

Supporting Information

for

Azobenzene dye-coupled quadruply hydrogen-bonding modules as colorimetric indicators for supramolecular interactions

Yagang Zhang and Steven C. Zimmerman*

Address: Department of Chemistry, 600 South Mathews Avenue, University of Illinois, Urbana, IL 61801, USA

Email: Steven C. Zimmerman* - sczimmer@illinois.edu

*Corresponding author

General experimental procedures, detailed synthetic procedures and characterization data

Table of contents

Experimental section

Additional general methods	S2
Synthetic procedures	S3
¹ H and ¹³ C NMR spectra	
Compound 1	S10
Compound 2	S11
Compound 3	S12
Compound 5	S13
Compound 6	S14
Compound 8	S15
Compound 10	S16

Compound 12	S17
Compound 13	S18
Compound 14	S19
Compound 15	S20
Compound 18	S21
Elemental composition reports of HR-ESI-MS	S22
References	S28

Experimental section

Additional general methods

Analytical thin-layer chromatography (TLC) was performed by using 0.2 mm silica 60 coated, plastic plates with F₂₅₄ indicator purchased from EM Science. Column chromatography was performed by using Ultra Pure SiliaFlash® P60 230–400 mesh (40–63 µm) silica gel (SiO₂) with fluorescence indicator green 254 nm (Fluka) following the general procedure reported by Still and co-workers [1]. Analytical size-exclusion chromatography (SEC) was performed in THF with a flow-rate of 1 mL min⁻¹ (30 °C) using Waters auto sampler, pump, photodiode array and refractive-index detector outfitted with Waters Styragel columns (HR5, HR4, HR3, HR1). Molecular weight was determined using conventional calibration against polystyrene standards. ¹H NMR spectra were acquired on a Varian Unity 500 MHz (¹³C, 126 MHz) spectrometer. ¹H NMR chemical shifts (δ) are reported in parts per million (ppm) and are referenced to the residual solvent peak at 7.26 ppm for CDCl₃, 2.50 ppm for DMSO-*d*₆ and 3.31 ppm for CD₃OD. ¹H NMR coupling constants (*J*) are reported in hertz (Hz). Several of the aromatic protons in the azobenzene units are listed as doublets although additional coupling and broadening was seen in most of the aromatic resonances. ¹³C NMR chemical shifts are reported in ppm and were referenced to the residual solvent peak at 77.16 ppm for CDCl₃, 39.52 ppm for DMSO-*d*₆ and 49.00 ppm for CD₃OD. All NMR spectra are original spectra obtained from the NMR instrument and then scanned. Mass spectra were obtained on Micromass Q-Tof Ultima (HR-ESI) and Micromass Quattro (LR-ESI) instruments. Melting points were determined by using a Thomas-Hoover melting point apparatus and are uncorrected.

Synthetic procedures

***tert*-Butyl 4-[(4-hydroxyphenyl)diazenyl]benzoate (1)**

To a stirred suspension of 4-aminobenzoic acid *tert*-butyl ester (2.40 g, 12.0 mmol) in 30 mL of a 2 M aqueous hydrochloric acid solution cooled in a 0 °C ice bath, was added dropwise 5 mL of an aqueous solution of NaNO₂ (900 mg, 13.0 mmol). The solution was stirred for 8 min and diluted with ice-cold MeOH (50 mL). The resultant yellow solution was added slowly to a solution of phenol (1.17 g, 12.4 mmol) and NaOH (960 mg, 24.0 mmol) in 14 mL MeOH cooled to 0 °C in an ice bath. The solution was stirred for 1 h and a concentrated aqueous solution of HCl was added until pH 1 to 2 was attained. The precipitate was filtered off and washed with an excess of water. The crude product was crystallized from EtOH–H₂O (1:10, v/v) to afford 2.23 g (60%) of **1** as an orange solid: mp 144-145 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.12 (d, *J* = 8.5, 2H), 7.92 (d, *J* = 9.0, 2H), 7.88 (d, *J* = 8.5, 2H), 6.97 (d, *J* = 9.0, 2H), 5.41 (s, 1H), 1.62 (s, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 165.5, 158.9, 155.2, 147.4, 133.4, 130.6, 125.5, 122.4, 116.0, 81.6, 28.4, 28.3; ESI-HR-MS calcd for (C₁₇H₁₈N₂O₃+H)⁺ 299.1396, found 299.1395.

***tert*-Butyl 4-[[4-(4-ethoxy-4-oxobutoxy)phenyl]diazenyl]benzoate (2)**

A mixture of **1** (1.00 g, 3.35 mmol), ethyl 4-bromobutyrate (2.75 g, 13.4 mmol), potassium carbonate (1.39 g, 10.1 mmol), and potassium iodide (500 mg, 0.301 mmol) in 50 mL of THF was stirred and heated under reflux for 16 h. The mixture was cooled, concentrated in vacuo, and dissolved in ca. 35 mL of CH₂Cl₂. The solution was washed with 30 mL of a 2.0 M aqueous solution of HCl and 30 mL of brine. The organic layer was concentrated in vacuo and the residue purified by silica gel column chromatography with MeOH–CH₂Cl₂ (1:99 v/v, *R*_f = 0.72) to afford 995 mg (72%) of **2** as an orange solid: mp 108-109 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.11 (d, *J* = 8.5, 2H), 7.93 (d, *J* = 9.0, 2H), 7.88 (d, *J* = 8.5, 2H), 7.00 (d, *J* = 9.0, 2H), 4.16 (q, *J* = 7.0, 2H), 4.10 (t, *J* = 6.0, 2H), 2.54 (t, *J* = 7.0, 2H), 2.14-2.18 (m, 2H), 1.62 (s, 9H), 1.26 (t, *J* = 7.0, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 173.2, 165.4, 162.0, 155.2, 147.2, 133.3, 130.5, 125.3, 122.3, 115.0, 114.8, 81.5, 67.3, 60.7, 30.8, 28.5, 28.4, 28.3, 24.7, 14.4, 14.3; ESI-HR-MS calcd for (C₂₃H₂₈N₂O₅+H)⁺ 413.2076, found 413.2079.

4-[[4-(4-Ethoxy-4-oxobutoxy)phenyl]diazenyl]benzoic acid (3)

Trifluoroacetic acid (4 mL, 52 mmol) was added to **2** (600 mg, 1.45 mmol). The mixture was stirred for 20 min. The excess trifluoroacetic acid was removed in vacuo and 2 mL acetone was

added. The solid was filtered off and was purified by silica gel column chromatography with MeOH-CH₂Cl₂ (5:95 v/v, R_f = 0.25) to afford 440 mg (85%) of **3** as an orange solid: mp 211-212 °C; ¹H NMR (500 MHz, DMSO-*d*₆) δ 13.20 (bs, 1H), 8.10 (d, *J* = 9.0, 2H), 7.91 (d, *J* = 9.5, 2H), 7.89 (d, *J* = 8.5, 2H), 7.13 (d, *J* = 9.0, 2H), 4.11 (t, *J* = 6.5, 2H), 4.07 (q, *J* = 7.0, 2H), 2.51 (t, *J* = 2.0, 2H), 1.99-2.02 (m, 2H), 1.18 (t, *J* = 7.0, 3H); ¹³C NMR (126 MHz, DMSO-*d*₆) δ 172.5, 161.8, 154.4, 146.2, 130.6, 125.0, 122.3, 115.2, 67.2, 60.0, 30.1, 24.1, 14.1; ESI-HR-MS calcd for (C₁₉H₂₀N₂O₅+H)⁺ 357.1450, found 357.1447.

Ethyl 4-{4-[4-(7-heptanamido-1,8-naphthyridin-2-ylcarbamoyl)phenyl]diazanyl}phenoxy)-butanoate (5**)**

To a stirred solution of **4** (38 mg, 0.140 mmol), DMAP (5 mg, 0.035 mmol), and **3** (100 mg, 0.280 mmol) in 12 mL of CH₂Cl₂ cooled to 0 °C in an ice bath, was added EDC (54 mg, 0.280 mmol), and the mixture was warmed to ambient temperature over several hours and stirred for 24 h. The solution was diluted with 20 mL of CH₂Cl₂ and was washed with water (3 × 20 mL), a saturated aqueous solution of sodium bicarbonate (2 × 20 mL), and brine (20 mL). The organic layer was dried over sodium sulfate, concentrated in vacuo and purified by silica gel column chromatography with MeOH-CH₂Cl₂ (5:95 v/v, R_f = 0.65) to afford **5** (61 mg, 71%) as an orange solid. mp 228-229 °C; ¹H NMR (500 MHz, CDCl₃) δ 9.14 (s, 1H), 8.61 (s, 1H), 8.59 (d, *J* = 8.5, 1H), 8.46 (d, *J* = 9.0, 1H), 8.18 (d, *J* = 8.5, 1H), 8.14 (d, *J* = 9.0, 1H), 8.05 (d, *J* = 8.5, 2H), 7.93 (d, *J* = 9.0, 2H), 7.92 (d, *J* = 8.5, 2H), 7.00 (d, *J* = 9.0, 2H), 4.16 (q, *J* = 7.5, 2H), 4.10 (t, *J* = 6.0, 2H), 2.54 (t, *J* = 7.0, 2H), 2.44 (t, *J* = 7.5, 2H), 2.13-2.18 (m, 2H), 1.68-1.74 (m, 2H), 1.34 (t, *J* = 7.0, 3H), 1.26-1.28 (m, 6H), 0.86 (t, *J* = 7.0, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 173.2, 172.6, 165.4, 162.2, 155.2, 154.2, 154.1, 153.8, 147.1, 139.3, 139.2, 134.8, 128.5, 125.4, 123.0, 118.6, 114.9, 113.8, 113.7, 67.2, 60.7, 38.1, 31.6, 30.8, 28.9, 25.3, 24.6, 22.6, 14.4, 14.1; ESI-HR-MS calcd for (C₃₄H₃₈N₆O₅+H)⁺ 611.2982, found 611.2984.

4-(4-{4-(*tert*-Butoxycarbonyl)phenyl}diazanyl}phenoxy)butanoic acid (6**)**

A suspension of **2** (0.148 g, 0.359 mmol) in 12 mL of a 1:1 (v/v) mixture of 10% aqueous LiOH and ethanol was stirred at room temperature for 1 h. The mixture was neutralized by dropwise addition of a 5% aqueous solution of HCl, concentrated in vacuo to remove ethanol, and extracted with CH₂Cl₂ (3 × 50 mL). The combined organic layers were washed with saturated aqueous sodium bicarbonate (20 mL), and brine (2 × 20 mL). The organic layer was dried over

sodium sulfate, concentrated in vacuo and purified by silica gel column chromatography MeOH–CH₂Cl₂ (5:95 v/v, *R_f* = 0.33) to afford 121 mg (88%) of **6** as an orange solid: mp 202-203 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.11 (d, *J* = 8.0, 2H), 7.94 (d, *J* = 9.0, 2H), 7.88 (d, *J* = 8.5, 2H), 7.01 (d, *J* = 9.0, 2H), 4.13 (t, *J* = 6.5, 2H), 2.63 (t, *J* = 7.0, 2H), 2.15-2.21 (m, 2H), 1.62 (s, 9H); ¹³C NMR (126 MHz, CDCl₃) δ 178.7, 165.5, 161.8, 155.2, 147.2, 133.3, 130.5, 125.3, 122.4, 114.9, 81.5, 67.0, 30.5, 28.3, 24.4; ESI-HR-MS calcd for (C₂₁H₂₄N₂O₅+H)⁺ 385.1763, found 385.1765.

***tert*-Butyl 4-({4-[4-(7-isobutyramido-1,8-naphthyridin-2-ylamino)-4-oxobutoxy]phenyl}diazanyl)benzoate (8)**

To a stirred solution of **7** (30 mg, 0.130 mmol), DMAP (4 mg, 0.033 mmol), and **6** (50 mg, 0.130 mmol) in 10 mL of CH₂Cl₂ cooled to 0 °C in an ice bath, was added EDC (27 mg, 0.143 mmol), and the reaction mixture was warmed to ambient temperature over several hours and stirred for 24 h. The solution was diluted with 20 mL of CH₂Cl₂ and washed with water (2 × 20 mL), a saturated solution of aqueous sodium bicarbonate (20 mL), and brine (20 mL). The organic layer was dried over sodium sulfate, concentrated in vacuo and purified by silica gel column chromatography with MeOH–CH₂Cl₂ (1:9 v/v, *R_f* = 0.72) to afford 47 mg (60%) of **8** as an orange solid: mp 208-209 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.86 (bs, 1H), 8.51 (bs, 1H), 8.46 (d, *J* = 8.5, 1H), 8.43 (d, *J* = 9.0, 1H), 8.13 (d, *J* = 9.0, 1H), 8.12 (d, *J* = 8.0, 1H), 8.10 (d, *J* = 8.0, 2H), 7.90 (d, *J* = 8.5, 2H), 7.86 (d, *J* = 8.5, 2H), 6.97 (d, *J* = 8.5, 2H), 4.13 (t, *J* = 6.0, 2H), 2.71 (t, *J* = 7.0, 2H), 2.60-2.64 (m, 1H), 2.23-2.27 (m, 2H), 1.61 (s, 9H), 1.26 (d, *J* = 7.0, 6H); ¹³C NMR (126 MHz, CDCl₃) δ 176.2, 171.7, 165.4, 161.8, 155.1, 154.2, 154.0, 147.2, 139.3, 139.2, 133.3, 130.6, 130.5, 125.3, 122.3, 118.5, 114.9, 113.7, 113.6, 81.5, 67.2, 37.0, 34.1, 28.3, 24.6, 19.4; ESI-HR-MS calcd for (C₃₃H₃₆N₆O₅+H)⁺ 597.2825, found 597.2825.

***tert*-Butyl 4-[4-{4-[7-(2-ethylhexanamido)-1,8-naphthyridin-2-ylamino]-4-oxobutoxy}phenyl]diazanyl]benzoate (10)**

To a stirred solution of **9** (31 mg, 0.108 mmol), DMAP (4 mg, 0.033 mmol) and **6** (42 mg, 0.108 mmol) in 10 mL CH₂Cl₂ cooled to 0 °C in an ice bath, was added EDC (25 mg, 0.119 mmol) and the mixture was warmed to ambient temperature over several hours and stirred for 24 h. The solution was diluted with 20 mL CH₂Cl₂ and was washed with water (2 × 20 mL), a saturated aqueous solution of sodium bicarbonate (20 mL), and brine (20 mL). The organic layer was dried over sodium sulfate, concentrated in vacuo and purified by silica gel column

chromatography with MeOH–CH₂Cl₂ (5:95 v/v, *R_f* = 0.41) to afford 47 mg (62%) of **10** as an orange solid: mp 198-199 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.88 (bs, 1H), 8.50 (bs, 1H), 8.48 (d, *J* = 9.0, 1H), 8.43 (d, *J* = 9.0, 1H), 8.14 (d, *J* = 9.0, 1H), 8.13 (d, *J* = 8.5, 1H), 8.10 (d, *J* = 8.0, 2H), 7.90 (d, *J* = 9.0, 2H), 7.87 (d, *J* = 8.5, 2H), 6.97 (d, *J* = 9.0, 2H), 4.13 (t, *J* = 6.0, 2H), 2.71 (t, *J* = 7.0, 2H), 2.23-2.27 (m, 2H), 2.17-2.21 (m, 1H), 1.67-1.74 (m, 2H), 1.61 (s, 9H), 1.48-1.55 (m, 2H), 1.25-1.30 (m, 2H), 0.93 (t, *J* = 7.5, 3H), 0.84 (t, *J* = 7.0, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 175.6, 171.7, 165.4, 161.8, 155.1, 154.1, 153.8, 147.2, 139.3, 139.2, 133.3, 130.6, 125.3, 124.3, 122.3, 119.3, 118.5, 114.9, 114.3, 113.9, 113.6, 81.5, 67.2, 60.0, 34.1, 32.5, 29.9, 28.3, 26.1, 24.6, 22.8, 14.0, 12.1; ESI-HR-MS calcd for (C₃₇H₄₄N₆O₅+H)⁺ 653.3451, found 653.3453.

***tert*-Butyl 4-[(4-{2-[2-(3-butylureido)-7-(cyclohexylmethyl)-4-oxo-4,7-dihydro-3*H*-pyrrolo-[2,3-*d*]pyrimidin-6-yl]acetoxy}phenyl)diazanyl]benzoate (**12**)**

To a stirred solution of **11** (36 mg, 0.089 mmol), DMAP (3 mg, 0.025 mmol) and **1** (19 mg, 0.064 mmol) in 10 mL of CH₂Cl₂ cooled to 0 °C in an ice bath, was added EDC (22 mg, 0.116 mmol) and the mixture was warmed to ambient temperature over several hours and stirred for 24 h. The solution was diluted with 20 mL CH₂Cl₂ and was washed with water (2 × 20 mL), a saturated aqueous solution of sodium bicarbonate (20 mL), and brine (20 mL). The organic layer was dried over sodium sulfate, concentrated in vacuo and purified by silica gel column chromatography with MeOH–CH₂Cl₂ (5:95 v/v, *R_f* = 0.24) to afford 47 mg (35%) of **12** as an orange-yellow solid: mp 150-151 °C; ¹H NMR (500 MHz, CDCl₃) δ 9.65 (bs, 1H), 9.36 (bs, 1H), 8.94 (bs, 1H), 8.13 (d, *J* = 8.5, 2H), 7.98 (d, *J* = 9.0, 1H), 7.91 (d, *J* = 8.5, 2H), 7.65 (bs, 1H), 7.31 (d, *J* = 9.0, 2H), 4.03 (s, 2H), 3.93 (d, *J* = 7.0, 2H), 3.45-3.49 (m, 2H), 1.80-1.84 (m, 2H), 1.75-1.77 (m, 1H), 1.63-1.68 (m, 4H), 1.62 (s, 9H), 1.46-1.51 (m, 2H), 1.14-1.19 (m, 4H), 0.99 (t, *J* = 7.5, 3H), 0.84-0.91 (m, 2H); ¹³C NMR (126 MHz, CDCl₃) δ 180.9, 168.1, 165.3, 154.8, 153.8, 153.0, 151.8, 149.2, 146.4, 134.0, 132.2, 130.5, 125.5, 124.5, 122.7, 122.3, 120.1, 116.1, 104.0, 81.7, 50.5, 49.5, 40.4, 40.3, 39.1, 38.9, 33.4, 32.3, 32.1, 30.9, 29.8, 28.3, 26.4, 25.8, 20.4, 14.0; ESI-HR-MS calcd for (C₃₇H₄₅N₇O₆+H)⁺ 684.3510, found 684.3508.

4-[(4-Hydroxyphenyl)diazanyl]benzoic acid (13**)**

To a suspension of 4-aminobenzoic acid (1.24 g, 9.00 mmol) in 25 mL of a 2.0 M aqueous solution of hydrochloric acid cooled to 0 °C in an ice bath was added dropwise a solution of

sodium nitrite (1.04 g, 15.1 mmol) in water (5.0 mL). The pale yellow mixture was stirred at 0 °C for 20 min and added slowly to an ice-cold solution of phenol (706 mg, 7.50 mmol), sodium hydroxide (260 mg, 6.50 mmol) and potassium carbonate (1.40 g, 10.1 mmol) in 20 mL of water. The mixture was stirred for 1 h while being cooled in an ice-water bath and then warmed to ambient temperature and stirred for 3 h. The brown suspension was neutralized with a 2 M aqueous solution of HCl and filtered. The solid was dissolved in MeOH and dry loaded on silica gel and purified by column chromatography with MeOH-CH₂Cl₂ (5:95 v/v, *R_f* = 0.32) to afford 1.18 g, (65%) of **13** as an orange solid: mp 274-275 °C; ¹H NMR (500 MHz, CD₃OD) δ 8.15 (d, *J* = 8.5, 2H), 7.88 (d, *J* = 8.5, 2H), 7.85 (d, *J* = 9.0, 2H), 6.92 (d, *J* = 9.0, 2H); ¹³C NMR (126 MHz, CD₃OD) δ 169.2, 162.9, 156.8, 147.6, 133.0, 131.8, 126.4, 123.2, 116.8; ESI-HR-MS calcd for (C₁₃H₁₀N₂O₃+H)⁺ 243.0770, found 243.0772.

4-[[4-(5-Bromopentyloxy)phenyl]diazanyl]benzoic acid (14)

A mixture of **13** (120 mg, 0.500 mmol), 1,5-dibromopentane (137 mg, 0.596 mmol), potassium carbonate (207 mg, 1.50 mmol), and potassium iodide (9 mg, 0.0542 mmol) in 15 mL anhydrous ethanol was stirred and heated under reflux for 20 h. The mixture was cooled to ambient temperature and concentrated in vacuo. The residue was dissolved in 50 mL of CH₂Cl₂ and washed with 30 mL of a 2 M aqueous solution of HCl and 30 mL of brine. The organic layer was concentrated in vacuo and purified by silica gel column chromatography with MeOH-CH₂Cl₂ (1:9 v/v, *R_f* = 0.43) to afford 120 mg (62%) of **14** as an orange solid: mp 216-217 °C; ¹H NMR (500 MHz, DMSO-*d*₆) δ 13.22 (bs, 1H), 8.12 (d, *J* = 8.5, 2H), 7.93 (d, *J* = 9.0, 2H), 7.91 (d, *J* = 8.0, 2H), 7.15 (d, *J* = 9.0, 2H), 4.11 (t, *J* = 6.5, 2H), 3.58 (t, *J* = 7.0, 2H), 1.86-1.92 (m, 2H), 1.76-1.82 (m, 2H), 1.53-1.59 (m, 2H); ¹³C NMR (126 MHz, DMSO-*d*₆) δ 166.8, 162.0, 154.4, 146.1, 132.2, 130.6, 125.0, 122.2, 115.2, 67.9, 35.1, 31.9, 27.7, 24.2; ESI-HR-MS calcd for (C₁₈H₁₉N₂O₃Br+H)⁺ 391.0657, found 391.0657.

Ethyl 4-[[4-(5-bromopentyloxy)phenyl]diazanyl]benzoate (15)

To a stirred solution of **14** (125 mg, 0.320 mmol), DMAP (12 mg, 0.098 mmol), and anhydrous ethanol (2.76 g, 60 mmol) in 12 mL of CH₂Cl₂ cooled to 0 °C in an ice bath, was added EDC (67 mg, 0.350 mmol). The mixture was warmed to ambient temperature over several hours, stirred for 24 h, and concentrated in vacuo. The residue was dissolved in 40 mL of CH₂Cl₂ and washed with water (2 × 20 mL), a saturated aqueous solution of sodium bicarbonate (2 × 20 mL),

and with brine (20 mL). The organic layer was dried over sodium sulfate, concentrated in vacuo and purified by silica gel column chromatography with MeOH-CH₂Cl₂ (1:99 v/v, *R_f* = 0.83) to afford **15** (102 mg, 76%) as an orange solid: mp 98-99 °C; ¹H NMR (500 MHz, CDCl₃) δ 8.17 (d, *J* = 8.5, 2H), 7.94 (d, *J* = 8.5, 2H), 7.90 (d, *J* = 8.0, 2H), 7.00 (d, *J* = 8.5, 2H), 4.41 (q, *J* = 7.0, 2H), 4.06 (t, *J* = 6.0, 2H), 3.45 (t, *J* = 6.5, 2H), 1.93-1.98 (m, 2H), 1.83-1.88 (m, 2H), 1.62-1.69 (m, 2H), 1.42 (t, *J* = 7.0, 3H); ¹³C NMR (126 MHz, CDCl₃) δ 166.3, 162.2, 155.4, 147.0, 131.6, 130.6, 125.3, 122.4, 114.9, 68.1, 61.3, 33.7, 32.5, 28.5, 24.9, 14.5; ESI-HR-MS calcd for (C₂₀H₂₃N₂O₃Br+H)⁺ 419.0970, found 419.0969.

Ethyl 4-[[4-(5-azidopentyloxy)phenyl]diazanyl]benzoate (16)

A mixture of **15** (100 mg, 0.238 mmol), sodium azide (372 mg, 0.572 mmol), tetrabutyl ammonium bromide (100 mg, 0.31 mmol) and five drops of acetonitrile in 15 mL of CH₂Cl₂ was stirred and heated under reflux for 4 h. The mixture was cooled to ambient temperature and concentrated in vacuo. The residue was taken up in 60 mL of CH₂Cl₂, washed with brine (3 × 30 mL), and concentrated in vacuo to afford 131 mg of crude **16** as an orange solid, which was used for next step without further purification. The product was confirmed with HR-ESI-MS calcd for (C₂₀H₂₃N₅O₃+H)⁺ 382.1879, found 382.1879.

Ethyl 4-[(4-{5-[4-(3-{2-[2-(3-butylureido)-7-(cyclohexylmethyl)-4-oxo-4,7-dihydro-3H-pyrrolo[2,3-*d*]pyrimidin-6-yl]acetoxy}propyl)-1H-1,2,3-triazol-1-yl]pentyloxy}phenyl)diazanyl]benzoate (18)

To a stirred solution of **16** (131 mg) and **17** (40 mg, 0.085 mmol) in 15 mL of CH₂Cl₂ was added a solution of copper(II)sulfate pentahydrate (10 mg, 0.04 mmol) and sodium ascorbate (36 mg, 0.182 mmol) in 2 mL water. The mixture was stirred at room temperature for 24 h. The water layer was separated and washed with 40 mL of CH₂Cl₂. The combined CH₂Cl₂ layers were washed with brine (2 × 20 mL), dried over sodium sulfate, concentrated in vacuo and purified by silica gel column chromatography with MeOH-CH₂Cl₂ (1:9 v/v, *R_f* = 0.46) to afford 52 mg (71%) of **18** as an orange yellow solid: mp 159-160°C; ¹H NMR (500 MHz, CDCl₃) δ 11.24 (bs, 1H), 9.39 (bs, 1H), 8.97 (bs, 1H), 8.15 (d, *J* = 8.0, 2H), 7.89 (d, *J* = 9.0, 2H), 7.87 (d, *J* = 9.0, 2H), 7.39 (bs, 1H), 6.94 (d, *J* = 9.0, 2H), 6.85 (bs, 1H), 4.40 (q, *J* = 7.0, 2H), 4.37 (t, *J* = 6.0, 2H), 4.23 (t, *J* = 7.5, 2H), 4.15 (t, *J* = 7.0, 2H), 3.99 (t, *J* = 7.5, 2H), 3.82 (d, *J* = 7.0, 2H), 3.73 (s, 2H), 2.25-2.29 (m, 2H), 2.03-2.09 (m, 2H), 2.00-2.04 (m, 1H), 1.96-2.05 (m, 2H), 1.80-1.83 (m, 2H),

1.84-1.86 (m, 2H), 1.69-1.72 (m, 4H), 1.62-1.66 (m, 4H), 1.54-1.58 (m, 2H), 1.46-1.52 (m, 4H), 1.41 (t, $J = 9.0$, 3H), 0.97 (t, $J = 7.0$, 3H); ^{13}C NMR (126 MHz, CDCl_3) δ 170.0, 166.3, 162.1, 159.3, 155.4, 154.1, 148.6, 147.0, 146.4, 131.6, 130.6, 127.5, 126.3, 125.3, 122.4, 114.8, 103.5, 103.3, 102.5, 83.1, 69.4, 67.9, 64.4, 64.0, 61.3, 50.3, 49.3, 40.2, 39.0, 33.2, 32.2, 31.2, 30.1, 28.6, 28.2, 27.4, 26.3, 25.9, 23.2, 22.0, 20.5, 15.3, 14.5, 14.0; ESI-HR-MS calcd for $(\text{C}_{45}\text{H}_{58}\text{N}_{10}\text{O}_7+\text{H})^+$ 851.4568, found 851.4562.

^1H and ^{13}C NMR spectra

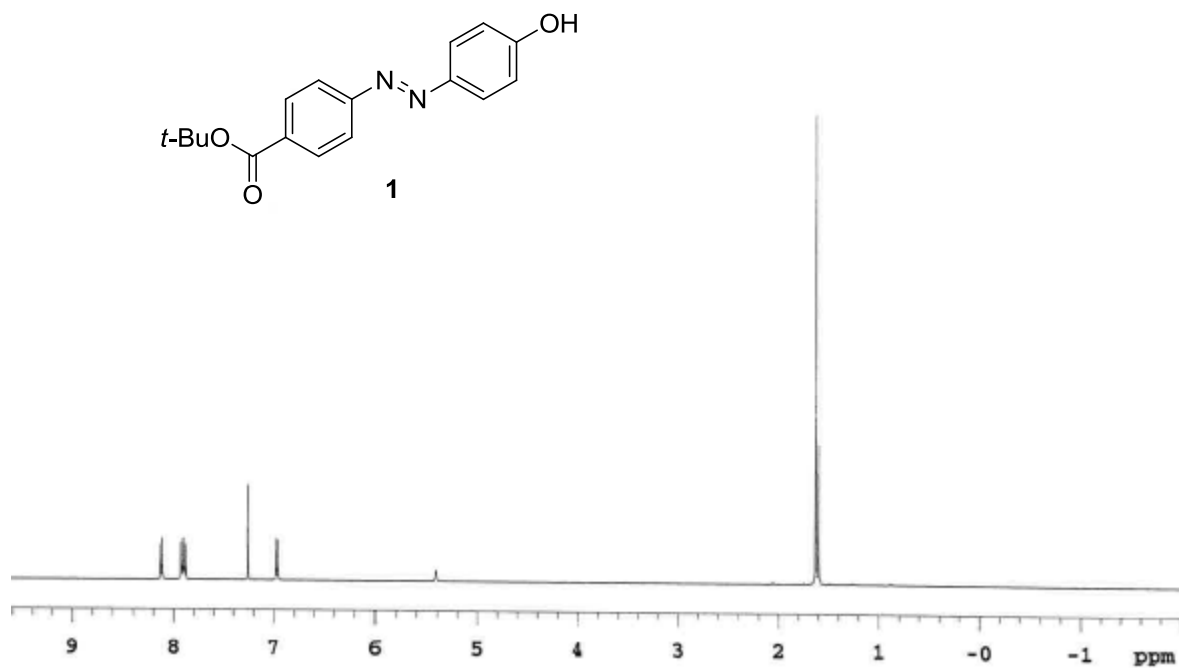


Figure S1: Compound **1**, ^1H NMR (500 MHz, CDCl_3).

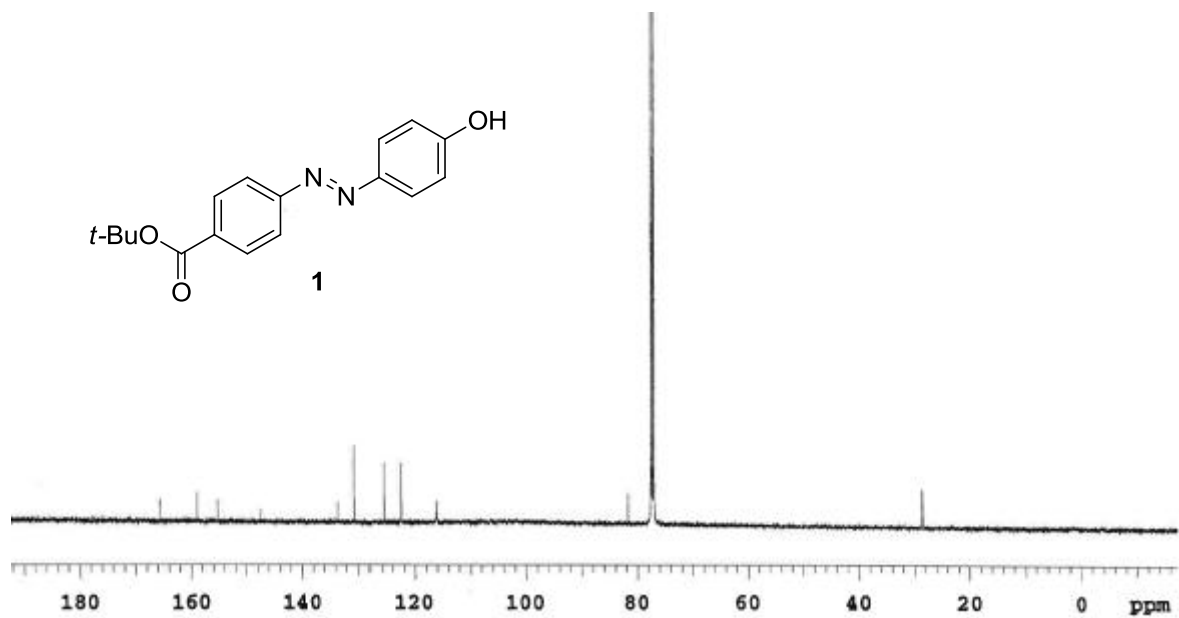


Figure S2: Compound **1**, ^{13}C NMR (126 MHz, CDCl_3).

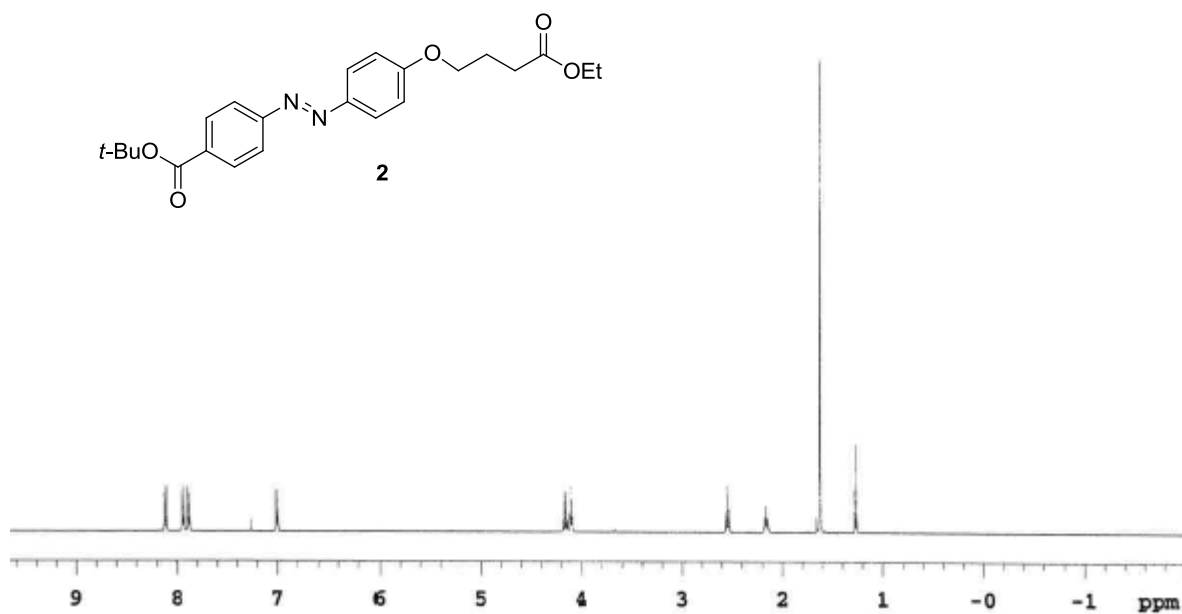


Figure S3: Compound 2, ¹H NMR (500 MHz, CDCl₃).

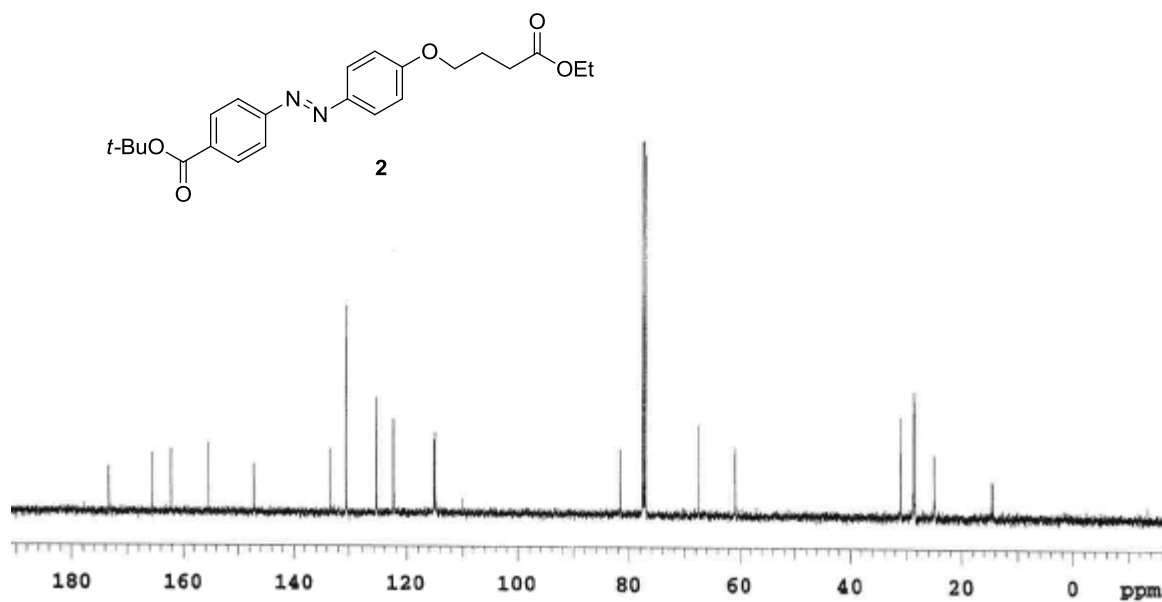


Figure S4: Compound 2, ¹³C NMR (126 MHz, CDCl₃).

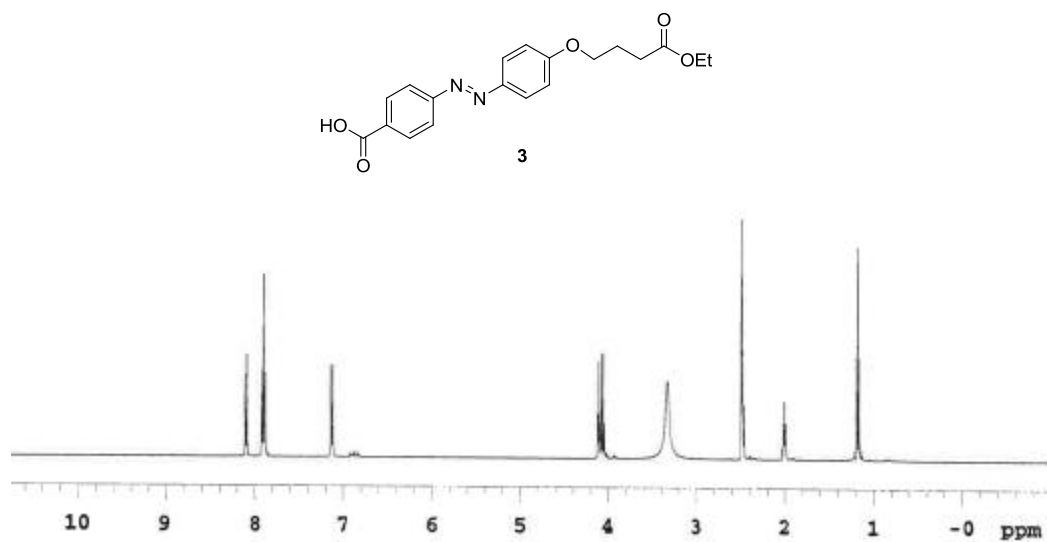


Figure S5: Compound 3, ¹H NMR (500 MHz, DMSO-*d*₆).

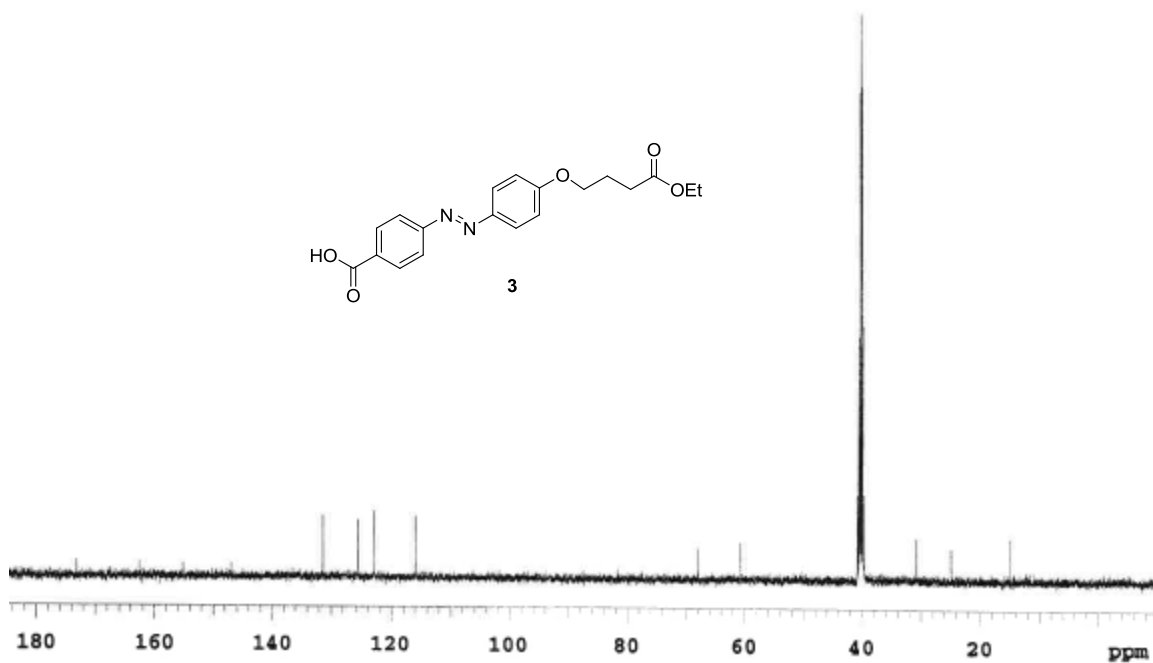


Figure S6: Compound 3, ¹³C NMR (126 MHz, DMSO-*d*₆).

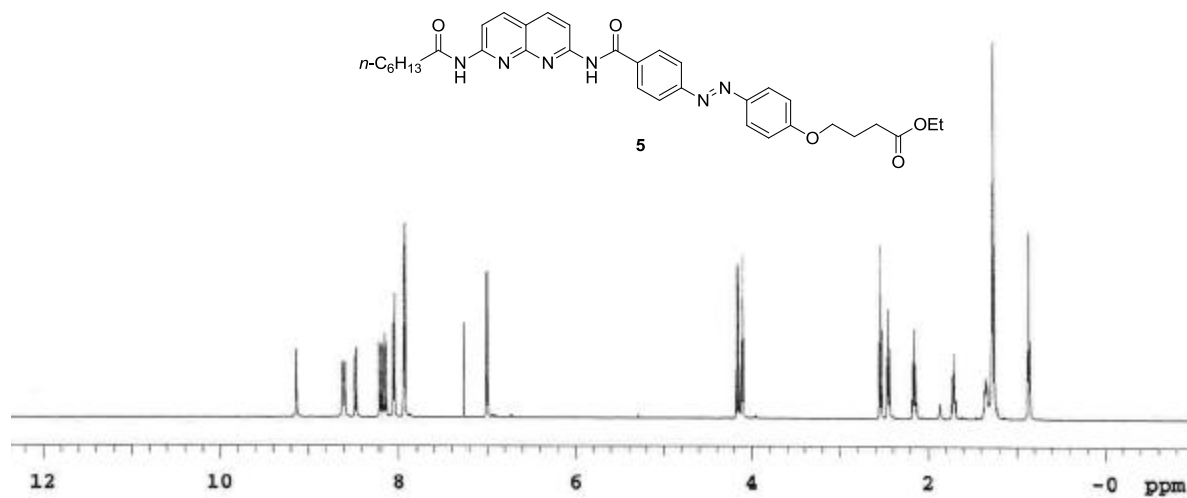


Figure S7: Compound **5**, ¹H NMR (500 MHz, CDCl₃).

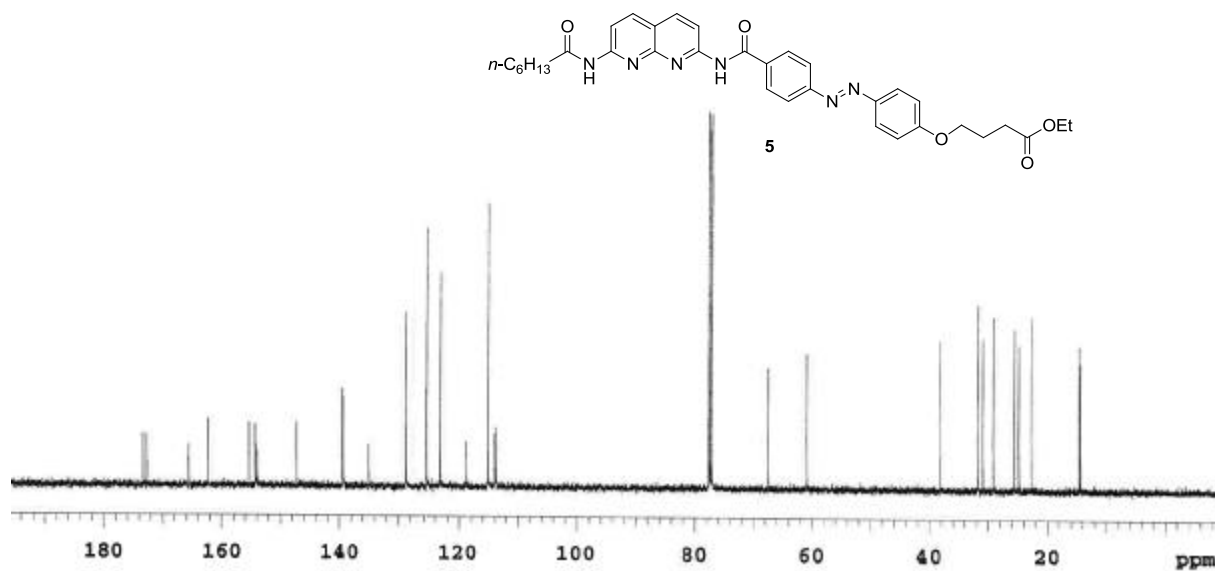


Figure S8: Compound **5**, ¹³C NMR (126 MHz, CDCl₃).



Figure S9: Compound 6, ¹H NMR (500 MHz, CDCl₃).

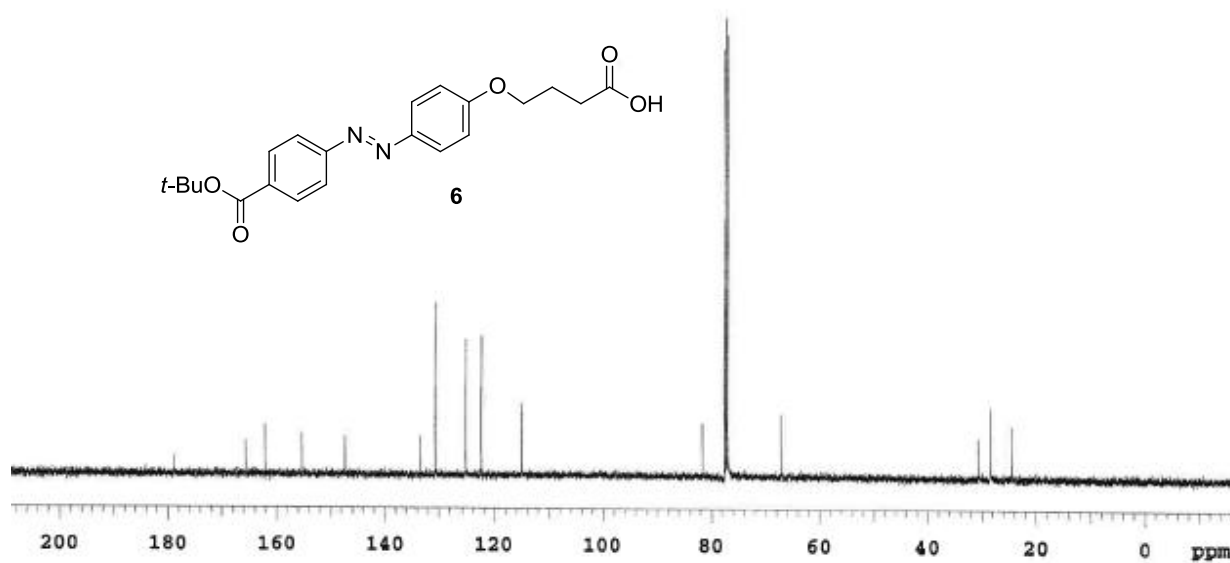


Figure S10: Compound 6, ¹³C NMR (126 MHz, CDCl₃).

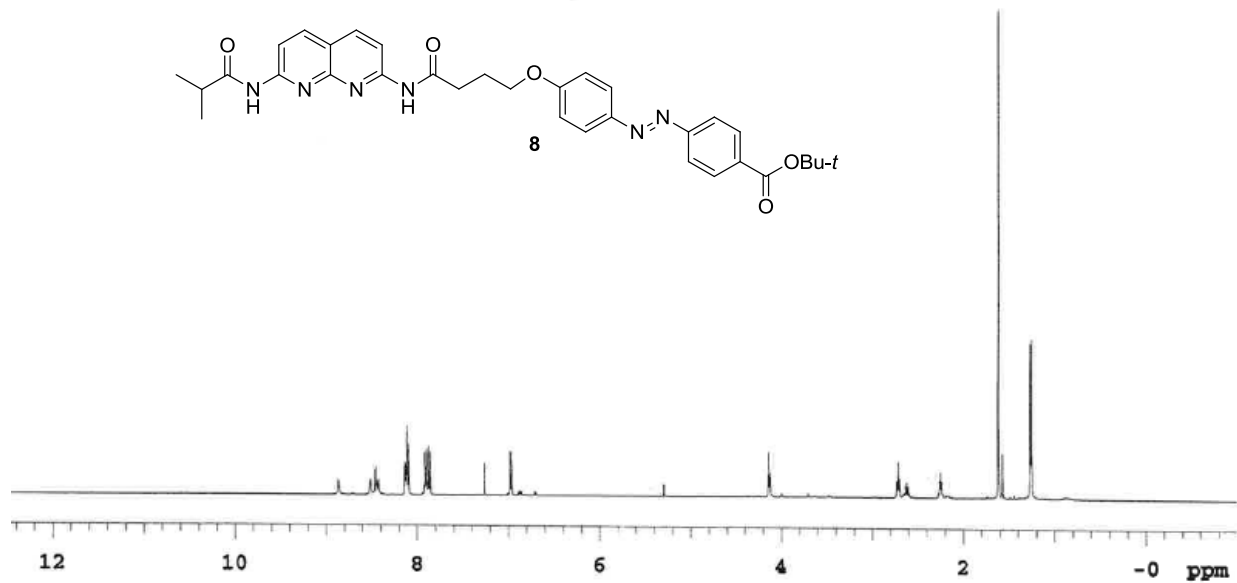


Figure S11: Compound 8, ¹H NMR (500 MHz, CDCl₃).

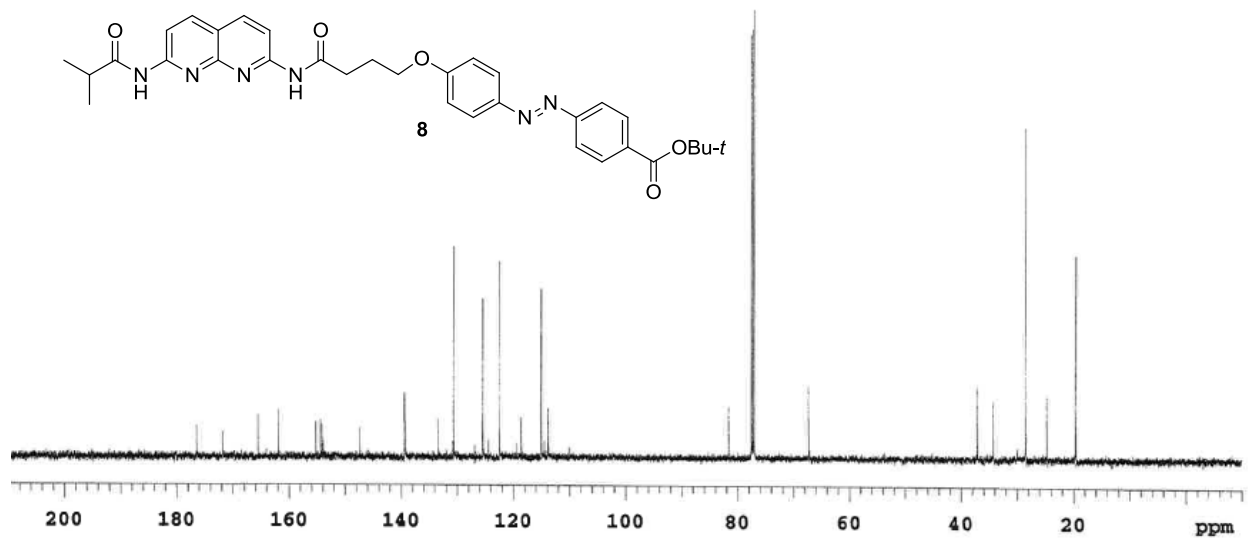


Figure S12: Compound 8, ¹³C NMR (126 MHz, CDCl₃).

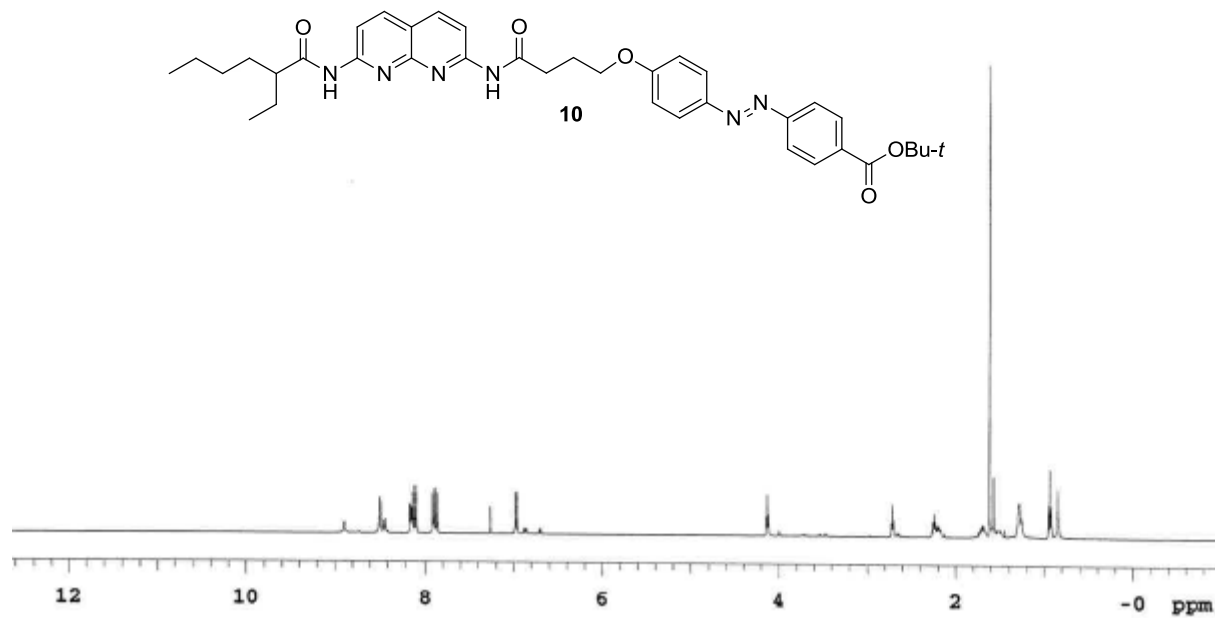


Figure S13: Compound 10, ¹H NMR (500 MHz, CDCl₃).

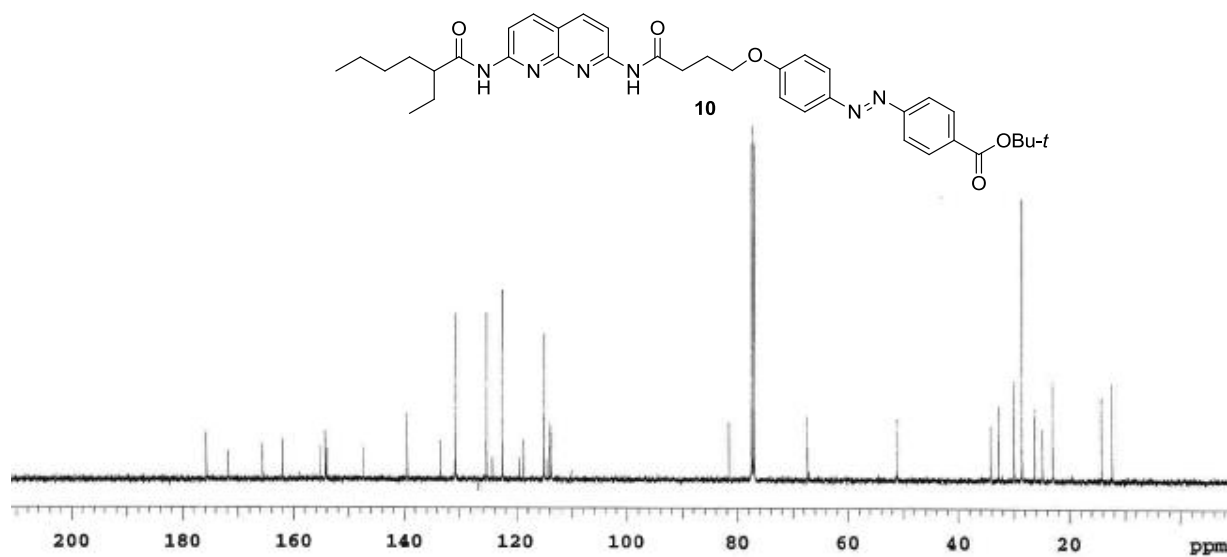


Figure S14: Compound 10, ¹³C NMR (126 MHz, CDCl₃).

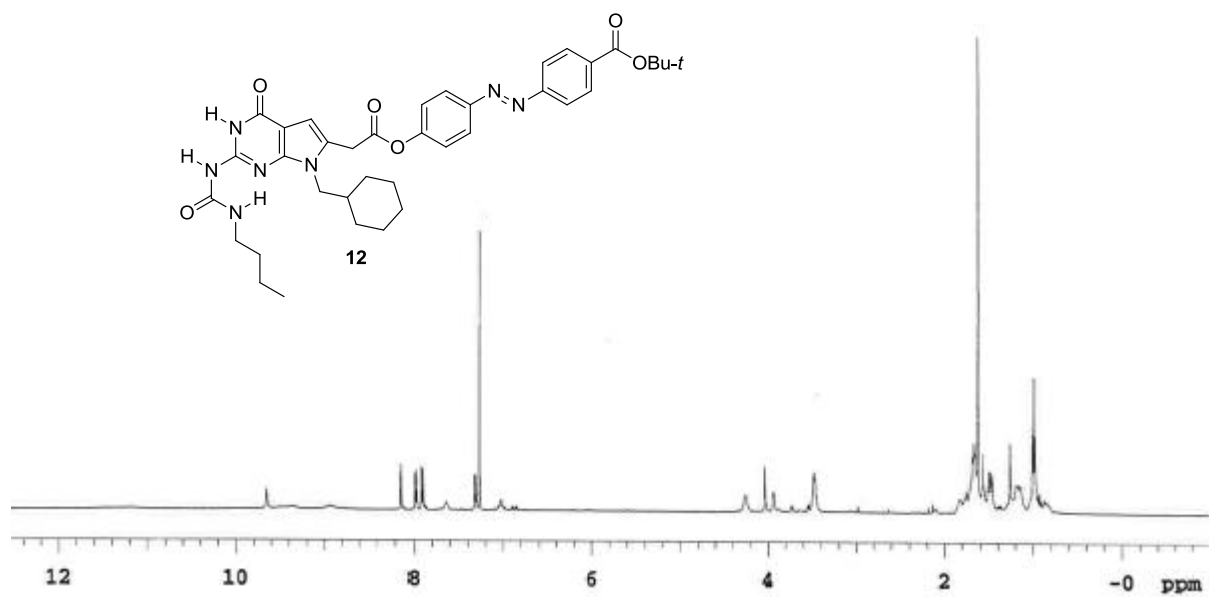


Figure S15: Compound 12, ¹H NMR (500 MHz, CDCl₃).

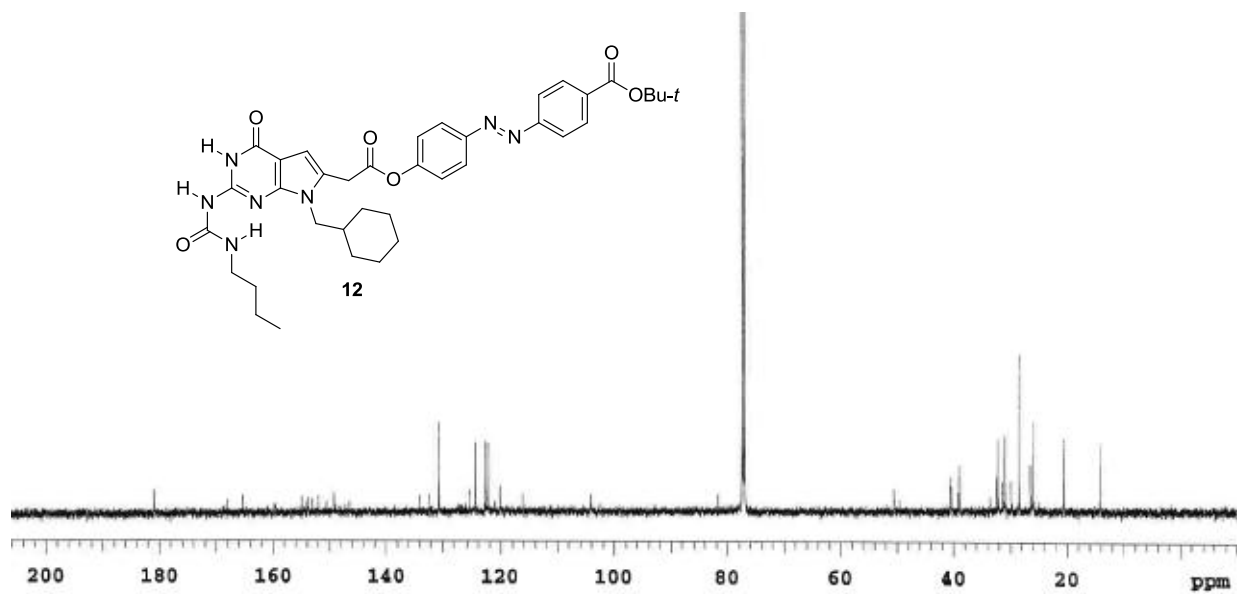


Figure S16: Compound 12, ¹³C NMR (126 MHz, CDCl₃).

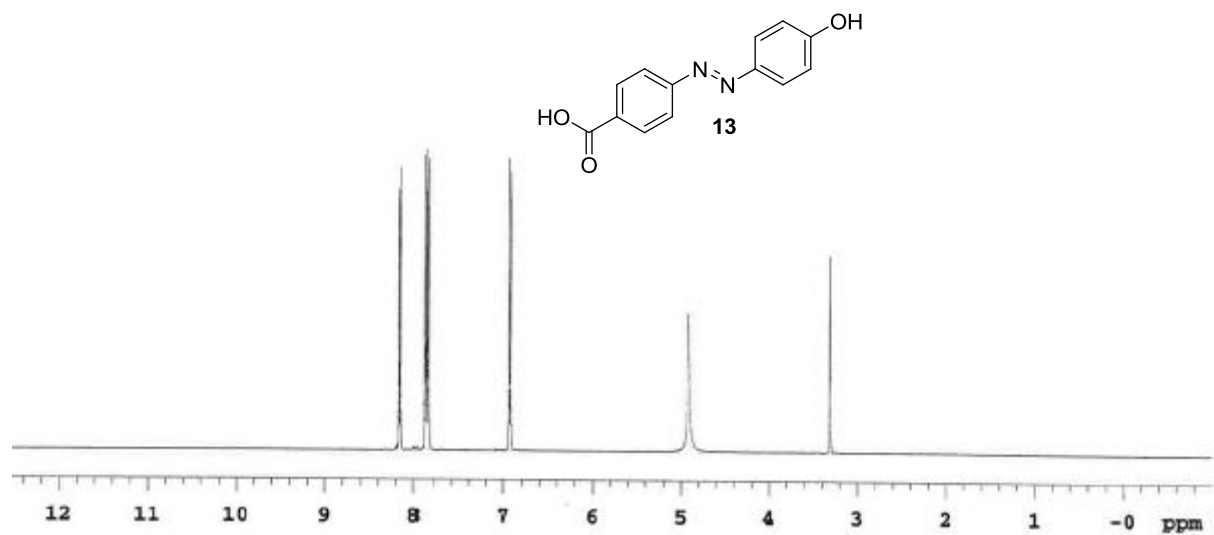


Figure S17: Compound 13, ^1H NMR (500 MHz, CD_3OD).

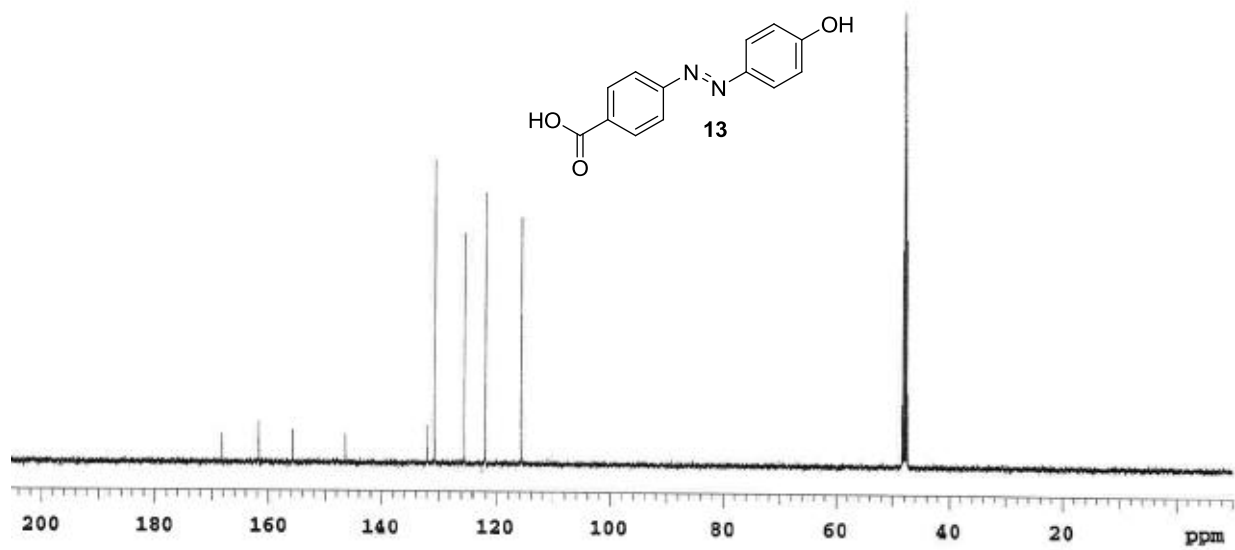


Figure S18: Compound 13, ^{13}C NMR (126 MHz, CD_3OD).

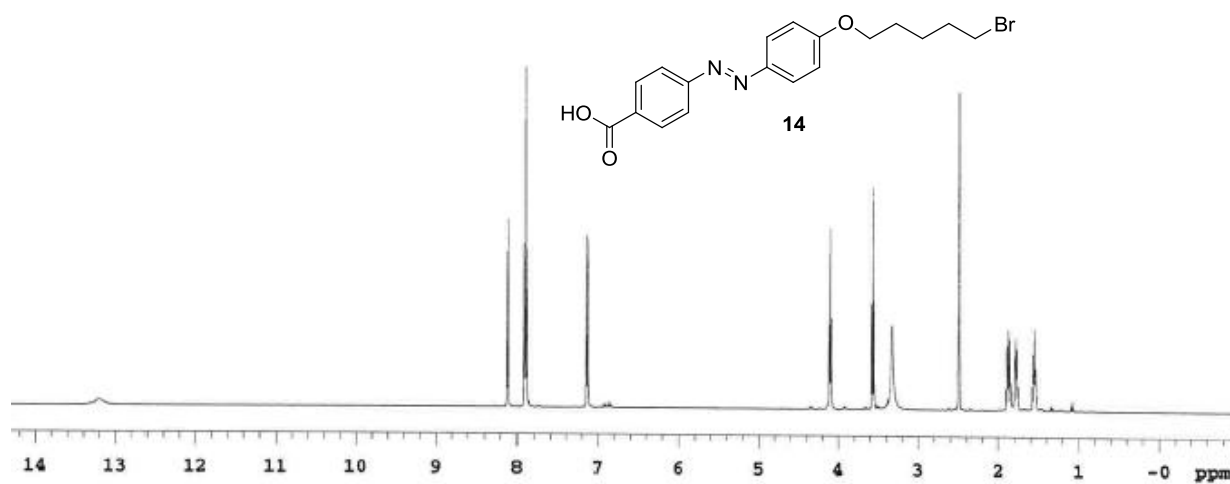


Figure S19: Compound 14, ¹H NMR (500 MHz, DMSO-*d*₆).

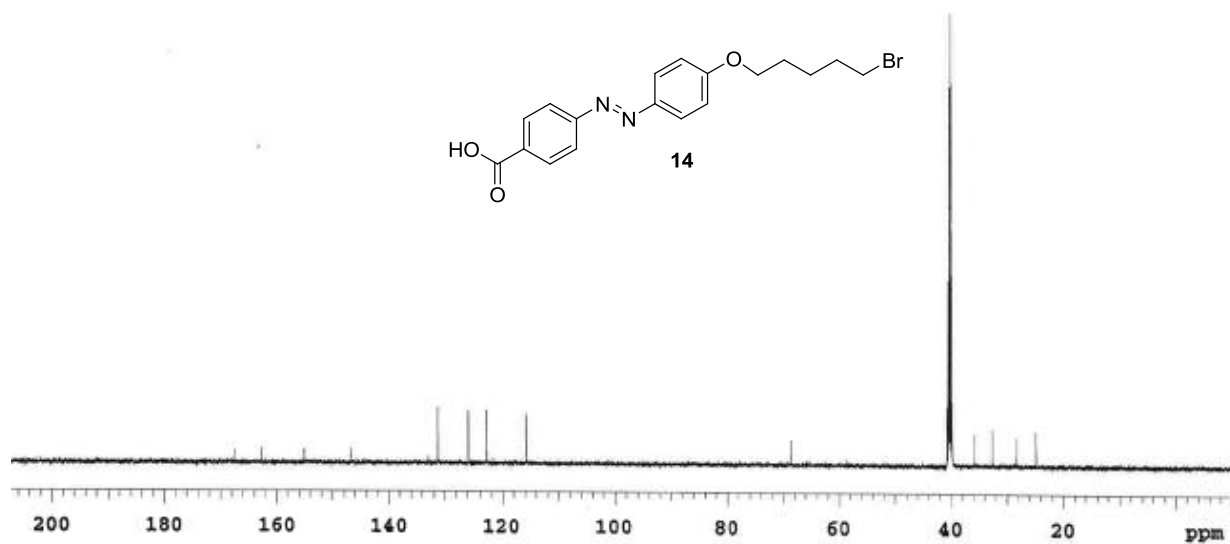


Figure S20: Compound 14, ¹³C NMR (126 MHz, DMSO-*d*₆).

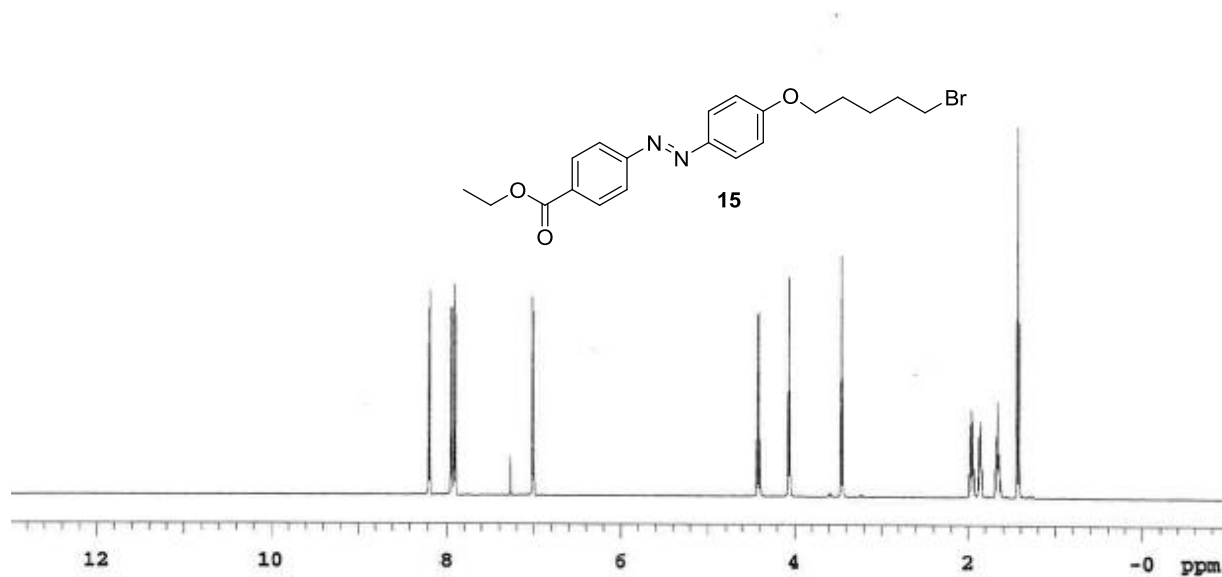


Figure S21: Compound 15, ¹H NMR (500 MHz, CDCl₃).

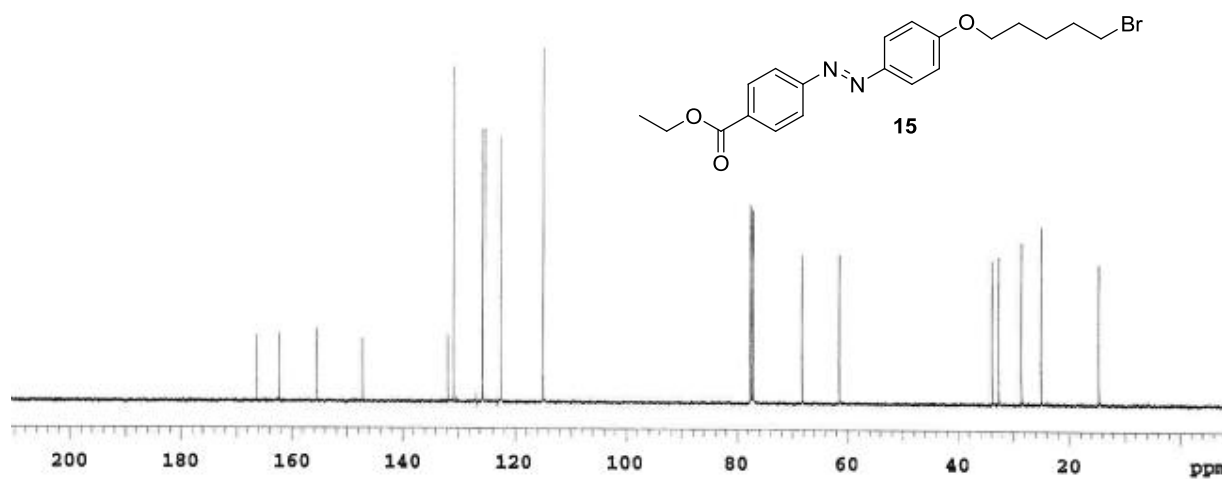


Figure S22: Compound 15, ¹³C NMR (126 MHz, CDCl₃).

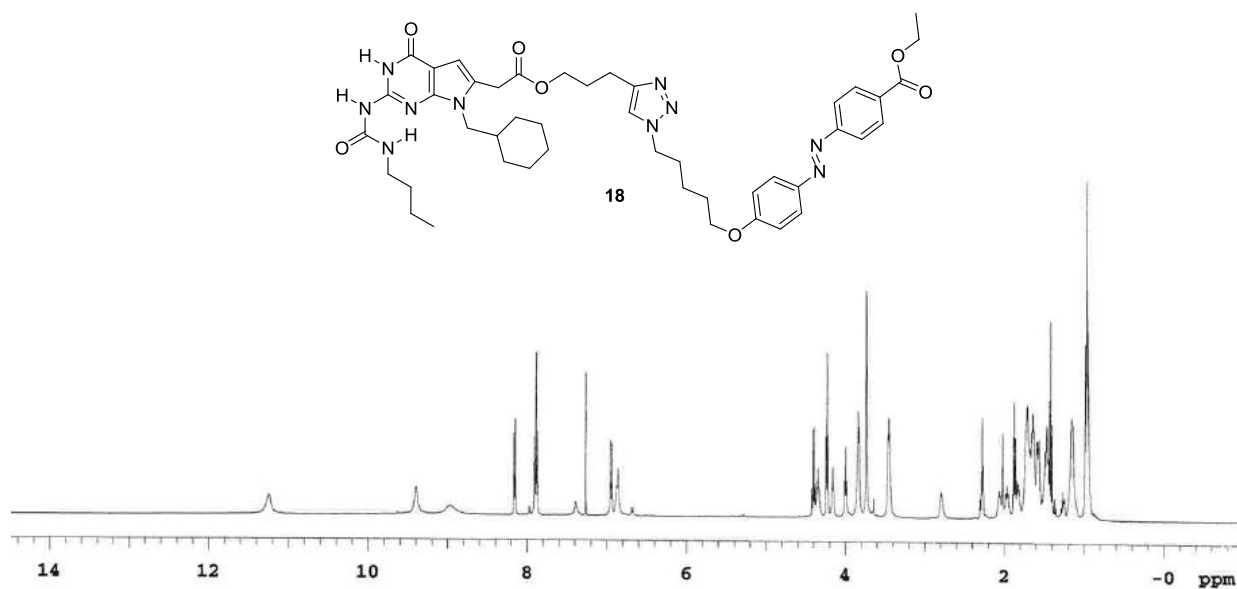


Figure S23: Compound 18, ¹H NMR (500 MHz, CDCl₃).

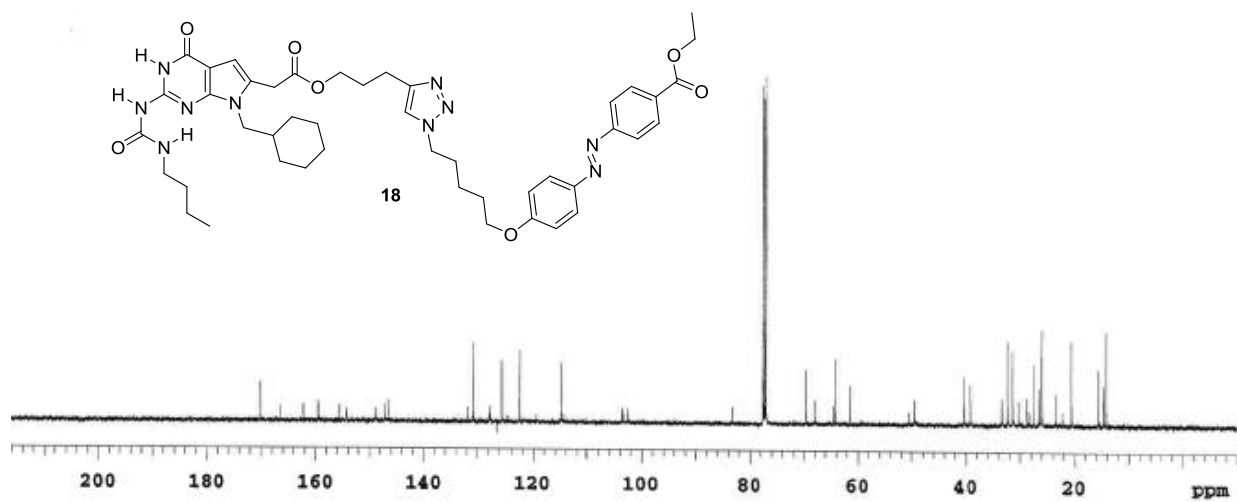


Figure S24: Compound 18, ¹³C NMR (126 MHz, CDCl₃).

Experimental composition reports of HR-ESI-MS

Elemental Composition Report

Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

265 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-100 H: 0-150 N: 0-5 O: 0-5 Na: 0-1

Zhang, Yagang, scz-zyg-3-3

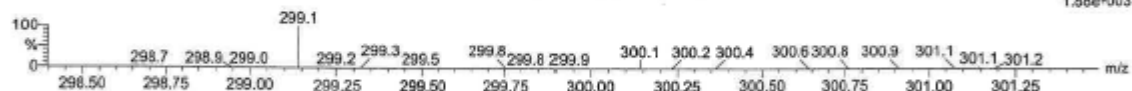
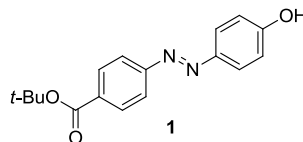
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_35844 45 (3.221) AM (Cen,3, 80.00, Ar,15000.0,716.46,0.70,LS 3); Sm (SG, 2x5.00); Cm (45:46)

Q-tof UE521

1: TOF MS ES+

1.88e+003



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
299.1395	299.1396	-0.1	-0.3	9.5	1.0	C17 H19 N2 O3
	299.1412	-1.7	-5.7	10.5	0.8	C20 H20 O Na
	299.1372	2.3	7.7	6.5	6.8	C15 H20 N2 O3 Na

Figure S25: Elemental composition report of compound 1 (HR-ESI-MS).

Elemental Composition Report

Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

363 formula(e) evaluated with 4 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-100 H: 0-150 N: 0-5 O: 0-5 Na: 0-1

Zhang, Yagang, scz-zyg-3-5

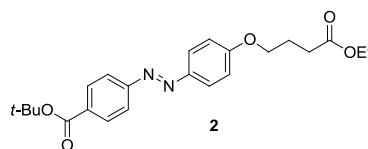
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_35845 31 (2.221) AM (Cen,3, 80.00, Ar,15000.0,716.46,0.70,LS 3); Sm (SG, 2x5.00); Cm (27:31)

Q-tof UE521

1: TOF MS ES+

1.84e+003



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
413.2079	413.2076	0.3	0.7	10.5	26.9	C23 H29 N2 O5
	413.2093	-1.4	-3.4	11.5	29.2	C26 H30 O3 Na
	413.2052	2.7	6.5	7.5	31.6	C21 H30 N2 O5 Na
	413.2117	-3.8	-9.2	14.5	30.8	C28 H29 O3

Figure S26: Elemental composition report of compound 2 (HR-ESI-MS).

Elemental Composition Report

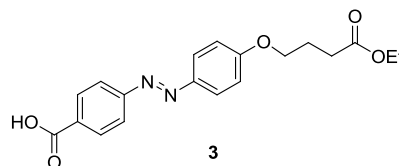
Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3



Monoisotopic Mass, Even Electron Ions

40 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-100 H: 0-150 N: 1-3 O: 4-6

Zhang, Yagang, scz-zyg-3-10

University of Illinois, SCS, Mass Spectrometry Lab

Q-tof UES21

Qtof_36005 51 (3.650) AM (Cen,3, 80.00, Ar,15000.0,716.46,0.70,LS 3); Sm (SG, 2x3.00); Cm (51:53)

1: TOF MS ES+
2.47e+003



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
357.1447	357.1450	-0.3	-0.8	10.5	12.6	C19 H21 N2 O5

Figure S27: Elemental composition report of compound 3 (HR-ESI-MS).

Elemental Composition Report

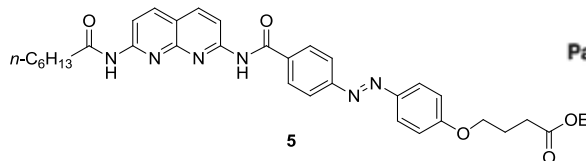
Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3



Monoisotopic Mass, Even Electron Ions

170 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-100 H: 0-150 N: 5-6 O: 0-5 Na: 0-1

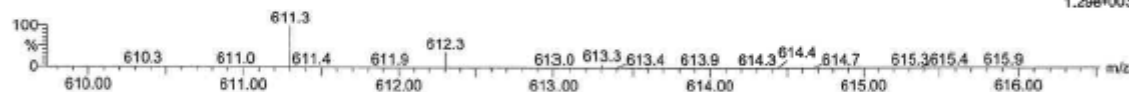
Zhang, Yagang, scz-zyg-3-9

University of Illinois, SCS, Mass Spectrometry Lab

Q-tof UES21

Ctof_35959 30 (2.150) AM (Cen,3, 80.00, Ar,15000.0,716.46,0.70,LS 3); Sm (SG, 2x3.00); Cm (30)

1: TOF MS ES+
1.29e+003



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
611.2984	611.2982	0.2	0.3	18.5	1.5	C34 H39 N6 O5
	611.2958	2.6	4.3	15.5	0.4	C32 H40 N6 O5 Na
	611.2923	6.1	10.0	27.5	15.7	C41 H35 N6

Figure S28: Elemental composition report of compound 5 (HR-ESI-MS).

Elemental Composition Report

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0
 Element prediction: Off
 Number of isotope peaks used for I-FIT = 3

Monoisotopic Mass, Even Electron Ions

133 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-150 H: 0-200 N: 2-6 O: 2-7

Zhang, Yagang, 3-17-2

University of Illinois, SCS, Mass Spectrometry Lab

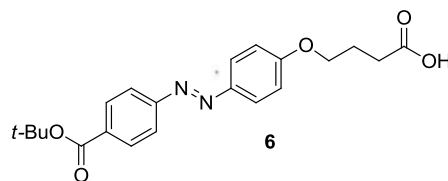
Qtof_36207 26 (1.864) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x5.00); Cm (23:26)

Q-tof UE521
 1: TOF MS ES+
 9.28e+002



Minimum: -1.5
 Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
385.1765	385.1763	0.2	0.5	10.5	0.7	C21 H25 N2 O5



Page 1

Figure S29: Elemental composition report of compound 6 (HR-ESI-MS).

Elemental Composition Report

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0
 Element prediction: Off
 Number of isotope peaks used for I-FIT = 3

Monoisotopic Mass, Even Electron Ions

357 formula(e) evaluated with 4 results within limits (all results (up to 1000) for each mass)

Elements Used:

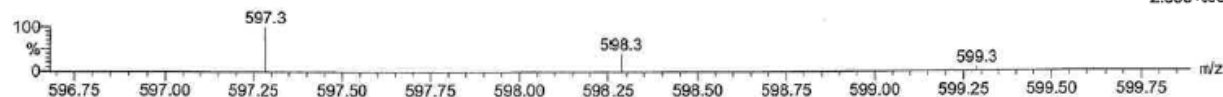
C: 0-150 H: 0-200 N: 0-6 O: 0-6

Zhang, Yagang, scz-zyg-3-21

University of Illinois, SCS, Mass Spectrometry Lab

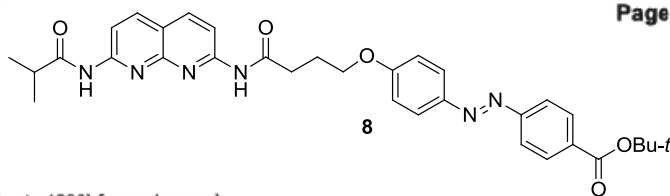
Qtof_36535 30 (2.150) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x3.00); Cm (29:30)

Q-tof UE521
 1: TOF MS ES+
 2.95e+003



Minimum: -1.5
 Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
597.2825	597.2825	0.0	0.0	18.5	0.1	C33 H37 N6 O5
597.2794	597.2794	3.1	5.2	26.5	34.7	C44 H37 O2
597.2866	597.2866	-4.1	-6.9	22.5	8.1	C38 H37 N4 O3
597.2767	597.2767	5.8	9.7	27.5	18.0	C40 H33 N6



Page 1

Figure S30: Elemental composition report of compound 8 (HR-ESI-MS).

Elemental Composition Report

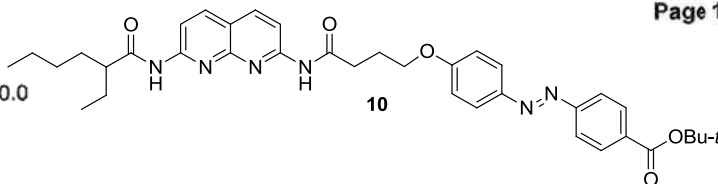
Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3



Monoisotopic Mass, Even Electron Ions

441 formula(e) evaluated with 5 results within limits (all results (up to 1000) for each mass)

Elements Used:

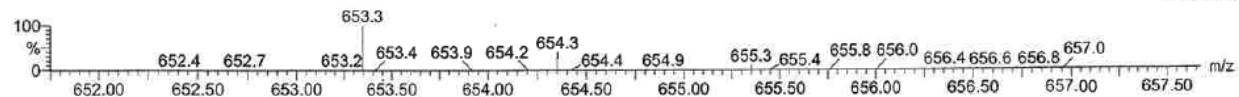
C: 0-150 H: 0-200 N: 0-6 O: 0-7

Zhang, Yagang, scz-zyg-3-20

University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36452 21 (1.507) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x3.00); Cm (20:21)

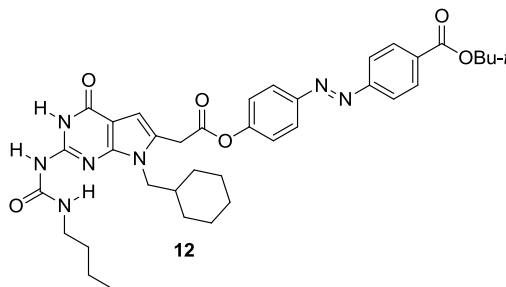
Q-tof UE521
1: TOF MS ES+
1.63e+003



Minimum: -1.5
Maximum: 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
653.3453	653.3451	0.2	0.3	18.5	0.9	C37 H45 N6 O5
	653.3478	-2.5	-3.8	17.5	4.7	C41 H49 O7
	653.3420	3.3	5.1	26.5	25.4	C48 H45 O2
	653.3492	-3.9	-6.0	22.5	8.5	C42 H45 N4 O3
	653.3393	6.0	9.2	27.5	15.5	C44 H41 N6

Figure S31: Elemental composition report of compound **10** (HR-ESI-MS).



Elemental Composition Report

Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

135 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

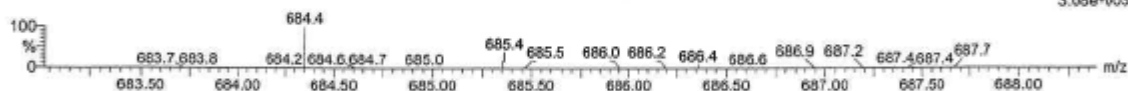
C: 0-150 H: 0-200 N: 5-7 O: 4-6 S: 0-1

Zhang, Yagang, scz-zyg-3-15-2

University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36134A 53 (3.793) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x3.00); Cm (53:54)

Q-tof UE521
1: TOF MS ES+
3.08e+003



Minimum: -1.5
Maximum: 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
684.3508	684.3510	-0.2	-0.3	18.5	2.9	C37 H46 N7 O6
	684.3543	-3.5	-5.1	13.5	13.1	C34 H50 N7 O6 S
	684.3550	-4.2	-6.1	22.5	16.7	C42 H46 N5 O4

Figure S32: Elemental composition report of compound **12** (HR-ESI-MS).

Elemental Composition Report

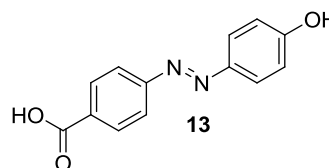
Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3



Monoisotopic Mass, Even Electron Ions

152 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-150 H: 0-200 N: 0-6 O: 0-6

Zhang, Yagang, scz-zyg-3-22

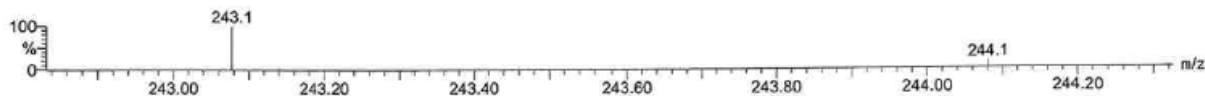
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36536 43 (3.079) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x3.00)

Q-tof UE521

1: TOF MS ES+

6.61e+002



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
243.0772	243.0770	0.2	0.8	9.5	n/a	C13 H11 N2 O3

Figure S33: Elemental composition report of compound **13** (HR-ESI-MS).

Elemental Composition Report

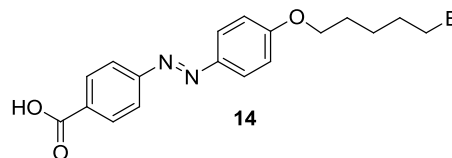
Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3



Monoisotopic Mass, Even Electron Ions

309 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-150 H: 0-200 N: 0-5 O: 0-6 Na: 0-1 Br: 1-1

Zhang, Yagang, scz-zyg-3-24-2

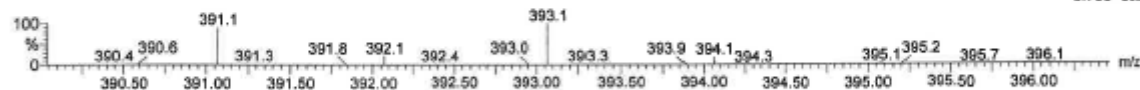
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36594 48 (3.436) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x3.00); Cm (48:49)

Q-tof UE521

1: TOF MS ES+

9.75e+002



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
391.0657	391.0657	0.0	0.0	9.5	2.5	C18 H20 N2 O3 Br
	391.0673	-1.6	-4.1	10.5	2.5	C21 H21 O Na Br
	391.0633	2.4	6.1	6.5	4.8	C16 H21 N2 O3 Na Br

Figure S34: Elemental composition report of compound **14** (HR-ESI-MS).

Elemental Composition Report

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

153 formula(e) evaluated with 3 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-150 H: 0-200 N: 0-5 O: 0-5 Br: 1-1

Zhang, Yagang, scz-zyg-3-26-1

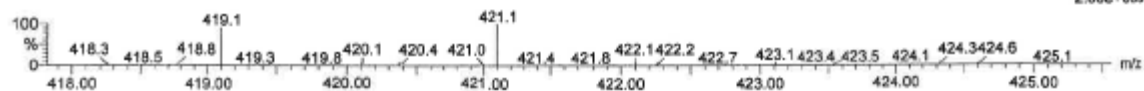
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36632_45 (3.221) AM (Cen,3, 80.00, Ar,15000.0,734.47,0.70,LS 3); Sm (SG, 2x3.00); Cm (45)

Q-tof UE521

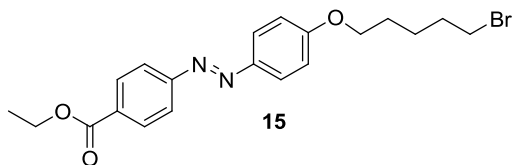
1: TOF MS ES+

2.80e+003



Minimum: -1.5
Maximum: 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
419.0969	419.0970	-0.1	-0.2	9.5	5.8	C20 H24 N2 O3 Br
	419.0930	3.9	9.3	5.5	15.0	C15 H24 N4 O5 Br
	419.1011	-4.2	-10.0	13.5	22.3	C25 H24 O Br



15

Page 1

Figure S35: Elemental composition report of compound **15** (HR-ESI-MS).

Elemental Composition Report

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

198 formula(e) evaluated with 1 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-150 H: 0-200 N: 0-5 O: 0-6

Zhang, Yagang, scz-zyg-3-27

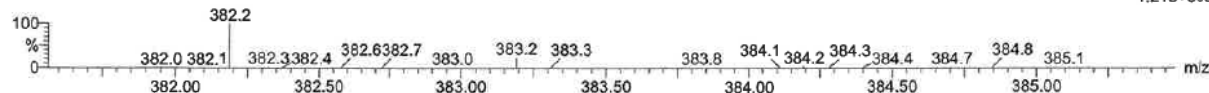
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36663_32 (2.293) AM (Cen,3, 80.00, Ar,15000.0,716.46,0.70,LS 3); Sm (SG, 2x3.00); Cm (32:34)

Q-tof UE521

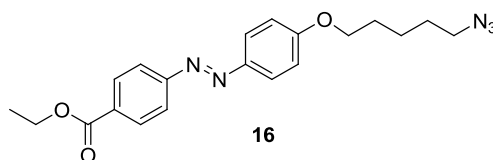
1: TOF MS ES+

1.21e+003



Minimum: -1.5
Maximum: 150.0

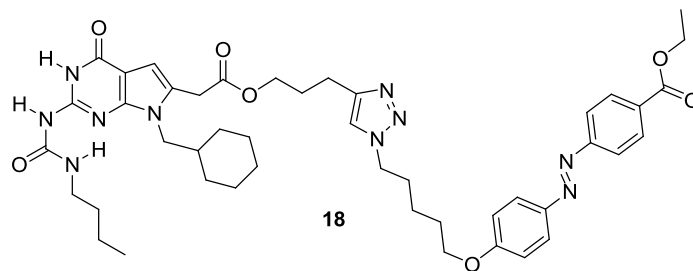
Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
382.1879	382.1879	0.0	0.0	11.5	1.1	C20 H24 N5 O3



16

Page 1

Figure S36: Elemental composition report of compound **16** (HR-ESI-MS).



Elemental Composition Report

Page 1

Single Mass Analysis

Tolerance = 10.0 PPM / DBE: min = -1.5, max = 150.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

67 formula(e) evaluated with 2 results within limits (all results (up to 1000) for each mass)

Elements Used:

C: 0-150 H: 0-200 N: 10-11 O: 7-8 Na: 0-1

Zhang, Yagang, scz-zyg-3-29

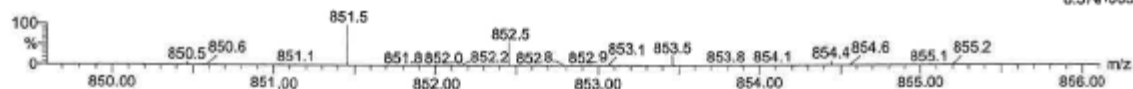
University of Illinois, SCS, Mass Spectrometry Lab

Qtof_36727 37 (1.576) AM (Cen,3, 80.00, Ar,15000.0,716.46,0.70,LS 3); Sm (SG, 2x3.00); Cm (37:40)

Q-tof UE521

1: TOF MS ES+

8.57e+003



Minimum: -1.5
Maximum: 5.0 10.0 150.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Formula
851.4562	851.4568	-0.6	-0.7	21.5	125.9	C45 H59 N10 O7
	851.4544	1.8	2.1	18.5	167.5	C43 H60 N10 O7 Na

Figure S37: Elemental composition report of compound **18** (HR-ESI-MS).

References

1. Still, W. C.; Kahn, M.; Mitra, A. *J. Org. Chem.* **1978**, *43*, 2923–2925.

doi:10.1021/jo00408a041