



## Supporting Information

for

### Sputtering onto liquids: a critical review

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### Sputtering onto liquids: Table with experimental parameters

Experimental parameters	Acronyms
<b>Target:</b> target material	EMIM – 1-ethyl-3-methylimidazolium
<b>Liquid:</b> host liquid	TMPA – N,N,N-trimethyl-n-propylammonium
<b>Amount of liquid substrate:</b> volume or weight of the host liquid plus parameters of the vessels	BMIM – 1-butyl-3-methylimidazolium
<b><math>d_t, \text{cm}</math>:</b> diameter of the target, centimeter	HyEMIM – 1-(2-hydroxyethyl)-3-methylimidazolium
<b>WD, cm:</b> working distance, centimeter	(BCN)MIM – 1-butyronitril-methylimidazolium
<b><math>p, \text{Pa}</math>:</b> pressure, Pascal	(MeOE)MIM – 1-methoxyethyl-methylimidazolium
<b><math>P, \text{W}</math>:</b> power, Watt	(HSE)MIM – 1-thioethyl-methylimidazolium
<b><math>V, \text{V}</math>:</b> voltage, volt	PP13 – 1-methyl-1-propylpiperidinium
<b><math>I, \text{mA}</math>:</b> current, millampere	TFSI – bis(trifluoromethanesulfonyl)imide (=NTf <sub>2</sub> )
<b><math>t_s, \text{min}</math>:</b> sputter time, minute	OTf – trifluoromethanesulfonate
<b><math>T_s, {}^\circ\text{C}</math>:</b> temperature, degree Celsius	FSA – bis(fluorosulfonyl)amide
<b><math>d_{NP}, \text{nm}</math>:</b> diameter of nanoparticles according to TEM, nanometer	OTD – 1-octadecane-thiol
<b>Ref.:</b> reference	PEMP – pentaerythritol tetrakis(3-mercaptopropionate)
	PVP – polyvinylpyrrolidone
	PEEL – pentaerythritol ethoxylate
	PEG – polyethylene glycol
	SMP – 3-mercaptopropionate
	MUA – 11-mercaptoundecanoic acid
	DES – deep eutectic solvents

**Table S1:** Table with experimental parameters.

IONIC LIQUID: EMIM – BF <sub>4</sub>												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	EMIM – BF <sub>4</sub>	0.60 cm <sup>3</sup> on 20 cm <sup>2</sup> glass plate	5	3.5	20 (air)	-	-	4.0	30	RT	5.5±0.86	[1]
Au	EMIM – BF <sub>4</sub>	80 mm <sup>3</sup> on 4 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	10	5	RT	2.9±0.8	[2]
Au	EMIM – BF <sub>4</sub>	0.60 cm <sup>3</sup> on 10 cm <sup>2</sup> glass plate	4.9	2.5	2 (Ar)	-	-	10	10	RT	2.2	[3]
W Mo Nb Ti	EMIM – BF <sub>4</sub>	0.60 cm <sup>3</sup> on 10 cm <sup>2</sup> glass slide	4.9	2	2 (Ar)	-	-	40	10	RT	WO <sub>x</sub> -3.2 -4.8 MoO <sub>x</sub> - 2.7 - 3.7 NbO <sub>x</sub> - 1.9 - 2.9 TiO <sub>x</sub> - 3.4 - 4.6	[4]
In	EMIM – BF <sub>4</sub>	0.60 cm <sup>3</sup> on 10 cm <sup>2</sup> glass slide	4.9	2	2 (Ar)	-	-	10	10	RT	In@In <sub>2</sub> O <sub>3</sub> : 8±2	[5]
IONIC LIQUID: TMPA – TFSI												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	TMPA – TFSI	0.6 cm <sup>3</sup> on 20 cm <sup>2</sup> glass plate	5	3.5	20 (air)	-	-	4.0	30	RT	1.9±0.46	[1]
Au	TMPA – TFSA	0.8 cm <sup>3</sup> on 4 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	10	5	RT	2.2±0.4	[2]
Au	TMPA – TFSI	0.6 cm <sup>3</sup> on 10 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	40	5	RT	2.3±0.3	[6]
Au Pt	TMPA – TFSI	? cm <sup>3</sup> on a glass plate 10 cm <sup>2</sup>	5	2	10 (Ar)	-	500	40	5	RT	Au: 2.9 ± 0.7 Pt: 1.0 ± 0.3	[7]
Pt	TMPA – TFSI	0.4 cm <sup>3</sup> on a glass plate (2.5 x 2.5 cm)	5.7	4.5	7 (Ar) 7 (N <sub>2</sub> )	-	-	40	5	RT	(Ar): 2.24±0.36 (N <sub>2</sub> ): 3.28±0.60	[8]
Pt	TMPA – TFSI	0.2 cm <sup>3</sup> on a glass plate (2.5 x 2.5 cm)	5.7	4.5	7 (Ar)	-	-	40	5; 15; 30	RT	2.28; 2.37; 2.27	[9]
IONIC LIQUIDS: BMIM – BF <sub>4</sub>												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	BMIM – BF <sub>4</sub>	0.8 cm <sup>3</sup> on a glass plate 4 cm <sup>2</sup>	5	3.5	20 (Ar)	-	-	10	5	RT	2.5±0.6	[2]
Au	BMIM – BF <sub>4</sub>	0.6 cm <sup>3</sup> on a glass plate (26mm x 38mm)	5	3.0	20 (Ar)	-	-	40	5	RT	2.6 ± 0.4	[10]
Au	BMIM – BF <sub>4</sub>	2.0 cm <sup>3</sup> on a stainless plate 15.9 cm <sup>2</sup>	5	2.5	12-13 (Ar)	-	1000	20	50	20 → 80	(SAXS): 0.6 → 3.5	[11]
Au	BMIM – BF <sub>4</sub>	2.0 cm <sup>3</sup> on a stainless plate 15.9 cm <sup>2</sup>	5	2.5; 5.0; 7.5	13-30 (Ar)	-	700→1000	20 → 40	50	20	From 1 to 3	[12]
Au	BMIM – BF <sub>4</sub>	1.23 g in a cylindrical glass support (3 cm)	5	5	2.1 (Ar)	-	335	40	2.5	RT	3.6 ± 0.4	[13]
In	BMIM – BF <sub>4</sub>	0.60 cm <sup>3</sup> on 10 cm <sup>2</sup> glass slide	4.9	2	2 (Ar)	-	-	10	10	RT	In@In <sub>2</sub> O <sub>3</sub> : 9.7±2.0	[5]

**Table S1:** Table with experimental parameters.

Pt	BMIM – BF <sub>4</sub>	0.4 cm <sup>3</sup> on 6.3 cm <sup>2</sup> glass plate	2.5	9.5 (Ar)	-	-	40	15	RT	1.2	[3]	
IONIC LIQUID: BMIM – PF <sub>6</sub>												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	BMIM – PF <sub>6</sub>	1.23 g in a cylindrical glass support (3 cm)	5	5	2.1 (Ar)	-	335	40	2.5	RT	3.7 ± 0.4	[13]
Au	BMIM – PF <sub>6</sub>	? cm <sup>3</sup> on 10 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	40	5	RT	2.6 ± 0.3	[14]
Au	BMIM – PF <sub>6</sub>	0.6 cm <sup>3</sup> on a glass plate (26mm x 38mm)	5	3.0	20 (Ar)	-	-	40	5	RT	2.3 ± 0.4	[10]
Au	BMIM – PF <sub>6</sub>	0.60 cm <sup>3</sup> on 5.7 cm <sup>2</sup> glass slide	5	3.5	20 (air)	-	-	40	5	RT	2.5	[15]
Au	BMIM – PF <sub>6</sub>	0.60 cm <sup>3</sup> on 10 cm <sup>2</sup> glass slide	5	3.5	20 (Ar)	-	-	40	5	RT	2.6±0.6	[16]
Au	BMIM – PF <sub>6</sub>	1.0 cm <sup>3</sup> on 6.1 cm <sup>2</sup> Teflon plate	5	3.5	20 (Ar)	-	-	40	5	RT	Au: 2.6±0.3	[17]
Au	BMIM – PF <sub>6</sub>	2.0 cm <sup>3</sup> on 15.9 cm <sup>2</sup> stainless plate	5	2.5	12-14 (Ar)	-	1000	20	50	20 → 80	(SAXS): 0.5 → 3	[18]
Au	BMIM – PF <sub>6</sub>	2.0 cm <sup>3</sup> on 15.9 cm <sup>2</sup> stainless plate	5	2.5	16-19 (Ar)	-	1000	20	50	RT	(SAXS): 1.0–1.5	[19]
Ag	BMIM – PF <sub>6</sub>	0.6 cm <sup>3</sup> on 10 cm <sup>2</sup> glass plate	5	8.5	5 (Ar)	-	-	10	5→45	RT	5.7±1.8	[20]
Ag	BMIM – PF <sub>6</sub>	0.6 cm <sup>3</sup> on 10 cm <sup>2</sup> glass plate	5	8.5	5 (Ar)	-	-	40	5	RT	11	
Ag	BMIM – PF <sub>6</sub>	1.0 cm <sup>3</sup> on 6.1 cm <sup>2</sup> Teflon plate	5	3.5	20 (Ar)	-	-	40	5	RT	Ag: 6.0±1.5	[17]
Ag	BMIM – PF <sub>6</sub>	multiple cavity holder (40 µL per cavity)	3.81		0.5 (Ar)	30	-	-	15	RT	6±3	[21]
Pd	BMIM – PF <sub>6</sub>	0.2 cm <sup>3</sup> on 2.5 x 2.5 cm glass plate	5.7	4.5	8 (Ar)	-	-	40	5	RT	3.0	[22]
IONIC LIQUID: BMIM – TFSI												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	BMIM – TFSI	0.6 cm <sup>3</sup> on a glass plate (26mm x 38mm)	5	3	20 (Ar)	-	-	40	5	RT	2.0 ± 0.4	[10]
Au	BMIM – TFSI	2.0 cm <sup>3</sup> on 15.9 cm <sup>2</sup> stainless plate	5	2.5	12-14 (Ar)	-	1000	20	50	20 → 80	(SAXS): 0.5 → 3	[18]
Au	BMIM – TFSI	1.23 g in a cylindrical glass support (ϕ = 3cm)	5	5	2.1 (Ar)	-	299	20	2.5	RT	3.2 ± 0.5	[13]
							322	30	2.5	RT	3.4 ± 0.5	
							335	40	2.5	RT	3.5 ± 0.6	
							358	60	2.5	RT	3.9 ± 0.8	
							410	110	2.5	RT	4.6 ± 0.7	
							335	40	5.0	RT	4.0 ± 0.9	
							335	40	7.5	RT	3.9 ± 0.8	
							335	40	10	RT	4.0 ± 0.8	
Au	BMIM – TFSI	0.6 cm <sup>3</sup> on 10 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	40	5	RT	2.3±0.3	[6]

**Table S1:** Table with experimental parameters.

Au	BMIM – TFSI	multiple cavity holder (40 $\mu\text{L}$ per cavity)	3.81	0.5 (Ar)	30	-	-	30	RT	Au: $2.1 \pm 0.7$	[23]	
Cu	BMIM – TFSI	multiple cavity holder (30 $\mu\text{L}$ per cavity)	3.81	0.5 (Ar)	30	-	-	120	RT	Cu: $2.8 \pm 1.1$	[23]	
IONIC LIQUID: BMIM – TFSI												
Cu	BMIM - TFSI	50 ml of IL in a petri dish (110 mm inner diameter $\times$ 20 mm height) with 30 rpm	3.81	9	0.5 (Ar)	30	405	75	24 hours	RT	$2.6 \pm 1$	
Pd	BMIM – TFSI	0.2 cm <sup>3</sup> on 2.5 $\times$ 2.5 cm glass plate	5.7	4.5	8 (Ar)	-	-	40	5	RT	2.2	
Ag	BMIM – TFSI	4.0 cm <sup>3</sup> into a cylindrical ceramic crucible ( $\phi = 4.5$ cm)	7.5	15	1.3 (Ar)	20	-	-	RT	5-20	[25]	
Ag	BMIM – TFSI	multiple cavity holder (40 $\mu\text{L}$ per cavity)	3.81	0.5 (Ar)	30	-	-	15	RT	$8 \pm 4$	[21]	
OTHER IONIC LIQUIDS												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	HyEMIM – BF <sub>4</sub> OMIM – BF <sub>4</sub> HyEMIM – TFSA EMIM – TFSA Ch – TFSA SBMI – TFSA EMIM – EtSO <sub>4</sub> EMIM – Ac BMIM – SCN BMIM – actate HyEA – formate	0.8 cm <sup>3</sup> on a glass plate 4 cm <sup>2</sup>	5	3.5	20 (Ar)	-	-	10	5	RT	5.1 $\pm 0.7$ 2.4 $\pm 0.8$ 4.3 $\pm 0.8$ 2.9 $\pm 0.6$ 5.5 $\pm 0.7$ 2.2 $\pm 0.6$ 2.5 $\pm 0.6$ 3.8 $\pm 1.3$ 2.8 $\pm 0.7$ 3.0 $\pm 0.7$ 5.2 $\pm 1.3$	[2]
Au	HyEMIM – BF <sub>4</sub>	0.80 cm <sup>3</sup> on 4.0 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	10	2.5	RT	4.2	[26]
Au	(BCN)MIM – TFSI	1.2 g on a Petri plate ( $\phi = 3$ cm)	5	5	2 (Ar)	2 $\rightarrow$ 45	275 $\rightarrow$ 410	-	2.5	RT	8.7 $\rightarrow$ 5.1	[27]
Au	[PY1,1O1] – BF <sub>4</sub> [N122,1O2] – BF <sub>4</sub>	0.6 cm <sup>3</sup> on a glass plate (26mm $\times$ 38mm)	5	3	20 (Ar)	-	-	40	5	RT	2.7 $\pm 0.5$ 3.5 $\pm 0.7$	[10]
Au	AMIM – TFSI AMIM – BF <sub>4</sub>	Layer of IL on Si(111)	-	-	3 (Ar)	50	-	-	5 sec	RT	5.1 $\pm 0.8$ 6.5 $\pm 0.7$	[28]
Au	EMIM – EtSO <sub>4</sub>	1.5 ml on a plate ( $\phi = 13$ cm)	5.1	15.2	0.53 (Ar)	30 (RF)	-	-	9	RT	$1.3 \pm 0.7$	[29]
Au	C <sub>n</sub> MIM – BF <sub>4</sub> (n=2,4,8) C <sub>n</sub> MIM – OTf (n=2,4,6) BMIM – FSA	2.0 cm <sup>3</sup> on 15.9 cm <sup>2</sup> stainless plate	5	2.5	12-14 (Ar)	-	1000	20	50	20 $\rightarrow$ 80 (SAXS): 0.5 $\rightarrow$ 3	[18]	
Au	C <sub>n</sub> MIM – BF <sub>4</sub> (n=2,4,8)	1 cm <sup>3</sup> on 6.1 cm <sup>2</sup> Teflon plane	5	2.5	10-15 (Ar)	-	1000	5	12 $\rightarrow$ 36	RT	(SAXS): 0.75 $\rightarrow$ 3.5	[30]

**Table S1:** Table with experimental parameters.

Au	BMIM – TFSI BMIM – N(CN) <sub>2</sub> BMIM – BF <sub>4</sub> BMIM – PF <sub>6</sub>	3.5 cm <sup>3</sup> in a PTFE container ( $\phi = 4$ cm)	5	4	10 (Ar)	-	-	50	1	RT	<2.5	[31] [32]
Au	PP13-TFSI	2.4 ml on a laboratory glass dish	5		3 (Ar)	-	-	20	60	RT	2.3±1.0	[33]
Au	BMIM – FAP	1.23 g in a cylindrical glass support (3 cm)	5	5	2.1 (Ar)	-	335	40	2.5	RT	4.9 ± 0.9	[13]
OTHER IONIC LIQUIDS												
Ag	BMIM – (Pf) <sub>2</sub> N EMIM – TFSI HMIM – TFSI BuPy – TFSI BmPyr – TFSI	multiple cavity holder (40 $\mu$ L per cavity)	3.81		0.5 (Ar)	30	-	-	15	RT	7±4 7±3 7±3 4±3 9±8	[21]
Ag	(BCN)MIM – TFSI (MeOE)MIM – TFSI (HSE)MIM – TFSI	3 g into a Petri dish ( $\phi=3$ cm)	5	5	2 (Ar)	-	335	40	2.5	RT	5.0 (1.8-14.3) 8.2 (4.0-15.5) 8.7 (5.9-12.0)	[34]
Pt	[Me <sub>3</sub> PrN, Et <sub>3</sub> OctN, MePrPyr, MeBuPyr, EtMelm, AllyEtIm] - TFSI	0.4 cm <sup>3</sup> on 2.5 x 2.5 cm glass plate	4.5		7 (Ar)	-	-	40	15	RT	1.4 - 2.4	[35]
Pt	HyEMIM – BF <sub>4</sub>	0.80 cm <sup>3</sup> on 4.0 cm <sup>2</sup> glass plate	5	3.5	20 (Ar)	-	-	10	2.5	RT	1.0	[26]
Pd	(BCN)MIM – TFSI (MeOE)MIM – TFSI (HSE)MIM – TFSI	3 g into a Petri dish ( $\phi=3$ cm)	5	5	2 (Ar)	-	-	80 → 320	5	RT	1.4-4.6 1.3-6.4 2.4-7.3	[36]
In	AMIM – BF <sub>4</sub> AEI-BF <sub>4</sub>	0.60 cm <sup>3</sup> on 10 cm <sup>2</sup> glass slide	4.9	2	2 (Ar)	-	-	10	10	RT	In@In <sub>2</sub> O <sub>3</sub> :18.2±4.0 In@In <sub>2</sub> O <sub>3</sub> :16.8±5.1	[5]
PEEL												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Cu	PEEL	7 g on a glass Petri plate ( $\phi = 6$ cm)			2 (Ar)	-	-	10 → 100	8 and 60	20	3 → 10	[37]
Ti	PEEL	4.0 cm <sup>3</sup> into a cylindrical ceramic crucible ( $\phi=4.5$ cm)	5.1	15	0.7 (Ar)	20	-	-			TiO <sub>2</sub> : 30-150	[25]
Ti	PEEL	2.5g on a Petri plate (5.7 cm <sup>2</sup> )	6		2 (Ar)	-	310-265	400	60-180		(Ti→TiO <sub>2</sub> ): 2.5-5	[38]
Au	PEEL		4		10 (air)	-	-	40	3	RT	2.1±0.7	[39]
Au	PEEL	2.5 ml into a watch glass	7.62	13	0.5	Au-20 Ag-17 Cu-75	-	60 49 175	5	RT	Au -3 Ag-16±8 CuOx – 1.2±0.8	[40]
Au Cu	PEEL	2.5 ml into a watch glass that was rotated (5 rpm)	7.62	13	0.5	17 75	-	49 175	5	RT	2.8 1.8 (CuOx)	[41]
PEGs (mostly PEG with MW~600)												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.

**Table S1:** Table with experimental parameters.

Au	PEG	2.0 cm <sup>3</sup> on a stainless plate 15.9 cm <sup>2</sup>	2.5	16-19 (Ar)	-	1000	20	50	20 → 60	2 → 8	[42]	
Au Ag	PEG (200,400,600)	2 ml of PEG	4.5-6.0	8 (Ar)	-	-	10-60	100-600 sec		Up to 50	[43]	
Au Ag	PEG	10 cm <sup>3</sup> on a Petri plate ( $\phi = 3.0\text{ cm}$ )	5	7.5	2 (Ar)	-	-	20 34	25	Au: 5.3±1.6 Ag: 5.2±1.7	[44]	
Cu	PEG	10 cm <sup>3</sup> on a Petri plate ( $\phi = 6.3\text{ cm}$ )	6	2 (Ar)	-	-	50	30	30	growth 3.1 → 4.1 Cu → CuO	[45]	
Au	PEG	10 cm <sup>3</sup> on a Petri plate ( $\phi = 6.3\text{ cm}$ )	5	11	2 (Ar)	-	-	10 20	30	2.6 ± 0.8 2.7 ± 0.9	[46]	
PEGs (mostly PEG with MW~600)												
Cu	PEG	10 cm <sup>3</sup> in 6.3cm $\phi$ Petri plate	5	11	2 (Ar)	-	-	40 50	30	30	1.0± 0.3 1.5 ± 0.4	
Pt	PEG	10 cm <sup>3</sup> on a Petri plate ( $\phi = 6.3\text{ cm}$ )	5	5	2 (Ar)	-	-	5 → 50	30	30	0.9 → 2.7	
Au Pt	PEG	2 ml on a Petri plate	5	8 (Ar)	-	-	30		RT	Au: 6.3 Pt: 3.9	[48]	
Au	PEG-600	2 ml	5	8 (Ar)	-	-	30	1.67→8.33	20	About 10 nm	[49]	
Ag	PEG PEG (+MUA)		0.5	5	2 (Ar)	-	-	30	20	RT	7.4±3.6 2.2±0.5	
Ag	PEG (+SMP)	10 g in a glass petri dish ( $\phi = 6.5\text{ cm}$ )	5	2 (Ar)	-	-	30	20	RT	2.7±0.5	[51]	
Au	PEG	10 g in a glass petri dish ( $\phi = 6.5\text{ cm}$ )	5	2 (Ar)	-	-	20	60		7.4±2.1 without and 3.7±0.9 with stirring	[52]	
Au	PEG+thiocholine	10 g in a glass petri dish ( $\phi = 6.5\text{ cm}$ )	5	2 (Ar)	-	-	30	20	30	1.7 ± 0.6	[53]	
Au	PEG PEG-SH PEG-S <sub>2</sub> H <sub>2</sub>	2 ml on a Petri plate ( $\phi = 4\text{ cm}$ )	5	10 (Ar)	-	420-430	30	5	RT	5.6±1.8 4.2±0.8 3.6±0.6	[54]	
Au	PEG PEG-SH PEG-NH <sub>2</sub>	2 ml on a Petri plate ( $\phi = 4\text{ cm}$ )	5	10 (Ar)	-	420-430	30	5	RT	5.6±1.9 1.5±0.3 2.5±0.8	[55]	
Cu	PEG PEG (+MUA)	7 g	6		-	-	20	60	40	2.6 ± 0.6 1.6 ± 0.3	[56]	
Au Ag	PEG (+MUTAB)	10.152 g of PEG + 0.04 g MUTAB		(Ar)	-	-	20 34	15		1.1 ± 0.3 1.4 ± 0.4	[57]	
Au Ag Cu	PEG (+MUTAB)	3.384 g PEG + 0.02 g of MUTAB on a Petri plate ( $\phi = 6.3\text{ cm}$ )	5	2 (Ar)	-	-	10	10		Au 1.2 ± 0.6 Ag 1.3 ± 0.6 Cu 1.0 ± 0.3	[58]	
Au	PEG PEG (+α-Thioglycerol)	10 cm <sup>3</sup> in 6.5 cm $\phi$ Petri plate (+0.05 g α-T)	2.5	2 -30 (air)	-	About 200	10→30	10→40	RT	no α-T: 4.7 - 5.5 α-T: 2.2-3.2	[59]	
GLYCEROL												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	Diglycerol (+thiocholine chloride)	12.8 g in 6.5cm $\phi$ Petri plate	2.5	20 (Ar)	-	-	30	30	25	6.7±3.2 (2.0±0.7)	[60]	

**Table S1:** Table with experimental parameters.

Au	Diglycerol +SH +NH <sub>2</sub> +C(O)OH	10 g on a Petri plate ( $\phi = 6\text{cm}$ )	5	2 (Ar)	-	-	20	20	RT	6.7±3.2 4.6±0.9 3.0±0.8 2.1±0.8	[61]	
Au	Glycerol		5.08	3.5	1.33 (Ar)	-	-	10	60, 120	RT	3.5 → 6.4	[62]
Pd	Glycerol	5 ml on a Petri plate ( $\phi = 4\text{cm}$ )	5	4-6 (Ar)	-	420-430	40	5	20	Pd=2.4±0.4	[63]	
Pt										Pt=1.7±0.3		
Au	Glycerol	3 ml in 4cm $\phi$ Petri plate	5	7 (Ar)	-	-	40	5	20	4-12	[64]	
Ag												
GLYCEROL												
Au	Glycerol	2 ml on a Petri plate ( $\phi = 4\text{cm}$ )	5	7 (Ar)	-	-	30	5	RT	3.5 ± 1.5 3.5 ± 2.4	[65]	
Ag	Glycerol +1% of PVP	3 cm <sup>3</sup> on a Petri plate ( $\phi = 4\text{cm}$ )		0.06 (Ar)	-	280 - 305	30	6	5 → 40	12.8 → 21.2	[66]	
VEGETABLE OILS												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Ag	Castor oil	1 cm <sup>3</sup> into a glass support ( $\phi = 3\text{ cm}$ )	5	2 (Ar)	13	320	-	2.5	RT	5.5±1.0	[67]	
	Canola oil				38	420	-	2.5		1.4 – 8.1		
Au	Castor oil	1.15 cm <sup>3</sup> into a glass support ( $\phi = 3\text{ cm}$ )	5	2 (Ar)	13	330	-	2.5	RT	3.6 ± 1.0 3.8 ± 1.1 2.4 → 3.7	[68,69]	
					13	330	10					
					260→450		2.5					
Ag	Castor oil	5 ml into plastic beaker ( $\phi = 2.8\text{ cm}$ )	5.1	20	0.067; 0.67; 2	20;40;60;80		1	RT	From 0.4 to 8 nm for primary NPs	[70]	
Au	Castor oil	4 g into plastic beaker ( $\phi = 2.8\text{ cm}$ )	5.1	20	0.067; 0.67; 2	20;40;60;80		1→10	RT	2.4 – 3.2	[71]	
Au	Sunflower oil	2 ml into a glass dish ( $\phi = 5\text{ cm}$ )	12	43	-	-	41	5 15 30	RT	2.9 ± 0.5 3.7 ± 0.7 3.4 ± 0.7	[72]	
SILICONE OILS												
Target	Liquid	Amount of liquid substrate	d <sub>t</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Ag	Silicone oil	3 mm in diameter pure silicone oil drop on a piece of glass	81	6	0.2 (Ar)	15, 30, 50 (RF)	-	-		17 → 100	film	[73]
Ag	Silicone oil	2 mm in diameter pure silicone oil drop on a piece of glass		8	0.2 (Ar)	50 (RF)	-	-	RT	film	[74]	
Ag	Silicone oils	Running liquid substrate		8	1→30 (Ar)	Ag: 500	-	-	Up to 240	Ag: 5-20	[75–77]	
Fe	Silicone oil +Koratine (80)					Fe:1000						
Al	Silicone oil	2 mm in diameter pure silicone oil drop on a piece of glass	81	6	0.2 (Ar)	-	-	30		18→120	film	[78]
Ag												

**Table S1:** Table with experimental parameters.

Fe	Silicone oil	resulting oil substrate with an area of about 25×18 mm <sup>2</sup> thickness of about 0.5 mm	6	9	0.8 (Ar)	10 → 100	-	-	0.16→12	film	[79]	
Fe	Silicone oil	resulting oil substrate with an area of about 25×18 mm <sup>2</sup> thickness of about 0.5 mm	6	9	0.8 (Ar)	10→120	-	-	0.08→12	film	[80]	
Au	silicone oil silicon oil +OTD molten OTD	1.0 g on a Petri plate ( $\phi = 3\text{cm}$ )	2.5	20 (Ar)	-	200	30	30	50	$4.9 \pm 1.4$ $1.9 \pm 0.4$ $1.3 \pm 0.3$	[81]	
PEMP												
Target	Liquid	Amount of liquid substrate	d <sub>l</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Cu	PEMP	7g			2 (Ar/air)	-	-	20	120	25	$1.1 \pm 0.2$	[82]
Ag	PEMP	7 g on a glass Petri plate ( $\phi = 6\text{ cm}$ )	5	2 (Ar)	-	-	50	20	0 → 100	1.8-3.4		[83]
Ag	PEMP	13 g on a glass Petri plate ( $\phi = 6\text{cm}$ )	4	10 (Ar)	-	-	30	20	RT	2.5		[84]
Au	PEMP		4	10 (air)	-	-	40	5 -15	RT	Less 1 nm		[39]
OTHER HOST LIQUIDS												
Target	Liquid	Amount of liquid substrate	d <sub>l</sub> , cm	WD, cm	p, Pa	P, W	V, V	I, mA	t <sub>s</sub> , min	T <sub>s</sub> , C	d <sub>NP</sub> , nm	Ref.
Au	MTAB	0.15 g of molten 6-MTAB in a 20 mm vessel	2.3	15 (air)	-	-	40	5 → 20	110	$1.3 \pm 0.3$		[85]
Au	DES		5.7	3.5	5 (Ar)	-	450	20	0.5→5		$5.0 \pm 0.5$	[86–88]
Au	Liquid crystals 4-pentyl-40-cyanobiphenyl	0.3 g in a cylindrical glass container ( $\phi=15\text{mm}$ )		20	(oxygen)	-	1000	-	20	RT	$2.9 \pm 0.6$	[89]

## References

- (1) Torimoto, T.; Okazaki, K. I.; Kiyama, T.; Hirahara, K.; Tanaka, N.; Kuwabata, S. *Appl. Phys. Lett.* **2006**, *89*, 243117. doi:10.1063/1.2404975
- (2) Sugioka, D.; Kameyama, T.; Kuwabata, S.; Torimoto, T. *Phys. Chem. Chem. Phys.* **2015**, *17*, 13150–13159. doi:10.1039/c5cp01602a
- (3) Torimoto, T.; Ohta, Y.; Enokida, K.; Sugioka, D.; Kameyama, T.; Yamamoto, T.; Shibayama, T.; Yoshii, K.; Tsuda, T.; Kuwabata, S. *J. Mater. Chem. A* **2015**, *3*, 6177–6186. doi:10.1039/C4TA06643J
- (4) Suzuki, T.; Suzuki, S.; Tomita, Y.; Okazaki, K. I.; Shibayama, T.; Kuwabata, S.; Torimoto, T. *Chem. Lett.* **2010**, *39*, 1072–1074. doi:10.1246/cl.2010.1072
- (5) Suzuki, T.; Okazaki, K.; Suzuki, S.; Shibayama, T.; Kuwabata, S.; Torimoto, T. *Chem. Mater.* **2010**, *22*, 5209–5215. doi:10.1021/cm101164r
- (6) Okazaki, K. I.; Kiyama, T.; Suzuki, T.; Kuwabata, S.; Torimoto, T. *Chem. Lett.* **2009**, *38*, 330–331. doi:10.1246/cl.2009.330
- (7) Suzuki, S.; Suzuki, T.; Tomita, Y.; Hirano, M.; Okazaki, K.; Kuwabata, S.; Torimoto, T. *CrystEngComm* **2012**, *14*, 4922. doi:10.1039/c2ce25235j
- (8) Tsuda, T.; KURIHARA, T.; HOSHINO, Y.; KIYAMA, T.; OKAZAKI, K.; TORIMOTO, T.; KUWABATA, S. *Electrochemistry* **2009**, *77*, 693–695. doi:10.5796/electrochemistry.77.693
- (9) Tsuda, T.; Yoshii, K.; Torimoto, T.; Kuwabata, S. *J. Power Sources* **2010**, *195*, 5980–5985. doi:10.1016/j.jpowsour.2009.11.027
- (10) Suzuki, S.; Ohta, Y.; Kurimoto, T.; Kuwabata, S.; Torimoto, T. *Phys. Chem. Chem. Phys.* **2011**, *13*, 13585–13593. doi:10.1039/c1cp20814d
- (11) Hatakeyama, Y.; Takahashi, S.; Nishikawa, K. *J. Phys. Chem. C* **2010**, *114*, 11098–11102. doi:10.1021/jp102763n
- (12) Hatakeyama, Y.; Onishi, K.; Nishikawa, K. *RSC Adv.* **2011**, *1*, 1815–1821. doi:10.1039/c1ra00688f
- (13) Wender, H.; De Oliveira, L. F.; Migowski, P.; Feil, A. F.; Lissner, E.; Precht, M. H. G.; Teixeira, S. R.; Dupont, J. *J. Phys. Chem. C* **2010**, *114*, 11764–11768. doi:10.1021/jp102231x
- (14) Khatri, O. P.; Adachi, K.; Murase, K.; Okazaki, K. I.; Torimoto, T.; Tanaka, N.; Kuwabata, S.; Sugimura, H. *Langmuir* **2008**, *24*, 7785–7792. doi:10.1021/la800678m
- (15) Okazaki, K. I.; Sakuma, J.; Yasui, J. I.; Kuwabata, S.; Hirahara, K.; Tanaka, N.; Torimoto, T. *Chem. Lett.* **2011**, *40*, 84–86. doi:10.1246/cl.2011.84
- (16) Kameyama, T.; Ohno, Y.; Kurimoto, T.; Okazaki, K.; Uematsu, T.; Kuwabata, S.; Torimoto, T. *Phys. Chem. Chem. Phys.* **2010**, *12*, 1804–1811. doi:10.1039/B914230D
- (17) Okazaki, K.; Kiyama, T.; Hirahara, K.; Tanaka, N.; Kuwabata, S.; Torimoto, T. *Chem. Commun.* **2008**, No. 6, 691–693. doi:10.1039/B714761A
- (18) Hatakeyama, Y.; Judai, K.; Onishi, K.; Takahashi, S.; Kimura, S.; Nishikawa, K. *Phys. Chem. Chem. Phys.* **2016**, *18*, 2339–2349. doi:10.1039/c5cp04123f
- (19) Iimori, T.; Hatakeyama, Y.; Nishikawa, K.; Kato, M.; Ohta, N. *Chem. Phys. Lett.* **2013**, *586*, 100–103. doi:10.1016/j.cplett.2013.09.010
- (20) SUZUKI, T.; OKAZAKI, K.; KIYAMA, T.; KUWABATA, S.; TORIMOTO, T. *Electrochemistry* **2009**, *77*, 636–638. doi:10.5796/electrochemistry.77.636
- (21) Meyer, H.; Meischein, M.; Ludwig, A. *ACS Comb. Sci.* **2018**, *20*, 243–250. doi:10.1021/acscombsci.8b00017
- (22) Oda, Y.; Hirano, K.; Yoshii, K.; Kuwabata, S.; Torimoto, T.; Miura, M. *Chem. Lett.* **2010**, *39*, 1069–1071. doi:10.1246/cl.2010.1069
- (23) Meischein, M.; Garzón-Manjón, A.; Frohn, T.; Meyer, H.; Salomon, S.; Scheu, C.; Ludwig, A. *ACS Comb. Sci.* **2019**. doi:10.1021/acscombisci.9b00140
- (24) Meischein, M.; Ludwig, A. *J. Nanoparticle Res.* **2021**, *23*, 1–15. doi:10.1007/s11051-021-05248-8
- (25) Carette, X.; Debièvre, B.; Cornil, D.; Cornil, J.; Leclère, P.; Maes, B.; Gautier, N.; Gautron, E.; El Mel, A. A.; Raquez, J. M.; Konstantinidis, S. *J. Phys. Chem. C* **2018**, *122*, 26605–26612. doi:10.1021/acs.jpcc.8b06987

- (26) Sugioka, D.; Kameyama, T.; Kuwabata, S.; Yamamoto, T.; Torimoto, T. *ACS Appl. Mater. Interfaces* **2016**, *8*, 10874–10883. doi:10.1021/acsami.6b01978
- (27) Wender, H.; Teixeira, S. R.; Migowski, P.; Feil, A. F.; Dupont, J.; Prechtl, M. H. G.; Machado, G.; de Oliveira, L. F.; Leal, R. *Phys. Chem. Chem. Phys.* **2011**, *13*, 13552. doi:10.1039/c1cp21406c
- (28) Calabria, L.; Fernandes, J. A.; Migowski, P.; Bernardi, F.; Baptista, D. L.; Leal, R.; Grehl, T.; Dupont, J. *Nanoscale* **2017**, *9*, 18753–18758. doi:10.1039/C7NR06167F
- (29) Hamm, S. C.; Basuray, S.; Mukherjee, S.; Sengupta, S.; Mathai, J. C.; Baker, G. A.; Gangopadhyay, S. *J. Mater. Chem. A* **2014**, *2*, 792–803. doi:10.1039/C3TA13431H
- (30) Hatakeyama, Y.; Okamoto, M.; Torimoto, T.; Kuwabata, S.; Nishikawa, K. *J. Phys. Chem. C* **2009**, *113*, 3917–3922. doi:10.1021/jp807046u
- (31) Vanecht, E.; Binnemans, K.; Patkovsky, S.; Meunier, M.; Seo, J. W.; Stappers, L.; Fransaer, J. *Phys. Chem. Chem. Phys.* **2012**, *14*, 5662. doi:10.1039/c2cp23677j
- (32) Vanecht, E.; Binnemans, K.; Seo, J. W.; Stappers, L.; Fransaer, J. *Phys. Chem. Chem. Phys.* **2011**, *13*, 13565–13571. doi:10.1039/c1cp20552h
- (33) Yoda, M.; Takashima, T.; Akiyoshi, K.; Torimoto, T.; Irie, H. *J. Chem. Phys.* **2020**, *153*, 014701. doi:10.1063/5.0010100
- (34) Qadir, M. I.; Kauling, A.; Calabria, L.; Grehl, T.; Dupont, J. *Nano-Structures and Nano-Objects* **2018**, *14*, 92–97. doi:10.1016/j.nanoso.2018.01.015
- (35) Yoshii, K.; Tsuda, T.; Arimura, T.; Imanishi, A.; Torimoto, T.; Kuwabata, S. *RSC Adv.* **2012**, *2*, 8262–8264. doi:10.1039/c2ra21243a
- (36) Qadir, M. I.; Kauling, A.; Ebeling, G.; Fartmann, M.; Grehl, T.; Dupont, J. *Aust. J. Chem.* **2019**, *72*, 49. doi:10.1071/CH18183
- (37) Nakagawa, K.; Narushima, T.; Udagawa, S.; Yonezawa, T. *J. Phys. Conf. Ser.* **2013**, *417*. doi:10.1088/1742-6596/417/1/012038
- (38) Porta, M.; Nguyen, M. T.; Yonezawa, T.; Tokunaga, T.; Ishida, Y.; Tsukamoto, H.; Shishino, Y.; Hatakeyama, Y. *New J. Chem.* **2016**, *40*, 9337–9343. doi:10.1039/c6nj01624c
- (39) Shishino, Y.; Yonezawa, T.; Udagawa, S.; Hase, K.; Nishihara, H. *Angew. Chemie* **2011**, *123*, 729–731. doi:10.1002/ange.201005723
- (40) Chauvin, A.; Sergievskaya, A.; Fucikova, A.; Antunes Corrêa, C.; Veselý, J.; Cornil, J.; Cornil, D.; Dopita, M.; Konstantinidis, S. *Nanoscale Adv.* **2021**. doi:10.1039/D1NA00222H
- (41) Chauvin, A.; Sergievskaya, A.; El Mel, A.-A.; Fucikova, A.; Antunes Corrêa, C.; Vesely, J.; Duverger-Nédellec, E.; Cornil, D.; Cornil, J.; Tessier, P.-Y.; Dopita, M.; Konstantinidis, S. *Nanotechnology* **2020**, *31*, 455303. doi:10.1088/1361-6528/abaa75
- (42) Hatakeyama, Y.; Morita, T.; Takahashi, S.; Onishi, K.; Nishikawa, K. *J. Phys. Chem. C* **2011**, *115*, 3279–3285. doi:10.1021/jp110455k
- (43) Slepčka, P.; Elashnikov, R.; Ulbrich, P.; Staszek, M.; Kolská, Z.; Švorčík, V. *J. Nanoparticle Res.* **2015**, *17*. doi:10.1007/s11051-014-2850-z
- (44) Nguyen, M. T.; Yonezawa, T.; Wang, Y.; Tokunaga, T. *Mater. Lett.* **2016**, *171*, 75–78. doi:10.1016/j.matlet.2016.02.047
- (45) Chau, Y. R.; Deng, L.; Nguyen, M. T.; Yonezawa, T. *MRS Adv.* **2019**, *4*, 305–309. doi:10.1557/adv.2019.55
- (46) Nguyen, M. T.; Zhang, H.; Deng, L.; Tokunaga, T.; Yonezawa, T. *Langmuir* **2017**, *33*, 12389–12397. doi:10.1021/acs.langmuir.7b03194
- (47) Deng, L.; Nguyen, M. T.; Yonezawa, T. *Langmuir* **2018**, *34*, 2876–2881. doi:10.1021/acs.langmuir.7b04274
- (48) Pišlová, M.; Kalbáčová, M. H.; Vrabcová, L.; Slepčka, P.; Kolská, Z.; Švorčík, V. *Dig. J. Nanomater. Biostructures* **2018**, *13*, 1035–1044
- (49) Fajstavr, D.; Karasová, A.; Michalcová, A.; Ulbrich, P.; Kasálková, N. S.; Siegel, J.; Švorčík, V.; Slepčka, P. *Nanomaterials* **2021**, *11*, 1–17. doi:10.3390/nano11061434
- (50) Ishida, Y.; Nakabayashi, R.; Matsubara, M.; Yonezawa, T. *New J. Chem.* **2015**, *39*, 4227–4230. doi:10.1039/c5nj00294j
- (51) Ishida, Y.; Nakabayashi, R.; Corpuz, R. D.; Yonezawa, T. *Colloids Surfaces A Physicochem. Eng. Asp.* **2017**, *518*, 25–29. doi:10.1016/j.colsurfa.2017.01.022

- (52) Ishida, Y.; Akita, I.; Sumi, T.; Matsubara, M.; Yonezawa, T. *Sci. Rep.* **2016**, *6*, 1–14. doi:10.1038/srep29928
- (53) Ishida, Y.; Morita, A.; Tokunaga, T.; Yonezawa, T. *Langmuir* **2018**, *34*, 4024–4030. doi:10.1021/acs.langmuir.8b00067
- (54) Rezníčková, A.; Slepicka, P.; Slavíkova, N.; Staszek, M.; Svorcík, V. *Colloids Surfaces A Physicochem. Eng. Asp.* **2017**, *523*, 91–97. doi:10.1016/j.colsurfa.2017.04.005
- (55) Rezníčková, A.; Slavíkova, N.; Kolska, Z.; Kolarova, K.; Belinova, T.; Hubalek Kalbacova, M.; Cieslar, M.; Svorcík, V. *Colloids Surfaces A Physicochem. Eng. Asp.* **2019**, *560*, 26–34. doi:10.1016/j.colsurfa.2018.09.083
- (56) Porta, M.; Nguyen, M. T.; Ishida, Y.; Yonezawa, T. *RSC Adv.* **2016**, *6*, 105030–105034. doi:10.1039/c6ra17291a
- (57) Corpuz, R. D.; Ishida, Y.; Nguyen, M. T.; Yonezawa, T. *Langmuir* **2017**, *33*, 9144–9150. doi:10.1021/acs.langmuir.7b02011
- (58) Corpuz, R. D.; Ishida, Y.; Yonezawa, T. *New J. Chem.* **2017**, *41*, 6828–6833. doi:10.1039/c7nj01369h
- (59) Sumi, T.; Motono, S.; Ishida, Y.; Shirahata, N.; Yonezawa, T. *Langmuir* **2015**, *31*, 4323–4329. doi:10.1021/acs.langmuir.5b00294
- (60) Ishida, Y.; Lee, C.; Yonezawa, T. *Sci. Rep.* **2015**, *5*, 8–13. doi:10.1038/srep15372
- (61) Akita, I.; Ishida, Y.; Yonezawa, T. *Bull. Chem. Soc. Jpn.* **2016**, *89*, 1054–1056. doi:10.1246/bcsj.20160187
- (62) Lee, S. H.; Jung, H. K.; Kim, T. C.; Kim, C. H.; Shin, C. H.; Yoon, T.-S.; Hong, A.-R.; Jang, H. S.; Kim, D. H. *Appl. Surf. Sci.* **2018**, *434*, 1001–1006. doi:10.1016/j.apsusc.2017.11.008
- (63) Staszek, M.; Siegel, J.; Kolářová, K.; Rimpelová, S.; Švorčík, V. *Micro Nano Lett.* **2014**, *9*, 778–781. doi:10.1049/mnl.2014.0345
- (64) Siegel, J.; Kolářová, K.; Vosmanská, V.; Rimpelová, S.; Leitner, J.; Švorčík, V. *Mater. Lett.* **2013**, *113*, 59–62. doi:10.1016/j.matlet.2013.09.047
- (65) Siegel, J.; Kvítek, O.; Ulbrich, P.; Kolská, Z.; Slepíčka, P.; Švorčík, V. *Mater. Lett.* **2012**, *89*, 47–50. doi:10.1016/j.matlet.2012.08.048
- (66) Staszek, M.; Siegel, J.; Polívková, M.; Švorčík, V. *Mater. Lett.* **2017**, *186*, 341–344. doi:10.1016/j.matlet.2016.10.036
- (67) Wender, H.; Gonçalves, R. V.; Feil, A. F.; Migowski, P.; Poletto, F. S.; Pohlmann, A. R.; Dupont, J.; Teixeira, S. R. *J. Phys. Chem. C* **2011**, *115*, 16362–16367. doi:10.1021/jp205390d
- (68) Wender, H.; De Oliveira, L. F.; Feil, A. F.; Lissner, E.; Migowski, P.; Meneghetti, M. R.; Teixeira, S. R.; Dupont, J. *Chem. Commun.* **2010**, *46*, 7019–7021. doi:10.1039/c0cc01353f
- (69) Castro, H. P. S.; Wender, H.; Alencar, M. A. R. C.; Teixeira, S. R.; Dupont, J.; Hickmann, J. M. *J. Appl. Phys.* **2013**, *114*, 183104. doi:10.1063/1.4831679
- (70) Sergievskaya, A.; O'Reilly, A.; Chauvin, A.; Veselý, J.; Panepinto, A.; De Winter, J.; Cornil, D.; Cornil, J.; Konstantinidis, S. *Colloids Surfaces A Physicochem. Eng. Asp.* **2021**, *615*, 126286. doi:10.1016/j.colsurfa.2021.126286
- (71) Sergievskaya, A.; O'Reilly, A.; Alem, H.; De Winter, J.; Cornil, D.; Cornil, J.; Konstantinidis, S. *Front. Nanotechnol.* **2021**, *3*. doi:10.3389/fnano.2021.710612
- (72) Michels, F. S.; Gonçalves, P. J.; Nascimento, V. A.; Oliveira, S. L.; Wender, H.; Caires, A. R. L. *Nanomaterials* **2021**, *11*, 1668. doi:10.3390/nano11071668
- (73) Feng, C.-M.; Ge, H. L.; Tong, M.-R.; Ye, G. X.; Jiao, Z.-K. *Thin Solid Films* **1999**, *342*, 30–34. doi:10.1016/S0040-6090(98)01151-1
- (74) Ye, G. X.; Feng, C. M.; Zhang, Q. R.; Ge, H. L.; Zhang, X. J. *Chinese Phys. Lett.* **1996**, *13*, 772–774. doi:10.1088/0256-307X/13/10/016
- (75) Wagener, M.; Günther, B. High Pressure DC-Magnetron Sputtering on Liquids: A New Process for the Production of Metal Nanosuspensions. In *Structure, Dynamics and Properties of Disperse Colloidal Systems*; Steinkopff: Darmstadt, 1998; Vol. 111, pp 78–81
- (76) Wagener, M.; Murty, B. S.; Günther, B. *MRS Proc.* **1996**, *457*, 149. doi:10.1557/PROC-457-149

- (77) Wagener, M.; Günther, B. *J. Magn. Magn. Mater.* **1999**, *201*, 41–44. doi:10.1016/S0304-8853(99)00055-4
- (78) Ge, H. L.; Feng, C.; Ye, G. X.; Ren, Y.; Jiao, Z. *J. Appl. Phys.* **1997**, *82*, 5469–5471. doi:10.1063/1.365574
- (79) Zhang, Y. J.; Yu, S. J. *Int. J. Mod. Phys. B* **2009**, *23*, 3147–3157. doi:10.1142/S0217979209049772
- (80) YU, S.-J.; Zhang, Y. J.; CHEN, J.-X.; Ge, H. L. *Surf. Rev. Lett.* **2006**, *13*, 779–784. doi:10.1142/S0218625X06008840
- (81) Ishida, Y.; Sumi, T.; Yonezawa, T. *New J. Chem.* **2015**, *39*, 5895–5897. doi:10.1039/c5nj01011j
- (82) Porta, M.; Nguyen, M. T.; Tokunaga, T.; Ishida, Y.; Liu, W.-R.; Yonezawa, T. *Langmuir* **2016**, *32*, 12159–12165. doi:10.1021/acs.langmuir.6b03017
- (83) Ishida, Y.; Udagawa, S.; Yonezawa, T. *Colloids Surfaces A Physicochem. Eng. Asp.* **2016**, *498*, 106–111. doi:10.1016/j.colsurfa.2016.03.044
- (84) Ishida, Y.; Udagawa, S.; Yonezawa, T. *Colloids Surfaces A Physicochem. Eng. Asp.* **2016**, *504*, 437–441. doi:10.1016/j.colsurfa.2016.05.035
- (85) Shishino, Y.; Yonezawa, T.; Kawai, K.; Nishihara, H. *Chem. Commun.* **2010**, *46*, 7211. doi:10.1039/c0cc01702g
- (86) Raghuvanshi, V. S.; Ochmann, M.; Hoell, A.; Polzer, F.; Rademann, K. *Langmuir* **2014**, *30*, 6038–6046. doi:10.1021/la500979p
- (87) Raghuvanshi, V. S.; Ochmann, M.; Polzer, F.; Hoell, A.; Rademann, K. *Chem. Commun.* **2014**, *50*, 8693–8696. doi:10.1039/c4cc02588a
- (88) O'Neill, M.; Raghuvanshi, V. S.; Wendt, R.; Wollgarten, M.; Hoell, A.; Rademann, K. *Zeitschrift für Phys. Chemie* **2015**, *229*, 221–234. doi:10.1515/zpch-2014-0644
- (89) Yoshida, H.; Kawamoto, K.; Kubo, H.; Tsuda, T.; Fujii, A.; Kuwabata, S.; Ozaki, M. *Adv. Mater.* **2010**, *22*, 622–626. doi:10.1002/adma.200902831