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1 **Nanostructure Mediated Enhancement of Antibacterial Activity of**
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3 **Morphological Changes Occuring Therein**

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32 **Abstract:**

33 *Staphylococcus aureus* is deliberated as one of the most challenging bacteria owing to its ability to develop
34 resistance against antibacterial drugs. In an attempt to explore new approaches for enhancing the activity
35 of antibiotics, here in this work, ampicillin is conjugated to Ag and Au nanoparticles (NPs) and its
36 antibacterial potential was investigated against *S. aureus*. The antibacterial activity was assessed and the
37 associated changes in the bacterial cell morphology were analyzed using atomic force microscopy (AFM)
38 as well as other characterization techniques. Results showed that the antibacterial activity of ampicillin
39 conjugated to gold and silver NPs was enhanced up to 10 and 5 times respectively, when compared with
40 the non-conjugated antibiotic. The kinetics of the conjugated ampicillin were improved. Bacterial
41 membrane destruction was scarcely evident after treating a cell culture with pure ampicillin for four hours.
42 However, Ag conjugates have severely disrupted the cell membranes and Au conjugates have completely
43 destroyed the cell morphology. The study gave an insight of the enhanced antimicrobial action of ampicillin
44 and can be exploited for the devising nanoparticle's based antimicrobial agents. More sophisticated
45 approaches such as faster and more efficient diagnostics, non-antimicrobial methodologies to prevent and
46 treat infections and a better understanding of staphylococcal pathogenesis will also be required to forestall
47 the future of the bacterial resistance.

48 **Keywords:** Bacterial resistance, ampicillin, antibacterial activity, Ag and Au nanoconjugates,
49 AFM, cell morphology

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60 **Introduction**

61 Nanotechnology has attracted significant attention because of the unique characteristics
62 and increasing importance of nanomaterials in various fields especially in nanomedicine [1]. Their
63 uniqueness is due to high surface area and more atoms at the particle boundaries. Among the
64 different metallic NPs, silver and gold NPs have comprehensive range of uses in nano-scale
65 strategies and tools due to their chemical inertness [2-5]. The worldwide increase in bacterial
66 resistance to existing medicines is a long-standing problem for human health. Bacterial resistance
67 to antimicrobial drugs has increased due to the irrational use of antibiotics, thus creating problems
68 in the treatment of bacterial infections. The development and spread of resistance to antibiotics
69 has compromised the clinical efficacy of currently existing antibiotics and highlighted the need for
70 new antibacterial compounds [6]. β -Lactam antibiotics are the most widely used antibiotics for
71 their effectiveness and safety profile, however occurrence of new, more antagonistic β -lactamases
72 has reached the point where several marketed β -lactams are no longer clinically effective [7].
73 Therefore, immediate approaches are needed to develop new antimicrobial drugs to handle this
74 problem. This has evoked a solid reaction from health consultants, who have implemented
75 initiatives to inspire the discovery of new antibiotics. One of the capable approaches for restricting
76 bacterial resistance is the application of metallic NPs as a powerful nano-weapon against multidrug
77 resistant bacteria [8-10], because metallic NPs has the ability to target several bacterial structures
78 [11]. There is a mounting evidence that the synergistic effect of antibiotics and NPs resulted in an
79 increase in antibacterial activity of antibiotics [12-16] and gold and silver alloy NPs, bound to
80 antibiotics displayed enhanced antibacterial potential [17]. The Ag NPs of antibiotics including
81 penicillin, vancomycin and amoxicillin, exhibited increased antibacterial activity against *S. aureus*
82 and *E. coli* [18]. Our previous work has shown that the antibacterial effect of ceftriaxone against
83 *E. coli* can be enhanced up to six times through conjugation with silver and gold NPs [19]. These
84 findings are very important because such potent antibiotics can be made active in comparatively a
85 small amount to treat infections, thereby decreasing side effects and minimizing the problem of
86 drug resistance. In this paper, we present the enhancement of the antibacterial potential of
87 ampicillin via conjugation to Au and Ag NPs. We have also explored the antibacterial action of

88 these nanoconjugates against *S. aureus* bacteria under atomic force microscope (AFM), which
89 enabled us to obtain detailed and exciting close-up images of the nanoconjugates involved in
90 various stages of antimicrobial actions. AFM is an appropriate tool for the study of living samples
91 and a distinct vantage is that samples can be analyzed without fixation, vacuum and conductive
92 coating. This technique is extremely efficient in getting images of tiny, highly fragile structures of
93 bacteria, morphological changes suggestive of antibacterial activity [20-24] and a further detailed
94 perception in the structure and mechanics of living specimens [25-26].

95 **Results and discussion**

96 The morphological analysis and mechanism of action of the antimicrobial activity of ampicillin
97 conjugated with AgNPs (Mpn-AgNPs) and AuNPs (Mpn-AuNPs) on *staphylococcus aureus* using
98 AFM was studied for the first time. *S. aureus* is a sensitive strain of bacteria that infect humans
99 and can cause respiratory diseases, food poisoning and skin infections [27]. *S. aureus* is notorious
100 for its capability to develop *resistance* to antibiotics and has created a worldwide problem in
101 clinical treatment [28]. Ampicillin was capped with Ag and Au NPs by mixing its aqueous solution
102 with ionic solutions of Ag and Au in the presence of triethylamine as a reducing agent. UV-visible
103 spectroscopy was used to monitor the conjugation of ampicillin with Ag and Au NPs. The UV-
104 visible spectra of the Mpn-AgNPs and Mpn-AuNPs exhibited surface plasmon bands (SPB) at 396
105 nm and 540 nm, respectively (**Fig. 4**), which can be correlated with the typical plasmonic
106 absorption of Ag and Au NPs [29-30]. The conjugation of ampicillin with Ag and Au NPs was
107 further confirmed by FT-IR spectroscopy (**Fig. 5**). The FTIR spectrum of ampicillin exhibited
108 absorption bands in region 3512 cm^{-1} and 3205 cm^{-1} which could be associated with stretching
109 vibrations of O–H and N–H groups, respectively. The band at 2968 cm^{-1} can be assigned to the
110 stretching vibrations of C–H groups, carbonyl group of the lactame ring showed the stretching
111 vibration at 1774 cm^{-1} and the amide carbonyl group exhibited band at 1688 cm^{-1} . The band at
112 1372 cm^{-1} could be assigned to the stretching vibrations of C–N of the lactame and thiazole.

113 The conjugation of ampicillin with Au and Ag NPs result in the decrease in absorbance intensities
114 and merging of bands of O–H (3512 cm^{-1}), N–H (3205 cm^{-1}) and C=O (1774 and 1688 cm^{-1})
115 stretching [31]. Ag and Au NPs were then characterized by AFM and their size were found to be
116 around 15-50 nm (**Fig. 6**).

117 The aim of this study was to examine the boosted antibacterial action and kinetics of the ampicillin
118 Ag and Au NPs through AFM against *S. aureus*, which has not yet been explored. The
119 membranolytic properties in the mechanisms of action of the antibiotics ampicillin, magainin and
120 human platelets extract have been studied by using *Bacillus cereus* and *Escherichia coli* as the
121 bacterial targets [32]. Similarly chitosan NPs of ampicillin trihydrate were synthesized and claimed
122 that they would be capable of sustained delivery of ampicillin [33]. Another study is based on
123 functionalized ampicillin with Ag and Au NPs and their antimicrobial activity against different
124 bacterial strains by determining their minimum bactericidal concentration (MBC) [34]. This paper
125 is offering the first description on visualizing the effect of ampicillin and its Ag and Au NPs on *S.*
126 *aureus* by AFM. The minimum inhibitory concentrations (MICs) of ampicillin and its Au and Ag
127 NPs were determined through a zone of inhibition [35]. The MICs of pure ampicillin and
128 conjugated ampicillin were found to be 50 ± 0.1 , $60 \pm 0.3 \mu\text{g mL}^{-1}$ (which corresponds to a 10.8
129 μg ampicillin) and $75 \pm 0.3 \mu\text{g mL}^{-1}$ (which correspond to a 4.52 μg ampicillin), respectively.
130 While the MICs of bare Ag and Au NPs were calculated to be 85 ± 0.3 and $100 \pm 0.2 \mu\text{g mL}^{-1}$,
131 respectively (**Fig. 7**).

132 The MIC for unconjugated ampicillin is in agreement with the literature value [36]. Although the
133 MICs of Ag and Au conjugates were more than pure ampicillin, the conjugates contain only a
134 small weight fraction of the ampicillin (18 % for Mpn-AgNPs and 6.03% for Mpn-AuNPs), which
135 specifies that ampicillin conjugated to Ag and Au NPs is about 5 and 10 times more active than
136 pure ampicillin, respectively. Further confirmation was carried by AFM which explored the more
137 persuasive and rapid action of the conjugates. Morphological characterization of the control *S.*
138 *aureus* samples showed typically round cells with normal shapes and flat membranes with a mean
139 length of 1.052 μm , mean width of 1.082 μm and mean height of 0.104 μm and with a maximum
140 height of 0.719 μm , as shown in **Fig. 8**. Bacterial cultures were then treated with pure ampicillin,
141 its Ag and Au conjugates and bare Ag and Au NPs to study the comparative action and kinetics
142 under AFM. Bacteria treated with MIC dose of unconjugated ampicillin for 1 hour showed slight
143 effect and only small lesions were seen on bacterial cell surface (**Fig. 9a**). Cell Morphological
144 degradation increased with time as a 2 hours treatment have further affected bacterial cells and
145 after 4 hours considerable damages of cell bodies were observed (**Fig. 10a, 11a**). After 8 hours
146 time period the cell morphologies were completely degraded and distorted (**Fig. 12a**). On the other

147 hand bacterial cultures treated with MIC dose of Mpn-AgNPs for 1 hour and 2 hours were found
148 to affect the cells more than pure ampicillin (**Fig. 9b, 10b**) with complete destruction of bacterial
149 cells after 4 hours treatment (**Fig. 11b**). A relatively stronger effect was observed in case of Mpn-
150 AuNPs of MIC dose on the bacterial cells in 1 hour and 2 hours treatment (**Fig. 9c, 10c**), and a
151 complete rupture of bacterial cells in 4 hours (**Fig. 11c**). Unconjugated Ag and Au NPs of MIC
152 doses did not show any observable effect but only minimal morphological changes and only a very
153 slight influence was observed even after treatment for 8 hours (**Fig. 12b, c**).

154 The interaction of NPs with a bacterial cell still needs further exploration, however many studies
155 have shown that at first metal NPs adsorb to surface of a microorganism due to resultant
156 electrostatic pressure and high affinity of metals towards Sulphur in the proteins [37]. After that,
157 NPs get inside into the cell causing perforations and lead to the release of the cellular matrix [38-
158 40]. Here in this case ampicillin reacts with the outer peptidoglycan layer of *S. aureus* thereby
159 enhancing the membrane's permeability. Subsequently the NPs get into the cells through
160 membranes and may be attached to the bacterial DNA and protein; thus, causing death of the cells
161 by disturbing metabolism and vital functions [41-43]. Consequently, the mutual action of
162 ampicillin and Ag and Au NPs lead to enhanced antibacterial potential [44] Transmission Electron
163 Microscopy was used for studying the antibacterial potential of silver NPs against *E. coli* [38], but
164 it represented *E. coli* when they were lifeless. Here in this study AFM explored noticeable
165 investigation of *S. aureus* by providing a thorough topographic demonstration of shape, surface
166 and phase imaging morphology that allowed analyses of height, width, length and boundary
167 stiffness.

168 **Conclusion**

169 Ampicillin was conjugated with Ag and Au NPs and were characterized by UV-visible,
170 FT-IR and AFM. The NPs were found to be very stable. The antibacterial potential of the
171 synthesized NPs was studied against *S. aureus* and it was found that conjugated ampicillin
172 exhibited antibacterial activity 5-10 times higher than the free drug. The kinetics and
173 morphological changes in the bacterial cell were studied under AFM. The study gave an insight of
174 the enhanced antimicrobial action of ampicillin and can be exploited for the devising
175 nanoparticle's based antimicrobial agents. More sophisticated approaches such as faster and more

176 efficient diagnostics, non-antimicrobial methodologies to prevent and treat infections and a better
177 understanding of staphylococcal pathogenesis will also be required to forestall the future of the
178 bacterial resistance.

179

180 **Experimental section**

181 *Materials*

182 Silver nitrate (AgNO_3) and Tetrachloroauric acid trihydrate ($\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$) was purchased
183 from Merck, triethylamine (TEA) from Scharlau and ampicillin (Mpn) were supplied by
184 Pharmagen Limited, Lahore, Pakistan. *Staphylococcus aureus* ATCC 11632 (provided by H.E.J.
185 Research institute of Chemistry (ICCBS), University of Karachi, Karachi Pakistan was used to
186 evaluate the antibacterial activity of ampicillin and its silver and gold nano-conjugates. We used
187 deionized water throughout experiment for the synthesis of NPs and further analysis.

188 *Synthesis of silver NPs stabilized with Ampicillin (Mpn-AgNPs)*

189 Solution of ampicillin (1 mM) and AgNO_3 (1 mM) were prepared in deionized water. These
190 two solutions were mixed using optimized ratio (9:1 Ag:ampicillin mole ratio). The reaction
191 mixture was stirred for 30 minutes and then 0.1 mL of triethylamine was added to it. The color of
192 the reaction mixture turned to yellowish red; the reaction was carefully monitored through UV-
193 visible spectroscopy. The reaction mixture was stirred for 2 hours then the suspensions were
194 centrifuged to collect NPs. Unreacted precursors and reaction by-products were removed by
195 washing the NPs repeatedly.

196 *Synthesis of gold NPs stabilized with Ampicillin (Mpn-AuNPs)*

197 1 mM solution of $\text{HAuCl}_4 \cdot 3\text{H}_2\text{O}$ and a 1 mM solution of ampicillin were prepared in
198 deionized water. These two solutions were mixed using optimized ratio (12:1 Au:ampicillin mole
199 ratio). The reaction mixture was stirred for 30 minutes and then 0.1 mL of triethylamine was added
200 to it. The reaction start immediately and colorless reaction mixture turned to purple red; the
201 reaction was monitored by UV-visible spectroscopy. The reaction mixture was stirred for 2 hours

202 and the suspensions were centrifuged to collect NPs. Unreacted precursors and reaction by-
203 products were removed by washing the NPs repeatedly.

204 *Characterization*

205 The synthesized ampicillin Ag and Au conjugates were characterized by UV-vis
206 spectroscopy; the spectra were collected by a Thermo Scientific Evolution 300 spectrophotometer.
207 FT-IR spectra were acquired with a Bruker Victor 22 spectrophotometer. Finally the shape and
208 size of NPs were determined by AFM (AFM, Agilent Technologies 5500, USA). The instrument
209 was used in ACAFM mode. The samples were dried on freshly cleaved mica surface for analysis
210 at ambient temperature. Si cantilever of force constant 42 N/m, length 125 μm and resonance
211 frequency 330 KHz was maintained throughout the analysis.

212 *Quantification of the weight of ampicillin in the conjugates.*

213 A known volume of suspension was centrifuged and the precipitated NPs were collected.
214 The supernatant was repeatedly centrifuged to remove the synthesized NPs. The supernatant was
215 then freeze-dried, and the residues weighed. Using this method the ampicillin was estimated as 18
216 wt% for Ag NPs and 6.03 wt% for Au NPs conjugates.

217 *Stability of the NPs*

218 UV-visible spectroscopy was used to describe temperature, salinity and pH stability of the
219 suspensions. Coagulation is usually accompanied by color change and shift of the surface plasmon
220 towards longer wavelengths [45]. The Ag and Au conjugates of ampicillin were found to be stable
221 at 100°C and 50°C temperature, respectively (**Fig. 1**), in a 3-12 pH range (**Fig. 2**) and salt
222 concentration up to 50 mM (**Fig. 3**).

223 *Minimum Inhibitory Concentration (MIC) by Agar well diffusion method.*

224 To calculate MICs, the agar-well diffusion method was employed [46]. MICs for ampicillin
225 were measured with or without silver and gold NPs. In brief, nutrient agar was used as a medium
226 to grow a lawn of *S. aureus* ATCC 11632 at a concentration of 10^6 cells in one mL and duplicate
227 dilutions were used to calculate minimum inhibition zones. The 60 mm well was made by using a

228 borer. The 500 $\mu\text{g ml}^{-1}$ stock solution of ampicillin and its Ag & Au NPs were used to avoid
229 nonspecific merged zones of inhibition. In each well different amounts of various concentrations
230 ranging from 500-5 $\mu\text{g ml}^{-1}$ were added. The plates were incubated at room temperature for 2 hours
231 to allow the diffusion process to take place before it was incubated for 24-48 hours at $37\text{ }^{\circ}\text{C} \pm 1$.
232 The zones of inhibition were measured by using a millimeter scale.

233

234 ***Antibacterial activity and Morphological changes of Staphylococcus aureus under AFM***

235 *S. aureus* ATCC 11632 were grown on Tryptic soya agar (Oxoid UK) at $37 \pm 0.5\text{ }^{\circ}\text{C}$ for 24
236 hours in static condition and marked as stock *S. aureus* culture. On freshly cleaved mica slide, 10
237 μL drop(s) of polylysine was added and left to dry. Then, freshly incubated culture of *S. aureus*
238 on tryptic soya agar (Oxoid UK) inoculated in sterilized distilled water to make 10^6 cfu of *S. aureus*
239 and 5-10 μL droplets of this solution were transferred onto a freshly cleaved mica surface. The
240 sample was characterized by atomic force microscopy to check its morphology of bacterial cells.
241 MIC (50 μg) dose of ampicillin were added into test tubes of nutrient broth containing 10^6 cfu of
242 *S. aureus* bacteria and incubated it for 1-8 hours respectively at $37 \pm 0.5\text{ }^{\circ}\text{C}$ after incubation 5-10
243 μL drops of each dose transferred on freshly cleaved mica coated with polylysine separately and
244 left it for dry and was characterized by AFM. The same procedure was applied for Ampicillin
245 conjugated with AgNPs, MIC (60 μg) dose was treated with 10^6 cfu of *S. aureus* for 1, 2, and 4
246 hours respectively and were characterized by AFM to check the cell changes and noted the effects
247 of these conjugates. On the other hand Mpn-AuNPs (75 μg) were treated with 10^6 cfu of *S. aureus*,
248 and incubated at $37 \pm 0.5\text{ }^{\circ}\text{C}$. 5-10 μL of this suspension was transferred on freshly cleaved mica
249 coated with polylysine and left it for dry and then was characterized by atomic force microscopy.
250 On this way we recorded, control, treated with ampicillin, ampicillin conjugated with Ag and Au
251 NPs and bare Ag & Au NPs images of *S. aureus* in similar condition using AFM (AFM, Agilent
252 Technologies 5500, USA) in the ACAFM mode. We used high frequency Si cantilever having
253 length of 125 μm , force constant 42 N/m and resonance frequency 330 KHz. All samples were
254 prepared and analyzed in a same condition.

255

256 **Abbreviations**

257 Nanoparticles (NPs)
258 Atomic force microscopy (AFM)
259 Silver nitrate (AgNO₃)
260 Tetrachloroauric acid trihydrate (HAuCl₄.3H₂O)
261 Triethylamine (TEA)
262 Ampicillin (Mpn)
263 NPs stabilized with Ampicillin (Mpn-AgNPs)
264 Fourier-transform infrared (FTIR)
265 Silver nanoparticles (Ag NPs)
266 Gold nanoparticles (Au NPs)
267 Minimum inhibitory concentration (MIC)

268 **Declarations**

269 **Ethics approval and consent to participate**

270 Not applicable

271 **Consent for publication**

272 Not applicable

273 **Availability of data and material**

274 All datasets on which the conclusions of the manuscript rely are presented in the paper.

275 **Authors' contributions**

276 MA, AK and AA supervised and designed the study. SA and SP performed all experiments. SA
277 and MRS were analyzed data. AL and MA were involved in writing, editing of manuscript. All
278 authors have read and approved the final version of the manuscript.

279 **Conflicts of interest:**

280 The authors declared that there is no conflict of interest.

281

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414 **Figures caption**

415 **Fig. 1:** Heat stability of Mpn-AgNPs (A) and Mpn-AuNPs (B)

416 **Fig. 2:** PH stability of Mpn-AgNPs (A) and Mpn-AuNPs (B)

417 **Fig. 3:** Salt stability of Mpn-AgNPs (A) and Mpn-AuNPs (B)

418 **Fig. 4:** UV-visible spectrum of Mpn-AgNPs (A) and Mpn-AuNPs (B)

419 **Fig. 5:** FT-IR spectra of Mpn-AgNPs (A) and Mpn-AuNPs (B)

420 **Fig. 6:** AFM images of Mpn-AgNPs (A) and Mpn-AuNPs (B)

421 **Fig. 7:** Minimum inhibitory concentration of Ampicillin (1), Mpn-AgNPs (2) Mpn-AuNPs (3)
422 bare AgNPs (4) and bare Au NPs (5)

423 **Fig. 8:** AFM images of *S. aureus* before treatment (control), Topography (A), 3D (B)

424 **Fig. 9:** AFM images of *S. aureus* treated for 1h with (A) ampicillin (B) Mpn-AgNPs (C) Mpn-
425 AuNPs

426 **Fig. 10:** AFM images of *S. aureus* treated for 2h with (A) Ampicillin (B) Mpn-AgNPs and (C)
427 Mpn-AuNPs

428 **Fig. 11:** AFM images of *S. aureus* treated for 4h with (A) ampicillin (B) Mpn-AgNPs and (C)
429 Mpn-AuNPs

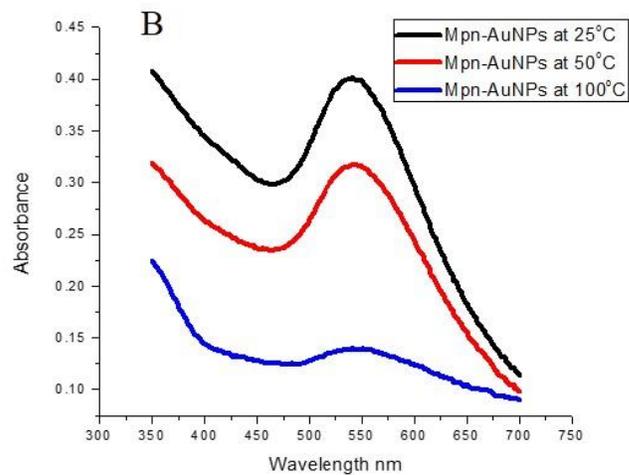
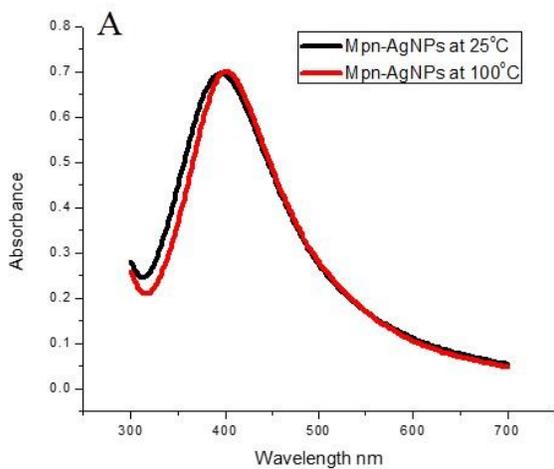
430 **Fig. 12:** AFM images of *S. aureus* treated for 8h with (A) ampicillin (B) bare AgNPs and (C)
431 bare AuNPs

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439 **Figure-1**

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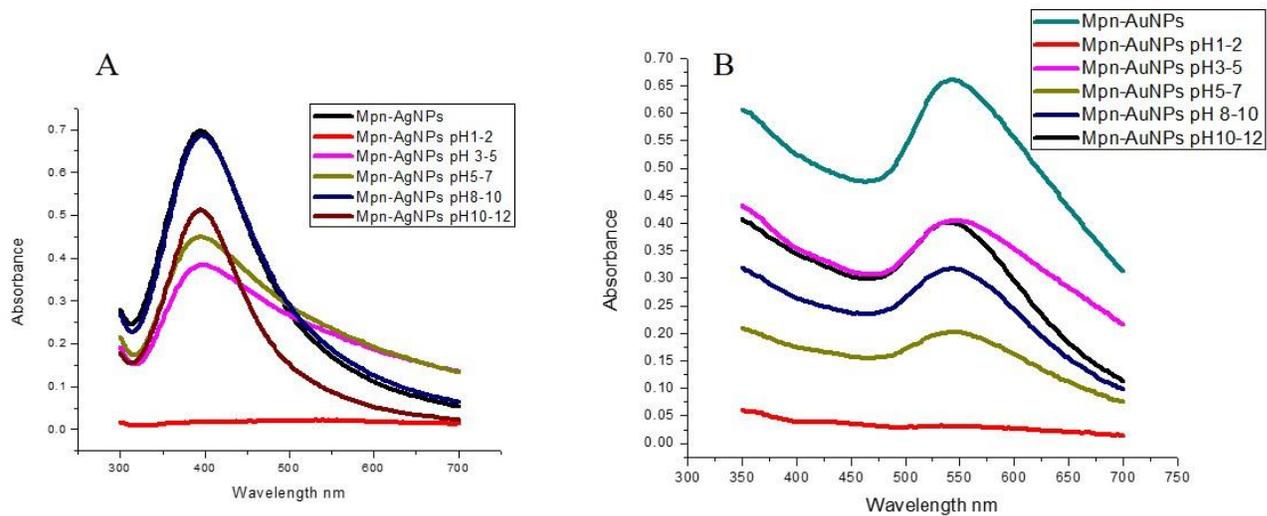
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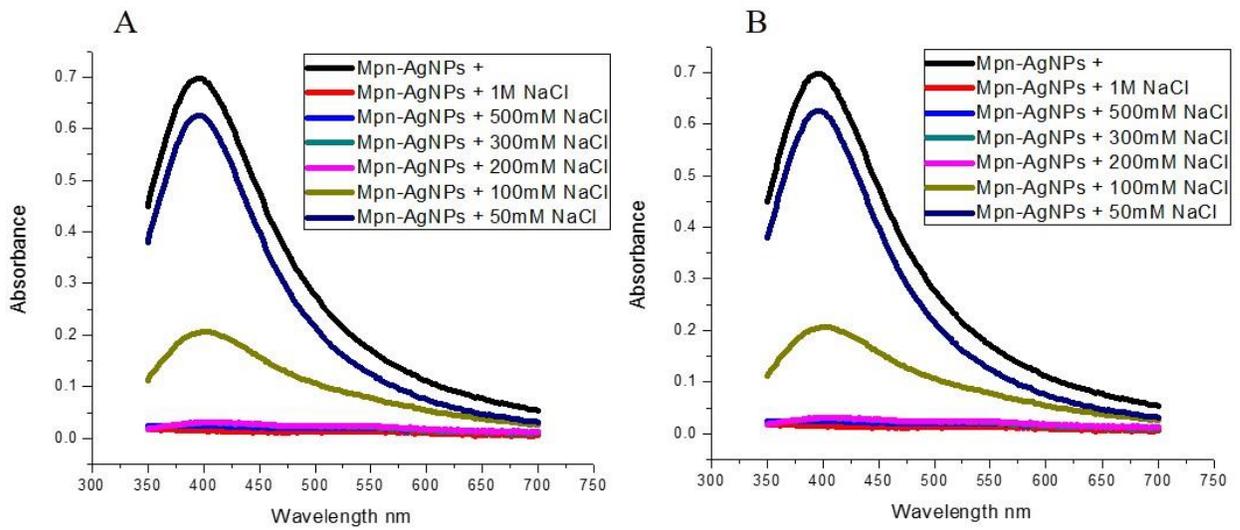
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460 **Figure-2**

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478 **Figure-3**

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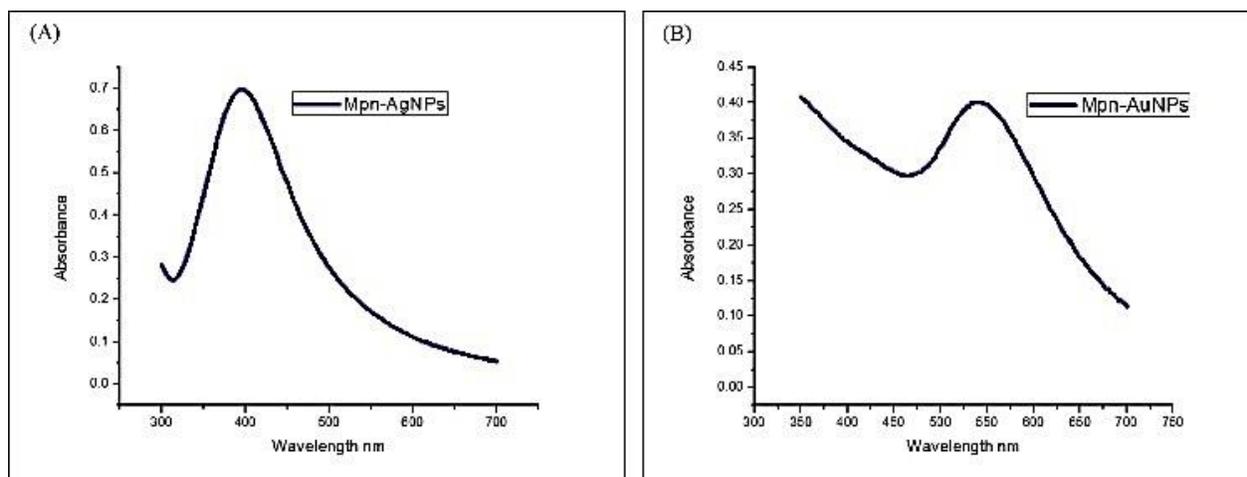
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497 **Figure-4**

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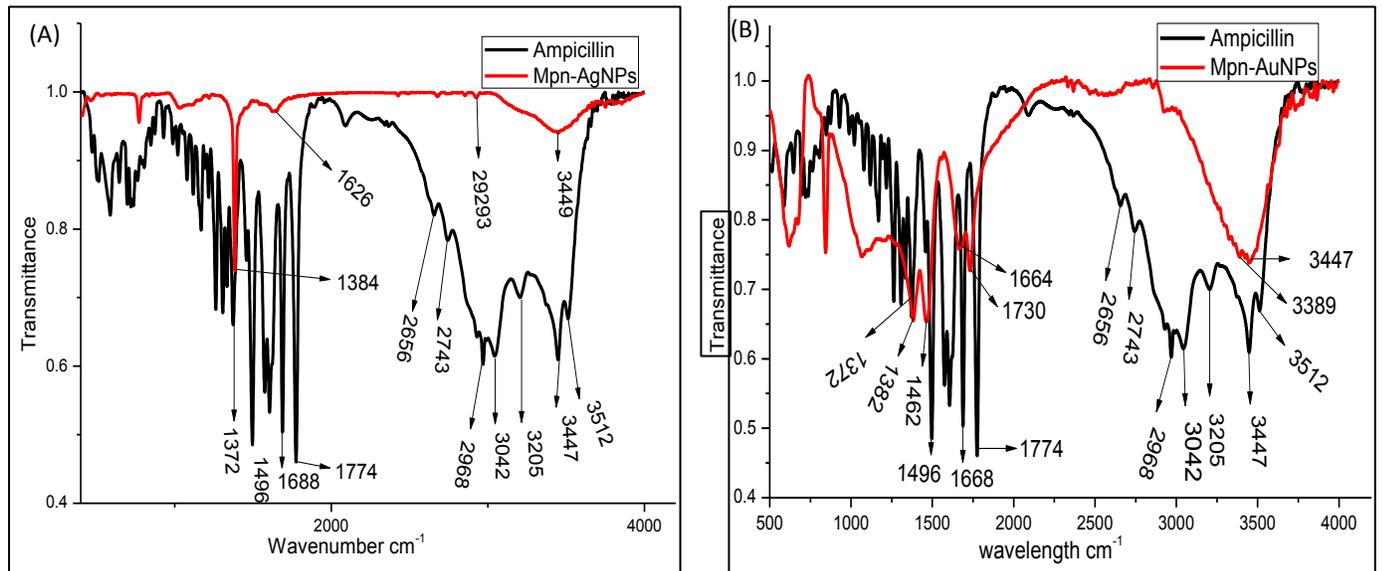
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520 **Figure-5**

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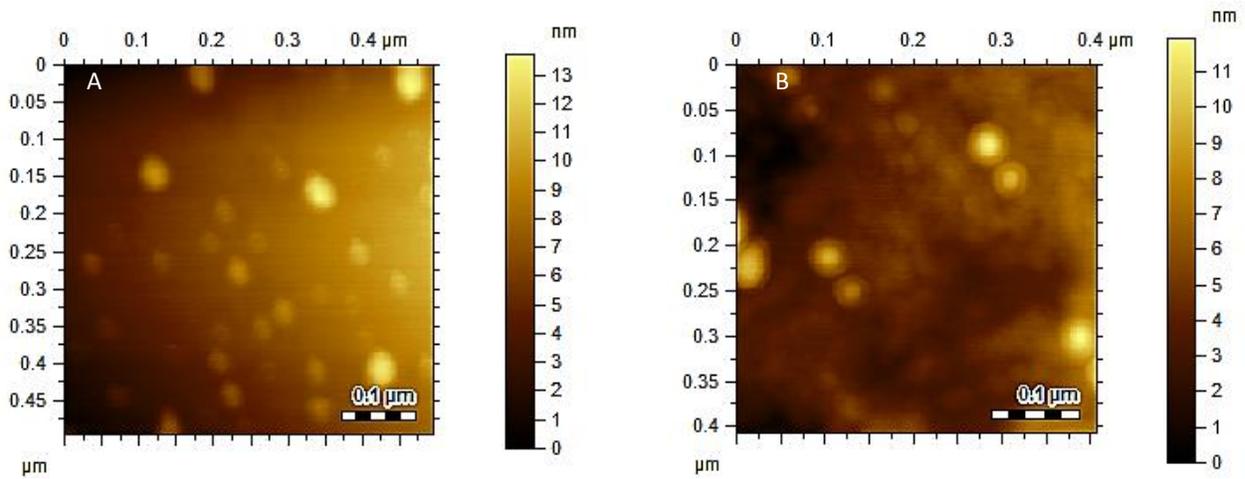
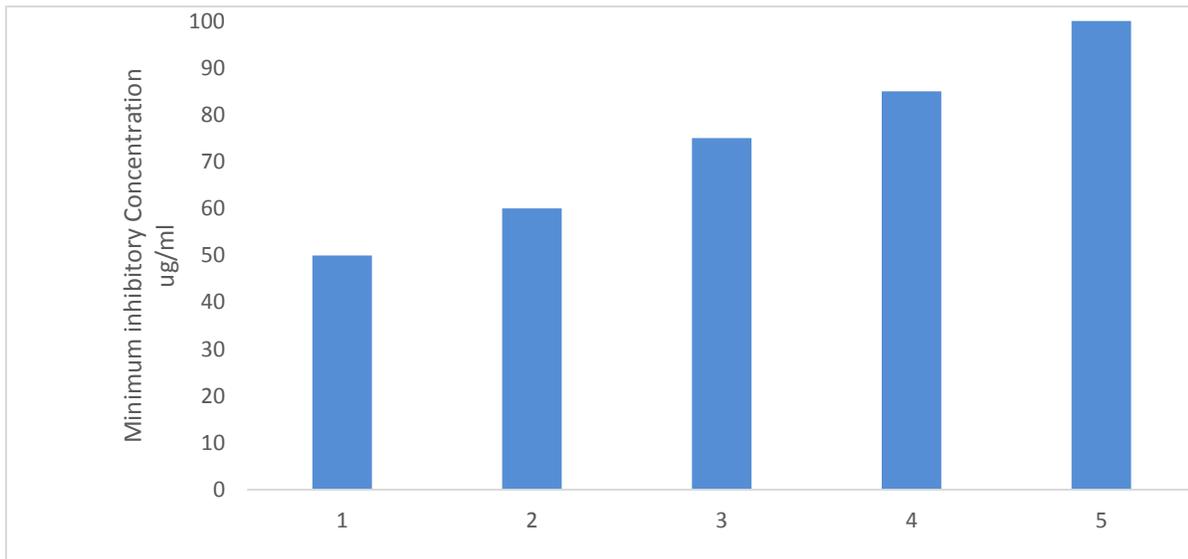


Figure-6

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Figure-7

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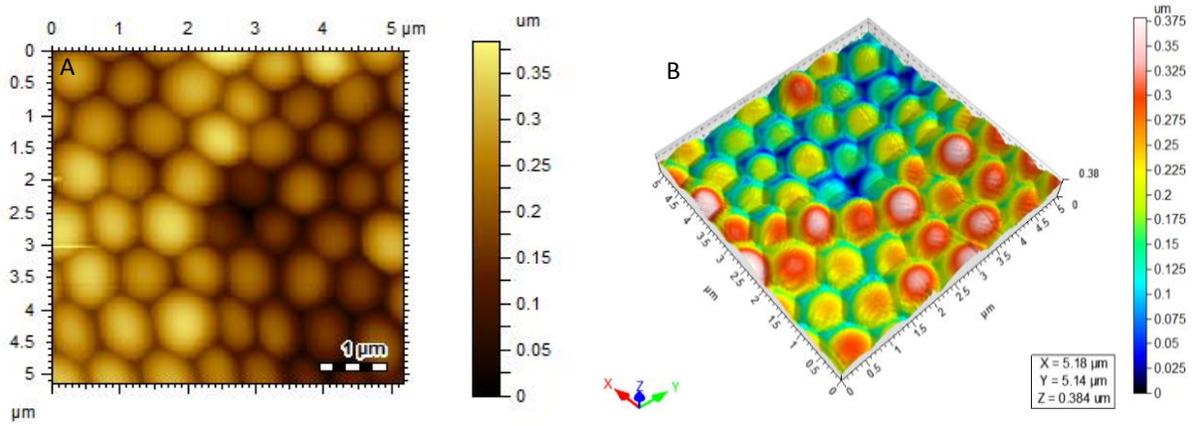


Figure-8

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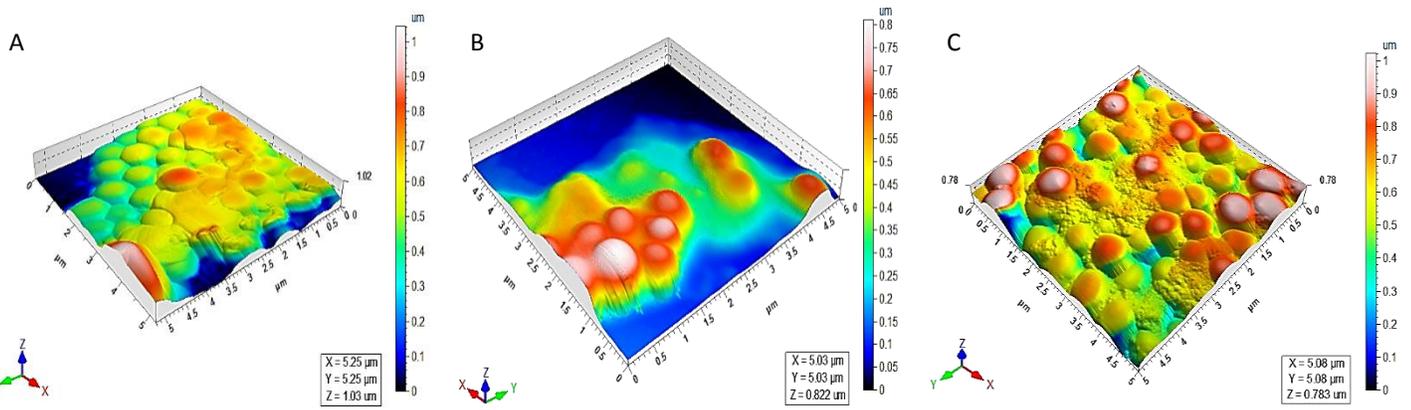


Figure-9

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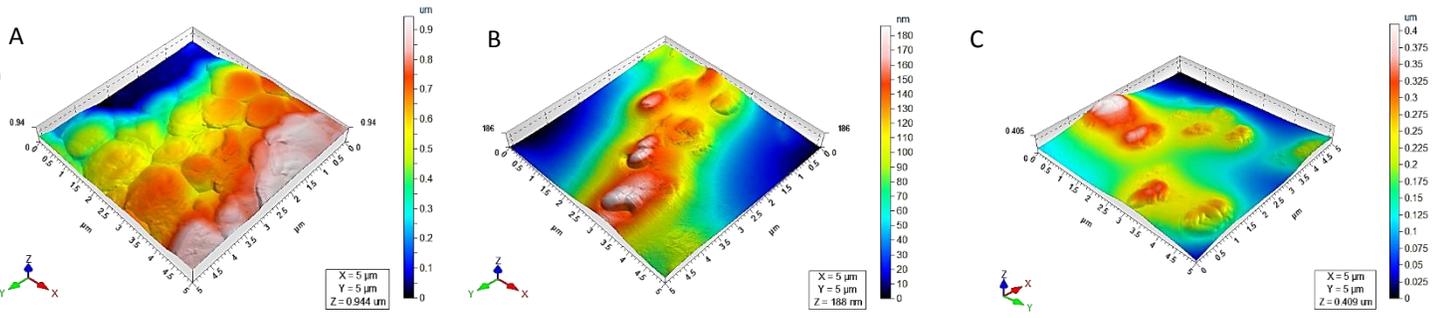


Figure-10

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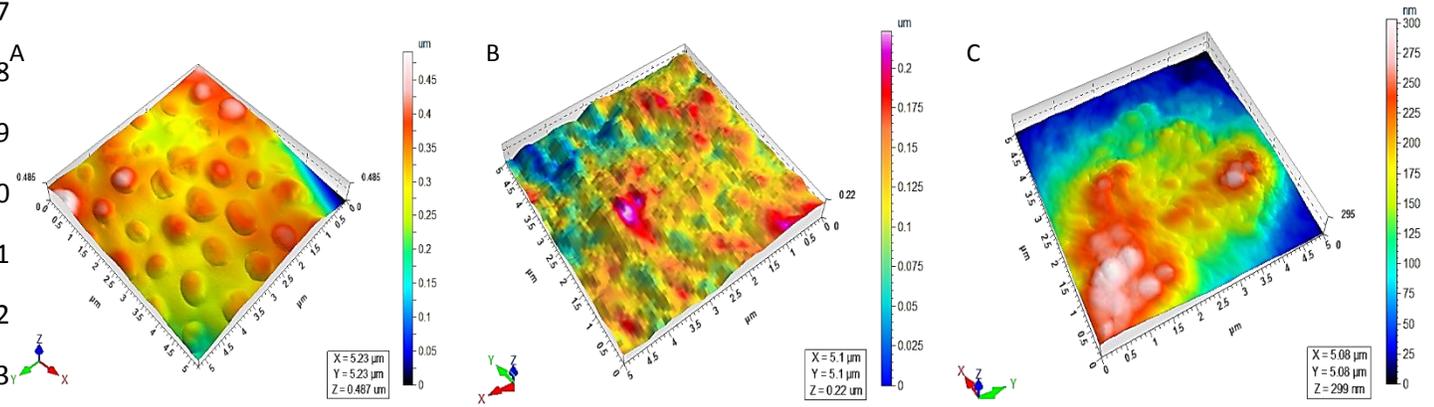


Figure-11

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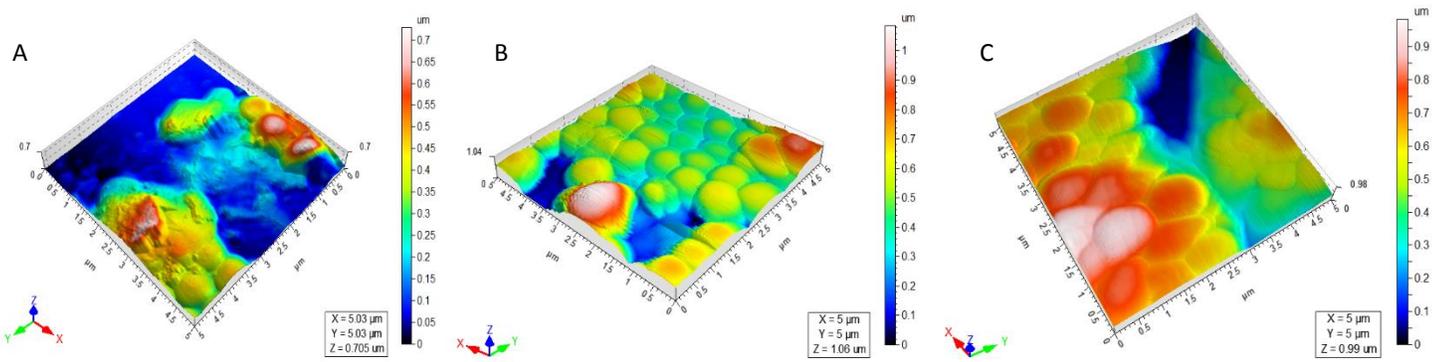


Figure-12