



## Supporting Information

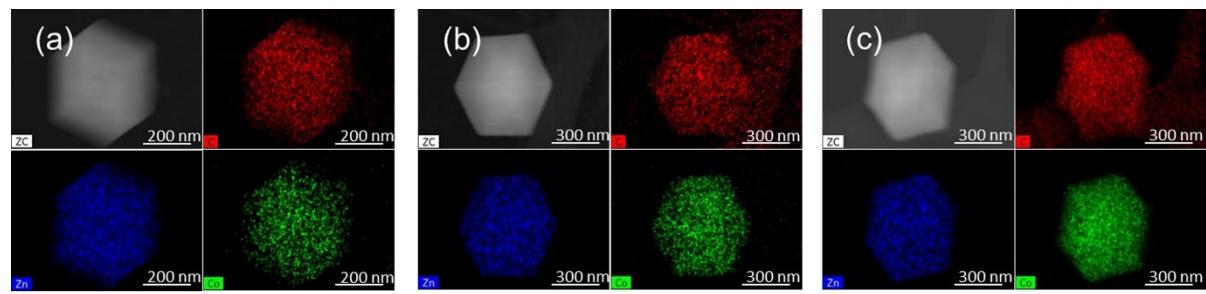
for

### Nanoarchitectonics of the cathode to improve the reversibility of Li–O<sub>2</sub> batteries

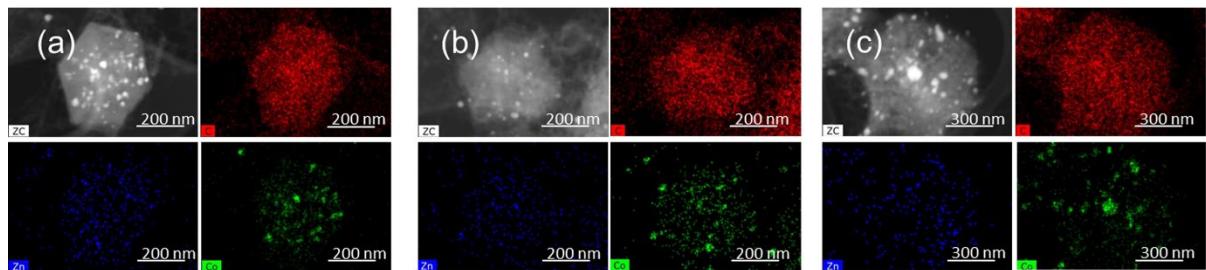
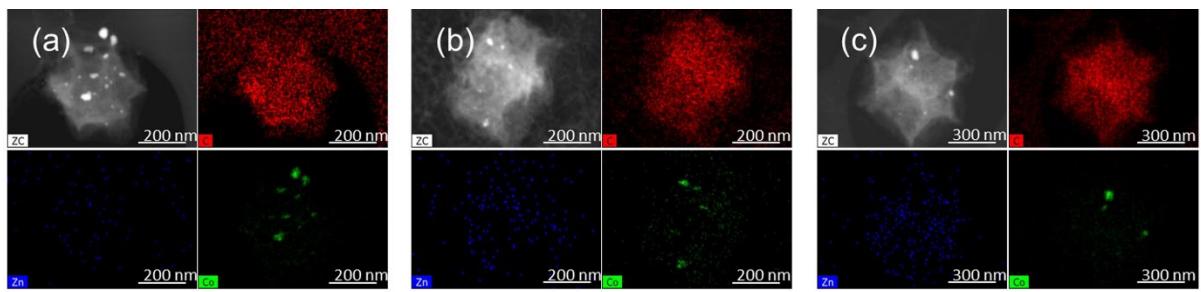
Hien Thi Thu Pham, Jonghyeok Yun, So Yeun Kim, Sang A Han, Jung Ho Kim, Jong-Won Lee and Min-Sik Park

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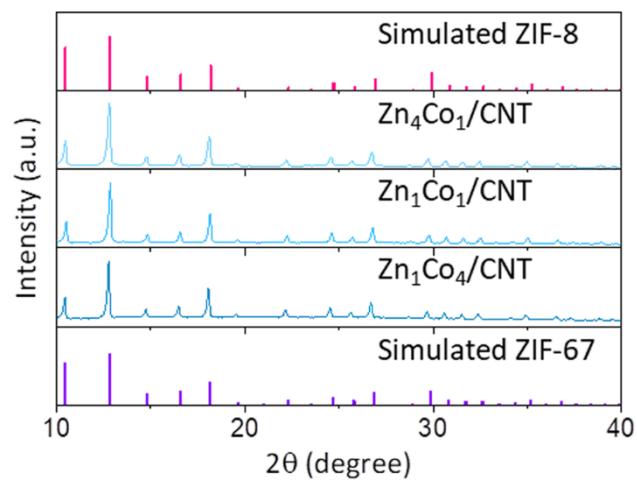
### Additional TEM, EDS, XRD, XPS, BET, SEM, and cathodic overpotential measurements



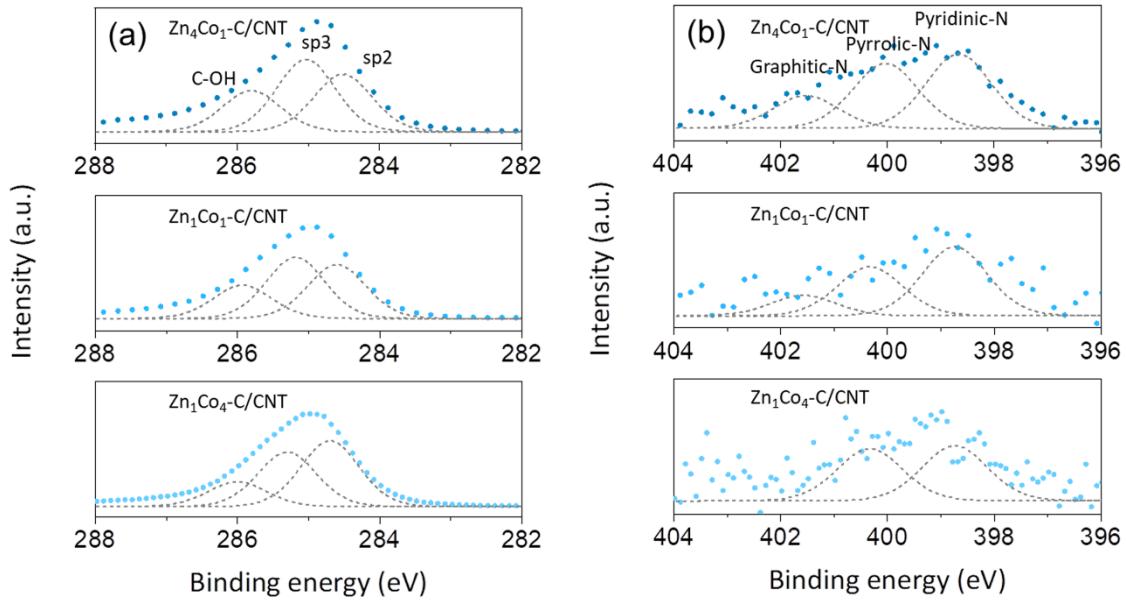
**Figure S1:** EDS elemental mapping results for (a)  $\text{Zn}_4\text{Co}_1/\text{CNT}$ , (b)  $\text{Zn}_1\text{Co}_1/\text{CNT}$ , and (c)  $\text{Zn}_1\text{Co}_4/\text{CNT}$ .



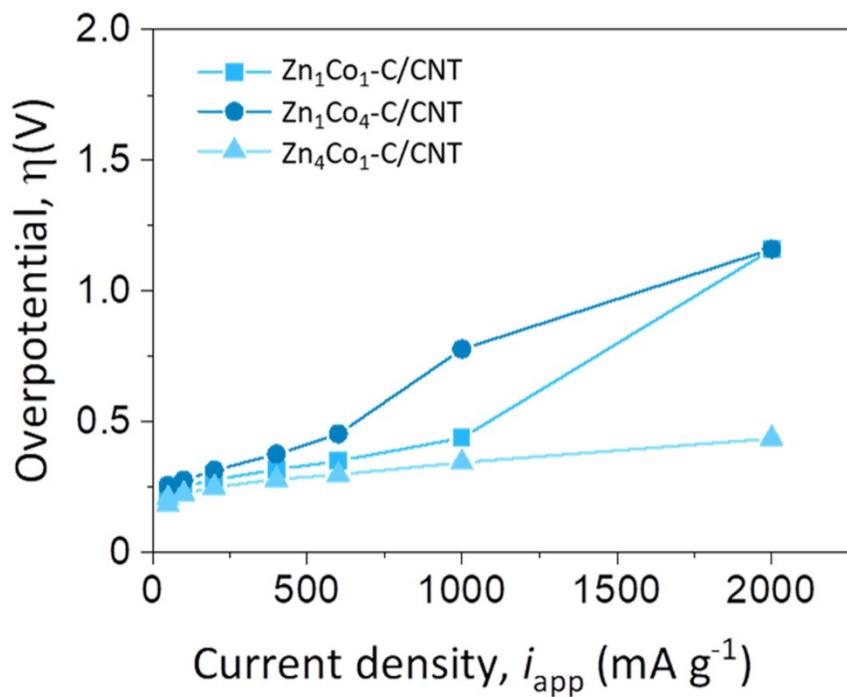
**Figure S2:** EDS elemental mapping results for (a)  $\text{Zn}_4\text{Co}_1$ -C/CNT, (b)  $\text{Zn}_1\text{Co}_1$ -C/CNT, and (c)  $\text{Zn}_1\text{Co}_4$ -C/CNT.



**Figure S3:** XRD patterns of as-synthesized  $Zn_xCo_y/CNT$  composites.



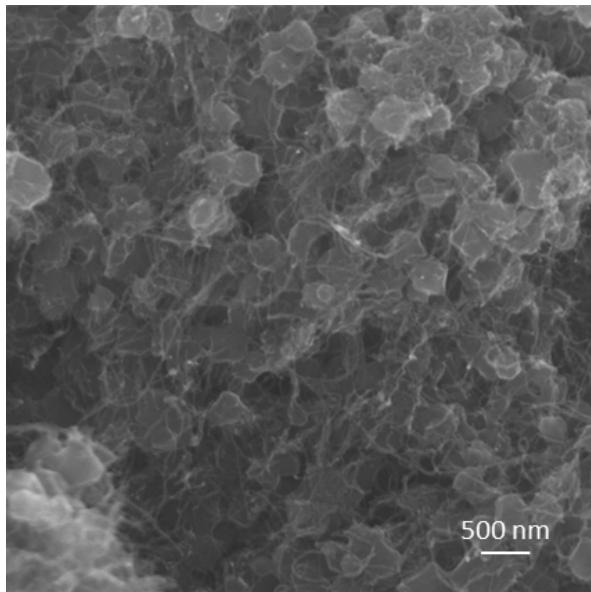
**Figure S4:** XPS spectra of  $\text{Zn}_x\text{Co}_y-\text{C}/\text{CNT}$ : (a) C 1s, and (b) N 1s.



**Figure S5:** Cathodic overpotential ( $\eta$ ) as a function of current density for LOBs with  $Zn_xCo_y$ -C/CNT electrodes. The cells were discharged in pure O<sub>2</sub> with a limited capacity of 500 mAh·g<sup>-1</sup> at various current densities in the range of 50–2,000 mA·g<sup>-1</sup>.

**Table S1:** Specific surface area and micro/meso/macropore volumes.

Material	BET surface area (m <sup>2</sup> ·g <sup>-1</sup> )	Micropore volume (cm <sup>3</sup> ·g <sup>-1</sup> )	Mesopore volume (cm <sup>3</sup> ·g <sup>-1</sup> )	Mesopore volume (cm <sup>3</sup> ·g <sup>-1</sup> )
Zn <sub>1</sub> Co <sub>4</sub> –C/CNT	305	0.04	1.23	0.05
Zn <sub>1</sub> Co <sub>1</sub> –C/CNT	357	0.03	0.95	0.05
Zn <sub>4</sub> Co <sub>1</sub> –C/CNT	489	0.07	0.88	0.04



**Figure S6:** SEM image of the Zn<sub>4</sub>Co<sub>1</sub>-C/CNT electrode (charged) after 100 cycles.

**Table S2:** Electrochemical performance of cathodes based on CNTs, MOFs, or mesoporous carbon for Li-O<sub>2</sub> batteries.

Material	Current density	Depth of discharge	Number of cycles	Reference
Multiwalled CNT paper	250 mA·g <sup>-1</sup>	1000 mAh·g <sup>-1</sup>	50	S1
Multiwalled CNT foam	372 mA·g <sup>-1</sup>	510 mAh·g <sup>-1</sup>	100	S2
Ordered mesoporous carbon	200 mA·g <sup>-1</sup>	2000 mAh·g <sup>-1</sup>	25	S3
CNT	0.05 mA·cm <sup>-2</sup>	0.5 mAh·cm <sup>-2</sup>	73	S4
Mn-MOF-74	250 mA·g <sup>-1</sup>	1000 mAh·g <sup>-1</sup>	30	S5
MOF-C/CNT	200 mA·g <sup>-1</sup>	500 mAh·g <sup>-1</sup>	137	This work

S1. Chen, Y.; Li, F.; Tang, D.-M.; Jian, Z.; Liu, C.; Golberg, D.; Yamada, A.; Zou, H. J. Mater. Chem. A **2013**, 1, 13076-13081.

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S3. Park, J. B.; Lee, J.; Yoon, C. S.; Sun, Y. K. ACS Appl. Mater. Interfaces **2013**, 5, 13426-13431.

S4. Nomura, A.; Ito, K.; Kubo, Y. Sci. Rep. **2017**, 7, 45596.

S5. Wu, D.; Guo, Z.; Yin, X.; Pang, Q.; Tu, B.; Zhang, L.; Wang, Y. G.; Li, Q. Adv. Mater. **2014**, 26, 3258-3262.