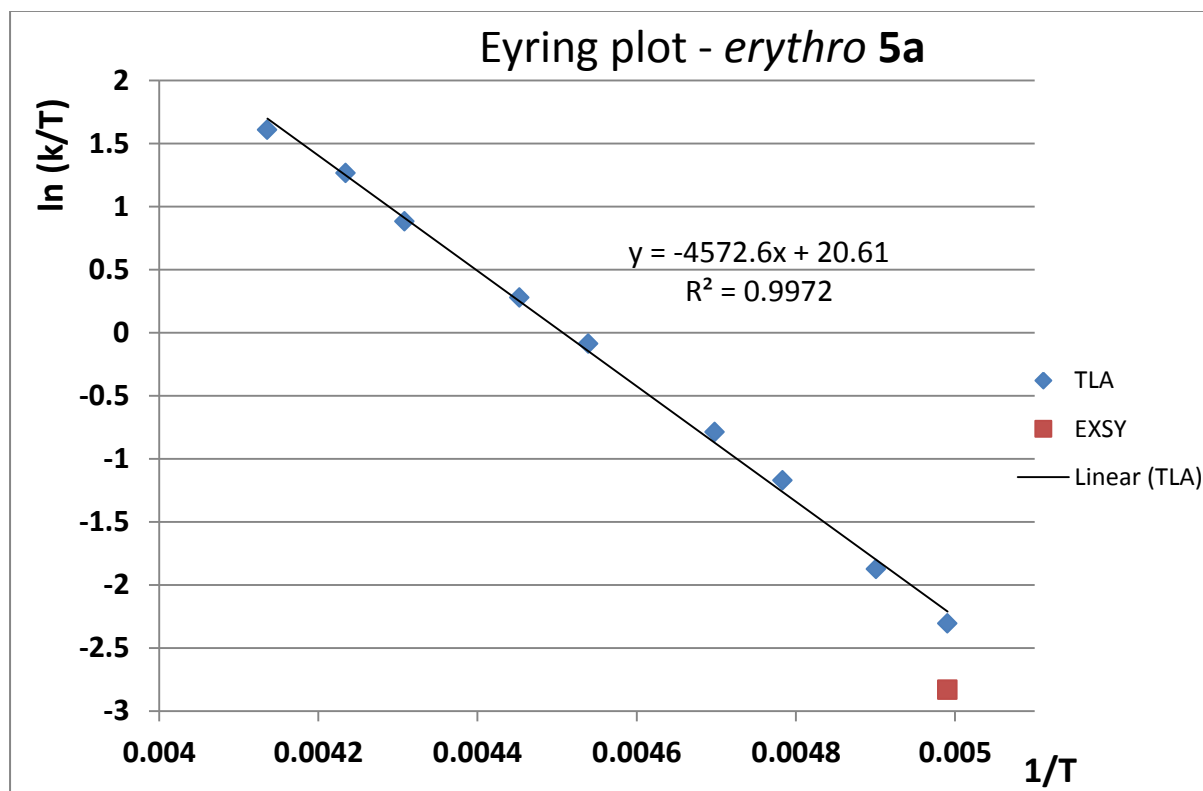
DSC of *threo*-cyclododecane 5b

Section 3 Variable-temperature NMR

Variable-temperature ^{19}F NMR experiments of *erythro*-fluorocyclododecane 5a.

Exp No	Chemical shifts			
	T [K]	k [Hz]	$1/T$	$\ln(k/T)$
1	200.38	11.8	0.004991	-2.83212
2	200.38	20.0	0.004991	-2.30448
3	204.05	31.3	0.004901	-1.87475
4	209.06	64.9	0.004783	-1.16977
5	212.85	96.8	0.004698	-0.78828
6	220.30	201.8	0.004539	-0.08776
7	224.59	296.8	0.004453	0.278782
8	232.11	561.7	0.004308	0.883751
9	236.16	837.7	0.004234	1.26618
10	241.79	1208.5	0.004136	1.609024

Fitting experimental data to the Eyring equation provides activation parameters of the ring interconversion process $\Delta G_{298}^{\ddagger} = 44.1 \pm 6.1 \text{ kJ}\cdot\text{mol}^{-1}$ ($10.5 \pm 1.5 \text{ kcal}\cdot\text{mol}^{-1}$), $\Delta G_{203}^{\ddagger} = 48.1 \text{ kJ}\cdot\text{mol}^{-1}$ ($11.4 \text{ kcal}\cdot\text{mol}^{-1}$), $\Delta H^{\ddagger} = 56.6 \pm 2.7 \text{ kJ}\cdot\text{mol}^{-1}$ ($13.5 \pm 0.6 \text{ kcal}\cdot\text{mol}^{-1}$), and $\Delta S^{\ddagger} = 42.0 \pm 11.3 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$ ($10.0 \pm 2.7 \text{ kcal}\cdot\text{mol}^{-1}$).

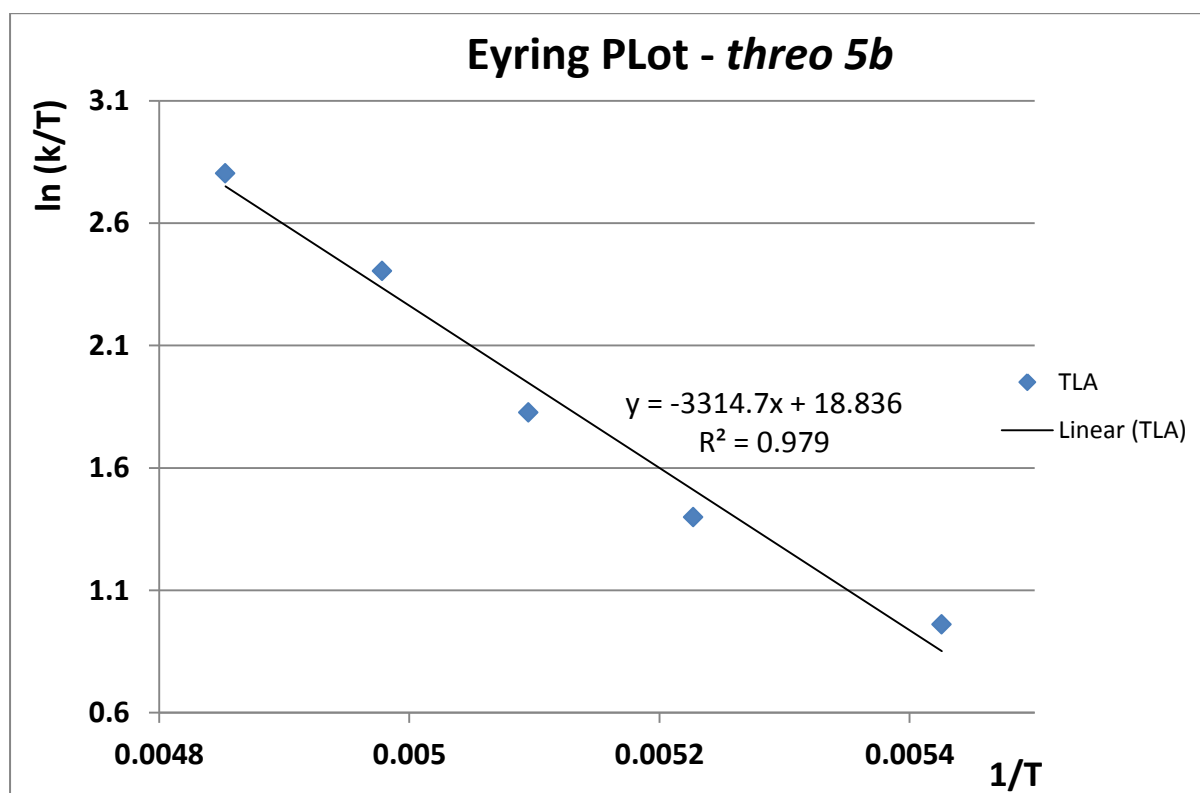


Variable-temperature ^{19}F NMR experiments of *threo*-difluorocyclododecane **5b**

The ^{19}F spectra in CD_2Cl_2 were recorded by using a Bruker AVANCE 500 MHz spectrometer equipped with 5 mm QNP-probe across the temperature range 193–219 K. The accurate temperatures for particular experiments were determined by using 4% MeOH in methanol- d_4 sample. Complete lineshape analysis was carried out by using a Bruker Topspin D-NMR module.

Exp No	Chemical shifts		T [K]	k [Hz]	$1/T$	$\ln(k/T)$
	A	B				
1	-192.6185	-194.1307	184.31	481.5	0.005426	0.960266
2	-192.5417	-194.1795	191.31	774.7	0.005227	1.398568
3	-192.4557	-194.2260	196.26	1219.2	0.005095	1.826534
4	-192.4435	-194.3793	200.87	2224.2	0.004978	2.404504
5	-192.5098	-194.4943	206.06	3400.6	0.004853	2.80354

Fitting experimental data to the Eyring equation provides activation parameters of the ring interconversion process $\Delta G^\ddagger_{298} = 39.8 \pm 4.8 \text{ kJ}\cdot\text{mol}^{-1}$ ($9.47 \pm 1.1 \text{ kcal}\cdot\text{mol}^{-1}$), $\Delta G^\ddagger_{203} = 35.9 \text{ kJ}\cdot\text{mol}^{-1}$ ($8.55 \text{ kcal}\cdot\text{mol}^{-1}$), $\Delta H^\ddagger = 27.6 \pm 1.2 \text{ kJ}\cdot\text{mol}^{-1}$ ($6.57 \pm 0.3 \text{ kcal}\cdot\text{mol}^{-1}$), and $\Delta S^\ddagger = -40.9 \pm 12.0 \text{ J}\cdot\text{K}^{-1}\cdot\text{mol}^{-1}$ ($-9.73 \pm 2.9 \text{ kcal}\cdot\text{mol}^{-1}$).



Section 4 Computational studies for each conformer of 1,2-difluorocyclododecane

Geometry 1	cC12-corner-aa B3LYP/6-311+G(2d,p)	cC12-corner-ae B3LYP/6-311+G(2d,p)	cC12-corner-ea B3LYP/6-311+G(2d,p)	cC12-corner-ee B3LYP/6-311+G(2d,p)	cC12-edge-aa B3LYP/6-311+G(2d,p)	cC12-edge-ee B3LYP/6-311+G(2d,p)	cC12-edge-ae B3LYP/6-311+G(2d,p)
E (a.u.)	-670.42504150	-670.43041090	-670.42599480	-670.42881540	-670.42055320	-670.42798090	-670.42658030
E (kcal/mol)	-420691.71354125	-420695.08283975	-420692.31173700	-420694.08166350	-420688.89713300	-420693.55801475	-420692.67913825
ZPE (kcal/mol)	204.76313000	204.46113000	204.76329000	204.46151000	204.74204000	204.70837000	204.63087000
E+ZPE (kcal/mol)	-420486.95041125	-420490.62170975	-420487.54844700	-420489.62015350	-420484.15509300	-420488.84964475	-420488.04826825
rel E (kcal/mol)	3.67	0.00	3.07	1.00	6.47	1.77	2.57

Geometry 1	cC12-corner-aa MP2/6-311+G(2d,p)	cC12-corner-ae MP2/6-311+G(2d,p)	cC12-corner-ea MP2/6-311+G(2d,p)	cC12-corner-ee MP2/6-311+G(2d,p)	cC12-edge-aa MP2/6-311+G(2d,p)	cC12-edge-ee MP2/6-311+G(2d,p)	cC12-corner-ae MP2/6-311+G(2d,p)
E (a.u.)	-668.62448390	-668.62992400	-668.62565740	-668.62858320	-668.61977200	-668.62786340	-668.62571840
E (kcal/mol)	-419561.86364725	-419565.27731000	-419562.60001850	-419564.43595800	-419558.90693000	-419563.98428350	-419562.63829600
ZPE (kcal/mol)	204.76313000	204.46113000	204.76329000	204.46151000	204.74204000	204.70837000	204.63087000
E+ZPE (kcal/mol)	-419357.10051725	-419360.81618000	-419357.83672850	-419359.97444800	-419354.16489000	-419359.27591350	-419358.00742600
rel E (kcal/mol)	3.72	0.00	2.98	0.84	6.65	1.54	2.81

Geometry 1	cC12-corner-aa M06-2X/6-311+G(2d,p)	cC12-corner-ae M06-2X/6-311+G(2d,p)	cC12-corner-ea M06-2X/6-311+G(2d,p)	cC12-corner-ee M06-2X/6-311+G(2d,p)	cC12-edge-aa M06-2X/6-311+G(2d,p)	cC12-edge-ee M06-2X/6-311+G(2d,p)	cC12-edge-ae M06-2X/6-311+G(2d,p)
E (a.u.)	-670.130765	-670.1349674	-670.1319447	-670.1342023	-670.1275112	-670.1333177	-670.132035
E (kcal/mol)	-420507.055	-420509.692	-420507.7953	-420509.2119	-420505.0133	-420508.6569	-420507.852
ZPE (kcal/mol)	206.97359	206.67512	207.02632	206.90772	207.18351	206.64104	207.00078
E+ZPE (kcal/mol)	-420300.0814	-420303.0169	-420300.769	-420302.3042	-420297.8298	-420302.0158	-420300.8512
rel E (kcal/mol)	2.94	0.00	2.25	0.71	5.19	1.00	2.17