

Supporting Information

for

Conformational analysis and intramolecular interactions in monosubstituted phenylboranes and phenylboronic acids

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¹H and ¹¹B NMR spectra for 2-fluorophenylboronic acid.

Potential energy surfaces for compounds **3–9**.

Table of Contents

Figure S1: ¹ H NMR spectrum for 1 in C ₆ D ₆ solution (2 mg mL ⁻¹).....	S2
Figure S2: ¹ H NMR spectrum for 1 in CD ₃ CN solution (20 mg mL ⁻¹).....	S2
Figure S3: ¹¹ B NMR spectrum for 1 in C ₆ D ₆ solution (2 mg mL ⁻¹).....	S3
Figure S4: Potential energy surface for 2-fluorophenylborane.....	S4
Figure S5: Potential energy surface for 2-chlorophenylborane.....	S4
Figure S6: Potential energy surface for 2-bromophenylborane.....	S4
Figure S7: Potential energy surface for 2-hydroxyphenylborane.....	S5
Figure S8: Potential energy surface for 2-sulfanylphenylborane.....	S5
Figure S9: Potential energy surface for 2-aminophenylborane.....	S6
Figure S10: Potential energy surface for 2-phosphanylphenylborane.....	S6

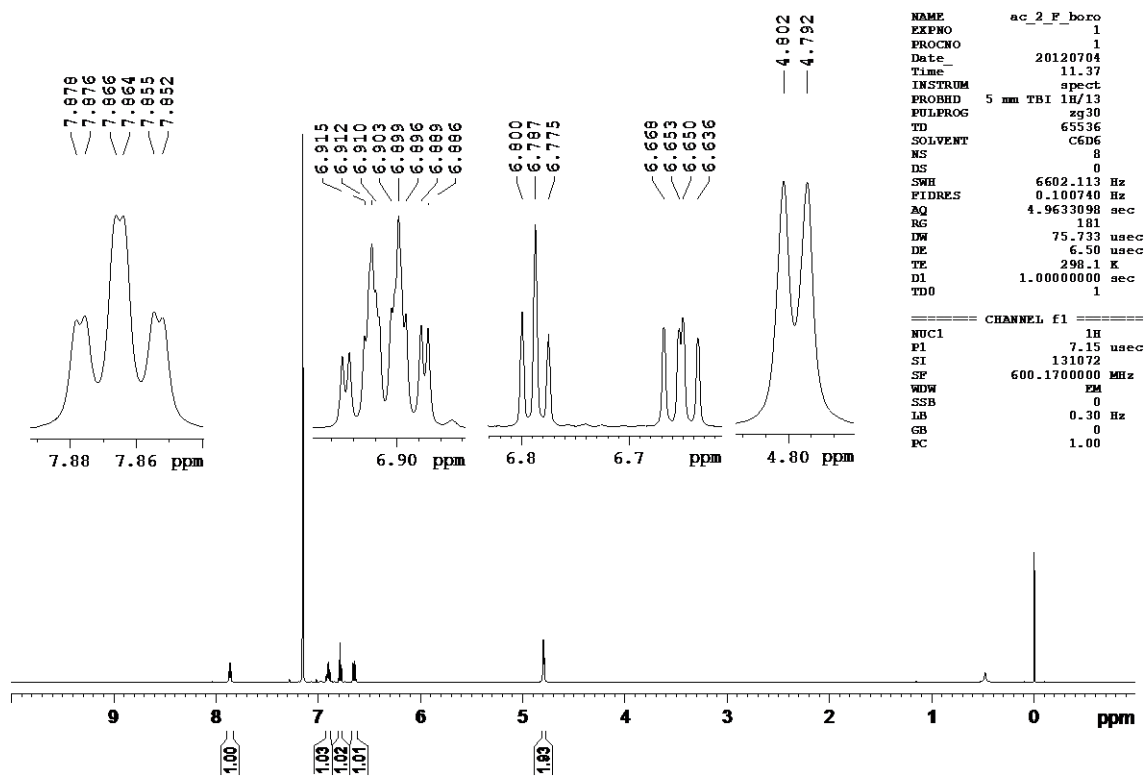


Figure S1: ^1H NMR spectrum for **1** in C_6D_6 solution (2 mg mL^{-1}).

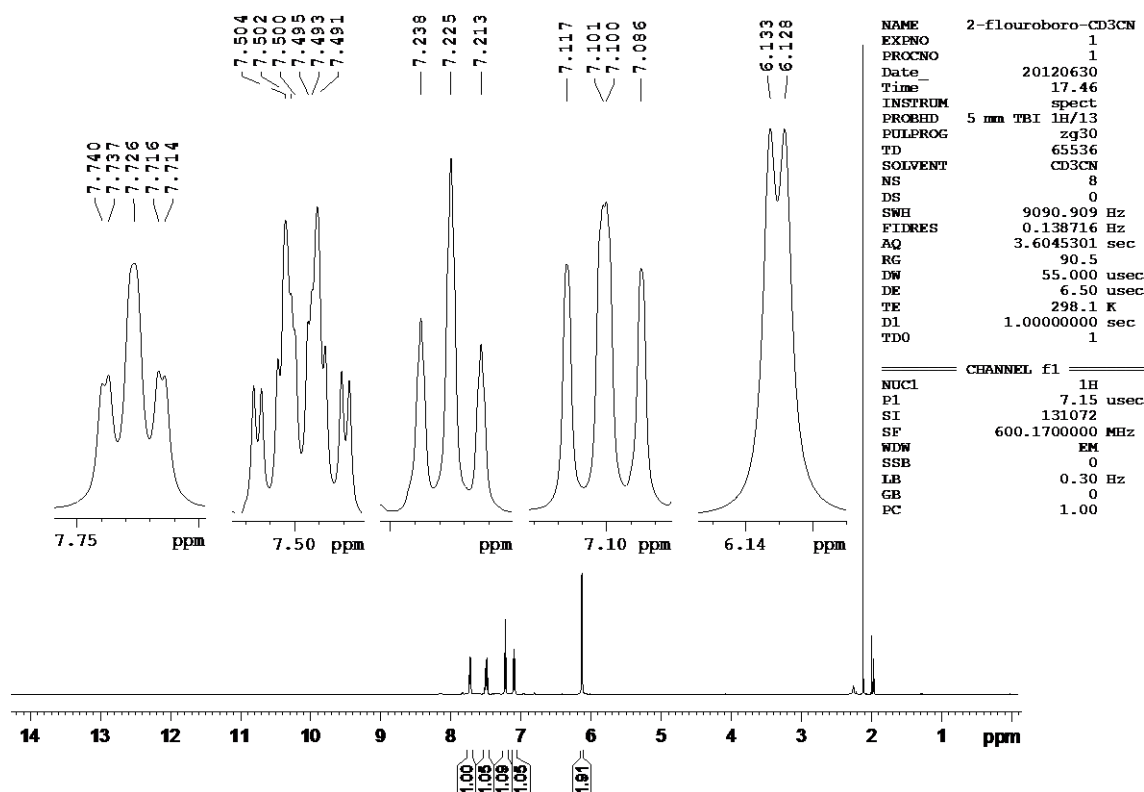
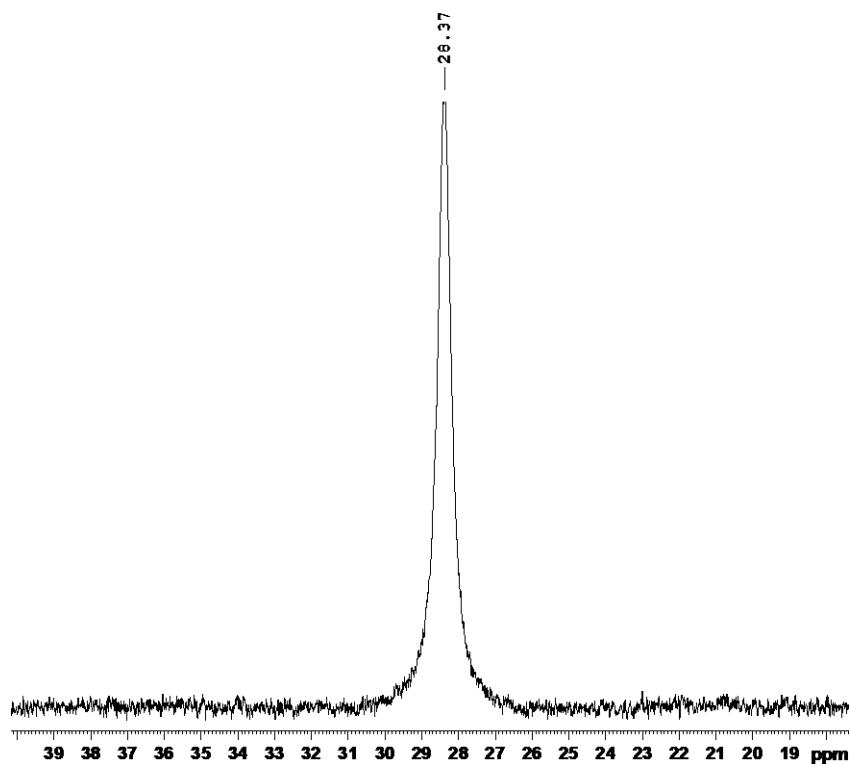


Figure S2: ^1H NMR spectrum for **1** in CD_3CN solution (20 mg mL^{-1}).



```

NAME      2-Fluoroboronic
EXPNO    1
PROCNO   1
Date_    20120717
Time     11.04
INSTRUM  spect
PROBHD   5 mm EATBO BB-
PULPROG  zg
TD       32768
SOLVENT  C6D6
NS       16
DS       0
SWH      4391.101 Hz
FIDRES   0.134006 Hz
AQ       3.7312329 sec
RG       203
LW       113.867 usec
DE       6.50 usec
TE       298.2 K
D1       5.0000000 sec
TD0      1
===== CHANNEL f1 =====
NUC1     11B
P1       18.38 usec
SI       32768
SF       192.5583870 MHz
WDW      EM
SSB      0
LB       2.00 Hz
GB       0
PC       1.40

```

Figure S3: ¹¹B NMR spectrum for **1** in C₆D₆ solution (2 mg mL⁻¹).

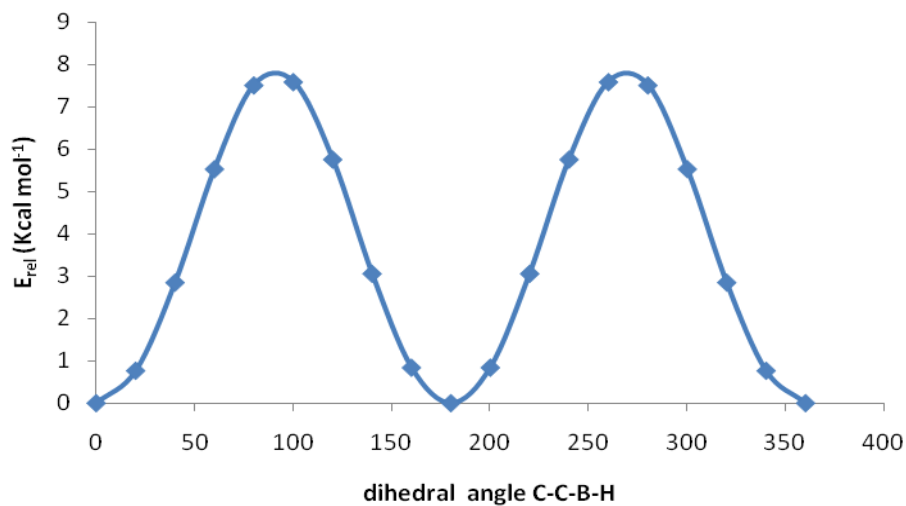


Figure S4: Potential energy surface for 2-fluorophenylborane (**3**).

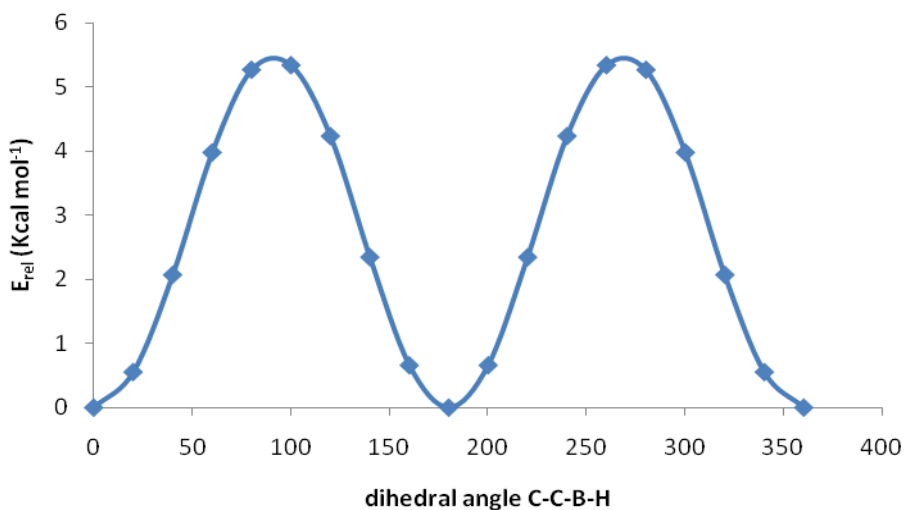


Figure S5: Potential energy surface for 2-chlorophenylborane (**4**).

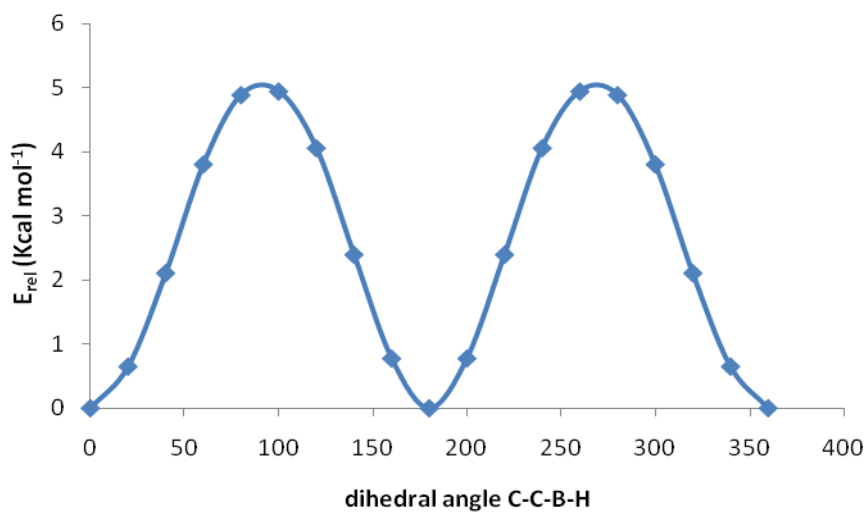


Figure S6: Potential energy surface for 2-bromophenylborane (**5**).

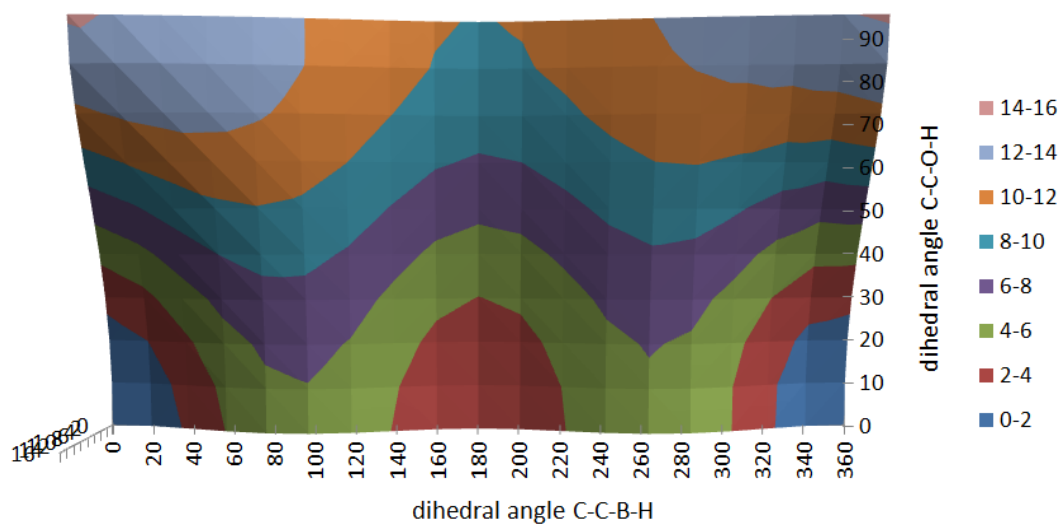


Figure S7: Potential energy surface for 2-hydroxyphenylborane (6).

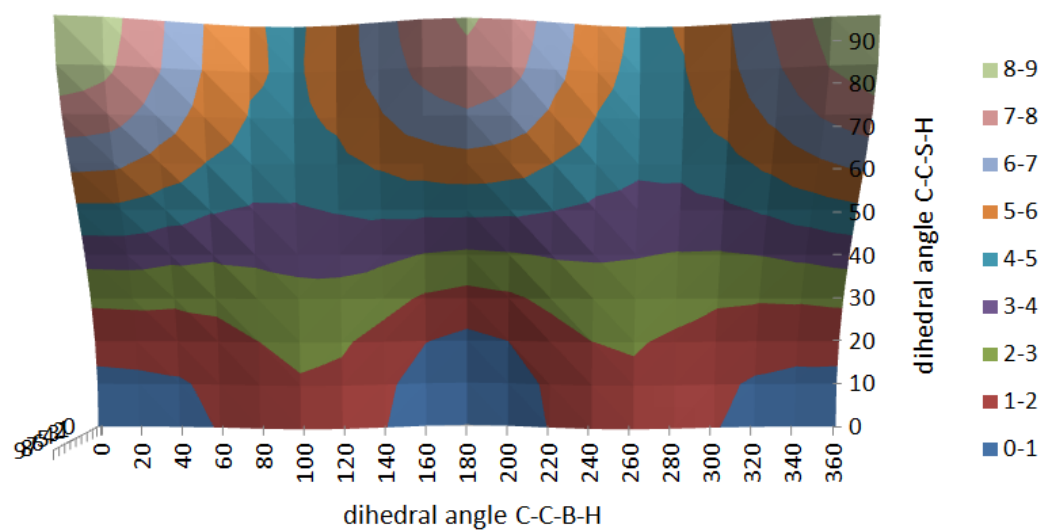


Figure S8: Potential energy surface for 2-sulfanylphenylborane (7).

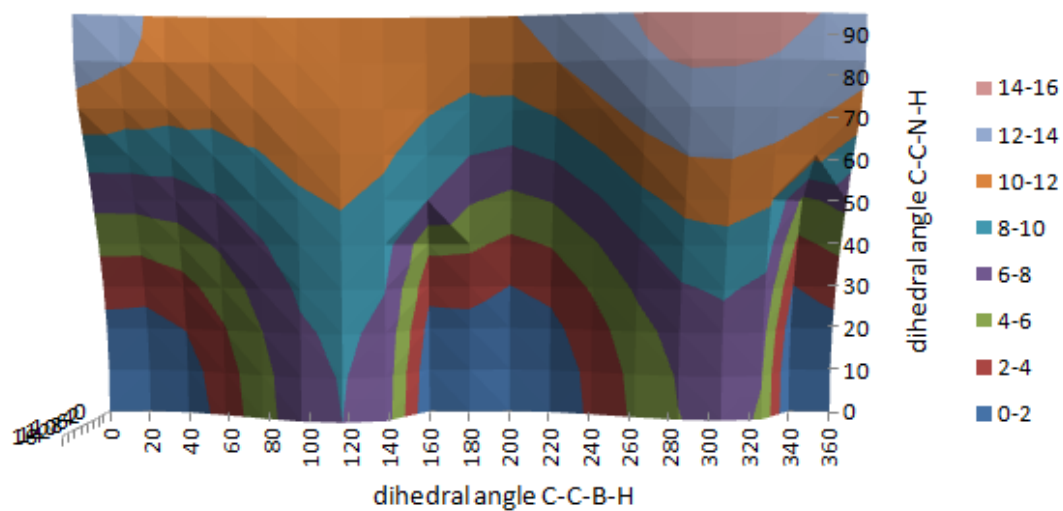


Figure S9: Potential energy surface for 2-aminophenylborane (8).

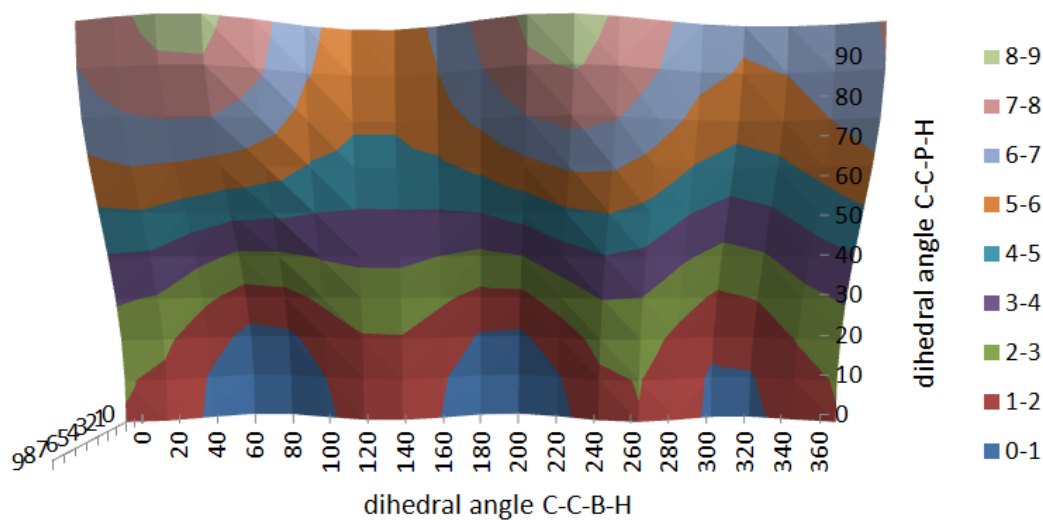


Figure S10: Potential energy surface for 2-phosphanylphenylborane (9).