

Supporting Information

Revealing the formation mechanism and band gap tuning of Sb₂S₃ nanoparticles

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1. Additional electron microscopy images

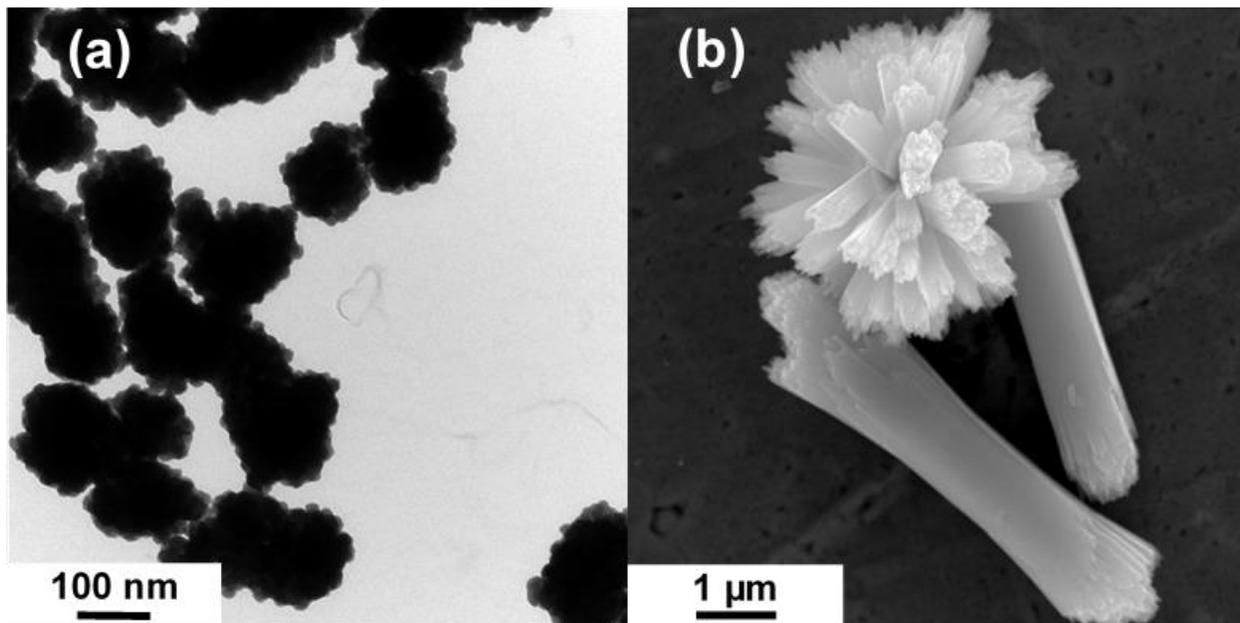


Figure S1: EM images of Sb_2S_3 nanoparticles received after different reaction times: (a) 10 min (TEM) and (b) 18 h (SEM)

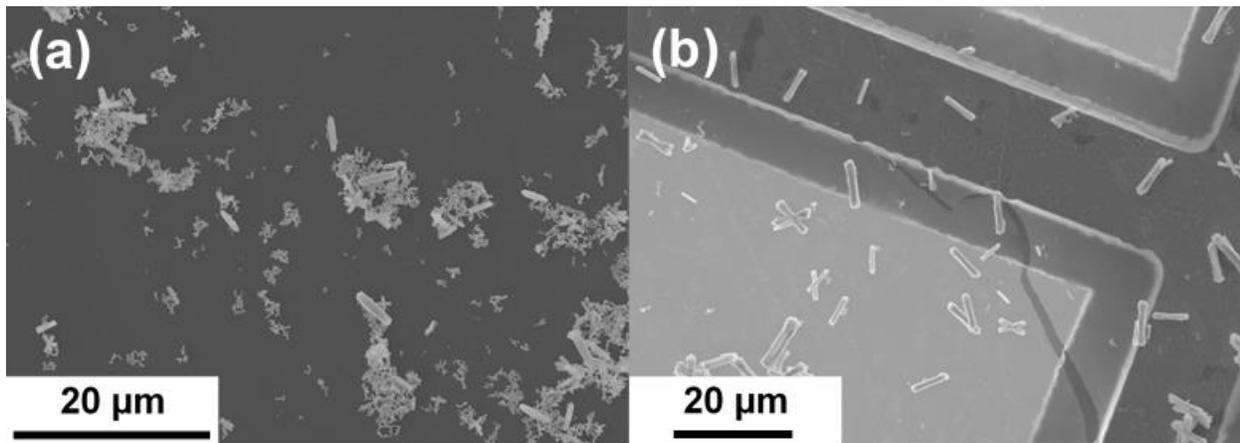


Figure S2: SEM images of Sb_2S_3 nanoparticles received after different reaction times: (a) 12 h and (b) 16 h. While there are a lot of small, spherical particles and little rods visible after 12 h, this is vice versa after 16 h

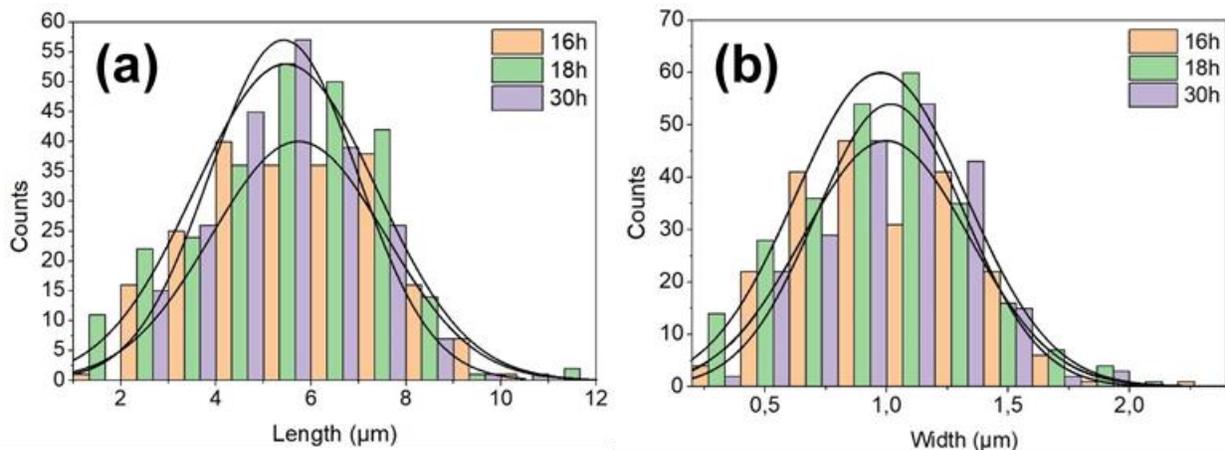


Figure S3: Histograms of the size distribution for the particles obtained after 16, 18, and 30 h for (a) length and (b) width, respectively. The size distribution curves were calculated assuming a Gaussian distribution.

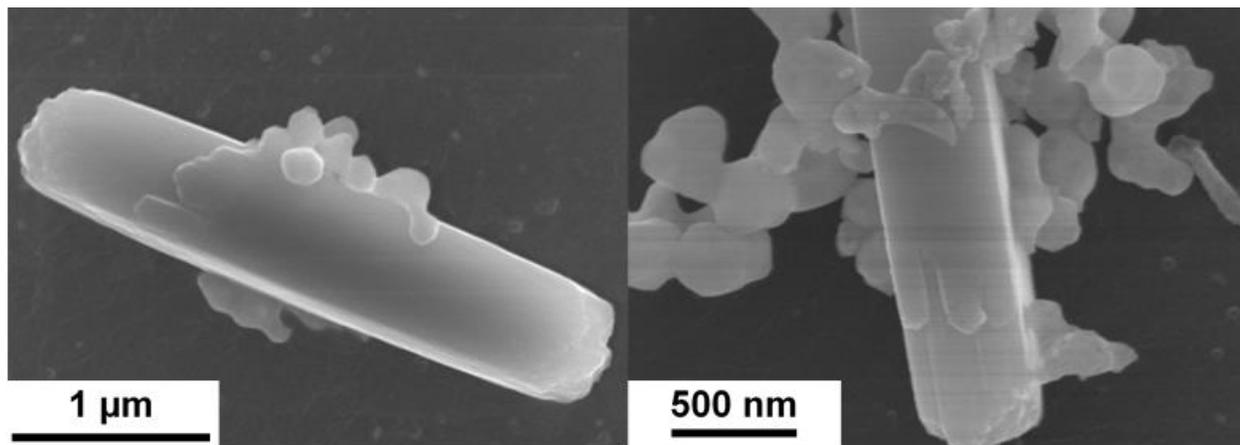


Figure S4: EM images of Sb_2S_3 nanoparticles: Layered growth after 12 h reaction time

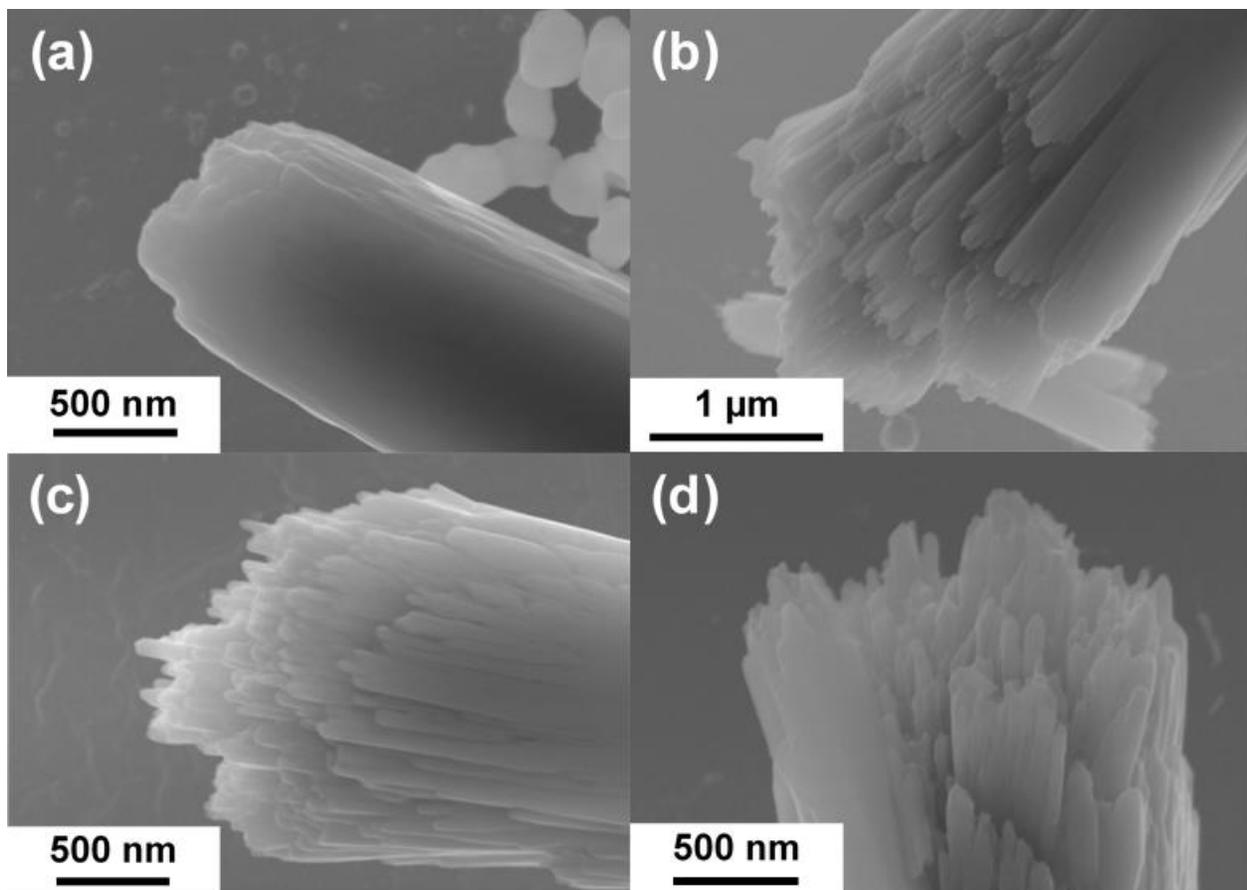


Figure S5: EM images of Sb_2S_3 nanoparticles: bristle-like tips after (a) 12 h, (b) 16 h (tip width: 40 ± 15 nm), (c) 18 h (tip width: 70 ± 15 nm), and (d) 30 h (top width: 80 ± 20 nm)

2. EDX data supporting the chemical composition

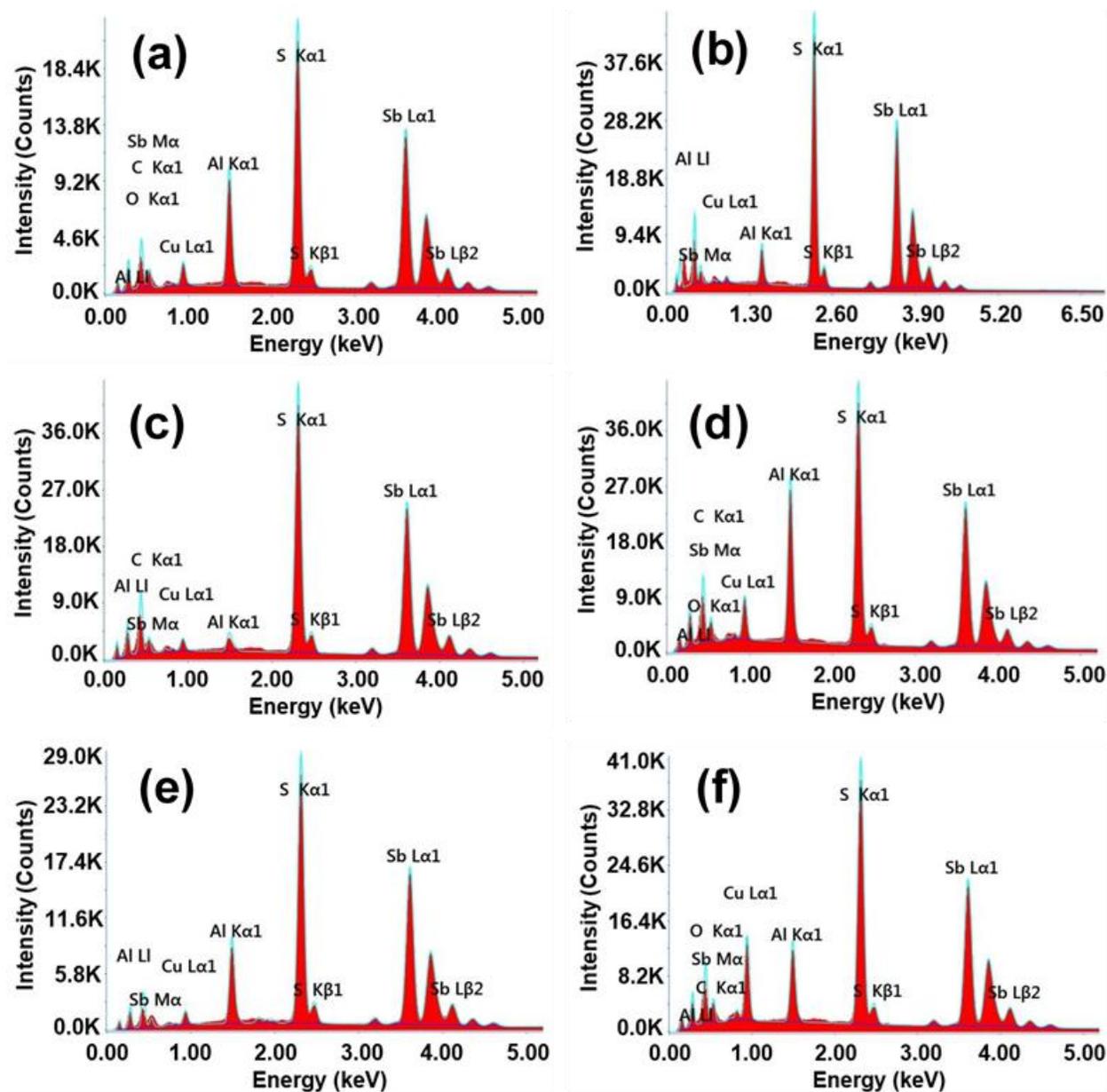


Figure S6: EDX data of Sb_2S_3 samples after different reaction times: (a) 10 min, (b) 30 min, (c) 12 h, (d) 16 h, (e) 18 h, and (f) 30 h. The samples were measured on carbon-coated copper grids placed on an aluminum holder, so these elements are found in all samples. Traces of oxygen may have been detected due to contamination or slight oxidation of the grid or the holder. The atomic ratios of Sb:S are given in Tab. 1.

3. Additional synthesis information

Table S1: Centrifugation speed and duration of the three centrifugation steps for the samples obtained after different reaction times

Reaction time	Step 1		Step 2		Step 3	
	Acceleration (g)	Duration (min)	Acceleration (g)	Duration (min)	Acceleration (g)	Duration (min)
<i>2 min</i>	150	5	250	10	2500	20
<i>5 min</i>	100	5	250	5	2500	10
<i>10 min</i>	100	5	250	5	2500	10
<i>30 min</i>	100	5	150	5	1000	10
<i>12 h</i>	50	5	100	5	500	5
<i>16 h</i>	50	5	100	5	500	5
<i>18 h</i>	50	5	100	5	500	5
<i>30 h</i>	50	5	100	5	500	5

For the centrifugation, 20 mL dispersion was filled in a 50 mL centrifugal tube. Subsequently, 30 mL isopropyl alcohol was added to precipitate the particles. Afterward, the mixture was centrifuged.